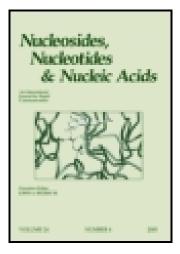
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Nucleosides and Nucleotides

Publication details, including instructions for authors and subscription information: http://www.tandfonline.com/loi/lncn19

Synthesis of N-Glycosylated Pyridines as New Antiviral Agents

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To cite this article: E. S. Ibrahim , G. E.H. Elgemeie , M. M. Abbasi , Y. A. Abbas , M. A. Elbadawi & A. M.E. Attia (1995) Synthesis of N-Glycosylated Pyridines as New Antiviral Agents, Nucleosides and Nucleotides, 14:6, 1415-1423, DOI: <u>10.1080/15257779508010701</u>

To link to this article: http://dx.doi.org/10.1080/15257779508010701

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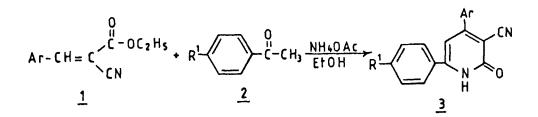
SYNTHESIS OF N-GLYCOSYLATED PYRIDINES AS NEW ANTIVIRAL AGENTS

E.S. Ibrahim⁴, G.E.H. Elgemeie^{4,+,b}, M.M. Abbasi^c, Y.A. Abbas^d, M.A. Elbadawi^c and A.M.E. Attia^d, ⁴Chemistry Department, Faculty of Science, Suez Canal University, Egypt ⁵Chemistry Department, Faculty of Science, Cairo University (Bani Suef Branch), Egypt ⁶Chemistry Department, Faculty of Science, Tanta University, Egypt ⁶Chemistry Department, Faculty of Education, Tanta University (Kafr El-Sheikh), Egypt

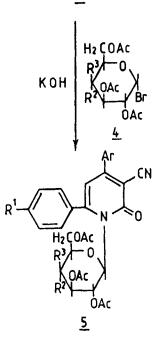
Abstract: A series of 3-cyanopyridine glycosides have been synthesized and evaluated for their inhibitory activity against human immunodeficiency virus (HIV) replication in MT-4 cells. Among the 3-cyanopyridine glycosides 6-(p-methylphenyl) and 6-(p-aminophenyl) were the most selective inhibitors of HIV replication.

3-Deazauridine and 3-deazacytidine were found to exert marked inhibitory effects on the growth of neoplastic and bacterial culture¹ and also have antiviral activity against RNA viruses². Also 3-deazauridine apparently blocks the action of an enzyme which catalyzes overall amination of UTP to CTP⁴. The 3-deazapyrimidine nucleosides constitute another logical class of analogues with potential biological activity^{4,5}. As a part of our program directed for development of new, simple and efficient procedures for the synthesis of antimetabolites⁶⁻¹¹, we report in this paper a novel synthesis of 3-deazapyrimidine glycosides 5 utilizing our previously reported pyridine-2(1*H*)-ones 3¹² as starting materials. Compounds 3 can be prepared by the reaction of α , β -unsaturated nitriles 1 with acetophenones 2 in boiling ethanol containing ammonium acetate. Compounds 3 reacted with 2,3,4,6-tetra-O-acetyl- α -D-gluco- and galactopyranosyl bromides 4 in the presence of aqueous potassium hydroxide to give the corresponding *N*-glucosides 5a-h and *N*-galactosides 5i-p, respectively. The structures of the reaction products 5 were established and confirmed on the basis of their elemental analyses and spectral data (MS, IR, ¹H NMR, ¹³C NMR). Thus the analytical data for 5n revealed a molecular formula C₃₁H₃₀N₂O₁₁ (m/z 606). The ¹H NMR spectrum showed a doublet at δ 6.6 ppm assigned to the anomeric proton of the galactose moiety with a spin-spin

