HETEROCYCLES, Vol. 75, No. 5, 2008, pp. 1225 - 1231. © The Japan Institute of Heterocyclic Chemistry Received, 14th December, 2007, Accepted, 1st February, 2008, Published online, 5th February, 2008. COM-07-11304

SYNTHESIS OF 2,4-DISUBSTITUTED ISOQUINOLIN-1(2*H*)-ONES BASED ON REACTIONS OF α -SUBSTITUTED 2-LITHIO- β -METHOXYSTYRENES WITH ISOCYANATES

Kazuhiro Kobayashi,* Kazutaka Hayashi, Changeng Nam, Shuhei Fukamachi, and Hisatoshi Konishi

Department of Materials Science, Faculty of Engineering, Tottori University, 4-101 Koyama-minami, Tottori 680-8552, Japan

Abstract - An efficient method for the preparation of 2,4-disubstituted isoquinolin-1(2*H*)-ones is described. The reaction of α -substituted 2-lithio- β -methoxystyrenes, generated by treating α -substituted 2-bromo- β methoxystyrenes with butyllithium, with isocyanates yields the corresponding α -substituted 2-(2-methoxyvinyl)benzamide derivatives, which in turn are transformed into 2,4-disubstituted isoquinolin-1(2*H*)-ones on treatment with a catalytic amount of concentrated hydriodic acid.

In a recent paper,¹ we demonstrated that α -substituted 2-lithio- β -methoxystyrenes, generated by a treatment of α -substituted 2-bromo- β -methoxystyrenes with butyllithium, reacted with various nitriles to give directly 1,4-disubstituted isoquinolines. As part of studies on the utility of these lithium products for organic synthesis, we have been interested in the reaction with isocyanates, which should give 2,4-disubstituted isoquinolin-1(2*H*)-ones. In this paper, we wish to report a convenient synthesis of 2,4-disubstituted isoquinolin-1(2*H*)-ones (**3**) from α -substituted 2-bromo- β -methoxystyrenes (**1**). We found that the reaction of α -substituted 2-lithio- β -methoxystyrenes with isocyanates gave only α -substituted 2-(2-methoxyvinyl)benzamide derivatives (**2**). However, these benzamides (**2**) could be converted into 2,4-disubstituted isoquinolin-1(2*H*)-ones (**3**) upon treatment with a catalytic amount of concentrated hydriodic acid. Recently, a number of efficient methods for the preparation of isoquinolin-1(2*H*)-one derivatives have been reported,² because of their biological activities.³ Moreover, some of them have been utilized as precursors for the synthesis of more complex and useful molecules.⁴ However, to date, there have been only a limited report on the synthesis of isoquinolin-1(2*H*)-one derivatives both at the 2- and 4-positions.⁵

Our two-step preparation of 2,4-disubstituted isoquinolin-1(2*H*)-ones (3) from α -substituted 2-bromo- β -methoxystyrenes (1) was conducted as illustrated in Scheme 1. The first step involves the reaction of α-substituted 2-lithio-β-methoxystyrenes, generated by the treatment of **1** with butyllithium in diethyl ether at 0 °C as described previously,¹ with various isocyanates, affording the corresponding α-substituted 2-(2-methoxyvinyl)benzamide derivatives (**2**). As summarized in Table 1, the yields of **2** are generally fair to good, though those using *t*-butyl isocyanate are somewhat lower (Entries 7 and 9). These lower yields are considered to arise from the less reactivity of this isocyanate due to the bulkiness of the *t*-butyl group. Compounds (**2**) were obtained as mixtures of stereoisomers, and the ratios were determined by their ¹H NMR spectra (see Experimental). However, the stereochemistry of each isomer could not be determined in every case.



Scheme 1

Entry	1	R ³ in RNCO	2 (Yield/%) ^{a,b}	3 (Yield/%) ^a
1	$1a (R^1 = H, R^2 = Ph)$	Ph	2a (73)	3a (67)
2	1a	o-Tol	2b (74)	3b (61)
3	1a	$4-ClC_6H_4$	2c (69)	3c (75)
4	1b ($\mathbf{R}^1 = \mathbf{H}, \mathbf{R}^2 = 4\text{-}\mathrm{ClC}_6\mathbf{H}_4$)	Ph	2d (71)	3d (66)
5	1b	$4-ClC_6H_4$	2e (62)	3e (73)
6	1b	$3-\text{MeOC}_6\text{H}_4$	2f (73)	3f (68)
7	1b	<i>t</i> -Bu	2g (51)	3g (64)
8	1c ($R^1 = H, R^2 = Me$)	$4-ClC_6H_4$	2h (65)	3h (71)
9	1c	<i>t</i> -Bu	2i (55)	3i (69)
10	1d ($R^1 = OMe, R^2 = Ph$)	Ph	2j (88)	3j (63)

 Table 1. Preparation of Isoquinolin-1(2H)-ones (3) via 2-Vinylbenzamides (2)

^aIsolated yields. ^bMixtures of stereoisomers.

