

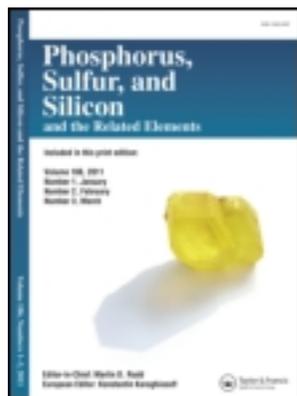
This article was downloaded by: [University of Guelph]

On: 17 September 2012, At: 10:21

Publisher: Taylor & Francis

Informa Ltd Registered in England and Wales Registered Number: 1072954

Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



Phosphorus, Sulfur, and Silicon and the Related Elements

Publication details, including instructions for authors and subscription information:

<http://www.tandfonline.com/loi/gpss20>

Synthesis and Molluscicidal Activity of Phosphorus-Containing Heterocyclic Compounds Derived from 5,6-Bis (4-bromophenyl)-3-hydrazino-1,2,4-triazine

T. E. Ali^a, R. M. Abdel-Rahman^a, F. I. Hanafy^a & S. M. El-Edfawy^a

^a Department of Chemistry, Ain Shams University, Roxy, Cairo, Egypt

Version of record first published: 26 Sep 2008.

To cite this article: T. E. Ali, R. M. Abdel-Rahman, F. I. Hanafy & S. M. El-Edfawy (2008): Synthesis and Molluscicidal Activity of Phosphorus-Containing Heterocyclic Compounds Derived from 5,6-Bis (4-bromophenyl)-3-hydrazino-1,2,4-triazine, *Phosphorus, Sulfur, and Silicon and the Related Elements*, 183:10, 2565-2577

To link to this article: <http://dx.doi.org/10.1080/10426500801967864>

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: <http://www.tandfonline.com/page/terms-and-conditions>

This article may be used for research, teaching, and private study purposes. Any substantial or systematic reproduction, redistribution, reselling, loan,

sub-licensing, systematic supply, or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae, and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand, or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

Synthesis and Molluscicidal Activity of Phosphorus-Containing Heterocyclic Compounds Derived from 5,6-Bis (4-bromophenyl)-3-hydrazino-1,2,4-triazine

T. E. Ali, R. M. Abdel-Rahman, F. I. Hanafy,
and S. M. El-Edfawy

Department of Chemistry, Ain Shams University, Roxy, Cairo, Egypt

Novel N¹-(phosphoryl moiety)-N²-(1,2,4-triazin-3-yl)hydrazines 4, 6, 8, 9 and 12, iminophosphorane 3, iminophosphane 5, 1,2,4-triazinyldiazaphospholine 7, 1,2,4,3-triazaphospholotriazines 2, 10, 11, and 1,2,4-triazino[3,2-c][1,2,4,5] triazaphosphine 13 have been obtained via treatment 5,6-bis(4-bromophenyl)-3-hydrazino-1,2,4-triazine (1) with various polyfunctional phosphorus reagents by stirring at room temperature or refluxing for long time in tetrahydrofuran. Structures of these compounds have been deduced upon the basis of elemental and spectral data. Molluscicidal activity of the prepared compounds against Biomphalaria Alexandrina snails (the intermediate host of Schistosoma mansoni) showed considerable activities.

Keywords Molluscicidal activity; phosphorus; synthesis; 1,2,4-triazine

INTRODUCTION

1,2,4-Triazines have recently been exploited as anti-AIDS,¹ anticancer,² and antimicrobial activities,³ which are attributed to the nature of substituents in 1,2,4-triazine ring. Furthermore, 5,6-diphenyl-3-hydrazino-1,2,4-triazine was shown to be useful starting material for synthesis of great variety of functionalized heteropolycyclic system.⁴ On the other hand, it is known that phosphorus containing heterocyclic compounds are very interesting biologically active.⁵ Synthesis of several phosphono substituted 1,2,4-triazines, triazinylphosphoramidates^{6,7} and series of stable phosphorus ylides incorporating 1,2,4-triazine moiety as

Received 6 November 2007; accepted 18 January 2008.

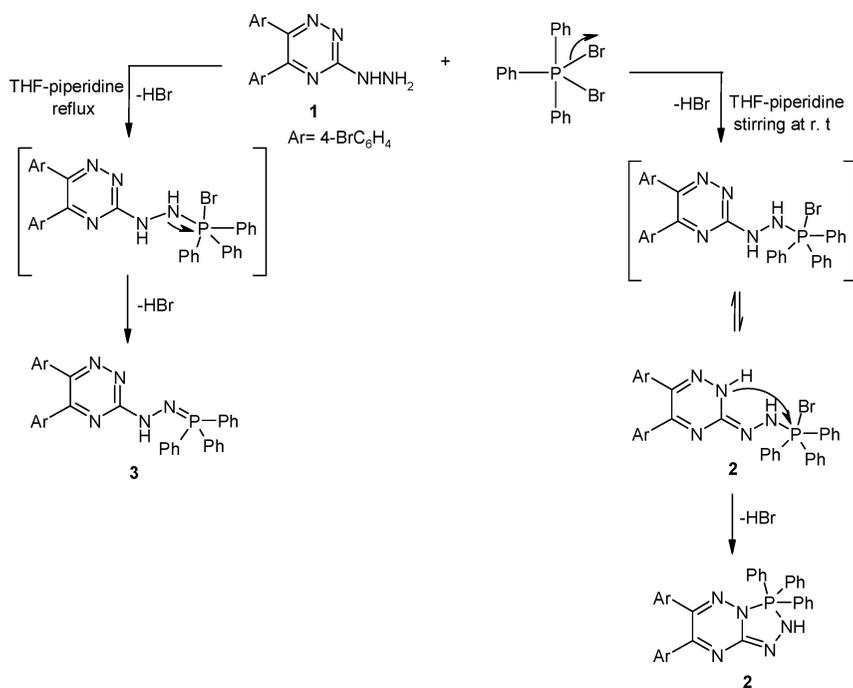
We thank institute of polymers, Bulgarian Academy of Science, for helping to measure ³¹P NMR spectra. Also, we thank Dr. Mortada El-Sayed for helping in testing the prepared compounds as molluscicidal agents in Theodor Bilharz Research Institute, Giza, Egypt.

