Ruthenium-Catalyzed Enantioselective Propargylation of Indoles with Propargylic Alcohols

Keiichiro Kanao, Hiroshi Matsuzawa, Yoshihiro Miyake, Yoshiaki Nishibayashi*

Institute of Engineering Innovation, School of Engineering, The University of Tokyo, Yayoi, Bunkyo-ku, Tokyo 113-8656, Japan Fax +81(3)58411175; E-mail: ynishiba@sogo.t.u-tokyo.ac.jp *Received 28 July 2008; revised 7 August 2008*



Abstract: Ruthenium-catalyzed enantioselective propargylation of 1-(triisopropylsilyl)-1*H*-indoles with propargylic alcohols gives the corresponding β -propargylated indoles in good yields with high enantioselectivity. Reactions with 1-(1-naphthyl)prop-2-yn-1-ol achieve the highest enantioselectivity (up to 95% ee).

Key words: asymmetric synthesis, Friedel–Crafts alkylation, ruthenium, indoles, propargylic alcohols



Scheme 1

Indoles represent a structural motif in a number of natural bioactive products. A variety of methods to obtain optically active indoles have been reported by Lewis acid and Brønsted acid catalyzed enantioselective Friedel–Crafts alkylation of indoles.^{1,2} We have recently disclosed the enantioselective propargylation of aromatic compounds such as 2-alkylfurans and *N*,*N*-dimethylaniline with propargylic alcohols catalyzed by a chiral thiolate-bridged diruthenium complex³ to afford the corresponding propar-

gylated products in good yields with high enantioselectivity (up to 94% ee).⁴ This is the first example of the enantioselective propargylation of aromatic compounds. As an extension of our study, we have more recently found the ruthenium-catalyzed enantioselective propargylation of indoles with propargylic alcohols to give the corresponding β -propargylated indoles in good to high yields (Scheme 1).⁵ In this reaction system, the introduction of a bulky group, such as the triisopropylsilyl (TIPS) moiety,



Scheme 2 The remarkable effect of the N-substituent on the enantioselective propargylation of indoles

SYNTHESIS 2008, No. 23, pp 3869–3873 Advanced online publication: 06.11.2008 DOI: 10.1055/s-0028-1083217; Art ID: Z17208SS © Georg Thieme Verlag Stuttgart · New York

Ar OH 1	5 mol% [Cp*RuCl(µ ₂ -SR*)] ₂ 10 mol% NH ₄ BF ₄ CICH ₂ CH ₂ Cl TIPS 40 °C	$Ar + H_2O$ 3 TIPS			
Entry	Ar of 1	Time (h)	Yield ^b (%)	ee ^c (%)	
1	Ph (1a)	7	3a , 77	78	
2	$4-MeC_{6}H_{4}(\mathbf{1b})$	10	3b , 70	71	
3	$4-ClC_{6}H_{4}(1c)$	7	3c , 72	79	
4	$4-PhC_{6}H_{4}(1d)$	10	3d , 76	90	
5	2-PhC ₆ H ₄ (1e)	30	3e , 63	83	
6	$3,5-Ph_2C_6H_3$ (1f)	7	3f , 98	80	
7	1-naphthyl (1g)	23	3g , 81	92	
8	2-naphthyl (1h)	7	3h , 82	84	

 Table 1
 Ruthenium-Catalyzed Enantioselective Propargylation of 1-(Triisopropylsilyl)-1H-indole with Propargylic Alcohols 1^a

^a Reaction conditions: **1a**–**h** (0.20 mmol), 1-(triisopropylsilyl)-1*H*-indole (0.60 mmol), ruthenium complex (0.010 mmol, generated in situ from $[Cp*RuCl]_4$ and **2**), NH₄BF₄ (0.020 mmol), DCE (5 mL).

^b Isolated yield of **3**.

