### The Decomposition Reaction of 4-Acetylsydnones Arylhydrazones

Wen-Fa Kuo<sup>a</sup> (•••), Chun-Yen Chiu<sup>b</sup> (•••), I-Tzu Lin<sup>a</sup> (•••), Shu-Ya Sheu<sup>a</sup> (•••), Kao-Hung Lin<sup>a</sup> (•••) and Mou-Yung Yeh<sup>\*a</sup> (•••) <sup>a</sup>Department of Chemistry, National Cheng Kung University, Tainan, Taiwan 70101, R.O.C. <sup>b</sup>Department of Environmental Engineering and Health, Tajen Institute of Technology, Ping Tung, Taiwan 907, R.O.C.

The acidic decomposition of 4-acetylsydnones arylhydrazones 2 results in the formation of 4-arylhydrazo-1,2-pyrazolin-5-ones 3 and 4-arylamino-1,2,3-triazoles 4, respectively. The reactions of 4-acetylsydnones 1 with arylhydrazines 5 afford compounds 3 as the only products in the absence of solvent.

#### INTRODUCTION

In the literature, only the substituted derivatives on the acyl functional group of 4-acylsydnones **1** has been investigated.<sup>1-4</sup> In this report, however, synthetic studies using compounds **1** through decomposition of the sydnone ring to give heterocyclic compounds are of great interest. We also present the decomposition of 4-acetylsydnones arylhydrazones **2** which would be transferred to two heterocyclic compounds, 4-arylhydrazo-1,2-pyrazolin-5-ones **3** and 4-amino-1,2,3-triazoles **4**, in the presence of hot HCl via three subsequent steps: N1-attack, sydnone ring opening and ring formation. In addition, the reactions of 4-acetylsydnones **1** with arylhydrazores **5** were also examined.

Various synthetic routes developed for 1,2-pyrazolin-5-ones<sup>5-9</sup> and 1,2,3-triazoles<sup>10-16</sup> have been reported. In this study, we report the decomposition of sydnone derivatives yielding compounds **3** and **4**.

#### **RESULTS AND DISCUSSION**

From the acidic decomposition of 4-acetylsydnones arylhydrazones **2** in the presence of hot HCl, 4-arylhydrazo-1,2-pyrazolin-5-ones **3** and 4-arylamino-1,2,3-triazoles **4** were obtained with considerably low yields. Instead, high yields of hydrolysis products, 4-acetylsydnones **1**, were obtained. In order to acquire satisfactory yields of compounds **3** and **4**, we modified the above synthetic method. Excess arylhydrazines **5** were used to react with compounds **1**. The resulting products **2** were not isolated and HCl<sub>(aq)</sub> was added to proceed acidic decomposition. Using this method, we obtained compound **3** and **4** with high yields. This is well understood that the condensations were reversible reactions and the excess hydrazines would result in a high quantity of hydrazones

in the reaction solution to perform acidic decomposition. The yields of decomposition are summarized in Table 1.

According to the spectral data and X-ray structure (Fig. 1), compounds **3a-3d** possess two aryl substituents which originated from corresponding arylhydrazines 5. On the other hand, we have also isolated arylamines as side products. On the spectral analysis of compounds 3, the N-H stretching was not obvious in the IR spectra. The N-H resonance signal of  ${}^{1}$ H NMR in each compound is broad and down field shift to 13 ppm. Besides, the C=O stretching was red shift to 1650 cm<sup>-1</sup> due to the intramolecular hydrogen bonding of N-H and C=O. Hence, we proposed the reaction mechanism as outlined in Scheme I. N1-attack resulted in the ring opening of sydnone and the formation of C=N double bond by denitrosolation. The C=N was then attacked by free arylhydrazine 5 to give 4-arylhydrazo-1,2-pyrazolin-5-ones 3. In addition, the structure identifications of 4-arylamino-1,2,3-triazoles 4 by various spectroscopic techniques were similar to those reported in the literature.<sup>16</sup> This proposed reaction mechanism of compounds 4 was also given in Scheme I, which refers to the photolysis mechanism of the sydnone ring reported by Marky.<sup>15</sup> In the presence of HCl, the sydnone ring opened and diazirine 6 was formed as an intermediate by decarboxylation. Then N1-attack resulted in heterolysis of N-N bond of diazirine 6 to give 4-arylamine-1,2,3-triazoles 4.