Address correspondence to T. E. Ali, Department of Chemistry, Faculty of Education, Ain Shams University, Roxy, 11711, Cairo, Egypt. E-mail: tarik_elsayed1975@yahoo.com

accessible intermediates for preparing new 1,2,4-triazine derivatives of therapeutically potencies have been reported.^{8,9} It is conceivable that molecular modification of 1,2,4-triazine rings introducing organophosphorus functionalities might be expected to exhibit the potential activities, depending on the position of the phosphoryl group to 1,2,4-triazine ring. The present work describes synthesis of phosphorus containing fused and isolated heterocyclic systems via reaction of 5,6-bis(4-bromophenyl)-3-hydrazino-1,2,4-triazines (**1**) with some phosphorus reagents in nonpolar solvent under different temperatures. Molluscicidal activity of the prepared compounds against *Biomphalaria Alexandrina* snails (the intermediate host of *Schistosoma mansoni*) showed considerable activities.

RESULTS AND DISCUSSION

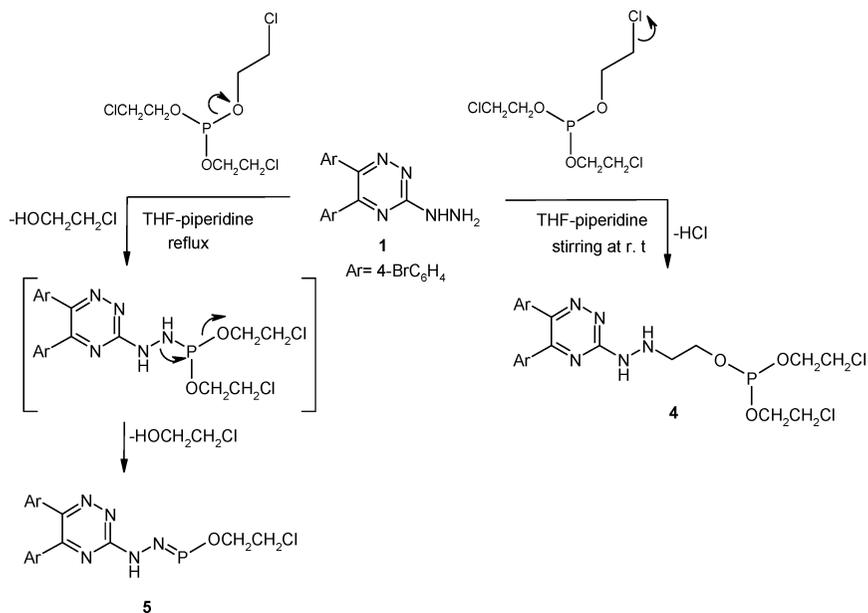
The synthesis of heterocyclic systems containing a 1,2,4-triazine moiety has gained much attention because of their rapid effect in various diseases.¹⁰ Thus, a facile route to synthesise of 6,7-bis(4-bromophenyl)-2,3-dihydro-3,3,3-triphenyl-3- λ^5 -1,2,4,3-triazaphosph-olino[4,5-b][1,2,



SCHEME 1

4]triazine (**2**) was deduced from stirring 5,6-bis(4-bromophenyl)-3-hydrazino-1,2,4-triazine (**1**) with dibromotriphenylphosphorane in THF containing piperidine at room temperature for 24 hours, while that reaction when carried out under refluxing conditions yielded 5,6-bis(4-bromophenyl)-3-[2-(triphenylphosphoranylidene)hydrazine]-1,2,4-triazine (**3**) (Scheme 1). Formation of **2** may be occurred via heterocyclization of the intermediate N^1, N^2 -disubstituted hydrazine through $N-2$ of a triazine moiety, while compound **3** is formed via iminophosphorane mechanism. The ^{31}P NMR spectrum of compound **2** showed singlet signal at δ 29.61 ppm.¹¹ Also, IR spectrum of iminophosphorane **3** showed N=P group as strong band at 1399 cm^{-1} while its ^{31}P NMR spectrum appeared at δ 15.87 ppm.¹²

We note that when phosphorus atom found in trivalent phosphine behaves as powerful nucleophile. Thus, stirring of 3-hydrazinotriazine **1** with tris(2-chloroethoxy)phosphine in THF containing piperidine at room temperature gave the corresponding hydrazidophosphite **4** which is formed via attacking of NH_2 group at the carbon atom which carries partial positive charge according to Horner-Emmons reactions (Scheme 2).^{13,14} On the other hand, refluxing of **1** with tris(2-chloroethoxy)phosphine in boiling THF and piperidine resulted