^c Determined by HPLC.

at the nitrogen of indole dramatically increased the enantioselectivity of the propargylated indoles (Scheme 2).⁵ Herein, we describe a practical method for the preparation of β -propargylated indoles from reactions of 1-(triisopropylsilyl)-1*H*-indole with propargylic alcohols catalyzed by a chiral thiolate-bridged diruthenium complex. As described in our previous paper, we have already found that the highest enantioselectivity was observed when 1-(1-naphthyl)prop-2-yn-1-ol (**1g**) was used as a substrate.⁵ Typical results are shown in Table 1. In fact, the reaction of 1-(1-naphthyl)prop-2-yn-1-ol (**1g**) with 1-(triisopropylsilyl)-1*H*-indole (3 equiv) in 1,2-dichloroethane in the presence of a catalytic amount of a chiral thi-



Scheme 3 A large-scale reaction of 1-(1-naphthyl)prop-2-yn-1-ol (1g) with 1-(triisopropylsilyl)-1H-indole

Synthesis 2008, No. 23, 3869-3873 © Thieme Stuttgart · New York

olate-bridged diruthenium complex, which was prepared in situ from the tetranuclear ruthenium(II) complex $[Cp*RuCl]_4$ and chiral disulfide 2^3 in tetrahydrofuran at room temperature for 12 hours, and ammonium tetrafluoroborate at 40 °C for 23 hours afforded 3-[1-(1-naphthyl)prop-2-ynyl]-1-(triisopropylsilyl)-1*H*-indole (**3g**) in 72% isolated yield with 93% ee (*R*)⁵ (Scheme 3). After recrystallization of the deprotected indole, enantiomerically pure β -propargylated indole **4** was obtained in 55% isolated yield.

Next, the propargylation of other 1-(triisopropylsilyl)-1Hindoles bearing a substituent in the 5- or 6-position of the indole ring with **1g** was carried out under the same reaction conditions. Typical results are shown in Table 2. The introduction of a methyl group at the 5- or 6-position of the indole ring gave a similar enantioselectivity (Table 2, entries 2 and 3). The same enantioselectivity was observed when a chloro moiety is presented at 5-position of the indole ring (Table 2, entry 4). On the other hand, the introduction of methoxy or fluoro moiety at the 5- or 6position of the indole ring gave a slightly lower enantioselectivity (Table 2, entries 5–7).

In summary, we have developed an efficient and practical method for the preparation of β -propargylated indoles from reactions of 1-(triisopropylsilyl)-1*H*-indole with propargylic alcohols catalyzed by a chiral thiolate-bridged diruthenium complex. This method provides a novel protocol for the asymmetric Friedel–Crafts alkyla-

Table 2Ruthenium-Catalyzed Enantioselective Propargylation of 1-(Triisopropylsilyl)-1H-indoles with 1g^a



^a Reaction conditions: **1g** (0.20 mmol), 1-(triisopropylsilyl)-1*H*-indoles (0.60 mmol), ruthenium complex (0.010 mmol, generated in situ from $[Cp*RuCl]_4$ and **2**), NH₄BF₄ (0.020 mmol), DCE (5 mL).

^b Isolated yield of **3**.

^c Determined by HPLC.

Synthesis 2008, No. 23, 3869-3873 © Thieme Stuttgart · New York

tion of indoles by using propargylic alcohols as a new type of electrophile.

¹H NMR (270 MHz) and ¹³C NMR (67.8 MHz) spectra were measured on a Jeol Excalibur 270 spectrometer using CDCl₃ as solvent. HPLC analyses were performed on a Hitachi L-7100 apparatus equipped with a UV detector using 25 cm \times 4.6 mm Daicel Chiralcel OD and Chiralpak IA columns. Mass spectra were measured on a Jeol JMS-700 mass spectrometer.

All reactions were carried out under a dry N₂ atmosphere. Chiral disulfide **2** was prepared according to our previous procedure.³ 1-Phenylprop-2-yn-1-ol (**1a**) is commercially available. Preparation of other propargylic alcohols was carried out according to literature methods.⁶ 1-(Triisopropylsilyl)-1*H*-indole was prepared according to the literature method.⁷ DCE and THF were distilled under N₂ over P₂O₅ and Na benzophenone ketyl, respectively, and degassed. All products were fully characterized.⁵

3-[1-(1-Naphthyl)prop-2-ynyl]-1-(triisopropylsilyl)-1*H*-indole (3g); Typical Procedure

