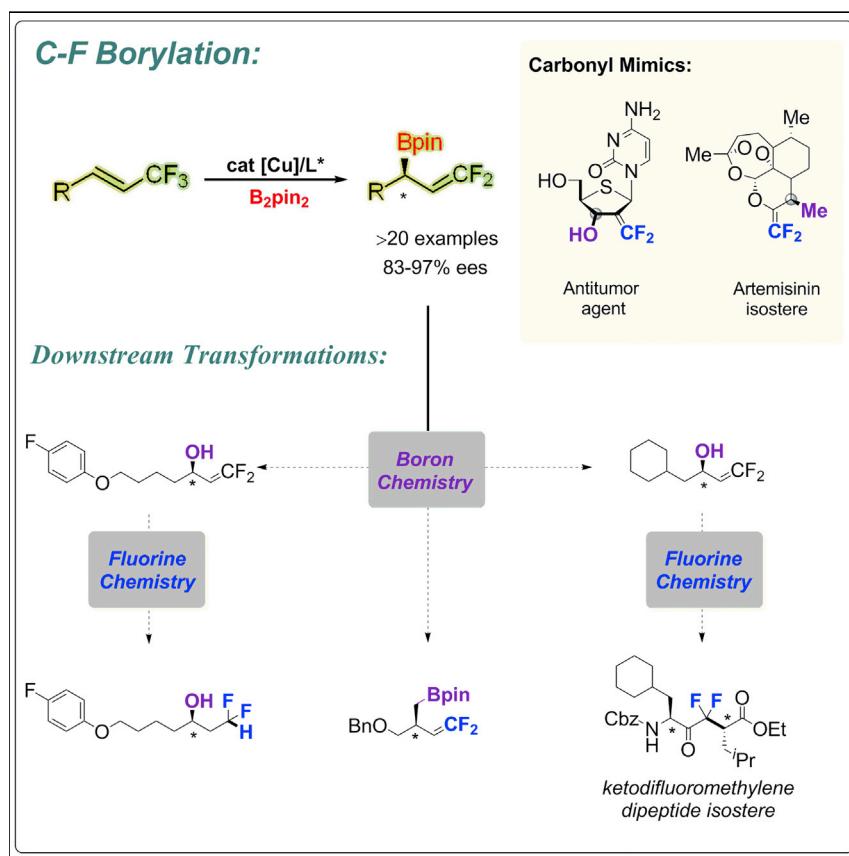


Article

Copper-Catalyzed Asymmetric Defluoroborylation of 1-(Trifluoromethyl)Alkenes



A mild catalytic system has been established for the asymmetric defluoroborylation of 1-(trifluoromethyl)alkenes via C-F activation. This route employs a low-cost copper catalyst and can be used to generate chiral gem-difluoroallylboronates with a broad substrate scope.

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HIGHLIGHTS

The first asymmetric defluoroborylation

Rapid access to optically active gem-difluoroallylboronates

Cheap copper catalysis under mild reaction conditions

Useful carbonyl mimic products based on the diversity of boron chemistry



Article

Copper-Catalyzed Asymmetric Defluoroborylation of 1-(Trifluoromethyl)Alkenes

Pan Gao,¹ Chengkai Yuan,¹ Yue Zhao,¹ and Zhuangzhi Shi^{1,2,*}

SUMMARY

gem-Difluoroalkenes have steric and electronic profiles similar to those of ketones, aldehydes, and esters, and consequently have been used widely as carbonyl isosteres in modern drug discovery. Although many attempts have been made to achieve gem-difluoroalkenes, the induction of enantioselectivity at the α position of a gem-difluorovinyl group still remains a challenge. Herein, an efficient method for the construction of gem-difluoroallylboronates with high enantiomeric excess via a copper-catalyzed defluoroborylation of 1-(trifluoromethyl)alkenes with B_2pin_2 is described. The reaction conditions were mild, and a variety of common functional groups, such as ether, fluoride, chloride, bromide, iodide, ester, cyano, sulfide, amino, and indoyl groups, were well tolerated. Furthermore, we not only applied this developed system as a powerful synthetic tool for the late-stage modification of complex compounds but also highlighted the utility of the formed compounds in synthesis.

