

Scheme 2 Diversity generating cleavage of the 5-thiotetrazole linker by metal catalysis. (a) PhMgBr (0.1 M), CuBr (0.06 M), THF, 50 °C, 5 h, 29%; (b) 1-Me indole (0.1 M), AgOTf (0.1 M), CH₂Cl₂, rt, 3 h, 23%; (c) PhSO₂Na (0.1 M), Pd(PPh₃)₄, THF–MeOH (2 : 1), 70 °C, 8 h, 38%; (yields were calculated after gravimetric analysis and are based on the initial loading of **1**).

side (Scheme 2).^{9a} Interestingly, the use of Pd(0) led to replacement with stereochemical net retention (double inversion) of the 5-thiotetrazole moiety if phenylsulfonates were employed,^{9b,c,16} resulting in the synthesis of *cis*-2,3-dehydrodecalins. This metal-dependent type of resin release significantly extends the scope of the linker system since it allows a diversity-generating cleavage of the polymer-supported decalin derivatives. All release reactions could be carried out under mild reaction conditions and gave the desired release products **13–15** in good to moderate overall yields of 23–38% for three steps, *i.e.* 61–67% average yield per step.

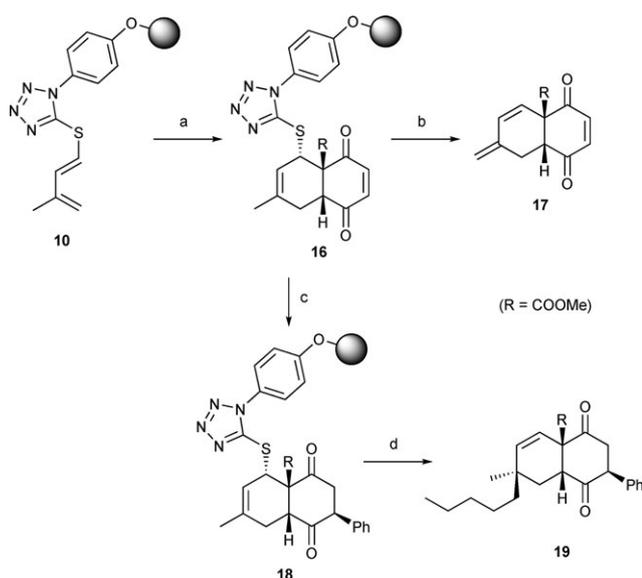
In order to further extend the scope of the reaction sequence, we attempted to carry out the Diels–Alder reaction in an enantioselective fashion and several chiral ligand–metal Lewis acid combinations and 2-methoxycarbonylquinone¹⁵ as a dienophile were employed (Scheme 3). The highest enantioselectivity (75% ee) and purity (>90%) for **16** was obtained if Gd(OTf)₃ and (*R,R*)-*i*PrPybox as a chiral ligand were used.¹⁷ Due to the presence of the α,β -unsaturated carbonyl system, the subsequent release of the 5-thiotetrazole system represented a formidable challenge. The use of Ag(I) salts in the absence of a nucleophile yielded olefin **17**.¹⁸ Alternatively, a cleavage with PhMgBr–CuBr in a two step process which consisted of a controlled Michael addition to the α,β -unsaturated Michael system to yield **18** followed by a traceless release with C₅H₁₁MgBr–CuBr in THF at 50 °C yielded the multi-substituted 1,2-dehydrodecalin derivative **19** in a 9% overall yield for five steps (*i.e.* 63% average yield per step) and 75% ee.

In summary, we have developed a traceless linker based on a 5-thiotetrazole moiety placed in an allylic position that can be easily cleaved with metal-assisted nucleophilic reactions employing Pd(0), Cu(I) and Ag(I) salts. The methodology proved suitable to perform even an enantioselective construction of the decalin scaffold on polymeric support.

Scheme 1 Synthesis of *cis*-decalins on solid support. (a) **3** (1.1 eq.), K₂CO₃ (1 eq.), DMF, 80 °C, 20 h, 97%; (b) 20% TFA–CH₂Cl₂, rt, 16 h, 93%; (c) TsCl (1.3 eq.), NEt₃ (2 eq.), DMAP (0.3 eq.), CH₂Cl₂, 0 °C, 4 h, 98%; (d) **9** (0.9 eq.), K₂CO₃ (1 eq.), KI (0.3 eq.), DMF, 80 °C, 6 h, 93% (based on **9**); (e) Pd(PPh₃)₄ (0.2 eq.), *N*-methyl morpholine (3 eq.), THF, rt, 1.5 h, 96%; (f) Rink amide or aminomethylated resin, **8** (0.1 M), HATU (0.1 M), DIEA (0.15 M), DMF, rt, 12 h; (g) methacrolein (0.3 M), LiHMDS (0.5 M), THF, rt, 16 h; (h) 2-methoxycarbonyl cyclohexenone **11** (0.1 M), Sc(OTf)₃ (0.02 M), CH₂Cl₂, 0 °C, 5 h.

