Facile P,N-heterocycle synthesis *via* tandem aminomethylation–cyclization of *H*-phosphinate building blocks[†]

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Various heterocycles containing phosphorus and nitrogen are easily synthesized from readily available *H*-phosphinate building blocks. Aminomethylation of these *H*-phosphinates is followed by *in situ* cyclization through substitution or cross-coupling to produce novel heterocycles in moderate to good yields.

Introduction

Perhaps surprisingly, few phosphorus–nitrogen heterocycles have been synthesized previously.¹ In recent work, we reported the formation of such heterocycles starting from ω -amino-*H*phosphinates (eqn (1)) through condensation with carbonyl compounds.² The method was made possible by synthetic methodologies we developed to access the starting materials.³ Herein, we shifted the focus to precursors which do not contain an amino group, but instead possess a reactive halide.



With the exception of our previous report (eqn (1)),²P,Ncontaining phosphinic heterocycles are rare in the literature. Since heterocycles have shown a variety of biological activities, and since cyclic P,N-phosphinates can be considered analogs of amino acids, we set out to expand the range of compounds accessible from simple building blocks. Below, we describe a general approach to P,Nheterocycles based on the Kabachnick–Fields aminomethylation⁴ of precursors, followed by *in situ* cyclization of the resulting amines (Scheme 1). Over the past several years, our laboratory has reported several general methods to prepare functionalized H-phosphinates.³ In the present work, such intermediates are prepared and their use in the synthesis of P,N-heterocycles is investigated.



Scheme 1 Strategy for the synthesis of P,N-heterocycles

† Electronic supplementary information (ESI) available: Additional procedures and spectral data. See DOI: 10.1039/b917428a

Results and discussion

Precursor synthesis



Several synthons were prepared through established reactions.³ Compounds **1**, **2** and **5** were prepared *via* hydrophosphinylation,^{3d-j} compound **3** *via* Pd-catalyzed cross-coupling,^{3k} and compound **4** *via* base-promoted alkylation.^{3r} The detailed conditions are shown in Schemes 2–4, and eqn (2) and (3). Synthon **1** (Scheme 2) was synthesized through our radical hydrophosphinylation using Et₃B as the initiator, as reported previously.^{3h} The resulting intermediate **6** was esterified⁵ directly to produce ester **1**. The low yield of compound **1** was attributed to difficulties during the purification of this polar compound using chromatography on silica gel. Nonetheless, the preparation of **1** was easily accomplished.



Homolog **2** was prepared in a single step through palladiumcatalyzed hydrophosphinylation^{3d} of 4-bromo-1-butene (eqn (2)). Synthon **2** was obtained in 67% yield. A similar reaction was used to prepare **5** from commercially available 2-bromostyrene (eqn (3)) in 64% yield.

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Synthon **3** required the use of our palladium-catalyzed cross-coupling of anilinium hypophosphite^{3k} with 1-bromo-2-iodobenzene. Esterification then proceeded in reasonable yield to form **3** (Scheme 3).



Scheme 3 Synthesis of ethyl (2-bromophenyl)-*H*-phosphinate 3. (a) 1-bromo-2-iodobenzene, anilinium hypophosphite (3 equiv.), $Pd(OAc)_2$ (2 mol%), dppp (2.2 mol%), CH_3CN , reflux, 16 h, 65%; (b) (EtO)₄Si, toluene, reflux, 24 h, 64%.

Finally, synthon **4** was synthesized through the LiHMDSpromoted alkylation^{3r} of 2-bromobenzyl bromide followed by deprotection of the acetal using TMSCl.⁶ The sequence produced **4** in 60% overall yield (Scheme 4).



Scheme 4 Synthesis of ethyl (2-bromobenzyl)-*H*-phosphinate 4. (a) 2-bromobenzyl bromide, LiHMDS, THF, -78 °C to rt, 3 h, 61%; (b) TMSCl, CH₂Cl₂, EtOH, rt, 16 h, 99%.

Reactivity and cyclization

With the above precursors in hand, their reactions with imines were investigated. Table 1 shows the results obtained with synthons 1

 Table 1
 P,N-heterocycle formation from 1 and 2

Entry	Synthon	Imine ^a	Product	Isolated yield (%) ^b
1	1	PhCH=NPh	Q EtO-P Ph N Ph	61
2	1	H ₂ C=NBn	Eto-P N Bn	58
3	2	PhCH=NPh	Ph N- Ph Ph	76
4	2	H ₂ C=NBn	EtO ⁻ P Bn	45

^{*a*} Conditions: *N*-benzylideneaniline, toluene, reflux, 16 h; or 1,3,5tribenzylhexahydro-1,3,5-triazine, toluene, reflux, 16 h. ^{*b*} Yield of pure compound after column chromatography.

and **2**. Reaction of the *H*-phosphinates with imines proceeded smoothly and cyclization then took place uneventfully.

The reactions of 2-bromophenyl-substituted H-phosphinate esters were investigated next (Table 2). Compounds 3 and 4 were subjected to similar reaction conditions, except that a base $(Cs_2CO_3, 1.5 \text{ equiv.})$ and a palladium catalyst $(Pd(PPh_3)_4, 2 \text{ mol}\%)$ were also added to the reaction mixtures. As expected, the aminomethylation took place easily, but, perhaps more surprisingly, the simplest Pd-catalyst delivered C-N bond formation in good yield, without the need for more sophisticated Buchwald-Hartwig-type cross-coupling catalysts.7 Thus, 5- and 6-membered P,N-heterocycles were obtained in a one-pot procedure (Table 2). Compound 3 reacted with a variety of imines (Table 2, entries 1-6). Interestingly, a reaction (entry 3) conducted with in situ formation of the imine (from paraformaldehyde and benzylamine) gave the expected product in only slightly lower yield than with the triazine precursor (entry 2). Compound 4 provided the 6membered heterocycle in comparable yields (entries 7 and 8). Again, Pd(PPh₃)₄ successfully catalyzed the cross-coupling step.

However, with synthon **5**, although the Kabachnik–Fields reaction took place in good yield, the resulting intermediates did not cyclize to give the 7-membered ring under otherwise identical conditions. The aminomethylated products were obtained with *N*-benzylideneaniline or 1,3,5-tribenzylhexahydro-1,3,5-triazine in 51 and 67% isolated yields, respectively.