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NH3 / CH3OH



Ar CN H ₂ COH R ³ O OH	_
<u>е</u> В ₃ , Г	

Ar	R¹	R²	R³		Ar	R۱	R²	R³
Phenyl	н	OAc	н	6a	Phenyl	н	он	н
Phenyl	СН3	OAc	н	b	Phenyl	CH3	он	н
Phenyl	OCH3	OAc	Н	C	Phenyl	OCH3	ОН	н
Phenyl	NH ₂	OAc	н	d	Phenyl	NH ₂	ОН	н
2-Furyl	Н	OAc	н	e	2-Furyl	н	ОН	н
2-Furyl	CH3	OAc	н	f	2-Furyl	CH₃	ОН	Н
2-Furyl	OCH3	OAc	н	9	2-Furyl	OCH3	ОН	н
2-Furyl	NHz	OAc	Н	h	2-Furyl	NH₂	ОН	н
Phenyl	н	Н	OAc	i	Phenyl	Н	Н	ОН
Phenyl	CH₃	н	OAc	j	Phenyl	СН3	Н	ОН
Phenyl	OCH3	н	OAc	k	Phenyl	OCH3	Н	ОН
Phenyl	NH₂	н	OAc	I	Phenyl	NH₂	Н	ОН
2-Furyl	н	н	OAc	m	2-Furyl	н	н	ОН
2-Furyl	CH3	н	OAc	n	2-Furyi	CH3	Н	ОН
2-Furyl	OCH3	н	OAc	0	2-Furyl	\mathbf{OCH}_{3}	н	ОН
2-Furyi	NH2	н	OAc	p	2-Furyl	NH₂	Н	ОН
	Phenyl Phenyl Phenyl 2-Furyl 2-Furyl 2-Furyl Phenyl Phenyl Phenyl Phenyl 2-Furyl 2-Furyl 2-Furyl	PhenylHPhenylCH3PhenylOCH3PhenylNH22-FurylH2-FurylCH32-FurylOCH32-FurylNH2PhenylHPhenylCH3PhenylCH3PhenylOCH3PhenylNH22-FurylNH22-FurylH2-FurylH2-FurylH2-FurylCH3PhenylOCH3PhenylOCH3OCH3OCH32-FurylCH32-FurylOCH3	PhenylHOAcPhenylCH3OAcPhenylOCH3OAcPhenylNH2OAc2-FurylHOAc2-FurylCH3OAc2-FurylOCH3OAc2-FurylOCH3OAc2-FurylNH2OAc2-FurylNH2OAcPhenylHHPhenylCH3HPhenylOCH3HPhenylNH2H2-FurylHH2-FurylHH2-FurylCH3H2-FurylCH3H2-FurylCH3H	PhenylHOAcHPhenylCH3OAcHPhenylOCH3OAcHPhenylNH2OAcH2-FurylHOAcH2-FurylCH3OAcH2-FurylCH3OAcH2-FurylCH3OAcH2-FurylOCH3OAcH2-FurylNH2OAcHPhenylHHOAcPhenylCH3HOAcPhenylOCH3HOAcPhenylNH2HOAc2-FurylHHOAc2-FurylHHOAc2-FurylCH3HOAc2-FurylCH3HOAc2-FurylCH3HOAc2-FurylCH3HOAc2-FurylCH3HOAc	PhenylHOAcH6aPhenylCH3OAcHbPhenylOCH3OAcHcPhenylOCH3OAcHd2-FurylHOAcHe2-FurylCH3OAcHf2-FurylOCH3OAcHg2-FurylOCH3OAcHhPhenylOCH3OAcHhPhenylOCH3OAcHhPhenylHHOAcjPhenylOCH3HOAckPhenylOCH3HOAcI2-FurylHHOAcn2-FurylCH3HOAcn2-FurylCH3HOAcn2-FurylOCH3HOAcn	PhenylHOAcH6aPhenylPhenylCH3OAcHbPhenylPhenylOCH3OAcHcPhenylPhenylOCH3OAcHdPhenylPhenylNH2OAcHdPhenyl2-FurylHOAcHe2-Furyl2-FurylCH3OAcHf2-Furyl2-FurylOCH3OAcHg2-Furyl2-FurylNH2OAcHh2-FurylPhenylHHOAciPhenylPhenylHHOAcjPhenylPhenylOCH3HOAckPhenylPhenylNH2HOAciPhenylPhenylNH2HOAciPhenylPhenylNH2HOAciPhenyl2-FurylHHOAciPhenyl2-FurylCH3HOAcn2-Furyl2-FurylCH3HOAcn2-Furyl2-FurylCH3HOAco2-Furyl2-FurylOCH3HOAco2-Furyl2-FurylOCH3HOAco2-Furyl2-FurylOCH3HOAco2-Furyl	PhenylHOAcH6aPhenylHPhenylCH3OAcHbPhenylCH3PhenylOCH3OAcHcPhenylOCH3PhenylNH2OAcHdPhenylNH22-FurylHOAcHe2-FurylH2-FurylCH3OAcHf2-FurylCH32-FurylCH3OAcHg2-FurylCH32-FurylOCH3OAcHg2-FurylOCH32-FurylNH2OAcHh2-FurylNH2PhenylHHOAciPhenylHPhenylHHOAciPhenylHPhenylCH3HOAckPhenylCH3PhenylHHOAciPhenylHPhenylOCH3HOAckPhenylOCH3PhenylNH2HOAciPhenylNH22-FurylHHOAcm2-FurylH2-FurylHHOAcm2-FurylH2-FurylCH3HOAcn2-FurylH2-FurylCH3HOAcn2-FurylH2-FurylCH3HOAcn2-FurylCH32-FurylOCH3HOAcn2-FurylCH32-FurylOCH3H <t< th=""><th>PhenylHOAcH6aPhenylHOHPhenylCH3OAcHbPhenylCH3OHPhenylOCH3OAcHcPhenylOCH3OHPhenylOCH3OAcHcPhenylOCH3OHPhenylNH2OAcHdPhenylNH2OH2-FurylHOAcHe2-FurylHOH2-FurylCH3OAcHf2-FurylCH3OH2-FurylOCH3OAcHg2-FurylOCH3OH2-FurylOCH3OAcHg2-FurylOCH3OH2-FurylNH2OAcHh2-FurylNH2OHPhenylHHOAciPhenylHHPhenylHHOAciPhenylHHPhenylCH3HOAckPhenylOCH3HPhenylOCH3HOAckPhenylOCH3HPhenylNH2HOAciPhenylNH2HPhenylNH2HOAciPhenylNH2HPhenylNH2HOAciPhenylNH2HPhenylNH2HOAciPhenylNH2H2-FurylHHOAcm2-FurylHH2-FurylH</th></t<>	PhenylHOAcH6aPhenylHOHPhenylCH3OAcHbPhenylCH3OHPhenylOCH3OAcHcPhenylOCH3OHPhenylOCH3OAcHcPhenylOCH3OHPhenylNH2OAcHdPhenylNH2OH2-FurylHOAcHe2-FurylHOH2-FurylCH3OAcHf2-FurylCH3OH2-FurylOCH3OAcHg2-FurylOCH3OH2-FurylOCH3OAcHg2-FurylOCH3OH2-FurylNH2OAcHh2-FurylNH2OHPhenylHHOAciPhenylHHPhenylHHOAciPhenylHHPhenylCH3HOAckPhenylOCH3HPhenylOCH3HOAckPhenylOCH3HPhenylNH2HOAciPhenylNH2HPhenylNH2HOAciPhenylNH2HPhenylNH2HOAciPhenylNH2HPhenylNH2HOAciPhenylNH2H2-FurylHHOAcm2-FurylHH2-FurylH