INTRODUCTION

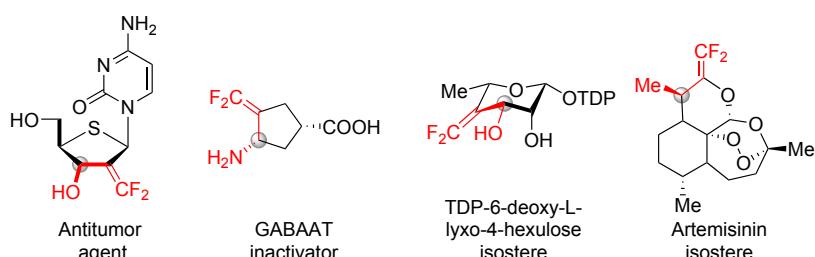
gem-Difluorovinyl groups, important structural motifs in medicinal design, exhibit steric and electronic profiles similar to those of the corresponding carbonyl substituents, and consequently have been used widely as carbonyl isosteres.¹ gem-Difluoroalkene mimics of bioactive ketones and esters have been prepared to improve pharmaceutical properties such as bioactivity, target specificity, and metabolic stability of the mimics over those of their bioactive parent structures (Figure 1A).^{2–6} Although impressive progress has been made in the synthesis of gem-difluoroalkenes in recent years, for example, the classical Wittig olefination and Julia-Kocienski reaction, some issues remain unresolved.^{7–18} First, most traditional approaches require strong bases, which leads to a limited substrate scope and low efficiency; second, the α position of a gem-difluorovinyl group in a drug candidate is usually a chiral carbon center, and the induction of enantioselectivity at this position is still challenging;¹⁹ third, the rapid and efficient installation of a range of functional groups such as hydroxyl, amino, and alkyl substituents to α position of a difluorovinyl group is in high demand. On the other hand, α -chiral boronate-substituted compounds are an important family of target molecules, because the C-B linkage provides an extremely useful stereogenic center.^{20–29} They have been shown to participate in C-C coupling reactions with excellent enantioselectivity.^{30,31} Furthermore, the formed chiral C-B bond may be readily converted into a C-O, C-N, and other C-heteroatom bond with retention of the configuration, allowing access to a wide array of functional groups that are common in valuable synthetic targets.³² Therefore, to address the aforementioned challenges, an efficient method is to construct α -chiral boronate-substituted gem-difluoroalkene building blocks.

Defluorinative functionalization of fluorine-containing compounds has shown promise as it confers synthetic versatility to inert C-F bonds.^{33–40} Defluoroborylations of

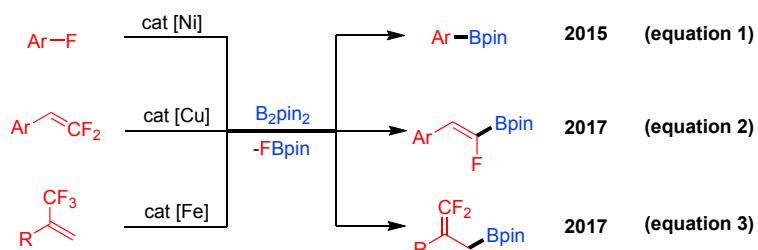
The Bigger Picture

Modern drug discovery relies on advance in chemical synthesis to address challenges from designing new pharmaceutical agents. Replacement of a carbonyl with the corresponding gem-difluoroalkene, a carbonyl isostere, has been demonstrated to provide bioactive substrates still recognized by their target. However, this potentially valuable carbonyl mimic has not been more extensively evaluated, possibly because the conventional routes to construct the gem-difluoroethylene motif involve a functional-group interconversion relying on highly reactive intermediates, organometallic reagents, or harsh reaction conditions. Our methodology enables C-F activation of 1-(trifluoromethyl)alkenes via a defluoroborylation process to produce a diverse array of enantioenriched gem-difluoroallylboronates. We anticipate that the strategy based on the diversity of boron chemistry will simplify the synthesis and structural elaboration of gem-difluoroalkene targets in chemistry, biology, and medicine.

A Biological activity of chiral gem-difluoroalkenes:



B Known processes for defluoroborylation:



C Enantioselective defluoroborylation:

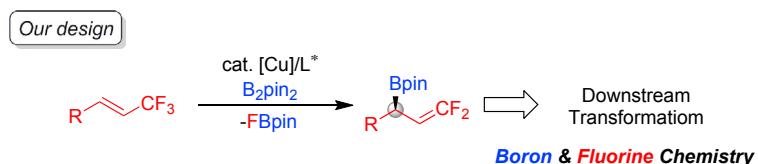


Figure 1. Development of a Protocol to Build Chiral gem-Difluoroallylboronates

unactivated fluoroarenes have been reported by several groups (Figure 1B, equation 1).^{41–44} Recent progress has been made for the construction of Z-fluoro-alkenylboronates by a Cu-catalyzed stereoselective monodefluoroborylation of gem-difluoroalkenes (Figure 1B, equation 2).^{45,46} Our group⁴⁷ and Ito's group⁴⁸ have also developed Cu-catalyzed stereoselective hydrodefluorinations of gem-difluoroalkenes to generate Z-fluoroalkenes, in which the Z-fluoro-alkenylboronates acted as the key intermediates. Notably, Fe-catalyzed syntheses of gem-difluoroallylboronates via the defluoroborylation of α -(trifluoromethyl)alkenes in the absence of ligand have also been developed (Figure 1B, equation 3).⁴⁹ Inspired by these results, here we report a mild and operationally simple strategy for the construction of α -chiral boronate-substituted gem-difluoroalkenes through a copper-catalyzed, enantioselective defluoroalkylation of 1-(trifluoromethyl)alkenes (Figures S35–S85) with B_2Pin_2 (Figure 1C). The challenge of this strategy was to cleave a stable C-F bond in a CF_3 group while simultaneously forming a readily transformable allylic C-B bond as part of a stereogenic center.

