S. S. FI SEVIER

Contents lists available at ScienceDirect

Carbohydrate Research

journal homepage: www.elsevier.com/locate/carres



Note

Regioselective synthesis of 4azido-Neu $2en5,7Ac_21Me$ and its intramolecular transformation to 4azido-Neu $2en5,9Ac_21Me$

Qingjie Zhao a, Yuanyuan Lou b, Ruisheng Xiong a, Haihong Li c, Jingshan Shen a,c,*

- ^a Shanghai Institute of Materia Medica, Chinese Academy of Sciences, Shanghai 201203, PR China
- ^b College of Materials Science and Engineering, Nanjing University of Technology, Nanjing 210009, PR China
- ^c Topharman Shanghai Co., Ltd, Shanghai 201209, PR China

ARTICLE INFO

Article history:
Received 30 January 2008
Received in revised form 11 June 2008
Accepted 12 June 2008
Available online 20 June 2008

Keywords: Neu5Ac Analogues Acetyl migration Azides X-ray crystal structure Carbohydrates

ABSTRACT

Methyl 5-N-acetyl-7-O-acetyl-4-azido-2,3-didehydro-2,4-dideoxy-neuraminic acid (4azido-Neu2en5,7 Ac₂1Me) was synthesized regioselectively starting from 4azido-Neu2en5Ac1Me in high yield. The transformation of 4azido-Neu2en5,7Ac₂1Me to the corresponding thermodynamically stable 4azido-Neu2en5,9Ac₂1Me via intramolecular acetyl migration was confirmed by single-crystal X-ray diffraction analysis. The proposed rearrangement mechanism is discussed.

© 2008 Published by Elsevier Ltd.

5-Acetamido-3,5-dideoxy-D-glycero-D-galacto-non-2-ulopyranoso nic acid (Neu5Ac, 1) and its derivatives are significant carbohydrates that play important roles in the construction of biological molecules. ¹⁻³ In an effort to learn more about the involvement of Neu5Ac derivatives substituted in the glycerol chain in biological processes with retention of amino or guanido groups at C-4, ⁴⁻¹⁰ an efficient preparation of this class of compounds would provide easy access to various structural congeners required for further structure–activity relationships study. 4Azido-Neu2en5Ac1Me (2) was a versatile intermediate for such kind of research since it can be easily transformed into amino or guanido derivatives. Herein, we report our efforts on the first efficient and regioselective synthesis of the 7-O-acetylated form of 4azido-Neu2en5Ac1Me (Schemes 1 and 2).

4-Azido-Neu2en5,7,8,9Ac₄1Me, prepared from Neu5Ac in good yield, ¹¹ was deacetylated with NaOMe in MeOH to give compound $\mathbf{2}^{12}$ in 96% yield. Without separation, compound $\mathbf{2}$ was reacted with acetone using Dowex 50W-X8 (H⁺) resin as a catalyst to give protected compound $\mathbf{3}^{13}$ in 80% yield. Treatment of compound $\mathbf{3}$ with either propionic, butyric, methoxyacetic, benzoic or acetic anhydride in pyridine under the catalysis of 4-dimethylaminopyridine (DMAP)¹⁴ afforded the corresponding acylated compounds $\mathbf{4a-e}$ in 95–98% yield.

Hydrolysis of **4a–d** with water in the presence of Dowex 50W-X8 (H^+) resin yielded the corresponding 7-O-acylated compounds. Due to their unstable character, these compounds were difficult to purify via crystallization and column chromatography, as indicated by LC–MS. Fortunately, compound **4e** could be converted into pure compound **5** by hydrolysis and crystallization from EtOAc. Compound **5** was stable when stored in desiccators or in common solvents such as MeOH, $\mathrm{CH_2Cl_2}$, and EtOAc at room temperature, while it was unstable in acidic media. It is interesting to note that compound **5** in MeOH began to transform into the thermodynamically stable compound **6** at 40 °C (Scheme 2). Furthermore, no 8-O-acetylated product was found during the rearrangement. It was postulated that the intramolecular transfer of the acetyl group proceeded via the cyclic intermediates of orthoesters **7** and **8** (Scheme 3). ¹⁵

All structures of new compounds were confirmed by elemental analysis, ¹H NMR, and ¹³C NMR spectroscopy. The evident

^{*} Corresponding author. E-mail address: jsshen@mail.shcnc.ac.cn (J. Shen).

