Generation and Reactivity of 1,2-Cyclohexadiene under Mild Reaction Conditions

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Dedicated to Professor Luis Castedo on the occasion of his 70th birthday

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An improved method to generate 1,2-cyclohexadienes by fluoride-induced β -elimination of 6-(trimethylsilyl)cyclohexenyl triflates is presented. This method allows the generation of these highly strained cyclic allenes in good yields and un-

der mild reaction conditions, as demonstrated by trapping experiments based on cycloaddition reactions. (© Wiley-VCH Verlag GmbH & Co. KGaA, 69451 Weinheim, Germany, 2009)

Introduction

In the last few decades fluoride-induced elimination of β -substituted organosilanes has emerged as a powerful method for the generation of unstable short-lived intermediates.^[1,2] For example, (*o*-bromophenyl)trimethylsilane (1, Scheme 1) has been used as a benzyne precursor and avoids the use of strong bases.^[3a] The key advance in the development of this methodology was the introduction of *o*-(trimethylsilyl)phenyl triflate (2), as the better leaving group character of the triflate allows the generation of benzyne in higher yields and under milder conditions.^[3b] This approach has led to a remarkable revival in the use of benzyne (3) over the past few years^[4,2d] and has also been used for the efficient generation of other strained molecules such as cyclohexyne^[5] and 1,2,3-cyclohexatriene.^[6a]



Scheme 1. Fluoride-induced generation of benzyne (3) and 1,2-cy-clohexadiene (6).

Cyclic allenes containing fewer than eight carbon atoms are strained molecules that are closely related to arynes and small cycloalkynes.^[1,2] These intermediates are potent elec-

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trophiles and undergo pericyclic reactions such as [2+2] and [4+2] cycloadditions with alkenes and dienes, respectively. 1,2-Cyclohexadiene (6, Scheme 1) is a highly reactive cyclic allene.^[7] The two most general methods for the generation of this molecule have for many years involved the base-induced elimination of HX from alkenyl halides and the decomposition of gem-dihalocyclopropanes by organolithium reagents (Doering-Moore-Skattebøl reaction).^[1] Both methods are hampered by the use of strong bases and highly nucleophilic reagents, and this limits their scope. Fluoride-induced decomposition of compound 4 to generate 6 was introduced by Shakespeare and Johnson in 1990,^[6a] albeit since then, this method has been sporadically used.^[8] Herein, we introduce 6-(trimethylsilyl)cyclohexenyl triflate (5), which allows the generation of 1,2-cyclohexadiene (6) in good yields and under mild reaction conditions as a result of the excellent leaving group character of the triflate group.

Results and Discussion

Vinyl triflate **5** was prepared from 2-(trimethylsilyl)cyclohexenone **7** (Scheme 2).^[6] Hydride conjugate addition to this enone with the use of L-Selectride followed by aqueous workup afforded cyclohexanone **8a**. Treatment of this compound with LDA at -78 °C and trapping of the resulting kinetic enolate with *N*-phenyltrifluoromethanesulfonimide led to cycloallene precursor **5** in 48% yield. Remarkably, alkyl-substituted cycloallene precursors such as **9** can be prepared in good yield by minor modifications of this procedure.

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Scheme 2. Synthesis of 1,2-cyclohexadiene precursors. Reagents and conditions: (a) i. L-Selectride, ii. NH_4Cl (aq.) for **8a** or MeI for **8b**; (b) i. LDA ii. PhNTf₂.

The ability of triflate **5** to form cycloallene **6** was confirmed when treatment of **5** with tetrabutylammonium fluoride (TBAF) in THF at room temperature afforded 1,2cyclohexadiene dimer (**10**) in 78% yield through a [2+2] cycloaddition of intermediate **6** (Scheme 3).^[9]



Scheme 3. Dimerization of 1,2-cyclohexadiene (6).

Compound **6** was also trapped in Diels–Alder reactions. The generation of cyclic allene **6** in the presence of furan (**11**) led to the isolation of a 95:5 mixture of the *endo-* and *exo-* [4+2] cycloaddition adducts **12**, respectively, in 53% yield (Scheme 4).^[10] In the presence of 1,3-diphenylisobenzofuran (**13**) a mixture of *endo-* and *exo-***14** was obtained in 66% yield (ratio *endo/exo* = 80:20). It should be mentioned that the generation of **6** by treatment of 1-bromocyclohexene with KOtBu in the presence of **13** afforded *endo/exo-***14** with the same ratio but in 38% yield.^[7a] The structure of *endo-***14** was confirmed by X-ray diffraction studies (Figure 1).



Scheme 4. [4+2] Cycloaddition reactions of 1,2-cyclohexadiene.