RESULTS AND DISCUSSION

Initial studies of the catalytic reaction conditions identified Cu salts in conjunction with an N-heterocyclic carbene (NHC) ligand as effective promoters of the addition of B_2Pin_2 to 1-(trifluoromethyl)alkene 1a (Table 1; see also Figures S35–S37). In the presence of 10 mol % $CuCl$, 11 mol % $ICy \cdot HCl$, 1.0 equiv NaO^tBu , and 2.0 equiv MeOH at 45°C under an argon atmosphere in CH_3CN , we indeed observed the

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Table 1. Reaction Development

The reaction scheme illustrates the asymmetric defluoroborylation of alkene **1a** (BnO-CH=CH-CF₃) to alcohol **4a** (BnO-CH(OH)-CH-CF₂). The reaction conditions involve 10 mol% [M], 11 mol% L*, 1.0–2.0 equiv base, 2.0 equiv MeOH, CH₃CN, 12 h, T, and B₂pin₂ (**2a**). Intermediate **3a** (BnO-CH(Bpin)-CH-CF₂) is formed, which is then oxidized by NaBO₃•4H₂O in THF/H₂O to yield **4a**.

The catalyst library includes:

- L1:** A chiral phosphine ligand with a bis(2-dimethylaminobenzylidene)diphenylphosphine core.
- L2:** A ferrocenyl-based phosphine ligand with two phenoxy groups.
- L3:** A ferrocenyl-based phosphine ligand with a diphenylphosphine group.
- L4:** A ferrocenyl-based phosphine ligand with two ferrocenylphosphine groups.
- L5:** A ferrocenyl-based phosphine ligand with two triphenylphosphine groups substituted with trifluoromethyl groups.
- L6:** A ferrocenyl-based phosphine ligand with two triphenylphosphine groups substituted with methoxy and methyl groups.
- L7:** A ferrocenyl-based phosphine ligand with two triphenylphosphine groups substituted with trifluoromethyl groups.

Entry	[M]	L*	Base (equiv)	Temp. (°C)	ee (%) ^a	Yield of 4a (%) ^b
1	CuCl	ICy·HCl	NaO <i>t</i> Bu (1.0)	45	—	18
2	CuI	ICy·HCl	NaO <i>t</i> Bu (1.0)	45	—	32
3	CuI	ICy·HCl	NaO <i>t</i> Bu (2.0)	45	—	84 (58) ^c
4	CuI	L1	NaO <i>t</i> Bu (2.0)	45	5	89
5	CuI	L2	NaO <i>t</i> Bu (2.0)	45	21	90
6	CuI	L3	NaO <i>t</i> Bu (2.0)	45	54	83
7	CuI	L4	NaO <i>t</i> Bu (2.0)	45	82	88
8	CuI	L5	NaO <i>t</i> Bu (2.0)	45	88	90
9	CuI	L6	NaO <i>t</i> Bu (2.0)	45	75	76
10	CuI	L7	NaO <i>t</i> Bu (2.0)	45	84	85
11	CuI	L5	NaO <i>t</i> Bu (2.0)	RT	89	91
12	CuI	L5	NaO <i>t</i> Bu (2.0)	0	92	68
13	CuI	L5	NaO <i>t</i> Bu (2.0)	0–RT	92	89 (65) ^c
14 ^d	CuI	L5	NaO <i>t</i> Bu (2.0)	0–RT	—	trace
15 ^e	CuI	L5	—	0–RT	—	trace
16	FeCl ₂	L5	NaO <i>t</i> Bu (2.0)	0–RT	—	0

Reaction conditions: **1a** (0.20 mmol), **2a** (0.40 mmol) in CH₃CN (2.0 mL), 12 hr, room temperature (RT), under Ar.

^aEnantiomeric excess values were determined by chiral high-performance liquid chromatography analysis.

^bIsolated yield after chromatography.

^cIsolated yield of **3a**.

^dWithout MeOH.

