DOI: 10.1002/ejoc.200901422

Total Synthesis of (-)-Fasicularin and (-)-Lepadiformine A Based on Zn-Mediated Allylation of Chiral *N-tert*-Butanesulfinyl Ketimine

San-Lin Mei^[a] and Gang Zhao*^[a]

Keywords: Total synthesis / Alkaloids / Fasicularin / Lepadiformine A / Zinc / Allylation

The stereoselective total synthesis of fasicularin (1) and lepadiformine A (2) is described, which features the utilization of a Zn-mediated allylation of a chiral, aliphatic, *N-tert*butanesulfinyl ketimine to construct the amino-substituted quaternary carbon center in good yield and with excellent diastereoselectivity. The azaspirocyclic scaffold was installed sequentially by a Sharpless dihydroxylation and an internal epoxide-opening reaction, and this scaffold was further converted into common intermediate **5**. Removing the tosyl (Ts) protecting group of **5** and reductively aminating using Luche's reagent completed the synthesis of (-)-fasicularin (1), while reducing **5** using L-selectride, deprotecting the Ts group, and finishing with an intramolecular amino alcohol cyclocondensation completed the total synthesis of (-)-lepadiformine A (2).

Introduction

Lepadiformines A–C and fasicularin, separately isolated by Biad et al. from *Clavelina lepadiformis* off the coast of Tunisia and later from *Clavelina moluccensis*^[1a–1c] and by Patil et al. from *Nephteis fasicularis* collected in Pompeii,^[1d] belong to a series of tricyclic, marine alkaloids bearing a common pyrrolo-/pyrido[1,2-*j*]quinoline framework including a *trans*-1-azadecalin, A–B ring system. In addition to their peculiar structures, these compounds also have some biologically interesting activities such as moderate in vitro cytotoxicity against several tumor cell lines, various cardiovascular effects in vitro and in vivo and antiarrythmia properties.^[1a,1c,2] Therefore, these complex structures have been the focus of many synthetic efforts.^[3,4] Chiral *N-tert*-butanesulfinamide is a versatile reagent in asymmetric synthesis, especially as an excellent chiral auxiliary for asymmetric induction in the preparation of all kinds of synthetically useful chiral amines.^[5] However, applications of this useful chiral auxiliary in the total synthesis of complex natural products are rare. With our continued interest in the total synthesis of natural alkaloids bearing an azaspirotricyclic skeleton with a sterically congested, quaternary carbon center,^[6] we hypothesized that chiral *N-tert*-butanesulfinamides should be highly useful for the synthesis of such alkaloids. Herein, we report the utilization of a Zn-mediated, allylation reaction of a chiral *Ntert*-butanesulfinyl ketimine as the key step in the synthesis of (–)-fasicularin and (–)-lepadiformine A (1 and 2, respectively, Figure 1).

 $\begin{array}{l} {\sf R}^1 = {\sf CH}_2 {\sf OH}, \ {\sf R}^2 = {\it n-C}_6 {\sf H}_{13}, \ {\sf Lepadiformine \ A} \ {\textbf (2)} \\ {\sf R}^1 = {\sf CH}_2 {\sf OH}, \ {\sf R}^2 = {\it n-C}_4 {\sf H}_9, \ \ {\sf Lepadiformine \ B} \ {\textbf (3)} \\ {\sf R}^1 = {\sf H}, \qquad {\sf R}^2 = {\it n-C}_6 {\sf H}_{13}, \ {\sf Lepadiformine \ C} \ {\textbf (4)} \end{array}$



Fasicularin (1)

 [a] Key Laboratory of Synthetic Chemistry of Natural Substances, Shanghai Institute of Organic Chemistry, Chinese Academy of Sciences, 345 Lingling Road, Shanghai 20032, P. R. of China

Fax: +86-21-64166128

E-mail: Zhaog@mail.sioc.ac.cn

Supporting information for this article is available on the WWW under http://dx.doi.org/10.1002/ejoc.200901422.

Results and Discussion

Retrosynthetic Analysis

We envisioned that both 1 and 2 could be obtained by different cyclization processes from a common intermediate 5, which, in turn, could be derived from azaspirobicyclic



1660



Scheme 1. Retrosynthetic analysis.

alcohol **6** (Scheme 1). The synthesis of **6** could be achieved by an intramolecular Sharpless dihydroxylation sequence from sulfonamide **7** with an amino-substituted, quaternary, carbon center. In view of the recent studies on the synthesis of chiral homoallylic amines by the Zn-mediated allylation of chiral *N-tert*-butanesulfinyl imines by Lin and coworkers,^[5b,5c] we envisaged that the allylation of the corresponding, chiral, *N-tert*-butanesulfinyl ketimine **9** would produce sulfinamide **8** with the desired chiral quaternary carbon center, which could be further converted to **7**.

Construction of the Quaternary, Amino-Substituted Carbon Center

Our synthetic route to the key sulfonamide intermediate 7 is outlined in Scheme 2. The condensation of chiral ketone $10^{[7]}$ (94% *ee*), and (*R*)-*tert*-butanesulfinamide with a known procedure^[8] using Ti(OEt)₄ as both the Lewis-acid promoter and water scavenger afforded desired ketimine 9 in 72% combined yield and with a *dr* of 9:1; these isomers could be separated by chromatography on a multigram scale. Next, efforts were directed to the construction of the amino-substituted, quaternary, carbon center. To the best

of our knowledge, there has been little study on the nucleophilic addition of organometallic reagents with chiral Ntert-butanesulfinyl imines derived from aliphatic cyclic ketones, notwithstanding the great usefulness of the resulting chiral amine products.^[5b,5c] At first, we attempted to synthesize 7 directly by the addition of homoallylic metal reagents. However, both the corresponding organolithium and Grignard reagents provided the desired product in very low yield.^[9] We then turned to the known, Zn-mediated, allylation system, because organozinc reagents are generally less basic. To our delight, treating ketimine 9 with preactivated Zn powder and allylic bromide in THF produced the desired sulfinamide 8 in 89% yield and with a dr of up to 12:1 in the absence of a Lewis acid. Notably, in the case of the ketimines derived from simple aromatic ketones, stoichiometric In(OTf)₃ was required as an additive for the reaction to proceed.^[5a] The stereochemical result of this transformation was tentatively explained by a chelated, chair-like, transition state model I (Scheme 2), in which the allyl-Zn could coordinate to the sulfinyl oxygen.^[5a] We note that the chiral sulfinyl group prevailed over chiral induction by the substrate in the stereochemical outcome; when the diastereomer of 9 with an (S) configuration at the tert-



Reagents and conditions: a) $Ti(OEt)_4/THF$, 76%, dr = 9:1; b) Zn, allyl bromide/THF, 89%, dr = 12:1; c) HCl in dioxirane/MeOH, 92%; d) TsCl, Et₃N, DMAP/CH₂Cl₂; e) 9-BBN, then H₂O₂/NaHCO₃, 86% over two steps; f) TPAP, NMO, 4Å sieves/CH₂Cl₂, 80%; g) DIBAL-H/CH₂Cl₂; h) NaHMDS, Ph₃PCH₃Br/THF, 92% over two steps.



FULL PAPER

butanesulfinyl group underwent the same reaction, the predominant configuration of the newly created quaternary center was (R). The subsequent removal of the *tert*-butanesulfinyl group generated primary amine **11**. After the protection by the tosyl (Ts) group and hydroboration-oxidation of **11**,^[10] the resulting primary alcohol **12** was converted into the corresponding lactam **13** by TPAP oxidation.^[11,12] Reduction with DIBAL-H followed by a Wittig reaction successfully gave desired sulfonamide **7**. The structure of **13** was confirmed by the X-ray analysis of its ester, which was obtained by replacing the benzyl group with a 4-nitrobenzoyl group (see the Supporting Information for details).

Preparation of Azaspirobicyclic Ketone 5

With sulfonamide 7 in hand, our designed strategy subsequently called for the construction of the azaspirobicyclic core. Disappointingly, the direct, intramolecular aminohydroxylation of 7 using m-CPBA^[13] as the oxygen source unselectively afforded spirocyclic alcohol 6 in 98% yield with a 6/13-epi-6 diastereoselectivity of 1:1; these isomers could not be separated by flash chromatography (Scheme 3). Attempts to improve the diastereoselectivity by an asymmetric epoxidation and ring-opening sequence using Shi's epoxidation^[14a] and Jacobsen's epoxidation^[14b] also failed. Finally, Sharpless dihydroxylation^[15a,15b] followed by selective protection^[15c] gave isomers 14 and 15 (14/15 = 2.8:1), which could be separated by column chromatography. Treating 14 with NaH gave 6 with the desired configuration, which was confirmed by the X-ray crystallographic analysis of the corresponding diester 16 (see the Supporting Information for details). Having established the azaspirobicyclic structure, the synthesis of ring-closure precursor 5 could be achieved by several routine steps. TBDPS protection of 6 followed by the removal of the benzyl group led to primary alcohol 17, which was further converted to mesylate 18. Treating 18 with excess potassium cyanide gave nitrile 19, which was treated with n-hexylmagnesium bromide to generate desired ketone 5 in excellent yield.

