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### Facile and One-Pot Synthesis of 1,2-Dihydroquinazolin-4(3H)-ones via Tandem Intramolecular Pinner/Dimroth Rearrangement

Jian-Hong Tang<sup>a</sup>, Da-Xin Shi<sup>a</sup>, Li-Jun Zhang<sup>b</sup>, Qi Zhang<sup>a</sup> & Jia-Rong Li<sup>a</sup>

<sup>a</sup> School of Chemical Engineering and Environment, Beijing Institute of Technology, Beijing, China

<sup>b</sup> School of Chemistry and Chemical Engineering, Tianjin University of Technology, Tianjin, China

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## FACILE AND ONE-POT SYNTHESIS OF 1,2-DIHYDROQUINAZOLIN-4(3H)-ONES VIA TANDEM INTRAMOLECULAR PINNER/DIMROTH REARRANGEMENT

Jian-Hong Tang,<sup>1</sup> Da-Xin Shi,<sup>1</sup> Li-Jun Zhang,<sup>2</sup> Qi Zhang,<sup>1</sup> and Jia-Rong Li<sup>1</sup>

<sup>1</sup>School of Chemical Engineering and Environment, Beijing Institute of Technology, Beijing, China

<sup>2</sup>School of Chemistry and Chemical Engineering, Tianjin University of Technology, Tianjin, China

*An efficient synthetic method for the preparation of quinazolin-4-one derivatives was designed. The facile condensation of aromatic o-aminonitriles with aromatic aldehydes catalyzed by Lewis acid give 1,2-dihydroquinazolin-4(3H)-ones in moderate to good yields under refluxing dimethylformamide.*

**Keywords:** Aromatic aldehyde; aromatic o-aminonitrile; catalyst; cyclocondensation; one-pot synthesis

### INTRODUCTION

1,2-Dihydroquinazolin-4(3H)-ones are important and useful nitrogen-containing heterocycles because of their diverse biological activities. They have been widely used as antitumor agents,  $\alpha_1$ -adrenoceptors antagonists, plant growth regulators, diuretics, and herbicides.<sup>[1]</sup> Some dihydroquinazolinones also show potential applications in flat-panel display as electro-luminescence material sandwiched in organic light-emitting devices.<sup>[2]</sup> The main synthetic approaches to 1,2-dihydroquinazolin-4(3H)-ones consist of preliminary amidation of 2-amino-benzonitrile, 2-amino-benzoic acid, or its derivatives followed by oxidative ring closure under basic conditions.<sup>[3]</sup> Other methods involve cycloaddition of anthranilic acid derivatives with a diverse range of substrates including imidates and iminoaldehydes and the aza-Witting reactions of  $\alpha$ -azido-substituted aromatic amides.<sup>[4,5]</sup> However, some of these methods are associated with drawbacks such as multistep procedures, costly reagents, harsh reaction conditions, and poor yields. Thus, there is still a need to develop efficient methods for the synthesis of dihydroquinazolinones.

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Address correspondence to Jia-Rong Li, School of Chemical Engineering and Environment, Beijing Institute of Technology, Beijing 100081, China. E-mail: jrli@bit.edu.cn

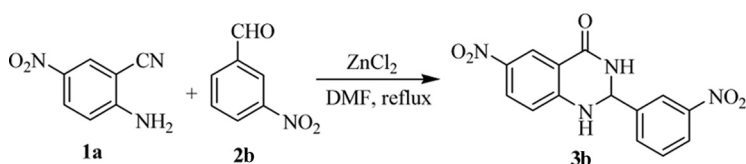
Aromatic *o*-aminonitrile is a versatile synthon for the synthesis of nitrogen-containing fused heterocycles **5**,<sup>[6]</sup> such as pyrimidine,<sup>[7]</sup> quinazoline,<sup>[8,9]</sup> quinazolinone,<sup>[10,11]</sup> and quin[3,4-*a*]carbazole.<sup>[12]</sup> In continuation of our previous work on the synthesis of a variety of nitrogen-containing heterocycles from 2-aminobenzonitriles,<sup>[13]</sup> we report herein a convenient and one-pot synthesis of 1,2-dihydroquinazolin-4(3H)-ones from the cyclocondensation of aromatic *o*-aminonitriles with aromatic aldehydes using anhydrous ZnCl<sub>2</sub> as the catalyst.

The reaction of 2-amino-5-nitrobenzonitrile **1a** with 3-nitrobenzaldehyde **2b** was chosen as a model (Scheme 1). After different conditions were screened, we were delighted to find that the 1,2-dihydroquinazolin-4(3H)-one derivative **3b** was obtained from this reaction with ZnCl<sub>2</sub> as catalyst in a suitable boiling solvent. After screening solutions, we found dimethylformamide (DMF) was the best solvent and dimethylsulfoxide (DMSO) was also suitable for this annulation, but no product was afforded in nonpolar solvents such as toluene and xylene.