Furthermore, the acidic decompositions of 4-acetyl-3arylsydnones 1 were also investigated. The results revealed that only the 4-acetyl-3-(4-chlorophenyl)sydnone 1a was transferred to compounds 3d and 4j with low yields (5% and 8%, respectively). For other 4-acetyl-3-arylsydnones, the acidic decomposition did not take place at all and collected compounds were only starting materials. This may be explained that compound 1a was more unstable than other 4-acetyl-3-arylsydnones 1. Hence a small amount of compound 1a was easily decomposed to 4-chlorophenylhydra-

	5		5 5			
Ar	Ar'	Compounds <b>3</b> <sup>a</sup>			Compounds 4 <sup>a</sup>	
			Yield%(1)	Yield%(2)		Yield%(1)
p-MeOC <sub>6</sub> H <sub>4</sub>	p-MeC <sub>6</sub> H <sub>4</sub>	<b>3</b> a	42	-	4a	35
p-MeC <sub>6</sub> H₄	p-MeC <sub>6</sub> H <sub>4</sub>	3a	16	43	4b	24
p-ClC <sub>6</sub> H <sub>4</sub>	p-MeC <sub>6</sub> H <sub>4</sub>	3a	35	42	4c	52
$C_6H_5$	p-MeC <sub>6</sub> H <sub>4</sub>	3a	26	-	4d	57
p-EtOC <sub>6</sub> H <sub>4</sub>	p-MeC <sub>6</sub> H <sub>4</sub>	3a	32	-	4e	25
p-MeOC <sub>6</sub> H <sub>4</sub>	C <sub>6</sub> H <sub>5</sub>	3b	40	-	4f	14
p-MeC <sub>6</sub> H <sub>4</sub>	$C_6H_5$	3b	30	61	4g	25
p-ClC <sub>6</sub> H <sub>4</sub>	$C_6H_5$	3b	28	59	4 <b>h</b>	27
p-EtOC <sub>6</sub> H <sub>4</sub>	$C_6H_5$	3b	32	-	4i	18
p-MeC <sub>6</sub> H <sub>4</sub>	$m-MeC_6H_4$	3c	-	67		-
p-ClC <sub>6</sub> H <sub>4</sub>	m-MeC <sub>6</sub> H <sub>4</sub>	3c	-	56		-
p-ClC <sub>6</sub> H <sub>4</sub>	p-ClC <sub>6</sub> H <sub>4</sub>	3d <sup>b</sup>	5	-	4j <sup>b</sup>	8

Table 1. Reactions of Sydnones 1 with Arylhydrazines 5

<sup>a</sup> Yields: (1) acidic decomposition; (2) reaction in the absence of solvent and acid.

<sup>b</sup> Starting material: 4-acetyl-3-(4-chlorophenyl)sydnone.

zine, which would further be condensed with remained 4acetyl-3-(4-chlorophenyl)sydnone (1a) and followed by subsequent decomposition with HCl as demonstrated in Scheme I. pyrazolin-5-ones 3 are presented in Table 1.

### CONCLUSION

Further study has been carried out on reactions between 4-acetylsydnones 1 and arylhydrazine 5 under more violent conditions (100 °C) in the absence of solvent and acid. 4-Acetylsydnones arylhydrazones 2 were formed as intermediates and 4-arylhydrazo-1,2-pyrazolin-5-ones 3 were then obtained as the only products with satisfactory yields. 4-Amino-1,2,3-triazoles 4 was not produced in this reaction because the decarboxylation of the sydnone ring did not proceed in the absence of HCl. The yields of 4-arylhydrazo-1,2-

A new synthetic route of 4-arylazo-1,2-pyrazolin-5ones **3** and 4-arylamino-1,2,3-triazoles **4** are presented in this report. These results suggest that excess arylhydrazines are needed for the acidic decomposition of 4-acetylsydnones arylhydrazones **2**. The preparation of 4-arylazo-1,2-pyrazolin-5-ones **3** from the reactions of compounds **1** with arylhydrazines **5** in the absence of solvent and HCl is also a new and more efficient route.



