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A Convenient Synthesis of Unsymmetrical N,N'-Disubstitutedα,ω-Diaminoalkanes

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A CONVENIENT SYNTHESIS OF UNSYMMETRICAL N,N'-DISUBSTITUTED α,ω-DIAMINOALKANES

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ABSTRACT: A general procedure is described for regiospecific construction of unsymmetrical N-alkyl (or aralkyl)-N'-aryl- α, ω -diaminoalkanes 3 (n=2,3,4) by reduction of N-(ω -arylaminoalkyl)amides 2 with borane. Compounds 2 are readily obtained by condensation of N-(ω -haloalkyl)amides 1 with aromatic amines.

In the course of our research on nitrogen heterocycles, we recently required to prepare several unsymmetrical N,N'-disubstituted α,ω -diaminoalkanes 3 (n= 2,3,4)

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Scheme
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	RNH-(CH ₂)n-NHCOF	צ'	
2 RNH ₂ 120 °C, 1h		2 78 - 87 % BH ₃ /THF 2 hs, reflux		
X-(CH ₂)n-NHCOR'		RNH-(CH ₂)n-NHCH ₂ R'		
	1		3	
Compound (1-3)	n	R	R'	Yield (%) (overall)
8	2	C ₆ H ₅	C ₆ H ₅	75
b	2	p-ClC ₆ H₄	C ₆ H ₅	67
c	2	β- C ₁₀ H ₇	C ₆ H ₅	64
d	2	p-NO ₂ C ₆ H ₄	C ₆ H ₅	55
e	2	p-CH₃OC ₆ H₄	C ₆ H ₅	62
f	3	p-ClC ₆ H₄	C ₆ H ₅	64
g	3	β-C ₁₀ H ₇	C ₆ H ₅	62
h	3	p-NO ₂ C ₆ H ₄	C ₆ H ₅	74
i	3	C ₆ H ₅	p-NO ₂ C ₆ H ₄	63
j	3	p-ClC₀H₄	CH₃	72
k	3	p-ClC₀H₄	C_2H_5	62
ł	3	p-ClC ₆ H₄	i-C ₃ H ₇	67
m	4	p-CH ₃ OC ₆ H ₄	C ₆ H ₅	77
n	4	p-NO ₂ C ₆ H ₄	C ₆ H ₅	70

as precursors of cyclic aminals (imidazolidines, hexahydropyrimidines, and hexahydro-1,3-diazepines). Such compounds are easily prepared by condensation of the suitable α,ω -diaminoalkanes with aldehydes, so the problem devolves to the preparation of unsymmetrical alkylenediamine derivatives. A careful examination of the literature showed only a few synthetic procedures, and none is general. Besides, compounds 3 are also of interest given their close relationship with biogenic polyamines.¹

Among accepted procedures to generate the secondary amine functional group, selective alkylation of polyamines is hardly feasible due to polyalkylation reactions which often complicate the isolation of the desired monoalkyl product from the reaction mixture. An alternative method consists in the reaction between an halogen derivative carrying the methylene chain and the suitable amine,² though this procedure may present the same limitations as the previous one. To obviate this problem, some methods have been developed to ensure selective polyamine monoalkylation³ or to generate the required secondary amines from other nitrogenous functions.^{2ab,4}

The present paper describes an improved approach to the synthesis of N-alkyl (or aralkyl)-N'-aryl- α,ω -diaminoalkanes **3a-n** (Scheme) in good overall yields, from easily available starting materials in two steps.

Treatment of N-(ω -haloalkyl)amides 1 with two moles of a primary aromatic amine affords N-(ω -arylaminoalkyl)amides 2a-n which are then readily converted to the corresponding compounds 3 by refluxing with borane in tetrahydrofurane. Instead, reduction of amides 2 with lithium aluminium hydride renders poorer yields. In agreement with Brown,⁵ benzaldehyde and benzyl alcohol are by-products. In the case of nitroaryl derivatives 3d,h,n, the preparation of compounds 2 by the above route requires very drastic conditions and affords low yields. In such cases, direct acylation of the nitroarylalkylenediamine is the most suitable procedure to obtain the intermediate amide $2.^{6}$

The extension of this method to the preparation of unsymmetrical N,N'-dialkyl α,ω -diaminoalkanes is limited by the low yield of the reaction of compounds 1 (n=3,4) with aliphatic amines, due to competitive intramolecular dehydro-halogenation reactions.^{6b,7}

