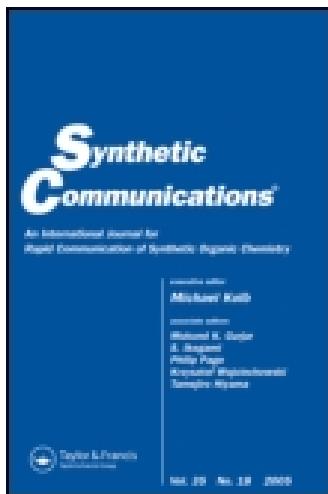


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SYNTHESIS OF 6-BENZYLIMIDAZO[2,1-*b*][1,3]THIAZOLE DURING SONOGASHIRA COUPLING

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*The reaction of 2-amino-3-(2-propynyl)-1,3-thiazolium bromide 2 with various iodobenzenes 3a-f in the presence of palladium catalyst leads to the formation of 6-benzylimidazo[2,1-*b*][1,3]thiazoles 4a-f.*

Keywords: Aryl iodide; imidazothiazole; Pd-catalyzed; Sonogashira coupling

INTRODUCTION

The imidazo[2,1-*b*]thiazole derivatives have been shown to possess antibacterial, antifungal, anthelmintic, antiviral, antiprotozoal, anti-inflammatory, anticonvulsant, anxiolytic, hypnotic, gastrointestinal, antiulcer, and immunomodulatory activities.^[1-5] Compounds containing the imidazo[2,1-*b*][1,3]thiazole skeleton have been used as anthelmintic agents, antihypertensives, anti-inflammatories, immunosuppressive agents, fungicides, herbicides, antitumor agents, and cardiotonic agents.^[6]

Considering the potent bioactivities of compounds possessing an imidazothiazole core, the development of a new strategy to synthesize 6-substituted imidazo[2,1-*b*][1,3]thiazoles efficiently attracted our attention.

Although several procedures have been developed for the synthesis of imidazo[2,1-*b*][1,3]thiazoles,^[7] no examples involving arylation of imidazothiazole by Pd-Cu-catalyzed (Sonogashira coupling) reactions have been reported in the literature.

The Sonogashira reaction (i.e., the palladium and copper cocatalyzed coupling of terminal alkynes with aryl and vinyl halides) is one of the most widely used C-C bond formation reactions.^[8] It provides an efficient route to aryl alkynes, which are interesting intermediates for the preparation of a variety of target compounds with applications ranging from natural products^[9] and pharmaceuticals^[10] to molecular organic materials.^[11] Because of the utility of the products, the development of new catalyst systems has received considerable attention. Palladium-catalyzed reactions have been immensely practical for both carboannulation^[12] and heteroannulation^[13]

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processes. In continuation of our recent studies^[14] on the synthesis of fused heterocycles and the Pd-catalyzed reaction of acetylenes leading to heterocyclic compounds of biological significance, we became interested in developing a synthetic route to 6-substituted imidazo[2,1-*b*][1,3]thiazoles.

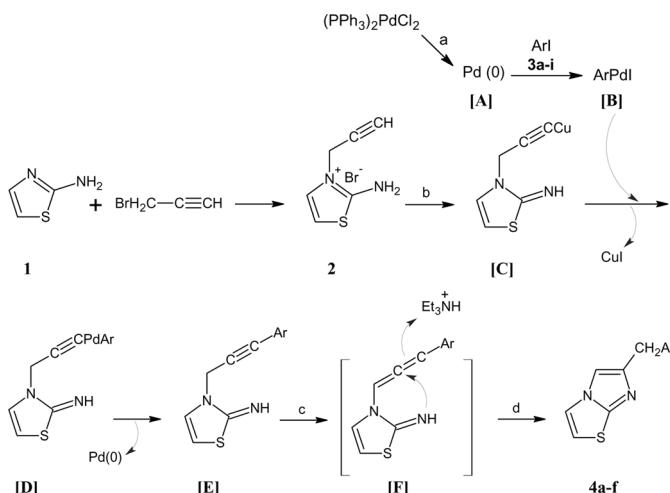
RESULTS AND DISCUSSION

In this article, we report that treatment of 2-amino-1,3-thiazole **1** with propargyl bromide in refluxing acetonitrile affords 2-amino-3-(2-propynyl)-1,3-thiazolium bromide **2** in good yield. The ¹H NMR spectrum of **2** showed a CH proton at 3.73 ppm, CH₂ protons at 5.03 ppm, and a single resonance for the NH₂ group at 9.78 ppm. This signal was removed on deuteration.