Scheme 1. Reagents and conditions: (a) Dowex 50W-X8 (H*) resin, Me₂CO, 80%; (b) anhydrides, Py, DMAP, 10-24 h; 95-98%.

Scheme 2. Reagents and conditions: (a) Dowex 50W-X8 (H⁺) resin, water, 76%; (b) MeOH, reflux, 80%.

Scheme 3. Plausible mechanism for the transformation of 5 to thermodynamically stable 6.

Table 1 Chemical shifts (ppm) in the 1 H NMR spectra (Me₂SO- d_6) of **2**, 12 **5**, and **6**

	4-H	5-H	6-H	7-H	8-H	9-H _a	9-H _b
2	4.50	3.95	4.15	3.44	3.65	3.65	3.44
5	4.30	4.02	4.86	4.37	3.79	3.33	3.20
6	4.42	3.94	3.41	4.17	3.83	4.27	3.94

difference between compounds **2**, ¹² **5**, and **6** was elucidated by the chemical shifts of C–H in the ¹H NMR spectra (Table 1). The obvious deviation of proton chemical shift was consistent with introduction and the position change of the *O*-acetyl group. An unambiguous confirmation of compound **6** was provided by X-ray crystal analysis. Figure 1 illustrates the structure, conformation, and atom numbering system of compound **6**.

1. Experimental

1.1. General methods

The reactions were performed with the use of commercial reagents and distilled solvents purified according to standard procedures. Optical rotations were recorded using a Perkin–Elmer 241 polarimeter. ^1H NMR spectra and ^{13}C NMR spectra were obtained in CDCl $_3$ or Me $_2\text{SO-}d_6$ on a Bruker AMX-400/600 equipment at 300 MHz or 400 MHz using TMS as an internal standard at room temperature. The chemical shifts (δ) are recorded in ppm relative to CDCl $_3$ (δ = 7.26) or Me $_2\text{SO-}d_6$ (δ = 2.50) for proton and CDCl $_3$ (δ = 77.0) or Me $_2\text{SO-}d_6$ (δ = 39.5) for carbon. Melting points were

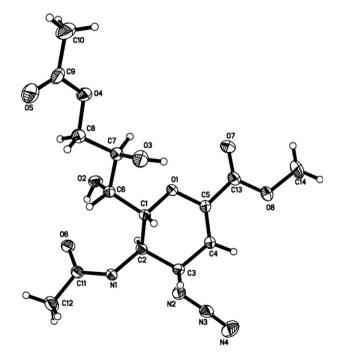


Figure 1. Crystal structure of ${\bf 6}$ showing 50% probability displacement for ellipsoids.

determined by using the capillary method on a Buchi-510 melting point apparatus and are uncorrected. The mass spectrum was recorded on a Finnigan MAT-95/711spectrometer. High-resolution mass spectra (HRMS) were recorded on an FTMS spectrometer. Elemental analyses were performed on a Carlo Erba 1106 analyzer.

1.2. Preparation of methyl 5-acetamido-2,6-anhydro-4-azido-3,4,5-trideoxy-8,9-0-isopropylidene-p-glycero-p-galacto-non-2-enonic acid (3)¹³