Subsequent intramolecular replacement of the methoxy substituent in the methoxyvinyl moiety with the amide nitrogen atom of 2 was easily achieved upon treatment of a catalytic amount of concentrated hydriodic acid in acetonitrile at 0 °C to room temperature to result in the formation of 3 in generally fair yields, independent of the substituents both at the 2-and 4-positions. These results are also summarized in Table 1.

In summary, we have shown that α -substituted 2-(2-methoxyvinyl)benzamide derivatives, which are readily prepared by reactions of α -substituted 2-lithio- β -methoxystyrenes with isocyanates, cyclize on treatment with concentrated hydriodic acid to afford 2,4-disubstituted isoquinoline-1(2*H*)-ones. Since these isoquinolinones derivatives are hard to obtain by conventional methods, the present new and simple method may find some value in organic synthesis. Further investigations on the development of methods

for the synthesis of heterocycles utilizing reactions of these lithium compounds with various electrophiles are now in progress in our laboratory, and results will be reported in near future.

EXPERIMENTAL

The melting points were determined on a Laboratory Devices MEL-TEMP II melting-point apparatus and are uncorrected. The IR spectra were recorded on a Shimadzu FTIR-8300 spectrometer. The ¹H NMR spectra were determined using SiMe₄ as an internal reference with a JEOL ECP500 FT NMR spectrometer operating at 500 MHz. The ¹³C NMR spectra were determined in CDCl₃ using TMS as an internal reference with a JEOL ECP500 FT NMR spectrometer operating at 125 MHz. Low-resolution mass spectra (EI) were recorded on a JEOL AUTOMASS 20 spectrometer. Thin-layer chromatography (TLC) was carried out on Merck Kieselgel 60 PF₂₅₄. Column chromatography was performed using Merck Kieselgel 60 (0.063–0.200 mm). All of the solvents used were dried over appropriate drying agents and distilled under argon prior to use.

Starting Materials. 2-Bromostyrene derivatives (1a),¹ (1b),¹ (1c),⁶ and (1d)¹ were prepared by appropriate reported procedures. All other chemicals used in this study were commercially available.

Typical Procedure for the Preparation of 2-(2-Methoxyvinyl)benzamides (2). 2-(2-Methoxy-1phenylethenyl)-*N*-phenylbenzamide (2a). To a stirred solution of 1a (0.29 g, 1.0 mmol) in Et₂O (4 mL) at 0 °C was added dropwise *n*-BuLi (1.6M in hexane; 1.0 mmol). After 1 h, PhNCO (0.12 g, 1.0 mmol) was added and stirring was continued for an additional 30 min. The mixture was quenched with saturated aqueous NH₄Cl (15 mL) and extracted with Et₂O three times (10 mL each). The combined extracts were dried over anhydrous Na₂SO₄ and evaporated. The residue was purified by column chromatography on silica gel to afford 2a (0.24 g, 73%); a pale-yellow viscous oil; R_f 0.40 (1:5 AcOEt–hexane); a mixture of stereoisomers (*ca.* 9:1); IR (neat) 3323, 1649, 1636, 1603 cm⁻¹; ¹H NMR δ 3.71 (0.3H, s), 3.77 (2.7H, s), 6.40 (0.9H, s), 6.69 (0.1H, s), 7.04 (0.9H, tt, *J* = 7.3, 1.4 Hz), 7.07 (0.1H, tt, *J* = 7.3, 1.4 Hz), 7.13–7.29 (7H, m), 7.35 (0.9H, d, *J* = 8.2 Hz), 7.36 (0.1H, d, *J* = 8.2 Hz), 7.42–7.51 (4H, m), 7.67 (0.9H, br s), 7.85 (1H, dd, *J* = 7.8, 1.8 Hz), 7.88 (0.1H, br s). Anal. Calcd for C₂₂H₁₉NO₂: C, 80.22; H, 5.81; N, 4.25. Found: C, 80.08; H, 5.85, N, 4.47.

