Asymmetric Synthesis of Thiadecalins via an Organocatalytic Triple Cascade/ Sulfa-Michael Sequence

Dieter Enders,* Bertram Schmid, Nico Erdmann, Gerhard Raabe

Institute of Organic Chemistry, RWTH Aachen University, Landoltweg 1, 52074 Aachen, Germany Fax +49(241)8092127; E-mail: enders@rwth-aachen.de *Received 5 May 2010*

Dedicated to Professor Rolf Huisgen on the occasion of his 90th birthday

Abstract: An efficient two-step asymmetric synthesis of highly substituted *cis*-configured thiadecalins and the corresponding hexahydrobenzothiophene core is described. Thiadecalin derivatives are known for their widespread biological activities, such as antimicrobial and neurotropic properties. The procedure is based on an organocatalytic triple cascade reaction followed by an intramolecular sulfa-Michael addition. In this manner six consecutive stereocentres are controlled and the target molecules are obtained in moderate yields, with virtually complete enantioselectivitiy (ee >99%) and after recystallisation in diastereomeric ratios of >97:3. The relative and absolute configuration was determined by NMR spectroscopy and X-ray structure analysis.

Key words: organocatalysis, asymmetric synthesis, Michael addition, thiadecalins, domino reaction

The research field of organocatalysis¹ has developed rapidly since about the turn of the millennium and can now be seen as a third pillar of asymmetric catalysis beside metal and biocatalysis. Numerous basic organocatalytic protocols for very efficient and highly stereoselective carbon–carbon and carbon–heteroatom bond formations are now part of the strategic arsenal of synthetic chemistry and can be used in the asymmetric total synthesis of natural products and bioactive compounds in general.

Thiadecalin derivatives have been studied intensively with regard to their antimicrobial and antiphage properties including the treatment of infected wounds.² It has also been shown that thiadecalins exhibit neutrotropic activity.³ The corresponding thiadecalin sulfonium salts showed sedative properties during hexenal-induced sleep. In addition, other thiadecalin derivatives are known to be antigestagens and antiglucocorticoids.⁴ These compounds are also mentioned in combination with the treatment of psychosis and addictive behaviour. The antiglucocorticoids can be used to treat spontaneous or narcotic-induced withdrawals, especially from heroin, morphine, methadone, cocaine, and their mixtures.⁵

Among the various subdisciplines of organocatalysis, domino reactions⁶ enabling the highly stereoselective construction of complex molecules in a single operation have been studied intensively in the last few years.⁷ In 2006, our group reported an efficient three-component domino reaction leading to tetrasubstituted cyclohexenecarbaldehydes.⁸ In order to synthesize the potentially bioactive thiadecalins 1 we envisaged extending the triple cascade method by a subsequent intramolecular sulfa-Michael addition⁹ via nucleophilic displacement of bromide 3 to form the sulfa-Michael precursor 2 (Scheme 1).



Scheme 1 Retrosynthetic analysis of the asymmetric thiadecalin synthesis

Employing 5-bromopentanal (4), readily available through DIBAL-H reduction of the commercially available bromo ester, in the cascade reaction with nitrostyrene (5) and cinnamaldehyde (6) under diphenylprolinol TMSether catalysis 7 gave the desired domino product 8 as a 3.8:1 diastereometric mixture of the α -nitro epimers and with practically complete enantiomeric excess (ee >99%) in 68% yield. The major diastereomer shown could be obtained as a pure stereoisomer (dr >97:3, ee >99%) after recrystallisation in 54% yield (Scheme 2).



Scheme 2 Triple domino reaction with 5-bromopentanal (4). *Reagents and conditions*: a) 0.2 M toluene, 20 mol% 7, 24 h 0 °C, then 48 h, r.t.

SYNTHESIS 2010, No. 13, pp 2271–2277 Advanced online publication: 27.05.2010 DOI: 10.1055/s-0029-1218804; Art ID: Z11510SS © Georg Thieme Verlag Stuttgart · New York

The cyclohexenecarbaldehyde **8** thus obtained was then converted into the more stable methyl ester **9** by Pinnick oxidation followed by esterification with HCl/methanol. The displacement of the bromide with potassium thioacetate in DMF proceeded smoothly, affording an 8.5:1 diastereomeric mixture of the corresponding thioester **10**. The minor diastereomer could be easily removed by column chromatography. Deprotection of the thiol with potassium carbonate in anhyd methanol provided the *cis*-thiadecalin **11** via intramolecular sulfa-Michael addition as a single stereoisomer in 62% yield after recrystallisation (Scheme 3).



Scheme 3 First route to thiadecalin 11. *Reagents and conditions*: a) NaClO₂, 2-methylbut-2-ene, buffer; b) SOCl₂, MeOH; c) KSAc, DMF; d) K_2CO_3 , MeOH; recrystallisation.

With the first route to the thiadecalin 11 in hand, we then shortened the protocol to a two-step version by employing the thioester containing aldehyde 12 as a substrate in the triple cascade. After optimisation, we were able to carry out the sulfa-Michael addition with the enal domino products 13. The thiadecalins 14 were obtained with complete enantioselectivity and in diastereomeric ratios ranging from 85:15 to >97:3. To our delight, one recrystallisation gave the virtually pure stereoisomers in moderate yields (Scheme 4).



Scheme 4 Optimised two-step route. *Reagents and conditions*: a) 20 mol% 7, toluene, 24 h, 0 °C, 48 h r.t.; b) K₂CO₃, MeOH.

Synthesis 2010, No. 13, 2271-2277 © Thieme Stuttgart · New York

The scope of the novel asymmetric thiadecalin synthesis was demonstrated by variation of the substituents R^1 and R^2 (Tables 1 and 2). In addition, by shortening the chain length (n = 0) of the aldehyde **12** opened an entry to the corresponding hexahydrobenzothiophene system as was exemplified in the case of **14e**.

Table I Scope of the Domino Reaction
--

13	R^1	R ²	n	Yield (%) ^a	l dr ^b	ee (%) ^c
a	Ph	Ph	1	51	>97:3 (80:20)	>99
b	4-ClC ₆ H ₄	Ph	1	45	>97:3 (95:5)	>99
c	$4-MeOC_6H_4$	Ph	1	43	>97:3 (85:15)	>99
d	4-methylfuran-2-yl	Ph	1	60	>97:3	>99
e	Ph	Ph	0	57	>97:3 (80:20)	>99
f	Ph	Η	1	32	>97:3 (80:20)	>99
f	Ph	Η	1	32	>97:3 (80:20)	>99

^a Yield after recrystallisation.