In a different (but related) approach, the heterocyclization of unsaturated ethyl cinnamyl-H-phosphinate was investigated through a tandem aminomethylation–Heck cyclization process (Scheme 5). Readily available cinnamyl-H-phosphinic acid⁸ 7 was esterified⁵ to **8** using our typical conditions. Ester **8** was then reacted with 2-iodoaniline and paraformaldehyde to form

Table 2 P,N-heterocycle formation from 3 and 4

Entry	Synthon	Imine ^a	Product	Isolated yield (%) ^b
1	3	PhCH=NPh	Ph N P O O OEt	61
2	3	H ₂ C=NBn ^c	P O OEt	63
3	3	$(CH_2O)_n + BnNH_2$	P O OEt	53
4	3	PhCH=NBn	Bn N P O OEt	74
5	3	PhCH=N- <i>t</i> Bu	N P O OEt	44
6	3	PhCH=NMe	Me N P ÓOEt	62
7	4	PhCH=NPh	Ph N Ph P=0 OEt	76
8	4	H ₂ C=NBn ^e	Bn N P=0 OFt	41

^{*a*} Conditions: imine or triazine (1 equiv.), Cs₂CO₃ (1.5 equiv.), Pd(PPh₃)₄ (2 mol%), toluene, reflux, 24 h. ^{*b*} Yield of pure compound after column chromatography. ^{*c*} 1,3,5-tribenzylhexahydro-1,3,5-triazine was used.

aminomethylated iodide **9** in moderate isolated yield. Compound **9** was subjected to Heck-type⁹ reaction conditions (Pd/dppf, 2 mol%) in DMF to produce the desired 7-membered ring heterocycle **10** in 35% isolated yield. We have not optimized this reaction, other than the use of dppf¹⁰ instead of PPh₃. While more work would be required to fully explore this strategy, Scheme 5 provides an interesting "proof of concept", and the basis for future experiments.

Along similar lines, we briefly attempted the Wacker-type cyclization¹¹ of unsaturated amino-*H*-phosphinate **12** (Scheme 6). Unfortunately, attempts on **12a** and **12b** were unsuccessful. The use



Scheme 5 Tandem aminomethylation–Heck cyclization. (a) $(EtO)_4$ Si, toluene, reflux, 24 h, 92%; (b) 2-iodoaniline (1.2 equiv.), paraformaldehyde (1.2 equiv.), toluene, reflux, 16 h, 46%; (c) Pd(OAc)₂ (2 mol%), dppf (2.2 mol%), *i*-Pr₂NEt (2 equiv.), DMF, 110 °C, 35%.



Scheme 6 Attempted Pd(II)-catalyzed cyclization. (a) 1,3,5-tribenzylhexahydro-1,3,5-triazine (0.4 equiv.), toluene, reflux, 16 h, **12a** 57%; **12b** 83%. (b) Pd(OAc)₂ (5 mol%), AcONa (2 equiv.), DMSO, O_2 , 80 °C, 60 h.

of ethyl cinnamyl-*H*-phosphinate **8** *via* sulfonamide **13** similarly failed, although Liu and Stahl reported the successful cyclization of the all-carbon analog **14**.^{11c}



Admittedly, we have not fully investigated the palladiumcatalyzed heterocyclizations of precursors 9, 12 and 13. In spite of the above failed experiments, tremendous flexibility remains available to synthesize phosphorus heterocycles from simple precursors. For example, Wolfe-type carboaminations¹² could also be tried on aminophosphinate precursors 12 and 13. Similarly, the use of allyl amine in the place of 2-iodoaniline (Scheme 5) could lead to a ring-closing metathesis precursor (the reaction of 8 with paraformaldehyde and allyl amine gives the corresponding diene in 53% yield). Our methodologies have made available a wide range of compounds through hydrophosphinylation, crosscoupling (halides, carboxylates, alcohols), alkylation, etc.,3 so that novel P-containing precursors could lead to facile syntheses of heterocyclic products. The preparation of P,O-heterocycles through the well-known addition of H-phosphinates to carbonyl compounds, using the precursors described in the present work, was not investigated because P,N-heterocycles are likely to be more interesting as amino acid analogs.13

Conclusions

The facile synthesis of P,N-heterocycles (substituted 3-hydroxy-1,3-azaphospholane and 3-hydroxy-1,3-azaphosphorinane-3-oxides) is described. With aryl bromide precursors, the cyclization proceeded well using $Pd(PPh_3)_4$ (2 mol%) as the cross-coupling catalyst. The availability of simple *H*-phosphinate building blocks opens up the possibility for the synthesis of various phosphorus heterocycles. Catalytic methods for the cyclization of other phosphinate precursors for the preparation of P,N- as well as other P-heterocycles will be the object of future studies.

Experimental

For general chemistry information, and additional details on the synthesis of precursors and intermediates, spectral data, and a copy of the ³¹P, ¹H and ¹³C NMR spectra, see the ESI.†

Ethyl (3-chloropropyl)-H-phosphinate (Scheme 2, compound 1)

To (3-chloropropyl)-*H*-phosphinic acid^{3g} (2.56 g, 18 mmol) in toluene (60 mL) was added tetraethyl orthosilicate (1.5 equiv., 5.63 g) under N₂, and the mixture was refluxed for 24 h. The solvent was removed *in vacuo* and the resulting oil was purified by column chromatography (silica, EtOAc 100%) to afford the desired product as a yellow oil (1.22 g, 40%): ¹H NMR (CDCl₃, 300 MHz) δ 7.17 (d, *J* = 533 Hz, 1H, H–P), 4.00-4.30 (m, 2H, –CH₂–O–), 3.64 (t, *J* = 5 Hz, 2H, –CH₂–Cl), 1.85-2.10 (m, 4H, 2 x –CH₂–), 1.39 (t, *J* = 7 Hz, 3H, CH₃–); ¹³C NMR (CDCl₃, 75.45 MHz) δ 62.8 (d, *J*_{PCC} = 6 Hz), 44.9 (d, *J*_{PCCCCI} = 17 Hz), 26.3 (d, *J*_{PC} = 95 Hz), 24.2 (d, *J*_{PCC} = 2 Hz), 16.4 (d, *J*_{POCC} = 6 Hz); ³¹P NMR (CDCl₃, 121.47 MHz) δ 37.7 (d, *J* = 533 Hz); HRMS (EI⁺) calc. for C₃H₁₂ClO₂P 171.0342, found 171.0346.

Ethyl (3-bromobutyl)-H-phosphinate (eqn (2), compound 2)^{3e}

To a solution of EtOP(O)H₂ in CH₃CN (2 equiv., 60 mmol, 120 mL) and 4-bromo-1-butene (30 mmol, 4.05 g, 3 mL) were added Pd₂dba₃ (0.5 mol%, 137 mg) and xantphos (1.1 mol%, 191 mg). After 16 h of reflux, the mixture was concentrated and the resulting oil was diluted in EtOAc (60 mL) and washed with brine $(1 \times 20 \text{ mL})$. The organic layer was dried and concentrated. The resulting oil was purified by column chromatography (silica, EtOAc 100%) to afford the desired product as a yellow oil (4.6 g, 67%): ¹H NMR (CDCl₃, 300 MHz) δ 7.13 (d, J = 530 Hz, 1H, H–P), 4.00-4.35 (m, 2H, $-CH_2-O$), 3.43 (t, J = 7 Hz, 2H, $-CH_2-O$ Br), 1.95-2.10 (m, 2H, -CH₂-), 1.65-1.90 (m, 4H, 2 x -CH₂-), 1.37 $(t, J = 7 \text{ Hz}, 3\text{H}, \text{CH}_{3})$; ¹³C NMR (CDCl₃, 75.45 MHz) δ 62.6 (d, $J_{POC} = 7$ Hz), 33.1 (d, $J_{PCC} = 15$ Hz), 32.7, 27.9 (d, $J_{PC} = 94$ Hz), 19.6, 16.4 (d, $J_{POCC} = 6$ Hz); ³¹P NMR (CDCl₃, 121.47 MHz) δ $38.8 (d, J = 530 Hz); HRMS (EI^+) calc. for C_6 H_{14} BrO_2 P 228.9993,$ found 228.9990.