Completion of the Total Synthesis of (-)-Fasicularin

Having the ring-closure precursor for the third cycle of the tricyclic alkaloids in hand, we first investigated its transformation to (–)-fasicularin (1). The removal of the Ts group of **5** with Na/naphthalene^[16] generated a cyclized intermediate enamine in situ, which was directly reduced in the next step without further purification because of its instability (Scheme 4). After performing some experimentation, the Luche reduction^[6,17] of the enamine was identified as the optimum method, affording the desired tricyclic tertiary amine **20** as a single isomer. The subsequent removal of the TBDPS group followed by Kibayashi's procedure (see the Experimental Section for details) completed the total synthesis of (–)-fasicularin (1), whose spectroscopic data were identical with those of reported values.^[4g]

Completion of the Total Synthesis of (-)-Lepadiformin

To synthesize (–)-lepadiformine A, ketone **5** was first reduced to the corresponding alcohol **22** (Scheme 5). Among the various conditions screened for this transformation, the reduction with L-Selectride^[4o] gave the desired product with the best *dr* of 6:1 in 99% yield. After removing the Ts group of **22**, an intramolecular, Mitsunobu-type reaction generated tricyclic **23** in excellent yield. We note that no desired product was found without the addition of a catalytic amount of DMAP to the standard reaction conditions of Ph₃P, CBr₄ and Et₃N.^[4g] Finally, the removal of the TBDPS group led to the total synthesis of (–)-lepadiformine A **(2)**, whose spectroscopic data were identical with reported values.^[4h]



Reagents and conditions: a) AD-mix- α , OsO₄, tBuOH/H₂O; b) TsCl, Bu₂SnO, Et₃N; c) NaH/THF, 0 C, 80% over three steps, dr = 2.8 : 1; d) TBDPSCl, imidazole, DMAP, DMF; e) Pd(OH)₂/C, H₂, 94% over two steps; f) MsCl, Et₃N, DMAP, CH₂Cl₂; g) KCN, DMSO, 91% over 2 steps; h) C₆H₁₃MgBr/Et₂O, reflux, 90%.

Scheme 3. Synthesis of the azaspirobicyclic ketone 5.



Reagents and conditions: a) Na, naphthalene/DME, –65 to –70 °C, 77%; b) NaBH₄/CeCl₃·7H₂O, –78 C, 89%; c) TBAF, 85%; d) PPh_{3.} NH₄SCN, DEAD,CH₂Cl_{2.} 89%; e) CH₃CN, 88%.

Scheme 4. Completion of the total synthesis of (-)-fasicularin (1).



Scheme 5. Completion of the total synthesis of (-)-lepadiformine A (2).

Conclusions

In summary, we have successfully developed a Zn-mediated allylation of a chiral, aliphatic, cyclic, *N-tert*-butanesulfinyl ketimine for the stereoselective total synthesis of lepadiformine A and fasicularin. This flexible strategy could also provide an amenable approach to the synthesis of other analogues of the azaspirotricyclic alkaloids of this family.

Experimental Section

Explanation of Acronyms: CBS: Corey–Bakshi–Shibata reagent, *m*-CPBA: *m*-chloroperbenzoic acid, (DHQD)₂PHAL: 1,4-bis(9-O-dihydroquinine)phthalazine, DIBAL-H: diisobutyaluminum hydride, DMP: Dess–Martin periodinane, DMAP: 4-(dimethylamino)pyridine, NaHMDS: sodium hexamethyldisilazane, NMO: *N*-methylmorpholine oxide, PCC: pyridinium chlorochromate, TPAP: tetra*n*-propylammonium perruthenate, TBDPS: *tert*-butyldiphenylsilyl.

Ketimine 9: To chiral ketone 10 (2.0 g, 8.61 mmol) in distilled THF (20 mL), were added (R)-tert-butanesulfinamide (1.46 g, 12.07 mmol) and Ti(OEt)₄ (3.6 mL, 17.17 mmol) at room temperature, and the reaction mixture was refluxed overnight. After the mixture was cooled to room temperature, the reaction was quenched with anhydrous MeOH (8 mL) and saturated aqueous NaHCO₃ (0.2 mL) and stirred for 2 h. After the mixture was filtered and concentrated, the reaction mixture was diluted with ethyl acetate (EA) and dried with anhydrous MgSO₄. After the mixture was filtered and concentrated, it was purified by flash chromatography [petroleum ether (PE)/EA, 5:1) to give ketimine 9 as a colorless oil (2.0 g) and 5-epi-9 (200 mg) in a combined yield of 76%. $R_{\rm f}$ = 0.42 (PE/EA = 3:1). $[a]_D^{22}$ = -42.6 (c = 0.96, CHCl₃). IR (neat): $\tilde{v} = 1625 \text{ cm}^{-1}$. ¹H NMR (300 MHz, CDCl₃): $\delta = 7.33-7.26 \text{ (m, 5)}$ H), 4.45 (s, 2 H), 3.52 (t, J = 6.5 Hz, 2 H), 3.45 (d, J = 12.9 Hz, 1 H), 2.56–2.44 (m, 1 H), 2.20–1.99 (m, 4 H), 1.73–1.48 (m, 5 H), 1.21 (s, 9 H) ppm. ¹³C NMR (75 MHz, CDCl₃): δ = 190.0, 138.3, 128.2, 127.5, 127.4, 72.7, 67.8, 56.0, 45.8, 35.2, 34.5, 30.7, 28.6, 25.0, 22.0 ppm. ESI-MS: calcd. for C₁₉H₂₉NO₂S 335.2; found *m*/*z* [M + Na]⁺ 358.2. HRMS (MALDI): *m*/*z* 358.1811. [M + Na]⁺; calcd. for C₁₉H₂₉NO₂SNa 358.1817.

5-*epi*-9: $R_{\rm f} = 0.35$ (PE/EA = 3:1). $[a]_{D}^{22} = -112.6$ (c = 0.24, CHCl₃). IR (neat): $\tilde{v} = 1618$ cm⁻¹. ¹H NMR (300 MHz, CDCl₃): $\delta = 7.37$ -7.27 (m, 5 H), 4.48 (AB, J = 12.6, 15.8 Hz, 2 H), 3.55–3.51 (m, 2 H), 3.33 (d, J = 13.5 Hz, 1 H), 2.58–2.49 (m, 1 H), 2.41–2.31 (m, 1 H), 2.27–2.16 (m, 1 H), 2.10–2.05 (m, 1 H), 1.94 (br., 1 H), 1.78 (br., 1 H), 1.65–1.51 (m, 3 H), 1.43–1.29 (m, 1 H), 1.22 (s, 9 H) ppm. ¹³C NMR (75 MHz, CDCl₃): $\delta = 190.3$, 138.1, 128.0, 127.3, 127.2, 72.6, 67.8, 55.6, 45.6, 34.5, 34.0, 30.6, 27.7, 24.4, 21.9 ppm. ESI-MS: calcd. for C₁₉H₂₉NO₂S 335.2; found *m/z* [M + Na]⁺; calcd. for C₁₉H₂₉NO₂SNa 358.1817.

