

SHORT
COMMUNICATIONS

Synthesis of *N*-Acylarenesulfonamides

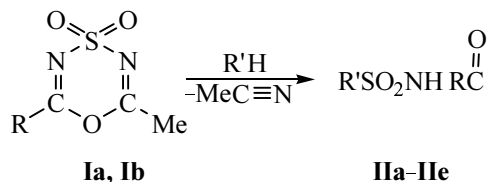
E. N. Utkina, A. A. Michurin, and A. V. Shishulina

Nizhegorodskii State Technical University, Nizhni Novgorod, 603600 Russia

Received April 4, 2003

2,6-Disubstituted 1,4,3,5-oxathiadiazine-4,4-dioxides of type **I** are known to be used for preparation of *N*-sulfonyl-substituted ureas, guanidines, amidines and other acyclic and cyclic compounds containing *N*-sulfonyl group, some of which possess high biological activity [1–3].

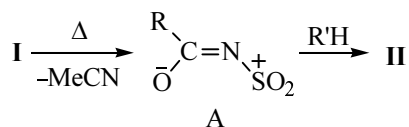
We demonstrated that dioxides **I** readily and selectively reacted with aromatic and heteroaromatic compounds affording the corresponding *N*-acylarenesulfonamides **II**.



R = CCl₃ (**Ia**, **IIa–d**), CBr₃ (**Ib**, **IIe**); R' = 2-thienyl (**IIa**), 4-MeC₆H₄ (**IIb**), 4-Me₂NC₆H₄ (**IIc**), 4-MeOC₆H₄ (**IId**, **e**).

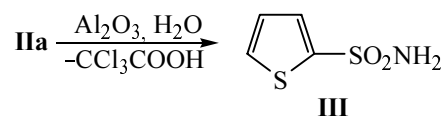
This reaction is a convenient procedure for building up an amidosulfonyl group in arene (hetarene).

It is presumable that one among possible paths of arenes (hetarenes) amidosulfonylation is a route involving generation from dioxides **I** of reactive species, *N*-sulfonylamides (A) [4] followed by their reaction with arenes (hetarenes).



The composition and structure of sulfonamides **II** were confirmed by elemental analyses, values of neutralization equivalents, and IR spectra.

The structure of sulfonamide **IIa** was also proved by its selective hydrolysis on Al₂O₃ [5] to afford 2-thiophenesulfonamide **III**.



N-Trichloroacetyl-2-thiophenesulfonamide (**IIa**).

To a solution of 0.23 g (0.66 mmol) of 6-methyl-2-trichloromethyl-1,4,3,5-oxathiadiazine-4,4-dioxide **Ia** in 9 ml of benzene was added 0.06 g (0.77 mmol) of thiophene, and the mixture was heated to 50°C for 6 h. Then the solvent was distilled off, and the residue was neutralized with Na₂CO₃ solution, 0.20 ml of concn. hydrochloric acid was added, and 0.27 g (98%) of *N*-trichloroacetyl-2-thiophenesulfonamide **IIa** was isolated, mp 154°C (from ethanol). IR spectrum, ν , cm⁻¹: 3310 (NH), 1745 (C=O), 1360, 1150 (SO₂). Found, %: N 4.43. Neutralization equivalent 306.40. C₆H₄Cl₃NO₃S₂. Calculated, %: N 4.54. Neutralization equivalent 306.86.

N-Trichloroacetyl-4-methylbenzenesulfonamide (**IIb**).

A solution of 0.15 g (0.57 mmol) of dioxide **Ia** in 5 ml of toluene was heated to 110°C for 5 h. The residue after evacuation was neutralized with Na₂CO₃ solution, 0.10 ml of concn. HCl was added, and 0.16 g (89%) of sulfonamide **IIb** was filtered off, mp 141°C (from ethanol) (publ.: mp 140–142°C [7]). IR spectrum, ν , cm⁻¹: 3350 (NH), 1755 (C=O), 1400, 1185 (SO₂). Found, %: N 4.39. Neutralization equivalent 315.80. C₉H₈Cl₃NO₃S. Calculated, %: N 4.43. Neutralization equivalent 316.50.

