

2,2'-Bipyridine-4-one (6). A mixture of chelate 7 and ligand 6 (~2:1, 0.21 g) was heated in a sealed glass tube for 9 h at 160–170 °C with a 6.9 *N* solution of HCl (4 mL) in butanol. Water (2 mL) was added to the reaction mixture. The aqueous layer was separated and water was distilled off *in vacuo* to give 0.17 g of 6·HCl. The obtained salt was dissolved in water (1 mL) and a 25% aqueous ammonia (0.3 mL) was added to the solution. The solution was extracted many times with CHCl₃ to give 0.08 g (62%) of bipyridinone 6, m.p. 146–147 °C (from acetone). Lit.⁷ m.p. 145 °C. MS, *m/z*: 172 [M]⁺. ¹H NMR (CDCl₃), δ: 6.54 (dd, 1 H, H-5); 7.15 (s, 1 H, H-3), 7.37 (dd, 1 H, H-6); 7.74 (d, 1 H, H-3'); 7.80–7.95 (m, 2 H, H-4', and H-5'); 8.63 (d, 1 H, H-6'). ¹³C NMR (CDCl₃), δ: 113.5 (C-3); 116.8 (C-5); 120.4 (C-3'); 124.8 (C-5'); 137.5 (C-4'); 139.2 (d, C-6, *J* = 179 Hz); 146.6 (C-2); 149.2 (d, C-6', *J* = 179 Hz); 149.7 (C-2'); 178.8 (C=O).

References

1. L. S. Vasil'ev, O. G. Azarevich, V. S. Bogdanov, B. I. Ugrak, and V. A. Dorokhov, *Izv. Akad. Nauk, Ser. Khim.*, 1994, **43**, 1342 [*Russ. Chem. Bull.*, 1994, **43**, 1282 (Engl. Transl.)].
2. L. S. Vasil'ev, O. G. Azarevich, V. S. Bogdanov, B. I. Ugrak, M. N. Bochkareva, and V. A. Dorokhov, *Zh. Org. Khim.*, 1994, **30**, 1702 [*J. Org. Chem.*, 1994, **30** (Engl. Transl.)].
3. V. A. Dorokhov, M. A. Present, and V. S. Bogdanov, *Izv. Akad. Nauk, Ser. Khim.*, 1995, **44**, 1118 [*Russ. Chem. Bull.*, 1995, **44**, 1080 (Engl. Transl.)].
4. L. S. Vasil'ev, F. E. Surzhikov, O. G. Azarevich, and V. S. Bogdanov, *Izv. Akad. Nauk, Ser. Khim.*, 1996, **45**, 2715 [*Russ. Chem. Bull.*, 1996, **45**, 2574 (Engl. Transl.)].
5. L. S. Vasil'ev, V. S. Bogdanov, E. M. Shashkova, Nguen Kyu Khoa, K. L. Cherkasova, and V. A. Dorokhov, *Izv. Akad. Nauk, Ser. Khim.*, 1993, **42**, 1323 [*Russ. Chem. Bull.*, 1993, **42**, 1266 (Engl. Transl.)].
6. V. A. Dorokhov, V. S. Bogdanov, L. S. Vasil'ev, and Nguen Kyu Khoa, *Izv. Akad. Nauk, Ser. Khim.*, 1996, **45**, 710 [*Russ. Chem. Bull.*, 1996, **45**, 671 (Engl. Transl.)].
7. P. V. Divekar, G. Read, and L. C. Vining, *Can. J. Chem.*, 1967, **45**, 1215.

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Heterocyclization of acetylketene *N,S*-acetals with benzoyl cyanamide

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N-Unsubstituted mono- and diacetylketene *N,S*-acetals undergo cyclization with benzoyl cyanamide to form 2-amino-4-methylthiopyrimidine derivatives.

Key words: mono- and diacetylketene *N,S*-acetals, benzoyl cyanamide, 2-amino-4-methylthiopyrimidines.

Mono- and diacetylketene *N,S*-acetals are used as convenient building blocks for the synthesis of heterocyclic systems.^{1–3}

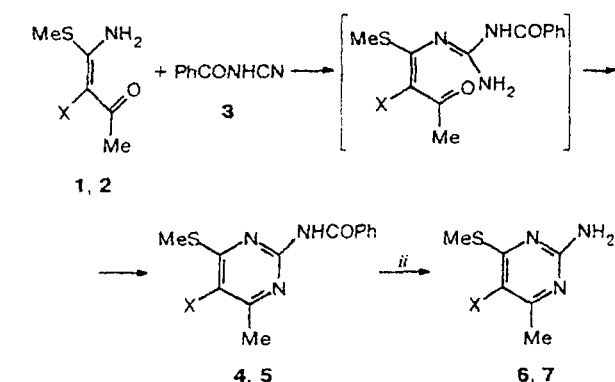
Previously we have shown that the scope of synthetic application for *N,S*-ketene acetals with unsubstituted NH₂ group can be significantly enlarged.^{4–7} The most effective reagents of this type are 3-[amino(methylthio)methylene]pentane-2,4-dione (**1**) (readily obtained from acetylacetone and MeSCN)⁸ and its decetylation product, 1-amino-1-methylthiobut-1-en-3-one (**2**).