These results prompted us to study the reactivity of cycloallene **6** in the presence of other unsaturated compounds as reaction partners. In this context, the reaction with tropone (**15**) would be of special interest due to the vast number of possible peri-, regio- and stereoisomers. It is known that the reaction of benzyne with tropone affords periselectively the [4+2] cycloaddition adduct as the major product, together with small amounts of the [6+2] cycloaddition product.^[11] By contrast, the reaction of tropone with acyclic allenes usually leads to mixtures of [8+2] and [4+2] cycloaddition adducts.^[12] Treatment of triflate **5** with CsF



Figure 1. X-ray structure of compound endo-14.

in the presence of tropone (15) selectively afforded a 96:4 mixture of the *endo*- and *exo*- [4+2] cycloaddition adducts 16, respectively, in good yield (75%, Scheme 5). The relative stereochemistry of compounds *endo*- and *exo*-16 was established on the basis of 2D NMR spectra (COSY and NOESY).^[13] In a similar way to the reaction with benzyne,^[11b] [6+2] cycloaddition product 17 was also isolated in 13% yield.



Scheme 5. Cycloadditions of 1,2-cyclohexadiene with tropone.

Analogously to arynes, cyclic allenes can be stabilized by complexation with a transition metal.^[14] This opens the possibility of employing cyclic allenes in transition-metalcatalyzed reactions.^[15] We therefore decided to study the reactivity of 1,2-cyclohexadiene (**6**) in the presence of catalytic amounts of a palladium complex. In particular, treatment of triflate **5** with CsF in the presence of dimethyl acetylenedicarboxylate (**18**) and 5 mol-% of Pd(PPh₃)₄ in MeCN at room temperature resulted in the formation of compounds **19** and **20** in 10 and 19% yield, respectively (Scheme 6). In the absence of the palladium catalyst these products were not detected by GC–MS, whereas in the absence of fluoride the starting material was untouched. The structure of compound **20** was unambiguously determined by single-crystal X-ray diffraction studies (Figure 2).



Scheme 6. Pd-promoted reaction of 1,2-cyclohexadiene with dimethyl acetylenedicarboxylate.

Presumably, a Pd^0 complex promotes the [2+2+2] cycloaddition of one molecule of cycloallene 6 with two molecules of alkyne 18 to give intermediate 21 (Scheme 7). Sub-



Figure 2. X-ray structure of compound 20.

sequent fluoride-induced deprotonation/rearomatization would give anion 22, which could evolve in two ways: either by protonation to afford tetrahydronaphthalene 19 or by conjugate addition to alkyne 18 followed by protonation to form compound 20. Alternatively, this compound could be produced through an ene reaction of intermediate 21 and alkyne 18. Remarkably, this is the first example of the participation of strained cyclic allenes in a metal-catalyzed reaction.



Scheme 7. Proposed mechanism for the formation of 19 and 20.

We next planned to confirm the ability of triflate **9** to form 1-methyl-1,2-cyclohexadiene (**23**, Scheme 8). Indeed, treatment of triflate **9** with CsF in the presence of diphenylisobenzofuran (**13**) afforded a mixture of three Diels–Alder adducts: *endolexo*-**24** in 56% yield (ratio *endolexo* = 68:32), resulting from the reaction of the methyl-substituted double bond of the allene with diene **13**,^[16] and *endo*-**25**, resulting from the reaction of the less-substituted double bond of cycloallene **23**. Remarkably, the other possible [4+2] cycloaddition product *exo*-**25** was not detected in the reaction mixture. As expected, the corresponding dimer of cyclic allene **23** was obtained in the absence of diene **13**.^[17]



Scheme 8. [4+2] Cycloaddition reaction of cyclic allene 23.

The stereochemistry of compound *endo-24* was confirmed by X-ray diffraction studies (Figure 3).



Figure 3. X-ray structure of compound endo-24.

Conclusions

1,2-Cyclohexadienes can be generated and trapped under mild reaction conditions by means of fluoride-induced β elimination of the corresponding 6-(trimethylsilyl)cyclohexenyl triflates. This method allows the generation of highly strained cyclohexallenes with higher efficiencies than classical approaches. Significantly, cyclic allenes generated by this method can participate not only in [4+2] cycloadditions with furan or tropone but also in palladium-promoted [2+2+2] cycloadditions with alkynes.

Experimental Section

General: All reactions were performed under an argon atmosphere by using oven-dried glassware and standard Schlenk techniques. Solvents were dried by distillation from a drying agent before use: MeCN from CaH₂, THF and toluene from Na. TLC was performed on Merck silica gel 60 F254; chromatograms were visualized with UV light (254 and 360 nm). Flash column chromatography was performed on Merck silica gel 60 (ASTM 230-400 mesh). ¹H and ¹³C NMR spectra were recorded at 250.13 and 62.83 MHz (Bruker DPX-250 instrument), 300 and 75 MHz (Varian Mercury-300 instrument) or 400 and 100 MHz (Varian Mercury-400 instrument), respectively. NOE, NOESY and COSY were recorded with a Bruker WM-500 instrument. Low-resolution mass spectra [LRMS (electron impact, EI)] were determined at 70 eV with an HP-5988A instrument. High-resolution mass spectra (HRMS) were obtained with a Micromass Autospec spectrometer. IR spectra were recorded with a Mattson Cygnus 100 spectrophotometer. GC-MS analyses were performed with an Agilent 6890N/5973inert instrument with a HP-5MS capillary column. Compound 7 was prepared from 2-cyclohexenone following a previously published procedure.^[6] Commercial reagents were used without further purification. CCDC-729036 (for endo-14), -729037 (for 20) and -729038 (for endo-24) contain the supplementary crystallographic data for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/ data_request/cif.