^b Diastereomeric ratio determined by NMR spectroscopy; dr of the crude product is given in parentheses.

^c The percentage of ee was determined by chiral stationary phase HPLC.

 Table 2
 Scope of the Thiadecalin Synthesis

14	R ¹	R ²	n	Yield (%) ^a	l dr ^b	ee (%)
a	Ph	Ph	1	57	>97:3 (90:10)	>99
b	$4-ClC_6H_4$	Ph	1	40	>97:3 (60:40)	>99
c	4-MeOC ₆ H ₄	Ph	1	52	>97:3 (88:12)	>99
d	4-methylfuran-2-yl	Ph	1	50	>97:3	>99
e	Ph	Ph	0	57	>97:3 (80:20)	>99
f	Ph	Н	1	40	85:15 (70:30)	>99

^a Yield after recrystallisation.

^b Diastereomeric ratio determined by NMR; dr of the crude product is given in parentheses.

The methylfuranyl group as R^1 gave the best results for the triple cascade reaction in terms of both yield and diastereoselectivity. The yield for the subsequent sulfa-Michael reaction was moderate, yet the diasteromeric ratio was better than 97:3, even before recrystallisation. In comparing the electron-donating and -withdrawing groups at the phenyl ring of R¹, no overall tendencies were observed. For n = 0, the reaction proceeded smoothly with 57% yield for the triple cascade and the sulfa-Michael addition, respectively. Further variation on R² was possible, but led to lower yields and diastereomeric ratios. Although a one-pot procedure for the two steps was feasible and led to good overall yields, the main drawback was the lack of diastereomeric excess obtained. For this reason the two-step procedure with purification of the intermediates 13 was preferred.

The relative and absolute configuration of the thiadecalin **14b** and the hexahydrobenzothiophene **14e** was determined by ¹H NMR spectroscopy (coupling constants) and X-ray crystallography (Figures 1 and 2).



Figure 1 Determination of the relative and absolute configuration of 14b (Ar = 4-ClC₆H₄) by ¹H NMR and X-ray analysis



Figure 2 Determination of the relative and absolute configuration of 14e by ¹H NMR and X-ray analysis

As shown in Figure 1, the ¹H NMR coupling constants indicate a chair conformation of the hexa-substituted cyclohexane ring of the *cis*-thiadecalin **14b**. This is also confirmed for the crystalline stage by the corresponding Röntgen structure. In the *cis*-configured hexahydrobenzothiophene derivative **14e**, however, the ¹H NMR and Röntgen data indicate a twist-boat conformation of the cyclohexane moiety (Figure 2).

In summary, we have developed a two-step organocatalytic asymmetric synthesis of functionalised, highly substituted cis-thiadecalins the corresponding and hexahydrobenzothiophene core. The novel entry is based on our diphenylprolinol TMS-ether catalysed triple cascade followed by an intramolecular sulfa-Michael addition. The target molecules are of pharmaceutical and medicinal interest and bear six consecutive stereocentres as part of a fully substituted cyclohexane ring. Besides moderate overall yields, virtually complete enantiomeric excesses (ee >99%) and after crystallisation, with one exception, diastereomeric ratios of >97:3 are obtained.

All reactions were carried out under argon in flame-dried glassware. Unless otherwise noted, all commercially available compounds were used without further purification. Toluene was freshly distilled under argon from Solvona[®] (Na on Celite). MeOH was freshly distilled under argon over Mg and I₂. The catalyst was prepared according to the previously described procedure.¹⁰ For the protocol to prepare 5-bromopentanal (4), see ref.¹¹ For preparative column chromatography SIL G-25 UV₂₅₄ from Macherey-Nagel, particle size 0.040–0.063 mm (230–240 mesh, flash) was used. Visualisation of the developed TLC plates was performed with ultraviolet irradiation (254 nm) or by staining with a KMnO₄ solution. Optical rotation values were measured on a Perkin-Elmer 241 polarimeter.

Mass spectra were measured on a Finnigan SSQ7000 (EI 70 eV) spectrometer and high-resolution mass spectra on a Thermo Fisher Scientific Orbitrap XL. IR spectra were recorded on a Perkin-Elmer FT-IR Spectrum 100 using an ATR-Unit. ¹H and ¹³C NMR spectra were recorded at ambient temperature on Varian Mercury 300 or Inova 400 spectrometers with TMS as an internal standard. Analyt-ical HPLC was performed on a Hewlett-Packard 1100 Series instrument using chiral stationary phases (Chiralcel OD, Chiralcel OJ, Chiralpak AD, Chiralpak AS, Chiralcel IA).

Organocatalytic Domino Reaction; General Procedure (GP 1)

Catalyst **7** (0.2 equiv) and nitroalkene **5** (1.0 equiv) were dissolved in toluene (0.5 M with respect to **12**) and cooled to 0 °C. Aldehyde **12** (1.0 equiv) was slowly added followed by a slow addition of the α,β -unsaturated aldehyde **6** (1.0 equiv). The reaction was kept at 0 °C until TLC showed complete consumption of **12** (usually 24 h) and then allowed to warm to r.t. The stirring was continued for 48 h. The domino product **13** was purified by column chromatography and recrystallisation.

Sulfa-Michael Addition; General Procedure (GP 2)

The domino product **13** (1 equiv) was dissolved in anhyd MeOH (0.02 M). Anhyd K_2CO_3 (1 equiv) was added and the suspension stirred at r.t. until **13** was consumed (usually 20 min). Brine (1 mL per 10 mg of **13**) was added, followed by H_2O (1 mL per 10 mg of **13**), and the product was extracted with CH_2Cl_2 (3 × 1 mL per 10 mg of **13**). The solvent was evaporated and the residue was recrystallised.

Bromocarbaldehyde 8

The synthesis of **8** following GP 1 yielded 4.2 g (54%) of the main diastereomer as colourless crystals; mp 122 °C; ee >99%; $R_f = 0.38$ (pentane–Et₂O, 1:1); $[\alpha]_D^{23}$ –1.1 (*c* 1, CHCl₃).

IR (film): 1648, 1546, 1451, 1401, 1366, 1285, 1240, 1166, 887, 772, 754, 699, 614, 552 cm⁻¹.