2-(2-Methoxy-1-phenylethenyl)-*N*-(**2-methylphenyl)benzamide** (**2b**): a pale-yellow viscous oil; $R_f 0.14$ (1:3 AcOEt–hexane); a mixture of stereoisomers (*ca.* 5:5); IR (neat) 3288, 1668, 1634 cm⁻¹; ¹H NMR δ 1.93 (1.5H, s), 2.04 (1.5H, s), 3.70 (1.5H, s), 3.78 (1.5H, s), 6.42 (0.5H, s), 6.70 (0.5H, s), 6.99–7.05 (1H, m), 7.07–7.19 (4H, m), 7.22–7.26 (3H, m), 7.28 (0.5H, dd, J = 7.8, 1.4 Hz), 7.34 (0.5H, d, J = 7.3 Hz), 7.42–7.55 (2H, m), 7.57 (0.5H, br s), 7.66 (0.5H, d, J = 7.8 Hz), 7.71 (0.5H, br s), 7.82 (0.5H, d, J = 8.2 Hz), 7.87 (0.5H, dd, J = 8.2, 1.8 Hz), 7.89 (0.5H, d, J = 7.8 Hz). Anal. Calcd for C₂₃H₂₁NO₂: C, 80.44; H, 6.16; N, 4.08. Found: C, 80.31; H, 6.17; N, 3.99.

N-(4-Chlorophenyl)-2-(2-methoxy-1-phenylethenyl)benzamide (2c): a pale-yellow viscous oil; $R_f 0.14$

(1:5 AcOEt–hexane); a mixture of stereoisomers (*ca.* 8:2); IR (neat) 3223, 1645, 1628 cm⁻¹; ¹H NMR δ 3.70 (2.4H, s), 3.78 (0.6H, s), 6.39 (0.2H, s), 6.69 (0.8H, s), 7.08–7.51 (12H, m), 7.66 (0.2H, br s), 7.82 (1H, d, *J* = 7.3 Hz), 7.88 (0.8H, br s). Anal. Calcd for C₂₂H₁₈ClNO₂: C, 72.62; H, 4.99; N, 3.85. Found: C, 72.47; H, 5.27; N, 3.86.

2-[1-(4-Chlorophenyl)-2-methoxyethenyl]-*N***-phenylbenzamide (2d):** a pale-yellow viscous oil; R_f 0.21 (1:4 AcOEt–hexane); a mixture of stereoisomers (*ca.* 7:3); IR (neat) 3288, 1652, 1636 cm⁻¹; ¹H NMR δ 3.71 (0.9H, s), 3.79 (2.1H, s), 6.43 (0.7H, s), 6.65 (0.3H, s), 7.04–7.51 (12H, m), 7.59 (0.7H, br s), 7.79 (0.3H, br s), 7.82 (1H, d, *J* = 7.8 Hz). Anal. Calcd for C₂₂H₁₈ClNO₂: C, 72.62; H, 4.99; N, 3.85. Found: C, 72.49; H, 5.01, N, 3.71.

N-(4-Chlorophenyl)-2-[1-(4-chlorophenyl)-2-methoxyethenyl]benzamide (2e): a white solid; mp 140–160 °C; a mixture of stereoisomers (*ca.* 9:1); IR (neat) 3320, 1654, 1631, 1601 cm⁻¹; ¹H NMR δ 3.71 (0.3H, s), 3.80 (2.7H, s), 6.42 (0.9H, s), 6.65 (0.1H, s), 7.14–7.22 (6H, m), 7.31–7.36 (3H, m), 7.44–7.51 (2H, m), 7.58 (0.9H, br s), 7.72 (0.1H, br s), 7.79 (1H, d, *J* = 7.3 Hz). Anal. Calcd for C₂₂H₁₇Cl₂NO₂: C, 66.34; H, 4.30; N, 3.52. Found: C, 66.24; H, 4.30, N, 3.49.