Synthesis of 1,2-Cyclohexadiene Precursors

2-(Trimethylsilyl)cyclohexanone (8a): L-Selectride (1.0 M in THF, 3.60 mL, 3.60 mmol) was added dropwise to a solution of **7** (600 mg, 3.57 mmol) in THF (18 mL) at -78 °C. The mixture was stirred under an argon atmosphere at this temperature for 2 h. Then, saturated aqueous NH₄Cl solution (45 mL) and Et₂O (30 mL) were added, the phases were separated and the aqueous

layer was extracted with Et₂O (2×40 mL). The combined organic layer was dried with Na₂SO₄, filtered and concentrated under reduced pressure. The residue was purified by column chromatography (SiO₂; Et₂O/hexane, 1:20) to afford **8a** (450 mg, 74%) as a colourless oil: ¹H NMR (400 MHz, CDCl₃): $\delta = 2.36$ (m, 1 H), 2.26–2.11 (m, 2 H), 2.01–1.92 (m, 3 H), 1.75–1.60 (m, 3 H), 0.10 (s, 9 H) ppm. ¹³C NMR (100 MHz, CDCl₃): $\delta = 212.5$ (C), 45.4 (CH), 41.3 (CH₂), 26.5 (CH₂), 25.3 (CH₂), 23.4 (CH₂), -1.6 (3 CH₃) ppm. MS (EI): *m/z* (%) = 170 (98), 155 (100). HRMS: calcd for C₉H₁₈OSi 170.1127; found 170.1120.

6-(Trimethylsilyl)-1-cyclohexenyl 1-Trifluoromethanesulfonate (5): nBuLi (2.25 M in hexane, 1.65 mL, 3.63 mmol) was added dropwise to *i*Pr₂NH (0.50 mL, 3.80 mmol) at -78 °C, leading to a gel that was dissolved in THF (6 mL). The mixture was stirred at -78 °C for 10 min and at room temperature for 20 min. Then, a solution of 8a (588 mg, 3.46 mmol) in THF (5 mL) was added at -78 °C. After stirring at -78 °C for 35 min, a solution of PhNTf₂ (1.30 g, 3.63 mmol) in THF (5 mL) was added, and the mixture was stirred at room temperature for 12 h. Then, 5% aqueous NaHCO3 solution (35 mL) was added, the phases were separated and the aqueous layer was extracted with Et_2O (2 × 35 mL). The combined organic layer was dried with Na₂SO₄, filtered and concentrated under reduced pressure. The residue was purified by column chromatography (SiO₂, pentane) to afford 5 (500 mg, 48%) as a colourless oil. ¹H NMR (400 MHz, CDCl₃): δ = 5.65 (ddd, J = 5.0, 3.6, 1.7 Hz, 1 H), 2.26–2.02 (m, 2 H), 2.00–1.90 (m, 2 H), 1.70–1.40 (m, 3 H), 0.10 (s, 9 H) ppm. ¹³C NMR (100 MHz, CDCl₃): δ = 152.8 (C), 118.6 (J = 320 Hz, CF₃) 115.2 (CH), 28.9 (CH), 25.5 (CH₂), 24.1 (CH₂), 21.0 (CH₂), -2.1 (3 CH₃) ppm. IR (CsI): $\tilde{v} = 2925$ cm⁻¹. MS (EI): m/z (%) = 302 (2), 79 (100). HRMS: calcd. for C₁₀H₁₇F₃O₃SiS 302.0620; found 302.0620.

2-Methyl-2-(trimethylsilyl)cyclohexanone (8b): L-Selectride (1.0 M in THF, 2.86 mL, 2.86 mmol) was added dropwise to a solution of 7 (400 mg, 2.38 mmol) in THF (12 mL) at -78 °C, and the mixture was stirred at this temperature for 2 h. Then, MeI (0.770 mL, 11.9 mmol) was added, and the mixture was stirred at room temperature for 5 h. Cold saturated NH₄Cl solution (30 mL) was added, the phases were separated and the aqueous layer was extracted with Et_2O (3×20 mL). The combined organic layer was dried with Na₂SO₄, filtered and concentrated under reduced pressure. The residue was purified by column chromatography (SiO₂; Et₂O/hexane, 1:20) to afford **8b** (430 mg, 98%) as a colourless oil. ¹H NMR (500 MHz, CDCl₃): $\delta = 2.44-1.89$ (m, 4 H), 1.73-1.64 (m, 3 H), 1.53 (m, 1 H), 1.13 (s, 3 H), 0.07 (s, 9 H) ppm. ¹³C NMR $(125 \text{ MHz}, \text{CDCl}_3): \delta = 214.7 \text{ (C)}, 44.1 \text{ (C)}, 40.9 \text{ (CH}_2), 35.7 \text{ (CH}_2),$ 25.5 (CH₂), 22.5 (CH₂), 20.8 (CH₃), -2.4 (3CH₃) ppm. IR (CsI): ṽ = 1680 cm⁻¹. MS (EI): m/z (%) = 184 (42), 169 (100). HRMS: calcd. for C₁₀H₂₀OSi 184.1283; found 184.1282.