Sulfinamide 8: To ketimine 9 (620 mg, 1.85 mmol) in distilled THF (10 mL) were added activated Zn powder (482 mg, 7.41 mmol) and a solution of allyl bromide (0.64 mL, 7.41 mmol) in THF (5 mL), and the mixture was stirred for 1 h at room temperature. The reaction mixture was quenched with brine (15 mL) and saturated aqueous NaHCO₃ (2 mL). After the mixture was stirred for 1 h, the mixture was filtered through celite, and the filtrate was separated and extracted with EA. The combined organic phases were washed with brine and dried with anhydrous Na₂SO₄. After the mixture was filtered and concentrated, the mixture was purified by flash chromatography (PE/EA, 5:1) to afford 8 as a colorless oil (622 mg) in 89% yield. $R_{\rm f} = 0.35$ (PE/EA, 3:1). $[a]_{\rm D}^{25} = -45.4$ (c = 0.81, CHCl₃). IR (neat): $\tilde{v} = 3290$, 1631, 1453, 1261, 1068 cm⁻¹. ¹H NMR (500 MHz, CDCl₃): δ = 7.34–7.25 (m, 5 H), 5.94–5.79 (m, 1 H), 5.24-5.14 (m, 2 H), 4.49 (AB, J = 12.0, 28.4 Hz, 2 H), 3.55-3.88 (m, 3 H), 2.57 (dd, J = 6.9, 14.4 Hz, 1 H), 2.08–1.93 (m, 8 H), 1.39-1.25 (m, 4 H), 1.19 (s, 9 H) ppm. ¹³C NMR (75 MHz, CDCl₃): $\delta = 138.5, 133.1, 128.3, 127.6, 127.5, 120.2, 73.0, 69.2, 59.6, 56.2,$ 40.8, 37.2, 36.6, 29.1, 28.1, 24.8, 22.9 ppm. ESI-MS: calcd. for C₂₂H₃₅NO₂S 377.2; found *m*/*z* [M + Na]⁺ 400.3. HRMS (MALDI): m/z 400.2281, m/z [M + Na]⁺ calcd. for C₂₂H₃₅NO₂SNa 400.2286.

Primary Amine 11: To 8 (3.45 g, 9.15 mmol) in distilled MeOH (69 mL) was added HCl (4 M in dioxane, 23 mL). After the mixture was stirred for 10 min, saturated aqueous K₂CO₃ (23 mL) was added, and the mixture was stirred for 10 min. After the mixture was concentrated, distilled water (50 mL) was added, and the mixture was extracted with EA ($15 \text{ mL} \times 3$). The combined organic phases were washed with brine and dried with anhydrous Na₂SO₄. After the mixture was filtered and concentrated, the mixture was purified by flash chromatography (PE/EA, 5:1) to afford primary amine 11 as a colorless oil (2.3 g) in 92% yield. $R_{\rm f} = 0.47$ (CH₂Cl₂/ EtOH, 9:1). $[a]_{D}^{27} = -32.6$ (*c* = 1.77, CHCl₃). IR (neat): $\tilde{v} = 1637$ cm⁻¹. ¹H NMR (300 MHz, CDCl₃): δ = 7.35–7.24 (m, 5 H), 5.91– 5.77 (m, 1 H), 5.12–5.05 (m, 2 H), 4.51 (AB, J = 11.2, 15.9 Hz, 2 H), 3.58–3.43 (m, 2 H), 2.32 (dd, J = 7.5, 13.5 Hz, 1 H), 2.06–1.89 (m, 2 H), 1.74–1.38 (m, 6 H), 1.26–1.03 (m, 6 H) ppm. $^{13}\mathrm{C}$ NMR $(75 \text{ MHz}, \text{CDCl}_3)$: $\delta = 134.1, 128.3, 127.5, 127.4, 118.0, 72.81, 69.4,$ 53.3, 45.1, 38.3, 37.4, 29.2, 28.0, 25.3, 22.6 ppm. ESI-MS: calcd. for $C_{18}H_{27}NO$ 273.2; found m/z [M + H]⁺ 274.2. HRMS (MALDI): m/z 274.2165, m/z [M + H]⁺ calcd. for C₁₈H₂₈NO 274.2171.

Alcohol 12: To primary amine 11 (2.3 g, 8.42 mmol) in distilled CH_2Cl_2 (50 mL) were added Et_3N (4.65 mL, 33.70 mmol), DMAP (103 mg, 0.84 mmol) and TsCl (3.21 g, 16.84 mmol) at room temperature. After the mixture was stirred overnight, the reaction was quenched by the addition of distilled water (25 mL). The layers

were separated, and the aqueous layer was extracted with EA (10 mL \times 3). The combined organic phases were washed with brine and dried with anhydrous Na₂SO₄. After the mixture was filtered and concentrated, the mixture was purified by flash chromatography (PE/EA, 25:1 to 3:1) to afford the protected amine product (3.6 g) in quantitative yield.

To the protected amine product (3.6 g, 7.83 mmol) in distilled THF (30 mL) was added 9-BBN (3.8 g, 31.15 mmol). After the mixture was stirred for 24 h, saturated aqueous NaHCO₃ (60 mL) and H_2O_2 (30 mL) were added, and the mixture was stirred for 3 h. The mixture was extracted with EA (20 mL × 3). The combined organic phases were washed with brine and dried with anhydrous Na₂SO₄. After the mixture was filtered and concentrated, the mixture was purified by flash chromatography (PE/EA, 3:1) to afford primary alcohol **12** as a colorless oil (3.0 g) in 86% yield.

$$\begin{split} R_{\rm f} &= 0.46 \; (\text{PE/EA}, \, 1:1). \; [a]_{2}^{24} = -33.0 \; (c = 1.17, \, \text{CHCl}_3). \; \text{IR (neat):} \\ \tilde{\nu} &= 3288, \, 1599, \, 1495, \, 1455 \; \text{cm}^{-1}. \; ^{1}\text{H} \; \text{NMR} \; (300 \; \text{MHz}, \; \text{CDCl}_3): \, \delta \\ &= 7.77 \; (d, \, J = 7.5 \; \text{Hz}, \, 2 \; \text{H}), \, 7.36 - 7.21 \; (m, \, 7 \; \text{H}), \, 5.70 \; (s, \, 1 \; \text{H}), \, 4.54 - \\ 4.45 \; (m, \, 2 \; \text{H}), \, 3.62 - 3.35 \; (m, \, 4 \; \text{H}), \, 2.39 \; (s, \, 3 \; \text{H}), \, 2.05 - 1.94 \; (m, \, 3 \; \text{H}), \\ 1.75 - 1.11 \; (m, \, 11 \; \text{H}), \, 0.91 - 0.84 \; (m, \, 1 \; \text{H}) \; \text{ppm}. \; ^{13}\text{C} \; \text{NMR} \; (75 \; \text{MHz}, \\ \text{CDCl}_3): \; \delta = 142.5, \; 141.1, \; 138.4, \; 129.4, \; 128.3, \; 127.8, \; 127.5, \; 126.8, \\ 72.9, \; 69.5, \; 62.7, \; 42.5, \; 32.1, \; 29.6, \; 28.6, \; 28.1, \; 25.4, \; 24.4, \; 22.1, \; 21.4 \\ \text{ppm}. \; \text{ESI-MS: calcd. for } C_{25}\text{H}_{35}\text{NO}_4\text{S} \; 445.2; \; \text{found} \; m/z \; [\text{M} \; + \\ \text{Na}]^+ \; 468.3. \; \text{HRMS} \; (\text{ESI): } m/z \; 446.2360, \; m/z \; [\text{M} \; + \; \text{H}]^+ \; \text{calcd. for} \\ C_{25}\text{H}_{36}\text{NO}_4\text{S} \; 446.2365. \end{split}$$

Lactam 13: To alcohol 12 (2.5 g, 5.62 mmol) in CH₂Cl₂ (50 mL) were added 4 Å molecular sieves (2.5 g) and NMO (1.58 g, 13.49 mmol). After the mixture was stirred for 15 min, TPAP (631 mg, 1.80 mmol) was added, and the mixture was stirred for an additional 15 min. After the mixture was filtered and concentrated, the mixture was purified by flash chromatography (PE/EA, 3:1) to afford lactam 13 as a white solid (1.98 g) in 80% yield. $R_{\rm f} = 0.41$ (PE/EA, 3:1); m.p. 121–123 °C $[a]_D^{19} = +17.8$ (c = 1.00, CHCl₃). IR (KBr): $\tilde{v} = 1728 \text{ cm}^{-1}$. ¹H NMR (300 MHz, CDCl₃): $\delta = 8.00 \text{ (d,}$ J = 8.3 Hz, 2 H), 7.37–7.23 (m, 7 H), 4.54 (AB, J = 11.8, 16.2 Hz, 2 H), 3.61 (t, J = 6.8 Hz, 2 H), 2.78–2.67 (m, 2 H), 2.40–2.30 (m, 5 H), 2.15–2.06 (m, 1 H), 1.90–1.68 (m, 6 H), 1.40–1.27 (m, 3 H), 1.13–1.04 (m, 1 H) ppm. ¹³C NMR (75 MHz, CDCl₃): δ = 174.8, 144.5, 138.5, 136.4, 129.2, 129.0, 128.3, 127.7, 127.5, 75.2, 72.9, 68.6, 40.7, 37.8, 30.0, 29.8, 28.9, 25.1, 23.3, 21.6 ppm. ESI-MS: calcd. for $C_{25}H_{31}NO_4S$ 441.2; found $m/z [M + Na]^+$ 464.3. HRMS (MALDI): m/z 464.1866, m/z [M + Na]⁺ calcd. for C₂₅H₃₁NO₄SNa 464.1871.