In a 200-mL round-bottomed flask were placed [Cp*RuCl]₄ (55.3 mg, 0.05 mmol) and bis[(*R*)-1-(6'-phenyl-1,1':4',1"-terphenyl-2'yl)propyl] disulfide (**2**, 76.4 mg, 0.10 mmol) under N₂. Anhyd THF (10 mL) was added and the mixture was magnetically stirred at r.t. for 12 h. The solvent was evaporated in vacuo. Then, NH₄BF₄ (21.8 mg, 0.20 mmol) and anhyd DCE (50 mL) were added under N₂, and the mixture was magnetically stirred at r.t. After the addition of **1g** (365 mg, 2.0 mmol) and 1-(triisopropylsilyl)-1*H*-indole (1.64 g, 6.0 mmol), the reaction flask was kept at 40 °C for 23 h. The solvent was concentrated under reduced pressure by an aspirator, and then the residue was purified by flash column chromatography (silica gel, hexane–EtOAc; hexane only to 100:1) to give **3g** (634 mg, 72%) as a pale yellow oil; 93% ee [HPLC (Daicel Chiralpak IA, hexane–*i*-PrOH, 99:1, flow rate = 0.5 mL/min, λ = 254 nm): *t*_R = 11.6 (major), 14.0 min (minor)].

3-[1-(1-Naphthyl)prop-2-ynyl]-1H-indole (4)

In a 20-mL round-bottomed flask was placed **3g** (634 mg, 1.45 mmol) under N₂. Anhyd THF (7.5 mL) was added and the mixture was cooled to 0 °C. Then, 1 M TBAF in THF (2.9 mL, 2.9 mmol) was added dropwise to the mixture and the mixture was magnetically stirred at 0 °C for 1 h. CH₂Cl₂ (20 mL) was added and the organic layer was washed with H₂O (3 × 20 mL). The aqueous layer was extracted with CH₂Cl₂ (3 × 10 mL) and the combined organic layers were dried (anhyd MgSO₄). The solvent was concentrated under reduced pressure by an aspirator, and the residue was purified by flash column chromatography (silica gel, hexane–EtOAc, 10:1) to give 4⁵ (360 mg, 88%) as a white solid; 93% ee. The product was further purified by recrystallization (hexane–EtOAc) to give enantiomerically pure 4 (223 mg, 55%); >99% ee [HPLC (Daicel Chiralpak IA, hexane–*i*-PrOH, 9:1, flow rate = 1.0 mL/min, l = 254 nm): $t_{\rm R}$ = 15.1 (major), 16.5 min (minor)].

 $[\alpha]_{D}^{27}$ –6.3 (*c* 1.10, CHCl₃).

6-Methyl-3-[1-(1-naphthyl)prop-2-ynyl]-1-(triisopropylsilyl)-1H-indole (3j)

Brown oil; yield: 83%; 92% ee [HPLC (Daicel Chiralpak OD, hexane–*i*-PrOH, 49:1, flow rate = 0.5 mL/min, λ = 254 nm): $t_{\rm R}$ = 13.7 (minor), 18.0 min (major)].

¹H NMR (270 MHz, CDCl₃): δ = 1.08 (dd, *J* = 5.1, 7.6 Hz, 18 H), 1.55–1.66 (m, 3 H), 2.42–2.44 (br, 4 H), 5.89 (s, 1 H), 6.84 (d, *J* = 7.9 Hz, 1 H), 7.02 (s, 1 H), 7.19–7.44 (m, 5 H), 7.56 (d, *J* = 7.1 Hz, 1 H), 7.75 (d, *J* = 8.2 Hz, 1 H), 7.82–7.86 (m, 1 H), 8.22–8.26 (m, 1 H). ¹³C NMR (67.8 MHz, CDCl₃): δ = 12.8, 18.0, 22.0, 32.0, 71.5, 84.7, 114.0, 116.6, 118.9, 121.2, 124.1, 125.4, 125.5, 125.6, 125.8, 127.4, 127.8, 128.7, 130.0, 131.1, 131.2, 134.0, 135.9, 142.2.

HRMS (EI): m/z [M] calcd for C₃₁H₃₇NSi: 451.2695; found: 451.2688.