6-Methyl-6-(trimethylsilyl)-1-cyclohexenyl 1-Trifluoromethanesulfonate (9): *n*BuLi (2.25 M in hexane, 1.01 mL, 2.28 mmol) was added dropwise to *i*Pr₂NH (0.31 mL, 2.39 mmol) at -78 °C, leading to a gel that was dissolved in THF (4 mL). The mixture was stirred at -78 °C for 10 min and at room temperature for 20 min. Then, a solution of **8b** (400 mg, 2.17 mmol) in THF (4 mL) was added at -78 °C. After stirring at -78 °C for 40 min, a solution of PhNTf₂ (820 mg, 2.28 mmol) in THF (5 mL) was added, and the mixture was stirred at room temperature for 18 h. Then, 5% aqueous NaHCO₃ solution (30 mL) was added, the phases were separated and the aqueous layer was dried with Et₂O (2 × 25 mL). The combined organic layer was dried with Na₂SO₄, filtered and concentrated under reduced pressure. The residue was purified by column chromatography (SiO₂, hexane) to afford **9** (350 mg, 51%) as a colourless oil. ¹H NMR (300 MHz, CDCl₃): δ = 5.68 (dd, J = 5.4, 2.9 Hz, 1 H), 2.28–1.98 (m, 2 H), 1.76–1.45 (m, 4 H), 1.18 (s, 3 H) 0.06 (s, 9 H) ppm. ¹³C NMR (75 MHz, CDCl₃): δ = 156.8 (C), 118.3 (J = 320 Hz, CF₃), 114.3 (CH), 33.3 (CH₂), 27.7 (C), 24.8 (CH₂), 19.5 (CH₃), 17.3 (CH₂), -3.5 (3 CH₃) ppm. MS (EI): m/z (%) = 183 (6) [M – Tf], 79 (100). HRMS: calcd. for C₁₀H₁₈OSi [M – Tf] 183.1160; found 183.1162.

Cycloaddition Reactions of 1,2-Cyclohexadienes

[2+2] Cycloaddition of 1,2-Cyclohexadiene (6): Tetrabutylammonium fluoride (TBAF; 1.0 M in THF, 280 µL, 0.28 mmol) was added dropwise to a solution of 5 (44 mg, 0.14 mmol) in MeCN (5 mL) at room temperature, and the mixture was stirred at this temperature for 1 h. Then, H₂O (5 mL) and Et₂O (5 mL) were added, the phases were separated and the aqueous layer was extracted with Et_2O (3×5 mL). The combined organic layer was dried with anhydrous Na₂SO₄, filtered and concentrated under reduced pressure. The residue was purified by column chromatography (SiO₂, hexane) to afford *trans*-1,2,3,6,7,8,8a,8b-octahydrobiphenylene $^{[9,18]}$ (10; 9 mg, 78%) as a colourless oil. $^1H~NMR$ (300 MHz, CDCl₃): δ = 5.39 (s, 2 H), 2.35–2.22 (m, 2 H), 2.16– 2.03 (m, 4 H), 2.01-1.90 (m, 2 H), 1.83-1.72 (m, 2 H), 1.52-1.30 (m, 2 H), 1.23–1.04 (m, 2 H) ppm. ¹³C NMR (75 MHz, CDCl₃): δ = 143.0 (C), 111.1 (CH), 46.8 (CH), 27.5 (CH₂), 24.5 (CH₂), 21.7 (CH_2) ppm. MS (EI): m/z (%) = 160 (73), 117 (100). HRMS: calcd. for C₁₂H₁₆ 160.1252; found 160.1246.

[4+2] Cycloaddition of 1,2-Cyclohexadiene (6) with Furan (11): TBAF (1.0 m in THF, 0.38 mL, 0.38 mmol) was added dropwise to a solution of 5 (97 mg, 0.32 mmol) in furan (11, 2.0 mL) at room temperature, and the mixture was stirred at this temperature for 24 h. Then, the mixture was concentrated under reduced pressure and the residue was purified by column chromatography (SiO₂, hexane) to afford a mixture of *endolexo*-1,4,4a,5,6,7-hexahydro-1,4epoxynaphthalene^[7d] (*endolexo*-12; 25 mg, 53%; 95:5 *endolexo*) and dimer 10 (4 mg, 15%). Data for *endo*-12: ¹H NMR (300 MHz, CDCl₃): δ = 6.34 (dd, J = 5.5, 1.0 Hz, 1 H), 6.02 (dd, J = 5.6, 1.1 Hz, 1 H), 5.57 (s, 1 H), 5.01 (s, 2 H), 2.34 (m, 1 H), 2.16 (m, 1 H), 1.99–1.74 (m, 3 H), 1.52 (m, 1 H), 0.35 (m, 1 H) ppm. MS (EI): *m/z* (%) = 148 (19), 91 (100).

