Chemistry of Polyhalogenated Nitrobutadienes, 6: A New Ring-Closure Approach to Perfunctionalized 5-Nitropyrimidines

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Abstract: The recently reported capability of pentachloro-2-nitrobutadiene and some closely related derivatives to allow the synthesis of highly substituted acyclic as well as five-membered (hetero)cyclic compounds is herein extended to pyrimidines with an exceptional substitution pattern. Our novel approach starts from a 1,1diamino-3,4,4-trichloro-2-nitrobuta-1,3-diene and an appropriate amidine. Some of the resulting perfunctionalized pyrimidines were subjected to further transformations to give promising candidates with respect to biological, especially pharmacological, activity.

Key words: nitro compounds, halides, pyrimidines, ring closure, nucleophilic substitution

The electronically remarkable pentachloro-2-nitrobuta-1,3-diene and similar polysubstituted butadienes are versatile starting materials for a series of synthetic applications. Among these, we have previously synthesized thiophenes, developed an unprecedented C_4 -to- C_3 cleavage of such a butadiene, and taken advantage of the dienophilic properties of these electron-deficient molecules. In addition, we have obtained imidazolidines, oxazolidines, and related five-membered 1,3-heterocycles from the corresponding thiolated butadiene derivatives.¹

Further efforts are underway to explore the entire synthetic potential of this type of compound. In this paper, we give the first account of the reaction of 1,1-bis(1*H*-benzotriazol-1-yl)-3,4,4-trichloro-2-nitrobuta-1,3-diene (1) or the 1-amino-1-(benzotriazol-1-yl)-3,4,4-trichloro-2-nitrobuta-1,3-dienes 2–5 with acetamidine or benzamidine hydrochlorides. The latter nitrobutadienes were prepared from the bis(benzotriazole) derivative 1 in up to 95% yield in accordance with the procedure of Zapol'skii and Kaberdin.² Thus, applying sodium hydride or lithium hydride, aqueous sodium hydroxide, or other bases in tetrahydrofuran or dimethyl sulfoxide to 2–5 we obtained the pyrimidines 6–10 in up to 85% yield (Table 1).

The optimized procedure includes the application of three equivalents of the amidine with a fourfold excess of sodium hydride in tetrahydrofuran and requires the reaction mixture to be initially maintained at -10 °C and then at room temperature. The proposed mechanism of this novel ring formation is depicted in Scheme 1.

SYNTHESIS 2008, No. 2, pp 0304–0310 Advanced online publication: 07.12.2007 DOI: 10.1055/s-2007-990948; Art ID: T13807SS © Georg Thieme Verlag Stuttgart · New York The free amidine is initially generated by the added base. The ring-closing reaction starts with vinylic nucleophilic substitution at C1, whereupon the amino group of the amidine acts as the nucleophile and, expectedly, the benzotriazolyl substituent plays the role of leaving group. Then, as the key step, the imino nitrogen within the amidine moiety of I forms the six-membered heteroring II in an intramolecular cyclization reaction. Subsequently, hydrochloric acid is eliminated from II by additional base, giving the exomethylene compound III. Finally, aromatization occurs that leads to the persubstituted pyrimidines 6-10. As an example, the structure of 7 was confirmed by X-ray crystallography (Figure 1).^{3,4}

Evidence for the mechanism proposed in Scheme 1 was obtained by isolating the initial substitution product 11 from the bis(benzotriazole) derivative 1 and benzamidine and then cyclizing it to the pyrimidine derivative 12 with base. In conclusion, 1 afforded 11 in 79% yield employing benzamidine hydrochloride and sodium hydrogen carbonate in acetonitrile. The subsequent ring closure was accomplished using potassium carbonate in tetrahydrofuran to give pyrimidine 12 (43%). In contrast, direct conversion of 1, i.e. without isolation of the intermediate 11, re-



Scheme 1 Proposed mechanism of pyrimidine formation

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Bzt Cl Cl	Bzt NO2 HNR ¹ R ² MeOH CI	Bzt Cl Cl 2-	NR ¹ R ² F NO ₂ Cl 5	NH NH ₂ ·HCI NaH, THF	$\begin{array}{c} CI & NO_2 \\ CI & II \\ N & N \\ R^3 \\ 6-10 \end{array}$	R ²	
Entry	NR ¹ R ²	2–5	Yield (%)	Base	Solvent	Product	Yield (%)
1		2	88	NaH	THF	6 , $R^3 = Ph$	73
2	N			NaOH	DMSO		61
3	NO	2	88	NaH	DMSO	7 , $R^3 = Me$	56
4				NaH	THF		54
5				LiH	THF		22
6				Cs ₂ CO ₃	DMSO		21
7	N	3	95	NaH	THF	8 , $R^3 = Ph$	76
8				NaOH	DMSO		40
9	N_N_F	4	95	NaH	THF	9 , $R^3 = Me$	35
10				NaOH	DMSO		32
11		5	57	NaH	THF	10 , $R^3 = Me$	85
12				NaOH	DMSO		30

Table 1 Synthesis of 4-Amino-6-(dichloromethyl)-5-nitropyrimidines

sulted in the same product, but with lower yield (Scheme 2).