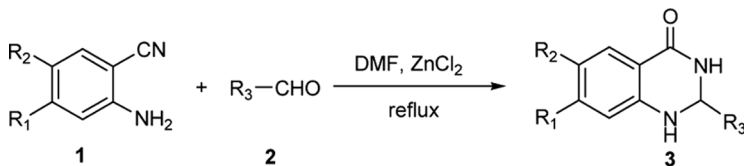
With the optimized reaction conditions in hand, various aldehydes **2**, including aromatic or heteroaromatic aldehydes with either electron-donating or electron-withdrawing substituents, were subjected to react with **1** to investigate the reaction scope, and the representative results are summarized in Table 1.

1,2-Dihydroquinazolin-4(3H)-one derivatives **3a–n** (Table 1) were formed by the reactions of aromatic *o*-aminonitriles with aldehydes in refluxed DMF in good yields. The position and electronic nature of the substituent on the phenyl ring of arylaldehydes had no relevance to quinazolinone yield. Although nitro- and methoxy- substituents have different electron effects on benzaldehydes, similar reaction rate and yield were observed. Heteroaromatic aldehydes were also readily cyclized with **1** to afford quinazolinones **3m–n** in 72–77% yield. The new conversion proceeded better when *o*-aminonitriles **1** had electron-withdrawing substituents. The structures of **3** were verified by instrument analysis, and **3l** was also further elucidated by x-ray crystallographic analysis (Fig. 1). Crystal data for **3l**: chemical formula, C<sub>14</sub>H<sub>11</sub>N<sub>3</sub>O<sub>3</sub>; formula weight, 269.26; temperature (K), 113(2); crystal system, monoclinic; space group, *Pnma*; and unit cell dimension, *a* = 1.09766(13) nm, *b* = 0.98626(9) nm, *c* = 1.17636(14) nm,  $\alpha$  = 90.00,  $\beta$  = 109.697(7)°,  $\gamma$  = 90°; *V* = 1.1990(2) nm<sup>3</sup>, *Z* = 4, and *D<sub>C</sub>* = 1.492 g cm<sup>−3</sup>.

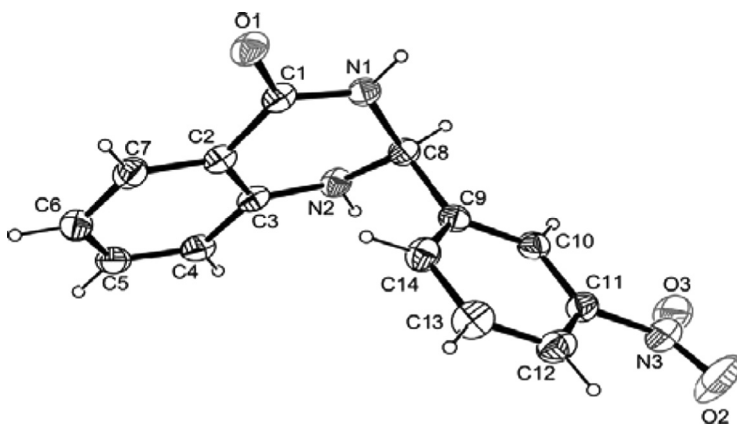
On the basis of these observations, a possible mechanism was proposed (Scheme 2). Addition of the amino group of the *o*-aminonitrile onto the carbonyl of the aldehyde gave the key intermediate **I**. The hydroxyl group of intermediate **I** then attacked the nitrile group (i.e., Pinner reaction<sup>[14]</sup>) to afford a benzoxazine **II**, which subsequently rearranged (Dimroth rearrangement<sup>[15]</sup>) to give the product **3**.

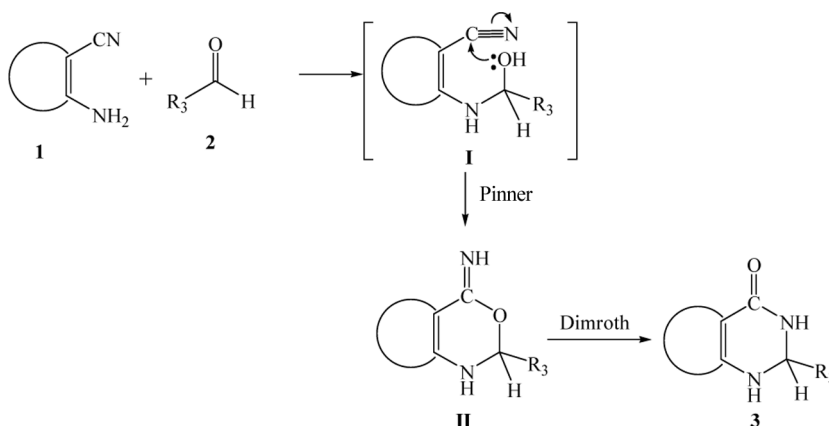


Scheme 1. Condensation of 2-amino-5-nitrobenzonitrile with 3-nitrobenzaldehyde.