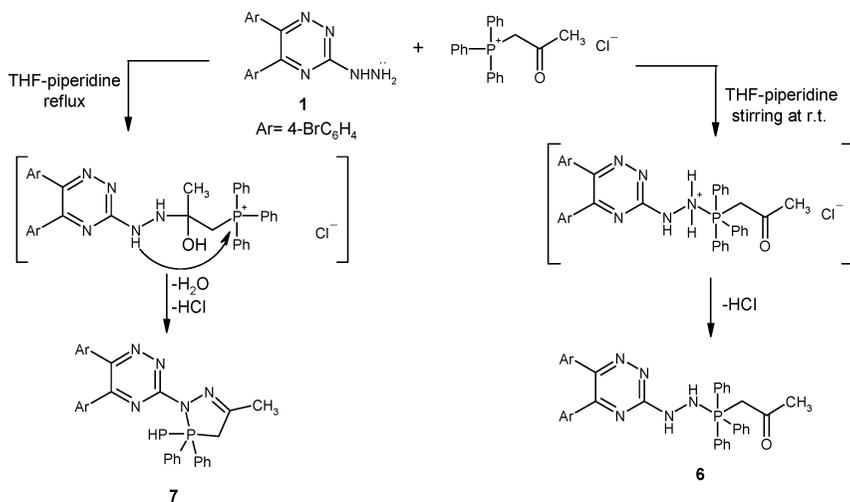


SCHEME 2

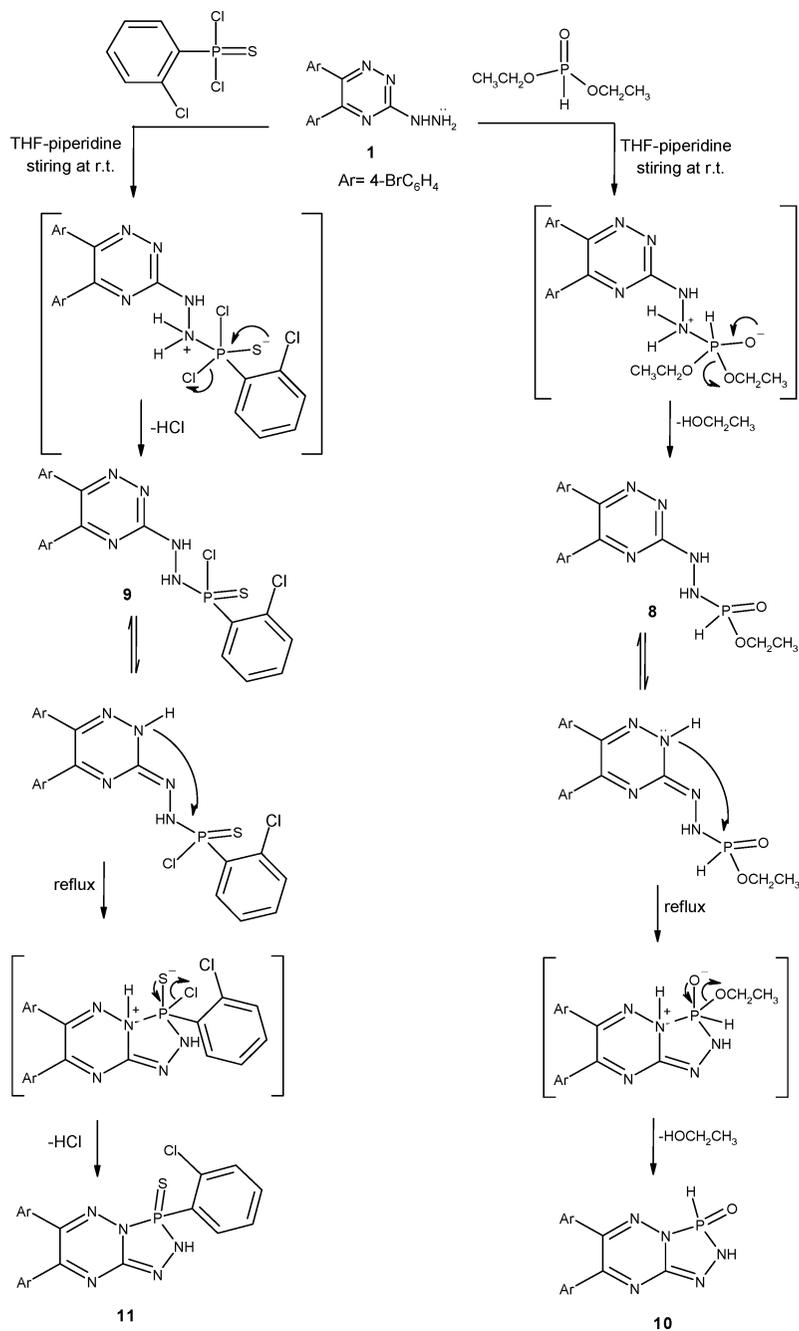
iminophosphane **5** (Scheme 2). The ^1H and ^{31}P NMR spectra of compounds **4** and **5** supported the postulated structures.¹¹

Some five-membered organophosphorus heterocyclic compounds have been synthesized by Mannich-type reactions of guanidine derivatives with phenyldichlorophosphine and ketones and/or aminoheterocyclic with aryl phosphoryl dichloride and aromatic aldehydes.^{15,16} Similarly, phosphorylation of 3-hydrazinotriazine **1** via treatment with acetyltriphenylphosphonium chloride under stirring in THF and few drops of piperidine for 24 hours at room temperature achieved 1-[2-[5,6-bis(4-bromophenyl)-1,2,4-triazin-3-yl]hydrazino}(triphenylphosphoranyl)acetone (**6**) (Scheme 3). Formation of **6** may be due to that phosphorus reagent such as phosphonium salt in which phosphorus atom is more electrophilic than carbon when one of the five groups is good leaving group. Also, the reaction of **1** with acetyltriphenylphosphonium chloride in boiling THF-piperidine afforded 5,6-bis(4-bromophenyl)-3-[(3,3,3-triphenyl)-5-methyl-3,4-dihydro-2H-1,2,3- λ^5 -diazaphosphol-2-yl]-1,2,4-triazine (**7**) (Scheme 3). Formation of **7** may be occurred via nucleophilic attack of hydrazine moiety at carbonyl group to loss molecule of water followed by leaving of HCl.

