Efficient Synthesis of Dihydrobenzofurans via a Multicomponent Coupling of Salicylaldehydes, Amines, and Alkynes

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Abstract: A simple and efficient synthesis of dihydrobenzofurans was developed using a multicomponent coupling of various salicylaldehydes, amines, and alkynes. The use of aliphatic alkynes containing a heteroatom is critical to the success of the reaction.

Key words: multicomponent coupling, C–H bond reactions, tandem reactions, heterocycles, catalysis

The dihydrobenzofuran core is an important subunit of many naturally occurring heterocycles, many of which are biologically active.¹ Therefore, chemists have developed many efficient methodologies for the synthesis of 2,3-di-hydrobenzofurans derivatives, such as the radical cyclization,² the Bartoli reaction³ or the dehydration method.⁴ On the other hand, transition-metal-catalyzed cascade reactions have emerged as the method of choice to access heterocycles.⁵ These cyclizations are often carried out under mild conditions with readily synthesized starting materi-

als affording a wide variety of complex molecules. The key step to these syntheses is often the intramolecular Heck reaction.⁶ Asymmetric variations of this reaction have also been reported.⁷ Recently, we reported an annulation of phenol with dienes leading to dihydrobenzo-furans.⁸ To further our interest in such compounds, we wish to report another approach to dihydrobenzofurans via a three-component coupling using readily available components (Scheme 1).

We started our investigation using salicylaldehyde, piperidine, and 3,3-dimethylbut-1-yne in acetonitrile with 30 mol% CuI as catalyst (Scheme 2). However, only the aldehyde–amine–alkyne coupling (A³-coupling) occurred and no cyclized product was obtained after stirring at 80 °C for 16 hours.⁹

We then hypothesized that using a propargyl alcohol as the alkyne would help the cyclization. The Cu(I) species



Scheme 1 Multicomponent coupling of salicyldehyde, amine, and alkyne to generate dihydrobenzofurans



Scheme 2 Multicomponent coupling of salicylaldehyde, amine, and 3,3-dimethylbut-1-yne





Scheme 3 Proposed mechanism for the intramolecular cyclization

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Scheme 4 Multicomponent coupling of salicylaldehyde, amine, and propargyl alcohol

would coordinate to both the π -bond of the alkyne and the oxygen (Scheme 3).

Recently, Yamamoto et al. proposed a similar activation of alkynes in which a Cu(I) species coordinated to a carbonyl oxygen and a π -bond of the alkyne.¹⁰ To much of our delight, the desired cyclized product was obtained exclusively with an excellent yield (Scheme 4). Although a similar reaction is known,¹¹ a stoichiometric amount of copper is required and the scope is limited.

Optimization of the reaction conditions was then carried out (Table 1). When AuI was used as a catalyst, only the A³-addition product was obtained under the same reaction conditions (Table 1, entry 2).¹² Both AgCl and AgBr were are also effective for the cascade reaction (Table 1, entries 3 and 4).¹³ Since CuI is much more economical than AgX and AuI, further investigations were carried out with CuI. Under microwave irradiation, we were able to shorten the reaction time, reduce the amount of catalyst, and eliminate the use of solvent (Table 1, entries 6 and 7).¹⁴ We found that microwave irradiation at 130 °C for 30 minutes with 5 mol% CuI as catalyst in the absence of solvent provided optimal conditions in terms of yield and reaction time. Further attempts to shorten reaction time resulted in incomplete conversion (Table 1, entries 8 and 9).

Subsequently, various salicylaldehydes, amines, and propargyl alcohols were tested under our optimized reaction conditions (Table 2). We were pleased to find that the reaction was not limited to propargyl alcohols. Indeed, a protected propargyl amine can be used, as well as homopropargyl alcohols (Table 2, entries 3 and 4). The use of 1-hexynol also affords the cyclized product, albeit in a lower yield due to competing dimerization (Table 2, entry 5). Unfortunately, primary amines were not effective for this transformation. It should be noted that under the standard conditions (oil bath at 80 °C for 16 h), the reaction is equally effective with 5% CuI. On the other hand, the use of AgCl as a catalyst under our optimized microwave conditions led to low conversions. Hence, the reaction had to be run in an oil bath.

