Synthesis of (7-polyfluoroalkyl)pyrazolo[1,5-*a*]pyrimidines based on lithium fluorine-containing β-diketonates

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The reactions of Li enolates of fluorine-containing β -diketones with 3-aminopyrazoles afforded (7-polyfluoroalkyl)pyrazolo[1,5-*a*]pyrimidines. The structure of 3-bromo-2-methyl-5-phenyl-7-trifluoromethylpyrazolo[1,5-*a*]pyrimidine was established by X-ray diffraction analysis.

Key words: Li enolates of fluorine-containing β -diketones, 3-aminopyrazoles, (7-fluoro-alkyl)pyrazolo[1,5-*a*]pyrimidines.

In recent years, the use of fluoroalkyl-containing synthons for insertion of fluorine-containing fragments into organic, including heterocyclic, compounds has attracted growing interest.

Recently,¹ we have proposed new fluorine-containing synthons, *viz.*, Li enolates of fluorinated β -diketones **1**. These compounds, which can readily be prepared according to the version of the Claisen condensation developed by us, serve as intermediates in the synthesis of the corresponding β -diketones and β -hydroxy ketones.^{2–4} Enolates **1** can be used as synthetic equivalents of fluorine-containing β -diketones for the preparation of fluoroalkyl-substituted pyrazoles, 5-hydroxy-4,5-dihydroisoxazoles, 2-aminopyrimidines, 1,5-benzodiazepines, 1,5-naphthodiazepines, pyrimidinethiols, and triazolopyridazines.¹

In the present study, we demonstrated that enolates 1 are convenient synthons for the preparation of fluoroalkylcontaining fused azaheterocycles with the bridgehead N atom, in particular, for the synthesis of pyrazolo[1,5-*a*]pyrimidines.*

Results and Discussion

Enolates 1a-c readily react with equimolar amounts of 3-aminopyrazoles 2a-d in AcOH at ~40 °C or with boiling (Scheme 1). The synthetic procedure is simple and the target products 3a-f can be obtained in 52-87% yields depending on the purity of the starting enolate. This approach can be applied to the synthesis of pyrazolo[1,5-*a*]pyrimidines whose pyrazole ring contains functional groups capable of undergoing further transformations.

In all cases, only one of the possible regioisomers was obtained as evidenced by the presence of the only set of resonance signals in the ¹H, ¹⁹F, and ¹³C NMR spectra of the resulting pyrazolo[1,5-*a*]pyrimidines.

The reactions of enolates of unsymmetrical β -dicarbonyl compounds **1** with aminopyrazoles can afford two different intermediates (**A** or **B**) followed by their condensation to form pyrazolopyrimidines **3** or **4**. It is more likely that the formation of isomer **3** proceeds *via* β -aminovinyl ketones **A** as evidenced by the results of our previous studies^{1,6} on the structures of the products obtained in reactions of unsymmetrical polyfluorinated β -diketones and their Li enolates with amines.

This suggestion was confirmed by the results of X-ray diffraction analysis of pyrazolopyrimidine **3c** (Fig. 1). The bond lengths and bond angles in compound **3c** are given in Tables 1 and 2, respectively.

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^{*} Nonfluorinated 3-substituted 5,7-dialkylpyrazolo[1,5-*a*]pyrimidines are selective inhibitors *in vitro* of cyclic adenosine monophosphate phosphodiesterase.⁵ In the cited study, fluorinecontaining compounds of this series, *viz.*, 5,7-bis(trifluoromethyl)- and 3-fluoro-5,7-dimethylpyrazolo[1,5-*a*]pyrimidines, were synthesized and their antidepressant properties were examined; however, their spectroscopic characteristics were lacking.

