

A One-Pot Synthesis of β -Lactams

Muneharu MIYAKE, Norio TOKUTAKE*, Makoto KIRISAWA

Department of Pharmacy, College of Science and Technology, Nihon University, 1-8, Kandasurugadai, Chiyoda-ku, Tokyo 101, Japan

The synthesis of β -lactams has been achieved by a variety of methods, among which the acid chloride/imine method has been frequently employed¹. Recently, the synthesis of β -lactams by the reaction of β -amino acids with methanesulfonyl chloride in chloroform/aqueous alkali solution in the presence of a phase transfer catalyst has been reported².

In connection with an earlier report³ dealing with the reaction of acid anhydrides with imines, the present work was undertaken to synthesize some β -lactams by the one-pot reaction of mixed acid anhydrides with imines in the presence of triethylamine. The mixed anhydrides **3**, synthetic equivalents to acid chlorides, were prepared from phthalimido- or *p*-chlorophenoxyacetic acids **1** and *p*-toluenesulfonyl chloride (**2**) in the presence of triethylamine. Under the mild reaction conditions employed, the *in situ* formed **3** react with imines **4** to give corresponding monocyclic β -lactams **5** in moderate yields.

The structure of the β -lactams **5** were confirmed by their I.R. and ¹H-N.M.R. spectra, and microanalyses (Table). The stereochemistry at C-3 and C-4 of the β -lactam ring was deduced from the coupling constants of the protons attached to these carbon atoms in their ¹H-N.M.R. spectra. Compound **5d** has the *cis*-configuration ($J=5$ Hz) whereas **5a**, **5b**, and **5c** are *trans*-isomers ($J=2$ Hz). No *cis/trans* isomeric mixture was observed in the ¹H-N.M.R. spectra of the products.

Since the advent of antibiotics exhibiting broad spectrum antibacterial activities, such as cephams and oxacephams, intensive synthetic studies have appeared in the literature. Tricyclic β -lactams have been synthesized¹ from azidoacetyl chloride and cyclic imines and cephams from substituted acetyl chlorides and dihydrothiazine derivatives⁴. The present procedure to prepare β -lactam compounds was then extended to the synthesis of some tricyclic β -lactams **7**, cephams **9**, and oxacephams **11**.

When a mixture of a 3,4-dihydroisoquinoline (**6**), which was utilized as the imine component, a mixed anhydride **3**, and triethylamine in anhydrous dichloromethane was stirred at room temperature for 48 h, the corresponding β -lactam compound **7** was obtained. Similar treatment of 2-phenyl-5,6-dihydro-4*H*-1,3-thiazine (**8**) and -oxazine (**10**) with **3** afforded β -lactams **9** and **11**, respectively. Spectral data and microanalyses support the structure of the products **7**, **9**, and **11** (Table).

The present method provides a simple and convenient preparation of the β -lactams under mild conditions. Further studies on the application of the method to organic synthesis are in progress.

The 3,4-dihydroisoquinolines **6a**⁵, **6b**⁶, **6c**⁶, 2-phenyl-5,6-dihydro-4*H*-1,3-thiazine (**8**)⁷ and 2-phenyl-5,6-dihydro-4*H*-1,3-oxazine (**10**)⁸ were prepared according to literature procedures.

β -Lactams **5**, Tricyclic β -Lactams **7**, Cephams **9**, and Oxacephams **11**; General Procedure:

A solution of the substituted acetic acid **1** (1 mmol), *p*-toluenesulfonyl chloride **2** (1 mmol), and triethylamine (2 mmol) in anhydrous dichloromethane (5 ml) is stirred at room temperature for 10 min. To this solution is added the imine **4** or 3,4-dihydroisoquinoline **6** or 1,3-thiazine **8** or 1,3-oxazine **10** (1 mmol) in anhydrous dichloromethane (2 ml). The reaction mixture is stirred at room temperature for 48 h, washed with 5% sodium hydrogen carbonate solution (3 ml), water (3 ml), and dried with anhydrous sodium sulfate. Removal of the solvent and subsequent trituration with ethanol under ice-cooling gives crude crystals, which on recrystallization from dichloromethane/ethanol afford the pure product (Table).

