J.C.S. CHEM. COMM., 1979

Synthesis of (-)-Shyobunone and its Epimers¹

By JOHN R. WILLIAMS* and JAMES F. CALLAHAN

(Department of Chemistry, Temple University, Philadelphia, Pennsylvania 19122)

Summary (-)-Shyobunone (10) and its epimers were synthesized from methylcyclobutene (1) and L-piperitone (2) in ca. 24% overall yield via irradiation followed by thermolysis of the reduced intermediate photoadduct (4).

(+)-SHYOBUNONE, whose structure is the mirror image of (10), was initially isolated from Acorus calamus L^2 and belongs to the elemene class of sesquiterpenes.³ It has been the topic of a number of recent syntheses.⁴ We here report a concise synthesis of (-)-shyobunone (10) and its C-2 and C-3 epimers, (11) and (12), respectively.

It occurred to us⁵ and to others⁶ that photochemical cycloaddition of cyclobutenes to cyclohexenones, followed by thermolysis of these photoadducts affords a convenient method for the synthesis of the germacrene and cis-1,2divinylcyclohexane ring systems. Therefore, by starting with an isoprene synthon, methylcyclobutene (1), and a commercially available chiral monoterpenoid, L-piperitone (2), $[\alpha]_D = 6.9^\circ$ [enantiomeric excess (e.e.) 14%],⁷ shyobunone and its epimers could be synthesized as shown in the Scheme.

Methylcyclobutene (1), which was obtained by the catalytic isomerization of methylenecyclobutane,8 and L-piperitone (2) were irradiated at -78 °C[†] until all (2) had disappeared.[‡] The photoadduct (3)§ was formed in 71% yield; i.r. (neat) 1688 cm⁻¹; n.m.r. (CDCl₃) δ 2.76 (s, 1H), 1.16 (s, 3H), 1.10 (s, 3H), 0.95 (d, 3H, J 7.6 Hz), and 0.80 (d, 3H, J 7.6 Hz). The photoadduct exhibited a positive Cotton effect, $[\theta]_{311} + 5660$, which supports the structure (3). Thermolysis of (3) yielded a substituted cadinene.9 However, if the ketone group in (3) was first reduced, this intramolecular reaction was avoided.

Reduction of (3) (NaBH₄, MeOH) afforded the alcohol (4) in 71% isolated yield; m.p. 72-72.5 °C; i.r. (KBr) 3313

[‡] Monitored by ¹H n.m.r. spectroscopy on an R-32 Perkin-Elmer 90 MHz n.m.r. spectrometer.

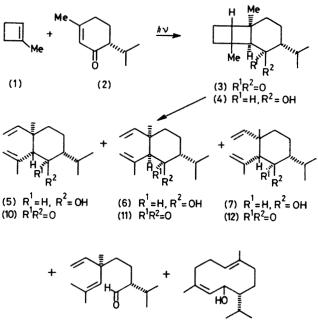
§ All new compounds have given satisfactory analytical and spectral data.

† Quartz apparatus, 450 W Hanovia Lamp, -78 °C (dry ice-propan-2-ol).

(6) $R^1 = H, R^2 = OH$ (7) $R^{1} = H, R^{2} = OH$ (11) $R^{1}R^{2}=0$ (12) $R^1 R^2 = 0$ (8) (9) SCHEME cm⁻¹; n.m.r. (CDCl₃) δ 3·71 (dd, 1H, J 8, 12 Hz), 2·77 (d, 1H, J 8 Hz), 1.38 (s, 3H), 1.01 (s, 3H), 0.97 (d, 3H, J 8 Hz), and 0.78 (d, 3H, J 8 Hz). Thermolysis of (4) in a sealed tube

(250 °C; xylene) yielded three major structural isomers: a

mixture of the elemene alcohols (5), (6), and (7) (relative



ratios 21:26:54, 48% yield); an aldehyde (8, 43%); and a germacrenol (with the probable structure (9, 6%)). The relative stereochemistry of the elemene alcohols (5)—(7)was elucidated by their spectral data and by their conversion (Jones oxidation) into the known ketones (10)-(12).^{2,4,10} A comparison of the Cotton effects exhibited in the c.d. spectra of the ketones with that exhibited in the o.r.d. of the naturally occurring compounds¹⁰ established the absolute stereochemistry of the products as (-)-shyobunone (10), $[\alpha]_{D} - 16^{\circ}$ (e.e. 14%),⁴ $[\theta]_{296\cdot5} - 1510$; 2-epi-(-)-shyobunone (11), $[\theta]_{301\cdot5} + 3040$; and 3-epi-(-)shyobunone (12), $[\theta]_{297\cdot 5} - 1830$. The aldehyde (8) was successfully converted into the