Experimental

Melting points were determined with a Büchi capillary apparatus and are ¹H NMR spectra were recorded on a Bruker MSL 300 MHz uncorrected. spectrometer, using deuteriochloroform as the solvent. Chemical shifts are reported in ppm (δ) relative to TMS as an internal standard. D₂O was employed to confirm exchangeable protons (ex.). Splitting multiplicities are reported as singlet (s), broad signal (bs), doublet (d), triplet (t), quartet (q) and multiplet (m). Mass spectra (EI) were recorded with a GC-MS Shimadzu QP-1000 spectrometer operating at 20 eV. TLC analyses were carried out on aluminium sheets silica gel 60 F 254 using chloroform-methanol (9:1) as the solvent. Column chromatographies were performed either on silica gel 60 or on aluminium oxide (90 active, neutral, activity I), with typically 30-50 g of stationary phase per gram substance. N-(ω -Haloalkyl)amides 1 were obtained by acylation of the corresponding ω bromoalkylamine hydrobromide (n=2,3) in Schotten-Baumann conditions,^{6a,b} and by a modified von Braun reaction on N-benzoylpyrrolidine in the case of $n=4.^{8}$

$N-(\omega - Arylaminoalkyl)$ amides 2:

Compounds 2a (mp 127°C) ^{6a}, 2b (mp 130°C) ⁹, 2c (mp 184°C) ^{6a}, 2d (mp 162°C) ^{6a}, 2e (mp 136°C) ^{6a}, 2f (mp 117°C) ^{6b}, 2g (mp 139°C) ^{6b}, 2h (mp 121°C) ^{6b}, 2i (mp 89°C) ¹⁰ and 2n (mp 145°C) ^{6c} were prepared following literature procedures.

Compounds 2j-m were obtained by reaction of the corresponding N-(ωhaloalquil)amides and arylamines.

General Procedure:

A mixture of the corresponding N-(ω -haloalkyl)amide 1 (10 mmol) and arylamine (20 mmol) was heated in an oil bath at 100-120 °C under reflux for 1 h. The crude product was treated three times with boiling water (20 ml) in order to eliminate arylamine hydrohalide.

N[3-(p-Chlorophenylamino)propyl]acetamide (2j).

The resulting oil was purified by column chromatography using silica gel 60 (chloroform: ethyl acetate, 8:2 to 1:1) to give 2j (85%); mp 91-93°C.

MS: $m/z= 227 (M^{+})$.

¹H RMN: δ =1.75 (2H, m, CH₂CH₂CH₂), 1.90 (3H, s, NHCOCH₃), 3.10 (2H, t, CH₂-NHR), 3.25-3.35 (2H, m, CH₂NHCO), 3.80 (1H, bs, ex., NHR), 6.25 (1H, bs, ex., NHCO), 6.50 (2H, dd, p-ClC₆H₄, 2 ortho H), 7.10 (2H, dd, p-ClC₆H₄, 2 meta H).

Anal. calcd. for C₁₁H₁₅N₂OCl: C 58.28, H 6.67, N 12.36, Cl 15.64; found: C 58.36, H 6.80, N 12.48, Cl 15.55.

N[3-(p-Chlorophenylamino)propyl]propanamide (2k).

The resulting oil was purified by column chromatography using silica gel 60

(chloroform: ethyl acetate, 8:2 to 1:1) to give 2k (74%); mp 61-63°C.

MS: $m/z= 241 (M^+)$.

¹H RMN: δ=1.10 (3H, t, CH₂CH₃), 1.70 (2H, m, CH₂CH₂-CH₂), 2.15 (2H, q, CH₂CH₃), 3.05 (2H, t, CH₂NHR), 3.20-3.35 (2H, m, CH₂NHCO), 4.10 (1H, bs, ex., NHR), 5.60 (1H, bs, ex., NHCO), 6.45 (2H, dd, p-ClC₆H₄, 2 ortho H), 7.05 (2H, dd, p-ClC₆H₄, 2 meta H).

Anal. calcd. for $C_{12}H_{17}N_2OCl$: C 59.87, H 7.12, N 11.64, Cl 14.73; found: C 60.04, H 7.34, N 11.50, Cl 14.85.