Sulfonamide 7: To lactam 13 (1.73 g, 3.92 mmol) in CH₂Cl₂ (40 mL) was added DIBAL-H (9.8 mL, 1.0 M in cyclohexane, 9.8 mmol) at -78 °C, and the mixture was stirred for another 2 h. The reaction was warmed to -40 °C, quenched with MeOH (1.0 mL) and saturated aqueous Rochelle's salt (100 mL) was added. The mixture was stirred for 1 h at room temperature, the layers were separated, and the aqueous layer was extracted with EA. The combined organic phases were washed with brine and dried with anhydrous Na₂SO₄. After the mixture was filtered and concentrated, the mixture was used directly in the next step without further purification.

To PPh₃CH₃Br (5.6 g, 15.68 mmol) in THF (20 mL) was slowly added NaHMDS (7.65 mL, 2.0 μ in THF, 15.3 mmol) at 0 °C, and stirring was continued for 0.5 h. After the mixture was cooled to -78 °C, a solution of the reduced lactam, prepared above, in THF (20 mL) was added, and stirring was continued at this temperature for 20 min. The reaction mixture was allowed to stir at 0 °C and then slowly warmed to room temperature and stirred for 2 h. The reaction was quenched with saturated aqueous NH₄Cl and extracted with EA. The combined organic phases were washed with

brine and dried with anhydrous Na₂SO₄. After the mixture was filtered and concentrated, the mixture was purified by flash chromatography (PE/EA, 9:1) to afford 7 as a colorless oil (1.6 g) in 92% yield over two steps. $R_{\rm f} = 0.45$ (PE/EA, 5:1). $[a]_{24}^{24} = -22.7$ (c = 1.29, CHCl₃). IR (neat): $\tilde{v} = 3284$, 1094 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): $\delta = 7.74$ (d, J = 8.2 Hz, 2 H), 7.36–7.26 (m, 5 H), 7.23 (d, J = 7.9 Hz, 2 H), 5.75–5.62 (m, 1 H), 5.22 (s, 1 H), 4.96–4.90 (m, 2 H), 4.52 (AB, J = 11.8, 19.4 Hz, 2 H), 3.56–3.42 (m, 2 H), 2.39 (s, 3 H), 2.07–1.07 (m, 15 H) ppm. ¹³C NMR (75 MHz, CDCl₃): $\delta = 142.6$, 141.0, 138.30, 138.26, 129.4, 128.3, 127.7, 127.5, 126.9, 114.8, 72.9, 69.4, 62.7, 42.8, 31.7, 31.4, 29.8, 28.7, 26.8, 24.4, 22.0, 21.4 ppm. ESI-MS: calcd. for C₂₆H₃₅NO₃S 441.2; found m/z [M + Na]⁺ calcd. for C₂₆H₃₅NO₃SNa 464.2235.

14 and 15: To $K_3Fe(CN)_6$ (3.22 g, 9.8 mmol), NaHCO₃ (821 mg, 9.7 mmol), K_2CO_3 (1.35 g, 9.8 mmol), (DHQD)₂PHAL (102 mg, 0.13 mmol), OsO₄ (9 mg, 0.035 mmol) and MeSO₂NH₂ (310 mg, 3.3 mmol) in *t*BuOH/H₂O (15:18, 33 mL), which was stirred at room temperature for 20 min, was added 7 (1.44 g, 3.26 mmol) in *t*BuOH (5 mL) at 0 °C, and stirring was continued overnight. The reaction was then quenched with Na₂SO₃ (6.0 g), and distilled water (10 mL) was added to dissolve the solid. The mixture was extracted with EA (3×10 mL), and the combined organic phases were washed with brine and dried with anhydrous MgSO₄. After the mixture was filtered and concentrated, the mixture was purified by flash chromatography (CH₂Cl₂/EtOH, 20:1) to afford the diol (1.54 g) in 99% yield, and the crude product was directly used in the next step.

To the diol (1.7 g, 3.57 mmol) in CH_2Cl_2 (40 mL) were added Bu_2SnO (889 mg, 3.57 mmol), Et_3N (0.59 mL, 4.28 mmol) and TsCl (715 mg, 3.75 mmol), and the mixture was stirred overnight. EA (40 mL) was added to dilute the mixture. After the mixture was filtered and concentrated, the mixture was purified by flash chromatography (PE/EA, 3:2) to afford **14** (1.49 g) and **15** (532 mg) in a combined yield of 89%.

14: $R_{\rm f} = 0.42$ (PE/EA, 1:1). $[a]_{19}^{19} = -15.2$ (c = 0.91, CHCl₃). IR (neat): $\tilde{v} = 3524$, 3305 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): $\delta = 7.80$ (d, J = 8.2 Hz, 2 H), 7.73 (d, J = 7.9 Hz, 2 H), 7.37–7.27 (m, 7 H), 7.22 (d, J = 8.2 Hz, 2 H), 5.63 (s, 1 H), 4.52–4.44 (m, 2 H), 3.91 (dd, J = 9.9, 3.2 Hz, 1 H), 3.80 (dd, J = 6.5, 10.0 Hz, 1 H), 3.62 (s, 1 H), 3.49–3.42 (m, 1 H), 3.40–3.35 (m, 1 H), 2.64 (d, J = 4.1 Hz, 1 H), 2.45 (s, 3 H), 2.39 (s, 3 H), 1.89–1.10 (m, 15 H) ppm. ¹³C NMR (100 MHz, CDCl₃): $\delta = 145.1$, 142.6, 140.9, 138.2, 132.5, 130.0, 129.4, 128.4, 127.9, 127.8, 127.6, 126.8, 73.7, 72.9, 69.5, 69.3, 62.4, 42.4, 32.0, 29.5, 28.3, 27.5, 25.5, 24.1, 22.0, 21.6, 21.5 ppm. ESI-MS: calcd. for C₃₃H₄₃NO₇S₂ 629.2; found *m*/*z* [M + Na]⁺ 652.1. HRMS (MALDI): *m*/*z* 652.2373, *m*/*z* [M + Na]⁺ calcd. for C₃₃H₄₃NO₇S₂Na 652.2379.

15: $R_{\rm f} = 0.5$ (PE/EA, 1:1). $[a]_{\rm D}^{19} = -13.7$ (c = 1.30, CHCl₃). IR (neat): $\tilde{v} = 3524$, 3299 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): $\delta = 7.81$ (d, J = 7.9 Hz, 2 H), 7.71 (d, J = 8.0 Hz, 2 H), 7.38–7.27 (m, 7 H), 7.22 (d, J = 8.2 Hz, 2 H), 5.58 (s, 1 H), 4.55 (AB, J = 11.7, 14.7 Hz, 2 H), 3.91 (dd, J = 9.6, 3.2 Hz, 1 H), 3.82 (dd, J = 6.7, 10.0 Hz, 1 H), 3.69 (s, 1 H), 3.52–3.42 (m, 1 H), 3.40–3.35 (m, 1 H), 2.45 (s, 4 H), 2.39 (s, 3 H), 1.82–1.11 (m, 15 H) ppm. ¹³C NMR (100 MHz, CDCl₃): $\delta = 145.1$, 142.7, 140.8, 138.2, 132.5, 130.0, 128.58, 128.4, 127.9, 127.8, 127.6, 126.8, 73.6, 72.9, 69.4, 69.3, 62.4, 42.8, 31.9, 29.8, 28.69, 27.4, 25.6, 24.3, 21.9, 21.6, 21.4 ppm. ESI-MS: calcd. for C₃₃H₄₃NO₇S₂ 029.2; found m/z [M + Na]⁺ calcd. for C₃₃H₄₃NO₇S₂Na 652.2379.