Abdel-Rahman,³ reported that refluxing of 3-hydrazino-1,2,4-triazine derivatives with α,β -bifunctional groups especially haloketone and/or oxoketone produced hydrazone. Thus, one of the most important of the present investigation is treatment of 3-hydrazinotriazine **1** with diethyl phosphite and/or 2-chlorophenyldichlorothiophosphate in



SCHEME 3



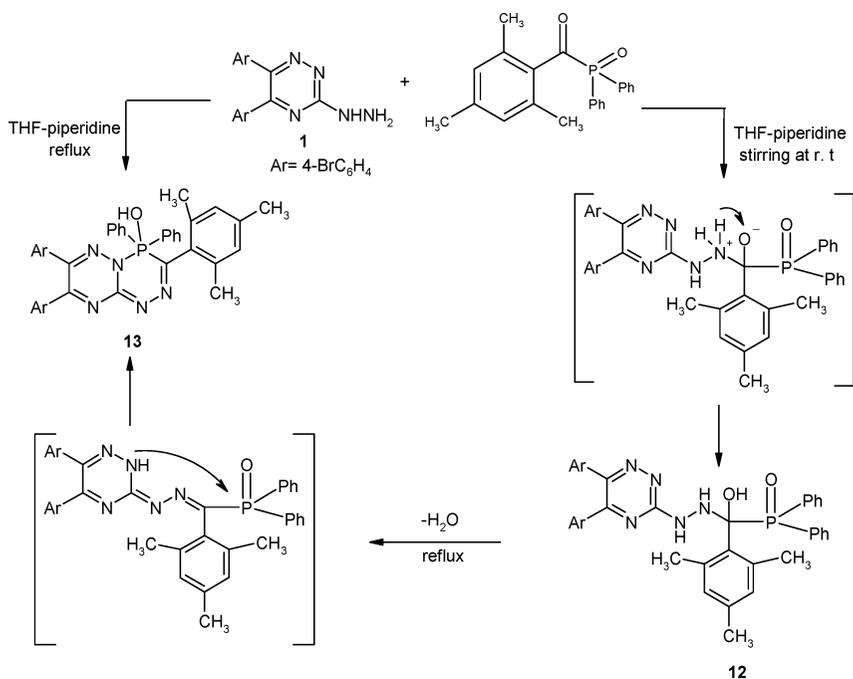
SCHEME 4

THF containing few drops of piperidine at room temperature yielded phosphonohydrazide **8** and phosphonohydrazidothioic acid **9**, respectively (Scheme 4). Also, repeat the same reactions under refluxing led to the direct formation of the 1,2,4,3-triazaphospholo[4,5-b] [1,2,4]triazine derivatives **10** and **11**, respectively. Formation of **10** and **11** may be occurred via nucleophilic attack of the primary NH₂ group of hydrazine on phosphorus atom to remove ethoxy and/or chlorine atom as good leaving groups followed by heterocyclization by loss one molecule of ethanol and/or HCl. The ¹H NMR spectra of **10** and **11** showed signals of NH protons at δ 10.92 and 10.34 ppm, respectively, while their ³¹P NMR spectra recorded signals at δ 14.52 and 47.72 ppm, respectively.^{17–19}

Phosphorus compounds with a nitrogen or oxygen heteroatom on the carbon α to phosphorus very often exhibited interesting biological properties.²⁰ Thus, introducing of P–C–O or P–C–N pattern into heterocyclic structure was achieved by interaction between 3-hydrazinotriazine **1** and diphenyl(2,4,6-trimethylbenzoyl)phosphorus oxide under stirring in THF at room temperature resulting (diphenylphosphoryl){[2-[5,6-bis(4-bromophenyl)-1,2,4-triazin-3-yl]hydrazino} (2',4',6'-trimethoxyphenyl)methanol (**12**), while on refluxing afforded 7,8-bis(4-bromophenyl)-4,4-diphenyl-3-(2',4',6'-trimethylphenyl)-4H-4λ⁵-1,2,4-triazino[3,2-c][1,2,4,5]triazaphosphinin-4-ol (**13**) (Scheme 5). Due to driving force of P=O bond is strong and phenyl groups are bad leaving groups, the nucleophilic attack of hydrazino moiety may be carried out at carbonyl group then P=O group. Structures of **12** and **13** were determined on basis of elemental analysis and spectral methods.

MOLLUSCIDAL ACTIVITY

Schistosomiasis (also called Bilharziasis) is the most important trematode disease in human. It is an endemic disease in many tropical and subtropical regions. The intermediate host of *Schistosoma mansoni*, which affects the intestinal system, is *Biomphalaria Alexandrina* snails (average shell diameter 6–8 mm) in Giza governorate in Egypt. The Snails were adapted to laboratory conditions three weeks before being used in toxicity, tests to be sure that the snails are strong and healthy. Snails were kept in plastic aquaria filled with dechlorinated tap water at room temperature 25–27°C. Dried lettuce leaves were added daily, and the water was changed weakly. Stock solution of the investigated compounds (500 ppm) were dissolved in the least amount of ethanol and diluted with dechlorinated tap water based on weight/volume. Series of dilutions of each compound were prepared.^{21–24} Table I indicates that under the employed experimental conditions the quantitative structure



SCHEME 5

activity relationship studies of these compounds. It is obvious that most of the prepared compounds on the tested snails depend on the type of N–P moieties which inter of their structures. Generally, the effect of the synthesized compounds is attributed to the following.