N[3-(p-Chlorophenylamino)propyl]-2-methylpropanamide (21).

The resulting oil was purified by column chromatography using silica gel 60 (chloroform: ethyl acetate, 8:2 to 1:1) to give **2k** (82%); mp 49-51°C.

MS: $m/z= 255 (M^+)$.

¹H RMN: δ=1.10 (6H, d, CH(CH₃)₂), 1.75 (2H, m, CH₂CH₂CH₂), 2.35 (1H, m, CH(CH₃)₂), 3.15 (2H, t, CH₂NHR), 3.30-3.40 (2H, m, CH₂NHCO), 4.20 (1H, bs, ex., NHR), 5.60 (1H, bs, ex., NHCO), 6.55 (2H, dd, p-ClC₆H₄, 2 ortho H), 7.10 (2H, dd, p-ClC₆H₄, 2 meta H).

Anal. calcd. for $C_{13}H_{19}N_2OCl$: C 61.29, H 7.52, N 11.00, Cl 13.92; found: C 61.15, H 7.74, N 10.85, Cl 13.78.

N[4-(p-Methoxyphenylamino)butyl]benzamide (2m).

The crude product was recrystallized from cyclohexane to give 2m (85%); mp 104°C.

MS: $m/z= 298 (M^{+})$.

¹H RMN: δ =1.75 (4H, m, CH₂CH₂CH₂CH₂), 2.61 (1H, bs, ex., NHR), 3.10 (2H, t,

*CH*₂NHR), 3.40-3.60 (2H, m, *CH*₂NHCO), 3.76 (3H, s, OCH₃), 6.10 (1H, bs, ex., NHCO), 6.55 (2H, d, p-CH₃OC₆H₄, 2 ortho H), 6.70 (2H, d, p-CH₃OC₆H₄, 2 meta H), 7.26-7.90 (5H, m, C₆H₅).

Anal. calcd. for C₁₈H₂₂N₂O₂: C 72.46, H 7.43, N 9.39, found: C 72.35, H 7.62, N 9.23.

N-Alkyl (or aralkyl)-N'-aryl- α, ω -diaminoalkanes (3a-n). General Procedure: Compounds 2 (5 mmol) were treated with BH₃/THF (10 ml satd. soln.)¹¹ and heated under reflux in a nitrogen atmosphere for 2 h. THF was evaporated *in vacuo* and the residue boiled with concentrated hydrochloric acid (10 ml) for 15 min after which solution was cooled. If the product crystallized, the resulting solid was filtered, washed with ethyl ether, dried and recrystallized (anhydrous ether/ethanol). If not, the acid solution was diluted with water (5 ml) and then made alkaline (pH=12) with sodium hydroxide pellets. The mixture was extracted with chloroform (3 x 20 ml) and the organic layer was washed with water (5 ml), filtered and dried with anhydrous sodium sulphate. The solution was concentrated *in vacuo* and the crude base purified by column chromatography.

N-Benzyl-N'-phenyl-1,2-diaminoethane (3a).

Product was isolated as dihydrochloride (87%); mp 188-9°C (lit.¹² mp 188-9°C).

Free base:

MS: $m/z= 226 (M^+)$.

¹H RMN: δ= 2.40 (1H, bs, ex. N*H*CH₂R'), 2.90 (2H, t, C*H*₂NHCH₂R'), 3.40 (2H, t, C*H*₂NHR), 3.50 (bs, ex., RN*H*CH₂), 3.90 (2H, s, NHC*H*₂R'), 7.00-7.40 (10H, m, aromatics).

N-Benzyl-N'-(p-chlorophenyl)-1,2-diaminoethane (3b).

Product was isolated as dihydrochloride (84%); m.p 224-226°C (lit.^{4b} monohydrochloride mp 225-7°C).

Anal. calcd. for $C_{15}H_{19}N_2Cl_3$: C 53.99, H 5.74, N 8.40, Cl 31.87; found: C 54.22, H 5.85, N ,8.44, Cl 31.60.

Free base:

MS: $m/z= 261 (M^+)$.

¹H RMN: δ=2.20 (1H, bs, ex., NHCH₂R'), 2.90 (2H, t, CH₂NHCH₂R'), 3.20 (2H,

t, CH2NHR), 3.70 (1H, bs, ex., RNH CH2), 3.80 (2H, s, NHCH2 R'), 6.40 (2H,

dd, p-ClC₆H₄, 2 ortho H), 7.10 (2H, dd, p-ClC₆H₄, 2 meta H), 7.40 (5H, s, C₆H₅).