¹H NMR (400 MHz, CDCl₃): δ = 1.62 (m, 1 H), 1.79 (m, 1 H), 1.95 (m, 1 H), 2.08 (m, 1 H), 3.01 (dd, *J* = 10.6, 3.4 Hz, 1 H), 3.36 (m, 2 H), 3.41 (m, 1 H), 4.47 (s, 1 H), 4.87 (dd, *J* = 3.4, 1.6 Hz, 1 H), 6.99 (m, 2 H), 7.20 (m, 2 H), 7.42–7.27 (m, 7 H), 9.61 (s, 1 H).

¹³C NMR (100 MHz, CDCl₃): δ = 29.6, 30.9, 32.7, 36.4, 42.9, 43.2, 92.3, 127.6, 127.7, 127.8, 128.1, 129.0, 129.1, 136.7, 137.6, 138.3, 153.0, 191.5.

MS (EI): m/z = 427.0 (M⁺).

Anal. Calcd for C₂₂H₂₂BrNO₃: C, 61.69; H, 5.18; N, 3.27. Found: C, 61.86; H, 4.76; N, 3.35.

Bromomethyl Ester 9

Compound 8 (4.0 g, 9.4 mmol) was dissolved in acetone (50 mL) and cooled to 0 °C. After the addition of 2-methylbut-2-ene (6.60 g, 10 equiv), a solution of NaClO₂ (2.13 g, 2.0 equiv) in H₃PO₄-aq NaHCO₃ buffer (15 mL, 1.25 M, pH 3.5) was added dropwise. The reaction mixture was allowed to warm to r.t. and stirred until completion of the reaction (TLC). Acetone was evaporated and the residue was treated with brine (40 mL). After extraction with EtOAc $(3 \times 50 \text{ mL})$, the combined organic phases were evaporated and the crude product was purified by flash chromatography (Et₂O) yielding the carboxylic acid in 3.38 g (81%) as a pale-brown solid. Most of the carboxylic acid (3.24 g) was redissolved in MeOH (70 mL) at 0 °C to obtain a 0.1 M solution. After the dropwise addition of SOCl_2 (3.47 g, 4.0 equiv), the solution was refluxed at 75 $^\circ\mathrm{C}$ until completion of the reaction (4 h). MeOH was evaporated, the residue was treated with H₂O (30 mL) and the product was extracted with EtOAc (3×30 mL). The crude product was purified by flash chromatography yielding 2.94 g (88%) of yellow crystals; mp 137 °C; ee >99%; dr >97:3; $R_f = 0.58$ (pentane–Et₂O, 1:1); $[\alpha]_D^{23}$ -20.1 (c 1, CHCl₃).

IR (CHCl₃): 2950, 1717, 1550, 1448, 1368, 1272, 758, 702 cm⁻¹.

¹H NMR (400 MHz, $CDCl_3$): $\delta = 1.54$ (m, 1 H), 1.73 (m, 1 H), 1.91 (m, 1 H), 2.97 (m, 1 H), 2.94 (dd, J = 10.4, 3.5 Hz, 1 H), 3.28 (m, 1 H), 3.35 (m, 2 H), 3.65 (s, 3 H), 4.52 (s, 1 H), 4.88 (dd, J = 3.5, 2.0 Hz, 1 H), 6.99 (m, 2 H), 7.23–7.42 (m, 8 H), 7.46 (d, J = 1.9 Hz, 1 H).

 ^{13}C NMR (100 MHz, CDCl₃): δ = 29.6, 31.2, 32.9, 35.8, 42.5, 44.8, 52.0, 92.6, 127.4, 127.6, 127.7, 127.9, 128.9, 129.0, 137.0, 139.4, 143.2, 165.8.

MS (EI): m/z = 457.1 (M⁺).

HRMS: *m/z* calcd for C₂₃H₂₄BrNO₄: 457.0889; found: 457.0885.

Methyl Ester 10

To a solution of **9** (0.50 g, 1.1 mmol) in DMF (6 mL) was added KSAc (0.37 g, 3.3 mmol) in portions. After stirring for 2 h at r.t., brine (10 mL) was added, followed by H₂O (300 mL). The product was extracted with EtOAc (3×40 mL). Evaporation of the solvent yielded a mixture of 2 diastereomers (dr = 8.5:1). Purification by flash chromatography (pentane–Et₂O, 2:1) yielded 0.39 g (78%) of the major diastereomer as colourless crystals; mp 117 °C; $R_f = 0.30$ (pentane–Et₂O, 2:1).

IR (CHCl₃): 3027, 2919, 2852, 1717, 1691, 1549, 1452, 1437, 1366, 1272, 1232, 1133, 1110, 756, 702, 626 cm⁻¹.

¹H NMR (400 MHz, $CDCl_3$): $\delta = 1.45$ (m, 1 H), 1.65 (m, 2 H), 1.80 (m, 1 H), 2.29 (s, 3 H), 2.82 (m, 2 H), 2.93 (dd, J = 10.3, 3.4 Hz, 1 H), 3.33 (m, 1 H), 3.64 (s, 3 H), 4.51 (s, 1 H), 4.87 (dd, J = 3.4, 2.2 Hz, 1 H), 6.98 (m, 2 H), 7.22–7.41 (m, 8 H), 7.45 (d, J = 2.2 Hz, 1 H).

 ^{13}C NMR (100 MHz, CDCl₃): δ = 26.5, 28.8, 30.6, 31.6, 36.1, 42.5, 44.8, 51.9, 92.6, 127.2, 127.6, 127.7, 127.8, 128.8, 129.0, 137.1, 139.5, 143.4, 165.8, 195.3.

MS (EI): m/z = 453.1 (M⁺).

Anal. Calcd for $C_{25}H_{27}NO_5S$: C, 66.20; H, 6.00; N, 3.09. Found: C, 65.87; H, 6.06; N, 3.02.

Thiadecaline Methyl Ester 11

The synthesis of **11** following GP 2 yielded 0.12 g (62%) of colourless crystals; mp 244 °C; $R_f = 0.16$ (pentane–Et₂O, 3:1).

IR (CHCl₃): 1726, 1548, 1453, 1436, 1366, 1279, 1237, 1197, 1171, 909, 831, 760, 733, 699 cm⁻¹.