N-GLYCOSYLATED PYRIDINES

Compd	EC _{s0} *	IC ₅₀ ^b	TI°
	μM	μΜ	(ratio IC ₅₀ /EC ₅₀)
5a	<u></u>	> 16.5	= <u>14 de anim + 20 d'a ser a 14 din a de a ser a ser (-</u>
5b		>11.7	
5c		>15.8	
5d		>11.6	
5e		>16.8	
5f		>8.6	
5g		>11.6	
5h	11.7	>11.8	>1
5j		>16.5	
5k		>11.4	
5m		>11.9	
5n		>16.4	
50		>11.6	
ба		>21.4	
6f	8.03	> 82.1	>10.2
бі		> 59.9	
6j	18.6	>22.3	>1.2
6k		>21.5	

 Table 1:
 Comparative potency and selectivity of 3-cyanopyridine-2-one glucosides and galactosides analogues as inhibitors of HIV replication in MT-4 cells.

a. Approximate values for 50% effective concentration of MT-4 cells against the cytopathic effect of HIV (EC₅₀)

- b. Inhibitor concentration for 50% (IC₅₀).
- c. The rapeutic index TI (IC_{50}/EC_{50}) .

coupling constant equal to 8.9 Hz which corresponds to a diaxial orientation for the H-1' and H-2' protons indicating the presence of only β -configuration. The other six protons of the galactopyranosyl ring resonate in the δ 4.12-5.64 ppm region. The remaining four acetoxy groups appeared as four singlets at δ 1.78-2.18 ppm region, while the methyl group of the aglycone resonated at δ 2.41 ppm. The ¹³C NMR spectrum of 5n was characterized by a signal at δ 87.1 ppm corresponding to the C-1' atom of the β -D-galactopyranose. Four signals appeared at δ 170.5-169.4 ppm due to the four acetoxy carbonyl carbon atoms of galactose, while a signal appearing at δ 20.33-20.26 ppm was attributed to the galactose methyl carbons. Another five