**Table 1.** Results of the reaction of o-aminonitriles with various aldehydes in DMF<sup>a</sup>

Entry	Product	R <sub>1</sub>	R <sub>2</sub>	R <sub>3</sub>	Time (h)	Yield <sup>b</sup> (%)
1	<b>3a</b>	H	NO <sub>2</sub>	C <sub>6</sub> H <sub>5</sub>	1.5	72
2	<b>3b</b>	H	NO <sub>2</sub>	<i>m</i> -O <sub>2</sub> NC <sub>6</sub> H <sub>4</sub>	1	80
3	<b>3c</b>	H	NO <sub>2</sub>	<i>p</i> -CH <sub>3</sub> OC <sub>6</sub> H <sub>4</sub>	1	82
4	<b>3d</b>	H	NO <sub>2</sub>	<i>p</i> -ClC <sub>6</sub> H <sub>4</sub>	1	80
5	<b>3e</b>	H	NO <sub>2</sub>	<i>p</i> -O <sub>2</sub> NC <sub>6</sub> H <sub>4</sub>	1	77
6	<b>3f</b>	Cl	H	<i>p</i> -O <sub>2</sub> NC <sub>6</sub> H <sub>4</sub>	2	72
7	<b>3g</b>	Cl	H	<i>p</i> -CH <sub>3</sub> OC <sub>6</sub> H <sub>4</sub>	2	74
8	<b>3h</b>	H	H	<i>p</i> -O <sub>2</sub> NC <sub>6</sub> H <sub>4</sub>	2.5	67
9	<b>3i</b>	H	H	<i>o</i> -HOC <sub>6</sub> H <sub>4</sub>	2.5	72
10	<b>3j</b>	H	H	<i>p</i> -CH <sub>3</sub> OC <sub>6</sub> H <sub>4</sub>	2.5	73
11	<b>3k</b>	H	H	<i>p</i> -Cl C <sub>6</sub> H <sub>4</sub>	2.5	70
12	<b>3l</b>	H	H	<i>m</i> -O <sub>2</sub> NC <sub>6</sub> H <sub>4</sub>	2.5	73
13	<b>3m</b>	H	NO <sub>2</sub>	furan-2-yl	2.5	72
14	<b>3n</b>	H	NO <sub>2</sub>	pyridin-4-yl	1	77

<sup>a</sup>All reactions were carried out using **1** (2.5 mmol), **2** (3.0 mmol), ZnCl<sub>2</sub>(3.0 mmol), and DMF (3 mL).<sup>b</sup>Isolated yield.**Figure 1.** Molecular structure of compound **3l**. Crystal structure: Selected bond length (Å<sup>a</sup>) and bond angles (°): O(1)–C(1) 1.2500(19), O(2)–N(3) 1.237(3), O(3)–N(3) 1.2247(19), N(1)–C(1) 1.346(2), N(1)–C(8) 1.465(2), N(2)–C(3) 1.374(2), N(2)–C(8) 1.456(2), C(1)–C(2) 1.479(2), C(2)–C(3) 1.410(2), C(1)–N(1)–C(8) 123.91(14), C(3)–N(2)–C(8) 119.55(13), O(3)–N(3)–O(2) 123.4(2), N(1)–C(1)–C(2) 116.11(14), C(3)–C(2)–C(1) 119.07(14), N(2)–C(3)–C(2) 119.70(15), N(2)–C(8)–N(1) 108.72(13), N(2)–C(8)–C(9) 112.72(13), N(1)–C(8)–C(9) 112.16(13).

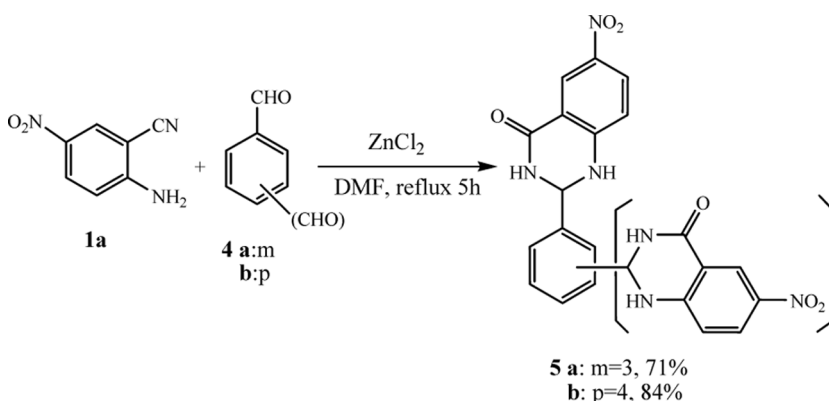


**Scheme 2.** Reaction of aromatic *o*-aminonitrile with aromatic aldehyde.

As an extension of the research, the reaction of 2-amino-benzonitrile **1a** with terephthalaldehyde **4** in the sealed reactor under the catalyst of  $\text{ZnCl}_2$  (0.1 mol) at  $200^\circ\text{C}$  afforded the expected symmetrical bisquinazolinone derivatives **5** (Scheme 3).