It is worthwhile to note that syntheses of 5-nitropyrimidines are rare in literature. One method, published in 1977, employs the condensation of sodium nitromalonaldehyde with amidines.⁵ The same salt is part of a threecomponent ring-closure reaction in combination with 2methylisothiourea sulfate and morpholine.⁶ An alternative access uses the reaction of 5-nitro-1*H*-pyrimidine-2,4-dione with phosphoryl chloride to obtain 2,4-dichloro-5-nitropyrimidine.⁷ Another synthesis starts from a mixture of 3-nitrobut-3-en-2-ones, 2-methylisourea, and 2,3-dichloro-5,6-dicyano-1,4-benzoquinone to afford 2-methoxy-5nitropyrimidine.⁸ In addition, 5-nitropyrimidines have been obtained by ring opening of a 3-nitrochromen-4one.⁹

Aside from the viewpoint of an organic chemist, such 5nitropyrimidines are interesting due to their structural similarity to natural compounds, such as bases in nucleosides and nucleotides, and their corresponding physiological activity. Various bioactivities of pyrimidines and other heterocycles are highlighted in a review.¹⁰

Among numerous applications, some examples are noteworthy: antitumor activity is documented^{11a-c} as well as the potential to inactivate human DNA repair.^{11d} The broad variety can be illustrated, for example by the activity against chronic obstructive pulmonary disease,^{11e} applicability against herpes simplex,^{11f} and other viral



Figure 1 X-ray crystal structure of pyrimidine $7^{3,4}$



i: benzamidine hydrochloride, NaHCO_3, MeCN, 79% ii: K_2CO_3, THF, 43% iii: i, 23%

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Scheme 2



Scheme 3

diseases.^{11g} Furthermore, one field of application of 5-nitropyrimidines uses their positive modulating effect of the GABAB receptor.^{11h,i}

During the course of our synthetic efforts, we obtained interesting derivatives of some of the aforementioned persubstituted pyrimidines **6–10**. Thus, in analogy to a recently found reaction,¹² the cyanopyrimidine **13** was obtained in 85% yield by applying sodium azide at 0 °C, whereas the same reagent afforded the tetrazole **14** (74%) at room temperature. Alternatively, **14** is obtained from **10** using an excess of sodium azide at room temperature (Scheme 3).

Moreover, the phenyl group in **6** was selectively nitrated at the *meta*-position using fuming nitric acid at 0 °C, to give **15** (Scheme 4).

Furthermore, treatment of **6** with tin(II) chloride in acetone converted the nitro into an amino group, but also resulted in an unexpected reduction of the dichloromethyl to a methyl group. Thus, the methylated aminopyrimidine **16** was isolated in 75% yield (Scheme 4), whereupon the assumed intermediate, i.e. the 5-amino-6-(dichloromethyl) derivative, was not observed. To the best of our knowledge, the reduction of a dichloromethyl group attached to a carbo- or hetero-cyclic ring to a methyl group with tin(II) chloride is unprecedented. Only a few reactions, somewhat similar to our conversion, have been previously published.

Among these, the complete reduction of a dichloroacetyl substitutent by zinc in acetic acid,^{13a} the reductive elimination of two chloro atoms during an indium(I) bromide mediated coupling of α,α -dichloro ketones to 1-arylbutane-1,4-diones,^{13b} and the conversion of a dichloroacetyl into an acetyl group upon the influence of tributyltin hydride,^{13c} are worthy of note.

Interestingly, such 5-aminopyrimidine derivatives exhibit pharmacological activities, e.g. against anoxia.¹⁴

In the course of our own investigations, the exhaustive acetylation of the amino group in **16** afforded **17** as expected (76% yield), but in the subsequent reaction with hydrazine hydrate no triazole was obtained, but one of the N-acetyl groups was removed reductively. The second acetyl substituent, in principle remained unaffected, but tautomerized to a mixture of the Z- and E-isomers (3:1) of acetimidic acid **18**, as evidenced by H NMR spectroscopy. Obviously, the exclusive presence of these enol tautomers is caused by the stabilizing conjugation of the enol double bond with the pyrimidine ring (Scheme 4).

In conclusion, we have found a novel synthesis of persubstituted 5-nitropyrimidines starting from the easily accessible 1,1-bis(1*H*-benzotriazol-1-yl)-3,4,4-trichloro-2-nitrobuta-1,3-diene (1) and an appropriate amidine. Furthermore, we investigated some subsequent conversions. Especially from a chemical point of view, the latter syntheses gave some remarkable results, but in addition, some products should offer interesting biological, in particular pharmacological properties. Therefore, corresponding tests of these promising compounds are presently underway.

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Melting points were determined with a Büchi apparatus 520 and are uncorrected. NMR spectra were referenced to the residual solvent peak: $CDCl_3$, $\delta = 7.26$ (¹H), $\delta = 77.0$ (¹³C); $DMSO-d_6$, $\delta = 2.50$



Scheme 4

Synthesis 2008, No. 2, 304-310 © Thieme Stuttgart · New York

(¹H), 39.7 (¹³C); acetone- d_6 , $\delta = 2.09$ (¹H), 29.8 (¹³C). IR spectral data were obtained on a Bruker Vector 22 FT-IR. Mass spectra were obtained on a Hewlett Packard MS 5989B spectrometer, usually in direct mode with EI (70 eV). In the case of chlorinated compounds, all peak values of molecular ions as well as fragments m/z refer to the isotope ³⁵Cl. HRMS-ESI were measured with a Bruker APEX IV 7 Tesla FT ion cyclotron resonance mass spectrometer. TLC was done on Merck TLC plates (aluminum backed) silica gel 60 F 254. Column chromatography was carried out on silica gel 60 (Merck).

Benzotriazole derivative 1 was prepared in 72% yield from pentachloro-2-nitrobuta-1,3-diene in Et₂O according to Kaberdin et al.¹⁵ Pentachloro-2-nitrobuta-1,3-diene was prepared in 53% yield (bp 69–71 °C).¹⁶

1,1-Bis(1*H*-benzotriazol-1-yl)-3,4,4-trichloro-2-nitrobuta-1,3-diene (1)

Mp 138-139 °C.

IR (KBr): 3099, 3080, 1798, 1650, 1574, 1536 (NO₂), 1459, 1426, 1317 (NO₂), 1295, 1209, 1173, 1019, 902, 862, 824, 792, 742, 652, 635, 615, 563, 432 cm⁻¹.

¹H NMR (CDCl₃): δ = 8.13-8.26 (m, 2 H), 7.37–7.55 (m, 4 H), 6.71–6.83 (m, 1 H), 6.50–6.57 (m, 1 H).