¹H NMR (400 MHz, CDCl₃): δ = 1.82 (m, 1 H), 1.93 (m, 2 H), 2.2 (m, 1 H), 2.36 (m, 1 H), 2.61 (m, 1 H), 2.79 (m, 1 H), 3.33 (dd, *J* = 12.1, 11.9 Hz, 1 H), 3.46 (s, 3 H), 3.59 (dd, *J* = 6.4, 1.5 Hz, 1 H), 3.69 (dd, *J* = 11.9, 4.3 Hz, 1 H), 4.13 (dd, *J* = 12.3, 12.1 Hz, 1 H), 5.33 (dd, *J* = 12.3, 6.3 Hz, 1 H), 7.20–7.46 (m, 10 H).

 ^{13}C NMR (100 MHz, CDCl₃): δ = 23.6, 28.0, 28.3, 37.7, 43.1, 45.7, 50.4, 51.9, 53.0, 87.3, 127.5, 128.0, 128.1, 128.9, 129.0, 138.0, 138.3, 172.7.

MS (CI): $m/z = 440.1 (M + C_2H_5^+), 412.1 (M + H^+).$

Anal. Calcd for $C_{23}H_{25}NO_4S$: C, 67.13; H, 6.12; N, 3.40. Found: C, 66.89; H, 6.12; N, 3.27.

S-(5-Oxopentyl) Ethanthioate 12 (n = 1)

Compound **4** (2.7 g, 16 mmol) was dissolved in DMF (30 mL) at 0 °C. After the addition of KSAc (4.5 g, 39 mmol), the solution was stirred for 30 min. H₂O (500 mL) was added and the mixture was extracted with Et₂O (3×50 mL). The solvent was evaporated and the crude **12** (n = 1) was purified by flash chromatography (pentane–Et₂O, 3:1) to give 1.76 g (67%) of a colourless oil; $R_f = 0.32$ (pentane–Et₂O, 2:1).

Synthesis 2010, No. 13, 2271–2277 © Thieme Stuttgart · New York

IR (CHCl₃): 2934, 2861, 1723, 1690, 1355, 1134, 958, 627 cm⁻¹.

¹H NMR (400 MHz, CDCl₃): δ = 1.66 (m, 4 H), 2.33 (s, 3 H), 2.47 (ddd, *J* = 7.2, 7.2, 1.5 Hz, 2 H), 2.88 (dd, *J* = 6.9, 6.9 Hz, 2 H), 9.77 (dd, *J* = 1.5 Hz, 1 H).

¹³C NMR (100 MHz, CDCl₃): δ = 21.0, 28.6, 29.0, 30.6, 43.3, 195.7, 201.9.

Anal. Calcd for $C_7H_{12}OS$: C, 52.47; H, 7.55. Found: C, 52.20; H, 7.56.

Carbaldehyde 13a

The synthesis of **13a** following GP 1 yielded 0.44 g (51%) of a colourless solid; mp 163 °C; ee >99%; $R_f = 0.40$ (pentane–Et₂O, 1:1); $[\alpha]_D^{23}$ –9.9 (*c* 1, CHCl₃).

IR (CHCl₃): 1681, 1654, 1451, 1359, 1163, 1139, 1110, 953, 774, 754, 700 cm⁻¹.

¹H NMR (400 MHz, $CDCl_3$): $\delta = 1.50$ (m, 1 H), 1.70 (m, 2 H), 1.82 (m, 1 H), 2.31 (s, 3 H), 2.82 (m, 2 H), 3.00 (dd, J = 10.6, 3.4 Hz, 1 H), 3.37 (m, 1 H), 4.46 (s, 1 H), 4.87 (dd, J = 3.4, 1.8 Hz, 1 H), 6.98 (m, 2 H), 7.19 (m, 2 H), 7.41–7.26 (m, 7 H), 9.60 (s, 1 H).

 ^{13}C NMR (100 MHz, CDCl₃): δ = 26.6, 28.7, 30.6, 31.3, 36.7, 42.9, 43.2, 92.3, 127.6, 127.7, 127.8, 128.0, 129.0, 136.8, 137.5, 138.4, 153.3, 191.6, 195.3.

MS (EI): m/z = 423.4 (M⁺).

Anal. Calcd for $C_{24}H_{25}NO_4S;\,C,\,68.06;\,H,\,5.95;\,N,\,3.31.$ Found: C, $68.15;\,H,\,5.97;\,N,\,3.29.$

Carbaldehyde 13b

The synthesis of **13b** following GP 1 yielded 0.935 g (45%) of a yellow solid; mp 132 °C; ee >99%; $R_f = 0.45$ (pentane–Et₂O, 1:1); $[\alpha]_D^{23}$ –13.6 (*c* 1, CHCl₃).

IR (CHCl₃): 2947, 1677, 1544, 1362, 1136, 706 cm⁻¹.

¹H NMR (400 MHz, CDCl₃): δ = 1.48 (ddt, *J* = 4.2, 6.7, 9.6 Hz, 1 H) 1.68 (m, 2 H), 1.81 (m, 1 H), 2.32 (m, 3 H), 2.83 (m, 2 H), 2.98 (dd, *J* = 3.4, 10.5 Hz, 1 H), 3.32 (m, 1 H), 4.46 (m, 1 H), 4.82 (dd, *J* = 1.8 Hz, 3.4 Hz, 1 H), 7.19 (m, 10 H), 9.59 (s, 1 H).

 ^{13}C NMR (100 MHz, CDCl₃): δ = 26.63, 28.77, 30.67, 31.34, 36.77, 42.83, 42.90, 92.11, 127.76–129.16, 133.99, 135.41, 137.55, 138.29, 152.85, 191.48, 195.37.

MS (CI): m/z = 458.2.

Anal. Calcd for $C_{24}H_{24}CINO_4S$: C, 62.94; H, 5.28; N, 3.06. Found: C, 63.04; H, 5.25; N, 3.06.

Carbaldehyde 13c

The synthesis of **13c** following GP 1 yielded 4.28 g (43%) of the major diastereomer as a yellow solid; ee >99%; $R_f = 0.27$ (pentane–Et₂O, 1:1); $[\alpha]_D^{23}$ -4.5 (*c* 1, CHCl₃).

IR (CHCl₃): 1684, 1546, 1512, 1452, 1362, 1305, 1250, 1181, 1132, 1109, 1031, 954, 837, 759, 702 cm⁻¹.