Fig. 1. ORTEP drawing of the X-ray structure of 3a.

Decomposition of 4-Acetylsydnones

#### Scheme I



#### **EXPERIMENTAL SECTION**

Melting points (Buchi 535 apparatus) are uncorrected. IR spectra were recorded on a Hitachi 270-30 infrared spectrometer. NMR spectra were measured on a Bruker AMX-200 NMR spectrometer with tetramethylsilane as internal stan-

dard. The mass spectra were registered on a Finnigan MAT TSQ-46C spectrometer at an ionizing potential 70 eV. Elemental analyses were performed on Heraeus CHN-O-Rapid and Tacussel Coulomax 78 analyzers. X-ray analysis was made with a Nonius CAD-4 diffractometer. Column chromatography was carried out on silica gel (Kieselgel 100, 230-400

mesh, E. Merck).

## Preparation of 4-arylhydrazo-1,2-pyrazolin-5-ones 3 and 4-arylamino-1,2,3-triazoles 4

A solution of 4-acetyl-3-(4-methoxyphenyl)sydnone (0.3 g, 1.3 mmol) and 4-methylphenylhydrazine hydrochloride (0.6 g, 3.8 mmol) in 10 mL of EtOH was stirred at 60 °C for 24 h. HCl (2 mL, 37%) was added and stirred continuously for 3 h. After reactions were complete, the precipitate was collected by filtration and recrystallization with EtOAc to give 0.17 g of **3a** (42%). The filtrate was removed under reduced pressure and the crude solid was subjected to chromatography (EtOAc : n-hexane = 1 : 4) to give 0.13 g of **4a** (35%).

#### The acidic decomposition of 4-acetyl-3-(4-chlorophenyl)sydnone 1a

A solution of compound 1a (0.5 g, 2.1 mmol) in 5 mL of EtOAc was added into 1.0 mL of HCl (37%). The reaction mixture was stirred at 60 °C for 72 h. The precipitated solid, **3d**, was collected by filtration and then recrystallized in EtOAc with 0.036 g of **3d** (5%). The filtrate was subjected to chromatography (EtOAc : n-hexane = 1 : 5) to give 0.054 g of **4j** (8%).

## The acidic decomposition of 4-acetyl-3-arylsydnones arylhydrazones 2

A solution of 0.3 g of compounds **2** and 1 mL of HCl (37%) in 5 mL of EtOH was stirred at 60 °C for 24 h. The precipitated solid was collected and recrystallized in EtOAc, and 8-12% of compound **3** was obtained. The filtrate was subjected to chromatography (EtOAc : n-hexane = 1 : 4) to give compounds **4** (3-7%) and 4-acetyl-3-arylsydnones (70-80%).

# The reactions of 4-acetyl-3-arylsydnones 1 with arylhydrazines 5 in the absence of HCl and solvent

0.25 g of 4-acety-3-(4-methylphenyl)sydnone (1.1 mmol) was added to 2 mL of phenylhydrazine (20 mmol). The solution was then heated to 100 °C and stirred for 5 h. After reactions were complete, the solution was dropped into acidic ice water (30 mL of water and 2 mL (37%) of HCl), precipitate was then collected and recrystallization was accomplished with EtOAc or acetone to give 0.2 g of **3b** (61%).

### 1-(4-Methylphenyl)-4-(4-methylphenylhydrazo)-3-methyl-1,2-pyrazolin-5-one (3a)

Red powder; mp 218-219 °C, IR (KBr), cm<sup>-1</sup>: 1653 (v C=O), 1593 (v C=N), <sup>1</sup>H NMR (DMSO-d<sub>6</sub>),  $\delta$  13.4 (br, 1H), 7.78 (d, *J* = 8.5 Hz, 2H), 7.49 (d, *J* = 8.4 Hz, 2H), 7.27-7.22 (m, 4H), 2.31 (s, 3H), 2.30 (s, 3H), 2.27 (s, 3H), EIMS (70 eV), *m/z* (%): 306 (M<sup>+</sup>, 100), 215 (21), 107 (31), 91 (78), 79 (27),