N-Benzyl-N'-(\beta-napthyl)-1,2-diaminoethane (3c).

Product was isolated as dihydrochloride (81%); mp 236-238°C.

Anal. calcd. for $C_{19}H_{22}N_2Cl_2$: C 65.33, H 6.35, N 8.02, Cl 20.30; found: C 65.51, H 6.55, N 8.07, Cl 20.10.

Free base:

MS: $m/z= 276 (M^{+})$.

¹H RMN: δ =2.50 (1H, bs, ex., NHCH₂R'), 3.10 (2H, t, CH₂ NHCH₂R'), 3.45 (2H,

t, CH₂NHR), 3.80 (1H, bs, ex., RNHCH₂), 3.90 (2H, s, NHCH₂R'), 6.60-7.50 (7H, m, β -naphtyl), 7.30 (5H, s, C₆H₅).

N-Benzyl-N'-(p-nitrophenyl)-1,2-diaminoethane (3d).

Product was isolated as dihydrochloride (82%); mp 189-191°C.

Anal. calcd. for $C_{15}H_{19}N_3O_2Cl_2$: C 52.34, H 5.56, N 12.21, Cl 20.60; found: C 52.49, H 5.60, N 12.08, Cl 20.46.

Free base:

MS: $m/z= 271 (M^+)$.

¹H RMN: δ=2.60 (1H, bs, ex., NHCH₂R'), 2.95 (2H, t, CH₂NHCH₂R'), 3.30 (2H, t, CH₂NHR), 3.80 (2H, s, NHCH₂R'), 3.90 (1H, bs, ex., RNHCH₂), 6.50 (2H, dd, p-NO₂C₆H₄, 2 ortho H), 7.35 (5H, s, C₆H₅), 8.10 (2H, dd, p-NO₂C₆H₄, 2 meta H). *N-Benzyl-N'-(p-methoxyphenyl)-1,2-diaminoethane* (**3e**)

Product was isolated as dihydrochloride (78%); mp 158-160°C (lit.^{4b} mp monohydrochloride 234-6°C).

Anal. calcd. for C₁₆H₂₂N₂OCl₂: C 58.36, H 6.73, N 8.51, Cl 21.53; found: C 58.50, H 6.77, N 8.56, Cl 21.30.

Free base:

MS: $m/z= 256 (M^{+})$.

¹H RMN: δ=2.55 (1H, bs, ex., NHCH₂R'), 2.90 (2H, t, CH₂NHCH₂R'), 3.20 (2H, t, CH₂NHR), 3.60 (1H, bs, ex., RNHCH₂), 3.70 (3H, s, CH₃O), 3.80 (2H, s, NHCH₂ R'), 6.60 (2H, dd, p-CH₃OC₆H₄, 2 ortho H), 6.80 (2H, dd, p-CH₃OC₆H₄, 2 meta H), 7.40 (5H, s, C₆H₅).

N-Benzyl-N'-(p-chlorophenyl)-1,3-diaminopropane (3f).

The crude material was purified by column chromatography using aluminium oxide (ethyl acetate:methanol, 9:1 to 0:10) to give **3f** (84%).

MS: $m/z= 275 (M^{+})$.

¹H RMN: δ=1.75 (2H, m, CH₂CH₂CH₂), 2.30 (1H, bs, ex., NHCH₂R'), 2.55 (2H, t, CH₂NHCH₂R'), 3.05 (2H, t, RNHCH₂), 3.60 (2H, s, CH₂R'), 3.75 (1H, bs, ex.,

NHR), 6.40 (2H, dd, p-ClC₆H₄, 2 ortho H), 7.05 (2H, dd, p-ClC₆H₄, 2 meta H), 7.30 (5H, s, C₆H₅).

Anal. calcd. for C₁₆H₁₉N₂Cl: C 69.93, H 6.97, N 10.19, Cl 12.90; found: C 69.75, H 7.19, N 10.06, Cl 12.70.

N-Benzyl-N'-(β -naphtyl)-1,3-diaminopropane (3g).