Eurjoc european Journal of Organic Chemist

Azaspirocyclic 6: To 14 (866 mg, 1.38 mmol) in THF (17 mL) was added NaH (133 mg, 3.33 mmol) at 0 °C, and the mixture was stirred for 12 h. The reaction was quenched with saturated aqueous $NH_4Cl (5.0 \text{ mL})$ and extracted with EA (3 × 10 mL). The combined organic phases were washed with brine and dried with MgSO₄. After the mixture was filtered and concentrated, the mixture was purified by flash chromatography (PE/EA, 5:3) to afford 6 (560 mg) in 89% yield. $R_{\rm f} = 0.56$ (PE/EA, 5:4); m.p. 120–122 °C. $[a]_{\rm D}^{24} =$ +41.1 (c = 0.98, CHCl₃). IR (KBr): $\tilde{v} = 3480$, 1153, 1095 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): δ = 7.79 (d, J = 7.9 Hz, 2 H), 7.37–7.31 (m, 5 H), 7.22 (d, J = 8.2 Hz, 2 H), 4.43 (s, 2 H), 3.80–3.62 (m, 3 H), 3.47–3.39 (m, 1 H), 3.22–3.14 (m, 1 H), 2.87 (s, 1 H), 2.59 (td, J = 3.2, 12.9 Hz, 1 H), 2.42–2.37 (m, 1 H), 2.10 (s, 1 H), 2.07–1.60 (m, 9 H), 1.25–0.96 (m, 6 H) ppm. ¹³C NMR (75 MHz, CDCl₃): δ = 143.1, 138.5, 138.4, 129.5, 128.4, 127.65, 127.56, 127.3, 75.7, 73.1, 69.2, 65.9, 63.2, 42.4, 39.7, 32.2, 31.1, 30.4, 28.0, 25.2, 24.6, 21.5 ppm. ESI-MS: calcd. for $C_{26}H_{35}NO_4S$ 457.2; found m/z [M + Na]⁺ 480.1. HRMS (MALDI): *m*/*z* 480.2179, *m*/*z* [M + Na]⁺ calcd. for C₂₆H₃₅NO₄SNa 480.2184.

Primary Alcohol 17: To 6 (400 mg, 1.38 mmol) in DMF (1.0 mL) was added imidazole (239 mg, 3.51 mmol) and DMAP (11 mg, 0.09 mmol) at 0 °C, and the mixture was stirred overnight at room temperature. The reaction was quenched with distilled water (5.0 mL) and extracted with EA (3×5 mL). The combined organic phases were washed with brine and dried with MgSO₄. After the mixture was filtered and concentrated, the mixture was directly used in the next step without further purification. To the crude material, dissolved in MeOH (12 mL), was added Pd(OH)₂/C under an atmosphere of H₂, and the mixture was stirred overnight. After the mixture was filtered and concentrated, the mixture was purified by flash chromatography (PE/EA, 10:1 to 2:1 to 1:4) to afford 17 (500 mg) in 94% yield over two steps. $R_{\rm f} = 0.69$ (PE/EA, 2:1); m.p. 140–142 °C. $[a]_{D}^{24} = -24.6$ (c = 1.10, CHCl₃). IR (KBr): $\tilde{v} = 3523 \text{ cm}^{-1}$. ¹H NMR (300 MHz, CDCl₃): $\delta = 7.62-7.58 \text{ (m, 6)}$ H), 7.44–7.38 (m, 6 H), 7.14 (d, J = 7.6 Hz, 2 H), 4.03–3.96 (m, 1 H), 3.71 (d, J = 9.6 Hz, 1 H), 3.63 (br., 1 H), 3.45-3.36 (m, 2 H), 2.47-2.29 (m, 5 H), 2.08-1.04 (m, 11 H), 1.00 (s, 9 H), 0.96-0.82 (m, 2 H) ppm. ¹³C NMR (100 MHz, CDCl₃): δ = 142.7, 140.0, 135.63, 135.61, 133.44, 133.40, 129.73, 129.69, 129.3, 127.67, 127.65, 127.2, 75.0, 64.4, 62.5, 61.2, 42.2, 39.8, 34.1, 31.8, 30.3, 26.9, 26.1, 25.3, 24.5, 21.4, 19.2 ppm. ESI-MS: calcd. for C₃₅H₄₇NO₄SSi 605.3; found *m*/*z* [M + Na]⁺ 628.2. HRMS (MALDI): m/z 628.2887, m/z [M + Na]⁺ calcd. for C₃₅H₄₇NO₄S-SiNa 628.2894.

Mesylate 18: To 17 (467 mg, 0.77 mmol) in CH₂Cl₂ (12 mL) were added Et₃N (0.32 mL, 2.32 mmol), DMAP (11 mg, 0.09 mmol) and MsCl (0.12 mL, 1.55 mmol) at 0 °C, and the mixture was stirred for 1 h. The reaction was quenched with distilled water (5 mL) and extracted with EA (3×5.0 mL). The combined organic phases were washed with brine and dried with MgSO4. After the mixture was filtered and concentrated, the mixture was directly used in the next step without further purification. $R_{\rm f} = 0.44$ (PE/EA, 3:1). $[a]_{\rm D}^{24} =$ -18.7 (c = 0.70, CHCl₃). IR (neat): $\tilde{v} = 1094$ cm⁻¹. ¹H NMR (300 MHz, CDCl₃): δ = 7.60–7.38 (12 H), 7.15 (d, J = 8.2 Hz, 2 H), 4.24-4.13 (m, 1 H), 4.04-3.92 (m, 2 H), 3.68 (dd, J = 3.7, 10.1 Hz, 1 H), 3.35 (t, J = 9.6 Hz, 1 H), 3.00 (s, 3 H), 2.51–2.42 (m, 2 H), 2.37 (s, 3 H), 2.05-1.83 (m, 2 H), 1.69-1.60 (m, 8 H), 1.25-1.02 (m, 2 H), 0.90 (s, 9 H), 0.88-0.81 (m, 1 H) ppm. ¹³C NMR (100 MHz, CDCl₃): δ = 143.0, 135.6, 133.4, 129.8, 129.7, 129.5, 127.69, 127.66, 127.03, 74.6, 68.5, 64.1, 62.2, 42.1, 39.9, 37.4, 31.8, 30.4, 30.2, 29.7, 26.9, 26.3, 25.1, 24.4, 21.5, 19.2 ppm. ESI-MS: calcd. for $C_{36}H_{49}NO_6S_2Si$ 683.3; found m/z [M + Na]⁺ 706.1.

FULL PAPER

HRMS (MALDI): m/z 706.2663, m/z [M + Na]⁺ calcd. for C₃₆H₄₉NO₆S₂SiNa 706.2668.

Nitrile 19: To 18 (the mixture described above) in DMSO (15 mL) was added KCN (301 mg, 4.63 mmol), and the mixture was heated at 60-70 °C for 5 h. After the mixture was cooled to room temperature, the reaction was quenched with distilled water (15 mL) and extracted with EA (3×10 mL). The combined organic phases were washed with brine and dried with MgSO₄. After the mixture was filtered and concentrated, the mixture was purified by flash chromatography (PE/EA, 4:1) to afford 19 (430 mg) in 91% yield over two steps. $R_{\rm f} = 0.5$ (PE/EA = 4:1); m.p. 137–139 °C. $[a]_{\rm D}^{24} =$ -40.4 (c = 0.84, CHCl₃). IR (KBr): $\tilde{v} = 2235$ cm⁻¹. ¹H NMR (300 MHz, CDCl₃): δ = 7.65–7.56 (m, 6 H), 7.47–7.41 (m, 6 H), 7.20 (d, J = 7.6 Hz, 2 H), 4.03–3.95 (br., 1 H), 3.87 (dd, J = 3.0, 10.0 Hz, 1 H), 3.45 (dd, J = 9.3, 10.2 Hz, 1 H), 2.45–2.37 (m, 4 H), 2.22-2.12 (m, 2 H), 2.02-1.88 (m, 2 H), 1.78-1.71 (m, 6 H), 1.61-1.11 (m, 3 H), 1.05 (s, 9 H), 1.00–0.88 (m, 2 H) ppm. ¹³C NMR $(100 \text{ MHz}, \text{ CDCl}_3): \delta = 143.3, 139.8, 135.7, 135.6, 133.4, 129.8,$ 129.7, 129.6, 127.71, 127.67, 126.7, 119.7, 74.2, 64.7, 62.6, 42.4, 42.3, 31.8, 29.2, 27.3, 26.9, 26.1, 25.0, 24.3, 21.5, 19.2, 15.0 ppm. ESI-MS: calcd. for $C_{36}H_{46}NO_3SSi 614.3$; found $m/z [M + H]^+$ 615.3. HRMS (ESI): m/z 615.3071, m/z [M + Na]⁺ calcd. for C₃₆H₄₇NO₃SSi 615.3077.