N-Trichloroacetyl-4-dimethylaminebenzenesulfonamide (**IIc**).

A solution of 0.20 g (0.75 mmol) of dioxide **Ia** and 0.14 g (1.13 mmol) *N,N*-dimethylaniline in 7 ml of benzene was heated to 110°C for 5 h. From the reaction mixture was filtered off 0.22 g (85%) of sulfonamide **IIc**, mp (decomp.) 230°C (from ethanol). IR spectrum, ν , cm⁻¹: 3280 (NH), 1735 (C=O), 1360, 1135 (SO₂). Found, %: N 8.02. Neutralization equivalent 348.06. C₁₀H₁₁Cl₃N₂O₃S. Calculated, %: N 8.10. Neutralization equivalent 348.67.

***N*-Trichloroacetyl-4-methoxybenzenesulfonamide (II_d)**. A solution of 0.10 g (0.38 mol) of dioxide **Ia** in 3 ml of anisole was heated to 100°C for 5 h. The residue after evacuation was neutralized with Na₂CO₃ solution, 0.10 ml of concn. HCl was added, and 0.11 g of sulfonamide **II_d** was filtered off, mp 168°C (from ethanol). IR spectrum, ν , cm⁻¹: 3390 (NH), 1785 (C=O), 1380, 1174 (SO₂). Found, %: N 4.30. Neutralization equivalent 334.31. C₉H₈Cl₃NO₄S. Calculated, %: N 4.21. Neutralization equivalent 332.50.

The reaction of dioxide **Ib** with anisole under similar conditions afforded ***N*-tribromoacetyl-4-methoxybenzenesulfonamide (II_e)** (93%), mp 153°C (from ethanol). IR spectrum, ν , cm⁻¹: 3390 (NH), 1750 (C=O), 1380 < 1150 (SO₂). Found, %: N 3.03. Neutralization equivalent 465.51. C₉H₈Br₃NO₄S. Calculated, %: N 3.01. Neutralization equivalent 465.70.

2-Thiophenesulfonamide (III). A solution of 0.20 g (0.64 mol) of sulfonamide **IIa** in 15 ml methylene chloride was maintained at stirring for 48 h at 15°C with 1.00 g of Al₂O₃. The precipitate of Al₂O₃ was filtered off and washed with acetone. The solvent was distilled off to afford 0.11 g (99%) of sulfonamide **III**, mp 142°C (from

ethanol), mp 142°C [6]. IR spectrum, ν , cm⁻¹: 3335 (NH), 1590 (C=C), 1360, 1170 (SO₂). The aluminum oxide was washed with 25% water solution of ammonia and filtered off. The water filtrate was evaporated to give 0.12 g (99%) of ammonium trichloroacetate IR spectra were recorded on spectrophotometer UR-20 from solutions in CH₂Cl₂.

REFERENCES

1. Michurin, A.A., Utkina, E.I., Zakharov, L.N., Fukin, G.K., and Shishulina, A.V., *Zh. Org. Khim.*, 2000, vol. 36, p. 1559.
2. Abramova, L.I., Titova, O.M., Michurin, A.A., and Zil'berman, E.N., *Vysokomol. Soed.*, 1994, vol. 36, p. 2084.
3. US Patent 4830663, 1987; *Ref. Zh. Khim.*, 1990, 14O423P.
4. Burgess, E.M. and Williams, W.M., *J. Org. Chem.*, 1973, vol. 38, p. 1244.
5. Utkina, E.N., Michurin, A.A., and Shishulina, A.V., *Zh. Org. Khim.*, 2000, vol. 36, p. 1263.
6. *Spravochnik khimika* (Handbook of Chemist), Moscow: Khimiya, 1964, vol. 2, p. 982.
7. Kretov, A.E. and Kremlev, M.M., *Ukr. Khim. Zh.*, 1959, vol. 25, p. 482.