Ethyl 2-(2-bromophenyl)ethyl-*H*-phosphinate (eqn (3), compound 5)

To a solution of EtOP(O)H₂ in CH₃CN (1.6 equiv., 32 mmol, 64 mL) and 2-bromostyrene (20 mmol, 3.66 g) were added Pd_2dba_3 (0.75 mol%, 137 mg) and xantphos (1.6 mol%, 185 mg). After 16 h of reflux, the mixture was concentrated and the resulting oil was

diluted in EtOAc (30 mL) and washed with brine (1 × 10 mL). The organic layer was dried and concentrated. The resulting oil was purified by column chromatography (silica, EtOAc–hexanes 7 : 3, v/v) to afford the desired product as a yellow oil (3.56 g, 64%): ¹H NMR (CDCl₃, 300 MHz) δ 7.53 (d, J = 8 Hz, 1H, *aro* CH), 7.20-7.35 (m, 2H, *aro* CH), 7.14 (d, J = 532 Hz, 1H, H–P), 7.00-7.15 (m, 1H, *aro* CH), 4.00-4.25 (m, 2H, –CH₂–O), 2.90-3.10 (m, 2H, –CH₂–P), 2.00-2.20 (m, 2H, –C–CH₂–), 1.37 (t, J = 14, 7 Hz, 3H, CH₃–); ¹³C NMR (CDCl₃, 75.45 MHz) δ 139.6 (d, $J_{PCC} = 16$ Hz), 133.2, 130.5, 128.6, 127.9, 124.3, 62.7 (d, $J_{POC} = 7$ Hz), 28.9 (d, $J_{PC} = 92$ Hz), 27.9, 16.5 (d, $J_{POCC} = 6$ Hz); ³¹P NMR (CDCl₃, 121.47 MHz) δ 37.4 (d, J = 532 Hz).

Ethyl (2-bromobenzyl)-H-phosphinate (Scheme 4, compound 4)

To ethyl (1,1-diethoxyethyl)-(2-bromobenzyl)phosphinate (3.03 g, 8 mmol) in 20 mL of 10% ethanol in dichloromethane was added chlorotrimethylsilane (1.5 equiv., 12 mmol, 1.3 mL) under N₂ and the clear solution was stirred for 16 h at room temperature.⁶ The solvent was removed in vacuo and the resulting oil was purified by column chromatography (silica, EtOAc-hexanes 3:7, v/v) to afford the desired product as a yellow oil (2.1 g, 99%): ¹H NMR $(CDCl_3, 300 \text{ MHz}) \delta 7.58 \text{ (d, } J = 8 \text{ Hz}, 1 \text{ H}, aro \text{ CH}), 7.20-7.40 \text{ (m,}$ 2H, aro CH), 7.10-7.20 (m, 1H, aro CH), 7.16 (d, J = 544 Hz, 1H, H–P), 4.00-4.25 (m, 2H, –CH₂–O–), 3.35-3.55 (m, 2H, C–CH₂–P), 1.32 (t, J = 14, 7 Hz, 3H, CH₃-); ¹³C NMR (CDCl₃, 75.45 MHz) δ 133.2 (d, J = 2 Hz), 132.1 (d, J = 6 Hz), 130.7 (d, J = 7 Hz), 129.2 (d, J = 4 Hz), 128.1 (d, J = 4 Hz), 124.8 (d, J = 7 Hz), 63.0 $(d, J_{POC} = 6 \text{ Hz}), 37.3 (d, J_{PC} = 89 \text{ Hz}), 16.4 (d, J_{POCC} = 6 \text{ Hz}); {}^{31}\text{P}$ NMR (CDCl₃, 121.47 MHz) δ 34.3 (d, J = 544 Hz); HRMS (EI⁺) calc. for C₉H₁₂BrO₂P 262.9837, found 262.9835.

General procedure for the cyclization from compound 1 or 2

To the compound **1** or **2** (1 mmol) in toluene (10 mL) was added *N*benzylideneaniline (1.2 equiv.) or 1,3,5-tribenzylhexahydro-1,3,5triazine (0.4 equiv) and the mixture was refluxed for 16 h. The solvent was removed *in vacuo*, and the resulting oil was diluted in EtOAc (30 mL) and washed with brine (1 × 10 mL). The organic layer was dried and concentrated. The resulting oil was purified by column chromatography (silica, EtOAc–hexanes 3:7, v/v) to afford the desired product.

1,2-Diphenyl-3-ethoxy-1,3-azaphosphorinane-3-oxide (Table 1, entry 1). Yellow oil, yield: 61%. ¹H NMR (CDCl₃, 300 MHz) δ 7.00-7.50 (m, 7H, aro CH), 6.50-6.80 (m, 3H, aro CH), 4.80-5.00 (m, 1H, -CH₂-), 4.55-4.75 (m, 1H, -P-CH-N-), 4.00-4.25 $(m, 2H, -CH_2-O_{-})_a$, 3.20-3.40 and 3.65-3.85 $(m, 2H, -CH_2-O_{-})_b$, 3.55-3.60 (m, 1H, -CH₂-), 3.40-3.50 (m, 1H, -CH₂-), 2.00-2.20 (m, 1H, -CH₂-), 1.60-2.00 (m, 2H, -CH₂-), 1.20-1.40 (m, 3H, CH₃-)_a, 1.03 (t, J = 14, 7 Hz, 3H, CH₃-)_b; ¹³C NMR (CDCl₃, 75.45 MHz) δ 146.4, 146.2, 135.9, 135.5, 129.5, 129.2, 128.9, 128.3, 128.2 (d, *J* = 4 Hz), 127.8 (d, *J* = 4 Hz), 118.8, 114.3 (d, *J* = 6 Hz), 62.4 (d, $J_{POC} = 7 \text{ Hz}_{a}, 62.1 \text{ (d}, J_{POC} = 7 \text{ Hz}_{b}, 57.9 \text{ (d}, J_{PC} = 93 \text{ Hz}_{b}, 57.7 \text{ (d}, J_{PC} = 93 \text{ Hz}_{b})$ $(d, J_{PC} = 96 \text{ Hz})_a, 45.4 (d, J_{PCC} = 5 \text{ Hz})_a, 45.1 (d, J_{PCC} = 6 \text{ Hz})_b,$ $25.4 (d, J_{PC} = 92 Hz)_a$, $25.2, 24.3 (d, J_{PC} = 95 Hz)_b$, $16.9 (d, J_{POCC} =$ 5 Hz_{a} , 16.6 (d, $J_{POCC} = 6 \text{ Hz}_{b}$; ³¹P NMR (CDCl₃, 121.47 MHz) δ 50.6 (s), 51.6 (s); HRMS (EI⁺) calc for C₁₈H₂₂NO₂P 315.1388, found 315.1388.