5-Fluoro-3-[1-(1-naphthyl)prop-2-ynyl]-1-(triisopropylsilyl)-1*H*-indole (3m)

Brown oil; yield: 51%; 80% ee [HPLC (Daicel Chiralpak OD, hexane–*i*-PrOH, 49:1, flow rate = 0.5 mL/min, λ = 254 nm): $t_{\rm R}$ = 12.4 (minor), 16.9 min (major)].

¹H NMR (270 MHz, CDCl₃): δ = 1.06 (dd, *J* = 4.9, 7.5 Hz, 18 H), 1.52–1.63 (m, 3 H), 2.47 (d, *J* = 2.4 Hz, 1 H), 5.85 (d, *J* = 2.4 Hz, 1 H), 6.85 (dt, *J* = 2.6, 9.1 Hz, 1 H), 7.07–7.12 (m, 2 H), 7.34–7.47 (m, 4 H), 7.59 (d, *J* = 6.9 Hz, 1 H), 7.78 (d, *J* = 8.2 Hz, 1 H), 7.84–7.87 (m, 1 H), 8.18–8.22 (m, 1 H).

¹³C NMR (67.8 MHz, CDCl₃): δ = 12.7, 17.9, 32.1, 71.8, 84.3, 104.4 (d, $J_{C-F} = 24$ Hz), 109.8 (d, $J_{C-F} = 26$ Hz), 114.5 (d, $J_{C-F} = 10$ Hz), 116.9 (d, $J_{C-F} = 5$ Hz), 124.0, 125.4, 125.5, 125.8, 125.9, 128.1, 128.8, 130.1 (d, $J_{C-F} = 10$ Hz), 131.1, 132.3, 134.1, 135.4, 138.1, 157.6 (d, $J_{C-F} = 235$ Hz).

HRMS (EI): m/z [M] calcd for C₃₀H₃₄FNSi: 455.2445; found: 455.2460.

6-Fluoro-3-[1-(1-naphthyl)prop-2-ynyl]-1-(triisopropylsilyl)-1*H*-indole (3n)

Yellow oil; yield: 78%; 81% ee [HPLC (Daicel Chiralpak OD, hexane–*i*-PrOH, 49:1, flow rate = 0.5 mL/min, λ = 254 nm): $t_{\rm R}$ = 12.9 (minor), 18.6 min (major)].

¹H NMR (270 MHz, CDCl₃): δ = 1.07 (dd, *J* = 5.1, 7.6 Hz, 18 H), 1.52–1.64 (m, 3 H), 2.45 (d, *J* = 2.5 Hz, 1 H), 5.88 (d, *J* = 2.5 Hz, 1 H), 6.78 (dt, *J* = 2.2, 9.1 Hz, 1 H), 7.03 (s, 1 H), 7.15 (dd, *J* = 2.2, 11.0 Hz, 1 H), 7.29–7.46 (m, 4 H), 7.56 (d, *J* = 7.1 Hz, 1 H), 7.77 (d, *J* = 8.2 Hz, 1 H), 7.84–7.87 (m, 1 H), 8.20–8.23 (m, 1 H).

¹³C NMR (67.8 MHz, CDCl₃): δ = 12.6, 18.0, 32.0, 71.7, 84.4, 100.4 (d, $J_{C-F} = 26$ Hz), 108.2 (d, $J_{C-F} = 24$ Hz), 116.9, 119.8 (d, $J_{C-F} = 10$ Hz), 124.0, 125.4, 125.5, 125.8, 125.9, 126.1, 128.0, 128.8, 130.8 (d, $J_{C-F} = 4$ Hz), 131.1, 134.0, 135.6, 141.7 (d, $J_{C-F} = 12$ Hz), 159.7 (d, $J_{C-F} = 237$ Hz).

HRMS (EI): m/z [M] calcd for C₃₀H₃₄FNSi: 455.2445; found: 455.2451.

Acknowledgment

This work was supported by Grants-in-Aid for Scientific Research for Young Scientist (S) (No. 19675002) and for Scientific Research on Priority Areas (No. 18066003) from the Ministry of Education, Culture, Sports, Science and Technology, Japan. K.K. acknowledges the Global COE Program for Chemistry Innovation.