¹³C NMR (CDCl₃): δ = 146.2 (C_{q,Bzl}), 146.1 (C_{q,Bzl}), 134.1 (CNO₂), 132.4 (C1), 131.4 (C_{q,Bzl}), 131.3 (C_{q,Bzl}), 130.9 (CH), 130.8 (CH), 130.7 (C3), 126.5 (CH), 126.4 (CH), 121.6 (CH), 121.5 (CH), 119.6 (C4), 109.6 (CH), 109.3 (CH).

MS: m/z (%) = 435 [M⁺] (1), 316 [M⁺ – benzotriazole] (2), 198 [M⁺ – 2 benzotriazole] (4), 51 (100).

1-(1*H*-Benzotriazol-1-yl)-3,4,4-trichloro-1-(morpholin-4-yl)-2nitrobuta-1,3-diene (2); Typical Procedure

To a suspension of bisbenzotriazole derivative **1** (5.00 g, 11.45 mmol) in MeOH (70 mL) at 5 °C was added a soln of morpholine (1.13 g, 13.0 mmol) in MeOH (5 mL) over 3 min. The resulting mixture was stirred at 10–15 °C for 2 h and at r.t. for 3 h. The precipitate was filtered off, washed with H₂O (2 × 50 mL) and cold MeOH (20 mL), and dried in vacuo to give **2**; yield: 4.07 g (88%); mp 164–166 °C (MeOH–acetone, 2:1) (Lit.¹⁷ mp 163–164 °C).

IR (KBr): 2986, 1599, 1558 (NO₂), 1467, 1345, 1283 (NO₂), 1230, 1110, 1012, 903, 875, 808, 768, 745, 713, 650, 573, 503 cm⁻¹.

¹H NMR (CDCl₃): δ = 8.16 (d, *J* = 8.3 Hz, 1 H), 7.44–7.83 (m, 3 H), 3.84–4.19 (m, 4 H, OCH₂), 3.17–3.62 (m, 4 H, NCH₂).

¹H NMR (acetone- d_6): δ = 8.24 (d, J = 8.3 Hz, 1 H), 7.53–8.02 (m, 3 H), 3.88–4.17 (m, 4 H, OCH₂), 3.31–3.80 (m, 4 H, NCH₂).

¹³C NMR (CDCl₃): δ = 147.6, 146.3, 132.4, 130.7 (CH), 126.1 (CH), 126.3 and 124.3 (CCl=CCl₂), 120.7 (CNO₂), 121.1 (CH), 110.3 (CH), 66.4 (2 C, OCH₂), 50.4 (2 C, NCH₂).

¹³C NMR (acetone- d_6): δ = 148.8, 147.2, 133.5, 131.2 (CH), 126.7 (CH), 125.9 and 125.2 (CCl=CCl₂), 124.3 (CNO₂), 121.4 (CH), 111.6 (CH), 67.0 (2 C, OCH₂), 51.1 (2 C, NCH₂).

MS: m/z (%) = 403 [M⁺] (1), 368 [M⁺ – Cl] (1), 93 (100).

1-(1*H*-Benzotriazol-1-yl)-3,4,4-trichloro-2-nitro-1-(piperidin-1-yl)buta-1,3-diene (3)

Following the typical procedure for **2** using piperidine and recrystallization (MeOH–acetone, 2:1); yield: 95%; mp 133–134 °C (Lit.³ mp 132–133 °C).

IR (KBr): 2946, 1600, 1560 (NO₂), 1498, 1453, 1285 (NO₂), 1251, 1160, 1025, 931, 918, 895, 803, 765, 751, 708, 665, 573, 461 cm⁻¹.

¹H NMR (CDCl₃): δ = 8.11 (d, *J* = 8.3 Hz, 1 H), 7.40–7.74 (m, 3 H), 3.00–3.67 (m, 4 H, NCH₂), 1.59–2.02 (m, 6 H, CH₂).

¹³C NMR (CDCl₃): δ = 148.2, 146.2, 132.4, 130.3 (CH), 125.8

(CH), 125.0 and 124.5 (CCl=CCl₂), 120.8 (CH), 115.0 (CNO₂), 110.4 (CH), 51.8 (2 C, NCH₂), 25.8 (2 C, NCH₂CH₂), 23.4 (CH₂CCH₂).

MS: m/z (%) = 401 [M⁺] (1), 93 (100).

1-(1*H*-Benzotriazol-1-yl)-3,4,4-trichloro-1-[4-(4-fluorophenyl)piperazin-1-yl]-2-nitrobuta-1,3-diene (4)

Following the typical procedure for **2** using 4-(4-fluorophenyl)piperazine and recrystallization (MeOH–acetone, 2:1); yield: 95%; mp 177–179 °C.

IR (KBr): 3050, 1601, 1562 (NO₂), 1504, 1452, 1301 (NO₂), 1005, 904, 812, 773, 704, 542 cm⁻¹.

¹H NMR (CDCl₃): δ = 8.17 (d, *J* = 8.1 Hz, 1 H), 7.40–7.75 (m, 3 H), 6.87–7.05 (m, 4 H), 3.26–3.65 (m, 8 H, NCH₂).

¹³C NMR (CDCl₃): δ = 158.1 (CF, ¹*J*_{C-F} = 244 Hz), 147.7, 146.7, 146.4, 132.4, 130.7 (CH), 126.3 (CCl), 126.0 (CH), 124.5 (CCl₂), 121.1 (CH), 119.3 (2 C, CH, ³*J*_{C-F} = 8 Hz), 116.0 (2 C, CH, ²*J*_{C-F} = 22 Hz), 111.8 (CNO₂), 110.3 (CH), 50.7 (2 C, NCH₂), 50.5 (2 C, NCH₂).

MS: m/z (%) = 496 [M⁺] (3), 479 [M⁺ – OH] (1), 123 (100).

HRMS (ESI): $m/z \, [M + H]^+$ calcd for $C_{20}H_{17}Cl_3FN_6O_2$: 497.04571; found: 497.04565.

1-(1*H*-Benzotriazol-1-yl)-3,4,4-trichloro-1-(4-chlorophenylamino)-2-nitrobuta-1,3-diene (5)

Following the typical procedure for **2** using 4-chloroaniline and recrystallization (MeOH–acetone, 2:1); yield: 57%; mp 112–113 °C.