Compd.	MP	Yield	Mol. formula	Found/Calcd. (%)			
	°C	%		С	н	N	m/z
5a	180	74	C ₃₂ H ₃₀ N ₂ O ₁₀	63.9	5.1	4.6	602
			(602)	63.8	5.0	4.7	
5b	196	76	C ₃₃ H ₃₂ N ₂ O ₁₀	64.6	5.1	4.7	616
			(616)	64.3	5.2	4.5	
5c	192	73	$C_{33}H_{32}N_2O_{11}$	62.9	5.3	4.5	632
			(632)	62.7	5.1	4.4	
5d	177	70	C ₃₂ H ₃₁ N ₃ O ₁₀	62.5	5.2	6.9	617
			(617)	62.2	5.0	6.8	
5e	222	71	$C_{30}H_{28}N_2O_{11}$	61.0	4.8	4.9	592
			(592)	60.8	4.7	4.7	
5f	206	75	$C_{31}H_{30}N_2O_{11}$	61.5	4.9	4.8	606
			(606)	61.4	5.0	4.6	
5g	190	70	$C_{31}H_{30}N_2O_{12}$	60.0	4.8	4.7	622
			(622)	59.8	4.8	4.5	
5h	226	67	$C_{30}H_{29}N_{3}O_{11}$	59.4	4.9	7.1	607
			(607)	59.3	4.8	6.9	
5i 140	146	72	$C_{32}H_{30}N_2O_{10}$	64.0	4.8	4.5	
			(602)	63.8	5.0	4.7	
5j 183 74	74	$C_{33}H_{32}N_2O_{10}$	64.1	5.2	4.4	616	
			(616)	64.3	5.2	4.5	
5k	168	71	$C_{33}H_{32}N_2O_{11}$	62.9	5.2	4.1	632
			(632)	62.7	5.1	4.4	
51	105	68	$C_{32}H_{31}N_3O_{10}$	62.1	5.1	7.0	
			(617)	62.2	5.0	6.8	
5m	180	70	$C_{30}H_{28}N_2O_{11}$	60.9	4.6	4.5	592
			(592)	60.8	4.7	4.7	
5n	220	73	$C_{31}H_{30}N_2O_{11}$	61.6	5.0	4.6	606
			(606)	61.4	5.0	4.6	
50	210	69	$C_{31}H_{30}N_2O_{12}$	59.9	4.9	4.6	
			(622)	59.8	4.8	4.5	
5p	170	65	$C_{30}H_{29}N_3O_{11}$	59.6	5.0	6.9	
			(607)	59.3	4.8	6.9	
6a	296	83	$C_{24}H_{22}N_2O_6$	66.6	5.0	6.7	434
			(434)	66.4	5.1	6.5	

Table 2: Physical and analytical data for compounds 5a-p and 6a-p.

Table 2. (Continued)

6b	263	85	$C_{23}H_{24}N_2O_6$	67.2	5.4	6.5	448
			(448)	67 .0	5.4	6.3	
6c	256	84	$C_{25}H_{24}N_2O_7$	64.5	5.3	6.3	464
			(464)	64.7	5.2	6.0	
6d	180	83	$C_{24}H_{23}N_3O_6$	64.3	5.1	9.2	
			(449)	64.1	5.1	9.4	
6e	295	84	$C_{22}H_{20}N_2O_7$	62.4	4.5	6.8	424
			(424)	62.3	4.7	6.6	
6f	297	85	$C_{23}H_{22}N_2O_7$	63.1	5.0	6.3	438
			(438)	63.0	5.0	6.4	
6g	285	81	$C_{23}H_{22}N_2O_8$	60.6	4,9	6.5	
			(454)	60.8	4.8	6.2	
6h	190	83	$C_{22}H_{21}N_3O_7$	60.0	4.8	9.7	439
			(439)	60.1	4.8	9.6	
6i	287	84	$C_{24}H_{22}N_2O_6$	66.5	5.3	6.6	
			(434)	66.4	5.1	6.5	
6j	261	85	$C_{25}H_{24}N_2O_6$	66.9	5.4	6.5	448
			(448)	67.0	5.4	6.3	
6k	264	82	$C_{25}H_{24}N_2O_7$	64.6	5.4	6.1	
			(464)	64.7	5.2	6.0	
61	173	84	C24H23N3O6	64.2	5.3	9.5	449
			(449)	64,1	5.1	9.4	
6m	288	83	$C_{22}H_{20}N_2O_7$	62.5	4.6	6.8	424
			(424)	62.3	4.7	6.6	
6n	285	84	$C_{23}H_{22}N_2O_7$	63.2	5.2	6.6	438
			(438)	63.0	5.0	6.4	
60	290	82	$C_{23}H_{22}N_2O_8$	60.9	4.9	6.4	454
			(454)	60.8	4.8	6.2	
6р	288	84	$C_{22}H_{21}N_{3}O_{7}$	60.3	5.0	9.5	
•			(439)	60.1	4.8	9.6	