¹H NMR (400 MHz, CDCl₃): $\delta = 1.50$ (m, 1 H), 1.69 (m, 2 H), 1.82 (m, 1 H), 2.32 (s, 3 H), 2.83 (m, 2 H), 2.95 (dd, J = 10.4, 3.5 Hz, 1 H), 3.30 (m, 1 H), 3.77 (s, 3 H), 4.44 (s, 1 H), 4.84 (dd, J = 3.5, 1.8 Hz, 1 H), 6.80 (m, 2 H), 6.90 (m, 2 H), 7.18 (m, 2 H), 7.40–7.27 (m, 4 H), 9.59 (s, 1 H).

 ^{13}C NMR (100 MHz, CDCl₃): δ = 26.6, 28.8, 30.6, 31.4, 37.0, 42.5, 42.8, 55.1, 92.4, 114.3, 127.6, 127.7, 128.8, 129.0, 137.5, 138.5, 153.4, 159.1, 191.6, 195.3.

MS (EI): m/z = 453.3 (M⁺).

Anal. Calcd for $C_{25}H_{27}NO_5S$: C, 66.20; H, 6.00; N, 3.09. Found: C, 66.01; H, 5.96; N, 3.07.

Carbaldehyde 13d

The synthesis of **13d** following GP 1 yielded 2.49 g (60%) of a palered oil; ee >99%; $R_f = 0.35$ (pentane–Et₂O, 2:1); $[\alpha]_D^{23}$ 24.1 (*c* 1, CHCl₃).

IR (film): 3360, 2920, 1689, 1551, 1451, 1362, 1218, 1135, 757, 625 $\rm cm^{-1}$

¹H NMR (400 MHz, CDCl₃): δ = 1.59 (m, 1 H) 1.72 (m, 1 H), 1.85 (m, 2 H), 2.19 (m, 3 H), 2.34 (m, 3 H), 2.90 (m, 2 H), 3.09 (dd, *J* = 3.5, 10.1 Hz, 1 H), 3.20 (m, 1 H), 4.49 (s, 1 H), 5.01 (dd, *J* = 2.2, 3.4 Hz, 1 H), 5.86 (dd, *J* = 1.0, 3.1 Hz, 1 H), 5.98 (m, 1 H), 7.28 (m, 5 H), 9.55 (s, 1 H).

¹³C NMR (100 MHz, CDCl₃): δ = 13.6, 26.9, 28.9, 30.8, 32.0, 36.9, 37.6, 42.5, 89.8, 106.4, 108.4, 127.9, 128.0, 129.2, 137.5, 138.2, 148.2, 152.0, 152.7, 191.6, 195.5.

MS (CI): $m/z = 428 (M + H^+)$.

HRMS (EI): m/z calcd for $C_{23}H_{25}NO_5S$: 427.1453. Found: 427.1436.

Carbaldehyde 13e

The synthesis of **13e** following GP 1 yielded 2.40 g (57%) of the major diastereomer as a yellow solid; mp 42 °C; ee >99%; $R_f = 0.29$ (pentane–Et₂O, 2:1); $[\alpha]_D^{23}$ –5.0 (*c* 1, CHCl₃).

IR (CHCl₃): 1684, 1546, 1365, 1131, 767, 699 cm⁻¹.

¹H NMR (300 MHz, CDCl₃): δ = 1.74 (m, 1 H), 1.87 (m, 1 H), 2.33 (s, 3 H), 3.01 (m, 3 H), 3.47 (m, 1 H), 4.48 (s, 1 H), 4.87 (dd, *J* = 1.7, 3.3 Hz, 1 H), 7.21 (m, 11 H), 9.63 (s, 1 H).

¹³C NMR (100 MHz, CDCl₃): δ = 26.0, 30.6, 32.4, 36.3, 43.2, 43.0, 92.4, 127.8–129.2, 136.7, 137.7, 138.5, 152.8, 191.8, 195.2.

Anal. Calcd for $C_{23}H_{23}NO_4S$: C, 67.46; H, 5.66; N, 3.42. Found: C, 67.31; H, 5.87; N, 3.24.

Carbaldehyde 13f

The synthesis of **13f** following GP 1 yielded 1.45 g (32%) of the major diastereomer as a colourless oil; ee >99%; $R_f = 0.27$ (pentane–Et₂O, 2:1); $[\alpha]_D^{23}$ –156.0 (*c* 1, CHCl₃).

IR (CHCl₃): 3025, 2921, 2852, 1682, 1548, 1371, 1134, 757, 627 $\rm cm^{-1}.$

¹H NMR (400 MHz, $CDCl_3$): $\delta = 1.57$ (m, 2 H), 1.73 (m, 2 H), 2.32 (m, 3 H), 2.82 (m, 4 H), 3.12 (m, 1 H), 3.32 (dd, J = 4.0, 6.3 Hz, 1 H), 4.89 (dt, J = 4.0, 5.8 Hz, 1 H), 6.99 (td, J = 1.6, 3.2 Hz, 1 H), 7.06 (m, 2 H), 7.32 (m, 3 H), 9.59 (s, 1 H).

¹³C NMR (100 MHz, CDCl₃): δ = 24.1, 26.9, 28.9, 30.6, 32.6, 39.1, 47.2, 83.4, 127.5, 128.1, 128.9, 136.6, 137.0, 151.5, 192.0, 195.4.

MS (EI): m/z = 347 (M⁺).

HRMS (EI): m/z calcd for $C_{18}H_{21}NO_4S$: 347.1191; found: 347.1187.

Thiadecaline 14a

The synthesis of **14a** following GP 2 yielded 0.05 g (57%) of colourless needles; mp 251 °C; ee >99%; $R_f = 0.42$ (pentane–Et₂O, 1:1); $[\alpha]_D^{23}$ –151.1 (*c* 1, CHCl₃).

IR (CHCl₃): 2930, 1720, 1544, 1496, 1452, 1367, 1333, 913, 828, 758, 736, 698 cm⁻¹.

¹H NMR (400 MHz, $CDCl_3$): $\delta = 1.82$ (m, 1 H), 1.92 (m, 2 H), 2.19 (m, 1 H), 2.48 (m, 1 H), 2.56 (m, 1 H), 2.83 (m, 1 H), 3.42 (ddd, J = 11.5, 10.8, 2.5 Hz, 1 H), 3.64 (dd, J = 10.8, 3.9 Hz, 1 H), 3.70 (dd, J = 6.9, 3.0 Hz, 1 H), 4.18 (dd, J = 12.1, 11.5 Hz, 1 H), 5.41 (dd, J = 12.1, 6.9 Hz, 1 H), 7.40–7.21 (m, 10 H), 9.68 (d, J = 2.5 Hz, 1 H).