^eWithout NaO*t*Bu.

formation of defluoroborylation product **3a** and isolated the corresponding alcohol product **4a** in 18% yield after Brown oxidation by NaBO₃ (entry 1). Other copper salts such as CuI showed better yields (entry 2). The use of 2.0 equiv of NaO*t*Bu resulted in

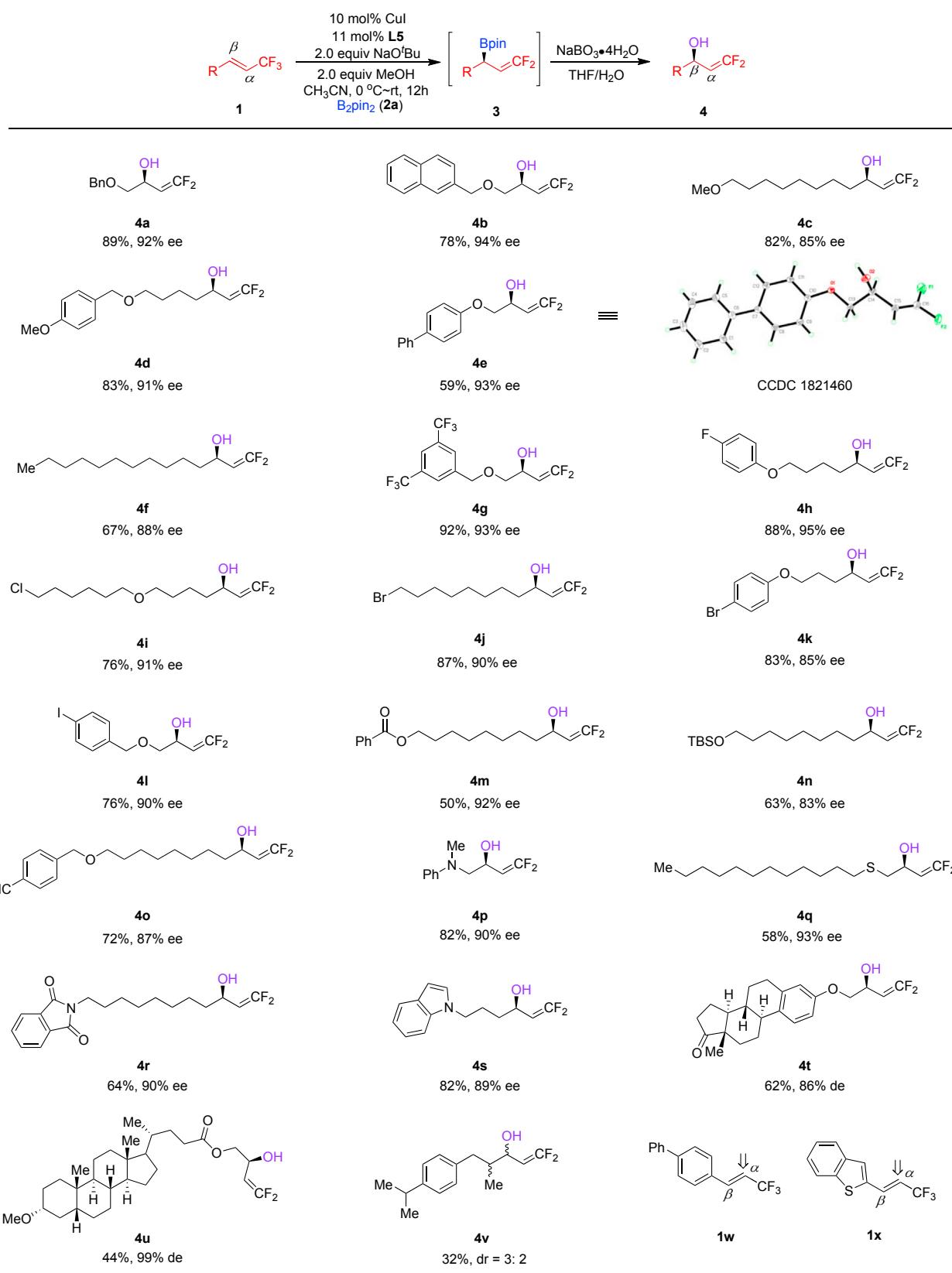
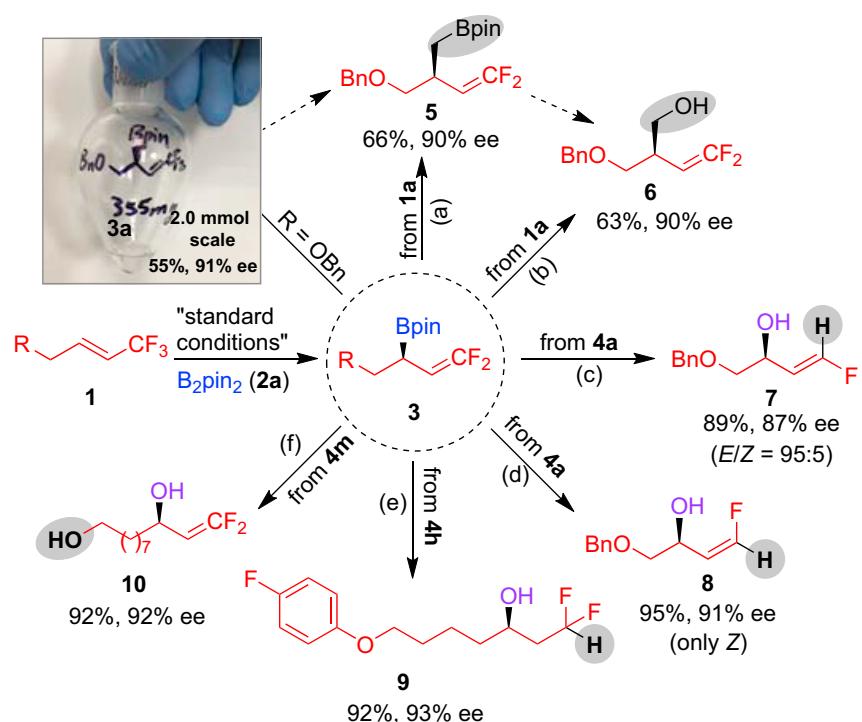