Ketone 5: To 19 (420 mg, 0.684 mmol) in Et₂O (15 mL) was added $n-C_6H_{13}MgBr$ (5.5 mL, 1.0 M in Et₂O, 5.5 mmol), and the mixture was refluxed for 5 h. After the mixture was cooled to room temperature, the reaction was quenched with saturated aqueous NH₄Cl (10 mL) and extracted with EA (3×5.0 mL). The combined organic phases were washed with brine and dried with MgSO₄. After the mixture was filtered and concentrated, the mixture was purified by flash chromatography (PE/EA, 20:1) to afford 5 (432 mg) in 90% yield. $R_{\rm f} = 0.5$ (PE/EA, 9:1). $[a]_{\rm D}^{24} = -40.4$ (c = 0.84, CHCl₃). IR (neat): $\tilde{v} = 2235 \text{ cm}^{-1}$. ¹H NMR (300 MHz, CDCl₃): $\delta = 7.62$ – 7.61 (m, 4 H), 7.55 (d, J = 8.2 Hz, 2 H), 7.42–7.36 (m, 6 H), 7.12 (d, J = 8.2 Hz, 2 H), 3.96-3.91 (m, 1 H), 3.86 (dd, J = 3.3, 9.5 Hz,1 H), 3.44 (t, J = 9.3 Hz, 1 H), 2.48-2.38 (m, 1 H), 2.34 (s, 1 H), 2.28-2.22 (m, 3 H), 2.13-1.67 (m, 24 H), 1.04 (s, 9 H), 0.95-0.86 (m, 4 H) ppm. ¹³C NMR (100 MHz, CDCl₃): δ = 210.9, 142.4, 140.1, 135.64, 135.62, 133.5, 129.7, 129.6, 129.3, 127.64, 127.61, 127.0, 75.0, 64.7, 62.6, 42.6, 42.5, 42.2, 40.7, 31.8, 31.6, 29.8, 28.9, 26.9, 26.2, 25.3, 24.8, 24.5, 23.7, 22.5, 21.4, 19.2, 14.0 ppm. ESI-MS: calcd. for C₄₂H₅₉NO₄SSi 701.4; found *m*/*z* [M + Na]⁺ 724.2. HRMS (ESI): m/z 724.3826, m/z [M + Na]⁺ calcd. for C42H59NO4SSiNa 724.3832.

Tertiary Amine 20: To 5 (12 mg, 17.1 µmol) in DME (1.0 mL) was added Na/naphthalene [which was prepared by stirring fresh sodium and naphthalene (13.2 mg, 103.1 µmol) in DME (1.0 mL) for 30 min at room temperature] at -68 °C, and the mixture was stirred for 10 min at this temperature. The reaction was quenched with saturated aqueous NH₄Cl (2.0 mL) and stirred at room temperature for awhile. The layers were separated, and the aqueous layer was extracted with EA (3×3.0 mL). The combined organic phases were washed with brine and dried with anhydrous K₂CO₃. After the mixture was filtered and concentrated, the mixture was purified by flash chromatography (PE/EA, 9:1) to afford the desired enamine (7 mg) in 77% yield.

To the enamine (7 mg, 12.66 μ mol) in THF/MeOH (1:2.5, 1.4 mL) was added CeCl₃·7H₂O (29 mg, 76.6 μ mol), and the mixture was stirred until the CeCl₃·7H₂O dissolved. To the mixture was added NaBH₄ (3 mg, 78.9 μ mol) at –78 °C, and the mixture was stirred at this temperature for 15 min. The reaction was quenched with distilled water (1.0 mL) and extracted with EA (3 × 2.0 mL). The com-

bined organic phases were washed with brine and dried with anhydrous MgSO₄. After the mixture was filtered and concentrated, the mixture was purified by flash chromatography (PE/EA, 5:1) to afford the desired tertiary amine **20** (6 mg) in 89% yield. $R_f = 0.46$ (PE/EA, 2:1). $[a]_{D}^{28} = -3.7$ (c = 0.80, CHCl₃). IR (neat): $\tilde{v} = 3070$, 2928, 2857, 1462, 1428 cm⁻¹. ¹H NMR (500 MHz, CDCl₃): $\delta = 7.67$ (d, J = 6.9 Hz, 2 H), 7.42–7.33 (m, 6 H), 3.82 (dd, J = 5.9, 10.0 Hz, 1 H), 3.53 (dd, J = 9.6, 10.1 Hz, 1 H), 3.03–2.98 (m, 1 H), 2.38–2.23 (m, 1 H), 2.11–2.04 (m, 3 H), 1.80–1.06 (m, 26 H), 1.05 (s, 9 H), 0.88 (t, J = 6.9 Hz, 3 H) ppm. ¹³C NMR (75 MHz, CDCl₃): $\delta = 135.6$, 129.5, 127.55, 127.52, 70.1, 67.6, 67.2, 62.6, 43.6, 38.6, 37.1, 32.0, 31.4, 30.8, 29.7, 27.0, 26.62, 26.55, 26.5, 26.3, 26.0, 24.7, 23.6, 22.6, 19.3, 14.1 ppm. ESI-MS: calcd. for C₃₅H₅₃NOSi 531.4; found m/z [M + H]⁺ 532.4. HRMS (MALDI): m/z 532.3969, m/z [M + H]⁺ calcd. for C₃₅H₅₄NOSi 532.3975.

Amino Alcohol 21: To 20 (19 mg, 0.036 mmol) in THF (1.0 mL) was added TBAF (0.18 mg, 0.18 mmol) at 0 °C, and the mixture was stirred for 48 h at room temperature. The reaction was quenched with iced water (1 mL) and extracted with EA $(3 \times 2.0 \text{ mL})$. The combined organic phases were washed with brine and dried with anhydrous K_2CO_3 . After the mixture was filtered and concentrated, the mixture was purified by flash chromatography (Et₂O/MeOH, 20:1) to afford **21** (9 mg) in 85% yield. $R_{\rm f}$ = 0.44 (EA/MeOH, 5:1). $[a]_D^{25} = +11.2$ (c = 0.50, MeOH). IR (neat): $\tilde{v} = 3346 \text{ cm}^{-1}$. ¹H NMR (400 MHz, CDCl₃): $\delta = 3.53 \text{ (dd, } J = 6.0,$ 10.5 Hz, 1 H), 3.38-3.32 (m, 1 H), 3.16-3.10 (m, 1 H), 2.84 (br., 1 H), 2.24 (br., 1 H), 1.97-1.88 (m, 2 H), 1.78-1.52 (m, 7 H), 1.44-1.08 (m, 18 H), 0.87 (t, J = 6.8 Hz, 3 H) ppm. ¹³C NMR (100 MHz, $CDCl_3$): $\delta = 67.2, 66.6, 64.9, 60.6, 41.8, 39.0, 37.6, 31.9, 31.2, 29.5, <math>\delta = 67.2, 66.6, 64.9, 60.6, 41.8, 39.0, 37.6, 31.9, 31.2, 29.5, \delta = 67.2, \delta =$ 27.8, 27.0, 26.3, 25.8, 25.7, 24.5, 23.8, 22.6, 14.1 ppm. ESI-MS: calcd. for $C_{19}H_{35}NO$ 293.2; found m/z [M + H]⁺ 294.2. HRMS (MALDI): m/z 294.2791, m/z [M + H]⁺ calcd. for C₁₉H₃₆NO 294.2797.

Fasicularin (1):^[4g] To PPh₃ (49 mg, 0.187 mmol), NH₄SCN (17 mg, 0.223 mmol) and DEAD (30 µL, 0.190 mmol) was added dropwise 21 (9 mg, 0.031 mmol) in CH₂Cl₂ (1.0 mL) at room temperature, and the mixture was stirred overnight at room temperature. The reaction was quenched with saturated aqueous NaHCO₃ (5 mL) and extracted with EA (3×2.0 mL). The combined organic phases were washed with brine and dried with anhydrous K₂CO₃. After the mixture was filtered and concentrated, the mixture was purified by flash chromatography (PE/EA, 100:1 to 10:1 to 3:1) to afford fasicularin (1, 4 mg) and a more polar isomer (4 mg) which was stirred in CH_3CN (1 mL) for 24 h to obtain fasicularin (1, 3.5 mg). $R_{\rm f} = 0.55$. $[a]_{\rm D}^{26} = -4.4$ (c = 0.34, MeOH). IR (neat): $\tilde{v} = 2152$ cm⁻¹. ¹H NMR (400 MHz, [D₅]-Py): δ = 3.63–3.54 (m, 1 H), 3.42 (dd, J = 14.2, 11.9 Hz, 1 H), 3.31 (dd, J = 4.1, 13.2 Hz, 1 H), 2.98–2.90 (m, 1 H), 2.57 (d, J = 13.3 Hz, 1 H), 1.98–1.96 (m, 1 H), 1.91–1.74 (m, 1 H), 1.61–1.06 (m, 24 H), 0.86 (t, J = 6.8 Hz, 3 H) ppm. ¹³C NMR (100 MHz, CDCl₃): δ = 111.6, 56.3, 52.3, 46.5, 46.1, 40.2, 34.3, 34.1, 32.3, 32.1, 30.3, 29.5, 27.7, 27.2, 26.3, 24.0, 22.9, 22.7, 19.3, 14.3 ppm. ESI-MS: calcd. for C₂₀H₃₄N₂S 334.2; found m/z $[M + H]^+$ 335.1. HRMS (MALDI): *m*/*z* 335.2516, *m*/*z* $[M + H]^+$ calcd. for C₂₀H₃₅N₂S 335.2521.