Compd.	IR(KBr)/cm ⁻¹	¹ H NMR (DMSO) δ/ppm
5a	2220 (CN), 1752 (CO ester), 1648 (CO pyridone)	1.82-2.08 (4s, 12H, 4CH ₃ CO), 4.21 (m, 2H, H-6',6"), 4.60 (m, 1H, H-5'), 5.32 (m, 2H, H-4' and H-3'), 5.78 (m, 1H, H-2'), 6.80 (d, $J_{1,2} = 8.38$ Hz, 1H, H-1'), 7.73 (s, 1H, Pyridine H-5), 7.90 (s, 5H, Ar-H), 8.13 (s, 5H, Ar-H).
5b	2222 (CN), 1750 (CO ester), 1636 (CO pyridone)	1.76-2.10 (4s, 12H, 4CH ₃ CO), 2.40 (s, 3H, CH ₃), 4.16 (m, 2H, H-6', 6 [*]), 4.53 (m, 1H, H-5'), 5.21 (m, 2H, H-4' and H-3'), 5.70 (t, 1H, H-2'), 6.76 (d, $J_{1.2} = 8.63$ Hz, 1H, H-1'), 7.39 (d, 2H, Ar-H), 7.65 (s, 3H, Ar-H), 7.76 (s, 1H, pyridine H-5), 7.96 (s, 2H, Ar-H), 8.30 (d, 2H, Ar-H)
5c	2218 (CN), 1755 (CO ester), 1640 (CO pyridone)	1.85-2.12 (4s, 12H, 4CH ₃ CO), 3.90 (s, 3H, OCH ₃), 4.19 (m, 2H, H-6', 6 [*]), 4.56 (t, 1H, H-5'), 5.18 (m, 2H, H-4' and H-3'), 5.70 (d, 1H, H-2'), 6.65 (d, $J_{1\cdot2} = 8.55$ Hz, 1H, H-1'), 7.11 (d, 2H, Ar-H), 7.66 (s, 3H, Ar-H), 7.77 (s, 1H, pyridine H-5), 7.95 (s, 2H, Ar-H), 8.30 (d, 2H, Ar-H).
5d	3460, 3380 (NH ₂), 2220 (CN), 1750(CO ester), 1642(CO pyridone)	1.88-2.09 (4s, 12H, 4CH ₂ CO), 4.25 (m, 2H, H-6',6"), 4.60 (m, 1H, H-5'), 5.28 (m, 2H, H-4' and H-3'), 5.80 (t, 1H, H-2'), 6.00 (s, 2H, NH ₂), 6.70 (d, $J_{1\cdot2} = 8.35$ Hz, 1H, H-1'), 7.64 (s, 5H, Ar-H), 7.71 (s, 1H, pyridine H-5), 8.23 (d, 4H, Ar-H).
5e	2220 (CN), 1756 (CO ester), 1610 (CO pyridone)	1.67-2.11 (4s, 12H, 4CH ₃ CO), 4.12 (d, 2H, H-6',6"), 4.31 (m, 1H, H-5'), 5.32 (m, 2H, H-4' and H-3'), 5.51 (m, 1H, H-2'), 6.38 (d, $J_{1,2} = 8.46$ Hz, 1H, H-1'), 6.72 (s, 1H, furan H-4), 7.58 (s, 5H, Ar-H), 7.71 (s, 1H, pyridine H-5), 8.03 (s, 1H, furan H-3), 8.18 (s, 1H, furan H-5).
51	2222 (CN), 1750 (CO ester), 1600 (CO pyridone)	1.64-2.05 (4s, 12H, 4CH ₃ CO), 2.46 (s,3H, CH ₃), 4.08 (m, 2H, H-6',6" and 1H, H-5'), 4.36 (m, 1H, H-4'), 5.34 (m, 2H, H-3' and H-2'), 6.32 (d, $J_{1:2} =$ 8.64 Hz, 1H, H-1'), 6.70 (s, 1H, furan H-4), 7.41 (d, 4H, Ar-H), 7.69 (s, 1H, pyridine H-5), 7.80 (s, 1H, furan H-3), 8.00 (d, 1H, furan H-5).
5g	2218 (CN), 1752 (CO ester), 1610 (CO pyridone)	1.62-2.10 (4s, 12H, 4CH ₃ CO), 3.88 (s, 3H, OCH ₃), 4.05 (m, 2H, H-6',6" and 1H, H-5'), 4.34 (m, 1H, H-4'), 5.30 (t, 1H, H-3'), 5.48 (m, 1H, H-2'), 6.31 (d, $J_{1.2}$ = 8.66 Hz, 1H, H-1'), 6.62 (s, 1H, furan H-4), 7.10 (d, 4H, Ar-H), 7.67 (s, 1H, pyridine H-5), 7.94 (s, 1H, furan H-3), 8.16 (d, 1H, furan H-5)
5h	3460, 3360 (NH ₂), 2220 (CN), 1760 (CO ester), 1610 (CO pyridone)	1.94-2.16 (4s, 12H, 4CH ₃ CO), 4.23 (m, 2H, H-6', 6"), 4.51 (s, 1H, H-5'), 5.22 (m, 2H, H-4' and H-3'), 5.72 (t, 1H, H-2'), 6.00 (s, 2H, NH ₂), 6.70 (d, $J_{1,2} = 8.56$ Hz, 1H, H-1'), 6.85 (s, 1H, furan H-4), 7.80 (d, 4H, Ar-H), 7.73 (s, 1H, pyridine H-5), 7.91 (s, 1H, furan H-3), 8.10 (s, 1H, furan H-5).
5i	2220 (CN), 1760 (CO ester), 1650 (CO pyridone)	1.86-2.12 (4s, 12H, 4CH ₃ CO), 4.21 (m, 2H, H-6', 6"), 4.63 (m, 1H, H-5'), 5.44 (m, 2H, H-4' and H-3'), 5.60 (t, 1H, H-2'), 6.61 (d, $J_{1:2} = 8.38$ Hz, 1H, H-1'), 6.90 (s, 1H, furan H-4), 7.65 (s, 5H, Ar-H), 7.78 (s, 1H, pyridine H-5), 8.20 (d, 1H, furan H-3), 8.41 (s, 1H, furan H-5)
5j	2222 (CN), 1758 (CO ester), 1650 (CO pyridone)	1.75-2.15 (4s, 12H, 4CH ₃ CO), 2.40 (s, 3H, CH ₃), 4.15 (m, 2H, H-6',6" and 1H, H-5'), 4.65 (t, 1H, H-4'), 5.57 (m, 2H, H-3' and H-2'), 6.65 (d, $J_{1,2} = 8.71$ Hz, 1H, H-1'), 6.85 (s, 1H, furan H-4), 7.35 (d, 4H, Ar-H), 7.75 (d, 1H, furan H-3), 8.10 (s, 1H, pyridine H-5), 8.34 (d, 1H, furan H-5).
5m	2218 (CN), 1754 (CO ester), 1650 (CO pyridone)	1.86-2.16 (4s, 12H, 4CH ₃ CO), 4.22 (m, 2H, H-6',6"), 4.64 (m, 1H, H-5'), 5.41 (m, 2H, H-4' and H-3'), 5.60 (t, 1H, H-2'), 6.58 (d, $J_{12} = 8.68$ Hz, 1H, H-1'), 6.88 (s, 1H, furan H-4), 7.65 (s, 5H, Ar-H), 7.76 (s, 1H, pyridine H-5), 8.20 (d, 1H, furan H-3), 8.34 (s, 1H, furan H-5).

