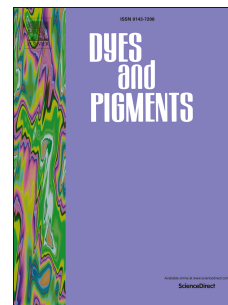


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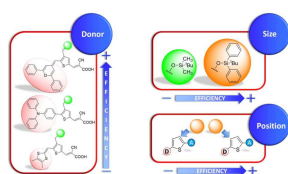
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ACCEPTED MANUSCRIPT

Organic sensitizers bearing a trialkylsilylether group for liquid dye sensitized solar cells

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Abstract

In this work we present the synthesis, optical characterization and performance of five metal-free sensitizers for dye-sensitized solar cells (DSSC). All dyes include, for the first time, a trialkylsilyl ether group in the π -conjugated bridge (a thiophene ring). The influence of different donor unities, like triarylamine (TPA), 4*H*-pyranylidene and dithiafulvene have been evaluated in DSSC with a liquid I₃/I⁻ electrolyte, obtaining the best results with the 4*H*-pyranylidene moiety. The size and the position of the bulky group have a great importance in the efficiency of the final devices.

In order to explain the recombination processes and electron life-time, charge extraction (CE) and transient photovoltaic (TPV) experiments have been carried out.

1. Introduction

Solar energy offers a clean, well-spread and inexhaustible energy source. Although the market is dominated by silicon-based photovoltaic devices, in recent years the interest on alternatives more environment-friendly has increased, specially focused on reducing mass during cell manufacture processes and the thickness of the final device.

Organic Photovoltaic Cells (OPVs)^{1, 2} and particularly Dye Sensitized Solar Cells (DSSCs)^{3, 4} constitute an interesting alternative due to their low manufacturing cost, flexibility of molecular design, light-weight and great aesthetic features, like color and transparency. The key element in a DSSC device is probably the sensitizer dye and over the last years, thousands of new dyes have been investigated. The most efficient organic sensitizers are based on Donor- π spacer-acceptor (D- π -A) structures,⁵ a type of push-pull systems which lead to effective photoinduced intramolecular charge-transfer (ICT). In these systems small variations in the different parts of the molecule (mostly in the donor and the π -bridge) may result in significant changes in the photovoltaic properties. Triphenylamine (TPA)-based metal free organic dyes are one of the most common donor groups in DSSCs,^{6, 7} as it presents several advantages, like a non-planar structure, which suppresses the formation of aggregates. Furthermore, the physical properties can be easily modulated by introduction of bulky or donor groups.⁸⁻¹³ Recently, proaromatic systems like 4*H*-pyranylidene¹⁴⁻¹⁷ and dithiafulvene¹⁸⁻²⁰ have been introduced as alternative and efficient donor unities in DSSCs, but no studies comparing their properties have been reported.

When designing a new sensitizer, one important factor to take into account is related to the minimization of aggregates by π - π stacking. This may be performed by using additives, such as deoxycholic acid (DCA)^{21, 22} or by the introduction of bulky groups both, in the donor or in the π -bridge.^{12, 23-26} However, it is relatively difficult to synthesize dyes with bulky chains in the conjugated spacer, requiring tedious and multiple reaction steps.

Silicon-based dyes have been very promising for DSSCs due to its photo and thermal stability.^{27, 28} Examples of organic sensitizers bearing a dithienosilole (DTS) as a π -linker have been reported with high efficiencies.²⁹⁻³⁴ However, to the best of our knowledge, organic dyes featuring a trialkylsilyl ether ($R_1R_2R_3SiO-$) were never used to preclude the π -aggregation on the TiO_2 in DSSCs. This popular protecting group can be easily introduced from alcohols³⁵ and the overall size and the stability depends on the nature of the R_1 , R_2 or R_3 . Moreover, the silylether group greatly enhances the solubility of the sensitizer that facilitates its adsorption on the TiO_2 surface.

In the present work, a series of five new metal-free organic sensitizers for DSSCs with a trialkylsilyloxy group have been designed, synthesized and characterized. In order to evaluate the donor influence on the photovoltaic properties a TPA and two proaromatic donor unities (4*H*-pyranylidene and dithiafulvene) have been used (Figure 1).

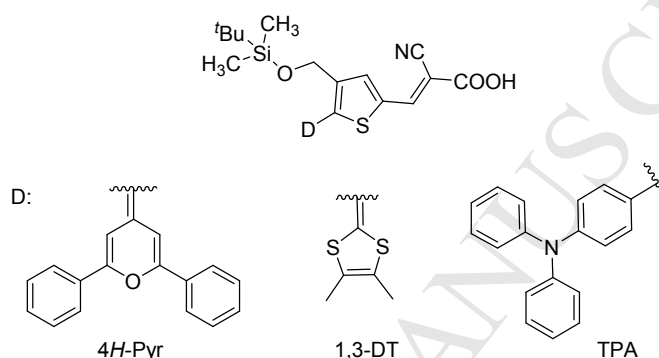


Fig 1. Molecular structures of TBDMS organic sensitizers

The influence of the size and the relative position on the heterocyclic linker of two bulky groups, *tert*-butyldimethyl (TBDMS) or *tert*-butyldiphenyl silylether (TBDPS) are also studied (Figure 2). Finally, the photophysical properties, molecular orbital calculations and the performance of DSSCs based on these organic dyes are reported.

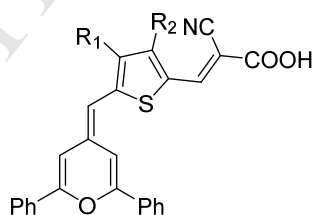


Fig. 2. Molecular structures of 4*H*-pyranylidene organic sensitizers

2. Experimental section

2.1. General information

Infrared measurements were carried out in KBr or neat using a Perkin-Elmer Fourier Transform Infrared 1600 spectrometer. Melting points were obtained on a Gallenkamp apparatus in open capillaries and are uncorrected. 1H - and ^{13}C -NMR spectra were recorded on a Bruker ARX300 or a Bruker AV400 at 300 or

400 MHz and 75 or 100 MHz respectively; δ values are given in ppm (relative to TMS) and J values in Hz. The apparent resonance multiplicity is described as s (singlet), d (doublet), and m (multiplet). ^1H - ^1H COSY and ^1H - ^{13}C -HSQC experiments were recorded on a Bruker ARX300 or a Bruker AV400 at 300 or 400 MHz in order to establish peaks assignment and spatial relationships. Electrospray mass spectra were recorded on a Bruker Microtof-Q spectrometer; accurate mass measurements were achieved using sodium formate as external reference. UV-Visible spectra were recorded with an UV-Vis UNICAM UV4 spectrophotometer. Pulse differential voltammetry measurements were performed with a μ -Autolab type III potentiostat using a glassy carbon working electrode, Pt counter electrode, and Ag/AgCl reference electrode. The experiments were carried out under argon in CH_2Cl_2 , with Bu_4NPF_6 as supporting electrolyte (0.1 mol L^{-1}). Scan rate was 0.01 V s^{-1} , modulation amplitude 0.025 V and modulation time 0.05 s^{-1} .

2.2. Synthesis

4-((*tert*-butyldimethylsilyloxy)methyl)-5-((2,6-diphenyl-4*H*-pyran-4-ylidene)methyl)thiophene-2-carbaldehyde (**7**)

A solution of 2,6-diphenyl-(4*H*-pyran-4-ylidene)-diphenylphosphine oxide **5** (680 mg, 1.56 mmol) in anhydrous THF (12 mL) was prepared, purged with argon and cooled to -78°C . To this solution, *n*-BuLi (1.6 M in hexanes) (1.2 mL, 2.08 mmol) was added dropwise and the resulting mixture was stirred for 15 min. Then 3-((*tert*-butyldimethylsilyloxy)methyl)thiophene-2-carbaldehyde **4** (472 mg; 1.84 mmol) in anhydrous THF (5 mL) was added dropwise and the mixture was warmed to 0°C for 3 hours (TLC monitoring using 10% EtOAc in hexanes). Saturated NH_4Cl solution was added to quench the reaction and the solvent was evaporated under reduced pressure. The organic layer was extracted with EtOAc (2×25 mL) and dried over anhydrous MgSO_4 . After the removal of the solvent, the residue was dissolved in EtOAc/hexanes (3/97) and filtered over silica gel to give the crude *tert*-butyl((2-((2,6-diphenyl-4*H*-pyran-4-ylidene)methyl)thiophen-3-yl)methoxy)dimethylsilane **6** as an intermediate. Then, a solution of 2,2,6,6-tetramethylpiperidine (0.27 mg, 1.58 mmol) in THF (8.4 mL) was prepared, purged with argon and cooled to -78°C . To this solution, $^t\text{BuLi}$ (1.7 M in pentane) (1.01 mL, 1.71 mmol) was added dropwise and the resulting mixture was stirred for 1 h and then a solution of **6** (676.0 mg, 1.41 mmol) in THF (33.6 mL) was added. The resulting mixture was stirred for an additional hour, DMF (0.28 mL, 3.70 mmol) was added dropwise and the mixture was warmed to -30°C . The reaction was quenched by the addition of saturated NH_4Cl solution and the solvent was evaporated under reduced pressure. The organic layer was extracted with EtOAc (2×20 mL) and dried over anhydrous MgSO_4 . After the removal of the solvent, the aldehyde **7** was purified by silica gel column chromatography (6% EtOAc in hexanes). Yield: red oil (583.0 mg, 1.16 mmol; 82%).

IR (neat): cm^{-1} 1648 (C=O), 1573 (C=C). ^1H NMR (400 MHz, CDCl_3): δ (ppm) 9.81 (s, 1H), 7.90–7.76 (m, 4H), 7.69 (s, 1H), 7.55–7.43 (m, 6H), 7.29 (d, $J = 2.0$ Hz, 1H), 6.55 (d, $J = 2.0$ Hz, 1H), 6.07 (s, 1H), 4.75 (s, 2H), 0.96 (s, 9H), 0.14 (s, 6H). ^{13}C NMR (100 MHz, CDCl_3): δ (ppm) 182.1, 155.3, 152.6, 146.6, 139.0, 137.9, 137.0, 132.8, 132.5, 132.5, 130.2, 129.7, 128.9, 128.8, 125.3, 124.7, 109.0, 104.7, 102.9, 59.7, 26.0, 18.4, -5.2 . HRMS (ESI $^+$): m/z calcd for $[\text{C}_{30}\text{H}_{33}\text{O}_3\text{SSi}]^+$: 501.1914, found: 501.1914 $[\text{M}+\text{H}]^+$; calcd for $[\text{C}_{30}\text{H}_{32}\text{NaO}_3\text{SSi}]^+$: 523.1734, found: 523.1721 $[\text{M}+\text{Na}]^+$.

4-((*tert*-butyldimethylsilyloxy)methyl)-5-((4,5-dimethyl-1,3-dithiol-2-ylidene)methyl)thiophene-2-carbaldehyde (**10**)

A solution of tributyl(4,5-dimethyl-1,3-dithiol-2-yl)phosphonium hexafluorophosphate **8** (317.7 mg, 0.66 mmol) and 3-((*tert*-butyldimethylsilyloxy)methyl)thiophene-2-carbaldehyde **4** (131.0 mg, 0.51 mmol) in anhydrous THF (16 mL) was prepared, purged with argon and cooled to -78°C . To this solution, Et_3N (352.4 μL , 2.51 mmol) was added dropwise and the resulting mixture was stirred for 15 min. After the removal of the solvent, the residue was dissolved in EtOAc/hexanes (1:9) and filtered over neutral aluminum oxide to give compound **9** as an intermediate. Then, a solution of **9** in THF (10 mL) was prepared, purged with argon

and cooled to $-45\text{ }^{\circ}\text{C}$. To this solution, *n*-BuLi (1.6 M in hexanes) (0.53 mL, 0.85 mmol) was added dropwise and the resulting mixture was stirred for 1h. DMF (0.09 mL, 1.2 mmol) was added dropwise and then, the mixture was warmed to room temperature over 24h. Saturated NH_4Cl solution was added to quench the reaction and the solvent was evaporated under reduced pressure. The organic layer was extracted with CH_2Cl_2 and dried over anhydrous MgSO_4 . After the removal of the solvent, the aldehyde **10** was purified by silica gel column chromatography (20% ethyl acetate in hexanes). Yield: red oil (136.2 mg, 0.34 mmol; 67%).

IR (neat): cm^{-1} 1649 (C=O), 1492 (C=C). ^1H NMR (400 MHz, CDCl_3): δ (ppm) 9.80 (s, 1H), 7.65 (s, 1H), 6.82 (s, 1H), 4.69 (s, 2H), 2.09 (s, 3H), 2.03 (s, 3H), 0.92 (s, 9H), 0.09 (s, 6H). ^{13}C NMR (100 MHz, CDCl_3): δ (ppm) 181.9, 146.6, 142.5, 137.9, 137.3, 135.8, 124.4, 123.4, 102.9, 59.6, 25.9, 18.3, 13.8, 13.1, -5.2 . HRMS (ESI^+): m/z calcd for $[\text{C}_{18}\text{H}_{27}\text{O}_2\text{S}_3\text{Si}]^+$: 399.0937, found: 399.0961 $[\text{M}+\text{H}]^+$; calcd for $[\text{C}_{18}\text{H}_{26}\text{NaO}_2\text{S}_3\text{Si}]^+$: 421.0756, found: 421.0784 $[\text{M}+\text{Na}]^+$.