## CONCLUSION

Quinazolin-4(3H)-one derivatives were conveniently obtained by the cyclization of *o*-aminonitriles with carbonyl compounds using zinc chloride as the catalyst of in DMF. The reaction scope is substantial, and a number of aryl aldehydes could be successfully applied to react with *o*-aminonitriles to give quinazolinone compounds with good yields. Currently work is actively under way to expand this synthetic methodology to other valuable *o*-aminonitrile systems.



**Scheme 3.** Reaction of 2-amino-5-nitrobenzonitrile with terephthalaldehyde.

## EXPERIMENTAL

### General Procedure

*o*-Aminobenzonitrile 1 (2.5 mmol) and aldehydes (2.5 mmol) were added to a solution of DMF (10 mL) and ZnCl<sub>2</sub> (3.0 mmol). The mixture was heated at reflux for the specified time (see Table 1). After completion of the reaction as indicated by thin-layer chromatography (TLC; eluent: ethyl acetate), the cooled reaction mixture was quenched with water (10 mL), and the precipitate was separated by filtration. Then the residue was dispersed into water and titrated to pH 12–13 by 20% sodium hydroxide. After filtration, the products **3a–n** were isolated by column chromatography (200- to 300-mesh silica gel, ethyl acetate–petroleum 1:2).

### 6-Nitro-2-phenyl-1,2-dihydroquinazolin-4(3*H*)-one (**3a**)

Mp 264–265°C; <sup>1</sup>H NMR (400 MHz, DMSO-*d*<sub>6</sub>) δ<sub>H</sub>: 6.02 (1H, t, *J* = 1.7 Hz, CH), 6.83 (1H, d, *J* = 8.8 Hz, ArH), 7.39 (1H, t, *J* = 7.2 Hz, ArH), 7.42 (2H, t, *J* = 7.2, 8.0 Hz, ArH), 7.49 (2H, d, *J* = 8.0 Hz, ArH), 8.11 (1H, dd, *J* = 2.8, 8.8 Hz, ArH), 8.44 (1H, d, *J* = 2.8 Hz, ArH), 8.57 (1H, s, NH), 8.75 (1H, s, NH). <sup>13</sup>C NMR (100 MHz, DMSO-*d*<sub>6</sub>) δ<sub>C</sub>: 66.30, 112.64, 114.24, 124.18, 126.54 (2C), 128.65 (2C), 128.88, 128.97, 137.12, 141.09, 152.13, 161.28; MS (ESI): *m/z* (%) = 270.1 (100) [M + H]<sup>+</sup>. IR (KBr): ν<sub>max</sub> 3385, 3166, 1690, 1618, 1530, 1329, 1140 cm<sup>−1</sup>. Anal. calcd. for C<sub>14</sub>H<sub>11</sub>N<sub>3</sub>O<sub>3</sub>: C, 62.45; H, 4.12; N, 15.61. Found: C, 62.28; H, 3.80; N, 15.25.

### 6-Nitro-(3-nitrophenyl)-1,2-dihydroquinazolin-4(3*H*)-one (**3b**)

Mp 294–296°C; <sup>1</sup>H NMR (400 MHz, DMSO-*d*<sub>6</sub>) δ<sub>H</sub>: 6.25 (1H, s, CH), 6.89 (1H, d, *J* = 8.8 Hz, ArH), 7.75 (1H, t, *J* = 8.0 Hz, ArH), 7.94 (1H, d, *J* = 8.0 Hz, ArH), 8.14 (1H, dd, *J* = 2.8, 8.8 Hz, ArH), 8.26–8.27 (1H, m, *J* = 1.6 Hz, ArH), 8.35 (1H, t, *J* = 1.6 Hz, ArH), 8.44 (1H, d, *J* = 2.8 Hz, ArH), 8.73 (1H, s, NH), 8.96 (1H, s, NH). <sup>13</sup>C NMR (100 MHz, DMSO-*d*<sub>6</sub>) δ<sub>C</sub>: 65.03, 112.69, 114.55, 121.38, 123.69, 124.16, 129.11, 130.45, 133.02, 137.50, 143.45, 147.80, 151.78, 161.19; MS (ESI): *m/z* (%) = 315.4 (100) [M + H]<sup>+</sup>. IR (KBr): ν<sub>max</sub> 3321, 3191, 1685, 1618, 1532, 1324 cm<sup>−1</sup>. Anal. calcd. for C<sub>14</sub>H<sub>10</sub>N<sub>4</sub>O<sub>5</sub>: C, 53.51; H, 3.21; N, 17.82. Found: C, 53.22; H, 3.27; N, 17.72.