alcohol (5) with SnCl₄,¹¹ thereby establishing the general structure of (8). This was confirmed by the following data: i.r. (neat) 1729 and 1630 cm⁻¹; n.m.r. (CDCl₃) δ 9.56 (d, 1H, J 2.4 Hz), 5.66-6.05 (m, 1H), 5.10 (br s, 1H), 4·8-5·1 (m, 2H), 1·68 (d, 3H, J 1 Hz), 1·60 (d, 3H, J 1 Hz), 1.13 (s, 3H) and 0.97 (d, 6H, J 7.6 Hz). The alcohol (5) was converted into (-)-shyobunone (10), $[\theta]_{296\cdot 5}$ -1429, to establish the absolute stereochemistry of the aldehyde as that of (8).

The intermediate photoadduct (4) probably is cleaved by a radical process¹² and may form the elemenes directly or, more probably, yields an intermediate germacrenol skeleton^{5,6} which undergoes a Cope rearrangement to yield the observed elemene alcohols. The elemene alcohols (5)—(7)rearrange on further heating to the aldehyde (8).

Since treatment of shyobunone with base causes the isopropenyl double bond to move into conjugation with the ketone, this also constitutes a synthesis of isoshyobunone (13).2,4,10



(13)

We thank Professor S. Yamamura for spectral data on shyobunone and its epimers, Temple University for the award of a University Fellowship to J.F.C., and the National Science Foundation for partial support.

(Received, 2nd January 1979; Com. 001.)

- ¹ For the previous paper in this series see J. R. Williams, H. Salama, and J. D. Leber, J. Org. Chem., 1977, 42, 102.
- ² S. Yamamura, M. Iguchi, A. Nishiyama, M. Niwa, H. Koyama, and Y. Hirata, *Tetrahedron*, 1971, 27, 5419. ³ T. K. Devon and A. I. Scott, 'Handbook of Naturally Occurring Compounds,' Vol. 2, Academic Press, New York, 1972.
- ⁴ K. Kato, Y. Hirata, and S. Yamamura, Tetrahedron, 1971, 27, 5987; C. Alexandre and F. Rouessac, Bull. Soc. Chim. France, 1977,
- 117; G. Fráter, Helv. Chim. Acta, 1978, 61, 2709.
- ⁵ J. R. Williams, R. Moore, P. Wepplo, and R. A. Paclin, 10th Middle Atlantic Regional Amer. Chem. Soc. Meeting, Philadelphia, Pennsylvania, 1976.
- ⁶G. L. Lange, M. D. Huggins, and E. Neidert, Tetrahedron Letters, 1976, 4409; P. A. Wender and J. C. Lechleiter, J. Amer. Chem. Soc., 1977, 99, 267; ibid., 1978, 100, 4321.
 - ⁹ 'Handbook of Chemistry and Physics,' Chemical Rubber Co., Cleveland, Ohio, 1976, p. C443.
- ¹ J. Shabtai and E. Gil Av, J. Org. Chem., 1963, 28, 2893.
 ⁹ J. R. Williams and J. F. Callahan, J.C.S. Chem. Comm., following communication.
 ¹⁰ M. Niwa, Y. Terada, M. Iguchi, and S. Yamamura, Chem. Letters, 1977, 1415; M. Niwa, M. Iguchi, and S. Yamamura, *ibid.*, p. 1419.
 ¹¹ J. A. Marshall, N. H. Andersen, and P. C. Johnson, J. Org. Chem., 1970, 35, 186.
 ¹² L. A. Paquette and J. A. Schwartz, J. Amer. Chem. Soc., 1970, 92, 3215; M. J. Goldstein and M. S. Benzon, *ibid.*, 1972, 94, 5119;
 E. N. Cain and R. K. Solly, *ibid.*, 1973, 95, 7884; A. Sinnema, F. van Rantwijk, A. R. DeKoning, A. M. van Wijk, and H. van Bekkum, L. C. Schem. Comm. 1972, 264.
- J.C.S. Chem. Comm., 1973, 364.