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Scheme 1

 $\begin{array}{l} \textbf{1:} \ R^{F} = CF_{3}\left(\textbf{a}\right), \ HCF_{2}\left(\textbf{b}\right), \ C_{4}F_{9}\left(\textbf{c}\right); \ R = Ph\left(\textbf{a},\textbf{b}\right), \ Me\left(\textbf{c}\right) \\ \textbf{2:} \ R^{1} = H\left(\textbf{a},\textbf{d}\right), \ Me\left(\textbf{b},\textbf{c}\right); \ X = H\left(\textbf{a},\textbf{b}\right), \ Br\left(\textbf{c}\right), \ COOEt\left(\textbf{d}\right) \\ \textbf{3:} \ R^{F} = CF_{3}\left(\textbf{a}-\textbf{d}\right), \ HCF_{2}\left(\textbf{e}\right), \ C_{4}F_{9}\left(\textbf{f}\right); \ R = Ph\left(\textbf{a}-\textbf{e}\right), \ Me\left(\textbf{f}\right); \\ R^{1} = H\left(\textbf{a},\textbf{d},\textbf{e}\right), \ Me\left(\textbf{b},\textbf{c},\textbf{f}\right); \ X = H\left(\textbf{a},\textbf{b},\textbf{f}\right), \ Br\left(\textbf{c}\right), \ COOEt\left(\textbf{d},\textbf{e}\right) \end{array}$

The bicyclic fragment is virtually planar (to within $\pm 0.079(4)$ Å), which is favorable for the formation of the common π -conjugated system of the pyrazole and pyrimidine rings. This conjugated system is characteristic of all the compounds of this type reported in the litera-



Fig. 1. Molecular structure of 3-bromo-2-methyl-5-phenyl-7-trifluorometylpyrazolo[1,5-*a*]pyrimidine (**3c**).

Table 1. Bond lengths (d) in the structure of 3c

Bond	$d/\text{\AA}$	Bond	$d/{ m \AA}$
Br-C(3)	1.864(3)	$\overline{C(7)-N(7a)}$	1.361(4)
N(1) - C(2)	1.343(5)	C(7) - C(15)	1.496(5)
N(1)-N(7a)	1.352(4)	C(9) - C(14)	1.376(5)
C(2) - C(3)	1.389(5)	C(9) - C(10)	1.386(5)
C(2) - C(8)	1.479(6)	C(10) - C(11)	1.370(5)
C(3) - C(3a)	1.376(5)	C(11) - C(12)	1.368(6)
C(3a) - N(4)	1.347(4)	C(12) - C(13)	1.372(6)
C(3a)–N(7a)	1.389(4)	C(13) - C(14)	1.383(5)
N(4) - C(5)	1.317(4)	C(15)-F(17)	1.314(5)
C(5) - C(6)	1.437(4)	C(15) - F(18)	1.323(4)
C(5) - C(9)	1.475(4)	C(15)-F(16)	1.328(5)
C(6)-C(7)	1.341(5)		

ture. $^{7-10}$ (Four structures containing the same bicyclic pyrazolopyrimidine framework⁷⁻¹⁰ were found in the Cambridge Structural Database¹¹). Comparison of the bond lengths with the statistical mean values for pyrazoles and pyrimidines¹² confirms the electron density redistribution within the bicyclic fragment. Thus the $C(6)^{---}C(7)$ and $C(5)^{---}N(4)$ bond lengths in the pyrimidine ring (1.341(5) and 1.317(4) Å, respectively) are smaller than the statistical mean lengths (1.387(18) and 1.339(15) Å, respectively) of the corresponding bonds in pyrimidines.¹² However, it should be noted that the $C(6)^{---}C(7)$ bond length is also smaller than the known bond lengths for pyrazolopyrimidines, which range from 1.355¹⁰ to 1.394 Å.⁷ The C(6)–C(5) and C(7)–N(7a) bond lengths in the pyrimidine ring (1.437(7) and 1.368(4) Å, respectively) are, on the contrary, larger than the statistical mean values (1.387(18) and 1.339(15) Å, respectively). In the pyrazole ring, the $N(1)^{---}C(2)$ and $C(3a)^{---}C(3)$ bond