### 6-Nitro-2-(4-methoxyphenyl)-1,2-dihydroquinazolin-4(3*H*)-one (**3c**)

Mp 242–243°C; <sup>1</sup>H NMR (400 MHz, DMSO-*d*<sub>6</sub>) δ<sub>H</sub>: 3.75 (3H, s, CH<sub>3</sub>), 5.97 (1H, s, CH), 6.82 (1H, d, *J* = 8.8 Hz, ArH), 6.98 (2H, d, *J* = 8.4 Hz, ArH), 7.39 (2H, d, *J* = 8.4 Hz, ArH), 8.11 (1H, dd, *J* = 2.8, 8.8 Hz, ArH), 8.43 (1H, d, *J* = 2.8 Hz, ArH), 8.50 (1H, s, NH), 8.68 (1H, s, NH). <sup>13</sup>C NMR (100 MHz, DMSO-*d*<sub>6</sub>) δ<sub>C</sub>: 55.19, 65.91, 112.62, 113.90 (2C), 114.20, 124.16, 127.94 (2C), 128.93, 132.94, 136.99, 152.19, 159.67, 161.36; MS (ESI): *m/z* (%) = 300.1 (100) [M + H]<sup>+</sup>. IR (KBr): ν<sub>max</sub> 3385, 3162, 1660, 1620, 1512, 1309 cm<sup>−1</sup>. Anal. calcd. for C<sub>15</sub>H<sub>13</sub>N<sub>3</sub>O<sub>4</sub>: C, 60.19; H, 4.38; N, 14.04. Found: C, 60.05; H, 4.40; N, 14.11.

**2-(4-Chlorophenyl)-6-nitro-1,2-dihydroquinazolin-4(3*H*)-one (3d)**

Mp 263–265°C;  $^1\text{H}$  NMR (400 MHz,  $\text{DMSO}-d_6$ )  $\delta_{\text{H}}$ : 6.06 (1H, s, CH), 6.84 (1H, d,  $J=8.8$  Hz, ArH), 7.50 (4H, s, ArH), 8.13 (1H, dd,  $J=2.8, 8.8$  Hz, ArH), 8.43 (1H, d,  $J=2.8$  Hz, ArH), 8.59 (1H, s, NH), 8.80 (1H, s, NH).  $^{13}\text{C}$  NMR (100 MHz,  $\text{DMSO}-d_6$ )  $\delta_{\text{C}}$ : 65.49, 112.62, 114.32, 124.13, 128.45 (2C), 128.62 (2C), 128.99, 133.40, 137.23, 140.05, 151.96, 161.20; MS (ESI):  $m/z$  (%) = 304.4 (100)  $[\text{M} + \text{H}]^+$ . IR (KBr):  $\nu_{\text{max}}$  3366, 3180, 1692, 1615, 1492, 1328  $\text{cm}^{-1}$ . Anal. calcd. for  $\text{C}_{14}\text{H}_{10}\text{N}_3\text{O}_3\text{Cl}$ : C, 55.37; H, 3.32; N, 13.83. Found: C, 55.59; H, 3.62; N, 13.44.

**6-Nitro-2-(4-nitrophenyl)-1,2-dihydroquinazolin-4(3*H*)-one (3e)**

Mp 263–265°C;  $^1\text{H}$  NMR (400 MHz,  $\text{DMSO}-d_6$ )  $\delta_{\text{H}}$ : 6.21 (1H, s, CH), 6.87 (1H, d,  $J=8.8$  Hz, ArH), 7.73 (2H, d,  $J=8.4$  Hz, ArH), 8.13 (1H, dd,  $J=2.8, 8.8$  Hz, ArH), 8.29 (2H, d,  $J=8.4$  Hz, ArH), 8.43 (1H, d,  $J=2.8$  Hz, ArH), 8.72 (1H, s, NH), 8.96 (1H, s, NH).  $^{13}\text{C}$  NMR (100 MHz,  $\text{DMSO}-d_6$ )  $\delta_{\text{C}}$ : 65.14, 112.63, 114.49, 123.92 (2C), 124.16, 127.80 (2C), 129.11, 137.44, 147.66, 148.29, 151.73, 161.09; MS (ESI):  $m/z$  (%) = 315.4 (100)  $[\text{M} + \text{H}]^+$ . IR (KBr):  $\nu_{\text{max}}$  3350, 3084, 1690, 1621, 1522, 1330  $\text{cm}^{-1}$ . Anal. calcd. for  $\text{C}_{14}\text{H}_{10}\text{N}_4\text{O}_5$ : C, 53.51; H, 3.21; N, 17.82. Found: C, 53.33; H, 3.31; N, 17.46.