[4+2] Cycloaddition of 1,2-Cyclohexadiene (6) with 1,3-Diphenylisobenzofuran (13): Finely powdered anhydrous CsF (76 mg, 0.50 mmol) was added to a solution of 5 (60 mg, 0.20 mmol) and 1,3-diphenylisobenzofuran (13; 81 mg, 0.30 mmol) in MeCN/THF (1:1, 2 mL), and the mixture was stirred at room temperature for 24 h. Then, the mixture was concentrated under reduced pressure and the residue was purified by column chromatography (SiO₂; hexane/CH₂Cl₂, 9:1) to afford endo/exo-9,10-diphenyl-1,2,3,9,9a,10-hexahydro-9,10-epoxyanthracene^[7a] (endo/exo-14; 46 mg, 66%; 80:20 endolexo) as a colourless oil. Data for endo-14: ¹H NMR (250 MHz, CDCl₃): δ = 7.91–7.82 (m, 2 H), 7.77–7.66 (m, 2 H), 7.53–7.33 (m, 6 H), 7.24–7.16 (m, 2 H), 7.15–7.07 (m, 2 H), 5.68 (dd, J = 3.3, 3.2 Hz, 1 H), 3.05 (m, 1 H), 1.69–1.52 (m, 2 H), 0.55–0.35 (m, 2 H), 1.82 (m, 1 H), 2.19 (m, 1 H) ppm. ¹³C NMR (62.8 MHz, CDCl₃): δ = 148.3 (C), 144.4 (C), 144.2 (C), 138.0 (C), 134.8 (C), 128.6 (2 CH), 128.5 (CH), 128.4 (2 CH), 128.2 (2 CH), 127.9 (CH), 127.0 (CH), 126.7 (2 CH), 125.8 (CH), 121.3 (CH), 120.2 (CH), 117.8 (CH), 90.2 (C), 89.8 (C), 48.2 (CH), 26.2 (CH₂), 24.9 (CH₂), 21.9 (CH₂) ppm. MS (CI): *m*/*z* (%) = 351 (40), 333 (100). Data for *exo*-14: ¹H NMR (250 MHz, CDCl₃): δ = 7.87– 7.61 (m, 4 H), 7.43 (m, 6 H), 7.31-7.22 (m, 2 H), 7.20-6.98 (m, 2 H), 5.62 (m, 1 H), 2.52 (m, 1 H), 2.11 (m, 1 H), 1.92 (m, 1 H), 1.82–1.64 (m, 2 H), 1.39 (m, 1 H), 0.80 (m, 1 H) ppm. MS (CI): m/z (%) = 351 (38), 333 (100).

Cycloaddition of 1,2-Cyclohexadiene (6) with Tropone (15): Finely powdered anhydrous CsF (74 mg, 0.48 mmol) was added to a solution of 5 (74 mg, 0.24 mmol) and tropone (15; 24 μ L, 0.24 mmol) in MeCN (2.5 mL), and the mixture was stirred at room temperature for 14 h. Then, the mixture was concentrated under reduced pressure and the residue was purified by column chromatography (SiO₂; AcOEt/hexane, 1:20) to afford endo/exo-16 (34 mg, 75%; 96:4 endolexo) and 17 (6.0 mg, 13%). Data for endo-16: ¹H NMR $(300 \text{ MHz}, \text{CDCl}_3): \delta = 6.75 \text{ (dd}, J = 11.1, 8.5 \text{ Hz}, 1 \text{ H}), 6.62 \text{ (ddd},$ J = 8.2, 7.2, 1.0 Hz, 1 H), 6.00 (ddd, J = 8.3, 7.1, 1.1 Hz, 1 H), 5.83-5.70 (m, 2 H), 3.94 (d, J = 6.6 Hz, 1 H), 3.25 (m, 1 H), 2.35(m, 1 H), 2.25–1.96 (m, 2 H), 1.91–1.75 (m, 2 H), 1.55 (m, 1 H), 1.08 (m, 1 H) ppm. ¹³C NMR (75 MHz, CDCl₃): δ = 193.0 (C), 148.7 (CH), 139.4 (CH), 132.3 (C), 130.4 (CH), 128.1 (CH), 124.0 (CH), 60.0 (CH), 41.9 (CH), 41.4 (CH), 28.9 (CH₂), 25.5 (CH₂), 22.5 (CH₂) ppm. MS (EI): m/z (%) = 186 (62), 131 (100). HRMS: calcd. for C₁₃H₁₄O 186.1044; found 186.1046. Data for 17: ¹H NMR (250 MHz, CDCl₃): δ = 6.01–5.83 (m, 2 H), 5.73 (m, 1 H), 5.56 (m, 1 H), 5.37 (m, 1 H), 3.39 (m, 1 H), 3.02–2.84 (m, 2 H), 2.17-2.03 (m, 2 H), 1.85-1.72 (m, 3 H), 0.88 (m, 1 H) ppm. ¹³C NMR (125 MHz, CDCl₃): δ = 214.0 (C), 144.1 (C), 128.9 (CH), 125.8 (2 CH), 125.6 (CH), 123.6 (CH), 57.0 (CH), 51.9 (CH), 51.7 (CH), 25.2 (CH₂), 22.0 (CH₂), 21.8 (CH₂) ppm. MS (EI): m/z (%) = 186 (70), 131 (100). HRMS: calcd. for $C_{13}H_{14}O$ 186.1044; found 186.1040.

