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# Phosphorus, Sulfur, and Silicon and the Related Elements

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### Synthesis of N-Protected o-Hydroxyl-phenyl—a-Aminophosphonic Monoester

Jianfeng Zhang  $^{\rm a}$  , Zhiwei Miao  $^{\rm a}$  , Zhanwei Cui  $^{\rm a}$  & Ruyu Chen  $^{\rm a}$ 

<sup>a</sup> State Key Laboratory of Elemento-Organic Chemistry, Research Institute of Elemento-Organic Chemistry, Nankai University, Tianjin, China Published online: 20 Jun 2008.

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## Synthesis of N-Protected o-Hydroxyl-phenyl— $\alpha$ -Aminophosphonic Monoester

### Jianfeng Zhang, Zhiwei Miao, Zhanwei Cui, and Ruyu Chen

State Key Laboratory of Elemento-Organic Chemistry, Research Institute of Elemento-Organic Chemistry, Nankai University, Tianjin, China

A series of N-protected O-hydroxylphenyl  $\alpha$ -aminophosphonic monoesters were synthesized via three-component Mannich-type reactions of phosphoramides, aldehydes (ketones) and 2-chlorobenzo [1,3,2] dioxaphospholes under solvent free and catalystfree conditions, followed by hydrolysis. It is an efficient and green method to the synthesis of N-phosphoramino O-hydroxyphenyl 1-aminoalkylphosphonic monoesters with high yields.

Keywords  $\alpha\text{-aminophosphonic monoesters;}$  Mannich-type reaction; no catalyst and solvent-free

 $\alpha$ -Aminophosphonic and phosphinic acids are the phosphorous analogues of  $\alpha$ -aminocarboxylic acids, and therefore have biological importance both in themselves and as building blocks for peptides.<sup>1</sup> They have acquired great attention in synthetic organic chemistry and a number of synthetic methods have been developed during past two decades. Of these methods, nucleophilic addition of phosphites to imines catalyzed by a base or an acid is the most convenient one. A variety of metal halides such as TiCl<sub>4</sub><sup>2</sup>, InCl<sub>3</sub><sup>3</sup>, TaCl<sub>5</sub>-SiO<sub>2</sub><sup>4</sup>, Mg(ClO<sub>4</sub>)<sub>2</sub><sup>5</sup> have been used as Lewis acid catalysts in methylene chloride or other organic solvent to promote this addition. To avoid these disadvantages of the use of organic solvents, a couple of modifications using montmorillonite clay<sup>6</sup> and alumina,<sup>7</sup>BiNO<sub>3</sub>·5H<sub>2</sub>O<sup>8</sup> in dry media under microwave irradiation have

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Address correspondence to Zhiwei Miao State Key Laboratory of Elemento-Organic Chemistry, Research Institute of Elemento-OrganicChemistry, Nankai University, Tianjin 300071, China. E-mail: miaozhiwei@nankai.edu.cn

4	$R_1$	$R_2$	Yield (%)	4	$R_1$	$R_2$	Yield (%)
4a	p-Cl-C <sub>6</sub> H <sub>5</sub>	Н	80	4f	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>5</sub>	Н	90
<b>4b</b>	o-Cl-C <sub>6</sub> H <sub>5</sub>	н	92	4g	p-NO <sub>2</sub> -C <sub>6</sub> H <sub>5</sub> H	Н	93
<b>4c</b>	p-Br-C <sub>6</sub> H <sub>5</sub>	н	89	4 <b>h</b>	$(CH_2)_5$	_	84
4d	o-Br-C <sub>6</sub> H <sub>5</sub>	н	85	<b>4i</b>	$CH_3$	$CH_3$	87
<b>4e</b>	$C_6H_5$	н	91	—	_	_	—

TABLE I Preparation of  $\alpha$ -aminophosphonic monoesters 4a-i

been reported. In 2002, Ranu<sup>9</sup> reported a more practical green alterative for the synthesis of  $\alpha$ -aminophosphonates by a three-component condensation of carbonyl compounds (aldehydes and ketones), amines and diethyl phosphite at 75–80°C in neat without any solvent and catalyst. We would like to disclose a practical and green method for the synthesis of N-protected *o*-hydroxyl-phenyl- $\alpha$ -aminophosphonic monoester under the condition of no catalyst and solvent-free (Scheme 1).



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