2-[1-(4-Chlorophenyl)-2-methoxyethenyl]-*N***-(3-methoxyphenyl)benzamide (2f):** a pale-yellow viscous oil; $R_f 0.13$ (1:4 AcOEt–hexane); a mixture of stereoisomers (*ca.* 5:5); IR (neat) 3300, 1659, 1634, 1605 cm⁻¹; ¹H NMR δ 3.75 (1.5H, s), 3.78 (3H, s), 3.79 (1.5H, s), 6.18 (0.5H, s), 6.42 (0.5H, s), 6.59–7.50 (11H, m), 7.58 (0.5H, br s), 7.78–7.82 (1.5H, m). Anal. Calcd for C₂₃H₂₀ClNO₃: C, 70.14; H, 5.12; N, 3.56. Found: C, 70.10; H, 5.26, N, 3.38.

2-[1-(4-Chlorophenyl)-2-methoxyethenyl]-*N***-(1,1-dimethylethyl)benzamide (2g):** a pale-yellow oil; $R_{\rm f}$ 0.21 (1:3 AcOEt–hexane); a mixture of stereoisomers (*ca*.7:3); IR (neat) 3400, 1653, 1634 cm⁻¹; ¹H NMR δ 1.13 (6.3H, s), 1.23 (2.7H, s), 3.74 (0.9H, s), 3.83 (2.1H, s), 5.71 (0.7H, br s), 5.93 (0.3H, br s), 6.34 (0.7H, s), 6.71 (0.3H, s), 7.05 (0.6H, d, J = 8.7 Hz), 7.18–7.25 (2.4H, m), 7.36–7.43 (4H, m), 7.67–7.71 (1H, m). Anal. Calcd for C₂₀H₂₂NO₄: C, 65.51; H, 7.90; N, 4.77. Found: C, 65.48; H, 7.94, N, 4.76.

N-(**4**-Chlorophenyl)-2-(2-methoxy-1-methylethenyl)benzamide (2h): a pale-yellow oil; R_f 0.40 (1:30 THF–benzene); a mixture of stereoisomers (*ca.* 7:3); IR (neat) 3263, 3229, 1655, 1645 cm⁻¹; ¹H NMR δ 1.88 (0.9H, d, J = 1.4 Hz), 1.94 (2.1H, d, J = 0.9 Hz), 3.59 (0.9H, s), 3.72 (2.1H, s), 6.12 (0.3H, q, J = 1.4 Hz), 6.26 (0.7H, q, J = 0.9 Hz), 7.23–7.55 (7H, m), 7.76 (0.3H, d, J = 7.3 Hz), 7.81 (0.7H, dd, J = 7.3, 1.4 Hz), 7.99 (0.7H, br s), 8.21 (0.3H, br s). Anal. Calcd for C₁₇H₁₆ClNO₂: C, 67.66; H, 5.34; N, 4.64. Found: C, 67.41; H, 5.40; N, 4.59.

N-(**1,1-Dimethylethyl**)-**2**-(**2-methoxy-1-methylethenyl**)**benzamide** (**2i**): a pale-yellow solid; mp 65–70 °C; a mixture of stereoisomers (*ca.* 7:3); IR (KBr) 3267, 1666, 1626 cm⁻¹; ¹H NMR δ 1.415 (2.7H, s), 1.420 (6.3H, s), 1.85 (0.9H, d, *J* = 1.4 Hz), 1.94 (2.1H, d, *J* = 1.4 Hz), 3.57 (0.9H, s), 3.69 (2.1H, s), 5.97 (0.7H, br s), 6.05 (0.3H, q, *J* = 1.4 Hz), 6.14 (0.7H, q, *J* = 1.4 Hz), 6.27 (0.3H, br s), 7.13–7.17 (1H, m), 7.28–7.39 (2H, m), 7.64 (0.7H, dd, *J* = 7.3, 1.4 Hz), 7.67 (0.3H, dd, *J* = 7.8, 1.4 Hz). Anal. Calcd for

C₁₅H₂₁NO₂: C, 72.84; H, 8.56; N, 5.66. Found: C, 72.59; H, 8.42; N, 5.53.