IR (KBr): 3263, 1620, 1578 (NO₂), 1488, 1461, 1375 (NO₂), 1342, 1188, 1045, 874, 830, 812, 744, 710, 636, 587 cm⁻¹.

¹H NMR (CDCl₃): δ = 11.55 (br s, 1 H, NH), 8.05 (d, *J* = 8.4 Hz, 1 H, Bzt-H), 7.48 (dd, *J* = 7.8, 7.6 Hz, 1 H, Bzt-H), 7.40 (dd, *J* = 8.4, 7.6 Hz, 1 H, Bzt-H), 7.36 (d, *J* = 7.8 Hz, 1 H, Bzt-H), 7.04 (d, *J* = 8.6 Hz, 2 H, CH=CCl), 6.72 (d, *J* = 8.6 Hz, 2 H, CH=CNH).

¹³C NMR (CDCl₃): δ = 146.0, 145.3, 133.6, 133.3, 131.8, 129.9 (CH), 129.8 (2 C, CHCCl), 125.6 (CH), 124.1 (2 C, CHCNH), 120.9 (CH), 120.2 (CNO₂), 109.6 (CH). The carbon atoms within the trichlorovinyl group were not detected.

MS: m/z (%) = 443 [M⁺] (1), 368 [M⁺ – Cl] (1), 93 (100).

HRMS (ESI): m/z [M + H]⁺ calcd for C₁₆H₁₀Cl₄N₅O₂: 443.95831; found: 443.95845.

4-(Dichloromethyl)-6-(morpholin-4-yl)-5-nitro-2-phenylpyrimidine (6); Typical Procedures

Method I (using aq NaOH, DMSO): To a suspension of **2** (1.00 g, 2.47 mmol) in DMSO (20 mL) at 0 °C was added over 3 min a mixture of benzamidine hydrochloride (1.16 g, 7.41 mmol), 30% aq NaOH (1.32 g, 9.88 mmol), and DMSO (10 mL). The combined slurry was stirred at 0–5 °C for 1 h and at r.t. for 3 h. Subsequently, the mixture was poured into a soln of concd HCl (10 mL) in cold H₂O (100 mL). After 10 min with stirring, the precipitate was filtered off, washed with H₂O (2 × 20 mL) and MeOH (2 × 10 mL), and dried in vacuo to give pyrimidine **6**; yield: 0.56 g (61%); mp 167–169 °C (MeOH–CHCl₃, 2:1).

IR (KBr): 3055, 2865, 1590, 1569 (NO₂), 1458, 1395, 1345 (NO₂), 1221, 1112, 1022, 999, 927, 865, 836, 811, 752, 699, 542 cm⁻¹.

¹H NMR (CDCl₃): δ = 8.40–8.50 (m, 2 H, Ph), 7.40–7.55 (m, 3 H, Ph), 7.11 (s, 1 H, CHCl₂), 3.78–3.86 (m, 4 H, OCH₂), 3.63–3.72 (m, 4 H, NCH₂).

 ^{13}C NMR (CDCl₃): δ = 162.9 (C2), 156.8, 154.6, 135.0 (C_{q,Ph}), 131.5 (C_{Ph}), 128.6 (2 C, C_{Ph}), 127.9 (2 C, C_{Ph}), 125.5 (CNO₂), 65.7 (2 C, OCH₂), 65.3 (CHCl₂), 46.3 (2 C, NCH₂).

MS: m/z (%) = 368 [M⁺] (30), 351 [M⁺ – OH] (17), 333 [M⁺ – Cl] (31), 322 [M⁺ – NO₂] (12), 295 (68), 105 (100).

HRMS (ESI): m/z [M + H]⁺ calcd for C₁₅H₁₅Cl₂N₄O₃: 369.05157; found: 369.05178.

Method II (with NaH in THF): To a suspension of NaH [0.071 g, 2.964 mmol; 0.118 g of 60% NaH in mineral oil washed with anhyd pentane (2 ×)] in anhyd THF (10 mL) was added benzamidine hydrochloride (0.348 g, 2.222 mmol). After cooling this slurry to -10 °C, a soln of **2** (0.300 g, 0.741 mmol) in THF (3 mL) was added dropwise. The resulting mixture was stirred at -10 °C for 1 h then at r.t. overnight. Subsequently, the mixture was poured into a soln of concd HCl (3 mL) in cold H₂O (30 mL). The mixture was stirred for 10 min and the crude product was extracted with CH₂Cl₂ (3 × 40 mL). The combined organic phases were washed with H₂O (2 × 20 mL) and dried in vacuo. Column chromatography (silica gel, petroleum ether–EtOAc, 4:1) gave pyrimidine **6**; yield: 0.199 g (73%).

4-(Dichloromethyl)-2-methyl-6-(morpholin-4-yl)-5-nitropyrimidine (7)

Following the typical procedure for **6**, Method II using **2** (0.300 g, 0.741 mmol), acetamidine hydrochloride (0.210 g, 2.221 mmol), and NaH (0.071 g, 2.964 mmol) with purification by column chromatography (petroleum ether–EtOAc, 1:1). Subsequent crystallization was induced by adding cold MeOH (5 mL) and drying the resultant precipitate in vacuo to give **7**; yield: 0.123 g (54%); mp 95–96 °C (MeOH–CHCl₃, 2: 1).

IR (KBr): 3031, 2861, 1576 (NO₂), 1510, 1405, 1339 (NO₂), 1209, 1114, 1064, 995, 874, 838, 826, 778, 741, 672, 532 cm⁻¹.

¹H NMR (CDCl₃): δ = 6.98 (s, 1 H, CHCl₂), 3.72–3.78 (m, 4 H, OCH₂), 3.54–3.60 (m, 4 H, NCH₂), 2.62 (s, 3 H, Me).

¹³C NMR (CDCl₃): δ = 168.7 (C2), 156.7, 154.5, 126.0 (CNO₂), 66.2 (2 C, OCH₂), 65.2 (CHCl₂), 46.6 (2 C, NCH₂), 26.1 (Me).