1-Benzyl-3-ethoxy-1,3-azaphosphorinane-3-oxide (Table 1, entry 2). Yellow oil, yield: 58%. ¹H NMR (CDCl₃, 300 MHz) δ 7.25-7.35 (m, 5H, *aro* CH), 3.90-4.20 (m, 2H, -CH₂-O-), 3.71 (dd, J = 12, 2 Hz, 1H, -CH₂-), 3.49 (dd, J = 13, 2 Hz, 1H, -CH₂-), 2.96 (t, J = 15 Hz, 1H, -CH₂-), 2.79 (d, J = 12 Hz, 1H, -CH₂-), 2.45 (d, J = 15 Hz, 1H, -CH₂-), 2.20-2.35 (m, 1H, -CH₂-), 1.85-2.10 (m, 3H, -CH₂-CH₂-), 1.60-1.80 (m, 1H, -CH₂-), 1.33 (t, J = 14, 7 Hz, 3H, CH₃-); ¹³C NMR (CDCl₃, 75.45 MHz) δ 137.4, 129.2 (2C), 128.5 (2C), 127.6, 64.4 (d, $J_{PCNC} = 15$ Hz), 60.4 (d, $J_{PCC} = 7$ Hz), 54.2 (d, $J_{PCCC} = 5$ Hz), 52.2 (d, $J_{PC} = 98$ Hz), 25.6 (d, $J_{PC} = 89$ Hz), 23.4 (d, $J_{PCC} = 6$ Hz), 16.7 (d, $J_{POCC} = 6$ Hz); ³¹P NMR (CDCl₃, 121.47 MHz) δ 44.2 (s); HRMS (EI⁺) calc. for C₁₃H₂₀NO₂P 253.1232, found 253.1230.

1,2-Diphenyl-3-ethoxy-1,3-azaphosphepane-3-oxide (Table 1, entry 3). Yellow oil, yield: 76%. ¹H NMR (CDCl₃, 300 MHz) δ 6.50-7.60 (m, 10H, aro CH), 4.69 (d, J = 17 Hz, 1H, -P-CH-N- $_{a}$, 4.61 (d, J = 16 Hz, 1H, $-P-CH-N-_{b}$, 3.70-3.85 and 4.00-4.30 $(m, 2H, -CH_2-O_-), 3.39 (t, J = 13, 6 Hz, 1H, -CH_2-), 3.29 (t, J =$ 13, 7 Hz, 1H, -CH₂-), 1.70-2.00 (m, 4H, -CH₂-CH₂-), 1.50-1.65 $(m, 2H, -CH_2)$, 1.28 $(t, J = 14, 7 Hz, 3H, -CH_3)$, 1.02 $(t, J = 14, 7 Hz, 3H, -CH_3)$ 7 Hz, 3H, -CH₃-)_b; ¹³C NMR (CDCl₃, 75.45 MHz) δ 146.3, 136.1, 135.6, 133.2, 132.1, 129.4, 129.1 (d, J = 2 Hz), 128.9 (d, J = 2 Hz), $128.3 (d, J = 4 Hz)_a$, $127.8 (d, J = 4 Hz)_b$, 118.9, 114.4, 111.8, 62.3 $(d, J_{POC} = 7 \text{ Hz})_a, 61.9 (d, J_{POC} = 7 \text{ Hz})_b, 57.9 (d, J_{PC} = 91 \text{ Hz})_a,$ 57.6 (d, $J_{PC} = 95$ Hz)_b, 33.6 (d, $J_{PCC} = 3$ Hz)_a, 33.4 (d, $J_{PCC} =$ 3 Hz_{b} , 33.0, 32.9, 26.9 (d, $J_{PC} = 78 \text{ Hz}_{a}$, 25.7 (d, $J_{PC} = 80 \text{ Hz}_{b}$, 20.7 (d, $J_{PCCC} = 5 \text{ Hz}_{a}$, 20.3 (d, $J_{PCCC} = 4 \text{ Hz}_{b}$, 16.9 (d, $J_{POCC} =$ 5 Hz_{a} , 16.7 (d, $J_{POCC} = 5 \text{ Hz}_{b}$; ³¹P NMR (CDCl₃, 121.47 MHz) δ 51.8 (s), 50.9 (s); HRMS (EI⁺) calc. for C₁₉H₂₄NO₂P 329.1544, found 329.1545.

1-Benzyl-3-ethoxy-1,3-azaphosphepane-3-oxide (Table 1, entry 4). Yellow oil, yield: 45%. ¹H NMR (CDCl₃, 300 MHz) δ 7.25-7.35 (m, 5H, *aro* CH), 3.90-4.00 (m, 1H, $-N-CH_2-P$), 3.60-3.80 (m, 3H, $-N-CH_2-P$ and $-CH_2-O$), 2.90-3.10 (m, 2H, $-N-CH_2-P$), 2.80-2.90 (m, 1H, $-CH_2-$), 2.60-2.70 (m, 1H, $-CH_2-$), 2.00-2.20 (m, 3H, $-CH_2-CH_2-$), 1.55-1.90 (m, 3H, $-CH_2-CH_2-$), 1.23 (t, J = 14, 7 Hz, 3H, $-CH_3-$); ¹³C NMR (CDCl₃, 75.45 MHz) δ 138.7, 129.2 (2C), 128.5 (2C), 127.5, 64.2 (d, $J_{PCNC} = 16$ Hz), 60.0 (d, $J_{POC} = 7$ Hz), 58.3, 54.5 (d, $J_{PCC} = 105$ Hz), 30.2, 29.1 (d, $J_{PC} = 88$ Hz), 19.9 (d, $J_{PCC} = 2$ Hz), 16.7 (d, $J_{POCC} = 6$ Hz); ³¹P NMR (CDCl₃, 121.47 MHz) δ 63.1 (s); HRMS (EI⁺) calc. for C₁₄H₂₂NO₂P 267.1388, found 267.1389.

General procedure for the cyclization from compound 3 or 4

To compound **3** or **4** (5 mmol) in toluene (50 mL) was added the corresponding imine (1.2 equiv.) and the mixture was refluxed for 16 h. Then, caesium carbonate (1.2 equiv.) and Pd(PPh₃)₄ (2 mol%) were added, and the mixture was refluxed for 24 h. The solvent was removed *in vacuo*, and the resulting oil was diluted in EtOAc (30 mL) and washed with brine (1 × 10 mL). The organic layer was dried and concentrated. The resulting oil was purified by column chromatography (silica, EtOAc–hexanes 3:7, v/v) to afford the desired product.