LiHMDS as a base.¹⁴ The subsequent Diels–Alder reaction with 2-methoxycarbonyl cyclohexenone **11**¹⁵ as the dienophile in the presence of Sc(OTf)₃ afforded the desired polymer-supported *cis*-1,2-dehydrodecalin derivative **12** with full conversion estimated by NMR spectroscopy after TFA release from resin.

Next, we tested several metals for catalyzing the nucleophilic release of the *cis*-1,2-dehydrodecalin derivative **12** from the resin. Release of the 5-thiotetrazole moiety as originally anticipated could be achieved with Ag(I) and Cu(I) salts by direct S_N2' displacement by *exo*-attack from the least hindered



Scheme 3 Enantioselective Diels–Alder reaction and traceless cleavage from solid support. (a) 2-methoxycarbonylquinone, $\text{Gd}(\text{OTf})_3$, (*R,R*)-*i*PrPybox, CH_2Cl_2 , -78°C , 3.5 h; (b) $\text{Ag}(\text{OTf})$, CH_2Cl_2 , rt, 5 h, 15%; (c) PhMgBr (0.15 M), CuBr (0.09 M), THF, rt, 5 h; (d) $\text{C}_5\text{H}_{11}\text{MgBr}$ (0.1 M), CuBr (0.06 M), THF, 50°C , 3 h, 9% (75% ee); (yields were calculated after gravimetric analysis and are based on the initial loading of 1).