Table 2. Selected bond angles (ω) in the structure of **3c**

Angle	ω/deg
C(2)-N(1)-N(7a)	104.4(3)
N(1) - C(2) - C(3)	111.1(3)
C(3) - C(2) - C(8)	128.6(4)
C(3a) - C(3) - C(2)	107.4(3)
C(2) - C(3) - Br	126.4(3)
N(4) - C(3a) - N(7a)	122.2(3)
C(3) - C(3a) - N(7a)	104.1(3)
C(5) - N(4) - C(3a)	118.1(3)
N(4) - C(5) - C(6)	121.3(3)
C(6) - C(5) - C(9)	120.8(3)
C(7) - C(6) - C(5)	119.8(3)
C(6) - C(7) - N(7a)	118.6(3)
C(6) - C(7) - C(15)	124.3(3)
N(1)-N(7a)-C(3a)	113.0(3)
C(7)-N(7a)-C(3a)	120.0(3)

lengths (1.343(5) and 1.376(5) Å, respectively) are larger (1.329(14) and 1.341(12) Å for pyrazoles¹²), whereas the N(7a)—N(1) and C(2)—C(3) bond lengths (1.352(4) and 1.389(5) Å, respectively) are smaller than the statistical mean values for pyrazoles (1.366(19) and 1.410(16) Å, respectively).¹² The N(7a)—C(3a) bond (1.389(4) Å) is elongated as compared to the analogous bonds in pyrazoles and pyrimidines (1.357(12) and 1.333(13) Å, respectively),¹² but this bond length is consistent with the data for pyrazolopyrimidines (from 1.390⁷ to 1.397 Å)¹⁰.

The angle between the planes of the bicyclic fragment and the phenyl ring is $29.3(2)^\circ$. The intermolecular Br...Br distance (3.659(1) Å) is smaller than the sum of the van der Waals radii $(3.9 \text{ Å} \text{ for Br}^{13})$.

The spectroscopic characteristics of compound **3c** correspond to those for CF₃-containing pyrazolo[1,5-*a*]pyrimidines.* For the latter compounds, a procedure has been recently proposed^{14,15} for the identification of the 5-R- and 7-R-regioisomers based on the chemical shifts of the protons of the Me groups in ¹H NMR spectra and of the C atoms of the alkyl groups, the C(5) and C(7) atoms of the pyrimidine ring, and the *ipso*-C atoms of the Ph group in ¹³C NMR spectra.

The ¹H and ¹³C NMR spectra of compounds **3a,b,d**—f were interpreted by comparing them with the spectra of compound **3c** and the published data.^{14,15} The assignment of the protonated C atoms in compound **3e** was made using the 2D HETCOR experiment based on correlations through the spin-spin coupling constants ${}^{1}J_{C,H}$.

The signals for the C(7) atoms in compounds **3a,c,d** occur as characteristic quartets at δ 134–135 with the spin-spin coupling constants ${}^{2}J_{C,F} \approx 37-38$ Hz. The signal for the C(7) atom in compound **3e** occurs as a triplet at δ 139.7 with the spin-spin coupling constant ${}^{2}J_{C,F} \approx 28$ Hz.

The ¹³C NMR spectra of compounds **3a,c,d,e** also have signals for the C(5) (δ 155–159) and C_{*ipso*}(Ph) (δ 135–136) atoms.

The mass spectra of compounds **3a,c** have molecular ion peaks and fragmentation peaks $[M - F]^+$, $[M - CF_3]^+$, $[Ph]^+$ (*m*/*z* 77), and $[CF_3]^+$ (*m*/*z* 69). The mass spectra of the compounds under consideration have no peaks responsible for specific fragmentation of the fused azaheterocycle.

To summarize, Li enolates of fluorinated β -diketones 1, like the corresponding β -diketones,^{14,15} react with 3-aminopyrazoles to form only one of the possible

regioisomers of pyrazolo[1,5-*a*]pyrimidines containing the fluoroalkyl substituent at position 7. However, enolates **1** are reagents of choice because they allow one to substantially simplify the synthetic procedure due to elimination of the necessity for purification and, in some cases, isolation of β -diketones.

Experimental

The synthesis of enolates 1a-c has been reported previously.¹ 3-Aminopyrazoles 2a-d were prepared according to known procedures.^{16–18} The course of the reactions was monitored by TLC on Silufol UV-254 plates (CHCl₃ and CCl₄ were used as eluents). The purities of the reaction products were checked by two-dimensional TLC. The compounds were visualized using aqueous solutions of Cu(OAc)₂ and KMnO₄.

The IR spectra were recorded on a Specord IR-75 spectrometer in thin layers (20 μ m) for liquid samples and in Nujol mulls for solids.