**7-Chloro-2-(4-nitrophenyl)-1,2-dihydroquinazolin-4(3*H*)-one (3f)**

Mp 243–244°C;  $^1\text{H}$  NMR (400 MHz,  $\text{DMSO}-d_6$ )  $\delta_{\text{H}}$ : 5.99 (1H, s, CH), 6.70 (1H, dd,  $J=2.0, 8.4$  Hz, ArH), 6.82 (1H, d,  $J=2.0$  Hz, ArH), 7.60 (2H, d,  $J=8.4$  Hz, ArH), 7.73 (2H, d,  $J=8.4$  Hz, ArH), 8.26 (1H, s, NH), 8.27 (1H, d,  $J=8.4$  Hz, ArH), 8.66 (1H, s, NH).  $^{13}\text{C}$  NMR (100 MHz,  $\text{DMSO}-d_6$ )  $\delta_{\text{C}}$ : 65.22, 113.59, 113.63, 117.43, 123.72 (2C), 127.95 (2C), 129.43, 138.09, 147.52, 148.20, 148.97, 162.42; MS (ESI):  $m/z$  (%) = 304.0 (100)  $[\text{M} + \text{H}]^+$ . IR (KBr):  $\nu_{\text{max}}$  3355, 3282, 1660, 1609, 1521, 1351  $\text{cm}^{-1}$ . Anal. calcd. for  $\text{C}_{14}\text{H}_{10}\text{N}_3\text{O}_3\text{Cl}$ : C, 55.37; H, 3.32; N, 13.84. Found: C, 55.72; H, 3.39; N, 13.86.

**7-Chloro-2-(4-methoxyphenyl)-1,2-dihydroquinazolin-4(3*H*)-one (3g)**

Mp 221–223°C;  $^1\text{H}$  NMR (400 MHz,  $\text{DMSO}-d_6$ )  $\delta_{\text{H}}$ : 3.75 (3H, s,  $\text{CH}_3$ ), 5.76 (1H, s, CH), 6.69 (1H, dd,  $J=2.0, 8.4$  Hz, ArH), 6.78 (1H, d,  $J=2.0$  Hz, ArH), 6.96 (2H, dd,  $J=1.6, 6.8$  Hz, ArH), 7.29 (1H, s, NH), 7.41 (2H, dd,  $J=1.6, 6.8$  Hz, ArH), 7.60 (1H, d,  $J=8.4$  Hz, ArH), 8.32 (1H, s, NH).  $^{13}\text{C}$  NMR (100 MHz,  $\text{DMSO}-d_6$ )  $\delta_{\text{C}}$ : 55.21, 66.23, 113.45, 113.71, 113.76 (2C), 117.01, 128.17 (2C), 129.36, 133.17, 137.79, 148.94, 159.56, 162.84; MS (ESI):  $m/z$  (%) = 289.0 (100)  $[\text{M} + \text{H}]^+$ . IR (KBr):  $\nu_{\text{max}}$  3299, 3181, 1653, 1609, 1512  $\text{cm}^{-1}$ . Anal. calcd. for  $\text{C}_{15}\text{H}_{13}\text{N}_2\text{O}_2\text{Cl}$ : C, 62.40; H, 4.54; N, 9.70. Found: C, 62.00; H, 4.55; N, 9.94.

**2-(4-Nitrophenyl)-1,2-dihydroquinazolin-4(3*H*)-one (3h)**

Mp 198–200°C;  $^1\text{H}$  NMR (400 MHz,  $\text{DMSO}-d_6$ )  $\delta_{\text{H}}$ : 5.91 (1H, s, CH), 6.68 (1H, t,  $J=7.6$  Hz, ArH), 6.76 (1H, d,  $J=8.0$  Hz, ArH), 7.26 (1H, t,  $J=7.6$  Hz,



ArH), 7.33 (1H, s, NH), 7.60 (1H, d,  $J = 8.0$  Hz, ArH), 7.73 (2H, d,  $J = 8.4$  Hz, ArH), 8.25 (2H, d,  $J = 8.4$  Hz, ArH), 8.51 (1H, s, NH).  $^{13}\text{C}$  NMR (100 MHz, DMSO- $d_6$ )  $\delta_{\text{C}}$ : 65.76, 115.02, 115.38, 117.92, 124.02 (2C), 127.86, 128.48 (2C), 134.00, 147.68, 147.91, 149.81, 163.71; MS (ESI):  $m/z$  (%) = 270.1 (100)  $[\text{M} + \text{H}]^+$ . IR (KBr):  $\nu_{\text{max}}$  3389, 3282, 1647, 1615, 1520, 1349  $\text{cm}^{-1}$ . Anal. calcd. for  $\text{C}_{14}\text{H}_{11}\text{N}_3\text{O}_3$ : C, 62.45; H, 4.12; N, 15.61. Found: C, 62.21; H, 4.51; N, 15.36.