Amino Alcohol 22: To 5 (108 mg, 0.154 mmol) in THF (5.0 mL) was added L-Selectride (1.6 mL, 1 μ in THF, 1.6 mmol) at -78 °C, and the mixture was stirred at this temperature for 5 h. The reaction was quenched with MeOH (1.0 mL) and stirred at room temperature for 0.5 h. After the mixture was concentrated, the mixture was purified by flash chromatography (PE/EA, 10:1 to 3:1) to afford 22 (91 mg) and the epimer of 22 (16 mg) in a combined yield of 99%. $R_f = 0.56$ (PE/EA, 6:1). $[a]_{D}^{25} = -42.2$ (c = 1.00, CHCl₃).

IR (neat): $\tilde{v} = 3548 \text{ cm}^{-1}$. ¹H NMR (300 MHz, CDCl₃): $\delta = 7.60$ (d, J = 7.9 Hz, 6 H), 7.45–7.35 (m, 6 H), 7.15 (d, J = 7.9 Hz, 2 H), 3.97–3.90 (m, 1 H), 3.78 (dd, J = 4.1, 9.9 Hz, 1 H), 3.38 (dd, J =9.7, 9.4 Hz, 1 H), 3.27–3.19 (br., 1 H), 2.48–2.40 (m, 1 H), 2.36 (s, 3 H), 2.23–2.15 (m, 1 H), 2.04–1.76 (m, 3 H), 1.59–1.26 (m, 10 H), 1.03–0.96 (13 H), 0.96 (s, 9 H), 0.89 (t, J = 5.9 Hz, 3 H) ppm. ¹³C NMR (100 MHz, CDCl₃): $\delta = 142.5$, 140.2, 135.64, 135.62, 133.53, 133.49, 129.68, 129.66, 129.3, 127.7, 127.6, 127.3, 75.6, 72.4, 64.5, 62.4, 42.9, 42.5, 37.4, 35.2, 31.84, 31.76, 30.1, 29.7, 29.3, 27.3, 26.9, 26.3, 25.4, 24.6, 22.6, 21.4, 19.2, 14.1 ppm. ESI-MS: calcd. for C₄₂H₆₁NO₄SSi 703.4; found *m*/*z* [M + Na]⁺ 726.3. HRMS (ESI): *m*/*z* 726.3983, *m*/*z* [M + Na]⁺ calcd. for C₄₂H₆₁NO₄SSiNa 726.3988.

Minor Epimer of 22: $R_{\rm f} = 0.36$ (PE/EA, 6:1). $[a]_{\rm D}^{22} = -42.2$ (c = 2.18, CHCl₃). IR (neat): $\bar{v} = 3545$ cm⁻¹. ¹H NMR (300 MHz, CDCl₃): $\delta = 7.59-7.56$ (m, 6 H), 7.45-7.26 (m, 6 H), 7.09 (d, J = 8.2 Hz, 2 H), 3.99-3.92 (m, 1 H), 3.61 (dd, J = 3.7, 9.6 Hz, 1 H), 3.34-3.27 (m, 2 H), 2.48-2.40 (m, 1 H), 2.34 (s, 3 H), 2.28-2.25 (m, 1 H), 2.05-1.17 (m, 24 H), 1.02 (s, 9 H), 1.02-0.95 (m, 1 H), 0.89 (t, J = 6.9 Hz, 3 H), 0.87-0.81 (m, 1 H) ppm. ¹³C NMR (100 MHz, CDCl₃): $\delta = 142.4$, 140.1, 135.6, 135.57, 133.5, 133.4, 129.7, 129.6, 129.3, 127.63, 127.60, 127.1, 75.9, 70.9, 64.3, 62.3, 42.3, 42.2, 37.9, 34.6, 31.9, 31.8, 29.8, 29.4, 26.9, 26.8, 26.4, 25.9, 25.4, 24.7, 22.6, 21.4, 19.1, 14.1 ppm. ESI-MS: calcd. for C₄₂H₆₁NO₄SSi 703.4; found m/z [M + Na]⁺ 726.2. HRMS (ESI): m/z 726.3983, m/z [M + Na]⁺ calcd. for C₄₂H₆₁NO₄SSiNa 726.3988.

Amine 23: To 22 (58 mg, 0.082 mmol) in DME (4.0 mL) was added Na/naphthalene [which was prepared by stirring fresh sodium and naphthalene (64 mg, 0.5 mmol) in DME (1.0 mL) for 30 min at room temperature] at -68 °C, and the mixture was stirred for 5 min at this temperature. The reaction was quenched with saturated aqueous NH₄Cl (4.0 mL) and stirred at room temperature for awhile. The layers were separated, and the aqueous layer was extracted with EA (3×3.0 mL). The combined organic phases were washed with brine and dried with anhydrous K₂CO₃. After the mixture was filtered and concentrated, the mixture was purified by flash chromatography (CH₂Cl₂/MeOH, 20:1) to afford the desired amino alcohol (45 mg) in quantitative yield.

To the above amino alcohol (40 mg, 0.073 mmol) in CH₂Cl₂ (12 mL) was added PPh₃ (115 mg, 0.439 mmol), CBr₄ (144 mg, 0.439 mmol) and DMAP (1.3 mg, 0.01 mmol) at 0 °C, and the mixture was stirred at room temperature for 24 h. The reaction was quenched with saturated aqueous NaHCO₃ (5.0 mL) and extracted with EA $(3 \times 2.0 \text{ mL})$. The combined organic phases were washed with brine and dried with anhydrous K₂CO₃. After the mixture was filtered and concentrated, the mixture was purified by flash chromatography (PE/EA, 3:1) to afford light yellow oil 23 (38 mg) in 98% over two steps. $R_{\rm f}$ = 0.29 (PE/EA, 3:1). $[a]_{\rm D}^{28}$ = -21.8 (c = 0.3, CHCl₃). IR (neat): \tilde{v} = 3072, 1466, 1427 cm⁻¹. ¹H NMR (500 MHz, CDCl₃): δ = 7.71–7.64 (m, 4 H), 7.41–7.34 (m, 6 H), 3.75 (dd, J = 3.7, 8.2 Hz, 1 H), 3.26 (dd, J = 9.6, 8.7 Hz, 1 H),3.21-3.14 (m, 1 H), 3.01-2.93 (m, 1 H), 2.12-2.03 (m, 2 H), 1.72-1.61 (m, 6 H), 1.51–1.40 (m, 4 H), 1.31–1.07 (m, 14 H), 1.05 (s, 9 H), 1.01-0.90 (m, 1 H), 0.85 (t, J = 6.8 Hz, 3 H) ppm.

(-)-Lepadiformine A (2): To 23 (38 mg, 0.072 mmol) in THF (1.0 mL) was added TBAF (0.36 mL, 0.36 mmol) at 0 °C, and the mixture was stirred for 24 h at room temperature. The reaction was quenched with iced water (1.0 mL) and extracted with EA (3×2.0 mL). The combined organic phases were washed with brine and dried with anhydrous K₂CO₃. After the mixture was filtered and concentrated, the mixture was purified by flash chromatography (Et₂O/MeOH, 20:1) to afford 2 as a colorless oil 2 (20 mg)



in 95% yield. $R_{\rm f} = 0.35$ (EA/MeOH, 4:1). $[a]_{28}^{28} = -14.7$ (c = 0.25, MeOH). IR (neat): $\tilde{v} = 3331 \text{ cm}^{-1}$. ¹H NMR (500 MHz, CDCl₃): $\delta = 3.42-3.30$ (m, 2 H), 3.23 (d, J = 8.7 Hz, 1 H), 3.16–3.10 (m, 1 H), 1.78–1.07 (m, 27 H), 1.02–0.94 (m, 1 H), 0.87 (t, J = 6.8 Hz, 3 H) ppm. ¹³C NMR (100 MHz, CDCl₃): $\delta = 67.6$, 62.2, 58.5, 53.4, 39.8, 38.1, 33.9, 31.8, 30.5, 29.5, 28.0, 27.6, 27.5, 26.2, 24.2, 23.2, 22.5, 22.4, 14.0 ppm. ESI-MS: calcd. for C₁₉H₃₅NO 293.2; found m/z [M + H]⁺ 294.1. HRMS (MALDI): m/z 294.2791, m/z [M + H]⁺ calcd. for C₁₉H₃₆NO 294.2797.