The ¹H and ¹³C NMR spectra were measured on a Bruker DRX-400 spectrometer (400 and 100 MHz) in CDCl₃ with Me₄Si as the internal standard. The ¹⁹F NMR spectra were recorded on a Tesla BS-587A spectrometer (75.3 MHz) in CDCl₃ with C₆F₆ as the internal standard.

The mass spectra (EI) were obtained on a MAT INCOS 50 spectrometer (the energy of ionizing electrons was 70 eV, direct inlet of the sample into the ion source).

5-Phenyl-7-trifluoromethylpyrazolo[1.5-*a*]pyrimidine (3a). A solution of enolate 1a (1 g, 4.5 mmol) and 3-aminopyrazole (2a) (0.58 g, 4.5 mmol) in glacial AcOH (5 mL) was refluxed for 10 h and then the reaction mixture was poured onto ice. The precipitate that formed was filtered off, dried, and recrystallized from hexane. Compound 3a was obtained in a yield of 0.96 g (85%), m.p. 105-106 °C. Found (%): C, 59.57; H, 3.11; F, 21.65; N, 15.93. C₁₃H₈F₃N₃. Calculated (%): C, 59.31; H, 3.04; F, 21.67; N, 15.97. IR, v/cm⁻¹: 1550 (C=C); 1620 (C=N). ¹H NMR, δ : 6.87 (s, 1 H, H(3)); 7.62 (s, 1 H, H(6); 7.52 and 8.09 (both m, 3 H + 2 H, Ph); 8.25 (s, 1 H, H(2)). ¹³C NMR, δ: 98.35 (s, C(3)); 103.68 (q, C(6), ${}^{3}J_{C,F} = 4.4$ Hz); 119.64 (q, CF₃, ${}^{1}J_{C,F} = 274.9$ Hz); 127.20 (s, o-C); 129.07 (s, m-C); 131.01 (s, p-C); 134.36 (q, C(7), ${}^{2}J_{C,F} = 37.1 \text{ Hz}$; 133.44 (s, C_{ipso}); 146.27 (s, C(2)); 155.30 (s, C(5)); 149.69 (s, C(3a)). ¹⁹F NMR, δ: 92.82 (CF₃). MS, m/z: 263 [M]⁺.

2-Methyl-5-phenyl-7-trifluoromethylpyrazolo[1,5-*a*]**pyrimidine (3b).** A solution of enolate 1a (1 g, 4.5 mmol) and 3-amino-5-methylpyrazole (2b) (0.22 g, 4.5 mmol) in AcOH (5 mL) was kept at 40 °C for one day and then the reaction mixture was poured onto ice. The precipitate that formed was filtered off, dried, and recrystallized from hexane. Compound **3b** was obtained in a yield of 0.9 g (72%), m.p. 123–124 °C. Found (%): C, 60.63; H, 3.69; N, 15.45. C₁₄H₁₀F₃N₃. Calculated (%): C, 60.65; H, 3.61; N, 15.16. IR, v/cm⁻¹: 1610 (C=N); 1545 (C=C). ¹H NMR, δ : 2.70 (s, 3 H, Me); 6.64 (s, 1 H, H(3)); 7.58 and 8.21 (both m, 3 H + 2 H, Ph); 7.79 (s, 1 H, H(6)).

3-Bromo-2-methyl-5-phenyl-7-trifluoromethylpyrazolo[1,5-a]pyrimidine (3c). A solution of enolate **1a** (1 g, 4.5 mmol)

^{*} The compounds were prepared by cyclocondensation of substituted aminopyrazoles with 1,1,1-trifluoromethyl-containing β -diketones in AcOH or by fusion of the reagents.^{14,15} However, the experimental details and the yields of the target products were not reported in the cited studies.