References

- For recent reviews, see: (a) Jørgensen, K. A. Synthesis 2003, 1117. (b) Bandini, M.; Melloni, A.; Umani-Ronchi, A. Angew. Chem. Int. Ed. 2004, 43, 550. (c) Bandini, M.; Emer, E.; Tommasi, S.; Umani-Ronchi, A. Eur. J. Org. Chem. 2006, 3527. (d) Poulsen, T. B.; Jørgensen, K. A. Chem. Rev. 2008, 108, 2903.
- (2) For recent examples of asymmetric Friedel–Crafts alkylation of indole derivatives, see: (a) Li, C.-F.; Liu, H.; Liao, J.; Cao, Y.-J.; Liu, X.-P.; Xiao, W.-J. Org. Lett. 2007, 9, 1847. (b) Yang, H.; Hong, Y.-T.; Kim, S. Org. Lett. 2007, 9, 2281. (c) Terada, M.; Sorimachi, K. J. Am. Chem. Soc. 2007, 129, 292. (d) Kang, Q.; Zhao, Z.-A.; You, S.-L. J. Am.

Chem. Soc. 2007, 129, 1484. (e) Dong, H.-M.; Lu, H.-H.; Lu, L.-Q.; Chen, C.-B.; Xiao, W.-J. Adv. Synth. Catal. 2007, 349, 1597. (f) Terada, M.; Yokoyama, S.; Sorimachi, K.; Uraguchi, D. Adv. Synth. Catal. 2007, 349, 1863. (g) Jia, Y.-X.; Zhong, J.; Zhu, S.-F.; Zhang, C.-M.; Zhou, Q.-L. Angew. Chem. Int. Ed. 2007, 46, 5565. (h) Evans, D. A.; Fandrick, K. R.; Song, H.-J.; Scheidt, K. A.; Xu, R. J. Am. Chem. Soc. 2007, 129, 10029. (i) Itoh, J.; Fuchibe, K.; Akiyama, T. Angew. Chem. Int. Ed. 2008, 47, 4016. (j) Desimoni, G.; Faita, G.; Toscanini, M.; Boiocchi, M. Chem. Eur. J. 2008, 14, 3630. (k) Liu, H.; Lu, S.-F.; Xu, J.; Du, D.-M. Chem. Asian J. 2008, 3, 1111. (1) Zhang, G.-W.; Wang, L.; Nie, J.; Ma, J.-A. Adv. Synth. Catal. 2008, 350, 1457. (m) Wanner, M. J.; Hauwert, P.; Schoemaker, H. E.; de Gelder, R.; van Maarseveen, J. H.; Hiemstra, H. Eur. J. Org. Chem. 2008, 180. (n) Nakamura, S.; Hyodo, K.;

Nakamura, Y.; Shibata, T.; Toru, T. *Adv. Synth. Catal.* **2008**, *350*, 1443. (o) Abid, M.; Teixeira, L.; Török, B. *Org. Lett.* **2008**, *10*, 933. (p) Liu, W.-B.; He, H.; Dai, L.-X.; You, S.-L. *Org. Lett.* **2008**, *10*, 1815. (q) Yuan, Z.-L.; Lei, Z.-Y.; Shi, M. *Tetrahedron: Asymmetry* **2008**, *19*, 1339.

- (3) Inada, Y.; Nishibayashi, Y.; Uemura, S. Angew. Chem. Int. Ed. 2005, 44, 7715.
- (4) Matsuzawa, H.; Miyake, Y.; Nishibayashi, Y. Angew. Chem. Int. Ed. 2007, 46, 6488.
- (5) Matsuzawa, H.; Kanao, K.; Miyake, Y.; Nishibayashi, Y. Org. Lett. 2007, 9, 5561; and supplementary information.
- (6) Nishibayashi, Y.; Milton, M. D.; Inada, Y.; Yoshikawa, M.; Wakiji, I.; Hidai, M.; Uemura, S. *Chem. Eur. J.* 2005, *11*, 1433.
- (7) Beswick, P. J.; Greenwood, C. S.; Mowlem, T. J.; Nechvatal, G.; Widdowson, D. A. *Tetrahedron* **1988**, 44, 7325.