1,2-Diphenyl-3-ethoxy-1,2-dihydro-benzo[1,3]azaphosphole-3oxide (Table 2, entry 1). Yellow oil, yield: 61%. ¹H NMR (CDCl₃, 300 MHz) δ 6.80-7.70 (m, 14H, *aro* CH), 5.13 (d, J = 15 Hz, 1H, -N-CH-P-)_b, 4.79 (d, J = 17 Hz, 1H, -N-CH-P-)_a, 4.20-4.30 (m, 2H, $-CH_2-O_{-a}$, 3.85-3.95 and 3.20-3.30 (m, 2H, $-CH_2-O_{-b}$, 1.38 (t, J = 14, 7 Hz, 3H, $CH_{3}-)_{a}$, 0.95 (t, J = 14 Hz, 3H, $CH_{3}-)_{b}$; ¹³C NMR (CDCl₃, 75.45 MHz) δ 154.7 (d, J = 22 Hz)_b, 152.9 (d, J = 22 Hz)_a, 143.2 (d, J = 8 Hz)_a, 142.3 (d, J = 10 Hz)_b, 134.8 (d, J = 2 Hz)_b, 134.6 (d, J = 2 Hz)_a, 134.1 (d, J = 7 Hz), 129.8, 129.7, 129.5 (d, J = 6 Hz), 129.1 (d, J = 2 Hz), 128.9 (d, J = 6 Hz), 128.8 (d, J = 2 Hz), 128.2 (d, J = 3 Hz), 128.18, 128.12, 127.6 (d, J = 5 Hz), 126.1, 125.6 (d, J = 83 Hz), 123.2, 120.3, 120.1, 119.9, 112.8 (d, J = 10 Hz)_a, 112.1 (d, J = 11 Hz)_b, 65.6 (d, $J_{PC} =$ 96 Hz)_b, 64.8 (d, $J_{PC} = 99$ Hz)_a, 62.6 (d, $J_{POC} = 7$ Hz)_b, 62.1 (d, $J_{POC} = 7$ Hz)_a, 16.9 (d, $J_{POCC} = 6$ Hz)_a, 16.4 (d, $J_{POCC} = 6$ Hz)_b; ³¹P NMR (CDCl₃, 121.47 MHz) δ 50.6 (s), 49.4 (s); HRMS (EI⁺) calc. for C₂₁H₂₀NO₂P 349.1232, found 349.1237.

1-Benzyl-3-ethoxy-1,2-dihydro-benzo[1,3]azaphosphole-3-oxide (**Table 2, entry 2).** Yellow oil, yield: 63%: ¹H NMR (CDCl₃, 300 MHz) δ 7.20-7.80 (m, 7H, *aro* CH), 6.75-6.85 (m, 2H, *aro* CH), 4.52 (q, J = 16 Hz, 2H, $-N-CH_2-P-$), 4.10-4.20 (m, 2H, $-CH_2-O-$), 3.37 (dd, J = 13, 5 Hz, 2H, $-N-CH_2-Ph$), 1.34 (t, J = 14, 7 Hz, 3H, CH₃-); ¹³C NMR (CDCl₃, 75.45 MHz) δ 155.3 (d, J = 23 Hz), 136.8, 135.0 (d, J = 2 Hz), 132.3 (d, J = 16 Hz), 129.1, 128.8, 128.7 (d, J = 5 Hz), 127.9, 127.4, 117.9 (d, J = 12 Hz), 13.2 (d, J = 131 Hz), 109.9 (d, J = 12 Hz), 61.9 (d, $J_{POC} = 6$ Hz), 52.5 (d, $J_{PCNC} = 6$ Hz), 47.6 (d, $J_{PC} = 102$ Hz), 16.8 (d, $J_{POCC} = 6$ Hz); ³¹P NMR (CDCl₃, 121.47 MHz) δ 52.4 (s); HRMS (EI⁺) calc. for C₁₆H₁₈NO₂P 287.1075, found 287.1080.

1-Benzyl-3-ethoxy-2-phenyl-1,2-dihydro-benzo[1,3]azaphosphole-3-oxide (Table 2, entry 4). Yellow oil, yield: 74%. ¹H NMR $(\text{CDCl}_3, 300 \text{ MHz}) \delta 6.65-7.70 \text{ (m, 14H, aro CH)}, 4.75 \text{ (d, } J =$ 16 Hz, 2H, -N-CH-Ph), 4.61 (d, J = 14 Hz, 1H, $-N-CH-P-)_b$, 4.41 (d, J = 16 Hz, 1H, $-N-CH_2-P-)_a$, 4.00-4.25 (m, 2H, $-CH_2-P-P-)_a$) O_{-} , 3.15-3.35 and 3.75-3.95 (m, 2H, $-CH_2-O_{-}$), 1.32 (t, J =14, 7 Hz, 3H, CH₃-)_a, 0.91 (t, J = 14 Hz, 3H, CH₃-)_b; ¹³C NMR $(CDCl_3, 75.45 \text{ MHz}) \delta 155.2 (d, J = 23 \text{ Hz})_b, 154.6 (d, J = 23 \text{ Hz})_a,$ 137.0, 136.4, 135.21, 135.18, 133.87, 133.82, 133.74, 132.3 (d, J = 10 Hz), 129.5 (d, J = 5 Hz), 129.2 (d, J = 2 Hz)_a, 129.1 (d, J = 2 Hz_{h} , 129.0, 128.5 (d, J = 2 Hz), 128.3 (d, J = 4 Hz), 127.73, $127.69, 127.63, 127.3, 118.7 (d, J = 12 Hz)_{b}, 118.2 (d, J = 12 Hz)_{a},$ 113.5, 113.1, 111.7, 111.4, 110.7 (d, $J = 11 \text{ Hz})_a$, 109.4 (d, J =11 Hz)_b, 63.4 (d, $J_{PC} = 98$ Hz)_b, 62.5 (d, $J_{POC} = 7$ Hz)_b, 61.95 (d, $J_{PC} = 101 \text{ Hz}_{a}$, 61.93 (d, $J_{POC} = 7 \text{ Hz}_{a}$, 49.1 (d, $J_{PCNC} = 8 \text{ Hz}_{b}$, 48.9 (d, $J_{PCNC} = 8 \text{ Hz}_{a}$, 17.0 (d, $J_{POCC} = 6 \text{ Hz}_{a}$, 16.4 (d, $J_{POCC} =$ 6 Hz)_b; ³¹P NMR (CDCl₃, 121.47 MHz) δ 51.1 (s), 49.8 (s); HRMS (EI⁺) calc. for C₂₂H₂₂NO₂P 363.1388, found 363.1395.

1-*tert*-**Buty1-3**-ethoxy-**2**-pheny1-**1**,**2**-dihydro-benzo[**1**,**3**]azaphosphole-**3**-oxide (**Table 2**, entry **5**). Yellow oil, yield: 44%. ¹H NMR (CDCl₃, 300 MHz) δ 6.70-7.80 (m, 9H, *aro* CH), 4.84 (d, J = 20 Hz, 1H, -N–CH–P–)_b, 4.66 (d, J = 20 Hz, 1H, -N–CH–P–)_a, 4.00-4.25 (m, 2H, -CH₂–O–)_b, 3.15-3.35 and 3.75-3.95 (m, 2H, -CH₂–O–)_b, 1.43 (s, 9H, -N–C(CH₃)₃)_a, 1.41 (s, 9H, -N–C(CH₃)₃)_b, 1.33 (t, J = 14, 7 Hz, 3H, CH₃–)_a, 0.80 (t, J = 14 Hz, 3H, CH₃–)_b; ¹³C NMR (CDCl₃, 75.45 MHz) δ 154.7 (d, J = 25 Hz)_b, 154.1 (d, J = 24 Hz)_a, 139.1 (d, J = 4 Hz), 137.1, 134.18, 134.15, 134.13, 132.3 (d, J = 10 Hz), 130.1 (d, J = 6 Hz)_a, 129.3 (d, J = 5 Hz)_b, 129.04, 129.01, 128.97, 127.8 (d, J = 3 Hz)_a, 127.6 (d, J = 3 Hz)_b, 126.9 (d, J = 4 Hz)_a, 115.5, 114.8 (d, J = 11 Hz)_b, 114.1 (d, J = 10 Hz)_b, 113.8, 62.7 (d, $J_{POC} = 7$ Hz)_b, 61.9 (d, $J_{PC} = 93$ Hz)_b, 61.6 (d, $J_{POC} = 7 \text{ Hz}_{a}$, 60.3 (d, $J_{PC} = 100 \text{ Hz}_{a}$, 57.7 (d, $J_{PCNC} = 6 \text{ Hz}_{b}$, 56.8 (d, $J_{PCNC} = 6 \text{ Hz}_{a}$, 29.6 (3C)_b, 29.5 (3C)_a,16.9 (d, $J_{POCC} = 6 \text{ Hz}_{a}$, 16.2 (d, $J_{POCC} = 6 \text{ Hz}_{b}$; ³¹P NMR (CDCl₃, 121.47 MHz) δ 52.3 (s), 51.3 (s); HRMS (EI⁺) calc. for C₁₉H₂₄NO₂P 329.1545, found 329.1546.