Methyl 5-acetamido-2,6-anhydro-4-azido-3,4,5-trideoxy-D-glycero-p-galacto-non-2-enonic acid 2 (724 g, 2.7 mmol) was dissolved in acetone (360 mL), and Dowex 50W \times 8X (H $^{+}$) resin (10 g) was added to the resulting soln while stirring at room temperature to adjust pH to 4. The mixture was then stirred for 3 h. Then, the soln was filtered and evaporated to give an oil, then saturated ag sodium hydrogen carbonate (100 mL) and CH₂Cl₂ (200 mL) were added. After separation, the ag layer was extracted with CH_2Cl_2 (3 × 100 mL). All organic layers were combined. washed with brine (200 mL), dried over anhyd sodium sulfate, and the solvent was removed by distillation under diminished pressure to give a white solid, which was of sufficient quality for use in the next step. An analytically pure sample was obtained via silica gel column chromatography (1:2 hexane-EtOAc). Yield: 21.5 g (80%); $[\alpha]_D^{20}$ +158 (c 0.42, CHCl₃), lit.¹³ +161; mp 95–97 °C; $R_{\rm f}$ 0.55 (20:1 CH₂Cl₂-MeOH); ¹H NMR (300 MHz, CDCl₃): δ 5.95 (d, 1H, J 2.5 Hz, H-3), 5.85 (d, 1H, J 8.0 Hz, NH), 5.35 (dd, 1H, J 2.0, 5.3 Hz, H-7), 4.67 (dd, 1H, J 2.5, 9.0 Hz, H-4), 4.64 (dd, 1H, J 1.9, 11.2 Hz, H-6), 4.38 (dd, 1H, J 6.1, 11.7 Hz, H-8), 4.14 (dd, 1H, J 6.2, 9.0 Hz, H-9), 3.96 (dd, 1H, J 6.1, 8.8 Hz, H-9), 3.81 (s, 3H, CO₂Me), 3.60 (dd, 1H, J 8.7, 18.6 Hz, H-5), 2.14 (s, 3H, OAc), 2.01 (s, 3H, NAc), 1.37 (s, 3H, CH₃), 1.35 (s, 3H, CH₃); ¹³C NMR (400 MHz, CDCl₃): δ 20.73, 20.85, 23.30, 48.67, 52.58, 57.56, 61.96, 67.73, 70.82, 75.69, 107.50, 145.15, 161.50, 170.42, 170.64; HRESIMS: calcd for C₁₅H₂₂N₄O₇Na [M+Na]⁺: 393.1386; found: *m*/*z* 393.1397.

1.3. General procedure for the preparation of methyl 5-acetamido-7-*O*-acyl-2,6-anhydro-4-azido-3,4,5-trideoxy-8,9-*O*-isopropylidene-p-glycero-p-galacto-non-2-enonic acid (4a–e)

A suspension of methyl 5-acetamido-2,6-anhydro-4-azido-3,4, 5-trideoxy-8,9-O-isopropylidene-D-glycero-D-galacto-non-2-enonic acid **3** (100 mg, 0.27 mmol) in pyridine (1.0 mL) was treated with DMAP (4 mg). The resulting mixture was cooled in an ice-water bath while propionic anhydride, butyric anhydride, methoxyacetic anhydride, benzoic anhydride, or acetic anhydride (0.54 mmol) was added dropwise, respectively. After stirring for 10-24 h at room temperature the mixture was concentrated to give an oil which was dissolved in EtOAc (6 mL) and washed sequentially with 1 M hydrochloric acid (2 \times 3 mL), saturated aq sodium hydrogen carbonate (3 \times 3 mL), and finally brine (3 mL). The organic layer was then dried over MgSO₄, concentrated, and the residues were purified by column chromatography or recrystallization.

1.3.1. Methyl 5-acetamido-2,6-anhydro-4-azido-3,4,5-trideoxy-8,9-0-isopropylidene-7-0-propionyl-p-glycero-p-galacto-non-2-enonic acid (4a)

The residue was chromatographed on silica gel with 1:1 EtOAcpetroleum ether to give **4a** (111 mg, 96%); white solid; [α]_D²⁰ +48.1 (c 0.48, MeOH); mp 63–64 °C; $R_{\rm f}$ 0.28 (1:1 EtOAc–petroleum ether); ¹H NMR (400 MHz, CDCl₃): δ 5.95 (d, 1H, J 2.6 Hz, 3-H), 5.85 (d, 1H, J 7.7 Hz, NH), 5.37 (dd, 1H, J 1.9, 5.2 Hz, 7-H), 4.72 (dd, 1H, J 2.7, 9.1 Hz, 4-H), 4.66 (dd, 1H, J 1.8, 10.6 Hz, 6-H), 4.37 (dd, 1H, J 6.2, 11.6 Hz, 8-H), 4.14 (dd, 1H, J 6.1, 9.0 Hz, 9-Ha), 3.95 (dd, 1H, J 6.3, 9.0 Hz, 9-Hb), 3.80 (s, 3H, CO₂Me), 3.53 (dd, 1H, J 8.7, 18.5 Hz, 5-H), 2.43 (m, 2H, CH₂), 2.01 (s, 3H, NAc), 1.36 (s, 3H, CH₃), 1.34 (s, 3H, CH₃), 1.16 (t, 3H, J 7.6 Hz, CH₃); ¹³C NMR