4-(3-((*tert*-butyldimethylsilyloxy)methyl)thiophen-2-yl)-*N,N*-diphenylaniline (**13**)

A solution of the stannyl derivative **12** (990.0 mg, 1.87 mmol) and compound **11** (668.2 mg, 1.87 mmol) in anhydrous toluene (24 mL) was prepared, purged with argon for 15 min before addition of $\text{Pd}(\text{PPh}_3)_4$ (112.3 mg, 0.10 mmol). The reaction mixture was refluxed for 15 h under an argon atmosphere. After addition of water (30 mL), the mixture was extracted with toluene (2×30 mL). The organic phase was washed with a saturated aqueous solution of NH_4Cl (30 mL) and water (3×10 mL) and dried over anhydrous MgSO_4 . After the removal of the solvent, the compound was purified by silica gel column chromatography (2% diethyl ether in hexanes). Yield: light yellow oil (392.4 mg, 0.83 mmol; 44%).

IR (neat): cm^{-1} 1592 (C=C). ^1H NMR (300 MHz, CD_2Cl_2): δ (ppm) 7.39–7.33 (m, 2H), 7.32–7.24 (m, 4H), 7.21 (d, $J = 5.2$ Hz, 1H), 7.15–7.02 (m, 9H), 4.68 (s, 2H), 0.89 (s, 9H), 0.05 (s, 6H). ^{13}C NMR (100 MHz, CD_2Cl_2): δ (ppm) 148.1, 148.0, 140.6, 137.6, 130.4, 130.3, 129.9, 128.4, 125.2, 123.7, 59.9, 26.3, 18.8, -5.02 . HRMS (ESI^+): m/z calcd for $[\text{C}_{29}\text{H}_{34}\text{NOSSi}]^+$: 472.2125, found: 472.2109 $[\text{M}+\text{H}]^+$; calcd for $[\text{C}_{29}\text{H}_{33}\text{NNaOSSi}]^+$: 494.1944, found: 494.1930 $[\text{M}+\text{Na}]^+$.

4-((*tert*-butyldimethylsilyloxy)methyl)-5-(4-(diphenylamino)phenyl)thiophene-2-carbaldehyde (**14**)

A solution of **13** (352.4 mg, 0.75 mmol) in anhydrous THF (17 mL) was prepared, purged with argon and cooled to $-30\text{ }^{\circ}\text{C}$. To this solution, *n*-BuLi (1.6 M in hexanes) (0.78 mL, 1.25 mmol) was added dropwise and the resulting mixture was stirred for 1 hour. Then, DMF (173 μL , 2 mmol) was added dropwise and the mixture was warmed to room temperature (TLC monitoring using 2% diethyl ether in hexanes). Saturated NH_4Cl solution was added to quench the reaction and the solvent was evaporated under reduced pressure. The organic layer was extracted with diethyl ether (15×3 mL) and dried over anhydrous MgSO_4 . After removal of the solvent, the aldehyde **14** was purified by column chromatography on silica gel (2% ethyl ether in hexanes). Yield: yellow oil (298.6 mg, 0.50 mmol; 80%).

IR (neat): cm^{-1} 1670 (C=O), 1445 (C=C). ^1H NMR (400 MHz, CD_2Cl_2): δ (ppm) 9.84 (s, 1H), 7.80 (s, 1H), 7.41–7.38 (m, 2H), 7.33–7.29 (m, 4H), 7.16–7.05 (m, 8H), 4.72 (s, 2H), 0.91 (s, 9H), 0.08 (s, 6H). ^{13}C NMR (100 MHz, CD_2Cl_2): δ (ppm) 183.2, 150.3, 149.4, 147.7, 141.0, 139.5, 139.2, 130.4, 130.0, 126.5, 125.7, 124.4, 122.7, 60.0, 26.3, 18.8, -4.5 . HRMS (ESI^+): m/z calcd for $[\text{C}_{30}\text{H}_{34}\text{NO}_2\text{SSi}]^+$: 500.2074, found: 500.2045 $[\text{M}+\text{H}]^+$.

3-(4-((*tert*-butyldimethylsilyloxy)methyl)-5-((2,6-diphenyl-4H-pyran-4-ylidene)methyl) thiophen-2-yl)-2-cyanoacrylic acid (**1**)

To a solution of aldehyde **7** (122 mg, 0.24 mmol) and 2-cyanoacetic acid (33 mg, 0.38 mmol) in chloroform (5 mL) was added piperidine (160 μL ; 1.62 mmol). The mixture was refluxed for 24 hours under an argon atmosphere and then cooled down to room temperature. HCl (1N) was added until pH=3, the organic phase was dried over MgSO_4 and concentrated under reduced pressure. Then, hexane was added and

a solid appeared. This compound was filtered under reduced pressure and washed with a mixture hexane/CH₂Cl₂ (9/1) to afford **1** Yield: dark purple solid (134 mg, 0.24 mmol; 96%).

Mp 215–218 °C. IR (KBr): cm⁻¹ 3100–2600 (COO–H), 2210 (C≡N), 1672 (C=O), 1652 (C=C). ¹H NMR (400 MHz, dmsO-d₆): δ (ppm) 8.33 (s, 1H), 8.02–7.89 (m, 4H), 7.86 (s, 1H), 7.63–7.45 (m, 6H), 7.37 (d, *J* = 1.5 Hz, 1H), 7.11 (d, *J* = 1.5 Hz, 1H), 6.37 (s, 1H), 4.76 (s, 2H), 0.90 (s, 9H), 0.11 (s, 6H). ¹³C NMR (75 MHz, dmsO-d₆): δ (ppm) 164.1, 154.8, 151.9, 145.7, 144.9, 139.3, 132.2, 131.6, 131.5, 130.5, 130.0, 129.9, 129.1, 128.9, 124.8, 124.5, 117.7, 109.1, 105.3, 102.4, 58.7, 25.7, 17.9, –5.4. HRMS (ESI⁺): *m/z* calcd for [C₃₃H₃₄NO₄SSi]⁺: 568.1972, found: 568.1945 [M+H]⁺.

3-(4-((*tert*-butyldimethylsilyloxy)methyl)-5-((4,5-dimethyl-1,3-dithiol-2-ylidene)methyl)thiophen-2-yl)-2-cyanoacrylic acid (2**)**

To a solution of aldehyde **10** (136.4 mg, 0.34 mmol) and 2-cyanoacetic acid (45.4 mg, 0.53 mmol) in chloroform (32 mL) was added piperidine (226.4 μL, 2.25 mmol). The mixture was refluxed for 85h under an argon atmosphere and then cooled down to room temperature. The reaction crude was chromatographed on reverse C18 silica gel (50 % AcONH₄ (20 mM) in acetonitrile). After the addition of acetic acid a solid appeared. This compound was filtered under reduced pressure to afford **2**. Yield: dark purple solid (68.7 mg, 0.15 mmol; 43%).

Mp 170–174 °C. IR (KBr): cm⁻¹ 2211 (C≡N), 1673 (C=O), 1531 (C=C). ¹H NMR (300 MHz, CDCl₃): δ (ppm) 8.21 (s, 1H), 7.68 (s, 1H), 6.84 (s, 1H), 4.69 (s, 2H), 2.14 (s, 3H), 2.07 (s, 3H), 0.92 (s, 9H), 0.09 (s, 6H). ¹³C NMR (75 MHz, CDCl₃): δ (ppm) 169.1, 149.1, 146.5, 145.5, 139.8, 136.6, 130.9, 125.6, 124.5, 116.9, 103.1, 91.5, 59.5, 25.9, 18.3, 14.0, 13.1, –5.3. HRMS (ESI⁺): *m/z* calcd for [C₂₁H₂₈NO₃S₃Si]⁺: 466.0995, found: 466.1040 [M+H]⁺. HRMS (ESI⁺): *m/z* calcd for [C₂₁H₂₇NNaO₃S₃Si]⁺: 488.0815, found: 488.0793 [M+Na]⁺.

3-(4-((*tert*-butyldimethylsilyloxy)methyl)-5-(4-(diphenylamino)phenyl)thiophen-2-yl)-2-cyanoacrylic acid (3**)**

To a solution of aldehyde **14** (139.2 mg; 0.29 mmol) and 2-cyanoacetic acid (45.4 mg; 0.53 mmol) in chloroform (26 mL) was added piperidine (200.6 μL; 1.90 mmol). The mixture was refluxed for 5 days under an argon atmosphere and then cooled down to room temperature. The reaction crude was chromatographed on reverse C18 silica gel (50 % AcONH₄ (20 mM) in acetonitrile). After the addition of acetic acid a solid appeared. This compound was filtered under reduced pressure to afford **3**. Yield: pinkish solid (113.1 mg, 0.20 mmol; 70%).

Mp 196–200 °C. IR (KBr): cm⁻¹ 2223 (C≡N), 1679 (C=O), 1566 (C=C). ¹H NMR (400 MHz, acetone-d₆): δ (ppm) 8.42 (s, 1H), 8.05 (s, 1H), 7.58–7.53 (m, 2H), 7.39–7.33 (m, 4H), 7.18–7.11 (m, 6H) 7.11–7.06 (m, 2H), 4.83 (s, 2H), 0.91 (s, 9H), 0.11 (s, 6H). ¹³C NMR (100 MHz, acetone-d₆): δ (ppm) 167.8, 151.5, 150.9, 148.9, 148.3, 143.4, 140.7, 134.9, 131.8, 131.5, 127.4, 127.2, 126.0, 123.6, 117.8, 100.1, 60.9, 27.3, 19.8, –4.1. HRMS (ESI⁺): *m/z* calcd for [C₃₃H₃₃N₂O₃SSi]⁺: 565.1987, found: 565.2014 [M–H]⁺.

3-((*tert*-butyldiphenylsilyloxy)methyl)thiophene-2-carbaldehyde (18**)**

A solution of *tert*-butyldiphenyl(thiophen-3-ylmethoxy)silane **17** (2.2 g, 6.2 mmol) in anhydrous diethyl ether (10 mL) was prepared, purged with argon and cooled to –10 °C. To this solution, *t*-BuLi (1.7 M in pentane) (4 mL, 6.2 mmol) was added dropwise for a period of 20 min and the resulting mixture was stirred at –10 °C for 1h. Then, DMF (0.75 mL, 9.6 mmol) was added dropwise and the mixture was warmed to 0 °C and stirred at this temperature overnight (TLC monitoring using 20% hexanes in dichloromethane). Saturated NH₄Cl solution was added to quench the reaction and the solvent was evaporated under reduced pressure. The organic layer was extracted with diethyl ether (10 × 3 mL) and dried over anhydrous MgSO₄. After removal of the solvent, the aldehyde **18** was purified by silica gel column chromatography (20% hexanes in dichloromethane). Yield: yellow oil (454.9 mg, 1.20 mmol; 20%).

IR (neat): cm^{-1} 1664 (C=O), 1588 (C=C). ^1H NMR (400 MHz, CD_2Cl_2): δ (ppm) 9.95 (d, $J = 0.9$ Hz, 1H), 7.68–7.66 (m, 5H), 7.48–7.37 (m, 6H), 7.25 (d, $J = 5.0$ Hz, 1H), 5.06 (s, 2H), 1.08 (s, 9H). ^{13}C NMR (100 MHz, CD_2Cl_2): δ (ppm) 182.9, 150.9, 138.1, 136.1, 134.5, 133.5, 130.5, 130.0, 128.5, 61.4, 27.2, 19.7. HRMS (ESI⁺): m/z calcd for $[\text{C}_{22}\text{H}_{25}\text{O}_2\text{SSi}]^+$: 381.1339, found: 381.1313 $[\text{M}+\text{H}]^+$; calcd for $[\text{C}_{22}\text{H}_{24}\text{NaO}_2\text{SSi}]^+$: 403.1158, found: 403.1139 $[\text{M}+\text{Na}]^+$.

4-((*tert*-butyldiphenylsilyloxy)methyl)thiophene-2-carbaldehyde (**19**)

A solution of *tert*-butyldiphenyl(thiophen-3-ylmethoxy)silane **17** (1.6 g, 4.5 mmol) in anhydrous THF (39 mL) was prepared, purged with argon and cooled to -10°C . To this solution, *n*-BuLi (1.6 M in hexanes) (2.8 mL, 4.5 mmol) was added dropwise and the resulting mixture was stirred at -10°C for 1 h and 30 min. Then, DMF (0.54 mL, 6.9 mmol) was added dropwise and the mixture was warmed to 0°C and stirred at this temperature overnight (TLC monitoring using 20% ethyl acetate in hexanes). Saturated NH_4Cl solution was added to quench the reaction and the solvent was evaporated under reduced pressure. The organic layer was extracted with diethyl ether (15×2 mL) and dried over anhydrous MgSO_4 . After removal of the solvent, the aldehyde **19** was purified by silica gel column chromatography (20% ethyl acetate in hexanes). Yield: white solid (465.3 mg, 1.22 mmol; 21%).

Mp $76\text{--}78^\circ\text{C}$. IR (KBr): cm^{-1} 1666 (C=O), 1466 (C=C). ^1H NMR (400 MHz, CD_2Cl_2): δ (ppm) 9.86 (d, $J = 1.3$ Hz, 1H), 7.69–7.65 (m, 5H), 7.59–7.58 (m, 1H), 7.47–7.37 (m, 6H), 4.77 (d, $J = 0.6$ Hz, 2H), 1.08 (s, 9H). ^{13}C NMR (100 MHz, CD_2Cl_2): δ (ppm) 183.5, 144.7, 144.5, 136.1, 135.8, 133.7, 130.8, 130.4, 128.4, 62.2, 27.1, 19.7. HRMS (ESI⁺): m/z calcd for $[\text{C}_{22}\text{H}_{25}\text{O}_2\text{SSi}]^+$: 381.1339, found: 381.1320 $[\text{M}+\text{H}]^+$; calcd for $[\text{C}_{22}\text{H}_{24}\text{NaO}_2\text{SSi}]^+$: 403.1158, found: 403.1144 $[\text{M}+\text{Na}]^+$.