### 2-(2-Hydroxyphenyl)-1,2-dihydroquinazolin-4(3H)-one (3i)

Mp 218~219°C;  $^1\text{H}$  NMR (400 MHz, DMSO- $d_6$ )  $\delta_{\text{H}}$ : 5.79 (1H, s, ArCH), 7.01~8.73 (8H, m, ArH), 8.36 (1H, s, NH), 8.49 (1H, s, NH), 12.72 (1H, s, OH).  $^{13}\text{C}$  NMR (100 MHz, DMSO- $d_6$ )  $\delta_{\text{C}}$ : 60.85, 111.89, 114.19, 123.90, 124.52, 128.68, 128.85, 130.09, 134.19, 135.00, 137.06, 146.95, 151.39, 165.89; MS (ESI):  $m/z$  (%) = 241.2 (100)  $[\text{M} + \text{H}]^+$ . IR (KBr):  $\nu_{\text{max}}$  3428, 3083, 1671, 1611  $\text{cm}^{-1}$ . Anal. Calcd. for  $\text{C}_{14}\text{H}_{12}\text{N}_2\text{O}_2$ : C, 69.99; H, 5.03; N, 11.66. found: C, 69.59; H, 5.53; N, 11.03.

### 2-(4-Methoxyphenyl)-1,2-dihydroquinazolin-4(3H)-one (3j)

Mp 255~257°C;  $^1\text{H}$  NMR (400 MHz, DMSO- $d_6$ )  $\delta_{\text{H}}$ : 3.86 (3H, s,  $\text{CH}_3$ ), 5.67 (1H, s, CH), 6.42 (2H, dd,  $J = 2.0, 7.6$  Hz, ArH), 6.89~6.90 (1H, m,  $J = 8.0$  Hz, ArH), 7.04 (1H, s, NH), 7.15 (1H, d,  $J = 8.0$  Hz, ArH), 7.20~7.21 (1H, m,  $J = 8.0$  Hz, ArH), 7.36 (1H, dd,  $J = 8.0$  Hz, ArH), 7.45 (2H, dd,  $J = 2.0, 7.6$  Hz, ArH), 8.21 (1H, s, NH).  $^{13}\text{C}$  NMR (100 MHz, DMSO- $d_6$ )  $\delta_{\text{C}}$ : 55.43, 65.36, 114.45, 114.71, 117.12 (2C), 117.69, 128.31 (2C), 128.75, 132.76, 133.79, 140.59, 147.56, 163.20; MS (ESI):  $m/z$  (%) = 255.1 (100)  $[\text{M} + \text{H}]^+$ . IR (KBr):  $\nu_{\text{max}}$  3269, 3171, 1678, 1602  $\text{cm}^{-1}$ . Anal. calcd. for  $\text{C}_{15}\text{H}_{14}\text{N}_2\text{O}_2$ : C, 70.85; H, 5.55; N, 11.02. Found: C, 70.88; H, 5.25; N, 10.81.

### 2-(4-Chlorophenyl)-1,2-dihydroquinazolin-4(3H)-one (3k)

Mp 227~228°C;  $^1\text{H}$  NMR (400 MHz, DMSO- $d_6$ )  $\delta_{\text{H}}$ : 5.78 (1H, s, CH), 6.67 (1H, t,  $J = 8.0$  Hz, ArH), 6.75 (1H, d,  $J = 8.0$  Hz, ArH), 7.15 (1H, s, NH), 7.25~7.27 (1H, m,  $J = 8.0$  Hz, ArH), 7.46~7.47 (2H, m,  $J = 2.0, 6.4$  Hz, ArH), 7.52 (2H, dd,  $J = 2.0, 6.4$  Hz, ArH), 7.62 (1H, dd,  $J = 8.0$  Hz, ArH), 8.34 (1H, s, NH).  $^{13}\text{C}$  NMR (100 MHz, DMSO- $d_6$ )  $\delta_{\text{C}}$ : 65.79, 114.48, 114.97, 117.30, 127.39, 128.33, 128.34, 128.78 (2C), 132.99, 133.42, 140.70, 147.67, 163.50; MS (ESI):  $m/z$  (%) = 259.1 (100)  $[\text{M} + \text{H}]^+$ . IR (KBr):  $\nu_{\text{max}}$  3325, 3188, 1658, 1609, 1483  $\text{cm}^{-1}$ . Anal. calcd. for  $\text{C}_{14}\text{H}_{11}\text{N}_2\text{OCl}$ : C, 65.00; H, 4.29; N, 10.83. Found: C, 65.38; H, 4.36; N, 10.88.