 ^{13}C NMR (100 MHz, CDCl₃): δ = 24.6, 26.8, 28.2, 36.1, 42.2, 42.8, 51.4, 53.4, 87.8, 127.8, 127.9, 128.0, 128.7, 128.8, 128.9, 137.7, 200.5.

MS (EI): m/z = 382.2 (M⁺).

Anal. Calcd for $C_{22}H_{23}NO_3S$: C, 69.26; H, 6.08; N, 3.67. Found: C, 69.50; H, 5.98; N, 3.66.

Thiadecaline 14b

The synthesis of **14b** following GP 2 yielded 0.04 g (40%) of colourless crystals, mp 126 °C; ee >99%; $R_f = 0.40$ (pentane–Et₂O 1:1); $[\alpha]_D^{23}$ –55.0 (*c* 1, CHCl₃).

IR (CHCl₃): 2926, 1720, 1546, 1369, 912, 835, 728, 700 cm⁻¹.

¹H NMR (300 MHz, CDCl₃): δ = 1.88 (m, 3 H), 2.17 (m, 1 H), 2.52 (m, 2 H), 2.82 (ddd, *J* = 3.0, 10.3, 13.5 Hz, 1 H), 3.39 (ddd, *J* = 2.4, 11.3, 10.3 Hz, 1 H), 3.58 (dd, *J* = 4.0, 10.3 Hz, 1 H), 3.70 (dd, *J* = 3.6, 7.0 Hz, 1 H), 4.11 (dd, *J* = 12.2, 11.3 Hz, 1 H), 5.41 (dd, *J* = 7.0, 12.2 Hz, 1 H), 7.29 (m, 9 H), 9.68 (d, *J* = 2.3 Hz, 1 H).

¹³C NMR (100 MHz, CDCl₃): δ = 24.8, 26.5, 28.2, 36.2, 41.9, 42.6, 50.4, 53.5, 87.8, 128.0–130.2, 134.2, 136.3, 137.8, 200.8.

HRMS (EI): m/z calcd for $C_{22}H_{22}CINO_3S$: 369.1080 (M⁺ – NO₂); found: 369.1065 (M⁺ – NO₂).

Thiadecaline 14c

The synthesis of **14c** following GP 2 yielded 0.51 g (52%) of the major diastereomer as colourless needles; mp 156 °C; ee >99%; $R_f = 0.29$ (pentane–Et₂O, 1:1); $[\alpha]_D^{23}$ –130.7 (*c* 1, CHCl₃).

IR (CHCl₃): 2917, 1717, 1609, 1548, 1511, 1455, 1368, 1250, 1180, 1109, 1032, 835, 812, 762, 700 cm⁻¹.

¹H NMR (300 MHz, CDCl₃): δ = 1.81 (m, 1 H), 1.89 (m, 2 H), 2.17 (m, 1 H), 2.48 (m, 1 H), 2.52 (m, 1 H), 2.81 (ddd, *J* = 13.7, 11.3, 3.0 Hz, 1 H), 3.4 (ddd, *J* = 12.0, 11.0, 2.6 Hz, 1 H), 3.62 (m, 1 H), 3.64 (m, 1 H), 3.79 (s, 3 H), 4.13 (dd, *J* = 12.2, 12.0 Hz, 1 H), 5.38 (dd, *J* = 12.2, 6.9 Hz, 1 H), 6.87 (m, 2 H), 7.32–7.20 (m, 7 H), 9.66 (d, *J* = 2.6 Hz, 1 H).

 ^{13}C NMR (100 MHz, CDCl₃): δ = 24.6, 26.8, 28.1, 36.2, 42.3, 42.7, 50.7, 53.4, 55.1, 87.9, 114.1, 127.9, 128.9, 129.7, 129.8, 137.8, 159.0, 200.6.

MS (EI): m/z = 411.2 (M⁺).

HRMS (EI): m/z calcd for C₂₃H₂₅NO₄S: 411.1499, found: 411.1506.

Thiadecaline 14d

The synthesis of **14d** following GP 2 yielded 0.045 g (50%) of a colourless solid; mp 140 °C; ee >99%; $R_f = 0.31$ (pentane–Et₂O, 2:1); $[\alpha]_D^{23}$ –153.0 (*c* 1, CHCl₃).

IR (CHCl₃): 2931, 1719, 1548, 1367, 1031, 791, 699 cm⁻¹.

¹H NMR (400 MHz, CDCl₃): δ = 1.88 (m, 3 H), 2.22 (m, 1 H), 2.33 (s, 3 H), 2.40 (m, 1 H), 2.66 (m, 1 H), 2.81 (m, 1 H), 3.37 (dt, *J* = 3.5, 12.0 Hz, 1 H), 3.58 (m, 1 H), 3.80 (dd, *J* = 4.2, 12.2 Hz, 1 H), 4.04 (t, *J* = 12.0 Hz, 1 H), 5.21 (dd, *J* = 5.6, 12.3 Hz, 1 H), 5.90 (m, 1 H), 6.07 (d, *J* = 3.1 Hz, 1 H), 7.25 (m, 5 H), 9.56 (d, *J* = 3.5 Hz, 1 H).

 ^{13}C NMR (100 MHz, CDCl₃): δ = 13.8, 23.3, 27.3, 27.4, 35.9, 42.1, 43.7, 46.3, 52.9, 87.2, 106.3, 110.2, 128.0, 128.3, 129.0, 137.3, 149.2, 152.8, 200.4.

MS (CI): $m/z = 386 (M + H^{+})$.

HRMS (EI): m/z calcd for $C_{21}H_{23}NO_4S$: 339.1419 (M - NO_2^+); found: 339.1417 (M - NO_2^+).

Hexahydrobenzothiophene Derivative 14e

The synthesis of **14e** following GP 2 yielded 0.055 g (57%) of the major diastereomer as a yellow solid; mp 166 °C; ee >99%; $R_f = 0.3$ (pentane–Et₂O, 1:1); $[\alpha]_D^{23}$ –139.0 (*c* 1, CHCl₃).

IR (CHCl₃): 1722, 1542, 1372, 767, 746, 703 cm⁻¹.