Figure 2. Substrate Scope

Reaction conditions: **1** (0.20 mmol), **2a** (0.40 mmol), CuI (10 mol % mmol), **L5** (11 mol % mmol), NaO^tBu (0.40 mmol), B₂pin₂ (0.4 mmol), MeOH (0.40 mmol) in CH₃CN (2.0 mL), 12 hr, 0°C to room temperature (RT), under Ar, isolated yield after chromatography. For **4j** and **4q**, 0°C, 12 hr. For **4v**, diastereoselectivity was determined by high-performance liquid chromatography analysis.

84% isolated yield of **4a**, despite precursor **3a** being unstable on silica gel column (only 58% isolated yield, entry 3). To achieve an enantioselective variation, we first evaluated chiral NHC ligands such as **L1**, which only afforded 5% enantiomeric excess (ee) (entry 4). We found bidentate phosphine ligands such as (S)-SEGPHOS (entry 5) to be more effective, and we discovered that the JOSIPHOS family of ligands provided the best reactivity and enantioselectivity (entries 6–10). Among them, the use of the (*R*, *S_p*)-**L5**, bearing P(4-CF₃Ph)₂ and P^tBu₂ substituents, provided the desired product **4a** in 90% yield and 88% ee (entry 8). With the more active catalyst system, it was found that reactions of **1a** could be conducted at room temperature (RT) (entry 11) and even 0°C (entry 12), leading to a slight increase in enantioselectivity. The greatest enhancement in reactivity (88%, 92% ee) was observed when the reaction was slowly warmed from 0°C to RT (entry 13; see also Figures S8 and S89–S91). Control experiments revealed that the absence of MeOH or NaO^tBu dramatically reduced the yield (entries 14 and 15). The choice of copper catalysis was critical, while the reaction could not work in the presence of FeCl₂ (entry 16).⁴⁹

With the optimized reaction conditions in hand, we then examined the scope of this defluoroborylation reaction (Figure 2). The naphthyl group of substrate **1b** (Figures S38–S40) did not hinder the defluoroborylation process, and the reaction afforded corresponding product **4b** (Figures S9 and S92–S94) in 78% yield and 94% ee. Asymmetric defluoroborylations of 1-(trifluoromethyl)alkenes with either ester (**4c**; Figures S10 and S95–S97) or anisole (**4d**; Figures S11 and S98–S100) moieties on a distal position proceeded smoothly. Biphenyl-containing substrate **1e** was tolerated and gave product **4e** (Figures S12 and S101–S103) in 93% ee. To determine the absolute configuration of our products, we grew crystals of compound **4e** and subjected them to X-ray crystallographic analysis (Figure 2; see also Figure S6 and Tables S1, S2, S3, S4, and S5). A non-activated linear olefin could be transformed successfully under the reaction conditions as well (**4f**; Figures S13 and S104–S106). Reactions of **1g** (Figures S47–S49) and **1h** (Figures S50–S52), which contain multiple sites for possible C–F borylation, showed extremely high selectivity for C–F activation at the vinyl-fluoro functionality over the alkyl terminal position (**4g**–**4h**; Figures S14, S15, and S107–S112). It should be noted that the chemoselectivity of the chiral copper catalyst was not affected by halo substituents such as Cl (**4i**; Figures S16 and S113–S115), Br (**4j**–**4k**; Figures S17, S18, and S116–S121), and I (**4l**; Figures S19, and S122–S124) on the substrates, highlighting the potential of this process in combination with subsequent conventional cross-coupling transformations. Adducts **4m** (Figures S20 and S125–S127) and **4n** (Figures S21 and S128–S130) were formed by the coupling of the corresponding Bz- and TBS-protected alcohols with B₂pin₂, respectively. Substrate **1o** (Figures S71–S73), with a cyano group, was found to be compatible with the reaction conditions and produced product **4o** (Figures S22 and S131–S133) in 72% yield with 87% ee. Notably, substrates with strong coordination abilities, such as those with amine (**4p**; Figures S23 and S134–S136), sulfide (**4q**; Figures S24 and S137–S139) and O-phthalimide (**4r**; Figures S25 and S140–S142) moieties, still underwent the defluoroborylation to afford the desired products in good yields. In addition, indole-containing compound **1s** (Figures S80–S82) was also accommodated with minimal impact on yield or enantioselectivity (82% yield, 89% ee; see also Figures S26 and S143–S145). Finally, having established an enantioselective defluoroborylation of