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Notes and references

- (a) D. R. Spring, *Chem. Soc. Rev.*, 2005, **34**, 472–482; (b) K. Hinterding, D. Alonso-Díaz and H. Waldmann, *Angew. Chem., Int. Ed.*, 1998, **37**, 688–749.
- A. Nören-Müller, I. Reis-Corrêa, H. Prinz, C. Rosenbaum, K. Saxena, H. J. Schwalbe, D. Vestweber, G. Cagna, S. Schunk, O. Schwarz, H. Schiewe and H. Waldmann, *Proc. Natl. Acad. Sci. U. S. A.*, 2006, **103**, 10606–10611.
- (a) P. Stahl, L. Kissau, R. Mazitschek, A. Giannis and H. Waldmann, *Angew. Chem., Int. Ed.*, 2002, **41**, 1174–1178; (b) M. A. Koch, A. Schuffenhauer, M. Scheck, S. Wetzel, M. Casaulta, A. Odermatt, P. Ertl and H. Waldmann, *Proc. Natl. Acad. Sci. U. S. A.*, 2005, **102**, 17272–17277; (c) L. Kissau, P. Stahl, R. Mazitschek, A. Giannis and H. Waldmann, *J. Med. Chem.*, 2003, **46**, 2917–2931; (d) P. Stahl, L. Kissau, R. Mazitschek, A. Huwe, P. Furet, A. Giannis and H. Waldmann, *J. Am. Chem. Soc.*, 2001, **123**, 11586–11593.
- (a) A. Ganesan, *Curr. Opin. Chem. Biol.*, 2008, **12**, 306–317; (b) R. Breinbauer, I. R. Vetter and H. Waldmann, *Angew. Chem., Int. Ed.*, 2002, **41**, 2879–2890.
- For examples of natural product-based compound collections from our laboratories, see: (a) A. Nören-Müller, W. Wilk, K. Saxena, H. Schwalbe, M. Kaiser and H. Waldmann, *Angew. Chem., Int. Ed.*, 2008, **47**, 5973–5977; (b) A. B. Garcia, T. Lessmann, J. D. Umarye, V. Mamane, S. Sommer and H. Waldmann, *Chem. Commun.*, 2006, 3868–3870; (c) M. A. Koch, L.-O. Wittenberg, S. Basu, D. A. Jeyaraj, E. Gourzoulidou, K. Reinecke, A. Odermatt and H. Waldmann, *Proc. Natl. Acad. Sci. U. S. A.*, 2004, **101**, 16721–16726; (d) O. Barun, S. Sommer and H. Waldmann, *Angew. Chem., Int. Ed.*, 2004, **43**, 3195–3199; (e) B. Meseguer, D. Alonso-Díaz, N. Griebenow, T. Herget and H. Waldmann, *Angew. Chem., Int. Ed.*, 1999, **38**, 2902–2906; (f) B. Sauerbrei, V. Jungmann and H. Waldmann, *Angew. Chem., Int. Ed.*, 1998, **37**, 1143–1146; (g) H. Waldmann, V. Khedkar, H. Dückert, M. Schürmann, I. M. Oppel and K. Kumar, *Angew. Chem., Int. Ed.*, 2008, **47**, 6869–6872; (h) review: K. Kumar and H. Waldmann, *Angew. Chem., Int. Ed.*, 2009, **48**, DOI: 10.1002/anie.200803437.
- (a) T. Leßmann and H. Waldmann, *Chem. Commun.*, 2006, 3380–3389; (b) I. Paterson and T. Temal-Laiieb, *Org. Lett.*, 2002, **4**, 2473–2476.
- For examples of traceless release from solid phase see: (a) C. Gil and S. Bräse, *Curr. Opin. Chem. Biol.*, 2004, **8**, 230–237; (b) F. Stieber, U. Grether and H. Waldmann, *Angew. Chem., Int. Ed.*, 1999, **38**, 1073–1077; (c) T. Takahashi, S. Tomida, H. Inoue and T. Doi, *Synlett*, 1998, 1261–1263; (d) K. C. Nicolaou, J. A. Pfefferkorn and G.-Q. Cao, *Angew. Chem., Int. Ed.*, 2000, **39**, 734–739; (e) W.-X. Gu, S.-X. Liu and R. B. Silverman, *Org. Lett.*, 2002, **4**, 4171–4174; (f) S. Bräse, *Acc. Chem. Res.*, 2004, **37**, 805–816; (g) review: F. Guillier, D. Orain and M. Bradley, *Chem. Rev.*, 2000, **100**, 2091–2157; (h) Diels–Alder reactions on solid support: W. C. Cheng, M. M. Olmstead and M. J. Kurt, *J. Org. Chem.*, 2001, **66**, 5528–5533; (i) L. Blanco, R. Bloch, E. Bugnet and S. Deloisy, *Tetrahedron Lett.*, 2000, **41**, 7875–7878; (j) D. Craig, M. J. Robson and S. J. Shaw, *Synlett*, 1998, 1381–1383.
- P. R. Blakemore, W. J. Cole, P. J. Kocienski and A. Morley, *Synlett*, 1998, 26–28.
- (a) K. Takeda, K. Tsuboyama, K. Torii, M. Murata and H. Ogura, *Tetrahedron Lett.*, 1988, **29**, 4105–4108; (b) K. Takeda, K. Torii and H. Ogura, *Tetrahedron Lett.*, 1990, **31**, 265–266; (c) H.-J. Gais, T. Jagusch, N. Spalthoff, F. Gerhards, M. Frank and G. Raabe, *Chem.–Eur. J.*, 2003, **9**, 4202–4221.
- (a) T. Anthonsen, M. S. Henderson, A. Martin, R. D. H. Murray, R. McCrindle and D. McMaster, *Can. J. Chem.*, 1973, **51**, 1332–1345; (b) M. S. Henderson, R. McCrindle and D. McMaster, *Can. J. Chem.*, 1973, **51**, 1346–1358; (c) S. Manabe and C. Nishino, *Tetrahedron*, 1986, **42**, 3461–3470.
- I. Kitagawa, M. Yoshihara, T. Tani and I. Yosioka, *Tetrahedron Lett.*, 1975, **16**, 23–26.
- I. Kitagawa, M. Yoshihara and T. Kamiguchi, *Tetrahedron Lett.*, 1977, **18**, 1221–1224.
- (a) H. Morita, S. Tashiro, M. Takeda, K. Fujimori, N. Yamada, Md. C. Shiekh and H. Kawaguchi, *Tetrahedron*, 2008, **64**, 3589–3595; (b) P. Mauleon, I. Alonso, M. R. Rodriguez and J. C. Carretero, *J. Org. Chem.*, 2007, **72**, 9924–9935.
- The use of NaHMDS or KHMDS resulted in *E/Z* mixtures and incomplete conversion.
- D. Liotta, C. Barnum, R. Puleo, G. Zima, C. Bayer and H. S. Kezar III, *J. Org. Chem.*, 1981, **46**, 2920–2923.
- For a previous example of traceless release from solid support employing a $\text{Pd}(0)$ -mediated allylic substitution, see: S. C. Schürer and S. Blechert, *Synlett*, 1998, 166–168.
- D. A. Evans and J. Wu, *J. Am. Chem. Soc.*, 2003, **125**, 10162–10163.
- The regiochemistry of **17** was assigned by NMR spectroscopy by comparison with known compounds, see e. g.: B. Witte, L. Meyer and P. Margaretha, *Helv. Chim. Acta*, 2000, **83**, 554–561.