4-((*tert*-butyldiphenylsilyloxy)methyl)-5-((2,6-diphenyl-4*H*-pyran-4-ylidene)methyl)thiophene-2-carbaldehyde (**21**)

A solution of **5** (22.6 mg, 0.51 mmol) in anhydrous THF (5 mL) was prepared, purged with argon and cooled to -78°C . To this solution, *n*-BuLi (1.6 M in hexanes) (0.32 mL, 0.51 mmol) was added dropwise and the resulting mixture was stirred for 20 min. Then 3-((*tert*-butyldiphenylsilyloxy)methyl)thiophene-2-carbaldehyde **18** (151.1 mg; 0.40 mmol) in anhydrous THF (5 mL) was added dropwise and the mixture was warmed to room temperature for 21 hours (TLC monitoring using 10% dichloromethane in hexanes). Saturated NH_4Cl solution was added to quench the reaction and the solvent was evaporated under reduced pressure. The organic layer was extracted with hexanes (25×2 mL) and dried over anhydrous MgSO_4 . After the removal of the solvent, the residue was dissolved in hexanes/ CH_2Cl_2 (9/1) and filtered over silica gel to give the compound **20** as an intermediate. A solution of **20** in THF (10 mL) was prepared, purged with argon and cooled to -78°C . To this solution, *n*-BuLi (1.6 M in hexanes) (0.36 mL, 0.58 mmol) was added dropwise and the resulting mixture was stirred for 1 h and 30 min. DMF (0.08 mL, 1.03 mmol) was added dropwise and the mixture stirred for 2 h. Saturated NH_4Cl solution was added to quench the reaction and the solvent was evaporated under reduced pressure. The organic layer was extracted with hexanes (15×2 mL) and dried over anhydrous MgSO_4 . After the removal of the solvent, the aldehyde **21** was purified by silica gel column chromatography (20% ethyl acetate in hexanes). Yield: red oil (136.4 mg, 0.22 mmol; 55%).

IR (neat): cm^{-1} 1643 (C=O), 1518 (C=C). ^1H NMR (400 MHz, CD_2Cl_2): δ (ppm) 9.77 (s, 1H), 7.92–7.89 (m, 2H), 7.80–7.78 (m, 2H), 7.75–7.67 (m, 4H), 7.64 (s, 1H), 7.55–7.38 (m, 12H), 7.27 (d, $J = 2.0$ Hz, 1H), 6.47 (d, $J = 2.0$ Hz, 1H), 6.00 (s, 1H), 4.81 (s, 2H), 1.09 (s, 9H). ^{13}C NMR (100 MHz, CD_2Cl_2): δ (ppm) 182.4, 155.8, 153.0, 147.1, 139.1, 138.5, 137.8, 136.2, 133.8, 133.2, 133.1, 133.0, 130.7, 130.4, 130.2, 129.4, 129.3, 128.4, 125.9, 125.2, 109.5, 105.4, 103.3, 60.9, 27.2, 19.7. HRMS (ESI⁺): m/z calcd for $[\text{C}_{40}\text{H}_{37}\text{O}_3\text{SSi}]^+$: 625.2227, found: 625.2235 $[\text{M}+\text{H}]^+$.

3-((*tert*-butyldiphenylsilyloxy)methyl)-5-((2,6-diphenyl-4*H*-pyran-4-ylidene)methyl)thiophene-2-carbaldehyde (**23**)

A solution of 2,6-diphenyl(4*H*-pyran-4-ylidene)diphenylphosphine oxide **5** (222.6 mg, 0.51 mmol) in anhydrous THF (5 mL) was prepared, purged with argon and cooled to -78°C . To the solution, *n*-BuLi (1.6 M in hexanes) (0.32 mL, 0.51 mmol) was added dropwise and the resulting mixture was stirred for 20 min. Then 4-((*tert*-butyldiphenylsilyloxy)methyl)thiophene-2-carbaldehyde **19** (150.0 mg; 0.39 mmol) in anhydrous THF (5 mL) was added dropwise and the mixture was warmed to room temperature for 21 h (TLC monitoring using 20% ethyl acetate in hexanes). Saturated NH_4Cl solution was added to quench the reaction and the solvent was evaporated under reduced pressure. The organic layer was extracted with EtOAc and dried over anhydrous MgSO_4 . After the removal of the solvent, the residue was dissolved in EtOAc/hexanes (2:8) and filtered over silica gel to give the crude *tert*-butyl((5-((2,6-diphenyl-4*H*-pyran-4-ylidene)methyl)thiophen-3-yl)methoxy)diphenylsilane **22** as an intermediate. A solution of **22** in THF (10 mL) was prepared, purged with argon and cooled to -78°C . To this solution, *n*-BuLi (1.6 M in hexanes) (0.42 mL, 0.68 mmol) was added dropwise and the resulting mixture was stirred for 2 h. DMF (0.09 mL, 1.2 mmol) was added dropwise and the mixture stirred for 2 h. Saturated NH_4Cl solution was added to quench the reaction and the solvent was evaporated under reduced pressure. The organic layer was extracted with EtOAc (15×3 mL) and dried over anhydrous MgSO_4 . After the removal of the solvent, the aldehyde **23** was purified by silica gel column chromatography (20% ethyl acetate in hexanes). Yield: dark red oil (138.9 mg, 0.22 mmol; 56%).

IR (neat): cm^{-1} 1643 (C=O), 1579 (C=C). ^1H NMR (400 MHz, CD_2Cl_2): δ (ppm) 9.86 (s, 1H), 7.90–7.88 (m, 2H), 7.82–7.80 (m, 2H), 7.73–7.70 (m, 4H), 7.50–7.41 (m, 12H), 7.22 (d, $J = 1.9$ Hz, 1H), 7.05 (s, 1H), 6.55 (d, $J = 1.9$ Hz, 1H), 6.14 (s, 1H), 5.02 (s, 2H), 1.11 (s, 9H). ^{13}C NMR (100 MHz, CD_2Cl_2): δ (ppm) 181.4, 155.6, 153.1, 151.5, 151.3, 136.1, 133.7, 133.5, 133.3, 133.2, 133.1, 130.7, 130.5, 130.2, 129.4, 129.3, 128.4, 127.8, 125.9, 125.3, 109.1, 107.6, 103.2, 61.2, 27.2, 19.7. HRMS (ESI $^+$): m/z calcd for $[\text{C}_{40}\text{H}_{37}\text{O}_3\text{SSi}]^+$: 625.2227, found: 625.2244 $[\text{M}+\text{H}]^+$.

3-(4-((*tert*-butyldiphenylsilyloxy)methyl)-5-((2,6-diphenyl-4*H*-pyran-4-ylidene)methyl)thiophen-2-yl)-2-cyanoacrylic acid (**15**)

To a solution of aldehyde **21** (97.9 mg, 0.157 mmol) and 2-cyanoacetic acid (20.8 mg, 0.24 mmol) in chloroform (16 mL) was added piperidine (109.9 μL , 1.04 mmol). The mixture was refluxed for 72 h under an argon atmosphere and then cooled down to room temperature. The reaction crude was chromatographed on reverse C18 silica gel (started 50 % AcONH_4 (20 mM) in acetonitrile and finished 30%). After the addition of acetic acid a solid appeared. This compound was filtered under reduced pressure to afford **15**. Yield: dark purple solid (83.4 mg, 0.12 mmol; 77%).

Mp $208\text{--}211^{\circ}\text{C}$. IR (KBr): cm^{-1} 2207 (C $\equiv$ N), 1652 (C=O), 1539 (C=C). ^1H NMR (300 MHz, $\text{THF-}d_8$): δ (ppm) 8.29 (s, 1H), 8.01–7.98 (m, 2H), 7.88–7.85 (m, 2H), 7.79 (s, 1H), 7.76–7.70 (m, 4H), 7.59–7.32 (m, 13H), 6.76 (d, $J = 1.6$ Hz, 1H), 6.12 (s, 1H), 4.87 (s, 2H), 1.09 (s, 9H). ^{13}C NMR (100 MHz, $\text{THF-}d_8$): δ (ppm) 164.9, 156.6, 153.8, 147.8, 146.1, 139.8, 139.8, 136.6, 134.3, 134.0, 133.7, 133.5, 131.9, 130.9, 130.6, 130.0, 129.7, 128.8, 126.2, 125.7, 117.9, 110.2, 106.2, 104.0, 96.6, 61.3, 27.4, 20.1. HRMS (ESI $^+$): m/z calcd for $[\text{C}_{43}\text{H}_{38}\text{NO}_4\text{SSi}]^+$: 692.2285, found: 692.2248 $[\text{M}+\text{H}]^+$; calcd for $[\text{C}_{43}\text{H}_{37}\text{NNaO}_4\text{SSi}]^+$: 714.2105, found: 714.2066 $[\text{M}+\text{Na}]^+$.

3-(3-((*tert*-butyldiphenylsilyloxy)methyl)-5-((2,6-diphenyl-4*H*-pyran-4-ylidene)methyl)thiophen-2-yl)-2-cyanoacrylic acid (**16**)

To a solution of aldehyde **23** (120.4 mg, 0.192 mmol) and 2-cyanoacetic acid (25.6 mg, 0.30 mmol) in chloroform (18 mL) was added piperidine (134.6 μL , 1.28 mmol). The mixture was refluxed for 20 h under an argon atmosphere, then cooled down to room temperature. The reaction crude was chromatographed on reverse C18 silica gel (started 50 % AcONH_4 (20 mM) in acetonitrile and finished 33%). After the addition of acetic acid a solid appeared. This compound was filtered under reduced pressure to afford **16**. Yield: dark purple solid (86.0 mg, 0.12 mmol; 65%).

Mp 219–226 °C. IR (KBr): cm^{-1} 2209 (C $\equiv$ N), 1651 (C=O), 1541 (C=C). ^1H NMR (400 MHz, THF- d_8): δ (ppm) 8.51 (s, 1H), 7.99–7.88 (m, 4H), 7.74–7.72 (m, 4H), 7.49–7.34 (m, 14H), 6.99 (s, 1H), 6.81 (br s, 1H), 4.92 (s, 2H), 1.08 (s, 9H). ^{13}C NMR (100 MHz, THF- d_8): δ (ppm) 164.9, 156.4, 154.0, 152.5, 151.7, 143.7, 136.6, 134.3, 134.0, 133.7, 133.5, 131.9, 131.2, 130.9, 130.7, 130.0, 129.8, 128.8, 128.1, 126.2, 125.8, 118.1, 109.8, 108.4, 103.9, 95.8, 61.3, 27.4, 20.0. HRMS (ESI $^+$): m/z calcd for $[\text{C}_{43}\text{H}_{38}\text{NO}_4\text{SSi}]^+$: 692.2285, found: 692.2313

2.3. Device preparation and characterization

The working and counter electrodes consisted of TiO_2 and thermalized platinum films, respectively, and were deposited onto F-doped tin oxide (FTO, Pilkington Glass Inc., with $15 \Omega \text{ sq}^{-1}$ sheet resistance) conducting glass substrates. Efficient DSC devices were made using $9 \mu\text{m}$ thick films consisting of 20 nm TiO_2 nanoparticles (Dyesol $^\circ$ paste) and a scattering layer of $4 \mu\text{m}$ of 400 nm TiO_2 particles (Dyesol $^\circ$ paste). Prior to the deposition of the TiO_2 paste, the conducting glass substrates were immersed in a solution of TiCl_4 (40 mM) for 30 minutes and then dried. The TiO_2 nanoparticle paste was deposited onto a conducting glass substrate using the screen printing technique. The TiO_2 electrodes were gradually heated under an air flow at $325 \text{ }^\circ\text{C}$ for 5 min, $375 \text{ }^\circ\text{C}$ for 5 min, $450 \text{ }^\circ\text{C}$ for 15 min and $500 \text{ }^\circ\text{C}$ for 15 min. The heated TiO_2 electrodes were immersed again in a solution of TiCl_4 (40 mM) at $70 \text{ }^\circ\text{C}$ for 30 min and then washed with ethanol. The electrodes were heated again at $500 \text{ }^\circ\text{C}$ for 30 min and cooled before dye adsorption. The active area for devices was 0.16 cm^2 . The counter electrode was made by spreading a 5 mM solution of H_2PtCl_6 in isopropyl alcohol onto a conducting glass substrate containing a small hole to allow the introduction of the liquid electrolyte using vacuum, followed by heating at $390 \text{ }^\circ\text{C}$ for 15 minutes. All films were sensitized in 0.3 mM dye solutions in dichloromethane for 2 hours at room temperature (optimized dye loading conditions). The sensitized electrodes were washed with dichloromethane and dried under air. Finally, the working and counter electrodes were sandwiched together using a thin thermoplastic (Surlyn) frame that melts at $100 \text{ }^\circ\text{C}$. Electrolyte LP1 was used: consisted of 0.5 M 1-butyl-3-methylimidazolium iodide (BMII), 0.1 M lithium iodide, 0.05 M iodine and 0.5 M *tert*-butylpyridine in acetonitrile.

The J/V curves of the cells were measured using a Sun 2000 solar simulator equipped with a 150W xenon lamp. The illumination intensity was measured to be 100 mW cm^{-2} with a calibrated silicon photodiode. The appropriate filters were utilized to faithfully simulate the AM 1.5G spectrum. The applied potential and cell current were measured using a Keithley 2400 digital source meter. The IPCE (incident photon to current conversion efficiency) was measured using a home-made set up consisting of a 150 W Oriel xenon lamp, a motorized monochromator and a Keithley 2400 digital source meter.