Table I Multicomponent Coupling	of Salicylaldehyde, Piperidine and 2-Methyl-3-butyn-2-ol
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1a	2a 3a	4					
Entry ^a	Catalyst (mol%)	Conditions	Yield (%) ^b				
1	CuI (30)	MeCN, 80 °C, 16 h	94				
2	AuI (30)	MeCN, 80 °C, 16 h	0				
3	AgBr (20)	MeCN, 80 °C, 16 h	88				
4	AgCl (20)	MeCN, 80 °C, 16 h	94				
5	CuI (30)	MW, 130 °C, 35 min, MeCN	74				
6	CuI (30)	MW, 130 °C, 35 min, neat	80				
7	CuI (5)	MW, 130 °C, 35 min, neat	90				
8	CuI (5)	MW, 130 °C, 15 min, neat	54				
9	CuI (5)	MW, 130 °C, 5 min, neat	28				

^a The reactions were performed with 1 equiv of **1a**, 2 equiv of **2a**, and equiv of **3a**.

^b NMR yield using nitromethane as an internal standard.

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Р ОН	H + N + H	cat. or ca	ICu] it. [Ag] R		
1a: R = H 1b: R = naph 1c: R = 5-Cl 1d: R = 3-ON	ithyl / le				
Entry ^a	Aldehyde	Amine	Alkyne	Isolated yield (%) ^c	
1	1a	2a	3a	85 (77)	
2	1a	2a	OH OH	81 (76)	
3	1a	2a		53 (42)	
4	1a	2a		88 (71)	
5	1a	2a	М	58 (44)	
6	1a		3a	44 (40)	
7	1a	$\left< \right>_{N}$	3a	88 (75)	
8 ^b	1b	2a	3a	80 (72)	
9 ^b	1c	2a	3a	71 (66)	
10 ^b	1d	2a	3a	82 (75)	

 Table 2
 Copper- and Silver-Catalyzed Multicomponent Coupling of Salicylaldehyde, Amine, and Alkyne

^a Reactions were run at 130 °C for 30 min in a microwave reactor using 1 equiv of aldehyde, 2 equiv of amine, and 2 equiv of alkyne under neat conditions with 5% CuI.

^b Acetonitrile was used as a solvent.

^c Value in parentheses correspond to reactions run in MeCN using 1 equiv of aldehyde, 2 equiv of amine, and 2 equiv of alkyne with 5% AgCl for 16 h in a oil bath at 80 °C.

The *Z*-isomer was obtained exclusively for all compounds (Figure 1).¹⁵ This is consistent with a 5-*exo*-dig mechanism as proposed in Scheme 3. The metal coordinates to the π -bond of the alkyne and the oxygen. Intramolecular attack then occurs, followed by protonolysis to afford the product.

In conclusion, we have developed an expedient route to dihydrobenzofurans using copper- and silver-catalyzed



Figure 1 Determination of stereochemistry by NOE experiment

multicomponent couplings under conventional heating as well as under microwave irradiations. The use of aliphatic alkynes bearing a heteroatom is crucial to the cyclization step. The scope, synthetic application, and mechanism of the method are under investigation in our laboratory.^{16–18}

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References and Notes

 For recent publications, see: (a) Van Miert, S.; Van Dyck, S.; Schmidt, T. J.; Brun, R.; Vlietinck, A.; Lemiere, G.; Pieters, L. *Bioorg. Med. Chem.* **2005**, *13*, 661. (b) Chu, G.-H.; Gu, M.; Cassel, J. A.; Belanger, S.; Graczyk, T. M.; DeHaven, R. N.; Conway-James, N.; Koblish, M.; Little, P. J.; DeHaven-Hudkins, D. L.; Dolle, R. E. *Bioorg. Med. Chem.* **2005**, *15*, 5114. (c) Shi, G. Q.; Drpinski, J. F.; Zhang, Y.; Santini, C.; Sahoo, S. P.; Berger, J. P.; MacNaul, K. L.; Zhou, G.; Agrawal, A.; Alvaro, R.; Cai, T.; Hernandez, M.; Wright, S. D.; Moller, D. E.; Heck, J. V.; Meinke, P. T. *J. Med. Chem.* **2005**, *48*, 5589.