3-Ethoxy-1-methyl-2-phenyl-1,2-dihydro-benzo[1,3]azaphosphole-3-oxide (Table 2, entry 6). White solid mp 114-116 °C, yield: 62%. ¹H NMR (CDCl₃, 300 MHz) δ 6.70-7.70 (m, 9H, aro CH), 4.45 (d, J = 15 Hz, 1H, $-N-CH-P-_{b}$, 4.32 (d, J = 15 Hz, 1H, -N-CH-P-)_a, 4.15-4.25 (m, 2H, -CH₂-O-)_a, 3.10-3.30 and 3.75-3.95 (m, 2H, -CH2-O-)b, 2.84 (s, 3H, -N-CH3)b, 2.83 (s, 3H, $-N-CH_3$)_a, 1.37 (t, J = 14, 7 Hz, 3H, CH_3 -)_a, 0.89 (t, J =14 Hz, 3H, CH₃-)_b; ¹³C NMR (CDCl₃, 75.45 MHz) δ 156.2 (d, $J = 23 \text{ Hz}_{a}$, 155.4 (d, $J = 23 \text{ Hz}_{b}$, 135.17, 135.15, 135.12, 135.09, 134.0 (d, J = 7 Hz), 133.9 (d, J = 4 Hz), 129.19, 129.17, 129.10, 129.07, 128.8 (d, J = 5 Hz), 128.47, 128.43, 128.39, 128.1 (d, J = 5 Hz), 127.8 (d, J = 4 Hz), 118.8 (d, J = 12 Hz)_b, 117.9 (d, J = 12 Hz_{a} , 113.3 (d, J = 20 Hz), 111.7 (d, J = 17 Hz), 110.3 (d, J =11 Hz)_b, 109.1 (d, J = 11 Hz)_a, 66.5 (d, $J_{PC} = 98$ Hz)_b, 64.8 (d, $J_{PC} = 101 \text{ Hz}_{a}$, 62.4 (d, $J_{POC} = 7 \text{ Hz}_{a}$, 61.9 (d, $J_{POC} = 7 \text{ Hz}_{b}$, 34.2 (d, $J_{PCNC} = 11 \text{ Hz}_{a}$, 33.5 (d, $J_{PCNC} = 10 \text{ Hz}_{b}$, 16.9 (d, $J_{POCC} =$ 6 Hz_{a} , 16.4 (d, $J_{POCC} = 6 \text{ Hz}_{b}$; ³¹P NMR (CDCl₃, 121.47 MHz) δ 50.5 (s), 49.1 (s); HRMS (EI⁺) calc. for C₁₆H₁₈NO₂P 287.1075, found 287.1073.

1,2-Diphenyl-3-ethoxy-1,2,3,4-tetrahydro-benzo[d][1,3]azaphosphinine (Table 2, entry 7). Yellow oil, yield: 76%: ¹H NMR $(\text{CDCl}_3, 300 \text{ MHz}) \delta 6.80-7.60 \text{ (m, 14H, aro CH)}, 5.26 \text{ (d, } J =$ 22 Hz, 1H, $-N-CH-P-_{b}$, 5.06 (d, J = 25 Hz, 1H, $-N-CH-P-_{a}$, 4.10-4.30 (m, 2H, -CH₂-O-)_a, 3.95-4.10 (m, 1H, -CH₂-O-)_b, 3.60-3.75 (m, 1H, -CH₂-O-)_b, 3.00-3.30 (m, 2H, -C-CH₂-P-), 1.28 (t, J = 14, 7 Hz, 3H, CH₃-)_a, 0.76 (t, J = 14, 7 Hz, 1H, CH₃-)_b; ¹³C NMR (CDCl₃, 75.45 MHz) δ 149.2 (d, J = 4 Hz)_a, 148.5 (d, $J = 4 \text{ Hz}_{b}$, 143.2 (d, $J = 9 \text{ Hz}_{a}$, 142.8 (d, $J = 9 \text{ Hz}_{b}$, 138.4, 136.6 (d, J = 6 Hz), 132.3 (d, J = 12 Hz)_a, 130.9 (d, J = 7 Hz)_b, 129.7, 129.2, 129.17, 129.15, 128.9 (d, J = 2 Hz), 128.5 (d, J =2 Hz), 128.2, 127.87, 127.84, 127.3 (d, J = 4 Hz), 126.8 (d, J =4 Hz), 126.6 (d, J = 10 Hz), 125.7 (d, J = 2 Hz), 125.1, 123.9, $122.2 (d, J = 9 Hz)_a$, 121.9, $121.6 (d, J = 7 Hz)_b$, 120.9, 117.3, 63.2 (d, $J_{PC} = 95 \text{ Hz})_b$, 62.8 (d, $J_{PC} = 91 \text{ Hz})_a$, 61.7 (d, $J_{POC} =$ 7 Hz)_b, 61.6 (d, $J_{POC} = 7$ Hz)_a, 31.5 (d, $J_{PC} = 84$ Hz)_b, 28.6 (d, $J_{PC} = 86 \text{ Hz}_{a}$, 16.7 (d, $J_{POCC} = 6 \text{ Hz}_{a}$, 16.4 (d, $J_{POCC} = 6 \text{ Hz}_{b}$; ³¹P NMR (CDCl₃, 121.47 MHz) δ 43.0 (s), 54.4 (s); HRMS (EI⁺) calc. for C₂₂H₂₂NO₂P 363.1388, found 363.1382.

1-Benzyl-3-ethoxy-1,2,3,4-tetrahydro-benzo[*d*][1,3]azaphosphinine (Table 2, entry 8). Yellow oil, yield: 41%. ¹H NMR (CDCl₃, 300 MHz) δ 7.10-7.80 (m, 7H, *aro* CH), 6.80-7.00 (m, 2H, *aro* CH), 4.35 (dd, J = 5, 2 Hz, 2H, -N-CH₂–P–), 3.85-4.20 (m, 2H, -CH₂–O–), 3.05-3.25 (m, 4H, -C–CH₂–P– and -N–CH₂–Ph), 1.24 (t, J = 14, 7 Hz, 3H, CH₃–); ¹³C NMR (CDCl₃, 75.45 MHz) δ 148.3 (d, J = 4 Hz), 137.3, 132.3 (d, J = 10 Hz), 132.2, 131.0 (d, J = 10 Hz), 128.76 (d, J = 12 Hz), 128.72 (d, J = 35 Hz), 128.4 (d, J = 2 Hz), 127.9, 123.0 (d, J = 7 Hz), 122.9, 115.9 (d, J = 2 Hz), 61.0 (d, $J_{PCC} = 7$ Hz), 57.2 (d, $J_{PCNC} = 12$ Hz), 48.3 (d, $J_{PC} = 111$ Hz), 31.6 (d, $J_{PC} = 87$ Hz), 16.7 (d, $J_{POCC} = 6$ Hz); ³¹P NMR (CDCl₃, 121.47 MHz) δ 57.2 (s); HRMS (EI⁺) calc. for C₁₇H₂₀NO₂P 301.1232, found 301.1227.