MS: m/z (%) = 306 [M⁺] (32), 271 [M⁺ - Cl] (25), 233 (70), 58 (100).

HRMS (ESI): m/z [M + H]⁺ calcd for C₁₀H₁₃Cl₂N₄O₃: 307.03592; found: 307.03593.

For comparison (see Table 1): NaH in DMSO or LiH in THF gave 7 in 56% and 22% yield, respectively. In addition, using Cs_2CO_3 in DMSO gave 7 in only 21% yield.

4-(Dichloromethyl)-5-nitro-2-phenyl-6-(piperidin-1-yl)pyrimidine (8)

Following the typical procedure for **6**, Method II gave **8** in 76% yield, whereas using Method I gave only 40% yield; mp 128–129 $^{\circ}$ C (MeOH–CHCl₃, 2:1).

IR (KBr): 3061, 2934, 2852, 1589, 1570 (NO₂), 1528, 1395, 1342 (NO₂), 1215, 1060, 1028, 921, 832, 771, 746, 705, 663 cm⁻¹.

¹H NMR (DMSO-*d*₆): δ = 8.34–8.42 (m, 2 H, Ph), 7.53–7.62 (m, 3 H, Ph), 7.50 (s, 1 H), 3.52–3.62 (m, 4 H, NCH₂), 1.59–1.71 (m, 6 H, CH₂).

¹³C NMR (DMSO- d_6): δ = 162.2 (C2), 156.9, 154.9, 135.7 (C_{q,Ph}), 132.3 (C_{Ph}), 129.0 (2 C, C_{Ph}), 128.8 (2 C, C_{Ph}), 125.6 (CNO₂), 67.6 (CHCl₂), 47.8 (2 C, NCH₂), 25.2 (2 C, NCH₂CH₂), 23.5 (CH₂CCH₂).

MS: m/z (%) = 366 [M⁺] (4), 349 [M⁺ – OH] (7), 294 (14), 105 (100).

HRMS (ESI): m/z [M + H]⁺ calcd for C₁₆H₁₇Cl₂N₄O₂: 367.07231; found: 367.07242.

4-(Dichloromethyl)-6-[4-(4-fluorophenyl)piperazin-1-yl]-2-methyl-5-nitropyrimidine (9)

Following the typical procedure for 6, Method II gave 9 in 35%

yield, whereas using Method I gave 32% yield; mp 97–98 °C (MeOH–CHCl₃, 2:1).

IR (KBr): 3068, 2820, 1568 (NO₂), 1510, 1460, 1333 (NO₂), 1219, 995, 922, 770, 748, 703, 673, 538 cm⁻¹.

 ^1H NMR (CDCl_3): δ = 6.87–7.05 (m, 4 H, Ph), 6.99 (s, 1 H, CHCl_2), 3.70–3.81 (m, 4 H, NCH_2), 3.58–3.70 (m, 4 H, NCH_2), 2.64 (s, 3 H, Me).

¹³C NMR (CDCl₃): δ = 168.7 (C2), 157.9 (CF, ¹*J*_{C-F} = 241 Hz), 156.7, 154.5, 146.6, 126.2 (CNO₂), 118.5 (2 C, ³*J*_{C-F} = 8 Hz), 115.9 (²*J*_{C-F} = 22 Hz), 65.2 (CHCl₂), 50.1 (2 C, NCH₂), 46.3 (2 C, NCH₂), 26.1 (Me).

MS: m/z (%) = 399 [M⁺] (9), 382 [M⁺ – OH] (1), 364 [M⁺ – Cl (2), 353 [M⁺ – NO₂] (2), 303 [M⁺ – PhF] (4), 150 (100).

HRMS (ESI): $m/z [M + H]^+$ calcd for $C_{16}H_{17}Cl_2FN_5O_2$: 400.07378; found: 400.07373.

4-(4-Chlorophenylamino)-6-(dichloromethyl)-2-methyl-5-nitropyrimidine (10)

Following the typical procedure for **6**, Method II gave **10** in 85% yield, whereas using Method I gave 30% yield; mp 145–146 °C (MeOH–CHCl₃, 2:1).

IR (KBr): 3320, 1613, 1568 (NO₂), 1511, 1407, 1382 (NO₂), 1284, 1204, 1097, 1001, 856, 833, 785, 748, 710, 649, 510 cm⁻¹.

¹H NMR (CDCl₃): δ = 9.73 (br s, 1 H, NH), 7.54 (d, *J* = 8.8 Hz, 2 H, CHCCl), 7.38 (d, *J* = 8.8 Hz, 2 H, CHCNH), 7.37 (s, 1 H, CHCl₂), 2.68 (s, 3 H, Me).

¹³C NMR (CDCl₃): δ = 171.3 (C2), 160.6 (C6), 153.1 (C4), 134.9 (HNC_{q,Ph}), 131.4 (ClC_{q,Ph}), 129.2 (2 C, CHCCl), 128.7 (CNO₂), 124.3 (2 C, CHCNH), 66.5 (CHCl₂), 26.5 (Me).

MS: m/z (%) = 346 [M⁺] (100), 329 [M⁺ – OH] (4), 311 [M⁺ – Cl] (2), 300 [M⁺ – NO₂] (8).

HRMS (ESI): m/z [M + H]⁺ calcd for $C_{12}H_{10}Cl_3N_4O_2$: 346.98639; found: 346.98646.

$N^1\mathchar`-[1-(1H-Benzotriazol-1-yl)-3,4,4-trichloro-2-nitrobuta-1,3-dienyl] benzamidine (11)$

To a suspension of **1** (5.00 g, 11.45 mmol) in MeCN (60 mL) was added benzamidine hydrochloride (2.07 g, 13.2 mmol) and NaHCO₃ (2.12 g, 25.2 mmol) suspended in MeCN (20 mL). The mixture was stirred at r.t. for 2 weeks. Subsequently, the precipitate was filtered off, washed with H₂O (2×50 mL), and dried in vacuo to give **11**; yield: 3.94 g (79%); mp 131–133 °C (MeOH–acetone, 3:1).