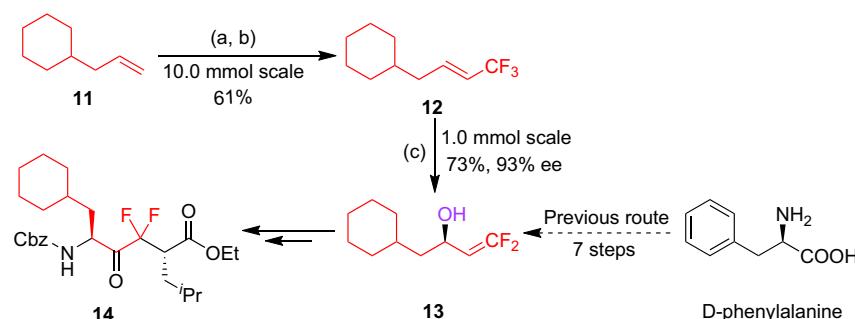


Scheme 1. Enantioenriched gem-Difluoroallylboronates as Versatile Intermediates

Reagents and conditions: (a) 1a (0.20 mmol), 2a (0.40 mmol), NaO^tBu (0.40 mmol), MeOH (0.40 mmol) in CH₃CN (1.0 mL), 12 hr, 0°C to RT, under Ar; solvent was removed and CH₂ICl (0.50 mmol) and ⁷LiBu (0.30 mmol) were added in THF (1.0 mL), 12 hr, 0°C; (b) then NaBO₃·4H₂O (0.40 mmol), THF/H₂O, 2 hr, RT, under Ar; (c) 4a (0.20 mmol), red Al (0.24 mmol), toluene, 12 hr, 80°C, under Ar; (d) 4a (0.20 mmol), CuTc (10 mol % mmol), Xantphos (10 mol % mmol), LiO^tBu (0.60 mmol), B₂pin₂ (0.60 mmol), H₂O (0.60 mmol), DMA (1.0 mL), 16 hr, 40°C, under Ar; (e) 4b (0.20 mmol), Pd/C (10 mol % mmol), 1 atm H₂, THF (1.0 mL), 6 hr, RT; (f) 4m (0.20 mmol), NaOH (2.0 mmol), MeOH/H₂O (20:1), 30 min, RT.

1-(trifluoromethyl)alkenes with relatively simple molecules, we applied this strategy to compounds derived from complex natural products. Substrate 1t, derived from estrone, allowed the diastereoselective construction of product 4t (Figures S27 and S146–S148) in 62% yield with 86% diastereomeric excess. In addition, lithocholic acid derivative 1u (Figures S83–S85) was borylated with excellent diastereoselectivity. The reactivity was sensitive to substrate structure, and defluoroborylation of trifluoropropene substrate 1v containing a secondary alkyl substituent only afforded a mixture of diastereoisomers 4v (Figures S1 and S152–S154) in low conversion. When β -(hetero)aryl-containing 1-(trifluoromethyl)alkenes such as 1w and 1x were subjected to the reaction, only a small amount of α -boronate addition products were generated at the current reaction conditions.

To showcase the practical utility of our copper-catalyzed asymmetric defluoroborylation process, we conducted a 2.0-mmol reaction and obtained gem-difluoroallylboronate 3a (Figures S7 and S86–S88) in 55% yield and 92% ee following flash column chromatography on silica gel (Scheme 1). As noted at the outset, enantioenriched products bearing both boronate and gem-difluorovinyl groups are extremely versatile intermediates in organic synthesis as they can be converted to other important families of compounds with retention of the ee at the boron-bound carbon. Several illustrative examples are provided. Boronate 5 (Figures S155–S157), with an additional methylene unit, can be generated from substrate



Scheme 2. Application of Our Developed Method as a Key Step in the Synthesis of a Ketodifluoromethylene Dipeptide Isostere