 Table 3:
 Spectral data for selected compounds listed in Table 2.

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Table 3: (Continued)

5n	2220 (CN), 1760 (CO ester), 1650 (CO pyridone)	1.78-2.18 (48, 12H, 4CH ₃ CO), 2.41 (s, 3H, CH ₃), 4.12 (m, 2H, H-6',6" and 1H, H-5'), 4.65 (t, 1H, H-4'), 5.51 (m, 2H, H-3' and H-2'), 6.60 (d, $J_{1,2}$ = 8.90 Hz, 1H, H-1') 6.85 9s, 1H, furan H-4), 7.32 (d, 4H, Ar-H), 7.78 (s, 1H, pyridine H-5), 8.10 (d, 1H, furan H-3), 8.33 (d, 1H, furan H-5).
6c	3600-3150 (OH), 2220 (CN), 1630 (CO pyridone)	3.25-3.78 (m, 6H, H-6',6", H-5', H-4', H-3' and H-2'), 3.88 (s, 3H, OCH,), 4.67 (t, 1H, 2'-OH), 5.18 (d, 1H, 3'-OH), 5.24 (s, 1H, 4'-OH), 5.50 (d, 1H, 6'-OH), 6.20 (d, J_{12} = 8.64 Hz, 1H, H-1'), 7.12 (d, 2H, Ar-H), 7.63 (m, 3H, Ar-H), 7.74 (m, 2H, Ar-H), 7.81 (s, 1H, pyridine H-5), 8.25 (d, 2H, Ar-H)
6d	3650-3200 (OH), 2222 (CN), 1635 (CO pyridone)	3.21-3.74 (m, 6H, H-6', 6", H-5', H-4', H-3' and H-2'), 4.65 (t, 1H, 2'-OH), 5.15 (d, 1H, 3'-OH), 5.21 (s, 1H, 4'-OH), 4.45 (d, 1H, 6'-OH), 5.88 (s, 2H, NH ₂), 6.18 (d, J ₁₋₂ = 8.55 Hz, 1H, H-1'), 6.68 (d, 2H, Ar-H), 7.56 (m, 3H, Ar-H), 7.62 (s, 1H, pyridine H-5), 7.72 (m, 2H, Ar-H), 8.00 (d, 2H, Ar-H).
6e	3600-3180 (OH) 2218 (CN) 1610 (CO Pyridone)	3.18-3.65 (m, 6H, H-6',6", H-5', H-4', H-3' and H-2'), 4.61 (t, 1H, 2'-OH), 5.18 (d, 2H, 3'-OH and 4'-OH), 5.46 (s, 1H, 6'-OH), 6.14 (d, $J_{1,2} = 8.63$ Hz, 1H, H-1'), 6.85 (dd, 1H, furan H-4), 7.54 (t, 3H, Ar-H), 7.72 (d, 1H, furan H- 3), 8.08 (s, 1H, pyridine H-5), 8.11 (d, 2H, Ar-H), 8.24 (dd, 1H, furan H-5).
61	3650-3150 (OH) 2220 (CN), 1610 (CO pyridone)	2.32 (s, 3H, CH ₃), 3.26-3.85 (m, 6H, H-6',6", H-5', H-4', H-3' and H-2'), 4.65 (t, 1H, 2'-OH), 5.15 (d, 2H, 3'-OH and 4'-OH), 5.45 (s, 1H, 6'-OH), 6.15 (d, $J_{1,2} = 8.52$ Hz, 1H, H-1'), 6.85 (m, 1H, furan H-4), 7.35 (d, 4H, Ar- H), 7.71 (d, 1H, furan H-3), 8.05 (s, 1H, pyridine H-5), 8.15 (d, 1H, furan H- 5).