1. The presence of a phosphorus atom bound to nitrogen, which contains the backbone of the skeleton.
2. The strong effect of the compounds **2**, **7**, **10**, **11**, and **13** on the tested snails appears in the cases of phosphorus heterobicyclic systems, due to facile donation and back donation in *d*-orbital of phosphorus atom.
3. The medial effect of the compounds **4**, **6**, **8**, and **12** on the tested snail exhibited in the cases, which recorded a conjugation system created.
4. A lethal effect of the evaluated compounds **3**, **5**, and **9** on the tested snail was deduced in the case where weak interactions between N–P were a poor donation and back donation is recorded.
5. In most cases, the total electron barrier of molecular structure of the evaluated compounds led to inhibition enzymatic effect on living

TABLE I The Molluscicidal Activity of the Prepared Compounds against *Biomphalaria Alexandrina* Snails

Compound no.	Mortality % of snails at concentration		
	100 ppm	50 ppm	25 ppm
2	100	30	20
3	0	0	0
4	30	20	10
5	0	0	0
6	20	20	10
7	60	20	10
8	30	20	10
9	10	0	0
10	100	20	10
11	100	30	30
12	20	10	10
13	80	30	20
Baylucide (reference standard)	100	100	100

processes for the tested snail via the contribution between apoenzyme and coenzyme, which may be loose that it and can be broken by a simple process of snails.

CONCLUSION

Synthesis of phosphorus containing fused and isolated heterocyclic systems as molluscicidal agents against *Biomphalaria Alexandrina* snails have been achieved via reaction of 5,6-bis(4-bromophenyl)-3-hydrazino-1,2,4-triazines (**1**) with some polyfunctional phosphorus reagents in tetrahydrofuran under stirring and reflux.

EXPERIMENTAL

Melting points were determined on Boetius apparatus and were uncorrected. IR spectra in KBr were recorded on a Perkin-Elmer 293 FT spectrophotometer (λ_{\max} cm⁻¹). UV absorption spectra in DMF were recorded on a Perkin-Elmer, Lambda 4B controller accessory interface. UV-VIS spectrophotometer (λ_{\max} in nm). ¹H NMR spectra were recorded on an EM NMR spectrometer 300 MHz PMR using CDCl₃ as a solvent and TMS as internal reference (chemical shift in ppm). ³¹P NMR spectra of were registered on a Varian Inova 500 MHz spectrometer at room temperature using TMS as internal standard and

85% H_3PO_4 as external reference. Mass spectrum recorded on a Gas Chromatographic GCMSqp 1000 ex Shimadzu instrument at 70 eV. Elemental analysis was carried out at microanalytical center, Cairo university. 5,6-Bis(4-bromophenyl)-3-hydrazino-1,2,4-triazine (**1**) was prepared according to the published method in literature.²⁵

1,2,4,3-Triazaphospholotriazine **2**, *N*¹-(phosphoryl moiety)-*N*²-(1,2,4-triazin-3-yl)hydrazines **4**, **6**, **8**, **9**, and **12**—General Method

A mixture of 3-hydrazinotriazine **1** (5 mmol) and phosphorus reagents such as dibromotriphenylphosphorane, tris(2-chloroethoxy) phosphine, acetonitriletriphenyl phosphonium chloride, diethyl phosphite, 2-chlorophenyldichlorothiophosphate and/or diphenyl-(2,4,6-trimethylbenzoyl)phosphorus oxide (5 mmol) in THF (50 ml) containing few drops of piperidine was stirred at room temperature (25–30°C) for 24 h. The solid separated was collected, washed with diethyl ether and crystallized from a suitable solvent affording the corresponding products **2**, **4**, **6**, **8**, **9**, and **12**, respectively.

6,7-Bis(4-bromophenyl)-2,3-dihydro-3,3,3-triphenyl-3- λ^5 -1,2,4,3-triazaphospholo [4,5-*b*][1,2,4]triazine (**2**)

Brown crystals from ethanol, 23% yield, m.p. = 130–131°C. IR (ν/cm^{-1}): 3420 (NH), 1578 (C=N). ¹H NMR (CDCl_3 , 300 MHz) δ : 7.26–7.70 (m, 23H, Ar-H), 11.60 (br, 1H, NH). ³¹P NMR (CDCl_3 , 101.25 MHz) δ : 29.61 ppm. MS (*m/z*, %): 681 (M+1, 10.23%), 680 (M⁺, 11.36), 550 (18.18), 456 (14.77), 388 (15.91), 362 (13.64), 277 (72.73), 183 (100), 157 (59.09), 75 (59.09). Anal. calcd. for $\text{C}_{33}\text{H}_{24}\text{Br}_2\text{N}_5\text{P}$ (681.37): C, 58.17; H, 3.55; P, 4.55%. Found: C, 57.82; H, 3.19; P, 4.32%.

Bis(2-chloroethyl) *N*-[(5,6-bis(4-bromophenyl)-1,2,4-triazin-3-yl)]hydrazido-phosphite (**4**)

Yellow crystals from ethanol, 35% yield, m.p. = 170–171°C. IR (ν/cm^{-1}): 3298, 3218 (NHNH), 1582 (C=N), 1067, 1006 (P–O–C). ¹H NMR (CDCl_3 , 300 MHz) δ : 1.56–1.66 (m, 4H, CH_2), 1.90 (br, 4H, CH_2), 3.13 (br, 4H, CH_2Cl), 6.32 (s, 1H, NH), 7.20–7.84 (m, 8H, Ar-H), 9.45 (br, 1H, NH). ³¹P NMR (CDCl_3 , 101.25 MHz) δ : 19.50 ppm. MS (*m/z*, %): 654 (M⁺, 75%), 313 (75), 185 (87.50), 156 (43.75), 137 (81.26), 74 (100). Anal. calcd. for $\text{C}_{21}\text{H}_{22}\text{Br}_2\text{Cl}_2\text{N}_5\text{O}_3\text{P}$ (654.13): C, 38.56; H, 3.39; P, 4.74%. Found: C, 37.99; H, 3.11; P, 4.66%.