Formation in situ of the imine from an aldehyde (Table 2, entry 3)

To the compound **3** (5 mmol) in toluene (50 mL) were added benzylamine (1.2 equiv., 0.65 mL) and paraformaldehyde (1.2 equiv., 198 mg), and the mixture was refluxed for 16 h. Then, caesium carbonate (1.2 equiv.) and Pd(PPh₃)₄ (2 mol%) were added and the mixture was refluxed for 24 h. The solvent was removed *in vacuo*, and the resulting oil was diluted in EtOAc (30 mL) and washed with brine (1 × 10 mL). The organic layer was dried and concentrated. The resulting oil was purified by column chromatography (silica, EtOAc–hexanes 3:7, v/v) to afford the desired product as a yellow oil (760 mg, 53%).

Heck product (Scheme 5, compound 10)

To compound 9 (0.5 mmol, 221 mg) in DMF (2.5 mL) was added N,N-diisopropylethylamine (2 equiv., 1 mmol, 0.2 mL), 1,1'-bis(diphenylphosphino)ferrocene (2.2 mol%, 6.1 mg) and Pd(OAc)₂ (2 mol%, 2.2 mg). After 40 h of reflux, the solvent was removed in vacuo, and the resulting oil was diluted in EtOAc (10 mL) and washed with brine (1 \times 20 mL). The organic layer was dried and concentrated. The resulting oil was purified by column chromatography (silica, EtOAc 100%) to afford the desired product as a pink oil (54.7 mg, 35%): ¹H NMR (CDCl₃, 300 MHz) δ 7.20-7.60 (m, 6H, aro CH), 7.13 (t, J = 15, 7 Hz, 1H, aro CH), 6.94 (d, J = 5 Hz, 1H, aro CH), 6.85 (t, J = 14, 7 Hz, 1H, aro CH),6.67 (d, J = 8 Hz, 1H, -C=CH-Ph), 4.00-4.25 (m, 1H, -NH-), 3.90-4.00 (m, 1H, -CH₂-O-), 3.45-3.70 (m, 3H, P-CH₂-N- and P-CH2-C-), 3.35-3.45 (m, 1H, -CH2-O-), 3.15-3.25 (m, 1H, P-CH₂-C-), 1.09 (t, J = 14, 7 Hz, 3H, CH₃-); ¹³C NMR (CDCl₃, 75.45 MHz) δ 145.5 (d, J = 2 Hz), 137.5 (d, J = 3 Hz), 133.0 (d, J = 7 Hz), 131.9 (d, J = 11 Hz), 131.56, 131.54, 129.6, 128.9 (d, *J* = 2 Hz), 128.88, 128.81, 128.5 (d, *J* = 2 Hz), 127.3, 119.9, 118.3, $60.9 (d, J_{POC} = 7 Hz), 44.3 (d, J_{PC} = 119 Hz), 31.4 (d, J_{PC} = 78 Hz),$ 16.6 (d, $J_{POCC} = 6$ Hz); ³¹P NMR (CDCl₃, 121.47 MHz) δ 54.2 (s); HRMS (EI⁺) calc. for C₁₈H₂₀NO₂P 313.1232, found 313.1230.

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Notes and references

- P,N-heterocycles: (a) L. Maier, Phosphorus Sulfur Relat. Elem., 1981, 11, 149; (b) D. G. Hewitt and M. W. Teese, Aust. J. Chem., 1984, 37, 205; (c) Y. G. Trishin, V. I. Namestnikov and V. K. Bel'skii, Russ. J. Gen. Chem., 1999, 69, 1588; (d) M. M. Campbell, N. I. Carruthers, S. J. Mickel and P. M. Winton, J. Chem. Soc., Chem. Commun., 1984, 200; (e) A. Yiotakis, S. Vassiliou, J. Jiracek and V. Dive, J. Org. Chem., 1996, 61, 6601; (f) J. Grembecka, A. Mucha, T. Cierpicki and P. Kafarski, J. Med. Chem., 2003, 46, 2641; (g) K. Issleib, H. Winkelmann and H.-P. Abicht, Synth. React. Inorg., Met.-Org., Nano-Met. Chem., 1974, 4, 191; (h) K. Issleib, H. Oehme and E. Leißring, Chem. Ber., 1968, 101, 4032; (i) F. Mathey, Phosphorus-Carbon Heterocyclic Chemistry: The Rise of a New Domain, Elsevier, Oxford, 2001 (ISBN: 978-0-08-043952-5).
- 2 C. Queffélec, P. Ribière and J.-L. Montchamp, J. Org. Chem., 2008, 73, 8987.
- Reviews: (a) L. Coudray and J.-L. Montchamp, *Eur. J. Org. Chem.*, 2008, 3601; (b) J.-L. Montchamp, *Speciality Chemicals Magazine*, 2006, 26, 44; (c) J.-L. Montchamp, *J. Organomet. Chem.*, 2005, 690, 2388. Hydrophosphinylation:; (d) S. Deprèle and J.-L. Montchamp, *J. Am.*

Chem. Soc., 2002, 124, 9386; (e) S. Deprèle and J.-L. Montchamp, Org. Lett., 2004, 6, 3805; (f) P. Ribière, K. Bravo-Altamirano, M. Antczak, J. D. Hawkins and J.-L. Montchamp, J. Org. Chem., 2005, 70, 4064; (g) M. I. Antczak and J.-L. Montchamp, Synthesis, 2006, 3080; (h) S. Deprèle and J.-L. Montchamp, J. Org. Chem., 2001, 66, 6745; (i) S. Gouault-Bironneau, S. Deprèle, A. Sutor and J.-L. Montchamp, Org. Lett., 2005, 7, 5909; (j) S. Deprèle and J.-L. Montchamp, J. Organomet. Chem., 2002, 643-644, 154. Cross-coupling:; (k) J.-L. Montchamp and Y. R. Dumond, J. Am. Chem. Soc., 2001, 123, 510; (1) Z. Huang, K. Bravo-Altamirano and J.-L. Montchamp, C. R. Chimie, 2004, 7, 763; (m) K. Bravo-Altamirano, Z. Huang and J.-L. Montchamp, Tetrahedron, 2005, 61, 6315; (n) K. Bravo-Altamirano, I. Abrunhosa-Thomas and J.-L. Montchamp, J. Org. Chem., 2008, 73, 2292; (o) K. Bravo-Altamirano and J.-L. Montchamp, Org. Lett., 2006, 8, 4169; (p) L. Coudray and J.-L. Montchamp, Eur. J. Org. Chem., 2008, 4101; (q) L. Coudray, K. Bravo-Altamirano and J.-L. Montchamp, Org. Lett., 2008, 10, 1123. Base-promoted alkylation:; (r) I. Abrunhosa-Thomas, C. E. Sellers and J.-L. Montchamp, J. Org. Chem., 2007, 72, 2851; (s) I. Abrunhosa-Thomas, P. Ribière, A. C. Adcock and J.-L. Montchamp, Synthesis, 2006, 325.