4-Methoxy-2-(2-methoxy-1-phenylethenyl)-*N*-phenylbenzamide (2j): a pale-yellow solid; mp 45–60 °C; a mixture of stereoisomers (*ca.* 5:5); IR (KBr) 3223, 1649, 1636 cm⁻¹; ¹H NMR δ 3.71 (1.5H, s), 3.748 (1.5H, s), 3.754 (1.5H, s), 3.76 (1.5H, s), 6.10 (0.5H, s), 6.48 (0.5H, s), 6.57 (0.5H, dd, *J* = 8.7, 2.7 Hz), 6.60 (0.5H, d, *J* = 2.7 Hz), 6.71 (0.5H, d, *J* = 2.7 Hz), 6.74 (0.5H, dd, *J* = 8.7, 2.7 Hz), 6.96–7.32 (10H, m), 7.41–7.45 (2H, m). Anal. Calcd for C₂₃H₂₁NO₃: C, 76.86; H, 5.89; N, 3.90. Found: C, 76.83; H, 5.95; N, 3.99.

Typical Procedure for the Preparation of Isoquinolinone Derivatives (3). 2,4-Diphenylisoquinolin-1(2*H*)-one (3a).⁵ To a stirred solution of 2a (0.24 g, 0.72 mmol) in MeCN (7 mL) at 0 °C was added a drop of concentrated HI. The mixture was allowed to warm to rt, and stirring was continued for 1.5 h before saturated aqueous NaHCO₃ (10 mL) was added. The organic materials were extracted with Et₂O (20 mL), and the extract was washed with brine and dried over anhydrous Na₂SO₄. Evaporation of the solvent gave a residue, which was purified by preparative TLC on silica gel to afford 3a (0.14 g, 67%); a white solid; mp 133–134 °C (hexane–Et₂O) (lit.,⁵ mp 132–133 °C). The IR and ¹H NMR data for this product were identical to those reported previously.⁵

2-(2-Methylphenyl)-4-phenylisoquinolin-1(2*H***)-one (3b): a white solid; mp 135–137 °C (hexane–Et₂O); IR (KBr) 1662, 1624, 1603 cm⁻¹; ¹H NMR \delta 2.24 (3H, s), 7.03 (1H, s), 7.29–7.37 (5H, m), 7.39–7.47 (4H, m), 7.56 (1H, ddd,** *J* **= 8.2, 6.9, 1.4 Hz), 7.63–7.68 (2H, m), 8.58 (1H, d,** *J* **= 7.8 Hz); ¹³C NMR \delta 17.84, 119.57, 124.79, 126.39, 127.11, 127.57, 127.73, 128.64 (two overlapped C's), 128.67, 128.86, 129.96 (two overlapped C's), 131.11, 132.40, 135.42, 136.10, 136.58, 140.40, 161.20; MS** *m/z* **311 (M⁺, 89), 294 (100). Anal. Calcd for C₂₂H₁₇NO: C; 84.86; H, 5.50; N, 4.50. Found: C, 84.67; H, 5.54; N, 4.46.**

2-(4-Chlorophenyl)-4-phenylisoquinolin-1(*2H*)-one (3c):⁵ a white solid; mp 148–150 °C (hexane–Et₂O) (lit.,² mp 148–149 °C); IR (KBr) 1661, 1628, 1601 cm⁻¹; ¹H NMR δ 7.13 (1H, s), 7.41–7.50 (9H, m), 7.57 (1H, ddd, *J* = 8.2, 6.9, 1.4 Hz), 7.60 (1H, d, *J* = 8.2 Hz), 7.66 (1H, ddd, *J* = 8.2, 6.9, 1.4 Hz), 8.57 (1H, dd, *J* = 8.2, 1.4 Hz).

4-(4-Chlorophenyl)-2-phenylisoquinolin-1(2*H***)-one (3d): a white solid; mp 145–147 °C (hexane–Et₂O); IR (KBr) 1661, 1626 cm⁻¹; ¹H NMR \delta 7.17 (1H, s), 7.40 (2H, d,** *J* **= 8.2 Hz), 7.43–7.58 (9H, m), 7.66 (1H, td,** *J* **= 7.3, 1.4 Hz), 8.58 (1H, dd,** *J* **= 7.3, 1.4 Hz); ¹³C NMR \delta 118.44, 124.46, 126.39, 126.83, 127.38, 128.22, 128.81, 128.91, 129.34, 131.24 (two overlapped C's), 132.60, 133.82, 134.58, 136.15, 141.13, 161.49; MS** *m***/***z* **331 (M⁺, 100). Anal. Calcd for C₂₁H₁₄ClNO: C; 76.36; H, 3.94; N, 4.24. Found: C, 76.17; H, 4.19; N, 4.08.**