1-[[2-[5,6-Bis(4-bromophenyl)-1,2,4-triazin-3-yl]hydrazino]{triphenylphosphor-anyl}]acetone (6)

Yellow crystals from methanol, 42% yield, m.p. = 220–222°C. IR (ν/cm^{-1}): 3318, 3209 (NHNH), 1685 (C=O), 1587 (C=N). ^1H NMR (CDCl_3 , 300 MHz) δ : 1.92 (s, 3H, CH_3), 3.35 (br, 2H, CH_2), 6.32 (s, 1H, NH), 7.18–7.89 (m, 23H, Ar-H), 9.62 (br, 1H, NH). ^{31}P NMR (CDCl_3 , 101.25 MHz) δ : 20.43 ppm. MS (m/z , %): 739 (M^+ , 55.56%), 616 (55.56%), 610 (55.56%), 556 (55.56%), 536 (55.56%), 390 (72.22%), 263 (72.22%), 196 (61.11%), 77 (100). Anal. calcd. for $\text{C}_{36}\text{H}_{30}\text{Br}_2\text{N}_5\text{OP}$ (739.46): C, 58.48; H, 4.09; P, 4.19%. Found: C, 57.94; H, 3.85; P, 3.69%.

Ethyl N-[[5,6-bis(4-bromophenyl)-1,2,4-triazin-3-yl]phosphonohydrazide (8)

Yellow crystals from ethanol, 25% yield, m.p. = 133–135°C. UV λ_{max} (ϵ): 210 (0.95), 280 (0.80), 320 (1.00), 385 (0.10). IR (ν/cm^{-1}): 3409 (br, NHNH), 1585 (C=N), 1216 (P=O), 1071 (P-O-C). ^1H NMR (CDCl_3 , 300 MHz) δ : 1.32 (t, 3H, $\text{J} = 6.9$ Hz, CH_3), 3.82 (q, 2H, $\text{J} = 7$ Hz, CH_2), 6.5 (br, 1H, P-H), 6.33 (s, 1H, NH), 7.19–7.86 (m, 8H, Ar-H), 9.52 (br, 1H, NH). ^{31}P NMR (CDCl_3 , 101.25 MHz) δ : 7.28 ppm. MS (m/z , %): 529 ($\text{M} + \text{O}$, 20.41%), 513 (M^+ , 28.57%), 460 (22.45%), 453 (24.49%), 367 (20.41%), 324 (24.49%), 257 (34.69%), 185 (61.22%), 155 (100%), 90 (65.31). Anal. calcd. for $\text{C}_{17}\text{H}_{16}\text{Br}_2\text{N}_5\text{O}_2\text{P}$ (513.13): C, 39.39; H, 3.14; P, 6.04%. Found: C, 38.98; H, 3.02; P, 5.67%.

N-[[5,6-bis(4-bromophenyl)-1,2,4-triazin-3-yl]P-[2-chlorophenyl]phosphono-hydrazidithioic chloride (9)}

Yellow crystals from benzene, 36% yield, m.p. = $>300^\circ\text{C}$. IR (ν/cm^{-1}): 3413, 3278 (NHNH), 1579 (C=N), 755 (P=S). ^1H NMR (CDCl_3 , 300 MHz) δ : 6.38 (s, 1H, NH), 7.20–7.91 (m, 12H, Ar-H), 9.82 (br, 1H, NH). ^{31}P NMR (CDCl_3 , 101.25 MHz) δ : 53.96 ppm. MS (m/z , %): 630 (M^+ , 70.59%), 619 (56.82%), 561 (64.71%), 471 (58.82%), 389 (64.71%), 382 (76.47%), 282 (58.82%), 256 (64.71%), 158 (100%), 122 (70.59%), 800 (70.59%). Anal. calcd. for $\text{C}_{21}\text{H}_{14}\text{Br}_2\text{Cl}_2\text{N}_5\text{PS}$ (630.13): C, 40.03; H, 2.24; P, 4.92%. Found: C, 39.94; H, 2.15; P, 4.52%.

(Diphenylphosphoryl){[2-[5,6-bis(4-bromophenyl)-1,2,4-triazin-3-yl]hydrazino}{2,4,6-trimethoxyphenyl}methanol (12)

Yellow crystals from methanol, 33% yield, m.p. = 224–226°C. UV λ_{max} (ϵ): 210–225 (0.95), 275–285 (1.25), 340–360 (0.15), 390 (0.05). IR (ν/cm^{-1}): 3143 (br, NH, OH), 1583 (C=N), 1211 (P=O). ^1H NMR (CDCl_3 , 300 MHz) δ : 1.56 (s, 9H, CH_3), 2.12 (br, 1H, OH), 6.25 (br, 1H, NH), 7.08–7.88 (m, 20H, Ar-H), 9.32 (br, 1H, NH). ^{31}P NMR (CDCl_3 , 101.25 MHz)

δ : 32.83 ppm. MS (m/z , %): 772 (M+3, 1.66%), 689 (2.15), 622 (1.82), 576 (1.99), 474 (1.99), 406 (2.15), 268 (2.15), 155 (100), 74 (65.23). Anal. calcd. for $C_{37}H_{32}Br_2N_5O_2P$ (769.48): C, 57.75; H, 4.19; P, 4.03%. Found: C, 57.94; H, 4.15; P, 3.72%.