In the present work, functionally substituted pyrimidines have been synthesized by heterocyclization of compounds **1** and **2** with benzoyl cyanamide (**3**). Thus, refluxing of *N,S*-acetal **1** and cyanamide **3** in toluene afforded 5-acetyl-2-benzoylamino-6-methyl-4-methyl-

thiopyrimidine (**4**). Debenzoylation of the latter with MeONa in MeOH gave *N*-unsubstituted aminopyrimidine **6** (Scheme 1).

In spite of the presence of the active C-nucleophilic center in molecule **2** (e.g., cyclization of *N,S*-acetal **2** and benzoyl isothiocyanate into 5-acetyl-6-methylthio-2-phenyl-3*H*-pyrimidine-4-thione proceeds with the primary formation of the C—C bond),⁹ compound **2** behaves also as an *N*-nucleophile with respect to cyanamide **3**. The reaction of compounds **2** and **3** in boiling toluene resulted in the condensation product **5**, which was isolated as an oil, identified by ¹H NMR, and converted (without any purification) into crystalline aminopyrimidine **7** by debenzoylation. The structures of functionally substituted pyrimidines **6** and **7** are con-

Scheme 1



Reagents and conditions: i. PhMe, Δ ; ii. MeONa, MeOH, Δ

firmed by spectral methods. Compound 7 has been previously¹⁰ synthesized from 2-amino-4-hydroxy-6-methylpyrimidine in three steps; spectral data have not been given.

Experimental

¹H and ¹³C NMR spectra were recorded on Bruker WM-250 and Bruker AM-300 instruments, respectively. IR spectra were obtained on a Perkin-Elmer 577 instrument, and mass spectra were recorded on a Varian MAT-311A spectrometer (EI, 70 eV). The starting *N,S*-acetals were synthesized by known procedures.⁸

5-Acetyl-2-amino-6-methyl-4-methylthiopyrimidine (6). A mixture of *N,S*-acetal 1 (0.35 g, 2 mmol) and benzoyl cyanamide 3 (0.44 g, 3 mmol) in toluene (10 mL) was refluxed for 3 h. The solvent was evaporated *in vacuo*, and the residue was chromatographed on a SiO₂ column (the eluent was C₆H₆, then a C₆H₆–EtOH mixture, 200 : 1 and 100 : 1). The appropriate fractions were collected and the solvent was evaporated to give 0.42 g (69%) of pyrimidine 4, m.p. 87–88 °C (from a 1 : 4 benzene–hexane mixture). IR (CHCl₃), ν /cm⁻¹: 3424 (NH), 1702 (CO), 1675 sh, 1568, 1534. ¹H NMR (CDCl₃), δ : 2.40 (s, 3 H, Me), 2.59 (s, 3 H, Me), 2.61 (s, 3 H, Me), 7.40–7.68 (m, 3 H, Ph), 7.88–7.98 (m, 2 H, Ph), 8.55 (br.s, 1 H, NH). ¹³C NMR (CDCl₃), δ : 12.81 (SMe), 22.13 (Me), 31.22 (COMe), 127.91 (C-5), 127.51, 128.52, 132.20, 134.21 (Ph), 155.50 (C-2), 162.36 (q, C-6, ²J = 6 Hz), 165.36 (NCO), 168.03 (q, C-4, ³J = 3 Hz), 201.64 (q, C=O, ²J = 6 Hz). MS, *m/z*: 301 [M]⁺. A mixture of pyrimidine 4 (0.42 g, 1.4 mmol) and MeONa (1.4 mmol) in MeOH (15 mL) was refluxed for 30 min. The solvent was evaporated *in vacuo*, and the residue was chromatographed on a SiO₂ column (the eluent was C₆H₆, then a 1 : 1 C₆H₆–CHCl₃ mixture and CHCl₃) to give 0.16 g