2,4-Bis(4-chlorophenyl)isoquinolin-1(2*H***)-one (3e):** a white solid; mp 180–183 °C (hexane–AcOEt); IR (KBr disk) 1657, 1622 cm⁻¹; ¹H NMR δ 7.11 (1H, s), 7.38 (2H, d, *J* = 8.7 Hz), 7.43 (2H, d, *J* = 8.7 Hz), 7.45 (2H, d, *J* = 8.7 Hz), 7.48 (2H, d, *J* = 8.7 Hz), 7.53 (1H, d, *J* = 8.2 Hz), 7.58 (1H, ddd, *J* = 8.2, 7.3, 1.4 Hz), 7.67 (1H, ddd, *J* = 8.2, 7.3, 1.4 Hz), 8.56 (1H, dd, *J* = 8.2, 1.4 Hz); MS *m*/*z* 365 (M⁺, 100). Anal.

Calcd for C₂₁H₁₃Cl₂NO: C, 68.87; H, 3.58; N, 3.82. Found: C, 68.93; H, 3.70; N, 3.73.

4-(4-Chlorophenyl)-2-(3-methoxyphenyl)isoquinolin-1(2*H***)-one (3f): a pale-yellow solid; mp 121–124 °C (Et₂O); IR (KBr) 1665, 1628, 1605 cm⁻¹; ¹H NMR \delta 3.85 (3H, s), 6.97 (1H, ddd,** *J* **= 8.2, 2.8, 0.9 Hz), 7.02–7.06 (2H, m), 7.16 (1H, s), 7.38–7.42 (3H, m), 7.45 (2H, d,** *J* **= 8.2 Hz), 7.54 (1H, d,** *J* **= 8.2 Hz), 7.57 (1H, ddd,** *J* **= 8.2, 7.3, 0.9 Hz), 7.66 (1H, ddd,** *J* **= 8.2, 7.3, 1.4 Hz), 8.58 (1H, dd,** *J* **= 8.2, 0.9 Hz); MS** *m***/***z* **361 (M⁺, 100). Anal. Calcd for C₂₂H₁₆ClNO₂: C, 73.03; H, 4.46; N, 3.87. Found: C, 73.07; H, 4.14; N, 3.84.**

4-(4-Chlorophenyl)-2-(1,1-dimethylethyl)isoquinolin-1(2*H***)-one (3g): a white solid; mp 183–185 °C (hexane–Et₂O); IR (KBr) 1643, 1616 cm⁻¹; ¹H NMR \delta 1.76 (9H, s), 7.30 (1H, s), 7.36 (2H, d,** *J* **= 8.2 Hz), 7.44–7.46 (3H, m), 7.49 (1H, ddd,** *J* **= 7.8, 7.3, 0.9 Hz), 7.58 (1H, ddd,** *J* **= 8.2, 7.3, 1.4 Hz), 8.51 (1H, dd,** *J* **= 8.2, 0.9 Hz); MS** *m***/***z* **311 (M⁺, 22), 255 (100). Anal. Calcd for C₁₉H₁₈ClNO: C, 73.19; H, 5.82; N, 4.49. Found: C, 72.99; H, 5.86; N, 4.32.**

2-(4-Chlorophenyl)-4-methylisoquinolin-1(*2H*)-one (3h): a white solid; mp 168–170 °C (hexane–Et₂O); IR (KBr) 1665, 1628, 1612 cm⁻¹; ¹H NMR δ 2.32 (3H, s), 6.98 (1H, s), 7.39 (2H, d, *J* = 8.7 Hz), 7.47 (2H, d, *J* = 8.7 Hz), 7.56 (1H, dd, *J* = 7.8, 7.3 Hz), 7.65 (1H, d, *J* = 7.8 Hz), 7.75 (1H, dd, *J* = 7.8, 7.3 Hz), 8.51 (1H, d, *J* = 7.8 Hz); MS *m*/*z* 269 (M⁺, 100). Anal. Calcd for C₁₆H₁₂ClNO: C; 71.38; H, 4.46; N, 5.20. Found: C, 71.32; H, 4.34; N, 5.04.