Transient photovoltage (TPV) and charge extraction (CE) measurements were carried out alike reported before.³⁶ In CE measurements, white light from a series of LEDs was used as the light source. When the LEDs are turned off, the cell is immediately short-circuited and the charge is extracted allowing the electron density in the cells to be calculated. By changing the intensity of the LEDs, the electron density can be estimated as a function of cell voltage. In TPV measurements, in addition to the white light applied by the LEDs, a diode pulse (660 nm, 10 mW) is applied to the sample inducing a change of 2–3 mV within the cell. The resulting photovoltage decay transients are collected and the τ values are determined by fitting the data to the equation $\exp(-t/\tau)$.

2.4. Computational details

Density Functional Theory (DFT) calculations were performed using Gaussian 09³⁷ with the ultrafine integration grid. Solvent effects were estimated using a Conductor-like Polarizable Continuum Model (CPCM).^{38, 39} Equilibrium geometries were optimized using the M06-2x hybrid meta-GGA exchange

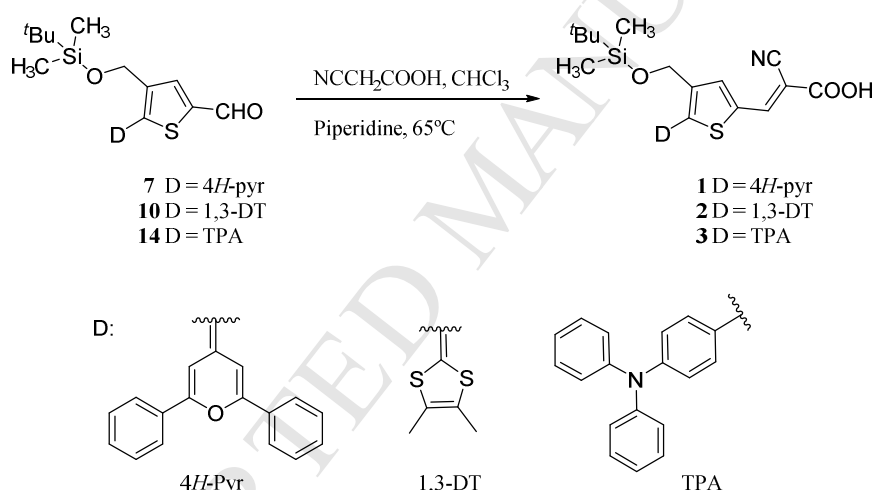
correlation functional⁴⁰ and the medium size 6-31G* base.⁴¹ Ground state geometries were characterized as minima by frequency calculations. Excitation energies were calculated by time-dependent single point calculations using the M06-2x/6-311+G (2d,p) model chemistry. Absorption spectra were estimated through the calculation of vertical excitations at the ground state geometry and emission spectra, through the calculation of vertical excitations at the excited state geometry. Due to the large computational cost required to calculate excited state vibrational energies, E_{0-0} were approached to adiabatic excitation energies (E_{Adia}). Ground state oxidation potentials (E_{OX}) were calculated using the M06-2x/6-311+G (2d,p) energies and calculating the thermal corrections to Gibbs free energy at the M06-2x/6-31G* level. Excited state oxidation potentials (E^*_{OX}) were estimated subtracting E_{0-0} from E_{OX} .

Molecular Orbital contour plots were obtained using the Avogadro software⁴² at 0.04 isosurface value.

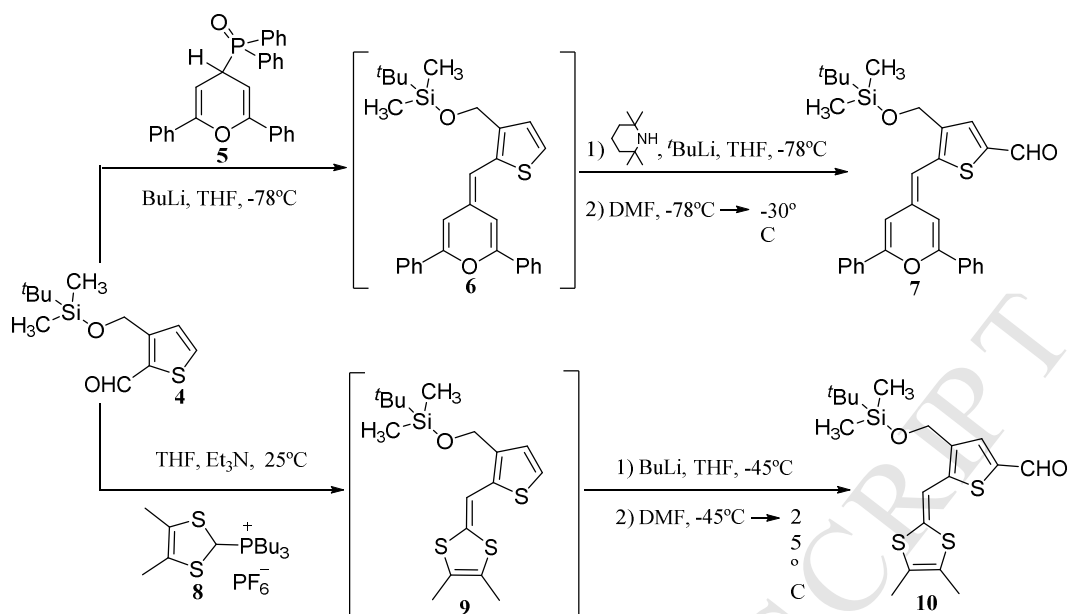
3. Results and discussion

3.1. Synthesis

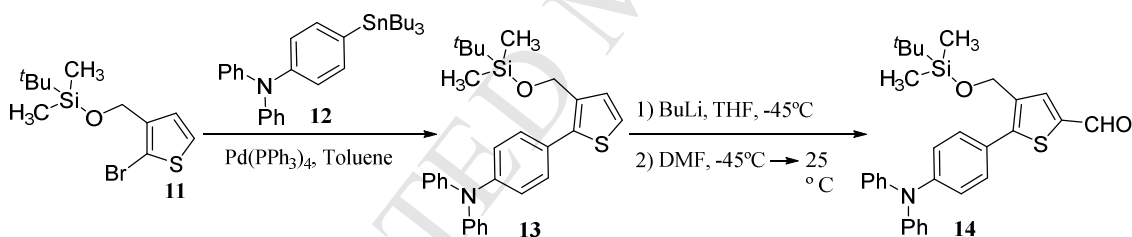
The target compounds **1**, **2**, and **3** were prepared via condensation of the corresponding aldehyde derivative (**7**, **10**, **14**) and cyanoacetic acid in the presence of piperidine (Scheme 1).



Scheme 1. Synthesis of dyes **1–3**

Scheme 2. Synthesis of aldehydes **7**, **10**

Aldehydes **7** and **10** were synthesized by a Wittig-Horner or Wittig reaction of the diphenylphosphine oxide **5**⁴³ or the phosphonium salt **8**⁴⁴ with aldehyde **4**,⁴⁵ followed by lithiation of the resulting intermediates **6** and **9** and reaction with anhydrous DMF (Scheme 2). A different approach was adopted for the aldehyde **14** (Scheme 3) which was prepared through a Stille reaction between the stannane **12**⁴⁶ and the compound **11**,⁴⁷ followed by lithiation and formylation with DMF.

Scheme 3. Synthesis of aldehyde **14**

3.2. Optical properties

The absorption spectra of the synthesized dyes were studied in 10^{-5} M CH_2Cl_2 solution (Fig. 3) and the relevant optical data are listed in Table 1. Two absorption bands can be observed for all dyes, one between 300–400 nm assigned to a π – π^* transition and the main band between 400–700 nm due to the ICT among the donor and the acceptor units. The inclusion of a proaromatic donor like 4*H*-pyranylidene or dithiafulvene results in a red-shift of the absorption when compared to the triarylamine dye. The molar extinction coefficients of the dyes increase in the order triarylamine < 4*H*-pyranylidene < dithiafulvene, being for the last one $37108 \text{ M}^{-1} \text{ cm}^{-1}$. When the dyes are attached to TiO_2 surface, they present broader bands than in solution and the maximum absorption peaks are around 30 nm blue-shifted. In general, the blue shifts of the absorption spectra on TiO_2 are ascribed to the deprotonation of the carboxylic acid when anchored on the titanium surface and/or to the formation of H-aggregates.⁴⁸

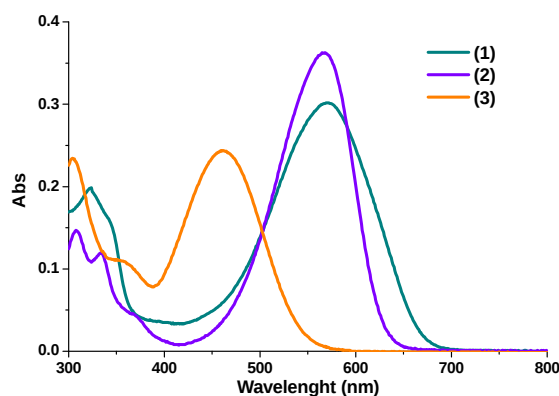


Fig. 3. Normalized UV-vis absorption of compounds 1–3

Table 1. Optical and electrochemical properties of dyes 1–3.

dye	λ_{abs} , nm (ϵ , $\text{M}^{-1}\text{cm}^{-1}$) ^a	λ_{abs} (nm) ^b	λ_{em} (nm) ^c	E_{ox}^d V (vs. NHE)	E_{0-0}^e (eV)	E_{ox}^{*f} V (vs. NHE)
1	570 (30370)	522	659	0.87	1.97	−1.10
2	568 (37108)	518	630	0.93	2.06	−1.12
3	460 (23958)	432	611	1.22	2.29	−1.08

^aAbsorption maxima in CH_2Cl_2 solution. ^bAbsorption maxima on TiO_2 films. ^cEmission spectra were recorded in CH_2Cl_2 . ^dFirst oxidation potentials were measured from three electrode electrochemical cell in CH_2Cl_2 containing 0.1 M TBAPF₆. A glassy carbon, Ag/AgCl (KCl 3M), and Pt were used as working, reference, and counter electrode respectively. ^eZeroth-zeroth transition energies estimated from the intersection of normalized absorption and emission spectra in CH_2Cl_2 solution. ^fExcited-state oxidation potentials of the dyes obtained from $E_{\text{ox}} - E_{0-0}$.

3.3. Electrochemical Properties

The oxidation potential (E_{ox}) of each dye was determined in solution by differential pulse voltammetry (DPV) methods. E_{ox}^* was estimated from $E_{\text{ox}} - E_{0-0}$ (Table 1, see also energy diagram shown in SI Figure S-33).

The donor unit affects significantly the oxidation potential values, with an increase on the ease of oxidation in the order $3 < 2 < 1$, pointing to the superior donor ability of the 4*H*-pyranylidene moiety. The regeneration of the oxidized dye after electron injection into the conduction band of TiO_2 is guaranteed in all cases, as the E_{ox} values are more positive than the potential of the iodide/triiodide redox couple (+0.42 vs NHE), indicating that the oxidized dyes formed after the electron injection into the TiO_2 electrode could thermodynamically accept electrons from I^- ions.

The resulting E_{ox}^* values for all the dyes were almost similar and sufficiently more negative than the energy conduction band edge energy level of TiO_2 (−0.5 V vs NHE), indicating an efficient injection of electrons into the TiO_2 from the excited dyes.

3.4. Photovoltaic properties of DSSCs

The device performance (with an effective area of 0.16 cm^2) were measured under sun-simulated AM 1.5 G irradiation (100 mW/cm^2). The optimized conditions for these dyes are determined to be 0.3 mM of dye in dichloromethane solution, 2 h of immersion and LP1 (0.5 M BMII/ 0.05M I_2 / 0.5M TBP/ 0.1M LiI in acetonitrile) as electrolyte. The relevant photovoltaic parameters V_{oc} , J_{sc} , ff , and solar-to-electrical energy

conversion efficiencies (η) are collected in Table 2. Moreover, current density-voltage (J - V) curves and incident photon- to- current conversion efficiencies (IPCE) of devices based on these dyes are represented in Fig. 4.

Table 2. Photovoltaic properties of DSSCs constructed using the dyes **1–3**.

dye	J_{sc} (mA cm ⁻²)	V_{oc} (V)	ff	η (%)
1	11.74	0.644	74	5.56
2	6.99	0.609	69	2.91
3	7.26	0.759	74	4.09

The IPCE tendencies of these dyes are in accordance with their UV-vis absorption spectra on the TiO₂ film. The onset of the IPCE spectra for dyes **1** and **2** are significantly broadened compared with the obtained for dye **3**. These results confirm the higher J_{sc} value obtained for devices based on dye **1**, with a broad and roughly constant photoresponse ($\sim 60\%$) in the range 450–600 nm. Dye **3** presents a maximum IPCE value of 93% at 475 nm and the highest V_{oc} value (0.759V) of all the molecules studied, suggesting a decrease in the charge recombination processes for this molecule. However, its J_{sc} is 60% lower than obtained for compound **1**, probably due to the narrower IPCE observed band.

Data in Table 2 indicate that the more efficient donor system is the 4*H*-pyranylidene ($\eta = 5.56\%$), followed by the triarylamine system ($\eta = 4.09\%$) and finally the 1,3-dithiole ($\eta = 2.91\%$).