- 4 (a) X. Cheng, R. Goddard, G. Buth and B. List, Angew. Chem., Int. Ed., 2008, 47, 5079; (b) E. D. Matveeva and N. S. Zefirov, Dokl. Chem., 2008, 420, 137; (c) V. V. Belakhov, Y. D. Shenin and B. I. Ionin, Russ. J. Gen. Chem., 2008, 78, 305; (d) N. S. Zefirov and E. D. Matveeva, ARKIVOC, 2008, (i), 1; (e) S. Bhagat and A. K. Chakraborti, J. Org. Chem., 2007, 72, 1263; (f) E. D. Matveeva, T. A. Podrugina, M. V. Prisyajnoy and N. S. Zefirov, Russ. Chem. Bull., 2006, 55, 1209; (g) H.-J. Cristau, A. Herve and D. Virieux, Tetrahedron, 2004, 60, 877; (h) E. D. Matveeva, T. A. Podrugina, E. V. Tishkovskaya, L. G. Tomilova and N. S. Zefirov, Synlett, 2003, 2321; (i) H.-J. Cristau, A. Coulombeau, A. Genevois-Borella, F. Sanchez and J.-L. Pirat, J. Organomet. Chem., 2002, 643–644, 381; (j) R. A. Cherkasov and V. I. Galkin, Russ. Chem. Rev., 1998, 67, 857.
- 5 Y. R. Dumond, R. L. Baker and J.-L. Montchamp, *Org. Lett.*, 2000, **2**, 3341.
- 6 (a) M. J. Gallagher and H. Honegger, *Tetrahedron Lett.*, 1977, 18, 2987;
 (b) M. J. Gallagher and H. Honegger, *Aust. J. Chem.*, 1980, 33, 287;
 (c) J. G. Dingwall, J. Ehrenfreund, R. G. Hall and J. Jack, *Phosphorus Sulfur Relat. Elem.*, 1987, 30, 571; (d) J. G. Dingwall, J. Ehrenfreund and R. G. Hall, *Tetrahedron*, 1989, 45, 3787; (e) E. K. Baylis, *Tetrahedron Lett.*, 1995, 36, 9385; (f) E. K. Baylis, *Tetrahedron Lett.*, 1995, 36, 9385;
 (g) W. Froestl, S. J. Mickel, R. G. Hall, G. von Sprecher, D. Strub,

D. P. A. Baumann, F. Brugger, C. Gentsch, J. Jaekel, H.-R. Olpe, G. Rihs, A. Vassout, P. C. Waldmeier and H. Bittiger, *J. Med. Chem.*, 1995, **38**, 3297.

- 7 Reviews: (a) D. S. Surry and S. L. Buchwald, Angew. Chem., Int. Ed., 2008, 47, 6338; (b) J. F. Hartwig, Acc. Chem. Res., 2008, 41, 1534; (c) V. T. Abaev and O. V. Serdyuk, Russ. Chem. Rev., 2008, 77, 177; (d) M. Kienle, S. R. Dubbaka, K. Brade and P. Knochel, Eur. J. Inorg. Chem., 2007, 4166; (e) J.-C. Hierso, M. Beaupérin and P. Meunier, Eur. J. Org. Chem., 2007, 3767; (f) B. Schlummer and U. Scholz, Speciality Chemicals Magazine, 2005, 25, 22; (g) B. Schlummer and U. Scholz, Adv. Synth. Catal., 2004, 346, 1599; (h) B. H. Yang and S. L. Buchwald, J. Organomet. Chem., 1999, 576, 125.
- 8 K. Bravo-Altamirano and J.-L. Montchamp, *Org. Synth.*, 2008, **85**, 96. See also ref. 30.
- 9 Reviews: (a) M. Oestreich, *The Mizoroki-Heck Reaction*, Wiley, Oxford, 2009 (ISBN: 0470033940); (b) M. Shibasaki, E. M. Vogl and T. Ohshima, *Adv. Synth. Catal.*, 2004, **346**, 1533; (c) J. T. Link, *Org. React.*, 2002, **60**, 157; (d) I. P. Beletskaya and A. V. Cheprakov, *Chem. Rev.*, 2000, **100**, 3009; (e) W. Cabri and I. Candiani, *Acc. Chem. Res.*, 1995, **28**, 2. For an interesting example using a phosphinate ester, see:; (f) F. Hong, J. Xia and Y. Xu, *J. Chem. Soc., Perkin Trans. 1*, 1994, 1665.
- 10 Y. Belabassi, S. Gouault-Bironneau, J.-L. Montchamp, 1,1'-Bis(diphenylphosphino)ferrocene, in *Encyclopedia of Reagents for Organic Synthesis (eEROS)*, update article, 2006.
- 11 (a) G. Zeni and R. C. Larock, Chem. Rev., 2006, 106, 4644; (b) G. Zeni and R. C. Larock, Chem. Rev., 2004, 104, 2285; (c) G. Liu and S. S. Stahl, J. Am. Chem. Soc., 2007, 129, 6328; (d) V. Kotov, C. C. Scarborough and S. S. Stahl, Inorg. Chem., 2007, 46, 1910; (e) M. M. Rogers, J. E. Wendlandt, I. A. Guzei and S. S. Stahl, Org. Lett., 2006, 8, 2257; (f) J. L. Brice, J. E. Harang, V. I. Timokhin, N. R. Anastasi and S. S. Stahl, J. Am. Chem. Soc., 2005, 127, 2868; (g) S. R. Fix, J. L. Brice and S. S. Stahl, Angew. Chem., Int. Ed., 2002, 41, 164.
- 12 (a) J. P. Wolfe, *Eur. J. Org. Chem.*, 2007, 571; (b) J. S. Nakhla, J. W. Kampf and J. P. Wolfe, *J. Am. Chem. Soc.*, 2006, **128**, 2893; (c) M. B. Bertrand, M. L. Leathen and J. P. Wolfe, *Org. Lett.*, 2007, **9**, 457; (d) M. B. Bertrand and J. P. Wolfe, *Tetrahedron*, 2005, **61**, 6447; (e) J. P. Wolfe and J. S. Thomas, *Curr. Org. Chem.*, 2005, **9**, 625; (f) J. E. Ney and J. P. Wolfe, *J. Am. Chem. Soc.*, 2005, **127**, 8644.
- 13 Biological activity of aminophosphinic acids: (a) N. J. Wardle, S. W. A. Bligh and H. R. Hudson, *Curr. Org. Chem.*, 2007, **11**, 1635; (b) F. Palacios, C. Alonso and J. M. de los Santos, *Chem. Rev.*, 2005, **105**, 899; (c) M. Collinsová and J. Jirácek, *Curr. Med. Chem.*, 2000, **7**, 629.