(400 MHz, CDCl₃): δ 8.94, 23.39, 25.21, 26.41, 27.43, 49.81, 52.46, 56.73, 65.52, 68.77, 74.72, 75.27, 107.48, 108.86, 145.01, 161.57, 170.82, 174.27; HRESIMS: calcd for C₁₈H₂₆N₄O₈Na [M+Na]⁺: 449.1648; found: m/z 449.1636. Anal. Calcd for C₁₈H₂₆N₄O₈: C, 50.70; H, 6.15; N, 13.14. Found: C, 50.71; H, 6.19; N, 12.88.

1.3.2. Methyl 5-acetamido-2,6-anhydro-4-azido-3,4,5-trideoxy-7-*O-n*-butyryl-8,9-*O*-isopropylidene-p-*glycero*-p-*galacto*-non-2-enonic acid (4b)

The residue was chromatographed on silica gel with 3:7 EtOAcpetroleum ether to give compound 4b (114 mg, 96%); white solid; $[\alpha]_D^{20}$ +41.6 (c 0.65, MeOH); mp 126–128 °C; R_f 0.41 (1:1 EtOAc– petroleum ether); δ 5.95 (d, 1H, J 7.9 Hz, NH), 5.94 (d, 1H, J 2.8 Hz, 3-H), 5.36 (dd, 1H, / 1.7, 5.1 Hz, 7-H), 4.71 (dd, 1H, / 2.7, 9.1 Hz, 4-H), 4.65 (dd, 1H, / 2.0, 10.4 Hz, 6-H), 4.37 (dd, 1H, / 6.1, 11.5 Hz, 8-H), 4.13 (dd, 1H, / 6.4, 9.0 Hz, 9-Ha), 3.93 (dd, 1H, / 6.2, 9.0 Hz, 9-Hb), 3.80 (s, 3H, CO₂Me), 3.52 (dd, 1H, 1 8.7, 17.4, 18.7 Hz, 5-H), 2.37 (m, 2H, CH₂), 2.00 (s, 3H, NAc), 1.65 (m, 2H, CH₂), 1.35 (s, 3H, CH₃), 1.33 (s, 3H, CH₃), 0.95 (s, 3H, 17.2 Hz, CH₃): 13 C NMR (400 MHz, CDCl₃): δ 13.66, 18.25, 23.45, 25.25, 26.50, 35.99, 49.80, 52.53, 56.84, 65.61, 68.73, 74.75, 75.37, 107.55, 108.93, 145.07, 161.65, 170.93, 173.47; HRESIMS: calcd for C₁₉H₂₈N₄O₈Na [M+Na]⁺: 463.1805; found: *m*/*z* 463.1804. Anal. Calcd for C₁₉H₂₈N₄O₈: C, 51.81; H, 6.41; N, 12.72. Found: C, 51.92; H, 6.34; N, 12.57.

1.3.3. Methyl 5-acetamido-2,6-anhydro-4-azido-3,4,5-trideoxy-8,9-0-isopropylidene-7-0-methoxylacetyl-p-glycero-p-galacto-non-2-enonic acid (4c)