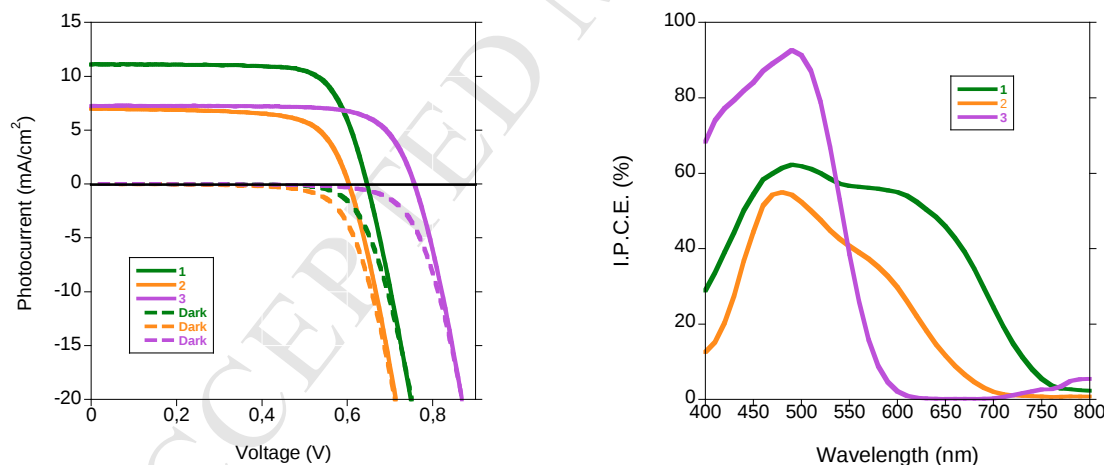
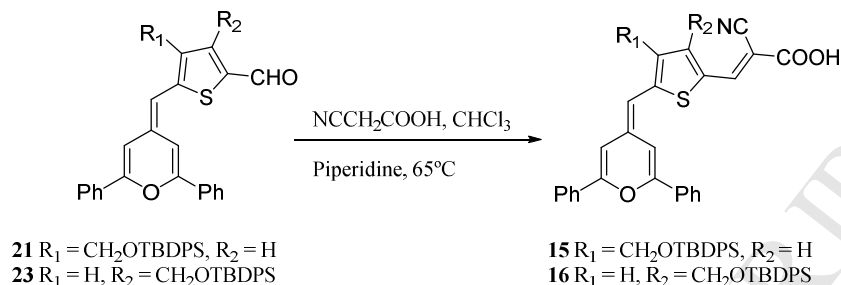


Fig. 4. J - V curves of compounds **1–3** (left) and IPCE curves of **1–3** (right)

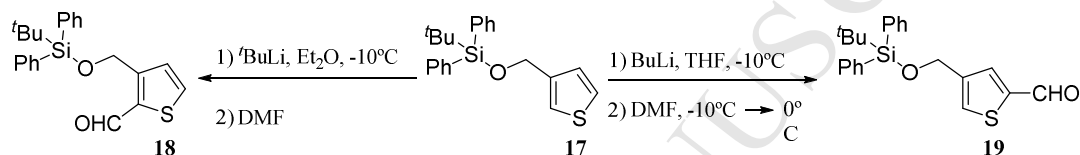
3.5. Structural modification of the thiophene ring and physical properties

With the aim of understanding the influence of a different bulky silyl ether group and its position in the thiophene ring, dyes **15** and **16** were synthesized by a Knoevenagel condensation of the unreported aldehydes **21** and **23** with 2-cyanoacetic acid (Scheme 4). We have chosen the pyranylidene system as donor group due to the best results obtained in the previous section. These precursors were obtained in a three-step synthetic route starting in both cases from compound **17**.⁴⁹ Lithiation conditions (base and solvent) were finely tuned in order to activate

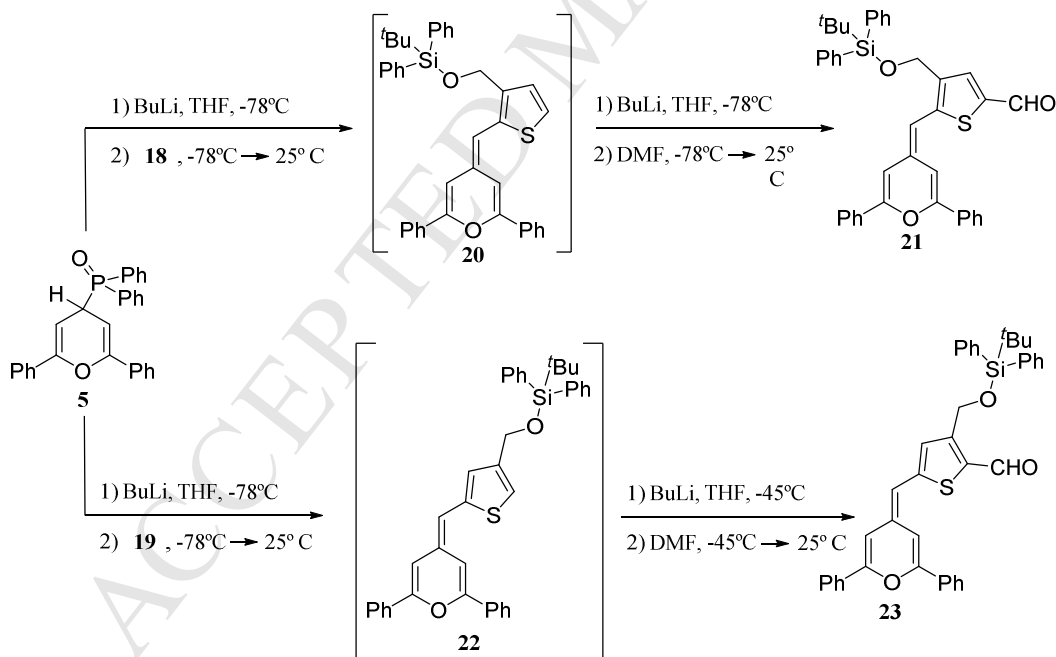
position 2 (*tert*-BuLi, Et₂O) or position 5 (BuLi, THF) (Scheme 5). Finally, compounds **18** and **19** were obtained by reaction of the organolithium intermediates with DMF. Next, a Horner reaction of diphenylphosphine oxide **5** with **18** or **19** afforded the corresponding pyranylidene-thiophene derivatives **20** and **22**, which by lithiation and reaction with DMF yielded final aldehydes **21** and **23** respectively (Scheme 6).



Scheme 4. Synthesis of dyes **15**, **16**



Scheme 5. Synthesis of thiophene-aldehydes **18**, **19**



Scheme 6. Synthesis of aldehydes **21**, **23**

The substitution of a TBDMS group (**1**) by a *tert*-butyldiphenylsilyl (TBDPS (**15**)) does not change significantly the maximum absorption peak (570 nm for **1** and 573 nm for **15**). By contrast, the molar extinction coefficient (ϵ) is enhanced on passing from TBDMS to TBDPS (30370 M⁻¹cm⁻¹ (**1**) to 33618 M⁻¹cm⁻¹ (**15**)), probably due to the incorporation of two additional phenyl groups. Moreover, a bigger enhancement of the ϵ is found when the silyloxy substituent

is located at the position 3 of the thiophene ring instead of the position 4 (from 33618 (**15**) to 39963 $\text{M}^{-1} \text{cm}^{-1}$ (**16**)).

Regarding the electrochemical properties, it has to be pointed that the E_{ox} values for systems **1**, **15** and **16** are very similar (+0.87, +0.90, and +0.88V respectively, measured in conditions indicated in Table 1) and hence, the influence of the bulky substituent and its position is fairly null (See SI Figure S-34).

We have tested the photovoltaic response of dyes **15** and **16** (Figure 5) in DSSCs and their results were compared with those of compound **1** (Table 3). The substitution of a TBDMS group (**1**) by the bulkier group (TBDPS) (**15**) improve slightly the efficiency (from 5.56% to 5.86%), mostly due to a better value of the V_{oc} for the last one. This result suggests that a TBDPS group retards the recombination of TiO_2 conduction band electrons with the electrolyte as it has been shown from charge extraction and transient photovoltage experiments (see below).⁵⁰

The influence of the position of the silyl ether group was also studied and we have found that the solar cell based on dye **16** (with a higher molecular extinction coefficient) shows an efficiency of 3.42%, 42% lower that obtained for dye **15**. The decrease in the efficiency is mainly due to a lower J_{sc} and V_{oc} , to a lesser extent. The remarkable decrease of J_{sc} was attributed to the lower injection efficiency, probably affected by the amount of dye absorbed to the TiO_2 surface.

Desorption experiments (0.1M NaOH, EtOH:H₂O (1:1)) determined that dye **16** is less adsorbed (106 nmol/cm^2) than dye **15** (140 nmol/cm^2) and this observation was attributed to the steric hindrance of the TBDPS group which is closer to the anchor group. Moreover, a lower amount of dye absorbed could result in higher I_3^- concentration in the interface TiO_2 /electrolyte, increasing the recombination processes.

Table 3. Photovoltaic properties of DSSCs constructed using the dyes **1**, **15** and **16**

dye	J_{sc} (mA cm^{-2})	V_{oc} (V)	ff	η (%)
1	11.74	0.644	74	5.56
15	11.44	0.689	74	5.86
16	7.71	0.624	71	3.42

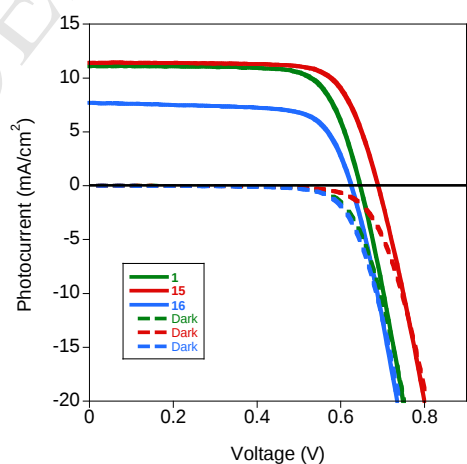


Fig. 5- J - V curves of compounds **1**, **15** and **16**

Figure 6 illustrates the charge density (charge accumulated at the device under different light bias) and the charge lifetime at a given charge. As it can be seen, the solar cells made using dye **15** have the slowest charge recombination under working conditions. Moreover, it is worthy to mention that a shift on the measured charge vs voltage is observed for the solar cells made using dye **16**. This shift can be attributed to the lower coverage of dyes at the surface of the TiO_2 , which has as a consequence the lower concentration of protons at the surface of the TiO_2 . In fact, this is also in good agreement with the lower photocurrent measured.

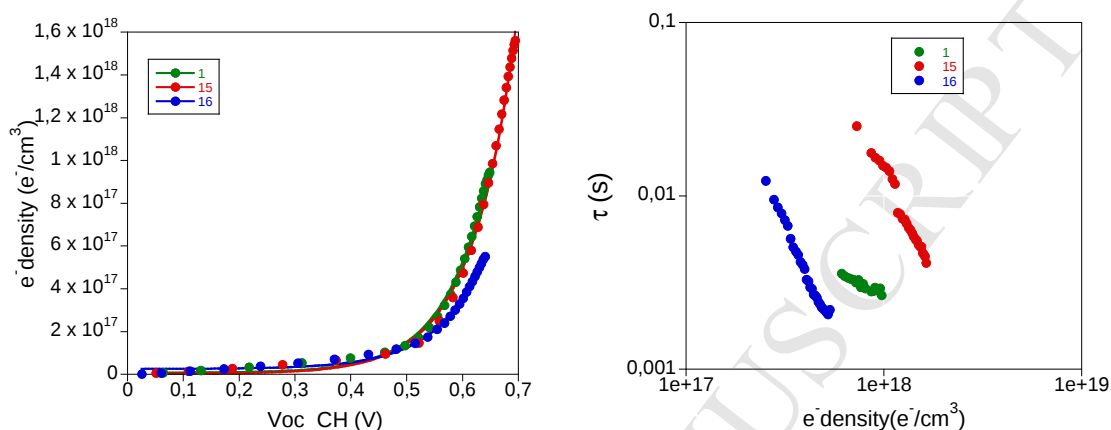


Fig. 6- Charge Extraction (left) and Transient Photovoltage (right) curves of compounds **1**, **15** and **16**

3.6. Theoretical calculations

The electronic structures of the new synthesized compounds have been studied using density functional theory methods and the most relevant parameters derived from these calculations are gathered in table 4. There is a good agreement with the experimental values described above, with estimated error of less than 0.25 eV in the excitation energies, less than 0.1 V in E_{OX} and less than 0.2 V in E_{OX}^* .

Table 4. Results of DFT Calculations.

	E_{HOMO} (eV)	E_{LUMO} (eV)	λ_{Abs} (nm)	f^a	λ_{em} (nm)	E_{0-0} (eV)	E_{OX} V (vs. NHE)	E_{OX}^* V (vs. NHE)
1	-6.38	-2.29	558	1.24	666	2.03	0.95	-1.08
2	-6.38	-2.13	528	1.00	605	2.18	1.00	-1.18
3	-6.58	-2.11	431	1.23	565	2.50	1.24	-1.26
15	-6.39	-2.30	559	1.23	666	2.03	0.93	-1.10
16	-6.40	-2.33	564	1.25	663	2.03	0.95	-1.08

^a Oscillator strength

Calculations describe the first excited state as the consequence of a one-electron transition from the HOMO to the LUMO (see Figure 7). Excitation involves some charge transfer from the donor to the cyanoacetic acid acceptor, with a large HOMO-LUMO overlap that gives rise to large oscillator strengths and therefore to great molar extinction coefficients.