IR (KBr): 3154, 1678, 1558 (NO₂), 1449, 1433, 1375 (NO₂), 1290, 1255, 1219, 1085, 1020, 940, 854, 785, 711, 644, 612, 551 cm⁻¹.

¹H NMR (CDCl₃): δ = 8.02 (d, *J* = 8.0 Hz, 1 H, Bzt-H), 7.58–7.68 (m, 3 H), 7.50–7.55 (m, 1 H), 7.40–7.48 (m, 2 H), 7.31–7.39 (m, 2 H), 6.55 (br s, 2 H, NH).

¹³C NMR (CDCl₃): δ = 165.1 (C=NH), 147.9 (NHCN), 145.6 (C_{q,Bzl}), 132.9 (CH, C_{Ph}), 132.1 (C_{q,Bzl}), 131.3 (C_{q,Ph}), 129.5 (CH, Bzt), 129.2 (CCl), 129.0 (2 CH, C_{Ph}), 127.4 (2 CH, C_{Ph}), 126.9 (CCl₂), 125.4 (CH, Bzt), 123.4 (CNO₂), 120.5 (CH, Bzt), 111.7 (CH, Bzt).

 $\text{MS:} \ m/z \ (\%) = 436 \ [\text{M}^+] \ (2), \ 401 \ [\text{M}^+ - \text{Cl}] \ (2), \ 78 \ (100).$

HRMS (ESI): m/z [M + H]⁺ calcd for C₁₇H₁₂Cl₃N₆O₂: 437.00818; found: 437.00842.

4-(1*H*-Benzotriazol-1-yl)-6-(dichloromethyl)-5-nitro-2-phenylpyrimidine (12)

Method I (from 1): To a suspension of 1 (5.00 g, 11.45 mmol) in THF (60 mL) was added over 3 min a suspension of benzamidine hydrochloride (2.07 g, 13.2 mmol) and K_2CO_3 (3.48 g, 25.2 mmol)

in THF (20 mL) at r.t. The mixture was stirred for 2 d and the solvent was removed in vacuo. The residue then was treated with dil aq 5% HCl (200 mL). Finally, the crude product was extracted with CH₂Cl₂ (3×50 mL), washed with H₂O (2×100 mL), and dried over (anhyd CaCl₂). The solvent was removed under reduced pressure and the crude product purified by column chromatography (petroleum ether–EtOAc, 10:1) to give **12**; yield: 1.06 g (23%).

Method II (from **11**): At r.t., K_2CO_3 (0.35 g, 2.51 mmol) was added to a suspension of **11** (1.00 g, 2.28 mmol) in THF (20 mL); the mixture was stirred for 1 d. Workup and purification as described for Method I gave **12**; yield: 0.39 g (43%); mp 179–181 °C (MeOH-acetone, 3:1).

IR (KBr): 3026, 1597, 1556 (NO₂), 1487, 1402, 1356 (NO₂), 1212, 1088, 1005, 855, 812, 750, 698, 629 cm⁻¹.

¹H NMR (CDCl₃): δ = 8.54–8.63 (m, 2 H), 8.49 (d, *J* = 8.3 Hz, 1 H), 8.20 (d, *J* = 8.3 Hz, 1 H), 7.76–7.87 (m, 1 H), 7.54–7.69 (m, 4 H), 7.07 (s, 1 H, CHCl₂).

 13 C NMR (CDCl₃): δ = 165.5 (C2), 158.2, 148.5, 146.2 (C_{q,Bzt}), 134.5 (C_{q,Bzt}), 133.4 (CH, C_{Ph}), 131.1 (C_{q,Ph}), 130.7 (CH, Bzt), 130.3 (CNO₂), 129.5 (2 CH, C_{Ph}), 129.2 (2 CH, C_{Ph}), 126.5 (CH, Bzt), 121.0 (CH, Bzt), 114.0 (CH, Bzt), 64.9 (CHCl₂).

MS: m/z (%) = 400 [M⁺] (15), 372 (10), 279 (28), 92 (100).

HRMS (ESI): m/z [M + H]⁺ calcd for C₁₇H₁₁Cl₂N₆O₂: 401.03151; found: 401.03148.

4-(4-Chlorophenylamino)-6-cyano-2-methyl-5-nitropyrimidine (13); Typical Procedure

To a soln of **10** (0.50 g, 1.44 mmol) in DMF (10 mL) at 0 °C was added NaN₃ (0.28 g, 4.31 mmol). The resulting mixture was stirred at 0 °C for 1 h and at r.t. for 30 min. Then ice cold H₂O (70 mL) was added and the soln was adjusted to pH 7 with 10% HCl soln. The resulting precipitate was filtered with suction, washed with H₂O (2 × 20 mL), dried in vacuo, and recrystallized (hexane–EtOAc, 1:1) to afford **13**; yield: 0.35 g (85%); mp 160–162 °C.

IR (KBr): 3319, 2127 (CN), 1609, 1576 (NO₂), 1520, 1494, 1408, 1273, 1215 (NO₂), 1090, 998, 857, 839, 788, 686, 510, 439 cm⁻¹.

¹H NMR (CDCl₃): δ = 10.11 (br s, 1 H, NH), 7.53 (d, *J* = 8.8 Hz, 2 H, CHCCl), 7.41 (d, *J* = 8.8 Hz, 2 H, CHCNH), 2.65 (s, 3 H, Me).

¹³C NMR (CDCl₃): δ = 172.2 (C2), 153.1 (C4), 138.6 (C6), 134.0 (HNC_{q,Ph}), 132.1 (ClC_{q,Ph}), 129.3 (2 C, CHCCl), 126.5 (CNO₂), 124.6 (2 C, CHCNH), 113.8 (CN), 26.3 (Me).

MS: m/z (%) = 289 [M⁺] (100), 272 [M⁺ – OH] (9), 254 [M⁺ – Cl] (8), 243 [M⁺ – NO₂] (12).