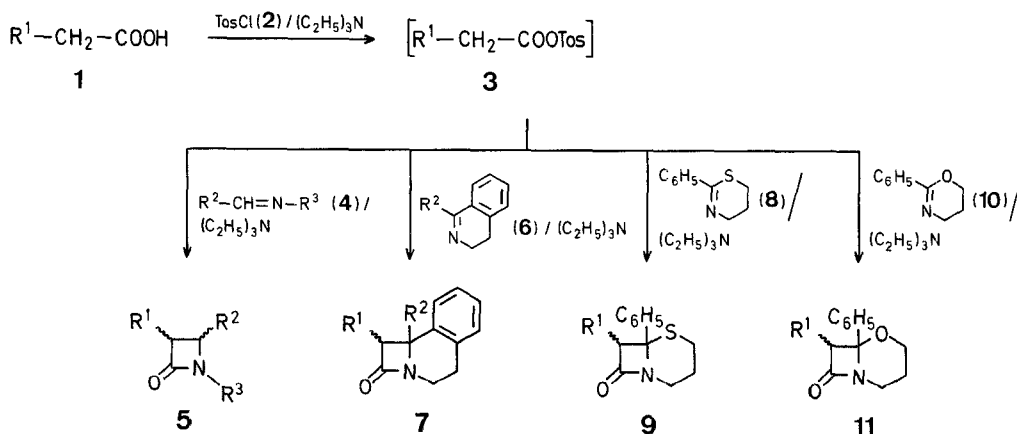


Table. β -Lactams **5**, Tricyclic β -Lactams **7**, Cepham **9**, and Oxacephams **11** prepared