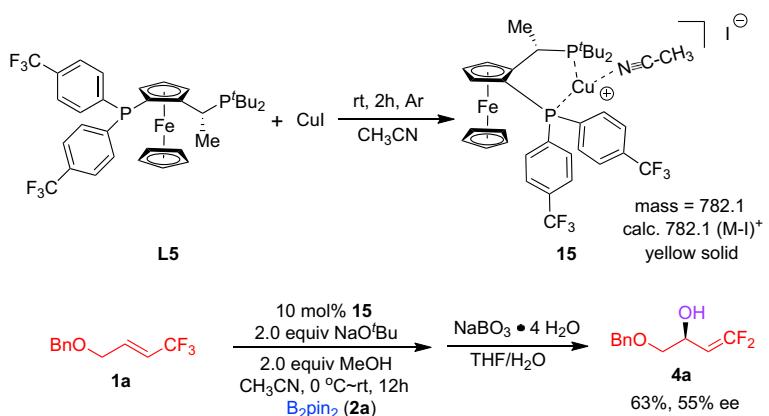
Reagents and conditions: (a) 11 (10.0 mmol), NaSO₂CF₃ (30.0 mmol), I₂O₅ (20.0 mmol), DCM/H₂O (4:1), 12 hr, 110°C; (B) DBU (20.0 mmol), DCM, 2 hr, RT; (C) 12 (1.0 mmol), CuI (10 mol %), L5 (11 mol %), NaO^tBu (2.0 mmol), B₂pin₂ (2.0 mmol), MeOH (2.0 mmol), CH₃CN (10.0 mL), 12 hr, 0°C to RT, under Ar.

1a through a tandem homologation with the *in situ*-formed CH₂ClLi reagent in 66% yield and 90% ee.^{50,51} This crude reaction mixture can be used to further generate corresponding homoallylic gem-difluoroallyl alcohol 6 (Figures S29 and S158–S160) in 63% yield and 90% ee. Reduction of 4a with red Al afforded E-fluoroalkene 7 (Figures S30 and S161–S163) in 89% yield and 87% ee (E/Z = 95:5). Alternatively, utilizing our recently developed method for copper-catalyzed hydrodefluorinations of gem-difluoroalkenes with water could afford the Z-fluoroalkene 8 (Figures S31 and S164–S166) in 95% yield without loss of ee.⁴⁷ The double bond in product 4h could be hydrogenated to yield CF₂H-substituted alkane 9 (Figures S32 and S167–S169) in 92% yield and 92% ee. In addition, hydrolysis product 4m afforded access to diol 10 (Figures S33 and S170–S172) without erosion of the enantiopurity.

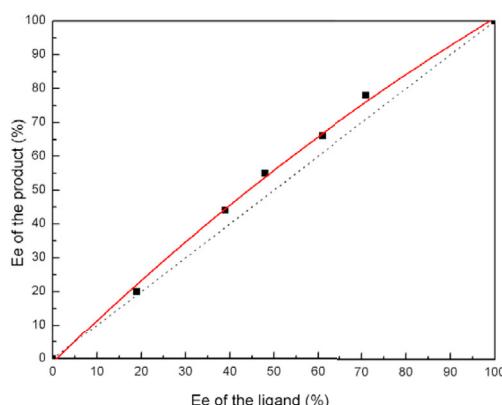
The utility of our strategy can also be exemplified by its use in a key step in the asymmetric synthesis of ketodifluoromethylene dipeptide isostere 14 (Scheme 2). Our synthesis commenced with the preparation of 1-(trifluoromethyl)alkene 12 (Figures S173–S175) from commercially available olefin 11. The reaction of 12 with B₂pin₂ (2a) under the optimized conditions formed key intermediate 13 (Figures S34 and S176–S178) in 73% yield and 94% ee, which was prepared from D-phenylalanine in seven steps in a previous route.⁵² We envisaged that this efficient method could be used for the diversity-oriented synthesis of ketodifluoromethylene dipeptide isosteres for medicinal applications in the future.

How bidentate phosphine ligands induce high levels of enantioselectivity in the transition metal-catalyzed asymmetric reactions has been a topic of interest for some time. Independent generation of the 1:1 Cu:L5 complex in CH₃CN afforded a yellow solid copper complex 15 (Figures S2–S4) confirmed by electrospray ionization mass spectrometry and ³¹P- and ¹H-nuclear magnetic resonance studies. However, copper complex 15 did not prove to be a competent catalyst in the presence of 1-(trifluoromethyl)alkene 1a and B₂pin₂ (2a), identical to the CuI/L5 system in terms of yield and enantioselectivity (Scheme 3A). This result indicates that complex 15 is not a true catalyst for our reaction. To provide more insights into the mechanism of the present catalytic system, we examined nonlinear effects (NLE)⁵³ of the copper-catalyzed asymmetric defluoroborylation of 1-(trifluoromethyl)alkenes using L5 with varying ee as the ligand under the optimized reaction conditions. A weak

A Investigation of a pregenerated copper complex 15



B Nonlinear relationship between enantioselectivity of L5 and 4a



Scheme 3. Mechanistic Experiments

positive NLE was observed (Scheme 3B), indicating that the copper(I) species involved in the catalysis should contain more than one bidentate phosphine ligand within or at the periphery of the catalytic cycle.⁵⁴