2-(1,1-Dimethylethyl)-4-methylisoquinolin-1(*2H*)-one (3i): a pale-yellow oil; R_f 0.41 (1:4 AcOEt-hexane); IR (neat) 1651, 1626, 1603 cm⁻¹; ¹H NMR δ 1.75 (9H, s), 2.28 (3H, d, J = 1.4 Hz), 7.19 (1H, q, J = 1.4 Hz), 7.47 (1H, ddd, J = 7.8, 7.3, 0.9 Hz), 7.55 (1H, d, J = 7.8 Hz), 7.66 (1H, ddd, J = 7.8, 7.3, 1.4 Hz), 8.46 (1H, dd, J = 7.8, 0.9 Hz); MS *m*/*z* 215 (M⁺, 18), 159 (100). Anal. Calcd for C₁₄H₁₇NO: C, 78.10; H, 7.96; N, 6.51. Found: C, 78.16; H, 8.05; N, 6.29.

6-Methoxy-2,4-diphenylisoquinolin-1(*2H*)-one (**3**j): a white solid; mp 153–155 °C (hexane–Et₂O); IR (KBr disk) 1722, 1636, 1605 cm⁻¹; ¹H NMR δ 3.81 (3H, s), 6.78 (1H, d, *J* = 2.3 Hz), 7.09 (1H, dd, *J* = 8.7, 2.3 Hz), 7.24 (1H, s), 7.39–7.41 (4H, m), 7.45–7.51 (6H, m), 8.33 (1H, d, *J* = 8.7 Hz); MS *m*/*z* (%) 327 (M⁺, 13), 251 (94), 223 (100). Anal. Calcd for C₂₂H₁₇NO₂: C, 80.71; H, 5.23; N, 4.28. Found: C, 80.68; H, 5.41; N, 4.46.

ACKNOWLEDGEMENTS

We would like to thank Mrs. Miyuki Tanmatsu of this Faculty for determining mass spectra and performing combustion analyses.

REFERENCES AND NOTES

- 1. K. Kobayashi, K. Hayashi, K. Miyamoto, O. Morikawa, and H. Konishi, Synthesis, 2006, 2934.
- 2. a) Z. Xiang, T. Luo, K. Lu, J. Cui, X. Shi, R. Fathi, J, Chen, and Z. Yang, Org. Lett., 2004, 6, 3155.

b) K. Cherry, A. Duchene, J. Thibonnet, J.-L. Parrain, and M. Abarbi, *Synthesis*, 2005, 2349. c) N. Coskun, and Y. Kizikusak, *Synth. Commun.*, 2005, **35**, 2435. d) J. F. Guastavino, S. M. Barolo, and R. A. Rossi, *Eur. J. Org. Chem.*, 2006, 3898. e) Z. He and A. K. Yudin, *Org. Lett.*, 2006, **8**, 5829.

- a) C. Y. Watson, W. J. D. Whish, and M. D. Threadgill, *Bioorg. Med. Chem.*, 1998, 6, 721. b) T. N. Li and W.-J. Cho, *Chem. Pharm. Bull.*, 2005, 53, 118. c) R. Fujita, T. Yoshisuji, S. Wakayanagi, H. Wakamatsu, and H. Matsuzaki, *Chem. Pharm. Bull.*, 2006, 54, 204. d) R. Fujita, S. Wakayanagi, H. Wakamatsu, and H. Matsuzaki, *Chem. Pharm. Bull.*, 2006, 54, 209. e) D. Brookings, R. J. Davenport, J. Davis, F. C. A. Galvin, S. Lloyd, S. R. Mark, R. Owens, V. Sabin, and J. Wynn, *Bioorg. Med. Chem. Lett.*, 2007, 17, 562.
- a) M. Treus, J. C. Estévez, L. Castedo, and R. Estévez, *Tetrahedron Lett.*, 2002, 43, 5323. b) T. N. Le, S. G. Gang, and W.-J. Cho, *J. Org. Chem.*, 2004, 69, 2768. c) Z. Ma and D. Y. W. Lee, *Tetrahedron Lett.*, 2004, 45, 6721. d) S. Peukert, U. Schwahn, S. Güssregen, H. Schreuder, and A. Hofmeister, *Synthesis*, 2005, 1550. e) E. R. Walker, S. Y. Leung, and A. G. Barrett, *Tetrahedron Lett.*, 2005, 46, 6537.
- 5. D. J. Dodsworth, M. Pia-Calcagno, U. Ehrmann, A. M. Quesada, O. Nuñez S., and P. G. Sammes, J. *Chem. Soc.*, *Perkin Trans. 1*, 1983, 1453.
- 6. B. Wuensch, G. Hoefner, and G. Bauschke, Arch. Pharm., 1993, 326, 513.