- (2) Jimenez, M. C.; Miranda, M. A.; Tormos, R. J. Org. Chem. **1998**, *63*, 1323.
- (3) Bartoli, G.; Bosco, M.; Caretti, D.; Dalpozzo, R.; Todesco, P. E. J. Org. Chem. 1987, 52, 4381.
- (4) (a) Bertolini, F.; Crotti, P.; Di Bussolo, V.; Macchia, F.; Pineschi, M. J. Org. Chem. 2007, 72, 7761. (b) Kuethe, J. T.; Wang, A.; Journet, M.; Davies, I. W. J. Org. Chem. 2005, 70, 3727.
- (5) Nakamura, I.; Yamamoto, Y. *Chem. Rev.* **2004**, *104*, 2127.
- (7) (a) Tietze, L. F.; Ila, H.; Bell, H. P. *Chem. Rev.* 2004, 104, 3453. (b) Shibasaki, M.; Vogl, E. M.; Ohshima, T. Adv. *Synth. Catal.* 2004, 346, 1533.
- (8) Nguyen, R. V.; Yao, X.; Li, C. J. Org. Lett. 2006, 8, 2397.
- (9) For recent examples of A³-coupling, see: (a) Sreedhar, B.; Reddy, P. S.; Krishna, C. S. V.; Babu, P. V. *Tetrahedron Lett.* 2007, 48, 7882. (b) Kidwai, M.; Bansal, V.; Mishra, N. K.; Kumar, A.; Mozumdar, S. *Synlett* 2007, 1581. (c) Li, P.; Wang, L. *Tetrahedron* 2007, 63, 5455. (d) Gommermann, N.; Knochel, P. *Chem. Eur. J.* 2006, 12, 4380. (e) Wei, C.; Mague, J. T.; Li, C. J. *Proc. Natl. Acad. Sci. U.S.A.* 2004, 101, 5749. For a review of the A³-coupling see: (f) Wei, C.; Li, Z.; Li, C. J. *Synlett* 2004, 1472.
- (10) Patil, N. T.; Wu, H.; Yamamoto, Y. J. Org. Chem. 2005, 70, 4531.
- (11) Ukhin, L. Y.; Komissarov, V. N.; Lendeman, S. V.; Khrustalyov, V. N.; Struchkov, Y. T. *Chem. Heterocycl. Compd.* **2002**, *38*, 1174.
- (12) (a) Yan, B.; Liu, Y. Org. Lett. 2007, 9, 4323. (b) Kidwai, M.; Bansal, V.; Kumar, A.; Mozumdar, S. Green Chem. 2007, 9, 742. (c) Lo, V. K. Y.; Liu, Y.; Wong, M. K.; Che, C. M. Org. Lett. 2006, 8, 1529. (d) Wei, C.; Li, C. J. J. Am. Chem. Soc. 2003, 125, 9584.
- (13) (a) Reddy, K. M.; Babu, N. S.; Suryanarayana, I.; Sai Prasad, P. S.; Lingaiah, N. *Tetrahedron Lett.* **2006**, *47*, 7563.
 (b) Yan, W.; Wang, R.; Xu, Z.; Xu, J.; Lin, L.; Shen, Z.; Zhou, Y. *J. Mol. Catal. A: Chem.* **2006**, *255*, 81. (c) Li, Z.; Wei, C.; Chen, L.; Varma, R. S.; Li, C. J. *Tetrahedron Lett.* **2004**, *45*, 2443. (d) Wei, C.; Li, Z.; Li, C. J. *Org. Lett.* **2003**, *5*, 4473.
- (14) For microwave-promoted A³-coupling, see: (a) Ju, Y.; Li,
 C. J.; Varma, R. S. *QSAR Comb. Sci.* 2004, *23*, 891.
 (b) Shi, L.; Tu, Y. Q.; Wang, M.; Zhang, F. M.; Fan, C. A. *Org. Lett.* 2004, *6*, 1001.
- (15) (a) Harkat, H.; Blanc, A.; Weibel, J. M.; Pale, P. J. Org. Chem. 2008, 73, 1620. (b) Liu, Y.; Song, F.; Song, Z.; Liu, M.; Yan, B. Org. Lett. 2005, 7, 5409. (c) Chaudhuri, G.; Kundu, N. G. J. Chem. Soc., Perkin Trans. 1 2000, 775. (d) Gabriele, B.; Salerno, G.; Lauria, E. J. Org. Chem. 1999, 64, 7687.
- (16) Typical Experimental Procedure for the Cu-Catalyzed Coupling of Aldehyde, Amine, and Alkyne under Microwave Irradiation