Pd-Promoted Reaction of 1,2-Cyclohexadiene (6) with Dimethyl Acetylenedicarboxylate (18): Finely powdered anhydrous CsF (94 mg, 0.61 mmol) was added to a solution of 5 (93 mg, 0.31 mmol), Pd(PPh₃)₄ (18 mg, 0.015 mmol) and dimethyl acetylenedicarboxylate (DMAD, 18; 94 µL, 0.76 mmol) in MeCN (3 mL), and the mixture was stirred at room temperature for 22 h. Then, the mixture was concentrated under reduced pressure, and the residue was purified by column chromatography (SiO₂; Et₂O/ hexane, 1:9 to 8:2) to afford $19^{[19]}$ (11 mg, 10%) as a colourless oil and 20 (28 mg, 19%) as a white solid. Data for 19: ¹H NMR $(250 \text{ MHz}, \text{CDCl}_3)$: $\delta = 3.88 \text{ (s, 6 H)}, 3.83 \text{ (s, 6 H)}, 2.85-2.73 \text{ (m, 1)}$ 4 H), 1.85–1.71 (m, 4 H) ppm. MS (EI): m/z (%) = 332 (100). HRMS: calcd. for C18H20O8 322.0896; found 332.0895. Data for **20**: ¹H NMR (250 MHz, CDCl₃): δ = 5.14 (d, J = 1.4 Hz, 1 H), 4.39 (m, 1 H), 3.90 (s, 3 H), 3.87-3.81 (m, 12 H), 3.67 (s, 3 H), 3.02 (m, 1 H), 2.70 (m, 1 H), 2.02 (m, 1 H), 1.88-1.71 (m, 3 H) ppm. ¹³C NMR (75 MHz, CDCl₃): δ = 168.0 (C), 167.2 (C), 166.8 (C), 166.3 (C), 166.2 (C), 165.0 (C), 150.9 (C), 139.2 (C), 136.4 (C), 135.5 (C), 135.1 (C), 129.8 (C), 129.4 (C), 123.9 (CH), 53.0 (CH₃), 53.0 (CH₃), 52.9 (CH₃), 52.8 (CH₃), 52.5 (CH₃), 51.8 (CH₃), 39.4 (CH), 26.3 (CH₂), 25.2 (CH₂), 16.4 (CH₂) ppm. MS (EI): m/z (%) = 474 (8), 207 (100). HRMS: calcd. for C₂₄H₂₆O₁₂ 474.1162; found 474.1171.

[4+2] Cycloaddition of 1-Methyl-1,2-Cyclohexadiene (23) with 1,3-Diphenylisobenzofuran (13): Finely powdered anhydrous CsF (67 mg, 0.44 mmol) was added to a solution of **9** (54 mg, 0.17 mmol) and 1,3-diphenylisobenzofuran (13; 70 mg, 0.26 mmol) in MeCN/THF (2:1, 2 mL), and the mixture was stirred at room temperature for 24 h. Then, the mixture was concentrated under reduced pressure, and the residue was purified by column chromatography (SiO₂; hexane/CH₂Cl₂, 9:1) to afford *endolexo*-9a-methyl-9,10-diphenyl-1,2,3,9,9a,10-hexahydro-9,10-epoxyanthracene (*endolexo*-24; 35.6 mg, 56%; 2.1:1 *endolexo*) as white solids and *endo*-4-methyl-9,10-diphenyl-1,2,3,9,9a,10-hexahydro-9,10-epoxyanthracene (*endo*-25; 24.2 mg, 38%) as a pale-brown solid. Data for *endo*-24: ¹H NMR (300 MHz, CDCl₃): δ = 7.94–7.84 (m, 2 H), 7.72–7.63 (m, 2 H), 7.56–7.40 (m, 5 H), 7.34 (m, 1 H), 7.25 (m, 1 H), 7.21–7.03 (m, 3 H), 5.59 (dd, *J* = 4.2, 2.8 Hz, 1 H), 2.18 (m, 1