## 1-Phenyl-4-phenylhydrazo-3-methyl-1,2-pyrazolin-5-one (3b)

Red powder; mp 153.5-154.5 °C, IR (KBr), cm<sup>-1</sup>: 1659 ( $\nu$  C=O), 1596 ( $\nu$  C=N), <sup>1</sup>H NMR (DMSO-d<sub>6</sub>),  $\delta$  13.27 (br, 1H), 7.92 (d, J = 7.4 Hz, 2H), 7.62 (d, J = 7.4 Hz, 2H), 7.48-7.18 (m, 6H), 2.29 (s, 3H), EIMS (70 eV), *m/z* (%): 278 (M<sup>+</sup>, 100), 252 (21), 201 (42), 173 (10), 160 (13), 93 (39), 77 (75). Anal. Calcd. for C<sub>16</sub>H<sub>14</sub>N<sub>4</sub>O: C, 69.05; H, 5.07; N, 20.13. Found C, 69.11; H, 5.13; N, 20.22.

#### 1-(3-Methylphenyl)-4-(3-methylphenylhydrazo)-3-methyl-1,2-pyrazolin-5-one (3c)

Red powder; mp 116-116.5 °C, IR (KBr), cm<sup>-1</sup>: 1650 (v C=O), 1590 (v C=N), <sup>1</sup>H NMR (DMSO-d<sub>6</sub>),  $\delta$  13.31 (br, 1H), 7.74-7.71 (m, 2H), 7.41-7.01 (m, 6H), 2.34 (s, 3H), 2.32 (s, 3H), 2.29 (s, 3H), EIMS (70 eV), *m/z* (%): 306 (M<sup>+</sup>, 100), 215 (21), 187 (4), 91 (31), 77 (6). Anal. Calcd. for C<sub>18</sub>H<sub>18</sub>N<sub>4</sub>O: C, 70.57; H, 5.92; N, 18.29. Found C, 70.56; H, 5.96; N, 18.30.

### 1-(4-Chlorophenyl)-4-(4-chlorophenylhydrazo)-3-methyl-1,2-pyrazolin-5-one (3d)

Red powder; mp 232-233 °C, IR (KBr), cm<sup>-1</sup>: 1659 (v C=O), 1595 (v C=N), <sup>1</sup>H NMR (DMSO-d<sub>6</sub>),  $\delta$  13.15 (br, 1H), 7.92 (d, *J* = 9.0 Hz, 2H), 7.75 (d, *J* = 8.9 Hz, 2H), 7.53-7.46 (m, 4H), 2.32 (s, 3H). EIMS (70 eV), *m/z* (%): 350 (M<sup>+</sup>+4, 11), 348 (M<sup>+</sup>+2, 65), 346 (M<sup>+</sup>, 100), 235 (27), 207 (10), 192 (6), 111 (39), 99 (14). Anal. Calcd. for C<sub>16</sub>H<sub>12</sub>N<sub>4</sub>OCl<sub>2</sub>: C, 55.35; H, 3.48; N, 16.14. Found C, 55.30; H, 3.56; N, 16.25.

### 2-(4-Methylphenyl)-4-(4-methoxyphenylamino)-5-methyl-2H-1,2,3-triazole (4a)

Yellow powder; mp 135.5-136.5 °C, IR (KBr), cm<sup>-1</sup>: 3382 (v NH), 1608 (v C=N), <sup>1</sup>H NMR (DMSO-d<sub>6</sub>),  $\delta$  8.30 (s, 1H), 7.74 (d, *J* = 8.4 Hz, 2H), 7.52 (d, *J* = 9.0 Hz, 2H), 7.28 (d, *J* = 8.4 Hz, 2H), 6.89 (d, *J* = 9.0 Hz, 2H), 3.70 (s, 3H), 2.32 (s, 6H). EIMS (70 eV), *m/z* (%): 294 (M<sup>+</sup>, 100), 279 (42), 147 (6), 119 (6), 105 (21), 91 (22). Anal. Calcd. for C<sub>17</sub>H<sub>18</sub>N<sub>4</sub>O: C, 69.37; H, 6.16; N, 19.03. Found C, 69.22; H, 6.18; N, 19.15.