From these experimental results and literature precedent, the proposed mechanism of the Cu-catalyzed chiral gem-difluoroallylboronate formation is shown in Figure 3. The reaction of the precatalyst (CuI), chiral ligand, and NaO*t*Bu forms the catalytically active species (A). This complex can easily undergo transmetalation with the anionic adduct of B₂pin₂ (2a')^{55,56} to generate intermediate B. Reaction of 1-(trifluoromethyl)alkenes (1, R = alkyl), was found to afford only C-B bonds at the β position. Such preferences were attributed to conjugate CF₃ substituent (C). While using 1-(trifluoromethyl)alkenes (1, R = (hetero)aryl) such as 1w-1x as substrates, C-Cu bond formation may be electronically favored at the benzylic position stabilization by the (hetero)aryl group (C').^{57–62} After further enantioselective formation of intermediate D, *cis* β-F elimination^{33–40,45–49} delivers the final product 3 and a CuF-like species E, which can undergo anion exchange with NaO*t*Bu to regenerate the copper catalyst.

Conclusion

In summary, we have developed an efficient asymmetric copper-catalyzed system that can activate the C-F bonds of 1-(trifluoromethyl)alkenes via a defluoroborylation process to produce a diverse array of enantioenriched gem-difluoroallylboronates.⁶³ Given the ubiquity of the gem-difluorovinyl group and its precursors in bioactive compounds, we anticipate that this strategy based on the diversity of boron chemistry will simplify the synthesis and structural elaboration of gem-difluoroalkene targets for research in chemistry, biology, and medicine.

EXPERIMENTAL PROCEDURES

Full experimental procedures are provided in the [Supplemental Information](#).

DATA AND SOFTWARE AVAILABILITY

The crystallography data have been deposited at the Cambridge Crystallographic Data Center (CCDC) under accession number CCDC: 1821460 (4e) and can be obtained free of charge from www.ccdc.cam.ac.uk/getstructures.

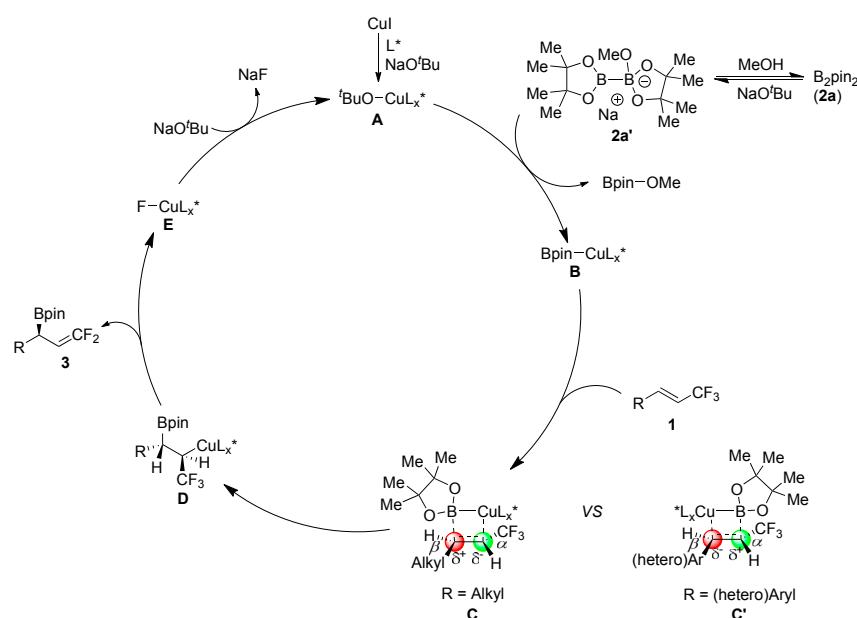


Figure 3. Proposed Catalytic Cycle

SUPPLEMENTAL INFORMATION

Supplemental Information includes Supplemental Experimental Procedures, 178 figures, 5 tables, and 1 data file and can be found with this article online at <https://doi.org/10.1016/j.chempr.2018.07.003>.

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AUTHOR CONTRIBUTIONS

Z.S. conceived and designed the study and wrote the paper. P.G. and C.Y. performed the experiments and mechanism study and analyzed the data. Y.Z. performed the crystallographic studies.

DECLARATION OF INTERESTS

The authors declare no competing interests.

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63. During the revision process for this manuscript, Ito et al. reported a related paper: Kojima, R., Akiyama, S., and Ito, H. (2018). A copper(I)-catalyzed enantioselective γ -boryl substitution of trifluoromethyl-substituted alkenes: synthesis of enantioenriched γ,γ -*gem*-difluoroallylboronates. *Angew. Chem. Int. Ed.* 57, 7196–7199.