бg	3600-3200 (OH) 2222 (CN), 1600 (CO pyridone)	3.21-3.77 (m, 6H, H-6',6", H-5', H-4', H-3' and H-2'), 3.86 (s, 3H, OCH ₃), 4.65 (s, 1H, 2'-OH), 5.20 (s, 2H, 3'-OH and 4'-OH), 5.54 s, 1H, 6'-OH), 6.18 (d, $J_{1,2} = 8.38$ Hz, 1H, H-1'), 6.88 (dd, 1H, furan H-4), 7.18 (d, 2H, Ar-H), 7.65 (d, 1H, furan H-3), 8.00 (s, 1H, pyridine H-5), 8.13 (m, 1H, furan H-5), 8.21 (d, 2H, Ar-H).
6j	3640-3180 (OH), 2220 (CN), 1630 (CO pyridone)	2.28 (s, 3H, CH ₃), 3.21-3.78 (m, 6H, H-6',6", H-5', H-4', H-3' and H-2'), 4.63 (t, 1H, 2'-OH), 5.16 (d, 1H, 3'-OH), 5.21 (s, 1H, 4'-OH), 5.42 (d, 1H, 6'-OH), 6.18 (d, $J_{1:2} = 9.30$ Hz, 1H, H-1'), 7.38 (d, 2H, Ar-H), 7.60 (m, 3H, Ar-H), 7.78 (m, 2H, Ar-H), 7.84 (s, 1H, pyridine H-5), 8.18 (d, 2H, Ar-H).
бк	3660-3200 (OH), 2220 (CN), 1620 (CO pyridone)	3.31-3.78 (m, 6H, H-6',6", H-5', H-4', H-3' and H-2'), 3.86 (s, 3H, OCH ₃), 4.61 (t, 1H, 2'-OH), 5.18 (d, 1H, 3'-OH), 5.25 (s, 1H, 4'-OH), 5.50 (d, 1H, 6'-OH), 6.20 (d, $J_{1,2} = 8.58$ Hz, 1H, H-1'), 7.11 (d, 2H, Ar-H), 7.60 (m, 3H, Ar-H), 7.70 (s, 1H, pyridine H-5), 7.81 (d, 2H, Ar-H), 8.05 (d, 2H, Ar-H).
ର	3650-3180 (OH, NH ₂), 2220 (CN), 1620 (CO pyridone)	3.28-3.87 (m, 6H, H-6', 6", H-5', H-4', H-3' and H-2'), 4.65 (t, 1H, 2'-OH), 5.12 (d, 1H, 3'-OH), 5.24 (s, 1H, 4'-OH), 5.48 (d, 1H, 6'-OH), 5.96 (s, 2H, NH ₂), 6.21 (d, $J_{1.2}$ = 8.66 Hz, 1H, H-1'), 6.68 (d, 2H, Ar-H), 7.58 (m, 3H, Ar-H), 7.68 (s, 1H, pyridine H-5) 7.85 (m, 2H, Ar-H), 8.05 (d, 2H, Ar-H).
60	3660-3200 (OH), 2222 (CN), 1625 (CO pyridone)	2.40 (s, 3H, CH ₃), 3.18-3.66 (m, 6H, H-6',6", H-5', H-4', H-3' and H-2'), 4.62 (t, 1H, 2'-OH), 5.18 (d, 2H, 3'-OH and 4'-OH), 5.45 (s, 1H, 6'-OH), 6.15 (d, $J_{1,2} = 8.48$ Hz, 1H, H-1'), 6.88 (m, 1H, furan H-4), 7.40 (d, 4H, Ar- H), 7.69 (d, 1H, furan H-3), 8.05 (s, 1H, pyridine H-5), 8.15 (d, 1H, furan H- 5).
бр	3600-3200 (OH, NH ₂), 2220 (CN), 1600 (CO pyridone)	3.22-3.68 (m, 6H, H-6', 6", H-5', H-4', H-3' and H-2'), 4.58 (l, 1H, 2'-OH), 5.08 (d, 1H, 3'-OH), 5.17 (s, 1H, 4'-OH), 5.49 (d, 1H, 6'-OH), 5.90 (s, 2H, NH ₂), 6.16 (d, $J_{1,2} = 8.55$ Hz, 1H, H-1'), 6.74 (d, 2H, Ar-H), 6.88 (dd, 1H, furan H-4), 7.60 (d, 1H, furan H-3), 7.78 (s, 1H, pyridine H-5), 8.00 (d, 2H, Ar-H), 8.14 (d, 1H, furan H-5).