H), 1.99 (m, 1 H), 1.91–1.62 (m, 2 H), 1.08 (s, 3 H), 0.95–0.72 (m, 2 H) ppm. ¹³C NMR (75 MHz, CDCl₃): δ = 148.5 (C), 148.2 (C), 145.2 (C), 137.8 (C), 135.2 (C), 128.8 (2 CH), 128.6 (CH), 128.4 (2 CH), 128.1 (2 CH), 127.1 (CH), 127.0 (CH), 125.6 (2 CH), 125.3 (CH), 121.5 (CH), 120.4 (CH), 117.7 (CH), 93.1 (C), 89.6 (C), 46.8 (C), 32.7 (CH₂), 23.4 (CH₂), 21.4 (CH₃), 18.2 (CH₂) ppm. MS (EI): m/z (%) = 364 (31), 346 (100). HRMS: calcd. for C₂₇H₂₄O 364.1827; found 364.1819. Data for exo-24: ¹H NMR (300 MHz, $CDCl_3$): $\delta = 7.87-7.61$ (m, 4 H), 7.56-7.45 (m, 4 H), 7.43-7.31 (m, 3 H), 7.27 (m, 1 H), 7.21–7.02 (m, 2 H), 5.57 (dd, J = 6.2, 3.0 Hz, 1 H), 2.15–1.92 (m, 2 H), 1.70 (m, 1 H), 1.37–1.09 (m, 3 H), 0.88 (s, 3 H) ppm. ¹³C NMR (75 MHz, CDCl₃): δ = 147.7 (C), 147.6 (C), 144.9 (C), 137.5 (C), 136.4 (C), 128.3 (2 CH), 128.2 (2 CH), 127.2 (CH), 127.1 (CH), 126.5 (CH), 126.3 (CH), 125.6 (2 CH), 125.4 (2 CH), 120.3 (CH), 119.8 (CH), 117.1 (CH), 91.8 (C), 89.4 (C), 46.3 (C), 31.9 (CH₂), 20.9 (CH₃), 20.7 (CH₂), 17.0 (CH₂) ppm. MS (EI): m/z (%) = 364 (40), 346 (100). HRMS: calcd. for C₂₇H₂₄O 364.1827; found 364.1821. Data for endo-25. ¹H NMR (300 MHz, CDCl₃): δ = 7.81 (dd, J = 7.9, 1.5 Hz, 2 H), 7.68 (dd, J = 8.2, 1.3 Hz, 2 H), 7.50–7.22 (m, 8 H), 7.20–7.11 (m, 2 H), 3.11 (m, 1 H), 2.26–1.98 (m, 2 H), 1.84–1.56 (m, 3 H), 1.33 (d, J = 1.7 Hz, 3 H), 0.33 (m, 1 H) ppm. ¹³C NMR (75 MHz, CDCl₃): δ = 146.6 (C), 144.7 (C), 138.0 (C), 137.3 (C), 135.6 (C), 129.8 (2 CH), 128.7 (CH), 128.4 (2 CH), 128.2 (2 CH), 127.9 (CH), 127.8 (C), 127.0 (2 CH), 126.8 (CH), 125.5 (CH), 121.9 (CH), 117.6 (CH), 90.4 (C), 90.1 (C), 48.6 (CH), 31.6 (CH₂), 26.2 (CH₂), 22.6 (CH₂), 19.7 (CH₃) ppm. MS (EI): m/z (%) = 364 (33), 346 (100). HRMS: calcd. for C₂₇H₂₄O 364.1827; found 364.1818.

Supporting Information (see footnote on the first page of this article): Determination of the relative stereochemistry of *endo*-16 and 17; copies of the ¹H and ¹³C NMR spectra.

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[4] For some recent examples, see: a) Y. Sato, T. Tamura, M. Mori, Angew. Chem. Int. Ed. 2004, 43, 2436–2440; b) U. K. Tambar,

For reviews on cyclic allenes, see: a) H. Hopf, "The Preparation of Allenes and Cumulenes" in *The Chemistry of Ketenes, Allenes, and Related Compounds* (Ed.: S. Patai), John Wiley & Sons, New York, **1980**; b) R. P. Johnson, *Chem. Rev.* **1989**, 89, 1111–1124; c) M. Balci, Y. Taskesenligil in *Advances in Strained and Interesting Organic Molecules* (Ed.: B. Halton), JAI Press, Stamford, CT, **2000**, vol. 8, p. 43; d) M. Christl, "Cyclic Allenes Up to Seven-Membered Rings" in *Modern Allene Chemistry* (Eds.: N. Krause, A. Stephen, K. Kashmi), Wiley-VCH, Weinheim, **2004**, pp. 243–357; e) T. Kawase in *Science of Synthesis* (Ed.: J. S. Siegel), Thieme, Stuttgart, **2008**, vol. 44, pp. 395–450.

^[2] For reviews on arynes and strained cycloalkynes, see: a) R. W. Hoffmann, *Dehydrobenzene and Cycloalkynes*, Academic Press, New York, **1967**; b) A. Krebs, J. Wilke, *Top. Curr. Chem.* **1983**, *109*, 189–233; c) H. Hart in *The Chemistry of Functional Groups, Suppl. C2: The Chemistry of Triple-Bonded Functional Groups* (Ed.: S. Patai), John Wiley & Sons, Chichester, **1994**, pp. 1017–1134; d) H. Pellissier, M. Santelli, *Tetrahedron* **2003**, *59*, 701–730.