HRMS (ESI): m/z [M + H]⁺ calcd for C₁₂H₉ClN₅O₂: 290.04393; found: 290.04391.

4-(4-Chlorophenylamino)-2-methyl-5-nitro-6-(1*H*-tetrazol-5yl)pyrimidine (14)

Following the typical procedure for 13 using 13 at r.t. for 10 h; yield: 74%; mp 155–157 °C (MeOH–acetone, 1:1).

Following the typical procedure for 13 using 10 and NaN₃ (5 equiv) at r.t. for 10 h; yield: 65%.

IR (KBr): 3365, 2186, 2126, 1609, 1567 (NO₂), 1491, 1343, 1211 (NO₂), 1092, 1012, 831, 504 $\rm cm^{-1}.$

¹H NMR (DMSO- d_6): $\delta = 9.50$ (br s, 1 H, NH-C₆H₄-Cl), 7.66 (d, J = 8.8 Hz, 2 H, CHCCl), 7.41 (d, J = 8.8 Hz, 2 H, CHCNH), 2.50 (s, 3 H, Me).

¹³C NMR (DMSO- d_6): $\delta = 167.4$ (C2), 156.2, 151.7, 148.5, 137.2 (HNC_{q,Ph}), 134.4 (CNO₂), 128.5 (2 C, CHCCl), 128.4 (ClC_{q,Ph}), 125.0 (2 C, CHCNH), 26.0 (Me).

$$\begin{split} \text{MS:} \ m/z\,(\%) &= 332\,[\text{M}^+]\,(6), 289\,(12), 286\,[\text{M}^+-\text{NO}_2]\,(4), 263\,[\text{M}^+-\text{CHN}_4]\,(10), 75\,(100). \end{split}$$

HRMS (ESI): m/z [M + H]⁺ calcd for $C_{12}H_{10}ClN_8O_2$: 333.06098; found: 333.06094.

6-(Dichloromethyl)-4-morpholino-5-nitro-2-(3-nitrophenyl)pyrimidine (15)

The nitration of pyrimidine **6** was accomplished by adding HNO₃ (5 mL) to **6** (0.20 g, 0.54 mmol) at 0 °C over 5 min. The mixture was stirred for 20 min at this temperature and an additional 10 min at 10 °C. Ice cold H₂O was added (50 mL) and the precipitate that formed was washed successively with H₂O (3×20 mL) and petroleum ether (2×10 mL). The crude product was dried and recrystallized (MeOH) to give dinitro compound **15**; yield: 0.16 g (73%); mp 162–163 °C.

IR (KBr): 2914, 2863, 1593, 1573, 1530 (NO₂), 1347 (NO₂), 1226, 1116, 1001, 951, 873, 837, 772, 737, 708, 659, 542 cm⁻¹.

¹H NMR (CDCl₃): δ = 9.19 (dd, *J* = 2.1, 1.6 Hz, 1 H), 8.82 (ddd, *J* = 7.9, 1.6, 1.2 Hz, 1 H), 8.39 (ddd, *J* = 8.2, 2.1, 1.2 Hz, 1 H), 7.70 (dd, *J* = 8.2, 7.9 Hz, 1 H), 7.08 (s, 1 H, CHCl₂), 3.80–3.87 (m, 4 H, OCH₂), 3.68–3.75 (m, 4 H, NCH₂).

¹³C NMR (CDCl₃): δ = 161.3 (C2), 157.6, 155.1, 148.6 (C_{q,Ph}NO₂), 137.4 (C_{q,Ph}), 134.9 (CH), 129.7 (CH), 126.6 (CNO₂), 126.4 (CH), 123.8 (CH), 66.2 (2 C, OCH₂), 65.5 (CHCl₂), 47.0 (2 C, NCH₂).

$$\begin{split} \text{MS:} \ m/z \ (\%) &= 413 \ [\text{M}^+] \ (11), \ 396 \ [\text{M}^+ - \text{OH}] \ (14), \ 378 \ [\text{M}^+ - \text{Cl}] \\ (10), \ 322 \ [\text{M}^+ - \text{NO}_2] \ (8), \ 234 \ (100). \end{split}$$

HRMS (ESI): m/z [M + H]⁺ calcd for C₁₅H₁₄Cl₂N₅O₅: 414.03665; found: 414.03678.

5-Amino-4-methyl-6-(morpholin-4-yl)-2-phenylpyrimidine (16) To a soln of **6** (1.00 g, 2.70 mmol) in acetone (50 mL) were added SnCl₂·2 H₂O (3.66 g, 16.20 mmol) and concd HCl (10 mL). The mixture was stirred at reflux for 6 h. The acetone was removed and then ice cold H₂O (200 mL) was added and the mixture was adjusted to pH 8 with 30% aq NaOH soln. The resulting precipitate was filtered off with suction, washed with H₂O (3 × 50 mL) and petroleum ether–Et₂O (5:1, 2 × 20 mL), and the crude product was recrystallized (hexane–EtOAc, 1:1) to give **16**; yield: 0.55 g (75%); mp 128– 129 °C.

IR (KBr): 3426, 3327 (NH₂), 2957, 2859, 1618, 1583, 1558, 1430, 1287, 1243, 1116, 974, 864, 773, 708, 670, 551, 528 cm⁻¹.

¹H NMR (CDCl₃): δ = 8.29–8.37 (m, 2 H, Ph), 7.35–7.47 (m, 3 H, Ph), 3.85–3.92 (m, 4 H, OCH₂), 3.52 (br s, 2 H, NH₂), 3.34–3.40 (m, 4 H, NCH₂), 2.45 (s, 3 H, Me).

 ^{13}C NMR (CDCl₃): δ = 154.9, 154.1, 149.5, 138.4 (C_{q,Ph}), 128.9 (CH, Ph), 128.2 (2 C, C_{Ph}), 127.7 (CNH₂), 127.1 (2 C, C_{Ph}), 66.9 (2 C, OCH₂), 48.1 (2 C, NCH₂), 19.8 (CH₃).