The crude material was purified by column chromatography using aluminium oxide (ethyl acetate:methanol, 9:1 to 0:10) to give **3g** (80%).

MS: $m/z= 290 (M^{+})$.

¹H RMN: δ=1.85 (2H, m, CH₂CH₂ CH₂), 2.40 (1H, bs, ex., NHCH₂R'), 2.60 (2H, t, CH₂NHCH₂R'), 3.20 (2H, t, RNHCH₂), 3.60 (2H, s, CH₂R'), 3.70 (1H, bs, ex., NHR), 6.60-6.80 (2H, m, β-naphtyl), 7.10-7.65 (5H, m, β-naphtyl), 7.30 (5H, s, C₆H₅).

Anal. calcd. for $C_{20}H_{22}N_2$: C 82.72, H 7.64, N 9.65; found: C 82.61, H 7.92, N 9.47.

N-Benzyl-N'-(o-nitrophenyl)-1,3-diaminopropane (3h).

The crude material was purified by column chromatography using aluminium oxide (ethyl acetate:methanol, 9:1 to 0:10) to give **3h** (82%).

MS: m/z= 285 (M^{+.}).

¹H RMN: δ =1.90 (2H, m, CH₂CH₂CH₂), 2.50 (1H, bs, ex., NHCH₂ R'), 2.80 (2H, t, CH₂ NHCH₂R'), 3.30 (2H, t, RNHCH₂), 3.70 (2H, s, CH₂R'), 3.85 (1H, bs, ex., NHR), 6.70 (1H, dd, o-NO₂C₆H₄, 1 ortho H), 6.75-6.80 (1H, m, o-NO₂C₆H₄, 1 para H), 7.30 (5H, s, C₆H₅), 7.35-7.45 (1H, m, o-NO₂C₆H₄, 1 meta H), 8.10 (1H, dd, o-NO₂C₆H₄, 1 meta H).

Anal. calcd. for C₁₆H₁₉N₃O₂: C 67.35, H 6.71, N 14.73; found: C 67.17, H 6.92, N 14.60.

N-(p-Nitrobenzyl)-N'-phenyl-1,3-diaminopropane (3i).

The crude material was purified by column chromatography using aluminium oxide (ethyl acetate:methanol, 9:1 to 0:10) to give **3i** (87%); mp 56-58°C.

MS: $m/z= 285 (M^+)$.

¹H RMN: δ =1.80 (2H, m, CH₂CH₂ CH₂), 2.50 (1H, bs, ex., NHCH₂R'), 2.75 (2H, t, CH₂NHCH₂R'), 3.20 (2H, t, RNHCH₂), 3.90 (2H, s, CH₂R), 3.80 (1H, bs, ex., NHR), 6.50-6.70 (3H, m, C₆H₅, 2 ortho and para H), 7.10-7.20 (2H, m, C₆H₅, 2 meta H), 7.45 (2H, dd, p-NO₂C₆H₄, 2 ortho H), 8.15 (2H, dd, p-NO₂C₆H₄, 2 meta H).

Anal. calcd. for $C_{16}H_{19}N_3O_2$: C 67.35, H 6.71, N 14.73; found: C 67.15, H 6.90, N 14.85.

N-(p-Chlorophenyl)-N'-ethyl-1,3-diaminopropane (3j).

The crude material was purified by column chromatography using aluminium oxide (ethyl acetate:methanol, 9:1 to 0:10) to give **3j** (85%); mp 206-208°C (dihydrochloride).

Anal. calcd. for C₁₁H₁₉N₂Cl₃: C 46.25, H 6.70, N 9.81, Cl 37.23; found: C 46.43, H 6.92, N 9.62, Cl 37.04.

Free base:

MS: $m/z= 213 (M^+)$.

¹H RMN: δ=1.05 (3H, t, CH₃), 1.75 (2H, m, CH₂CH₂CH₂), 2.25 (1H, bs, ex., NHCH₂R'), 2.60 (2H, q, CH₂CH₃), 2.70 (2H, t, CH₂NHCH₂R'), 3.10 (2H, t,

RNHCH₂), 3.50 (1H, bs, ex., NHR), 6.45 (2H, dd, p-ClC₆H₄, 2 ortho H), 7.05 (2H, dd, p-ClC₆H₄, 2 meta H).

N-(p-Chlorophenyl)-N'-propyl-1,3-diaminopropane (3k).