¹H NMR (300 MHz, $CDCl_3$): $\delta = 9.46$ (s, 1 H, J = 1.9 Hz), 7.29 (m, 10 H), 5.25 (dd, J = 6.0, 9.3 Hz, 1 H), 4.13 (dd, J = 7.4, 11.2 Hz, 1 H), 3.74 (dd, J = 9.3, 12.3 Hz, 1 H), 3.69 (t, J = 6.3, 6.0 Hz, 1 H), 3.22 (ddd, J = 1.9, 11.2, 12.3 Hz, 1 H), 3.09 (m, 1 H), 2.94 (ddd, J = 3.6, 7.1, 10.7 Hz, 1 H), 2.86 (m, 1 H), 2.37 (dtd, J = 3.7, 6.1, 9.7 Hz, 1 H), 2.10 (m, 1 H).

¹³C NMR (100 MHz, CDCl₃): δ = 29.9, 35.8, 43.9, 44.6, 45.0, 46.6, 57.9, 92.2, 127.6, 128.2, 128.3, 129.0, 129.3, 137.2, 137.4, 201.1.

MS (EI): m/z = 367 (M⁺).

HRMS (EI): m/z calcd for $C_{21}H_{21}NO_3S$: 321.1313 (M - NO₂⁺), found: 321.1297(M - NO₂⁺).

Thiadecaline 14f

The synthesis of **14f** following GP 2 yielded 0.032 g (40%) of the major diastereomer as a colourless oil; ee >99%; $R_f = 0.31$ (pentane–Et₂O, 1:1); $[\alpha]_D^{23}$ –178.0 (*c* 1, CHCl₃).

IR (film): 2935, 2854, 1724, 1684, 1547, 1448, 1375, 910, 763, 731, 700 $\rm cm^{-1}$

¹H NMR (400 MHz, CDCl₃): δ = 1.43 (m, 2 H), 1.59 (m, 2 H), 2.23 (m, 1 H), 2.52 (m, 2 H), 2.73 (m, 2 H), 2.87 (m, 1 H), 3.84 (m, 1 H), 4.20 (dd, *J* = 11.5, 11.5 Hz, 1 H), 4.74 (dt, *J* = 4.7, 11.5 Hz, 1 H), 7.30 (m, 5 H), 9.70 (s, 1 H).

 ^{13}C NMR (100 MHz, CDCl₃): δ = 20.5, 25.9, 28.2, 29.5, 39.6, 42.2, 43.9, 51.6, 91.0, 127.7, 128.9, 137.0, 198.5.

HRMS (EI): m/z calcd for $C_{16}H_{19}NO_3S$: 305.1086; found: 305.1087.

Acknowledgment

This work was supported by the Deutsche Forschungsgemeinschaft (priority program Organocatalysis) and the Fonds der Chemischen Industrie (Kekulé stipend to N.E.). We thank the former Degussa AG and BASF AG for the donation of chemicals. The collection of the X-ray data of **14e** by Dr. Dix, Bruker AXS GmbH, Karlsruhe is gratefully acknowledged.

References

(1) (a) Berkessel, A.; Gröger, H. Asymmetric Organocatalysis; Wiley-VCH: Weinheim, 2005. (b) Seayad, J.; List, B. Org. Biomol. Chem. 2005, 3, 719. (c) Marigo, M.; Jørgensen, K. A. Chem. Commun. 2006, 2001. (d) List, B. Chem. Commun. 2006, 819. (e) Lelais, G.; MacMillan, D. W. C. Aldrichimica Acta 2006, 39, 79. (f) List, B. Chem. Rev. 2007, 107, 5413. (g) de Figueiredo, R. M.; Christmann, M. Eur. J. Org. Chem. 2007, 2575. (h) Dalko, P. I. Enantioselective Organocatalysis. Reactions and Experimental Procedures; Wiley-VCH: Weinheim, 2007. (i) Enders, D.; Niemeier, O.; Henseler, A. Chem. Rev. 2007, 107, 5606. (j) Enders, D.; Balensiefer, T. Acc. Chem. Res. 2004, 37, 534. (k) Dondoni, A.; Massi, A. Angew. Chem. Int. Ed. 2008, 47, 4638; Angew. Chem. 2008, 120, 4716. (l) Melchiorre, P.; Marigo, M.; Carlone, A.; Bartoli, G. Angew. Chem. Int. Ed. 2008, 47, 6138; Angew. Chem. 2008, 120, 6232. (m) Enders, D.; Narine, A. A. J. Org. Chem. 2008, 73, 7857. (n) Buckley, B. Ann. Rep. Prog. Chem., Sect. B: Org. Chem. 2009, 105, 113. (o) Bella, M.; Gasperi, T. Synthesis 2009, 1583. (p) Bertelsen, S.; Jørgensen, K. A. *Chem. Soc. Rev.* **2009**, *38*, 2178. (q) Enders, D.; Wang, C.; Liebich, J. X. *Chem. Eur. J.* **2009**, *15*, 11058. (r) Merino, P.; Marqués-Lopez, E.; Tejero, T.; Herrera, R. P. *Synthesis* **2010**, 1.