(58%) of pyrimidine 6, m.p. 153–154 °C (from a 1 : 4 benzene–hexane mixture). Found (%): C, 48.46; H, 5.55; N, 20.96; S, 15.82. C₈H₁₁N₃OS. Calculated (%): C, 48.71; H, 5.62; N, 21.30; S, 16.25. IR (CHCl₃), ν /cm⁻¹: 3520 and 3415 (NH₂), 1682 (C=O), 1598, 1532. ¹H NMR (CDCl₃), δ : 2.31 (s, 3 H, Me), 2.47 (s, 3 H, Me), 2.54 (s, 3 H, Me), 5.25 (br.s, 2 H, NH₂). ¹³C NMR (CDCl₃), δ : 12.20 (SMe), 22.74 (Me), 31.47 (COMe), 121.41 (C-5), 160.74 (C-2), 163.43 (q, C-6, ²J = 6.3 Hz), 167.23 (q, C-4, ³J = 3.0 Hz), 200.46 (q, C=O, ²J = 6 Hz). MS, *m/z*: 197 [M]⁺.

2-Amino-6-methyl-4-methylthiopyrimidine (7). Pyrimidine 5 was synthesized in the same way as compound 4 from *N,S*-acetal 2 and benzoyl cyanamide 3 and isolated as an oil in 75% yield by SiO₂ chromatography (the eluent was C₆H₆, then a C₆H₆–CHCl₃ mixture, 2 : 1 and 1 : 1). ¹H NMR (CDCl₃), δ : 2.42 (s, 3 H, Me), 2.59 (s, 3 H, Me), 6.80 (s, 1 H, H-5), 7.45–7.61 (m, 3 H, Ph), 7.88–7.98 (m, 2 H, Ph), 8.55 (br.s, 1 H, NH). Debenzoylation of crude product 5 (similar to the transformation of 4 into 6) afforded pyrimidine 7 in 56% yield. m.p. 146–148 °C (cf., Ref. 11). IR (CHCl₃), ν /cm⁻¹: 3520 and 3410 (NH₂), 1598, 1563. ¹H NMR (CDCl₃), δ : 2.25 (s, 3 H, Me), 2.48 (s, 3 H, Me), 5.12 (br.s, 2 H, NH₂), 6.40 (s, 1 H, H-5). ¹³C NMR (CDCl₃), δ : 12.20 (SMe), 23.49 (Me), 107.62 (d, C-5, ¹J = 169 Hz), 162.30 (C-2), 165.67 (q, C-6, ²J = 7.0 Hz), 171.21 (q, C-4, ³J = 3.0 Hz). MS, *m/z*: 155 [M]⁺.

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References

1. R. K. Dieter, *Tetrahedron*, 1986, **42**, 3029.
2. H. Junjappa, H. Ila, and C. V. Asokan, *Tetrahedron*, 1990, **46**, 5423.
3. M. Gelbin and D. Martin, *J. Prakt. Chem.*, 1987, **329**, 753.
4. V. A. Dorokhov, M. F. Gordeev, A. V. Komkov, and V. S. Bogdanov, *Izv. Akad. Nauk, Ser. Khim.*, 1991, **40**, 2593 [*Bull. Acad. Sci. USSR, Div. Chem. Sci.*, 1991, **40**, 2267 (Engl. Transl.)].
5. V. A. Dorokhov, A. V. Komkov, and B. I. Ugrak, *Izv. Akad. Nauk, Ser. Khim.*, 1993, **42**, 1429 [*Russ. Chem. Bull.*, 1993, **42**, 1364 (Engl. Transl.)].
6. V. A. Dorokhov and M. A. Prezent, *Izv. Akad. Nauk, Ser. Khim.*, 1992, **41**, 1455 [*Bull. Russ. Acad. Sci., Div. Chem. Sci.*, 1992, **41**, 1139 (Engl. Transl.)].
7. V. A. Dorokhov, M. A. Prezent, and V. S. Bogdanov, *Izv. Akad. Nauk, Ser. Khim.*, 1995, **44**, 1118 [*Russ. Chem. Bull.*, 1995, **44**, 1080 (Engl. Transl.)].
8. V. A. Dorokhov, M. F. Gordeev, E. M. Shashkova, A. V. Komkov, and V. S. Bogdanov, *Izv. Akad. Nauk SSSR, Ser. Khim.*, 1991, 2600 [*Bull. Acad. Sci. USSR, Div. Chem. Sci.*, 1991, **40**, 2274 (Engl. Transl.)].
9. A. V. Komkov, PhD Thesis (chemistry), Institute of Organic Chemistry, Russian Academy of Sciences, Moscow, 1995.
10. K. Takagi and T. Ueda, *Chem. Pharm. Bull.*, 1963, **11**, 1382.

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