**Iminophosphorane 3, Iminophosphane 5,
1,2,4-Triazinylidiazaphospholine 7,
1,2,4,3-Triazaphospholinotriazine 10, 11, and
1,2,4,5-Triazaphosphinine 13—General Method**

A mixture of 3-hydrazinotriazine **1** (5 mmol) and dibromotriphenylphosphorane, tris(2-chloroethoxy)phosphine, acetyltriphenylphosphonium chloride, diethyl phosphite, 2-chlorophenyldichlorothiophosphate and/or diphenyl-(2,4,6-trimethylbenzoyl)phosphorus oxide (5 mmol) in THF (50 ml) containing few drops of piperidine was refluxed for 12 h. The solvent was removed under vacuum, and the residue was treated with methanol or ethyl acetate and crystallized from a suitable solvent to give the corresponding products **3**, **5**, **7**, **10**, **11**, and **13**, respectively.

5,6-Bis(4-bromophenyl)-3-[2-(triphenylphosphoranylidene)hydrazino]-1,2,4-triazine (3)

Yellow crystals from ethanol, 88% yield, m.p. = 240–242°C. IR (ν/cm^{-1}): 3408 (br, NH), 1584 (C=N), 1399 (N=P). 1H NMR ($CDCl_3$, 300 MHz) δ : 8.45 (br, 1H, NH), 7.23–7.73 (m, 23H, Ar-H). ^{31}P NMR ($CDCl_3$, 101.25 MHz) δ : 15.87 ppm. MS (m/z , %): 683 (M+2, 4.74%), 682 (M+1, 4.01), 681 (M⁺, 3.28), 586 (5.47), 462 (3.28), 402 (4.72), 292 (7.30), 277 (100), 199 (25.55), 77 (42.34). Anal. calcd. for $C_{33}H_{24}Br_2N_5P$ (681.37): C, 58.17; H, 3.55; P, 4.55%. Found: C, 57.94; H, 4.29; P, 4.13%.

2-Chloroethyl{2-[5,6-bis(4-bromophenyl)-1,2,4-triazin-3-yl]hydrazono} phosphinite (5)

Yellow crystals from toluene, 36% yield, m.p. = 170–171°C. UV λ_{max} (ϵ): 205 (0.9), 245 (1.15), 290 (1.00), 320 (0.65), 330 (0.60). IR (ν/cm^{-1}): 3218 (NH), 1583 (C=N), 1334 (N=P), 1066 (P-O-C). 1H NMR ($CDCl_3$, 300 MHz) δ : 1.60 (br, 2H, CH_2), 3.71 (t, 2H, CH_2), 6.33 (s, 1H, NH), 7.19–7.84 (m, 8H, Ar-H). ^{31}P NMR ($CDCl_3$, 101.25 MHz) δ : 14.45 ppm. MS (m/z , %): 529 (M⁺, 5.37%), 527 (5.37), 424 (5.37), 382 (15.12), 313 (4.88), 233 (4.88), 199 (98.05), 155 (100), 76 (62.93). Anal. calcd. for $C_{17}H_{13}Br_2ClN_5OP$ (529.56): C, 38.56; H, 2.47; P, 5.85%. Found: C, 37.98; H, 2.31; P, 5.48%.

5,6-Bis(4-bromophenyl)-3-[(3,3,3-triphenyl)-5-methyl-3,4-dihydro-2H-1,2,3- λ^5 -diazaphosphol-2-yl]-1,2,4-triazine (7)

Brown crystals from methanol, 22% yield, m.p. = 213–214°C. IR (ν/cm^{-1}): 3090 (C-H_{arom}), 2810 (C-H_{aliph}), 1586 (C=N). ¹H NMR (CDCl₃, 300 MHz) δ : 1.54 (s, 3H, CH₃), 3.75 (br, 2H, CH₂), 7.25–8.00 (m, 23H, Ar-H). ³¹P NMR (CDCl₃, 101.25 MHz) δ : 25.90 ppm. MS (m/z, %): 723 (M+2, 58.82%), 722 (M+1, 64.71), 567 (64.71), 534 (94.12), 404 (82.35), 348 (70.59), 279 (82.35), 206 (88.24), 103 (100), 55 (94.12). Anal. calcd. for C₃₆H₂₈Br₂N₅P (721.44): C, 59.94; H, 3.91; P, 4.29%. Found: C, 57.84; H, 3.75; P, 3.90%.

6,7-Bis(4-bromophenyl)-2,3-dihydro-1,2,4,3-triazaphospholo[4,5-b][1,2,4] triazine-3-oxide (10)

Yellow crystals from methanol, 19% yield, m.p. = 198–199°C. UV λ_{max} (ϵ): 220 (1.05), 280 (0.70), 330 (0.75), 390 (0.10). IR (ν/cm^{-1}): 3415 (NH), 1587 (C=N), 1218 (P=O). ¹H NMR (CDCl₃, 300 MHz) δ : 6.61 (br, 1H, P-H), 7.18–7.88 (m, 8H, Ar-H), 10.92 (br, 1H, NH). ³¹P NMR (CDCl₃, 101.25 MHz) δ : 14.52 ppm. MS (m/z, %): 467 (M⁺, 11.96%), 466 (10.87), 419 (11.96), 368 (16.30), 325 (14.13), 269 (16.30), 226 (21.74), 183 (100), 155 (98.91), 76 (75). Anal. calcd. for C₁₅H₁₀Br₂N₅OP (467.06): C, 38.57; H, 2.16; P, 6.63%. Found: C, 38.04; H, 1.95; P, 6.80%.