The crude material was purified by column chromatography using aluminium oxide (ethyl acetate:methanol, 9:1 to 0:10) to give **3k** (84%); mp 51-53°C.

MS: $m/z= 227 (M^+)$.

¹H RMN: δ =0.90 (3H, t, CH₃),1.65 (2H, m, CH₂CH₃), 1.90 (2H, m, CH₂CH₂ CH₂), 2.50 (1H, bs, ex., NHCH₂R'), 2.65 (2H, t, NHCH₂R'), 2.85 (2H, t, CH₂NHCH₂R'), 3.20 (2H, t, RNHCH₂), 3.80 (1H, bs, ex., NHR), 6.55 (2H, dd, p-ClC₆H₄, 2 ortho H), 7.10 (2H, dd, p-ClC₆H₄, 2 meta H).

Anal. calcd. for C₁₂H₁₉N₂Cl: C 63.56, H 8.45, N 12.35, Cl 15.64; found: C 63.67, H 8.64, N 12.17, Cl 15.83.

N-(p-Chlorophenyl-N'-(2-methylpropyl)-1,3-diaminopropane (31).

The crude material was purified by column chromatography using aluminium oxide (ethyl acetate:methanol, 9:1 to 0:10) to give **31** (82%); mp 243-245°C (dihydrochloride).

Anal. calcd. for C₁₃H₂₃N₂Cl₃: C 49.77, H 7.39, N 8.93, Cl 33.90; found: C 49.98, H 7.58, N 8.95, Cl 33.88.

Free base:

MS: $m/z= 241 (M^+)$.

¹H RMN: δ=0.95 (6H, d, CH₃), 1.75 (1H, m, CH), 1.80 (2H, m, CH₂CH₂CH₂), 2.45 (3H, bs, NHCH₂ R'), 2.75 (2H, t, CH₂NHCH₂R'), 3.15 (2H, t, RNHCH₂), 3.40 (1H, bs, ex., NHR), 6.50 (2H, dd, p-ClC₆H₄, 2 ortho H), 7.10 (2H, dd, p-ClC₆H₄, 2 meta H).

N-Benzyl-N'-(p-methoxyphenyl)-1,4-diaminobutane (3m).

The crude material was purified by column chromatography using silica gel 60 (chloroform:methanol, 9:1) to give **3m** (90%).

MS: $m/z = 284 (M^+)$.

¹H RMN: δ =2.20-2.25 (4H, m, CH₂CH₂CH₂CH₂), 2.60 (1H, bs, ex., NHCH₂R'), 2.90 (2H, t, CH₂NHCH₂R'), 3.30 (2H, t, RNHCH₂), 3.60 (1H, bs, ex., NHR), 3.80 (2H, s, NHCH₂R'), 6.65 (2H, dd, p-CH₃OC₆H₄, 2 ortho H), 6.85 (2H, dd, p-CH₃OC₆H₄, 2 meta H), 7.55 (5H, s, C₆H₅).

Anal. calcd. for C₁₈H₂₄N₂O: C 76.02, H 8.51, N 9.85; found: C 76.10, H 8.75, N 9.94.

N-Benzyl-N'-(p-nitrophenyl)-1,4-diaminobutane (3n).

The crude material was purified by column chromatography using silica gel 60 (chloroform:methanol, 9:1) to give **3n** (78%).

MS: $m/z= 299 (M^+)$.

¹H RMN: δ =2.20-2.15 (4H, m, CH₂CH₂CH₂CH₂), 2.65 (1H, bs, ex., NHCH₂R'), 2.95 (2H, t, CH₂NHCH₂R'), 3.45 (2H, t, RNHCH₂), 3.80 (2H, s, NHCH₂R'), 3.90 (1H, bs, ex., NHR), 6.55 (2H, dd, p-NO₂C₆H₄, 2 ortho H), 7.45 (5H, s, C₆H₅), 8.15 (2H, dd, p-NO₂C₆H₄, 2 meta H).

Anal. calcd. for C₁₇H₂₁N₃O₂: C 68.20, H 7.07, N 14.04; found: C 68.43, H 7.26, N 14.00.

Reduction of compounds 2 with lithium aluminium hydride afforded moderate yields of 3 (45-60%). Two by-products (benzaldehyde and benzyl alcohol) were detected by GC-MS.

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