and 3-amino-4-bromo-5-methylpyrazole (2c) (0.80 g, 4.5 mmol) in AcOH (5 mL) was kept at 40 °C for one week. Then the solution was neutralized with NaHCO₃ and the product was extracted with CHCl₃. The extract was dried with MgSO₄, the solvent was evaporated, and the residue was recrystallized from hexane. Compound **3c** was obtained in a yield of 1.3 g (87%), m.p. 178-180 °C. Found (%): C, 46.35; H, 2.53; F, 15.33; N, 11.30. C₁₄H₉BrF₃N₃. Calculated (%): C, 47.19; H, 2.53; F, 16.01; N, 11.80. IR, v/cm⁻¹: 1545 (C=C); 1610 (C=N). 1 H NMR, δ : 2.57 (s, 3 H, Me); 7.53 and 8.14 (both m, 3 H + 2 H, Ph); 7.57 (s, 1 H, H(6)). ¹³C NMR, δ: 13.52 (s, CH₃); 87.33 (s, C(3)); 103.57 (q, C(6), ${}^{3}J_{C,F} = 4.1$ Hz); 119.41 (q, CF₃, ${}^{1}J_{C,F} = 274.7$ Hz); 127.42 (s, o-C); 129.18 (s, m-C); 131.36 (s, *p*-C); 134.29 (q, C(7), ${}^{2}J_{C,F} = 37.2$ Hz); 135.90 (s, C_{*ipso*}); 146.66 (s, C(2)); 155.88 (s, C(5)); 155.35 (s, C(3a)). MS, m/z: 356 [M]⁺.

Ethyl 5-phenyl-7-trifluoromethylpyrazolo[1,5-a]pyrimidine-3-carboxylate (3d). A solution of enolate 1a (1g, 4.5 mmol) and ethyl 3-aminopyrazole-4-carboxylate (2d) (0.7 g, 4.5 mmol) in AcOH (5 mL) was refluxed for 10 h and then the reaction mixture was poured onto ice. The precipitate that formed was filtered off and reprecipitated with hexane from CHCl₃. Compound 3d was obtained in a yield of 0.78 g (52%), m.p. 129 °C. Found (%): C, 56.51; H, 3.87; N, 12.94. C₁₆H₁₂F₃N₃O₂. Calculated (%): C, 57.31; H, 3.58; N, 12.53. IR, v/cm⁻¹: 1573 (C=C); 1635 (C=N); 1700 (C=O). ¹H NMR, δ : 1.47 (t, 3 H, Me, J = 7.2 Hz); 4.42 (q, 2 H, CH_2 , J = 7.2 Hz); 7.26 and 7.56 (both m, 3 H + 2 H, Ph); 7.48 (s, 1 H, H(6)); 7.78 (s, 1 H, H(2)). 13 C NMR, δ : 14.37 (s, OCH₂CH₃); 60.48 (s, OCH₂CH₃); 104.38 (s, C(3)); 104.69 (q, C(6), ${}^{3}J_{C,F} = 4.1$ Hz); 119.20 (q, CF₃, ${}^{1}J_{C,F} = 275.0 \text{ Hz}$; 127.67 (s, o-C); 129.16 (s, m-C); 131.93 (s, *p*-C); 135.10 (q, C(7), ${}^{2}J_{C,F} = 37.8$ Hz); 135.33 (s, C_{ipso}); 148.52 (s, C(2)); 158.51 (s, C(5)); 148.50 (s, C(3a)); 161.98 (s, CO₂Et). ¹⁹F NMR, δ: 93.39 (CF₃).