3-(2-Chlorophenyl)-6,7-bis(4-bromophenyl)-2,3-dihydro-1,2,4,3-triazaphospholo [4,5-b][1,2,4]triazine-3-sulfide (11)

Yellow crystals from methanol, 38% yield, m.p. = 219–220°C. IR (ν/cm^{-1}): 3309 (NH), 1577 (C=N), 753 (P=S). ¹H NMR (CDCl₃, 300 MHz) δ : 7.09–7.78 (m, 12H, Ar-H), 10.34 (br, 1H, NH). ³¹P NMR (CDCl₃, 101.25 MHz) δ : 47.72 ppm. MS (m/z, %): 592 (M-H, 27.50), 460 (25), 37 (35), 292 (37.50), 213 (35), 178 (100), 151 (37.50), 55 (35). Anal. calcd. for C₂₁H₁₃Br₂ClN₅PS (593.67): C, 42.49; H, 2.21; P, 5.22%. Found: C, 42.05; H, 2.15; P, 4.84%.

7,8-Bis(4-bromophenyl)-4,4-diphenyl-3-(2,4,6-trimethylphenyl)-4H-4 λ^5 -1,2,4-triazino[3,2-c][1,2,4,5]triazaphosphinin-4-ol (13)

Brown crystals from methanol, 27% yield, m. p. = 129–130°C. UV λ_{max} (ϵ): 210 (0.45), 240 (0.95), 280 (1.15), 350 (0.50). IR (ν/cm^{-1}): 3414 (br, OH), 1581 (C=N), 1070 (P-OH). ¹H NMR (CDCl₃, 300 MHz) δ : 1.57 (s, 9H, CH₃), 4.19 (s, 1H, OH), 7.09–7.85 (m, 20H, Ar-H). ³¹P NMR (CDCl₃, 101.25 MHz) δ : 25.70 ppm. MS (m/z, %): 752.30 (M+1, 2.43), 644 (3.10), 533 (4.87), 441 (3.32), 334 (6.64), 323 (4.65), 173 (7.30), 82 (100). Anal. calcd. for C₃₇H₃₀Br₂N₅OP (751.47): C, 59.14; H, 4.02; P, 4.12%. Found: C, 58.94; H, 3.88; P, 3.74%.

REFERENCES

- [1] R. M. Abdel-Rahman, *Pharmazie*, **56**, 275 (2001).
- [2] R. M. Abdel-Rahman, *Farmaco*, **46**, 379 (1991).
- [3] R. M. Abdel-Rahman, *Farmaco*, **47**, 319 (1992).
- [4] R. M. Abdel-Rahman, *Trends in Heterocyclic Chemistry (Indian)*, **6**, 126 (1999).
- [5] L. Wan, I. Alkorta, J. Elguero, J. Sun, and W. Zheng, *Tetrahedron*, **63**, 9129 (2007).
- [6] Z. Wang and J. G. Verkade, *Tetrahedron Lett.*, **39**, 9331 (1998).
- [7] R. M. Abdel-Rahman, *Trends in Heterocyclic Chemistry (Indian)*, **8**, 187 (2002).
- [8] H. Wanoff, M. Tzanova, *Arkivoc*, 98 (2003).
- [9] J. L. Pirat, J. Monbrun, D. Virieux, H. J. Cristau, *Tetrahedron* **61**, 7029 (2005).
- [10] Z. El-Gendy, J. M. Morsy, H. A. Allimony, W. R. Abdel-Monem, and R. M. Abdel-Rahman, *Pharmazie*, **56**, 376 (2001).
- [11] Y. Haribaba, M. A. Kumar, K. Srinivasulu, C. S. Reddy, and C. N. Raju, *Arkivoc*, 189 (2006).
- [12] I. Yavari, M. Adib, and L. Hojabri, *Tetrahedron*, **58**, 6895 (2007).
- [13] L. Horner, H. Hoffman, and H. G. Wippel, *Ber.*, **91**, 61 (1958).
- [14] W. S. Wadsworth and W. D. Emmons, *J. Am. Chem. Soc.*, **83**, 1733 (1961).
- [15] R. Lu, H. Yang, and Z. Shang, *Phosphorus, Sulfur Silicon Relat. Elem.*, **108**, 197 (1996).
- [16] J. Zhou, Y. Qiu, K. Feng, and R. Y. Chen, *Synth. Commun.*, **28**, 4673 (1998).
- [17] M. Maffei and G. Buono, *Tetrahedron*, **59**, 8821 (2003).
- [18] M. R. Bryce and R. S. Matthews, *J. Organomet. Chem.*, **325**, 153 (1987).
- [19] J. Hernandez, F. M. Goycoalea, D. Z. Rivera, J. J. Onofre, K. Martinez, J. Lizardi, M. S. Reyes, B. Gordillo, C. V. Contreras, O. G. Barradas, S. C. Sanchez, and Z. Dominguez, *Tetrahedron*, **62**, 2520 (2006).
- [20] H. J. Cristu, J. L. Pirat, D. Virieux, J. Monbrun, C. Ciptadi, and Y. A. Boko, *J. Organomet. Chem.*, **690**, 2472 (2005).
- [21] World Health Organization: Snail Control in Prevention of Bilharziasis, *WHO Monograph Ser.*, **50**, 124 (1965).
- [22] R. Montegazza, R. Tomenasini, R. Fusco, and S. Rossi, *Arch. Intern. Pharmacodynamic*, **95**, 123 (1953).
- [23] N. Mishriky, Y. A. Ibrahim, A. S. Girgis, and N. G. Fawzy, *Pharmazie*, **54**, 738 (1999).
- [24] A. S. Girgis, *Pharmazie*, **55**, 426 (2000).
- [25] P. V. Laakso, R. Robinson, and H. P. Vandrewala, *Tetrahedron*, **1**, 103 (1957).