- (2) (a) Klimenko, S. K.; Stolbova, T. V.; Kulikova, L. K.; Shub, G. M. *Pharm. Chem. J.* 2001, *35*, 661. (b) Klimenko, S. K.; Stolbova, T. V.; Kulikova, L. K.; Shub, G. M. *Pharm. Chem. J.* 2001, *35*, 370. (c) Klimenko, S. K.; Stolbova, T. V.; Kulikova, L. K. *Pharm. Chem. J.* 2001, *35*, 22.
- (3) Klimenko, S. K.; Stolbova, T. V.; Makarov, V. V. *Pharm. Chem. J.* **2002**, *36*, 583.
- (4) Scheidges, C.; Ottow, E.; Neef, G.; Beier, S.; Elger, W. German Patent DE 3820948 A1 19891221, 1989; *Chem. Abstr.* 1989, 112, 198891.
- (5) (a) Oberlander, C.; Piazza, P. V. PCT Int. Appl. WO 9826783 A1 19980625, **1998**; *Chem. Abstr.* **1998**, *129*, 50105. (b) Petit, F.; Philibert, D.; Ulmann, A. Eur. Pat. Appl. EP 676203 A1 19951011, **1995**; *Chem. Abstr.* **1995**, *124*, 22540.
- (6) (a) Tietze, L. F. Chem. Rev. 1996, 96, 115. (b) Tietze, L. F.; Brasche, G.; Gericke, K. Domino Reactions in Organic Synthesis; Wiley-VCH: Weinheim, 2006. (c) Pellissier, H. Tetrahedron 2006, 62, 1619. (d) Pellissier, H. Tetrahedron 2006, 62, 2143. (e) Nicolaou, K. C.; Edmonds, D. J.; Bulger, P. G. Angew. Chem. Int. Ed. 2006, 45, 7134; Angew. Chem. 2006, 118, 7292. (f) Chapman, C. J.; Frost, C. G. Synthesis 2007, 1. (g) Davies, H.; Sorensen, E. Chem. Soc. Rev. 2009, 38, 2969.
- (7) (a) Enders, D.; Grondal, C.; Hüttl, M. R. M. Angew. Chem. Int. Ed. 2007, 46, 1570; Angew. Chem. 2007, 119, 1590. (b) Yu, X.; Wang, W. Org. Biomol. Chem. 2008, 6, 2037. (c) Halland, N.; Aburel, P. S.; Jørgensen, K. A. Angew. Chem. Int. Ed. 2004, 43, 1272; Angew. Chem. 2004, 116, 1292. (d) Yang, J. W.; Fonseca, M. T. H.; List, B. J. Am. Chem. Soc. 2005, 127, 15036. (e) Huang, Y.; Walji, A. M.; Larsen, C. H.; MacMillan, D. W. C. J. Am. Chem. Soc. 2005, 127, 15051. (f) Marigo, M.; Schulte, T.; Franzén, J.; Jørgensen, K. A. J. Am. Chem. Soc. 2005, 127, 15710. (g) Enders, D.; Hüttl, M. R. M.; Runsink, J.; Raabe, G.; Wendt, B. Angew. Chem. Int. Ed. 2007, 46, 467; Angew. Chem. 2007, 119, 471. (h) Enders, D.; Narine, A. A.; Benninghaus, T. R.; Raabe, G. Synlett 2007, 1667. (i) Carlone, A.; Cabrera, S.; Marigo, M.; Jørgensen, K. A. Angew. Chem. Int. Ed. 2007, 46, 1101; Angew. Chem. 2007, 119, 1119. (j) Hayashi, Y.; Okano, T.; Aratake, S.; Hazelard, D. Angew. Chem. Int. Ed. 2007, 46, 4922; Angew. Chem. 2007, 119, 5010. (k) Zhao, G.-L.; Rios, R.; Vesley, J.; Eriksson, L.; Córdova, A. Angew. Chem. Int. Ed. 2008, 47, 8468; Angew. Chem. 2008, 120, 5896. (1) Enders, D.; Hüttl, M. R. M.; Raabe, G.; Bats, J. W. Adv. Synth. Catal. 2008, 350, 267. (m) Enders, D.; Wang, C.; Bats, J. W. Angew. Chem. Int. Ed. 2008, 47, 7539; Angew. Chem. 2008, 120, 7649. (n) Kotame, P.; Hong, B.-C.; Liao, J.-H. Tetrahedron Lett. 2009, 50, 704. (o) Zhang, F.-L.; Xu, A.-W.; Gong, Y.-F.; Wei, M.-H.; Yang, X.-L. Chem. Eur. J. 2009, 15, 6815. (p) Enders, D.; Krüll, R.; Bettray, W. Synthesis 2010, 567. (q) Rueping, M.; Kuenkel, A.; Tato, F.; Bats, J. W. Angew. Chem. Int. Ed. 2009, 48, 3699; Angew. Chem. 2009, 121, 3754. (r) Reyes, E.; Talavera, G.; Vicario, J. L.; Badia, D.; Carrillo, L. Angew. Chem. Int. Ed. 2009, 48, 5701; Angew. Chem. 2009, 121, 5811. (s) Zhu, D.; Lu, M.; Dai, L.; Zhong, G. Angew. Chem. Int. Ed. 2009, 48, 6089; Angew. Chem. 2009, 121, 6205. (t) Wu, L.-Y.; Bencivenni, G.; Mancinelli, M.; Mazzanti, A.; Bartoli, G.; Melchiorre, P. Angew. Chem. Int. Ed. 2009, 48, 7196; Angew. Chem. 2009, 121, 7332. (u) Alba, A.-N.; Companyo, X.; Viciano, M.; Rios, R. Curr. Org. Chem. 2009, 13, 1432. (v) Enders, D.; Wang, C.; Bats,

J. W. Synlett **2009**, 1777. (w) Enders, D.; Wang, C.; Raabe, G. Synthesis **2009**, 4119. (x) Franzén, J.; Fisher, A. Angew. Chem. Int. Ed. **2008**, 48, 787; Angew. Chem. **2008**, 121, 801. (y) Enders, D.; Jeanty, M.; Bats, J. W. Synlett **2009**, 3175. (z) Hong, L.; Sun, W.; Liu, C.; Wang, L.; Wang, R. Chem. Eur. J **2010**, 15, 440. (aa)Grondal, C.; Jeanty, M.; Enders, D. Nat. Chem. **2010**, 2, 167.

- (8) Enders, D.; Hüttl, M. R. M.; Grondal, C.; Raabe, G. *Nature* 2006, 441, 861.
- (9) (a) Enders, D.; Lüttgen, K.; Narine, A. A. Synthesis 2007, 959. (b) Jørgensen, K. A.; Brandau, S.; Maerten, E. J. Am.

Chem. Soc. **2006**, *128*, 14986. (c) Rios, R.; Sundén, H.; Ibrahem, I.; Zhao, G.-L.; Córdova, A. *Tetrahedron Lett.* **2006**, *47*, 8679. (d) Wang, W.; Li, H.; Wang, J.; Zu, L. *J. Am. Chem. Soc.* **2006**, *128*, 10354. (e) Li, H.; Zu, L.; Xie, H.; Wang, J.; Jiang, W.; Wang, W. *Org. Lett.* **2007**, *9*, 1833.

- (10) Marigo, M.; Wabnitz, T.; Fielenbach, D.; Jørgensen, K. A. Angew. Chem. Int. Ed. 2005, 44, 794; Angew. Chem. 2005, 117, 804.
- (11) Enders, D.; Scherer, H. J.; Runsink, J. Chem. Ber. 1993, 126, 1929.