Ethyl 7-difluoromethyl-5-phenylpyrazolo[1,5-a]pyrimidine-3carboxylate (3e). A solution of enolate 1b (1 g, 4.9 mmol) and ethyl 3-aminopyrazole-4-carboxylate (2d) (0.76 g, 4.9 mmol) in AcOH (5 mL) was refluxed for 10 h and then the reaction mixture was poured onto ice. The precipitate that formed was dried and reprecipitated with hexane from CHCl₂. Compound **3e** was obtained in a yield of 0.99 g (64%), m.p. 137 °C. Found (%): C, 59.21; H, 4.44; N, 13.18. C₁₆H₁₃F₂N₃O₂. Calculated (%): C, 60.56; H, 4.10; N, 13.24. IR, v/cm⁻¹: 1560 (C=C); 1624 (C=N); 1710 (C=O). ¹H NMR, δ: 1.46 (t, 3 H, Me, J =7.2 Hz); 4.49 (q, 2 H, CH_2 , J = 7.2 Hz); 7.38 (t, 1 H, HCF_2 , J = 52.4 Hz); 7.54 and 8.24 (both m, 3 H + 2 H, Ph); 7.72 (s, 1 H, H(6)); 8.59 (s, 1 H, H(2)). ¹³C NMR, δ: 14.41 (s, OCH_2CH_3 ; 60.40 (s, OCH_2CH_3); 103.18 (t, C(6), ${}^{3}J_{CF} =$ 4.9 Hz); 104.04 (s, C(3)); 107.96 (t, HCF₂, ${}^{1}J_{C,F} = 241.6$ Hz); 127.71 (s, o-C); 129.09 (s, m-C); 131.69 (s, p-C); 135.78 (s, C_{ipso}); 139.67 (t, C(7), ${}^{2}J_{C,F} = 28.1$ Hz); 147.93 (s, C(3a)); 148.24 (s, C(2)); 159.07 (s, C(5)); 162.22 (s, <u>CO₂Et</u>). ¹⁹F NMR, δ: 37.36 (d, HCF₂, ${}^{2}J_{F,H} = 52.4$ Hz).

2,5-Dimethyl-7-nonafluorobutylpyrazolo[1,5-*a***]pyrimidine (3f). A solution of enolate 1c (1 g, 3.2 mmol) and 3-amino-5methylpyrazole (2b) (0.31 g, 3.2 mmol) in AcOH (5 mL) was refluxed for 6 h, poured onto ice, and extracted with ether. The solvent was evaporated and the residue was chromatographed on a column with silica gel L 100/250 (CHCl₃ as the eluent). The solvent was evaporated and yellow compound 3f was obtained in** a yield of 0.9 g (78%), m.p. 101 °C. Found (%): C, 39.40; H, 2.11; N, 11.43. $C_{12}H_8F_9N_3$. Calculated (%): C, 39.45; H, 2.19; N, 11.50. IR, v/cm⁻¹: 1620 (C=N); 1555 (C=C). ¹H NMR, δ : 2.51 and 2.64 (both s, 3 H each, Me); 6.48 (s, H(3)); 6.88 (s, H(6)). ¹⁹F NMR, δ : 81.19 (tt, C(4')F₃, *J* = 9.8 Hz, *J* = 2.4 Hz); 48.99 (t, C(1')F₂, *J* = 12.7 Hz); 41.58 (m, C(2')F₂); 36.17 (m, C(3')F₃).

X-ray diffraction study of compound 3c. X-ray diffraction data were collected on a Bruker P4 diffractometer (graphite monochromator, Mo-K α radiation, $\theta/2\theta$ scanning technique, $2\theta < 50^{\circ}$). The pale-yellow crystals are monoclinic, a = 9.958(2), b = 13.112(3), c = 10.370(2) Å, $\beta = 90.13(1)^{\circ}, V = 1354.0(5)$ Å³, space group $P2_1/c$, $C_{14}H_9BrF_3N_3$, Z = 4, molecular weight 356.15, $d_{\text{calc}} = 1.747 \text{ g cm}^{-3}$, $\mu = 3.066 \text{ mm}^{-1}$. The intensities of 2336 independent reflections were measured from a crystal with dimensions of $0.33 \times 0.54 \times 0.80$ mm. The absorption corrections were applied taking into account the crystal habitus (transmission was 0.23–0.39). The structure was solved by direct methods using the SHELXS-86 program package and refined by the least-squares method in the anisotropic-isotropic approximation using the SHELXL-97 program package to $wR_2 = 0.1043$, S = 1.041 for all reflections (R = 0.0392 for 1622 $F > 4\sigma$). The positions of the H atoms were revealed from the difference electron density synthesis and refined isotropically. The molecular structure of 3-bromo-2-methyl-5-phenyl-7-trifluoromethylpyrazolo[1,5-a]pyrimidine **3c** is shown in Fig. 1. The complete tables of the atomic coordinates and thermal parameters were deposited with the Cambridge Structural Database.

The mass spectra were measured within the framework of the Free Spectra Program of the Oak Ridge National Laboratory (USA) (Grant IPP, Project ORS-T2-0175-RU) and the Chemical Block Int. Company.

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