Compound **2** was treated in dimethylformamide (DMF) with aryl iodides **3a-f** and triethylamine in the presence of bis(triphenylphosphine)palladium(II) chloride and cuprous iodide at room temperature to give 6-substituted imidazo[2,1-*b*][1,3]thiazoles **4a-f** in moderate to high yields (Scheme 1, Table 1). The reactions were carried out under an argon atmosphere, and DMF and triethylamine were degassed prior to use.

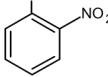
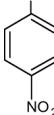
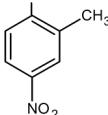
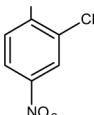
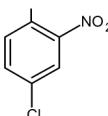
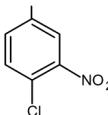
Mechanistically, the formation of 6-benzylimidazo[2,1-*b*][1,3]thiazoles involves the following steps (as shown in Scheme 1): (i) formation of ArPdI **[B]** through oxidative addition of Pd(0) **[A]** to ArI^[15]; (ii) transmetalation of ArPdI with the Cu salt of **[C]**, generating the alkynyl palladium species **[D]**; (iii) extrusion of Pd(0) to yield the alkynes **[E]**; and (iv) isomerization to the allenic intermediates^[16] **[F]** which then cyclize to products **4a-f**.

In conclusion, we have developed a successful palladium-catalyzed reaction for the synthesis of 6-benzylimidazo[2,1-*b*][1,3]thiazoles from readily available starting materials.



Scheme 1. Plausible mechanism for the formation of 6-benzylimidazo[2,1-*b*][1,3] thiazoles at room temperature. Reagents and conditions: (a) reduction of Pd(II) to Pd(0) with alkyne and Et₃N; (b) CuI, Et₃N; (c) isomerization to an allene with CuI, Et₃N; (d) nucleophilic attack on the allene (F) to generate the 2-benzylimidazo[2,1-*b*][1,3]thiazoles **4a-f**.

Table 1. Melting points and yields of 6-benzylimidazo[2,1-*b*][1,3]thiazoles **4a–f**

Product	Ar	Mp (°C)	Yield (%)
4a		224–225	72
4b		229–230	87
4c		254–255	85
4d		209–210	90
4e		231–232	95
4f		243–244	86

EXPERIMENTAL

Synthesis of 2-Amino-3-(2-propynyl)-1,3-thiazolium Bromide **2**

A mixture of 2-aminothiazole **1** (2 g, 20 mmol) and propargyl bromide (2 mL, 24 mmol) in acetonitrile (10 mL) was heated under reflux for 1 h. The precipitate formed was filtered off and recrystallized from acetonitrile to afford the title compound. Yield 95%; mp 159–160°C; ¹H NMR (500 MHz, DMSO-*d*₆): δ = 3.73 (t, *J* = 2.4 Hz, 1H, CH), 5.03 (d, *J* = 2.3 Hz, 2H, CH₂), 7.07 (d, *J* = 4.5 Hz, 1H, CH of thiazole), 7.50 (d, *J* = 4.5 Hz, 1H, CH of thiazole), 9.78 (s, 2H, NH₂); IR, ν (KBr disc): 3300, 3250, 2150 cm⁻¹; MS *m/z* 139 (M⁺). Anal. calcd. for C₆H₇BrN₃S: C, 32.89; H, 3.22; N, 12.79; S, 14.63. Found: C, 32.65; H, 3.03; N, 12.55; S, 14.47.

Synthesis of 6-Substituted Imidazo[2,1-*b*][1,3]thiazoles **4a–f**

A mixture of the aryl iodide **3a–f** (0.75 mmol), (PPh₃)₂PdCl₂ (0.05 mmol), CuI (0.1 mmol), and triethylamine (3 mmol) was stirred in DMF (5 mL) at room temperature under an argon atmosphere. 2-Amino-3-(2-propynyl)-1,3-thiazolium bromide **2** (1.27 mmol, 0.28 g) was then added, and the mixture was stirred at room temperature

for 18 h. After completion of the reaction, the resulting solution was concentrated in vacuo, and the crude product was subjected to silica-gel column chromatography using $\text{CHCl}_3\text{-CH}_3\text{OH}$ (95:5) as the eluent to afford the pure product (Table 1).