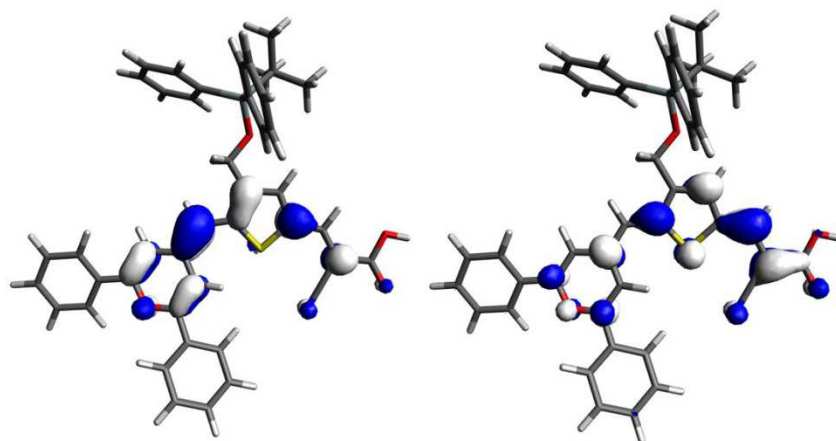


Figure 7. HOMO (left) and LUMO (right) for compound **15**.

Unfortunately, a direct correlation between the oscillator strengths (proportional to the peak area) and the molar extinction coefficients (related to peak height) is not possible for compounds giving rise to absorption bands of different width. Thus, compound **2** having the largest ϵ gives rise to the narrowest absorption band and also the lowest f .

4H-pyranylidene compounds **1**, **15** and **16** display similar HOMO and LUMO energies and therefore their excitation energies and oxidation potentials are also similar. Compound **2**, with a weak donor group has a lower HOMO energy compared to the other studied dyes and, therefore, displays higher excitation energies and E_{ox} .

Conclusions

A series of five new metal free sensitizers (D- π -A) for DSSC with a trialkylsilyl ether group in the spacer have been synthesized and characterized. Several donor unities have been studied, obtaining the best performances with *4H*-pyranylidene, followed by triphenylamine (TPA) and dithiafulvene moieties respectively. Two trialkylsilyl ether groups (TBDMS and TBDPS) were compared in order to study the influence of their molecular size in the photovoltaic properties. The best efficiencies values were obtained for a TBDPS group at the position 4 of the thiophene ring. This observation indicates that trialkylsilyl ethers can be promising substituents for future more efficient sensitizers.

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Appendix. Supplementary material

Supplementary data associated with this article can be found, in the online version at

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Synthesis and characterization of organic sensitizers with a trialkylsilyloxy group on the π -spacer

DSSC efficiency was highly dependent on the size and position of the silyloxy group

Best efficiencies were obtained with the 4H-pyranylidene, followed by TPA and dithiafulvene systems

Organic sensitizers bearing a trialkylsilylether group for liquid dye sensitized solar cells

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General Experimental Methods:

Infrared measurements were carried out in KBr or neat using a Perkin-Elmer Fourier Transform Infrared 1600 spectrometer. Melting points were obtained on a Gallenkamp apparatus in open capillaries and are uncorrected. ^1H - and ^{13}C -NMR spectra were recorded on a Bruker ARX300 or a Bruker AV400 at 300 or 400 MHz and 75 or 100 MHz respectively; δ values are given in ppm (relative to TMS) and J values in Hz. The apparent resonance multiplicity is described as s (singlet), d (doublet), and m (multiplet). ^1H - ^1H COSY and ^1H - ^{13}C -HSQC experiments were recorded on a Bruker ARX300 or a Bruker AV400 at 300 or 400 MHz in order to establish peaks assignment and spatial relationships. Electrospray mass spectra were recorded on a Bruker Q-ToF spectrometer; accurate mass measurements were achieved using sodium formate as external reference. UV-Visible spectra were recorded with an UV-Vis UNICAM UV4 spectrophotometer. Pulse differential voltammetry measurements were performed with a μ -Autolab type III potentiostat using a glassy carbon working electrode, Pt counter electrode, and Ag/AgCl reference electrode. The experiments were carried out under argon in CH_2Cl_2 , with Bu_4NPF_6 as supporting electrolyte (0.1 mol L^{-1}). Scan rate was 0.01 V s^{-1} , modulation amplitude 0.025 V and modulation time 0.05 s^{-1} .

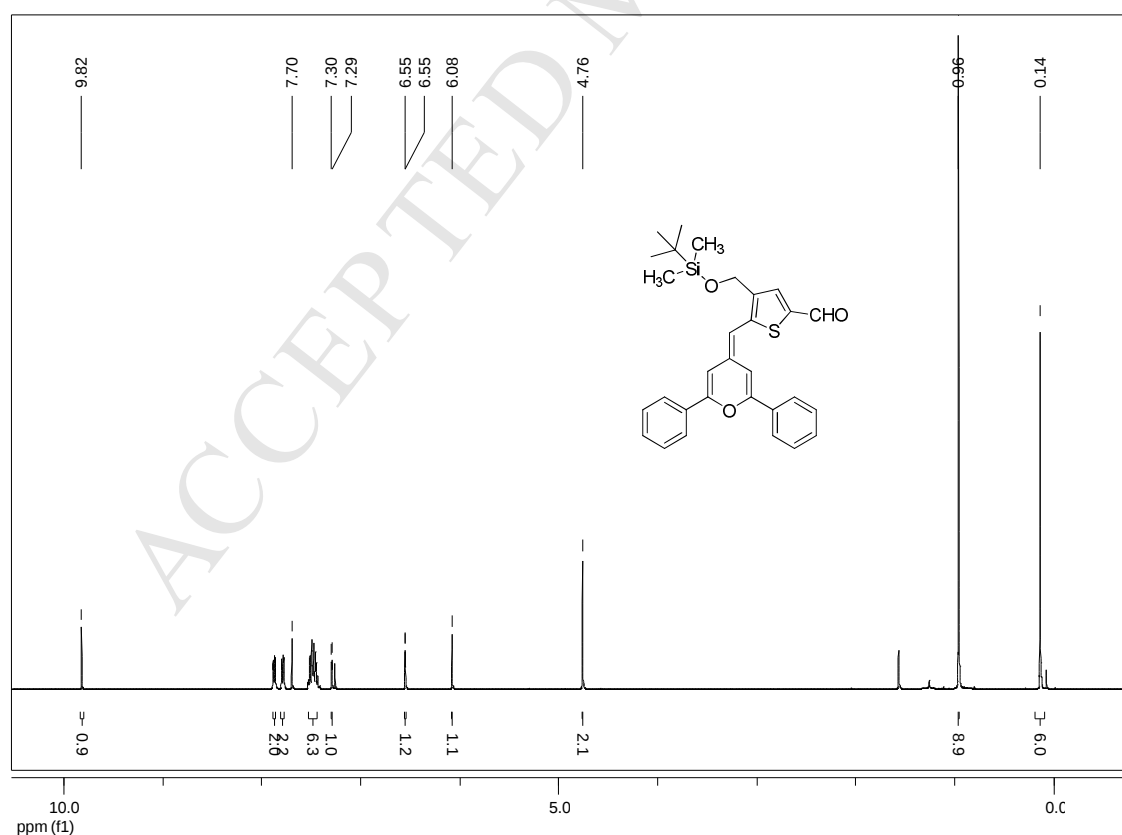


Figure S-1: ^1H NMR spectrum of compound 7 (400 MHz, CDCl_3).

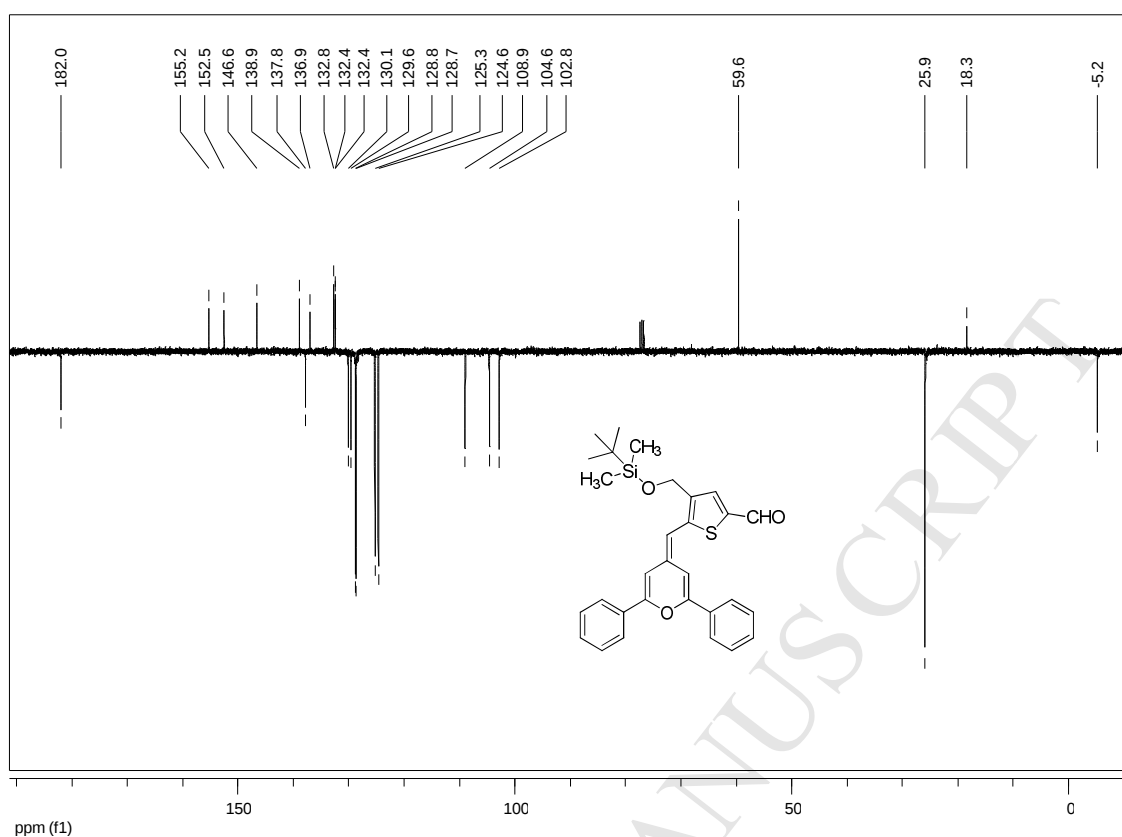


Figure S-2: ¹³C NMR (APT) spectrum of compound **7** (100 MHz, CDCl₃).

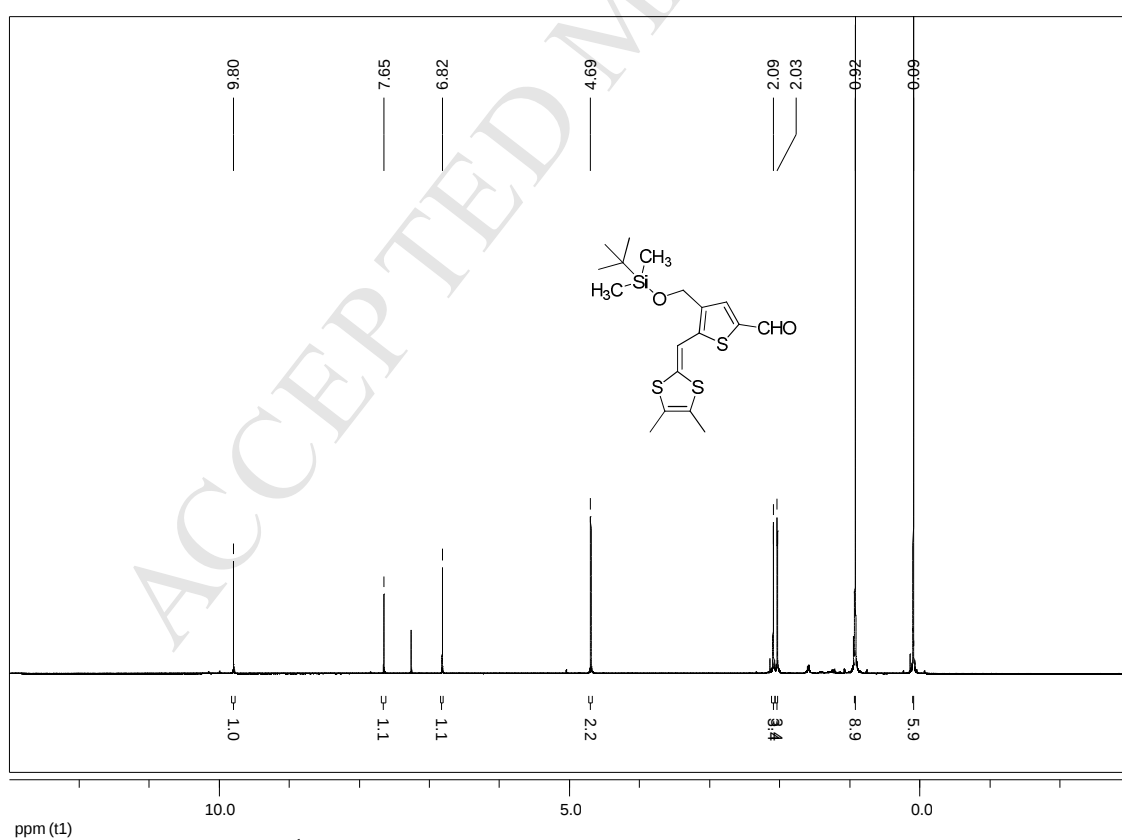


Figure S-3: ¹H NMR spectrum of compound **10** (400 MHz, CDCl₃).