 ^[3] a) R. F. Cunico, E. M. Dexheimer, J. Organomet. Chem. 1973, 59, 153–160; b) Y. Himeshima, T. Sonoda, H. Kobayashi, Chem. Lett. 1983, 1211–1214.

FULL PAPER

B. M. Stoltz, J. Am. Chem. Soc. 2005, 127, 5340–5341; c) Z.
Liu, X. Zhang, R. C. Larock, J. Am. Chem. Soc. 2005, 127, 15716–15717; d) T. T. Jayanth, M. Jeganmohan, M.-J. Cheng, S.-Y. Chu, C.-H. Cheng, J. Am. Chem. Soc. 2006, 128, 2232–2233; e) J. L. Henderson, A. S. Edwards, M. F. Greaney, J. Am. Chem. Soc. 2006, 128, 7426–7427; f) H. Yoshida, H. Fukushima, J. Ohshita, A. Kunai, J. Am. Chem. Soc. 2006, 128, 11040–11041; g) D. Peña, D. Pérez, E. Guitián, Angew. Chem. Int. Ed. 2007, 46, 3323–3325; i) T. Gerfaud, L. Neuville, J. Zhu, Angew. Chem. Int. Ed. 2009, 48, 572–577; j) F. Sha, X. Huang, Angew. Chem. Int. Ed. 2009, 48, 3458–3461.

- [5] a) B. Iglesias, D. Peña, D. Pérez, E. Guitián, L. Castedo, *Synlett* 2002, 486–488; b) D. Peña, B. Iglesias, I. Quintana, D. Pérez, E. Guitián, L. Castedo, *Pure Appl. Chem.* 2006, 78, 451–455.
- [6] a) W. C. Shakespeare, R. P. Johnson, J. Am. Chem. Soc. 1990, 112, 8578–8579; b) C. Shih, E. L. Fritzen, J. S. Swenton, J. Org. Chem. 1980, 45, 4462–4471.
- [7] a) G. Wittig, P. Fritze, Angew. Chem. Int. Ed. Engl. 1966, 5, 846–848; b) G. Wittig, P. Fritze, Justus Liebigs Ann. Chem. 1968, 711, 82–87; c) W. R. Moore, W. R. Moser, J. Am. Chem. Soc. 1970, 92, 5469–5477; d) A. T. Bottini, L. L. Hilton, J. Plott, Tetrahedron 1975, 31, 1997–2001; e) M. Balci, W. M. Jones, J. Am. Chem. Soc. 1980, 102, 7607–7608; f) M. W. Schmidt, R. O. Angus Jr., R. P. Johnson, J. Am. Chem. Soc. 1982, 104, 6838–6839; g) M. Christl, M. Schreck, Angew. Chem. Int. Ed. Engl. 1987, 26, 449–451.
- [8] Y. Sütbeyaz, M. Ceylan, H. Seçen, J. Chem. Res. (S) 1993, 293.
- [9] It has been proposed that the dimerization of 1,2-cyclohexadiene occurs through a stepwise path via a diallylene intermedi-

ate; see: W. R. Moore, P. D. Mogolesko, D. D. Traficante, J. Am. Chem. Soc. 1972, 94, 4753–4754.

- [10] Computational studies on the Diels–Alder reaction of strained cyclic allenes suggest that diradical stepwise pathways are preferred over the concerted paths; see: M. Nendel, L. M. Tolbert, L. E. Herring, Md. N. Islam, K. N. Houk, J. Org. Chem. 1999, 64, 976–983.
- [11] a) J. Ciabattoni, J. E. Crowley, A. S. Kende, J. Am. Chem. Soc. 1967, 89, 2778–2779; b) T. Miwa, M. Kato, T. Tamano, Tetrahedron Lett. 1969, 10, 1761–1764.
- [12] a) K. Hayakawa, H. Nishiyama, K. Kanematsu, J. Org. Chem. 1985, 50, 512–517; b) M. P. S. Ishar, R. P. Gandhi, *Tetrahedron* 1993, 49, 6729–6740.
- [13] See Supporting Information for details.
- [14] a) J. P. Visser, J. E. Ramakers, J. Chem. Soc., Chem. Commun. 1972, 178–179; b) J. Yin, K. A. Abboud, W. M. Jones, J. Am. Chem. Soc. 1993, 115, 3810–3811.
- [15] For a review of palladium-catalyzed cycloaddition reactions of arynes, see: E. Guitián, D. Pérez, D. Peña in *Topics in Organometallic Chemistry* (Ed.: J. Tsuji), Springer, Berlin, 2005, vol. 14, pp. 109–146.
- [16] This finding suggests a diradical stepwise pathway for this reaction. See ref.^[10]
- [17] M. Christl, M. Schreck, Chem. Ber. 1987, 120, 915-920.
- [18] The structure of dimer 10 was established by comparison with previously reported spectroscopic data: W. R. Moore, L. N. Bell, G. P. Daumit, J. Org. Chem. 1971, 36, 1694–1695.
- [19] G. Maier, R. Wilmes, H. Fuchs, M. Leinweber, *Chem. Ber.* 1993, 126, 1827–1833.

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