The Synthesis of (±)-Pyrenolide C Using the Oxazole-Singlet Oxygen Reaction. Assignment of Stereochemistry.

Harry H. Wasserman* and K. Spencer Prowse

Department of Chemistry, Yale University, New Haven, CT 06511 USA

Key Words: macrolides; singlet oxygen; oxazoles; triamides; antifungal agents

Abstract: Pyrenolide C has been synthesized for the first time using an oxazole template to construct the framework of the ten-membered unsaturated lactone system. The stereochemical assignment, based on the mode of synthesis, has been confirmed by NMR studies.

The morphogenic substances recently isolated by Nukina from *Pyrenophora teres*, pyrenolides A (1), B (2) and C,(3) are highly functionalized unsaturated lactones which are of special interest as antifungal agents.¹ These systems have a secondary methyl group in the (R) configuration at C-9, and the 2,3-Z, 5,6-E-dienedione unsaturated grouping in common, and differ only in the pattern of substitution at the C₇.C₈ positions.



We have previously studied the use of oxazoles as protecting groups for carboxylic acids which may be regenerated in activated (triamide) form by singlet oxygen oxidation,² and have now explored the application of this methodology to the synthesis of the pyrenolides. In this paper, we report the first total synthesis of (\pm) -pyrenolide C by a route which resolves the stereochemical ambiguity at C-7.

The initial objective in our synthesis was the α,β -unsaturated macrolide 4. We hoped to convert 4 to the β,γ -isomer 5 which could then be transformed to a trans-epoxide 6 for subsequent ring opening to the allylicalcohol system in the product 3a, (Scheme 1) which, as will be shown, contains the desired stereochemistry corresponding to the natural product.





We began our synthesis of 4 using a Wittig coupling of the aldehyde 7 with the ylide partner 8 to form the E enone 9 (95%) incorporating the complete framework of the macrolide target. Compound 7 was efficiently formed from commercially available 6-methyl-5-heptene-2-ol 10, and the ylide component 8 was prepared by the reaction of 4-chloromethyloxazole 11³ with the enolate 12 (Scheme 2).⁴ We found it advantageous to protect the ketone 9 by converting it to the ethylene ketal 13 with ethylene glycol in the presence of p-TsOH. Under these conditions, the double bond shifted into the favorable β,γ -position (4:1/ β,γ : α,β) with respect to the ketal.



In the next phase, the Troc protecting group was removed with zinc in 10% acetic acid/THF to give the alcohol 14 (89%). Dye-sensitized photooxidation (Sensitox, sensitizer) then transformed 14 to the triamide 15 (93%) which was slowly added to refluxing benzene in the presence of collidine p-toluenesulfonate, under modified high dilution, to yield the ten membered lactone 16 (47%) as a mixture of *cis* and *trans* isomers⁵ (Scheme 3). Treatment of 16 with lithium bis-trimethylsilylamide and diphenyldiselenide yielded α -selenolactones 17 (71%) as a mixture of isomers, which, on removal of the ketal protecting group, (acetone/H₂SO₄, 60°, 2 days) yielded two isolable products: the *cis* selenium derivative 18 (43%) and the *trans* isomer 19 (57%).



Scheme 4



Oxidation of each isomer with m-CPBA followed by treatment with pyridine, yielded the two Z-enediones 20 and 21⁶ (Scheme 4). Both *cis* and *trans* alkenes (20 and 21) were oxidized with m-CPBA to give the corresponding epoxides (22 and 23) in essentially quantitative yields. On treatment with base (NaOt-Bu) or Lewis acids (FeCl₃-SiO₂), the *cis* epoxide 21 gave only products of decomposition. By contrast, the *trans* epoxide 23 was readily transformed by silica gel to an allylic alcohol (Scheme 5) which was identical (TLC, IR, ¹H MNR, MS spectrum) with natural pyrenolide C.⁷

Scheme 5







Assignment of the syn-relationship between the allylic hydroxyl and the secondary methyl group in 3a followed from the mode of generating the allyl alcohol in the last step. Thus, in the opening of the epoxide by a β -elimination⁸ as shown in Scheme 6, a *trans*-enone system would be formed in which the newly formed hydroxyl group would bear a syn orientation with respect to the methyl group. This stereochemical feature was confirmed by decoupling experiments in which the pair of protons, H_B and H_D, as well as the pair, H_A and H_B were shown to be 160-180° apart. The geometrical assignment was based on examination of coupling constants and the corresponding angles estimated from a Karplus diagram for vicinal protons.⁹ Finally, in NOE experiments, irradiation of protons D and A gave enhancements in complete accord with the stereochemistry shown in 3a.

Scheme 6



References and Footnotes

- a) Nukina, M.; Ikeda, M.; Sassa, T. Agric. Biol. Chem. 1980, 44, 2761. b) Nukina, M.; Sassa, T.; Ikeda, M. Tet. Lett. 1980, 21, 301. c) Suzuki, S.; Tanaka, A.; Yamashita, K. Agric. Biol. Chem. 1987, 51, 3095.
- a) Wasserman, H.H.; Gambale, R.J. J. Am. Chem. Soc. 1985, 107, 1423; b) Wasserman, H.H.; Gambale, R.J.; Pulwer, M.J. Tetrahedron 1981, Symposium in Print, 37, 4059; c) Wasserman, H.H.; Gambale, R.J. Tetrahedron Lett. 1981, 22, 4849; d) Wasserman, H.H.; Gambale, R.J.; Pulwer, M.J. Tetrahedron Lett. 1981, 22, 1737; e) Wasserman, H.H.; Pickett, J.E.; Vinick, F.S. Heterocycles 1981, 15, 1068; f) Wasserman, H.H.; Floyd, M.B. Tetrahedron, Supplement 7 1966, 441; g) Wasserman, H.H.; Lu, T.-J. Tetrahedron Lett. 1982, 23, 3831.
- 3. Goto, Y.; Yamuzaki, M.; Hamana, M.; Chem. Pharm. Bull. 1971, 49, 2050.
- 4. In initial work, we found that with 12, the reaction of the 2-chloromethyl isomer yielded sideproducts resulting from o-alkylation and lithium-halogen exchange.
- 5. Deketalization of this mixture by exchange with acetone (H₂SO₄, catalyst) permitted separation into the *cis* product (30%) and the *trans* isomer (62%).
- a) Grieco, P.A..; Pogonowski, C.S.; Burke, S. J. Org. Chem. 1975, 40, 542. b) Reich, H.J.; Reich, I.L.; Renga, J.M. J. Am. Chem. Soc. 1973, 95, 5183. c) Sharpless, K.B.; Lauer, R.F.; Teranishi, A.Y. J. Am. Chem. Soc. 1973, 95, 6137.
- 7. We thank Professor M.Nukina for generously proving us with samples of natural pyrenolides A, B, and C.
- 8. Fujitsawa, T.; Takeuchi, M.; Sato, T. Chem. Lett. 1982, 1975.
- 9. Spectrometric Identification of Organic Compounds, 4th Edition; Silverstein, R.M.; Bassler, G.C.; Morrill, T.C.; John Wiley & sons, Inc.: New York, 1981, p. 210.

Acknowledgement: The support of this work by Grant GM-13854 from the General Medical Sciences Institute of the N.I.H. is gratefully acknowledged. (Received in UK 28 May 1992)