The residue was chromatographed on silica gel with EtOAc (60%) in petroleum ether to give compound **4c** (116 mg, 97%); white solid; $[\alpha]_D^{20}$ +42.0 (*c* 0.23, MeOH); mp 80–81 °C; R_f 0.15 (1:1 EtOAc-petroleum ether); 1 H NMR (400 MHz, CDCl₃): δ 6.50 (d, 1H, J 8.5 Hz, NH), 5.92 (d, 1H, J 2.4 Hz, 3-H), 5.48 (dd, 1H, J 1.5, Hz, 7-H), 4.60 (dd, 1H, J 1.5, 10.5 Hz, 6-H), 4.56 (dd, 1H, J 2.6, 9.5 Hz, 4-H), 4.36 (dd, 1H, J 5.9, 10.7 Hz, 8-H), 4.19 (d, 1H, J 16.8 Hz, CH-H), 4.13 (dd, 1H, J 6.4, 9.1 Hz, 9-Ha), 4.02 (d, 1H, J 16.8 Hz, CH-H), 3.95 (dd, 1H, J 6.2, 9.1 Hz, 9-Hb), 3.78 (s, 3H, CO₂Me), 3.70 (dd, 1H, J 9.2, 19.0 Hz, 5-H), 3.45 (s, 3H, OCH₃), 1.99 (s, 3H, NAc), 1.32 (s, 3H, CH₃), 1.31 (s, 3H, CH₃); ¹³C NMR (400 MHz, CDCl₃): δ 23.30, 25.09, 26.42, 49.69, 52.48, 56.81, 59.40, 65.40, 69.36 (2C), 74.67, 75.23, 107.65, 108.85, 144.99, 161.49, 170.25, 170.99; HRESIMS: calcd for $C_{18}H_{26}N_4O_9N_a$ $[M+Na]^+$: 465.1597; found: m/z 465.1579. Anal. Calcd for C₁₈H₂₆N₄O₉: C, 48.87; H, 5.92; N, 12.66. Found: C, 48.89; H, 5.95; N, 12.49.

1.3.4. Methyl 5-acetamido-2,6-anhydro-4-azido-7-0-benzoyl-3,4,5-trideoxy-8,9-0-isopropylidene-p-glycero-p-galacto-non-2-enonic acid (4d)

The residue was chromatographed on silica gel with 2:3 EtOAcpetroleum ether to give **4d** (126 mg, 98%); white solid; $[\alpha]_D^{20}$ +51.0 (c 0.24, MeOH); mp 166–167 °C; R_f 0.37 (1:1 EtOAcpetroleum ether); ¹H NMR (400 MHz, CDCl₃): δ 8.06 (d, 2H, J 7.2 Hz, 2-H and 2'-H), 7.61 (t, 1H, J 7.5 Hz, 4'-H), 7.47 (d, 1H, J 7.4 Hz, 3-H and 3'-H), 6.04 (d, 1H, J 7.1 Hz, NH), 5.94 (d, 1H, J 2.7 Hz, 3-H), 5.62 (d, 1H, J 1.6, 4.7 Hz, 7-H), 5.08 (dd, 1H, J 2.7, 9.3 Hz, 4-H), 5.00 (dd, 1H, J 1.5, 10.3 Hz, 6-H), 4.51 (dd, 1H, J 6.0, 11.0 Hz, 8-H), 4.23 (dd, 1H, J 6.2, 9.1 Hz, 9-Ha), 4.16 (dd, 1H, J 6.1, 8.9 Hz, 9-Hb), 3.82 (s, 3H, CO₂Me), 3.20 (ddd, 1H, J 7.3, 9.6, 17.1 Hz, 5-H), 2.06 (s, 3H, NAc), 1.34 (s, 3H, CH₃), 1.25 (s, 3H, CH₃); ¹³C NMR (400 MHz, CDCl₃): δ 23.57, 25.28, 26.47, 51.78, 52.49, 55.80, 65.62, 69.93, 74.89, 75.29, 107.70, 109.04, 128.64 (2C), 129.03, 130.01 (2C), 133.74, 145.05, 161.66, 166.37, 171.34; HRESIMS: calcd for $C_{22}H_{26}N_4O_8Na$ [M+Na]*: 497.1648; found: m/z

497.1649. Anal. Calcd for $C_{22}H_{26}N_4O_8$: C, 55.69; H, 5.52; N, 11.81. Found: C, 55.83; H, 5.63; N, 11.55.

1.3.5. Methyl 5-acetamido-7-*O*-acetyl-2,6-anhydro-4-azido-3,4,5-trideoxy-8,9-*O*-isopropylidene-p-*glycero*-p-*galacto*-non-2-enonic acid (4e)