In a 5 mL Biotage microwave vial, salicylaldehyde ($209 \mu L$, 2 mmol), morpholine ($348 \mu L$, 4 mmol), 2-methyl-3-butyn-2-ol ($387 \mu L$, 4 mmol), and CuI (19 mg, 0.10 mmol) were mixed. The vial was then capped and was subjected to microwave irradiation for 30 min at 130 °C. The crude mixture was directly isolated via column chromatography on

- (17) Typical Experimental Procedure for the Ag-Catalyzed Coupling of Aldehyde, Amine, and Alkyne under Normal Heating Conditions
 Salicylaldehyde (209 μL, 2 mmol), morpholine (348 μL, 4 mmol), 2-methyl-3-butyn-2-ol (387 μL, 4 mmol), and AgCl (14.3 mg, 0.10 mmol) were mixed in a reaction flask at 80 °C in a oil bath in MeCN overnight. The solvent was removed under reduced pressure, and the crude mixture was isolated via column chromatography on SiO₂ (gradient eluent: hexane–EtOAc = 10:1 to 1:1) to give 220 mg (44% yield) of product (Table 2, entry 6).
 (18) 2-Methyl-1-(3-piperidin-1-yl-3H-benzofuran-2
 - ylidene)propan-2-ol (Table 2, Entry 1) IR (neat): 3419 (br), 2933, 1696, 1463, 1158 cm⁻¹. ¹H NMR (400 MHz, CDCl₃): δ = 7.36 (m, 1 H), 7.24 (m, 1 H), 6.99 (m, 1 H) 5.13 (s, 1 H), 4.79 (s, 1 H), 2.58 (m, 2 H), 2.40 (m, 2 H), 1.53–1.39 (m, 12 H) ppm. ¹³C NMR (75 MHz, CDCl₃): δ = 157.7, 152.2, 129.5, 126.3, 125.4, 122.2, 113.8, 109.9, 70.6, 67.8, 49.6, 30.9, 30.8, 26.5, 24.6 ppm. MS (EI): *m/z* (%) = 223 [M⁺], 214, 189. HRMS: *m/z* calcd for C₁₇H₂₃O₂N: 273.1729: found: 273.17.

1-(3-Piperidin-1-yl-3*H*-benzofuran-2-ylidenemethyl)cyclohexanol (Table 2, Entry 2)

IR (neat): 3419 (br), 2948, 1698, 1612, 1464, 1289 cm^{-1.} ¹H NMR (400 MHz, CDCl₃): δ = 7.33 (m, 1 H), 7.17 (m, 1 H), 6.91 (m, 1 H), 6.85 (m, 1 H), 5.08 (s, 1 H), 4.74 (s, 1 H), 2.53 (m, 2 H), 2.32 (m, 2 H), 1.85–1.36 (m, 16 H) ppm. ¹³C NMR (100 MHz, CDCl₃): δ = 157.6, 152.9, 129.4, 126.2, 125.2, 122.1, 112.4, 109.9, 71.7, 67.8, 49.6, 30.0, 26.5, 25.8, 24.6, 23.0 ppm. MS (EI): *m/z* (%) = 313 [M⁺], 229, 214. HRMS: *m/z* calcd for C₂₀H₂₇O₂N: 313.2036; found: 313.2042. *N*-[1-Ethyl-1-(3-piperidin-1-yl-3H-benzofuran-2-

ylidenemethyl)propyl]-4-methylbenzenesulfonamide (Table 2, Entry 3)

IR (neat): 2970, 2802, 1696, 1612, 1594, 1461, 1157, 1094 cm⁻¹. ¹H NMR (400 MHz, CDCl₃): δ = 7.64 (m, 2 H), 7.25 (m, 2 H), 6.92 (m, 2 H), 6.92 (m, 4 H), 5.26 (s, 1 H), 4.53 (s, 1 H), 4.19 (s, 1 H), 2.39 (m, 1 H), 2.27 (m, 1 H), 2.19 (s, 3 H), 2.03 (m, 3 H), 1.74 (m, 3 H), 1.34 (m, 6 H), 0.84 (m, 6 H) ppm. ¹³C NMR (75 MHz, CDCl₃): δ = 157.5, 153.6, 142.5, 139.9, 129.6, 129.0, 127.4, 126.0, 125.3, 122.2, 110.0, 108.0, 67.8, 62.7, 49.6, 30.3, 26.3, 24.5, 21.6, 8.4 ppm. MS (EI): *m/z* (%) = 454 [M⁺], 283, 214, 155. HRMS: *m/z* calcd for C₂₆H₃₄O₃N₂S: 454.2290; found: 454.2282. **4-(3-Piperidin-1-yl-3H-benzofuran-2-ylidene)butan-2-ol** (**Table 2, Entry 4**)