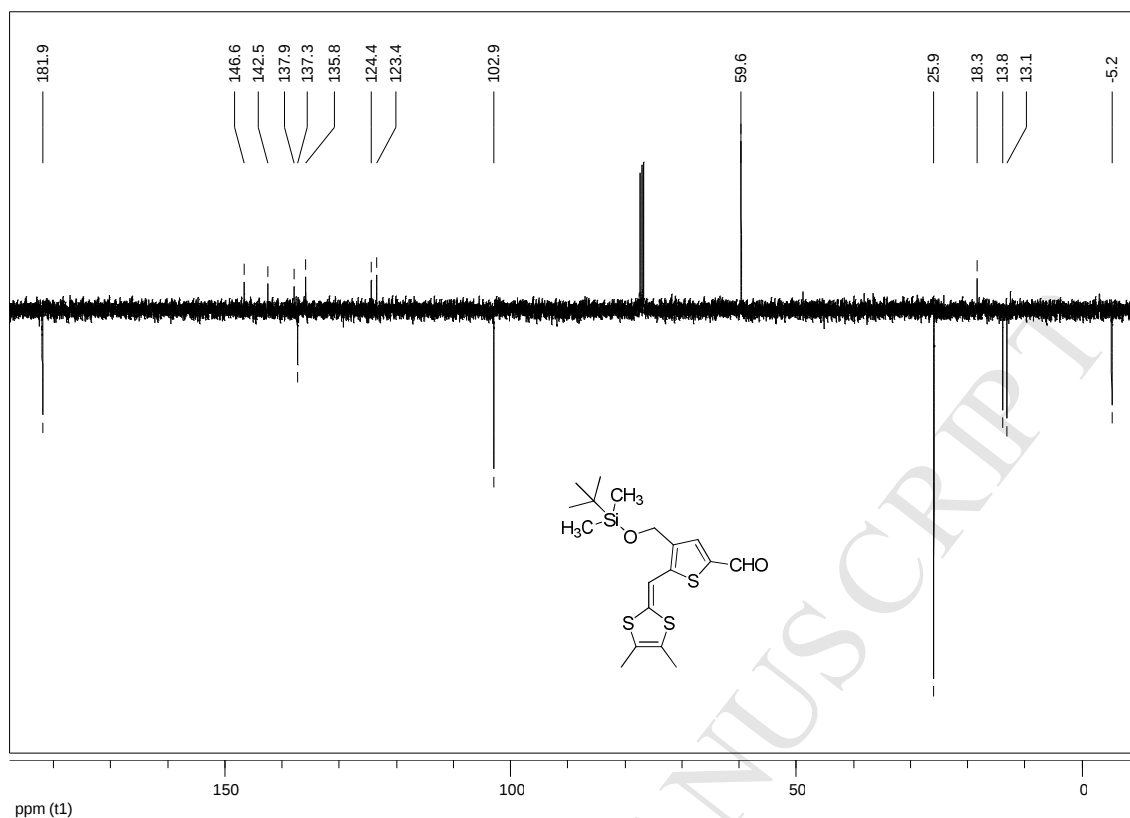


Figure S-4: ¹³C NMR (APT) spectrum of compound **10** (100 MHz, CDCl₃).

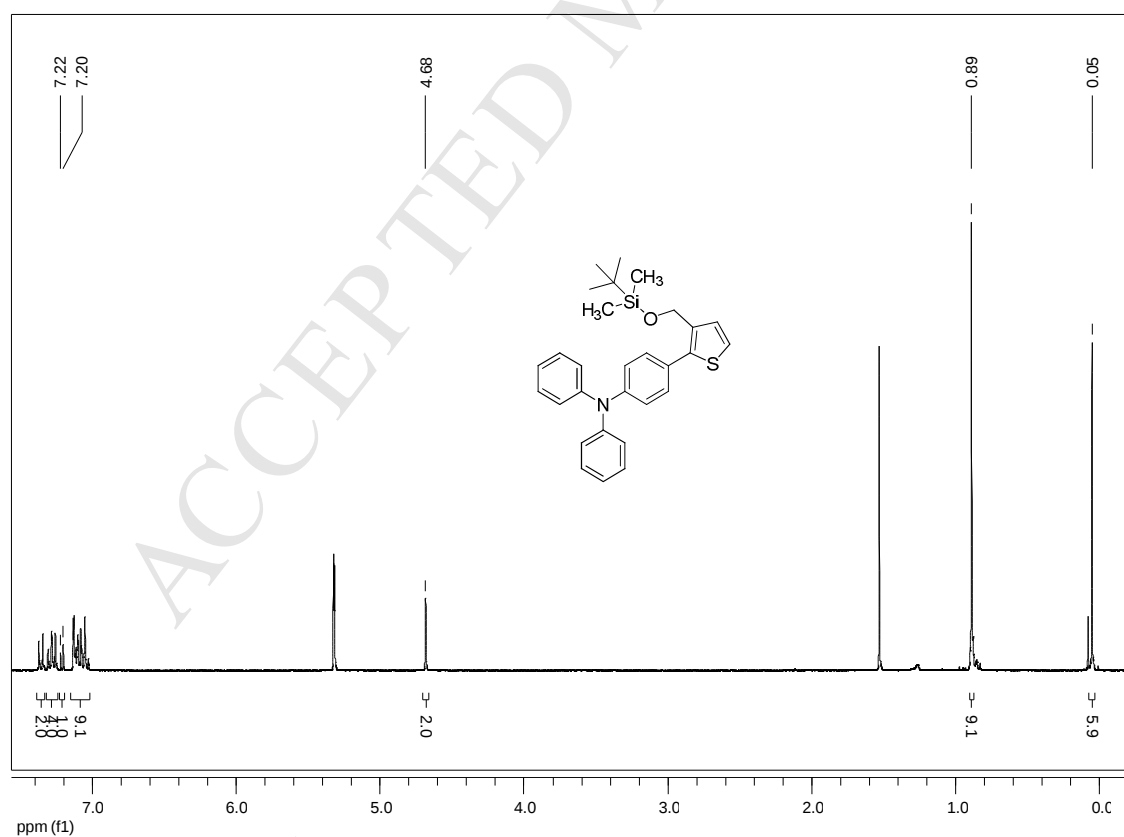


Figure S-5: ¹H NMR spectrum of compound **13** (300 MHz, CD₂Cl₂).

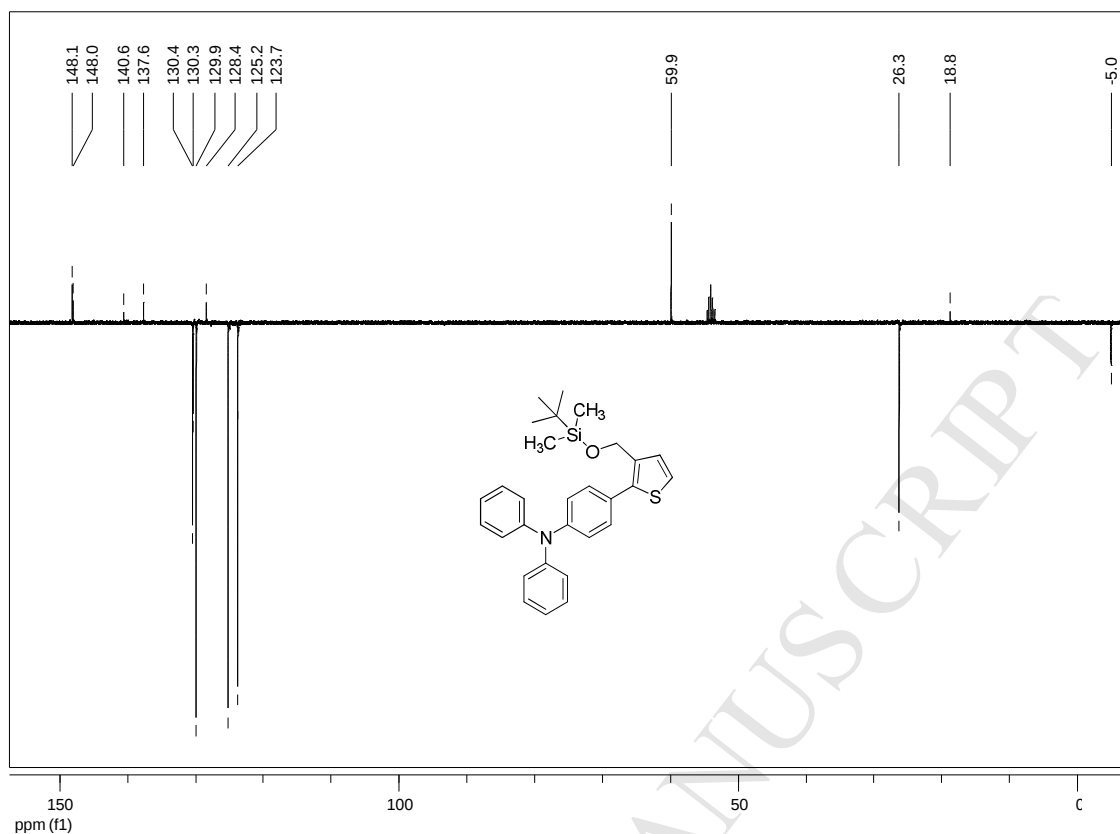


Figure S-6: ¹³C NMR (APT) spectrum of compound **13** (100 MHz, CD₂Cl₂).

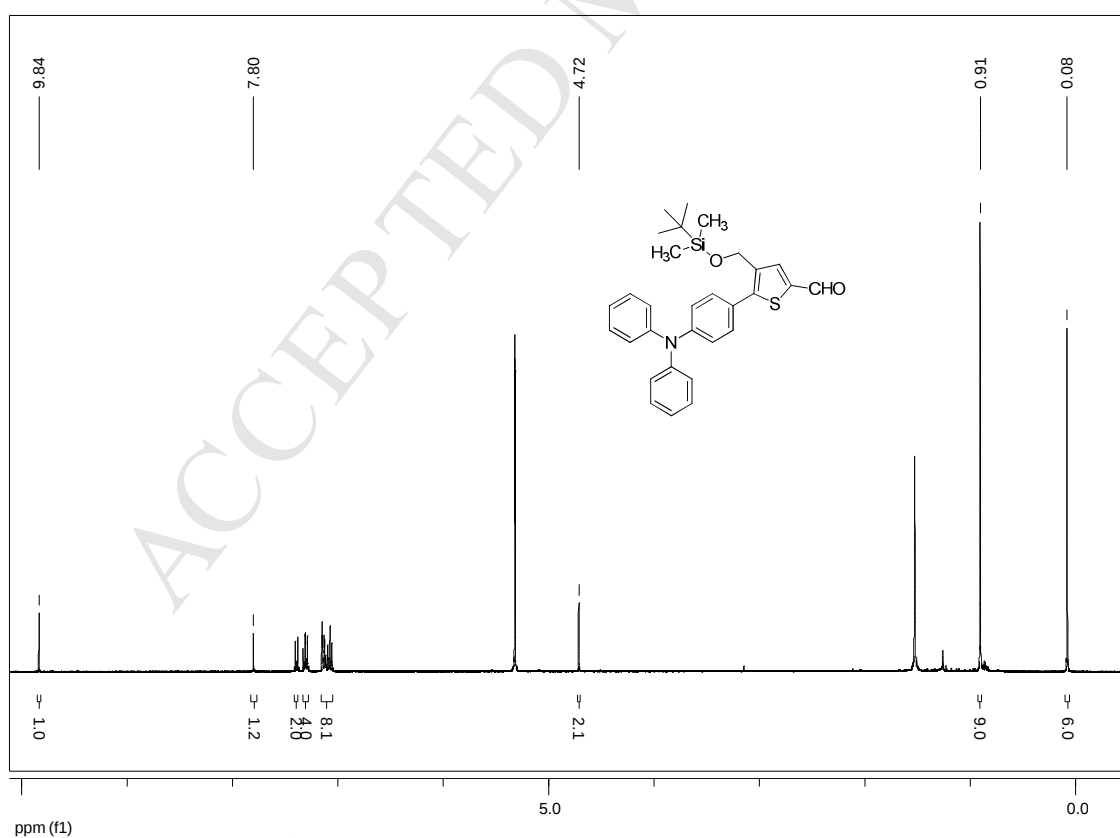
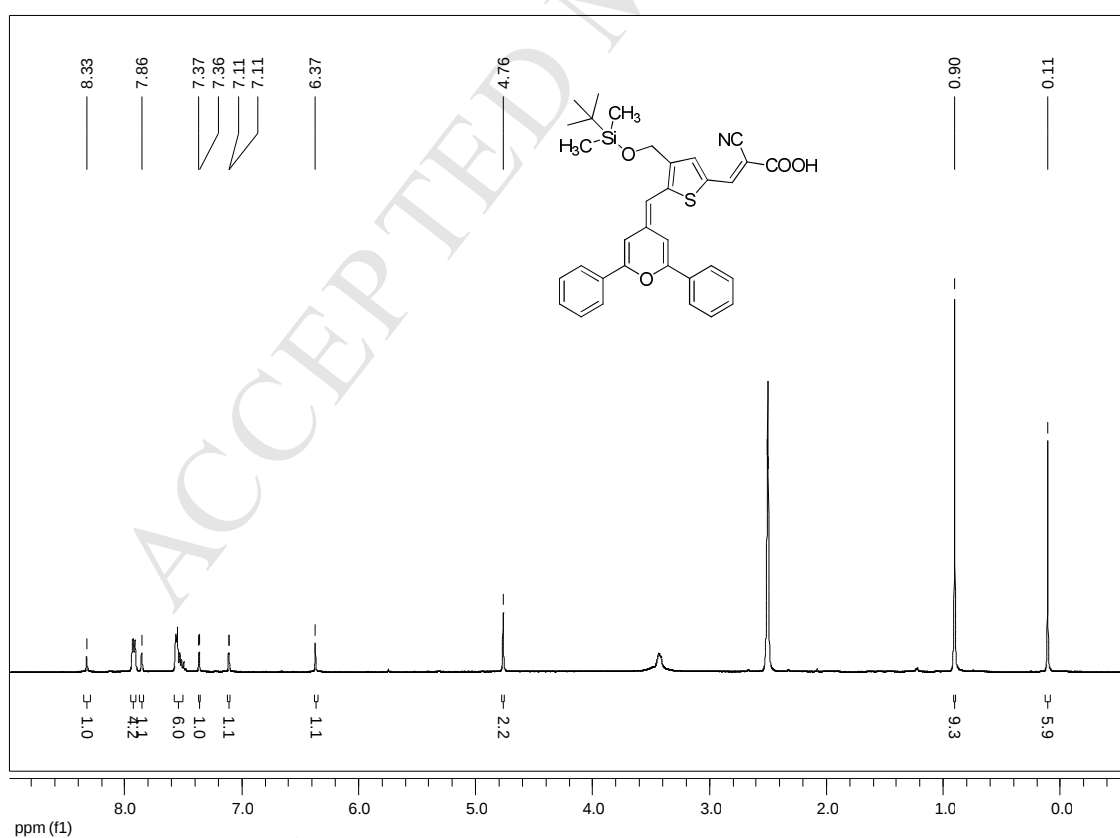
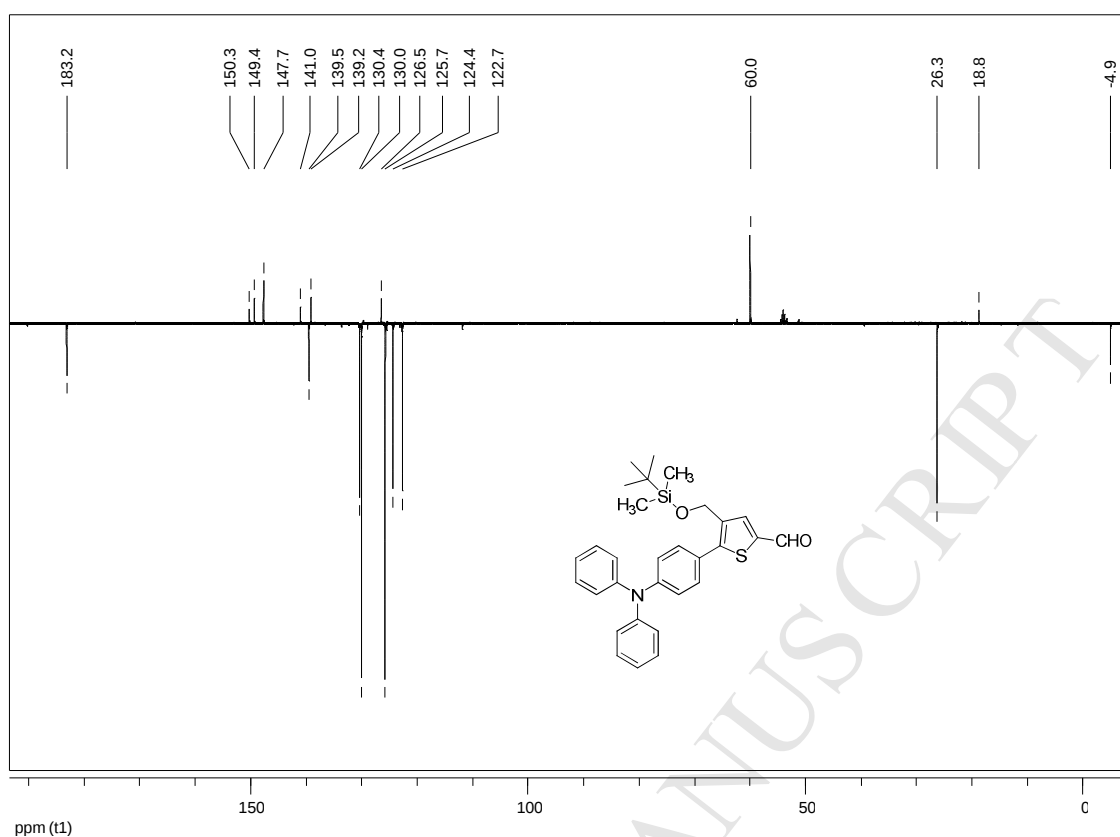