The residue was crystallized with EtOAc–petroleum ether to give **4e** (106 mg, 95%); white solid; $[\alpha]_D^{20}$ +60.8 (c 0.38, MeOH); mp 155–156 °C; R_f 0.55 (20:1 CH₂Cl₂–MeOH); ¹H NMR (400 MHz, CDCl₃): δ 5.95 (d, 1H, J 2.7 Hz, H-3), 5.77 (d, 1H, J 8.1 Hz, NH), 5.35 (dd, 1H, J 1.9, 5.4 Hz, H-7), 4.70 (dd, 1H, J 2.6, 9.1 Hz, H-4), 4.65 (dd, 1H, J 1.6, 10.4 Hz, H-6), 4.38 (dd, 1H, J 6.2, 10.6 Hz, H-8), 4.14 (dd, 1H, J 6.3, 9.0 Hz, H-9), 3.96 (dd, 1H, J 6.3, 9.0 Hz, H-9), 3.81 (s, 3H,CO₂Me), 3.57 (dd, 1H, J 8.7, 18.3 Hz, H-5), 2.14 (s, 3H, OAc), 2.01 (s, 3H, NAc), 1.37 (s, 3H, CH₃), 1.35 (s, 3H, CH₃); ¹³C NMR (400 MHz, CDCl₃): δ 20.94, 23.36, 25.31, 26.47, 52.54, 52.71, 57.40, 65.62, 68.96, 74.81, 75.66, 107.62, 108.96, 145.18, 161.66, 170.85, 170.99; HRESIMS: calcd for C₁₇H₂₅N₄O₈: 413.1672 [M+H]⁺; found: m/z 413.1651. Anal. Calcd for C₁₇H₂₄N₄O₈: C, 49.51; H, 5.87; N, 13.59. Found: C, 49.17; H, 6.07; N, 13.59.

1.4. Methyl 5-acetamido-7-*O*-acetyl-2,6-anhydro-4-azido-3,4,5-trideoxy-p-*glycero*-p-*galacto*-non-2-enonic acid (5)

Methyl 5-acetamido-7-O-acetyl-2,6-anhydro-4-azido-3,4,5-trideoxy-8,9-O-isopropylidene-p-glycero-p-galacto-non-2-enonic acid 4e (13 g, 37 mmol) was dissolved in water (390 mL), and then Dowex $50W \times 8X$ (H⁺) resin (5 g) was added to adjust pH to 4. After reaction for 3 h, the resulting soln was filtered and evaporated to give a foam, then EtOAc was added to crystallize the title compound **5** (8.9 g, 76%); white solid; $[\alpha]_{\rm D}^{20}$ +55.9 (*c* 0.49, MeOH); mp 174–175 °C; $R_{\rm f}$ 0.18 (EtOAc); ¹H NMR (400 MHz, Me₂SO- $d_{\rm 6}$): δ 8.10 (d, 1H, J 9.8 Hz, NH), 5.85 (d, 1H, J 2.5 Hz, 3-H), 5.74 (s, 1H, OH), 4.86 (dd, 1H, J 1.9, 9.5 Hz, 6-H), 4.37 (dd, 1H, J 1.9, 10.3 Hz, 7-H), 4.30 (dd, 1H, J 2.6, 9.3 Hz, 4-H), 4.02 (ddd, 1H, / 8.6, 19.2 Hz, 5-H), 3.79 (ddd, 1H, / 3.7, 7.3, 10.6 Hz, 8-H), 3.74 (s, 3H, CO₂Me), 3.33 (dd, 1H, I 3.5, 11.5 Hz, 9-H), 3.20 (dd, 1H, I 7.3, 11.2 Hz, 9-H), 1.94 (s, 3H, OAc), 1.76 (s, 3H, NAc); ¹³C NMR (400 MHz, Me₂SO- d_6): δ 20.84, 22.67, 46.26, 52.42, 59.05, 63.24, 68.25, 69.45, 75.97, 108.11, 145.14, 161.51, 169.25, 169.67; HRE-SIMS: calcd for $C_{14}H_{21}N_4O_8$: 373.1359 [M+H]⁺; found: m/z373.1356. Anal. Calcd for C₁₄H₂₀N₄O₈ 0.1 MeCOOEt: C, 45.38; H, 5.50; N, 14.70; Found: C, 45.67; H, 5.49; N, 14.47.