### 2-(4-Chlorophenyl)-1,2-dihydroquinazolin-4(3H)-one (3l)

Mp 210~212°C;  $^1\text{H}$  NMR (400 MHz, DMSO- $d_6$ )  $\delta_{\text{H}}$ : 5.95 (1H, s, CH), 6.70 (1H, t,  $J = 7.6$  Hz, ArH), 6.79 (1H, d,  $J = 8.0$  Hz, ArH), 7.29 (1H, t,  $J = 8.0$  Hz, ArH), 7.35 (1H, s, NH), 7.62 (1H, dd,  $J = 7.6$  Hz, ArH), 7.70 (1H, t,  $J = 7.6$  Hz, ArH), 7.94 (1H, d,  $J = 7.6$  Hz, ArH), 8.21~8.22 (1H, m,  $J = 1.4, 1.4$  Hz, ArH), 8.36 (1H, t,  $J = 1.8, 1.8$  Hz, ArH), 8.53 (1H, s, NH).  $^{13}\text{C}$  NMR (100 MHz, DMSO- $d_6$ )

$\delta_C$ : 65.20, 114.61, 114.97, 117.55, 121.59, 123.29, 127.43, 130.06, 133.39, 133.59, 144.32, 147.32, 147.73, 163.36; MS (ESI):  $m/z$  (%) = 270.1 (100)  $[M+H]^+$ ; IR (KBr):  $\nu_{\max}$  3296, 3188, 1653, 1610, 1532, 1353  $\text{cm}^{-1}$ . Anal. calcd. for  $\text{C}_{14}\text{H}_{11}\text{N}_3\text{O}_3$ : C, 62.45; H, 4.12; N, 15.61. Found: C, 62.16; H, 4.20; N, 15.24.

### 2-(Furan-2-yl)-6-nitro-1,2-dihydroquinazolin-4(3H)-one (3m)

Mp 267~269°C;  $^1\text{H}$  NMR (400 MHz,  $\text{DMSO}-d_6$ )  $\delta_H$ : 6.04 (1H, t,  $J=2.8$  Hz, CH), 6.37 (1H, d,  $J=2.8$  Hz, FuH), 6.42–6.43 (1H, m, FuH), 6.86 (1H, d,  $J=9.2$  Hz, ArH), 7.65 (1H, s, NH), 8.12 (1H, dd,  $J=2.8, 9.2$  Hz, ArH), 8.44 (1H, d,  $J=2.8$  Hz, ArH), 8.65 (1H, s, NH), 8.83 (1H, d,  $J=1.6$  Hz, FuH).  $^{13}\text{C}$  NMR (100 MHz,  $\text{DMSO}-d_6$ )  $\delta_C$ : 60.58, 108.19, 111.17, 113.63, 115.15, 124.75, 129.57, 138.09, 144.05, 152.53, 154.19, 161.99; MS (ESI):  $m/z$  (%) = 260.1 (100)  $[M+H]^+$ ; IR (KBr):  $\nu_{\max}$  3382, 3188, 3072, 1674, 1615, 1530, 1497, 1331, 1299, 1117  $\text{cm}^{-1}$ . Anal. Calcd. for  $\text{C}_{12}\text{H}_9\text{N}_3\text{O}_4$ : C, 55.60; H, 3.50; N, 16.21. found: C, 55.74; H, 3.40; N, 16.15.

### 6-Nitro-2-(pyridin-4-yl)-1,2-dihydroquinazolin-4(3H)-one (3n)

Mp 231~234°C;  $^1\text{H}$  NMR (400 MHz,  $\text{DMSO}-d_6$ )  $\delta_H$ : 6.38 (1H, t,  $J=2.8$  Hz, CH), 7.82 (1H, d,  $J=8.4$  Hz, PyH), 7.95 (1H, s, NH), 8.01–8.04 (2H, m, PyH), 8.38 (1H, dd,  $J=2.4, 8.4$  Hz, ArH), 8.68–8.71 (3H, ArH, PyH), 8.98 (1H, s, NH);  $^{13}\text{C}$  NMR (100 MHz,  $\text{DMSO}-d_6$ )  $\delta_C$ : 61.52, 120.30(2C), 122.31, 122.53, 129.19, 130.08, 140.07, 145.89, 153.06(2C), 154.63, 162.15; MS (ESI):  $m/z$  (%) = 271.1 (100)  $[M+H]^+$ ; IR (KBr):  $\nu_{\max}$  3340, 3183, 3072, 1679, 1619, 1567, 1466, 1347, 1143  $\text{cm}^{-1}$ ; Anal. Calcd. for  $\text{C}_{13}\text{H}_{10}\text{N}_4\text{O}_3$ : C, 57.78; H, 3.73; N, 20.73. found: C, 57.98; H, 3.40; N, 20.09.

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