6-(2-Nitrobenzyl)imidazo[2,1-*b*][1,3]thiazole 4a

^1H NMR (500 MHz, DMSO- d_6): δ = 4.25 (s, 2H, CH_2), 6.95–7.96 (m, 6H, thiazol & ArH), 8.63 (s, 1H, CH of imidazole); IR, ν (KBr disc): 1520, 1340 cm^{-1} , MS m/z 259 (M^+). Anal. calcd. for $\text{C}_{12}\text{H}_9\text{N}_3\text{O}_2\text{S}$: C, 55.59; H, 3.50; N, 16.21; S, 12.37. Found: C, 55.32; H, 3.36; N, 16.40; S, 12.21.

6-(4-Nitrobenzyl)imidazo[2,1-*b*][1,3]thiazole 4b

^1H NMR (500 MHz, DMSO- d_6): δ = 4.18 (s, 2H, CH_2), 7.25–8.15 (m, 6H, thiazol & ArH), 8.31 (s, 1H, CH of imidazole); IR, ν (KBr disc): 1525, 1340 cm^{-1} , MS m/z 259 (M^+). Anal. calcd. for $\text{C}_{12}\text{H}_9\text{N}_3\text{O}_2\text{S}$: C, 55.59; H, 3.50; N, 16.21; S, 12.37. Found: C, 55.37; H, 3.32; N, 16.45; S, 12.18.

6-(2-Methyl-4-nitrobenzyl)imidazo[2,1-*b*][1,3]thiazole 4c

^1H NMR (500 MHz, DMSO- d_6): δ = 2.50 (s, 3H, CH_3), 4.10 (s, 2H, CH_2), 7.22–8.06 (m, 5H, thiazol & ArH), 8.14 (s, 1H, CH of imidazole); IR, ν (KBr disc): 1530, 1345 cm^{-1} , MS m/z 273 (M^+). Anal. calcd. for $\text{C}_{13}\text{H}_{11}\text{N}_3\text{O}_2\text{S}$: C, 57.13; H, 4.06; N, 15.37; S, 11.73. Found: C, 57.31; H, 3.90; N, 15.51; S, 11.61.

6-(2-Chloro-4-nitrobenzyl)imidazo[2,1-*b*][1,3]thiazole 4d

^1H NMR (500 MHz, DMSO- d_6): δ = 4.28 (s, 2H, CH_2), 7.24–8.28 (m, 5H, thiazol & ArH), 8.32 (s, 1H, CH of imidazole); IR, ν (KBr disc): 1525, 1340 cm^{-1} , MS m/z 293 (M^+), 295 ($\text{M} + 2$). Anal. calcd. for $\text{C}_{12}\text{H}_8\text{ClN}_3\text{O}_2\text{S}$: C, 49.07; H, 2.75; N, 14.31; S, 10.92. Found: C, 49.29; H, 2.87; N, 14.20; S, 11.02.

6-(4-Chloro-2-nitrobenzyl)imidazo[2,1-*b*][1,3]thiazole 4e

^1H NMR (500 MHz, DMSO- d_6): δ = 4.34 (s, 2H, CH_2), 7.26–7.93 (m, 5H, thiazol & ArH), 8.08 (s, 1H, CH of imidazole); IR, ν (KBr disc): 1520, 1340 cm^{-1} , MS m/z 293 (M^+), 295 ($\text{M} + 2$). Anal. calcd. for $\text{C}_{12}\text{H}_8\text{ClN}_3\text{O}_2\text{S}$: C, 49.07; H, 2.75; N, 14.31; S, 10.92. Found: C, 48.88; H, 2.62; N, 14.47; S, 10.77.

6-(4-Chloro-3-nitrobenzyl)imidazo[2,1-*b*][1,3]thiazole 4f

^1H NMR (500 MHz, DMSO- d_6): δ = 4.13 (s, 2H, CH_2), 7.25–7.95 (m, 5H, thiazol & ArH), 8.02 (s, 1H, CH of imidazole); IR, ν (KBr disc): 1530, 1350 cm^{-1} , MS m/z 293 (M^+), 295 ($\text{M} + 2$). Anal. calcd. for $\text{C}_{12}\text{H}_8\text{ClN}_3\text{O}_2\text{S}$: C, 49.07; H, 2.75; N, 14.31; S, 10.92. Found: C, 49.23; H, 2.90; N, 14.45; S, 10.81.

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