signals at δ 61.7, 67.7, 67.9, 70.3 and 71.3 ppm were assigned to C-6', C-4', C-2', C-3' and C-5', respectively. On the other hand, the signal of the carbonyl carbon atom of aglycone appeared at δ 162.7 ppm and the nitrile carbon atom appeared at δ 115.1 ppm. The preparation of crystalline 1-(β -D-glycopyranosyl)3-cyanopyridine-2-ones **6a-p** was achieved by removing the blocking acetyl groups on treatment with methanolic ammonia at 0°C. The structures of compounds **6** were established on the basis of elemental analyses and spectral data. Thus, the analytical data for **6f** revealed a molecular formula $C_{23}H_{22}N_2O_7$ (m/z 438). The ¹HNMR spectra showed the anomeric proton as a doublet at δ 6.18 ppm ($J_{1,2} = 9.3$ Hz) indicating the presence of only the B-configuration. The other six glucose protons appeared as a multiplet at δ 3.21-3.78 ppm, while the four hydroxy groups of the glucose moiety resonated at δ 4.63-5.48 ppm (exchangeable by D₂O). The ¹³C NMR spectra was characterized by a signal at δ 96.8 ppm corresponding to the C-1' atom of β -D-glucopyranose. Another five signals at δ 60.6, 69.7, 72.9, 77.1 and 78.2 ppm were assigned to C-6', C-4', C-2', C-3' and C-5' of the glucose moiety, respectively.

In summary, we have achieved a regiospecific synthesis of interesting 3-deazapyrimidine nucleosides by the reaction of pyridine-2(1*H*)-ones with α -halomonosugars. These nucleosides can be utilized as an excellent starting materials for the synthesis of other carbohydrate derivatives.

Antiviral Activity:

The anti-HIV activity and cytotoxicity of some 3-cyano-4 and 6-disubstituted-2-pyridone glucosides and galactosides are shown in Table 1. Among these types of glycosides the 4-furyl-6(p-methylphenyl) glucoside **6f** turned out to be the most selective anti-HIV agent, followed by 4-phenyl-6(p-methylphenyl) galactosides **6j** and 4-furyl-6 (p-aminophenyl) glucose tetraacetate **5h**., while; 4-furyl-6-phenyl- and 4, 6-diphenyl-, 4-furyl-6 (p-methylphenyl)-, and 4-phenyl-6(p-methylphenyl) glucosides or galactosides were devoid any theraputic index at the indicated concentration.

EXPERIMENTAL:

1-(2,3,4,6-Tetra-O-acetyl-B-D-gluco-and D-galactopyranosyl)3-cyanopyridine-2-ones; 5

General Coupling Procedure:

To a solution of 2(1H)-pyridineones **3a-h** (0.01 mol) in aqueous potassium hydroxide (0.56 g, 0.01 mol, in 6 mL of distilled water) was added a solution of 2,3,4,6-tetra-O-acetyl- α -D-gluco- or galactopyranosyl bromide 4 (4.521 g, 0.011 mol) in acetone (30 mL). The reaction mixture was stirred at room temperature until judged complete by TLC (8-12h), then evaporated under reduced pressure at 40°C and the residue washed with distilled water to remove the formed potassium bromide. The product was dried and crystallized from ethanol to afford colourless crystals (cf. Table 2).

1-(B-D-Gluco- and D-galactopyranosyl)3-cyanopyridine-2-ones; 6

General Procedure for Nucleoside Deacylation:

Dry gaseous ammonia was passed through a solution of protected nucleosides 5 (0.5 gm) in dry methanol (20 mL) at 0°C for about 0.5 hour. The reaction mixture was then stirred at 0°C until judged complete by

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TLC (20-24 h). The mixture was evaporated under reduced pressure at 40°C giving a solid residue which was crystallized from methanol to afford colourless crystals (cf. Table 2).

Biological Procedure

The compounds 5a-p and 6a-p were dissolved in dimethyl sulfoxide then diluted 1:100 in cell culture medium before preparing serial half- Log_{10} dilutions. T₄ Lymphocytes were added and after a brief interval HIV-1 is added, resulting in 1:200 final dilution of the compound. Uninfected cells with the compound serve as a toxicity control, and infected and uninfected cells without the compound serve as basic controls. Cultures are incubated at 37° in a 5% carbon dioxide atmosphere for 6 days. The tetrazolium salt, XTT, is added to all wells, and cultures are incubated to allow formazan color development by viable cells. Individual wells are analyzed spectrophotometrically to quantitate formazan production, and in addition are viewed microscopically for detection of viable cells and confirmation of protective activity.

Drug-treated virus-infected cells are compared with drug-treated noninfected cells and with other appropriate controls on the same plate. Data are reviewed in comparison with other tests done at the same time and a determination about activity is made.

Acknowledgements

The authors are greatly indebted to Dr. John P. Bader, Dr. V. L. Narayanan and Dr. Michael R. Boyd, AIDS Department and Drug Synthesis & Chemistry Branch of Health and Human Services, National Institutes of Health for their valuable in vitro anti-HIV activity and antitumor activities of our samples.

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Received August 3, 1994 Accepted February 17, 1995