#### 2-(4-Methylphenyl)-4-(4-methylphenylamino)-5-methyl-2H-1,2,3-triazole (4b)

Yellow powder; mp 115.5-116 °C, IR (KBr), cm<sup>-1</sup>: 3400 (v NH), 1602 (v C=N), <sup>1</sup>H NMR (DMSO-d<sub>6</sub>),  $\delta$  8.36 (s, 1H), 7.74 (d, *J* = 8.4 Hz, 2H), 7.44 (d, *J* = 8.4 Hz, 2H), 7.28 (d, *J* = 8.4 Hz, 2H), 7.08 (d, *J* = 8.4 Hz, 2H), 2.32 (s, 6H), 2.22 (s, 3H). EIMS (70 eV), *m/z* (%): 278 (M<sup>+</sup>, 100), 105 (38), 91 (30), 78 (13), 65 (18), 51 (5). Anal. Calcd. for C<sub>17</sub>H<sub>18</sub>N<sub>4</sub>: C, 73.35;

H, 6.52; N, 20.13. Found C, 73.24; H, 6.59; N, 20.21.

#### 2-(4-Methylphenyl)-4-(4-chlorophenylamino)-5-methyl-2*H*-1,2,3-triazole (4c)

Yellow powder; mp 113-114 °C, IR (KBr), cm<sup>-1</sup>: 3424 ( $\nu$  NH), 1602 ( $\nu$  C=N), <sup>1</sup>H NMR (DMSO-d<sub>6</sub>),  $\delta$  8.68 (s, 1H), 7.76 (d, *J* = 8.5 Hz, 2H), 7.56 (d, *J* = 8.9 Hz, 2H), 7.33-7.27 (m, 4H), 2.33 (s, 6H). EIMS (70 eV), *m/z* (%): 300 (M<sup>+</sup>+2, 34), 298 (M<sup>+</sup>, 100), 152 (7), 105 (42), 91 (19), 78 (10), 65 (11). Anal. Calcd. for C<sub>16</sub>H<sub>15</sub>N<sub>4</sub>Cl: C, 64.32; H, 5.06; N, 18.75. Found C, 64.30; H, 5.11; N, 18.79.

## 2-(4-Methylphenyl)-4-phenylamino-5-methyl-2*H*-1,2,3-triazole (4d)

Yellow powder; mp 112.5-113.5 °C, IR (KBr), cm<sup>-1</sup>: 3436 (v NH), 1599 (v C=N), <sup>1</sup>H NMR (DMSO-d<sub>6</sub>),  $\delta$  8.50 (s, 1H), 7.77 (d, *J* = 8.4 Hz, 2H), 7.55 (d, *J* = 8.2 Hz, 2H), 7.32-7.23 (m, 4H), 6.87-6.80 (m, 1H), 2.33 (s, 3H), 2.34 (s, 3H). EIMS (70 eV), *m/z* (%): 264 (M<sup>+</sup>, 100), 187 (6), 142 (5), 118 (6), 107 (23), 105 (36), 91 (21), 77 (20), 65 (11). Anal. Calcd. for C<sub>16</sub>H<sub>16</sub>N<sub>4</sub>: C, 72.70; H, 6.10; N, 21.19. Found C, 72.85; H, 6.36; N, 21.46.

### 2-(4-Methylphenyl)-4-(4-ethoxyphenylamino)-5-methyl-2H-1,2,3-triazole (4e)

Yellow powder; mp 120-121 °C, IR (KBr), cm<sup>-1</sup>: 3388 (v NH), 1602 (v C=N), <sup>1</sup>H NMR (DMSO-d<sub>6</sub>),  $\delta$  8.27 (s, 1H), 7.73 (d, *J* = 8.1 Hz, 2H), 7.49 (d, *J* = 8.9 Hz, 2H), 7.28 (d, *J* = 8.1 Hz, 2H), 6.87 (d, *J* = 8.9 Hz, 2H), 3.96 (q, *J* = 6.9 Hz, 2H), 2.32 (s, 6H), 1.30 (t, *J* = 6.9 Hz, 3H). EIMS (70 eV), *m/z* (%): 308 (M<sup>+</sup>, 100), 279 (65), 105 (29), 91 (30), 78 (10), 65 (14), 57 (9). Anal. Calcd. for C<sub>18</sub>H<sub>20</sub>N<sub>4</sub>O: C, 70.11; H, 6.54; N, 18.17. Found C, 70.19; H, 6.67; N, 18.31.