IR (neat): 3382 (br), 2931, 1701, 1612, 1475, 1218, 1151 cm⁻¹. ¹H NMR (400 MHz, CDCl₃): $\delta = 7.36$ (m, 1 H), 7.20 (m, 1 H), 6.94 (m, 1 H), 6.80 (m, 1 H), 4.96 (m, 1 H), 4.78 (s, 1 H), 3.93 (sept, 1 H, *J* = 6.0 Hz), 2.39 (m, 6 H), 1.53–1.38 (m, 9 H) ppm. ¹³C NMR (75 MHz, CDCl₃): $\delta = 158.1$, 154.9, 129.5, 126.4, 125.9, 121.7, 109.8, 101.7, 68.1, 67.2, 49.6, 35.4, 26.4, 24.6, 23.1 ppm. MS (EI): *m/z* (%) = 273 [M⁺], 214, 188. HRMS: *m/z* calcd for C₁₇H₂₃O₂N: 273.1729; found: 273.1723.

5-(3-Diethylamino-3*H*-benzofuran-2-ylidene)pentan-1ol (Table 2, Entry 5)

IR (neat): 3412 (br), 2935, 1711, 1611, 1463, 1234 cm⁻¹. ¹H NMR (400 MHz, CDCl₃): δ = 7.27 (m, 1 H), 7.11 (m, 1 H), 6.84 (m, 1 H), 6.74 (m, 1 H), 4.78 (t, 1 H, *J* = 6.4 Hz), 4.62 (s, 1 H), 3.46 (m, 2 H), 2.45–2.17 (m, 4 H), 1.56–1.26 (m, 12 H) ppm. ¹³C NMR (75 MHz, CDCl₃): δ = 158.2, 152.6, 129.4, 126.4, 125.7, 121.5, 109.5, 66.8, 61.9, 49.2, 35.3, 32.0, 29.8, 25.3, 20.1 ppm. MS (EI): *m/z* (%) = 287 [M⁺], 214, 202. HRMS: *m/z* calcd for C₁₈H₂₅O₂N: 287.1872; found: 287.1875.

2-Methyl-1-(3-morpholin-4-yl-3*H*-benzofuran-2-ylidene)propan-2-ol (Table 2, Entry 6)

IR (neat): 3419 (br), 2967, 1612, 1462, 1374, 1232, 1115 cm⁻¹. ¹H NMR (400 MHz; CDCl₃): δ = 7.41 (m, 1 H), 7.26 (m, 1 H), 7.01 (m, 1 H), 6.94 (m, 1 H), 5.15 (s, 1 H), 4.79 (s, 1 H), 3.66 (m, 4 H), 2.63 (m, 2 H), 2.41 (m, 2 H), 1.50 (s, 3 H), 1.48 (s, 3 H) ppm. ¹³C NMR (75 MHz CDCl₃): δ = 157.7, 151.4, 136.1, 129.9, 126.3, 122.4, 114.5, 110.1, 70.6, 67.4, 48.6, 30.8, 30.8 ppm. MS (EI): *m/z* (%) = 275 [M⁺], 216, 189, 131, 86. HRMS: *m/z* calcd for C₁₆H₂₁O₃N: 275.1521; found: 275.1519.

2-Methyl-1-(3-pyrrolidin-1-yl-3*H*-benzofuran-2ylidene)propan-2-ol (Table 2, Entry 7)

IR (neat): 3430 (br), 2911, 1672, 1561, 1341, 1160 cm⁻¹. ¹H NMR (400 MHz, CDCl₃): δ = 7.35 (m, 1 H), 7.19 (m, 1 H), 6.93 (m, 1 H), 6.87 (m, 1 H), 5.12 (s, 1 H), 4.74 (s, 1 H), 2.57 (m, 2 H), 2.36 (m, 2 H), 1.54–1.24 (m, 10 H) ppm. ¹³C NMR (75 MHz, CDCl₃): δ = 157.6, 152.2, 129.5, 125.4, 122.2, 113.7, 109.9, 70.6, 67.8, 49.6, 30.9, 26.5, 24.6 ppm. MS (EI): *m/z* (%) = 258 [M⁺], 189, 131. HRMS: *m/z* calcd for C₁₆H₂₁O₂N: 258.1494; found: 258.1491.