Supporting Information (see also the footnote on the first page of this article): Procedures for the synthesis of **6**, **16** and **24**, HPLC analysis of **10**, X-ray crystal structure of **16** and **24** and copies of of the ¹H and ¹³C NMR spectra of the prepared compounds.

Acknowledgments

Research support from the National Natural Science Foundation of China (no. 20172064, 203900502 and 20532040), the Basic Research Program (973 Program) of China (no. 2010CB833204), and Excellent Young Scholars Foundation of National Natural Science Foundation of China (no. 20525208) is acknowledged.

- a) J. F. Biad, S. Guyot, C. Roussakis, J. F. Verbist, J. Vercauteren, J. F. Weber, K. Boukef, *Tetrahedron Lett.* **1994**, *35*, 2691;
 b) M. Juge, N. Grimaud, J. F. Biad, M. P. Sauviat, M. Nabil, J. F. Verbist, J. Y. Petit, *Toxican.* **2001**, *39*, 1231; c) M. P. Sauviat, J. Vercauteren, N. Grimaud, M. Juge, M. Nabil, J. Y. Petit, J. F. Biad, *J. Nat. Prod.* **2006**, *69*, 558; d) A. D. Patil, A. J. Freyer, R. Reichwein, B. Carte, L. B. Killmer, L. Faucette, R. K. Johnson, D. J. Faukkner, *Tetrahedron Lett.* **1997**, *38*, 363.
- [2] S. Dutta, H. Abe, S. Aoyagi, C. Kibayashi, K. S. Gates, J. Am. Chem. Soc. 2005, 127, 15004.
- [3] For excellent reviews, see: a) S. M. Weinreb, Acc. Chem. Res. 2003, 36, 59; b) C. Kibayashi, S. Aoyagi, H. Abe, Bull. Chem. Soc. Jpn. 2003, 76, 2059; c) P. Schär, S. Cren, P. Renaud, Chimia 2006, 60, 131; d) S. M. Weinreb, Chem. Rev. 2006, 106, 2531; e) J. Liu, R. P. Hsung, Chemtracts: Org. Chem. 2005, 18, 321.
- [4] For the total synthesis of lepadiformine A, see: a) H. Abe, S. Aoyagi, C. Kibayashi, J. Am. Chem. Soc. 2000, 122, 4583; b) T. J. Greshock, R. L. Funk, Org. Lett. 2001, 3, 3511; c) P. Sun, C. Sun, S. M. Weinreb, Org. Lett. 2001, 3, 3507; d) H. Abe, S. Aoyagi, C. Kibayashi, Angew. Chem. Int. Ed. 2002, 41, 3017; e) P. Sun, C. Sun, S. M. Weinreb, J. Org. Chem. 2002, 67, 4337; f) J. Liu, R. P. Hsung, S. D. Peters, Org. Lett. 2004, 6, 3989; g) H. Abe, S. Aoyagi, C. Kibayashi, J. Am. Chem. Soc. 2005, 127, 1473; h) P. Schär, P. Renaud, Org. Lett. 2006, 8, 1569; i) J. J. Caldwell, D. Craig, Angew. Chem. Int. Ed. 2007, 46, 2631; j) B. Lygo, H. M. E. Kirton, C. Lumley, Org. Biomol. Chem. 2008, 6, 3085. For formal syntheses of lepadiformine A, see: k) M. Lee, T. Lee, E.-Y. Kim, H. Ko, D. Kim, S. Kim, Org. Lett. 2006, 8, 745; 1) H. Mihara, T. Shibuguchi, A. Kuramochi, T. Ohshima, M. Shibasaki, Heterocycles 2007, 72, 421. For total syntheses of fasicularin, see: m) J.-H. Maeng, R. L. Funk, Org. Lett. 2002, 4, 331. For the formal synthesis of fasicularin, see: n) M. D. B. Fenster, G. R. Dake, Org. Lett. 2003, 5, 4313; o) M. D. B. Fenster, G. R. Dake, Chem. Eur. J. 2005, 11, 639. For the total syntheses of the Cylindricines family, see: p) B. B. Snider, T. Liu, J. Org. Chem. 1997, 62, 5630; q) J. F. Liu, C. H. Heathcock, J. Org. Chem. 1999, 64, 8263; r) G. A. Molander, M. Rönn, J. Org. Chem. 1999, 64, 5183; s) B. M. Trost, M. T. Rudd, Org. Lett. 2003, 5, 4599; t) T. Arai, H. Abe, S. Aoyagi, C. Kibayashi, Tetrahedron Lett. 2004, 45, 5921; u) S. Canesi, D. Bouchu, M. A. Ciufolini, Angew. Chem. Int. Ed. 2004, 43, 4336; v) J. Liu, J. J. Swidorski, S. D. Peters, R. P. Hsung, J. Org. Chem. 2005, 70, 3898; w) J. Wang, J. J. Swidorski, N. Sydorenko, R. P. Hsung, H. A. Coverdale, J. M. Kuyava, J. Liu, Het-

FULL PAPER

erocycles **2006**, *70*, 423; x) T. Shibuguchi, H. Mihara, A. Kuramochi, S. Sakuraba, T. Ohshima, M. Shibasaki, *Angew. Chem. Int. Ed.* **2006**, *45*, 4635; y) J. J. Swidorski, J. Wang, R. P. Hsung, *Org. Lett.* **2006**, *8*, 777; z) A. C. Flick, M. J. A. Caballero, A. Padwa, *Org. Lett.* **2008**, *10*, 1871.

- [5] a) J. A. Ellman, T. D. Owens, T. P. Tang, Acc. Chem. Res. 2002, 35, 984; b) X.-W. Sun, M.-H. Xu, G.-Q. Lin, Org. Lett. 2006, 8, 4979; c) G.-Q. Lin, M.-H. Xu, Y.-W. Zhong, X.-W. Sun, Acc. Chem. Res. 2008, 41, 831.
- [6] a) H. Wu, H. L. Zhang, G. Zhao, *Tetrahedron* 2007, 63, 6454;
 b) H. L. Zhang, G. Zhao, Y. Ding, B. Wu, J. Org. Chem. 2005, 70, 4954.
- [7] G. Y. Wang, C. W. Zheng, G. Zhao, *Tetrahedron: Asymmetry* 2006, 17, 2074. The preparation of 10 was achieved from the corresponding γ-keto ester by catalyzed CBS reduction in three steps in a 27% overall yield.
- [8] G. C. Liu, D. A. Cogan, T. D. Owens, T. P. Tang, J. A. Ellman, J. Org. Chem. 1999, 64, 1278.
- [9] a) D. A. Cogan, G. C. Liu, J. A. Ellman, *Tetrahedron* 1999, 55, 8883; b) D. A. Cogan, J. A. Ellman, *J. Am. Chem. Soc.* 1999, 121, 268–269.
- [10] H. C. Brown, P. A. Tierney, J. Am. Chem. Soc. 1958, 80, 1552.

- [11] a) W. P. Griffith, S. V. Ley, G. P. Whitecombe, A. D. White, J. Chem. Soc., Chem. Commun. 1987, 1625; b) J. P. Schmidt, S. Beltrán-Rodil, R. J. Cox, D. G. McAllister, M. Reid, R. J. K. Taylor, Org. Lett. 2007, 9, 4041.
- [12] Other oxidative conditions including PCC oxidation, Swern oxidation, and DMP oxidation were also tried but gave inferior results.
- [13] A. Nuhrich, J. Moulines, Tetrahedron 1991, 47, 3075.
- [14] a) H. Tian, X. She, H. Yu, L. Shu, Y. Shi, J. Org. Chem. 2002, 67, 2435; b) M. Palucki, J. G. McCormick, E. N. Jacobsen, Tetrahedron Lett. 1995, 36, 5457.
- [15] a) H. Takahata, M. Kubota, K. Ihara, N. Okamoto, T. Momose, N. Azer, A. T. Eldefrawi, M. E. Eldefrawi, *Tetrahedron: Asymmetry* **1998**, *9*, 3289; b) C. H. Kolb, M. S. Van-Nieuwenhze, K. B. Sharpless, *Chem. Rev.* **1994**, *94*, 2483; c) T. Miyazaki, S. Yokoshima, S. Simizu, H. Osada, H. Tokuyama, T. Fukuyama, *Org. Lett.* **2007**, *9*, 4737.
- [16] T. Hudlicky, X. Tian, K. Königsberger, R. Maurya, J. Rouden, B. Fan, J. Am. Chem. Soc. 1996, 118, 10752–10765.
- [17] M. Periasamy, P. Thirumalaikumar, J. Organomet. Chem. 2000, 609, 137.

Received: December 7, 2009 Published Online: February 9, 2010