### 4-(4-Methoxyphenylamino)-2-phenyl-5-methyl-2*H*-1,2,3-triazole (4f)

Yellow powder; mp 97.5-98.5 °C, IR (KBr), cm<sup>-1</sup>: 3370 (v NH), 1611 (v C=N), <sup>1</sup>H NMR (DMSO-d<sub>6</sub>),  $\delta$  8.34 (s, 1H), 7.85 (d, *J* = 7.8 Hz, 2H), 7.56-7.43 (m, 4H), 7.20-7.27 (m, 1H), 6.89 (d, *J* = 9.1 Hz, 2H), 3.71 (s, 3H), 2.33 (s, 3H). EIMS (70 eV), *m/z* (%): 280 (M<sup>+</sup>, 100), 265 (43), 92 (11), 91 (19), 77 (23). Anal. Calcd. for C<sub>16</sub>H<sub>16</sub>N<sub>4</sub>O: C, 68.55; H, 5.75; N, 19.99. Found C, 68.48; H, 5.83; N, 20.06.

# 4-(4-Methylphenylamino)-2-phenyl-5-methyl-2*H*-1,2,3-triazole (4g)

Yellow powder; mp 110-111 °C, IR (KBr), cm<sup>-1</sup>: 3406 ( $\nu$  NH), 1599 ( $\nu$  C=N), <sup>1</sup>H NMR (DMSO-d<sub>6</sub>),  $\delta$  8.42 (s, 1H), 7.86 (d, *J* = 8.0 Hz, 2H), 7.53-7.43 (m, 4H), 7.28-7.20 (m, 1H), 7.09 (d, *J* = 8.2 Hz, 2H), 2.34 (s, 3H), 2.23 (s, 3H). EIMS (70

eV), *m/z* (%): 264 (M<sup>+</sup>, 100), 222 (16), 132 (13), 91 (39), 77 (26). Anal. Calcd. for C<sub>16</sub>H<sub>16</sub>N<sub>4</sub>: C, 72.70; H, 6.10; N, 21.20. Found C, 72.58; H, 6.15; N, 21.33.

### 4-(4-Chlorophenylamino)-2-phenyl-5-methyl-2*H*-1,2,3triazole (4h)

Yellow powder; mp 91.5-92.5 °C, IR (KBr), cm<sup>-1</sup>: 3400 ( $\nu$  NH), 1599 ( $\nu$  C=N), <sup>1</sup>H NMR (DMSO-d<sub>6</sub>),  $\delta$  8.73 (s, 1H), 7.89 (d, *J* = 7.7 Hz, 2H), 7.63-7.46 (m, 4H), 7.35-7.24 (m, 3H), 2.36 (s, 3H). EIMS (70 eV), *m/z* (%): 286 (M<sup>+</sup>+2, 43), 284 (M<sup>+</sup>, 100), 92 (9), 77 (30), 64 (12). Anal. Calcd. for C<sub>15</sub>H<sub>13</sub>N<sub>4</sub>Cl: C, 63.27; H, 4.60; N, 19.68. Found C, 63.21; H, 4.77; N, 19.78.

# 4-(4-Ethoxyphenylamino)-2-phenyl-5-methyl-2*H*-1,2,3-triazole (4i)

Yellow powder; mp 102.5-103 °C, IR (KBr), cm<sup>-1</sup>: 3436 ( $\nu$  NH), 1599 ( $\nu$  C=N), <sup>1</sup>H NMR (DMSO-d<sub>6</sub>),  $\delta$  8.32 (s, 1H), 7.84 (d, *J* = 7.8 Hz, 2H), 7.55-7.40 (m, 4H), 7.26-7.19 (m, 1H), 6.87 (d, *J* = 8.9 Hz, 2H), 3.95 (q, *J* = 6.9 Hz, 2H), 2.33 (s, 3H), 1.29 (t, *J* = 6.9 Hz, 3H). EIMS (70 eV), *m/z* (%): 294 (M<sup>+</sup>, 100), 266 (32), 265 (76), 92 (21), 77 (36). Anal. Calcd. for C<sub>17</sub>H<sub>18</sub>N<sub>4</sub>O: C, 69.37; H, 6.16; N, 19.03. Found C, 69.31; H, 6.20; N, 19.06.