1.5. Methyl 5-acetamido-9-*O*-acetyl-2,6-anhydro-4-azido-3,4,5-trideoxy-p-*glycero*-p-*galacto*-non-2-enonic acid (6)

A stirred soln of methyl 5-acetamido-7-O-acetyl-2,6-anhydro-4-azido-3,4,5-trideoxy-D-glycero-D-galacto-non-2-enonic acid **5** (300 mg, 0.81 mmol) in MeOH (50 mL) was heated at reflux for 48 h. The solvent was evaporated under diminished pressure, and the resulting mixture was chromatographed (silica gel, 50:1 CH₂Cl₂-MeOH) to give **6** as a white solid. Single crystals suitable for X-ray analysis were obtained by slow evaporation of an EtOAc–MeOH soln at 298 K. Yield: 240 mg (80%); $[\alpha]_D^{20}$ +74.2 (c 0.99, MeOH); mp 194–195 °C; $R_{\rm f}$ 0.36 (EtOAc); $^1{\rm H}$ NMR (300 MHz, Me₂SO- d_6): δ 8.10 (d, 1H, J 9.8 Hz, NH), 5.85 (d, 1H, J2.5 Hz, 3-H), 5.74 (s, 1H, OH), 4.86 (dd, 1H, 1, 1.9, 9.5 Hz, 6-H), 4.37 (dd, 1H, / 1.9, 10.3 Hz, 7-H), 4.30 (dd, 1H, / 2.6, 9.3 Hz, 4-H), 4.02 (ddd, 1H, J 8.6, 19.2 Hz, 5-H), 3.79 (ddd, 1H, J 3.7, 7.3, 10.6 Hz, 8-H), 3.74 (s, 3H, CO₂Me), 3.33 (dd, 1H, J 3.5, 11.5Hz, 9-H), 3.20 (dd, 1H, J 7.3, 11.2 Hz, 9-H), 1.94 (s, 3H, OAc), 1.76 (s, 3H, NAc); 13 C NMR (400 MHz, Me₂SO- d_6): δ 20.83, 22.68, 47.56, 52.34, 58.09, 66.76, 66.86, 68.21, 76.40, 107.27, 145.23, 161.65, 170.47, 171.92; HRESIMS: calcd for C₁₄H₂₁N₄O₈ [M+Na]⁺:

373.1359; found: m/z 373.1359. Anal. Calcd for $C_{14}H_{20}N_4O_8$: C, 45.16; H, 5.41; N, 15.05. Found: C, 45.24; H, 5.34; N, 15.09.

1.6. X-ray crystallographic data

Crystal data were collected using a Bruker SMART APEX CCD area detector diffractometer. A full sphere of the reciprocal space was scanned by phi–omega scans. Semi-empirical absorption correction based on redundant reflections was performed by the program sadabs. The structures were solved by direct methods using shells-97. The structures were solved by direct methods using shells-97. The hydrogen atom treatment varied depending on the crystal quality. All H atoms were placed in geometrically idealized positions and constrained to ride on their parent atoms, with C–H = 0.93–0.96 Å, N–H = 0.86 Å, and $U_{\rm iso}$ = 1.2 $U_{\rm eq}$ (C) or (N). The crystal structure of **6** was refined to R_1 = 0.0413 (4579 reflections, all unique) and R_1 = 0.0364 (3007 reflections with F_0 > 2 σ (F_0)) by the full-matrix least-squares method using the shells-9719 program based on 241 parameters.

Supplementary data

Complete crystallographic data for the structural analysis have been deposited (Deposition No. CCDC 676214) with the Cambridge Crystallographic Data Center. These data may be obtained, on request, from The Director, CCDC, 12 Union Road, Cambridge, CB2 1EZ, UK (tel.: +44-1223-336408; fax: +44-1223-336033; e-mail: deposit@ccdc.cam.ac.uk or *via* www: http://www.ccdc.cam.ac.uk).