Product No. R ¹	R ²	R ³	Yield [%]	m.p. ^a [°C]	Molecular formula ^b	I.R. (KBr) $\nu_{C=O}$ [cm ⁻¹]	¹ H-N.M.R. (CDCl ₃) δ [ppm]
5a			48	217-218°	C ₂₃ H ₁₄ Cl ₂ N ₂ O ₃ (437.3)	1710, 1760, 1780	5.23 (d, 1H, <i>J</i> =2 Hz); 5.33 (d, 1H, <i>J</i> =2 Hz); 7.22-7.81 (m, 12H)
5b			52	262-262.5°	C ₂₆ H ₂₀ N ₂ O ₅ (440.4)	1700, 1720, 1765	1.38 (t, 3H, <i>J</i> =7 Hz); 4.35 (q, 2H, <i>J</i> =7 Hz); 5.32 (d, 1H, <i>J</i> =2 Hz); 5.48 (d, 1H, <i>J</i> =2 Hz); 6.95-8.45 (m, 8H)
5c			41	124-125°	C ₂₂ H ₁₈ ClNO ₂ (363.8)	1750	2.42 (s, 3H); 5.11 (d, 1H, <i>J</i> =2 Hz); 5.2 (d, 1H, <i>J</i> =2 Hz); 6.9-7.5 (m, 13H)
5d			41	157-158°	C ₂₃ H ₂₀ ClNO ₃ (393.9)	1740	2.25 (s, 3H); 3.74 (s, 3H); 5.29 (d, 1H, <i>J</i> =5 Hz); 5.37 (d, 1H, <i>J</i> =5 Hz); 6.62-7.41 (m, 12H)
7a		—	49	124-125°	C ₂₃ H ₁₈ ClNO ₂ (375.8)	1750	2.5-3.0 (m, 2H); 3.55- 3.98 (m, 2H); 5.42 (s, 1H); 6.65-7.51 (m, 13H)
7b		—	53	261-262°	C ₂₃ H ₁₇ N ₃ O ₅ (439.4)	1710, 1760, 1780 ^c	2.7-3.0 (m, 2H); 3.75- 4.15 (m, 2H); 5.68 (s, 1H); 7.08-8.23 (m, 12H)
7c		—	55	135-136°	C ₂₃ H ₁₇ ClN ₂ O ₄ (420.8)	1760 ^c	2.5-3.15 (m, 2H); 3.6- 4.0 (m, 2H); 5.52 (s, 1H); 6.75-8.36 (m, 12H)
9a	—	—	39	157-158°	C ₁₃ H ₁₆ ClNOS (329.8)	1760	1.6-2.15 (m, 2H); 2.5- 3.39 (m, 3H); 4.0-4.5 (m, 1H); 4.31 (s, 1H); 6.79- 7.49 (m, 9H)
9b	—	—	46	121-121.5°	C ₁₈ H ₁₆ NO ₂ S (345.8)	1780	1.6-2.15 (m, 2H); 2.55- 3.3 (m, 3H); 3.85-4.4 (m, 1H); 5.27 (s, 1H); 6.5- 7.71 (m, 9H)
9c	—	—	64	208-209°	C ₂₀ H ₁₅ N ₃ O ₅ S (409.4)	1712, 1762, 1785 ^d	1.65-2.2 (m, 2H); 2.5- 2.9 (m, 2H); 3.05-3.6 (m, 1H); 4.1-4.6 (m, 1H); 5.5 (s, 1H); 7.0-8.2 (m, 8H)
11a	—	—	44	239-240°	C ₂₀ H ₁₆ N ₂ O ₄ (348.3)	1710, 1760, 1790	1.25-2.25 (m, 2H); 3.0- 4.32 (m, 4H); 5.35 (s, 1H); 7.14-7.72 (m, 9H)
11b	—	—	43	138-139°	C ₁₈ H ₁₆ ClNO ₃ (329.8)	1760	1.2-2.69 (m, 3H); 2.7- 3.5 (m, 1H); 3.52-4.38 (m, 2H); 5.25 (s, 1H); 6.5-7.68 (m, 9H)
11c	—	—	51	217-218°	C ₂₀ H ₁₅ N ₃ O ₆ (393.3)	1712, 1780, 1795 ^d	1.05-2.2 (m, 2H); 2.4- 2.7 (m, 1H); 2.9-4.25 (m, 3H); 5.25 (s, 1H); 6.95- 8.45 (m, 8H) ^c

^a Not corrected.^b Satisfactory microanalyses obtained: C \pm 0.30, H \pm 0.18, N \pm 0.30.^c ν_{NO_2} = 1520 cm⁻¹.^d ν_{NO_2} = 1540 cm⁻¹.^e Measured in DMSO-*d*₆.

Received: March 23, 1983

² Y. Watanabe, T. Mukaiyama, *Chem. Lett.* **1981**, 443.³ M. Miyake, M. Kirisawa, N. Tokutake, *Synthesis* **1982**, 1053.¹ A. K. Bose, B. Anjaneyulu, S. K. Bhattacharya, M. S. Manhas, *Tetrahedron* **23**, 4769 (1967).⁴ A. K. Bose, V. Sudarsanam, B. Anjaneyulu, M. S. Manhas, *Tetrahedron* **25**, 1191 (1969).

- ⁵ M. Lora-Tamayo, R. Madronero, Guillermo, *Chem. Ber.* **93**, 289 (1960).
- ⁶ V. M. Rodionov, E. V. Yavorskaya, *J. Gen. Chem. U.S.S.R.* **11**, 446 (1941).
V. M. Rodionov, E. V. Yavorskaya, *J. Gen. Chem. U.S.S.R.* **13**, 491 (1943).
- ⁷ G. Pinkus, *Chem. Ber.* **26**, 1077 (1893).
- ⁸ J. H. Boyer, J. Hamer, *J. Am. Chem. Soc.* **77**, 951 (1955).