# 2-(4-Chlorophenyl)-4-(4-chlorophenylamino)-5-methyl-2*H*-1,2,3-triazole (4j)

Yellow powder; mp 127.5-128.5 °C, IR (KBr), cm<sup>-1</sup>: 3436 (v NH), 1602 (v C=N), <sup>1</sup>H NMR (DMSO-d<sub>6</sub>),  $\delta$  8.77 (s, 1H), 7.87 (d, *J* = 8.8 Hz, 2H), 7.60 (d, *J* = 8.5 Hz, 2H), 7.55 (d, *J* = 8.8 Hz, 2H),7.32 (d, *J* = 8.5 Hz, 2H), 2.35 (s, 3H). EIMS (70 eV), *m*/*z* (%): 322 (M<sup>+</sup>+4, 11), 320 (M<sup>+</sup>+2, 63), 318 (M<sup>+</sup>, 100), 264 (10), 219 (11), 193 (19), 179 (10), 163 (13), 152 (15), 149 (10), 138 (16), 127 (35),. Anal. Calcd. for C<sub>15</sub>H<sub>12</sub>N<sub>4</sub>Cl<sub>2</sub>: C, 56.44; H, 3.79; N, 17.55. Found C, 56.37; H, 3.82; N, 17.59.

#### ACKNOWLEDGMENT

We thank the National Science Council of the Republic of China for financial support of the work.

Received February 25, 2000.

#### **Key Words**

Acidic decomposition; 4-Acetylsydnones arylhydrazones; 4-Arylazo-1,2-pyrazolin-5-ones; 4-Arylamino-1,2,3-triazoles.

#### REFERENCES

- Tien, H. J.; Kanda, K.; Chinone, A.; Ohta, M. Bull. Chem. Soc. Japan 1973, 46, 3304.
- 2. Yeh, M. Y.; Tien, H. J.; Fuchigami, T.; Nonaka, T. J. Chin. Chem. Soc. **1986**, *33*, 61.
- 3. Yeh, M. Y.; Tien, H. J. J. Chin. Chem. Soc. 1986, 33, 83.
- Kuo, W. F.; Lee, C. Y.; Yeh, M. Y. J. Chin. Chem. Soc. 2000, 47, 227.
- 5. Boese, A. B. Jr. Ind. Eng. Chem. 1940, 32, 16.
- 6. Levin, P. A. Zh. Prik. Khim. 1961, 34, 2808.
- 7. Dykhanov, N. N. Med. Prom. S. S. S. R. 1961, 15(1), 42.
- Jones, R.; Ryan, A. J.; Sternhwll, S.; Wright, S. E. *Tetra*hedron 1963, 19, 1497.

- 9. Arriau, J.; Campillo, J. P. Tetrahedron 1974, 1345.
- 10. Hann, R. M.; Hundson, C. S. J. Am. Chem. Soc. 1944, 66, 735.
- 11. Benson, F. R.; Savel, W. L. Chem. Rev. 1950, 46, 1.
- 12. Wittig, C.; Bangert, F.; Kleiner, H. Ber. 1928, 61B, 1140.
- 13. Ruccia, M.; Vivona, N.; Spinelli, D. Adv. Hetercycl. Chem. 1981, 29, 141.
- Frenna, V.; Vivona, N.; Consiglio, G; Spinelli, D. J. Chem. Soc. Perkin Trans 2, 1984, 541.
- Marky, M.; Meier, H.; Wunderli, A.; Heimgartner, H.; Schmid, H.; Hansen, H. J. *Helv. Chim. Acta* **1978**, *61*, 1477.
- Begtrup, M.; Holm, J. J. Chem. Soc. Perkin Trans 1, 1981, 503.