2-Methyl-1-{3-piperidin-1-yl-3*H*-naphtho[2,3-*b*]furan-2-ylidene}propan-2-ol (Table 2, Entry 8)

IR (neat): 3581 (br), 2934, 1688, 1522, 1249, 1161 cm⁻¹. ¹H NMR (400 MHz CDCl₃): δ = 8.18 (m, 1 H), 7.83 (m, 1 H), 7.75 (m, 1 H), 7.52 (m, 1 H), 7.38 (m, 1 H), 2.53 (m, 4 H), 1.64–1.32 (m, 12 H) ppm. ¹³C NMR (75 MHz, CDCl₃): δ = 155.2, 151.2, 131.2, 130.8, 130.5, 128.8, 127.9, 124.0, 124.0, 117.9, 115.1, 111.4, 70.9, 68.2, 49.3, 31.2, 31.1, 26.8,

25.0 ppm. MS (EI): m/z (%) = 323 [M⁺], 264, 181. HRMS:

m/z calcd for C₂₁H₂₅O₂N: 323.1885; found: 323.1879. 1-(5-Chloro-3-piperidin-1-yl-3H-benzofuran-2-ylidene)-2-methylpropan-2-ol (Table 2, Entry 9) IR (neat): 3401 (br), 2936, 1702, 1608, 1469, 1234, 1154 cm⁻¹. ¹H NMR (400 MHz, CDCl₃): δ = 7.34 (s, 1 H), 7.20 (m, 1 H), 6.86 (m, 1 H), 5.14 (m, 1 H), 4.76 (s, 1 H), 2.58 (m, 2 H), 2.39 (m, 2 H), 1.54–1.46 (m, 12 H) ppm. ¹³C NMR (75 MHz, CDCl₃): δ = 156.2, 152.1, 129.5, 127.4, 126.3, 114.2, 110.9, 70.6, 69.8, 67.8, 49.6, 30.9, 26.5, 24.5 ppm. MS (EI): m/z (%) = 307 [M⁺], 248, 223. HRMS: m/z calcd for C₁₇H₂₂O₂N³⁵Cl: 307.1339; found: 307.1344. HRMS: *m/z* calcd for C₁₇H₂₂O₂N³⁷Cl: 309.1312; found: 309.1312. 1-(6-Methoxy-3-piperidin-1-yl-3H-benzofuran-2ylidene)-2-methylpropan-2-ol (Table 2, Entry 10) IR (neat): 3580 (br), 2938, 1736, 1699, 1620, 1444, 1280 cm⁻¹. ¹H NMR (400 MHz, CDCl₃): δ = 7.22 (m, 1 H), 6.50 (m, 1 H), 5.09 (m, 1 H), 4.67 (s, 1 H), 3.75 (s, 3 H), 2.53 (m, 2 H), 2.31 (m, 2 H), 1.49–1.36 (m, 12 H) ppm. ¹³C NMR (75 MHz, CDCl₃): δ = 161.2, 158.9, 153.0, 126.5, 117.2, 113.8, 108.1, 96.1, 70.6, 67.4, 55.7, 49.4, 30.9, 30.8, 26.4, 24.6 ppm. MS (EI): *m/z* (%) = 303 [M⁺], 244, 219. HRMS: *m/z* calcd for C₁₈H₂₅O₃N: 303.1834; found: 303.1831. 2-(4,4-Dimethyl-1-piperidin-1-yl-pent-2-ynyl)phenol (1b) IR (neat): 2936, 2233, 1644, 1465, 1458, 1362, 1244, 1153 cm⁻¹. ¹H NMR (400 MHz, CDCl₃): δ = 7.51 (m, 1 H), 7.22 (m, 1 H), 6.87 (m, 2 H), 4.88 (s, 1 H), 2.64 (m, 4 H), 1.68 (m, 6 H), 1.39 (s, 9 H) ppm. ¹³C NMR (75 MHz, CDCl₃): δ =

158.0, 129.3, 128.6, 122.1, 110.0, 116.4, 99.1, 95.41, 71.3, 60.7, 31.6, 28.0, 26.2, 24.4 ppm. MS (EI): m/z (%) = 271 [M⁺], 186, 84. HRMS: m/z calcd for C₁₈H₂₅ON: 271.1936; found: 271.1933.

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