 $\text{MS:}\ m/z\ (\%) = 270\ [\text{M}^+]\ (100),\ 225\ (48),\ 213\ (82).$

HRMS (ESI): m/z [M + H]⁺ calcd for C₁₅H₁₉N₄O: 271.15534; found: 271.15533.

5-(Diacetylamino)-4-methyl-6-(morpholin-4-yl)-2-phenylpyrimidine (17)

To Ac₂O (50 mL) was added **16** (0.55 g, 2.03 mmol). The resulting mixture then was stirred at 105–110 °C for 3 h. The mixture was cooled to r.t. and ice cold H₂O (200 mL) was added. The resulting precipitate was filtered off with suction and treated with H₂O (3×50 mL) and petroleum ether (2×20 mL) and dried in vacuo to give **17**; yield: 0.55 g (76%); mp 149–150 °C (MeOH–CHCl₃, 2:1).

IR (KBr): 2964, 2865, 1716 (C=O), 1550, 1434, 1367, 1239, 1207, 1116, 1090, 983, 855, 776, 716, 586, 539 cm⁻¹.

¹H NMR (CDCl₃): δ = 8.36–8.44 (m, 2 H, Ph), 7.43–7.49 (m, 3 H, Ph), 3.74–3.80 (m, 4 H, OCH₂), 3.52–3.58 (m, 4 H, NCH₂), 2.32 (s, 6 H, 2 Ac), 2.31 (s, 3 H, Me).

¹³C NMR (CDCl₃): δ = 173.0 (C=O), 165.5, 162.4, 160.8, 137.2 (C_{q,Ph}), 130.7 (CH, Ph), 128.4 (2 C, C_{Ph}), 128.3 (2 C, C_{Ph}), 118.3 (C5), 66.7 (2 C, OCH₂), 47.7 (2 C, NCH₂), 25.8 (2 C, CH₃), 20.6 (CH₃).

MS: m/z (%) = 354 [M⁺] (92), 311 [M⁺ – Ac] (68), 268 [M⁺ – morpholine] (73), 254 [M⁺ – NAc₂] (39), 87 (100).

HRMS (ESI): m/z [M + H]⁺ calcd for $C_{19}H_{23}N_4O_3$: 355.17647; found: 355.17667.

N-[4-Methyl-6-(morpholin-4-yl)-2-phenylpyrimidin-5-yl]acetimidic Acid (18)

To a suspension of **17** (0.20 g, 0.56 mmol) in MeOH (10 mL) was added 64% hydrazine hydrate (0.084 g, 1.68 mmol) at 20 °C. The resulting mixture was stirred at 40–45 °C for 6 h and then cooled to r.t. Subsequently, the solvent was removed under reduced pressure. The mixture then was treated with ice cold H₂O (50 mL) to obtain a precipitate that was filtered off with suction and washed with H₂O (2 × 50 mL) then dried in vacuo to give **18**; yield: 0.16 g (93%); mixture of *E*- and *Z*-isomers, ratio 3:1 (¹H NMR); mp 160–161 °C (MeOH–CHCl₃, 1:1).

IR (KBr): 3226, 2847, 1651, 1560, 1438, 1371, 1289, 1224, 1118, 1021, 986, 835, 760, 701, 588, 549 $\rm cm^{-1}.$

¹H NMR (CDCl₃): δ = 8.32–8.42 (m, 2.72 H, Ph), 7.40–7.51 (m, 4.08 H, Ph), 6.98 (br s, 0.36 H, OH), 6.70 (br s, 1 H, OH), 3.75–3.83 (m, 5.44 H, OCH₂), 3.62–3.73 (m, 1.36 H, NCH₂). 3.50–3.59 (m, 4.08 H, NCH₂). 2.45 (s, 1.08 H, Me), 2.37 (s, 3 H, Me), 2.15 (s, 3 H, Me), 1.86 (s, 1.08 H, Me).

¹³C NMR (CDCl₃): δ (major isomer) = 168.8, 165.3, 161.4, 161.0, 137.7 (C_{q,Ph}), 130.3 (1CH, Ph), 128.3 (2 CH, Ph), 128.1 (2 CH, Ph), 115.6 (C5), 66.7 (OCH₂), 47.9 (NCH₂), 23.2 (CH₃), 21.2 (CH₃); δ (minor isomer) = 172.9, 165.8, 161.6, 160.9, 137.3 (C_{q,Ph}), 130.6 (1CH, Ph), 128.4 (2 C, C_{Ph}), 128.2 (2 C, C_{Ph}), 115.4 (C5), 66.8 (2 C, OCH₂), 47.5 (2 C, NCH₂), 21.1 (CH₃), 20.2 (CH₃).

MS: m/z (%) = 312 [M⁺] (14), 297 [M⁺ – Me] (6), 254 [M⁺ – (CH₃C(OH)=N)] (12), 226 [M⁺ – morpholine] (8), 78 (100).

HRMS (ESI): m/z [M + H]⁺ calcd for $C_{17}H_{21}N_4O_2$: 313.16590; found: 313.16587.

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- *a* = 921.5(3) pm, *b* = 1230.6(5) pm, *c* = 1369.6(6) pm, *V* = 1315.1(9) × 10⁶ pm³, *Z* = 4, *d*_{calc} = 1.551 g cm⁻³, *F*(000) = 632, absorption coefficient = 0.503 mm⁻¹ using 4346 independent reflections and 439 parameters. *R*1 = 0.0420, *wR*2 = 0.1011 [*I* > 2 σ (*I*)], goodness of fit on *F*² = 1.082, residual electronic density = 0.463 and -0.341 e A⁻³. Crystallographic data have been deposited with the Cambridge Crystallographic Data Centre as supplementary publication no. CCDC-645171. Copies of the data can be obtained free of charge on application to CCDC, 12 Union Road, Cambridge CB2 1EZ, UK (fax: +44(1223)336033, email: fileserv@ccdc.ac.uk or http://www.ccdc.cam.ac.uk).
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