Acknowledgment

This work was supported by National Great Project of Scientific and Technical Supporting Programs Funded by Ministry of Science & Technology of China (No. 2006BAI01B02).

References

- 1. DeNinno, M. P. Synthesis 1991, 583-593.
- 2. Hemeon, I.; Bennet, A. J. Synthesis 2007, 1899–1926.
- 3. von Itzstein, M.; Keifel, M. J. Chem. Rev. **2002**, 102, 471–490.
- 4. Smith, P. W.; Sollis, S. L.; Howes, P. D.; Cherry, P. C.; Starkey, I. D.; Cobley, K. N.; Weston, H.; Scicinski, J.; Merritt, A.; Whittington, A.; Wyatt, P.; Taylor, N.; Green, D.; Bethell, R.; Madar, S.; Fenton, R. J.; Morley, P. J.; Pateman, T.; Beresford, A. J. Med. Chem. 1998, 41, 787–797.
- Taylor, N. R.; Cleasby, A.; Singh, O.; Skarzynski, T.; Wonacott, A. J.; Smith, P. W.; Sollis, S. L.; Howes, P. D.; Cherry, P. C.; Bethell, R.; Colman, P.; Varghese, J. J. Med. Chem. 1998, 41, 798–807.
- Andrews, D. M.; Cherry, P. C.; Humber, D. C.; Jones, P. S.; Keeling, S. P.; Martin, P. F.; Shaw, C. D.; Swanson, S. Eur. J. Med. Chem. 1999, 34, 563–574.
- Masuda, T.; Shibuya, S.; Arai, M.; Yoshida, S.; Tomozawa, T.; Ohno, A.; Yamashita, M.; Honda, T. Bioorg. Med. Chem. Lett. 2003, 13, 669–673.
- 8. Honda, T.; Masuda, T.; Yoshida, S.; Arai, M.; Kaneko, S.; Yamashita, M. *Bioorg. Med. Chem. Lett.* **2002**, *12*, 1925–1928.
- 9. Honda, T.; Masuda, T.; Yoshida, S.; Arai, M.; Kobayashi, Y.; Yamashita, M. *Bioorg. Med. Chem. Lett.* **2002**, *12*, 1921–1924.
- Smith, P. W.; Starkey, I. D.; Howes, P. D.; Sollis, S. L.; Keeling, S. P.; Cherry, P. C.; von Itzstein, M.; Wu, W.-Y.; Jin, B. Eur. J. Med. Chem. 1996, 31, 143–150.
- Jin, W.-X.; Zhao, Q.-J.; Lai, Q.-L.; Wei, Y.-B. Chinese J. Pharm. 2007, 38, 321–324.
- Chandler, M.; Bamford, M. J.; Conroy, R.; Lamont, B.; Patel, B.; Patel, V. K.; Steeples, I. P.; Storer, R.; Weir, N. G.; Wright, M.; Williamson, C. J. Chem. Soc., Perkin Trans. 1 1995, 1173–1180.
- 13. Honda, T.; Kobayashi, Y.; Masuda, T.; Yomashida, M.; Arai, M. U.S. Patent 6,340,702, 2002; *Chem. Abstr.*, **2003**, *1*36, 118703.
- 14. Zhdanov, R. I.; Zhenodarova, S. M. Synthesis 1975, 222-245.
- 15. Doerschuk, A. P. J. Am. Chem. Soc. 1952, 74, 4202-4203.
- 16. Sheldrick, G. M. SADABS; Bruker AXS: Madison, WI 53711, 2000.
- Sheldrick, G. M. SHELXS-97: Program for Crystal Structure Solution; Göttingen 1997.
- 18. Sheldrick, G. M. SHELXS-97: Program for Crystal Structure Refinement, Release 2; Göttingen 1997.
- Sheldrick, G. M. SHELXL-97. Program for Crystal Structure Refinement; University of Göttingen: Göttingen, Germany, 1997.