Figure S-7: ¹H NMR spectrum of compound **14** (400 MHz, CD₂Cl₂).



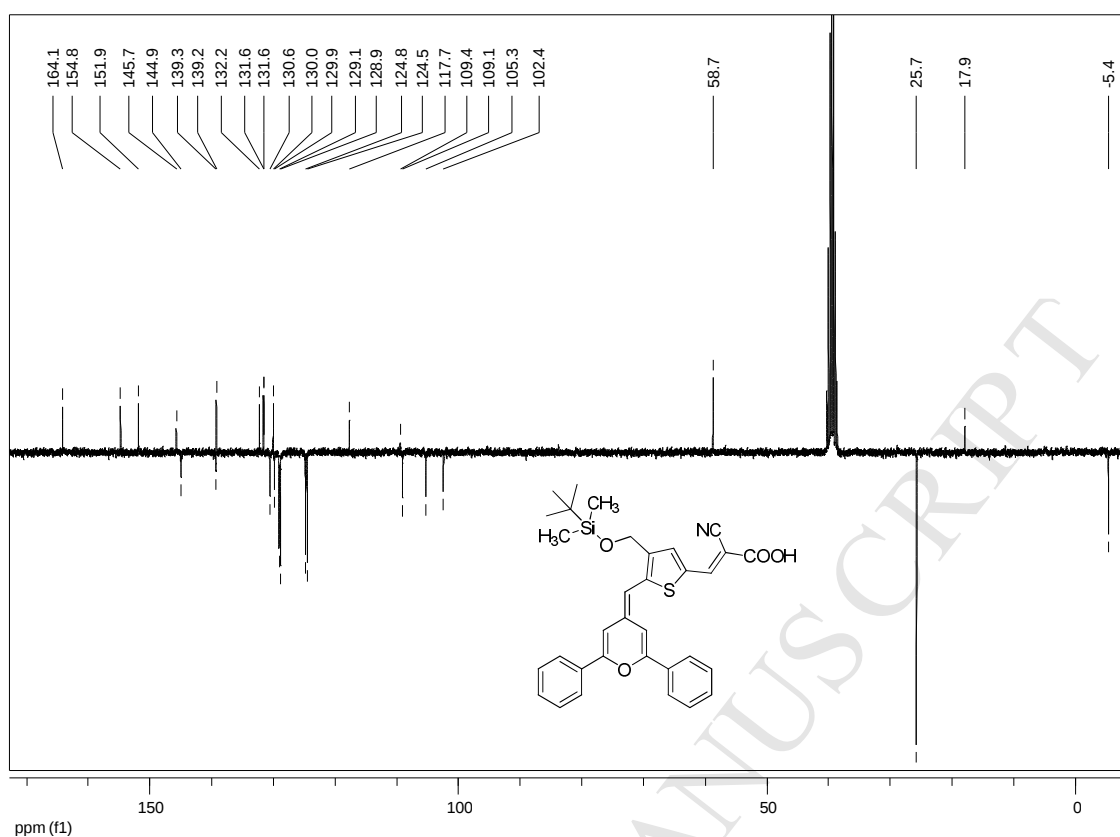


Figure S-10: ¹³C NMR (APT) spectrum of compound 1 (75 MHz, dmso-d₆).

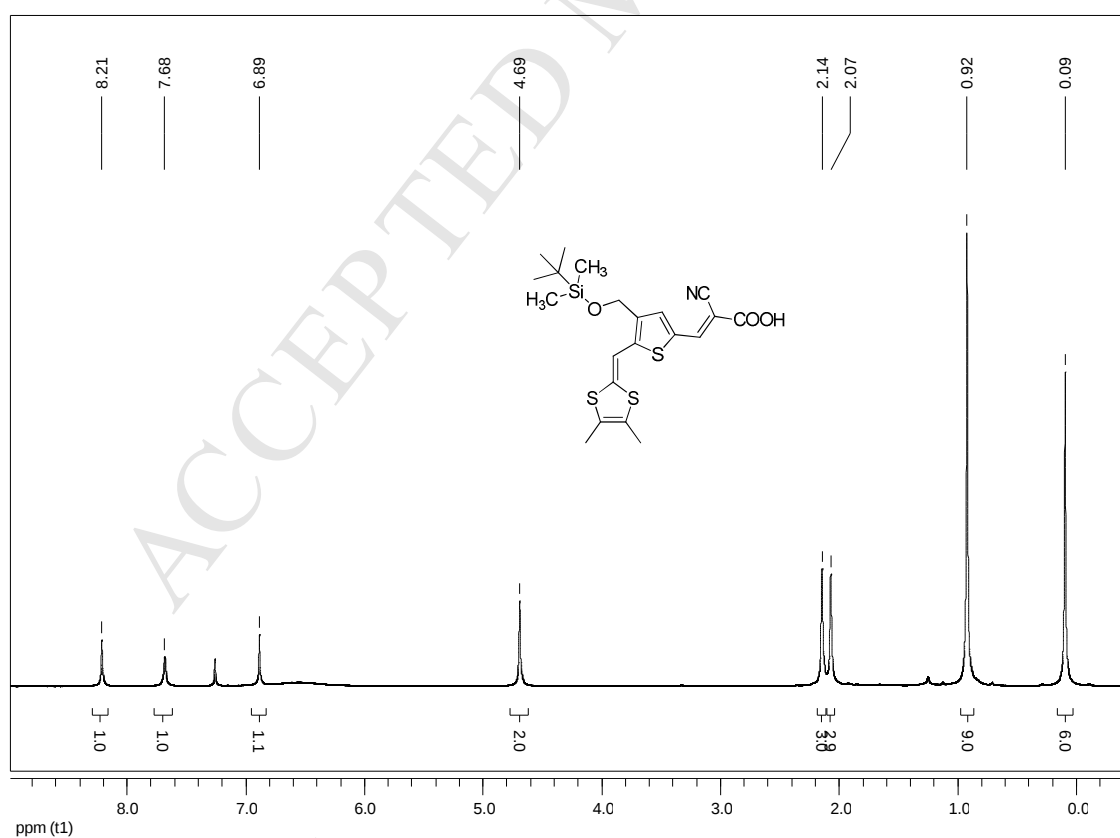


Figure S-11: ¹H NMR spectrum of compound 2 (300 MHz, CDCl₃).

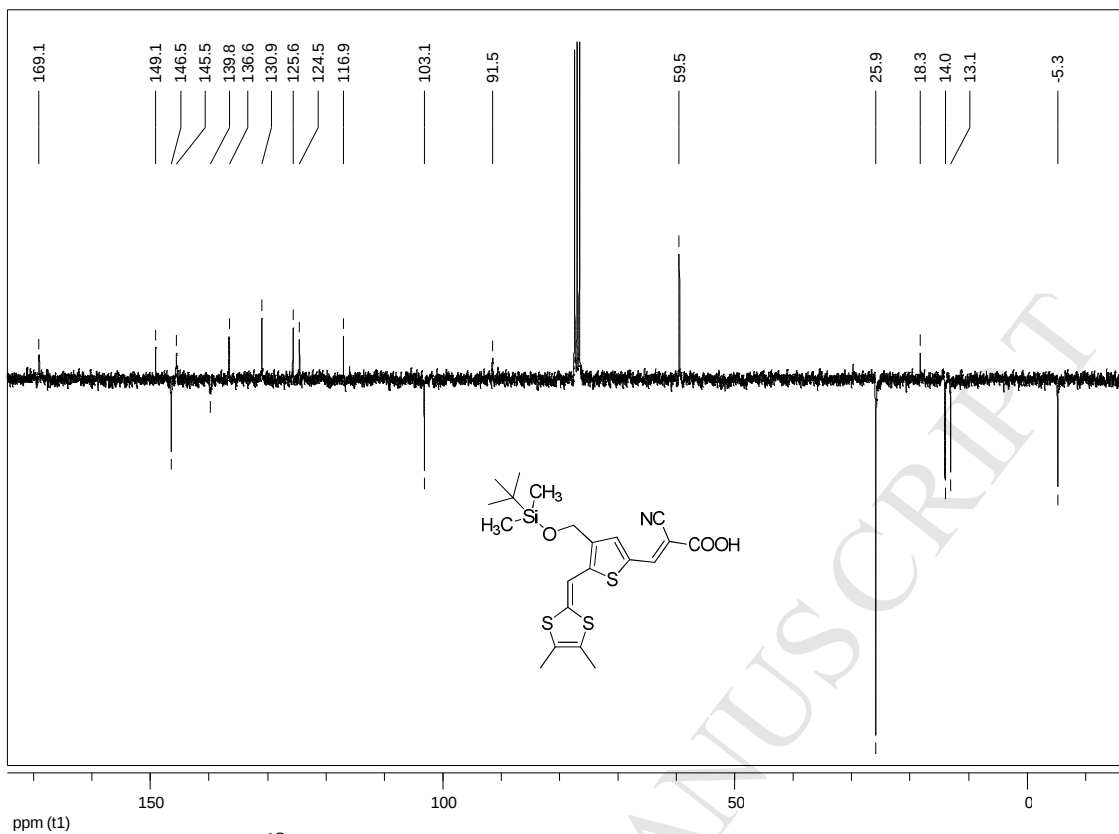


Figure S-12: ^{13}C NMR (APT) spectrum of compound **2** (75 MHz, CDCl_3).

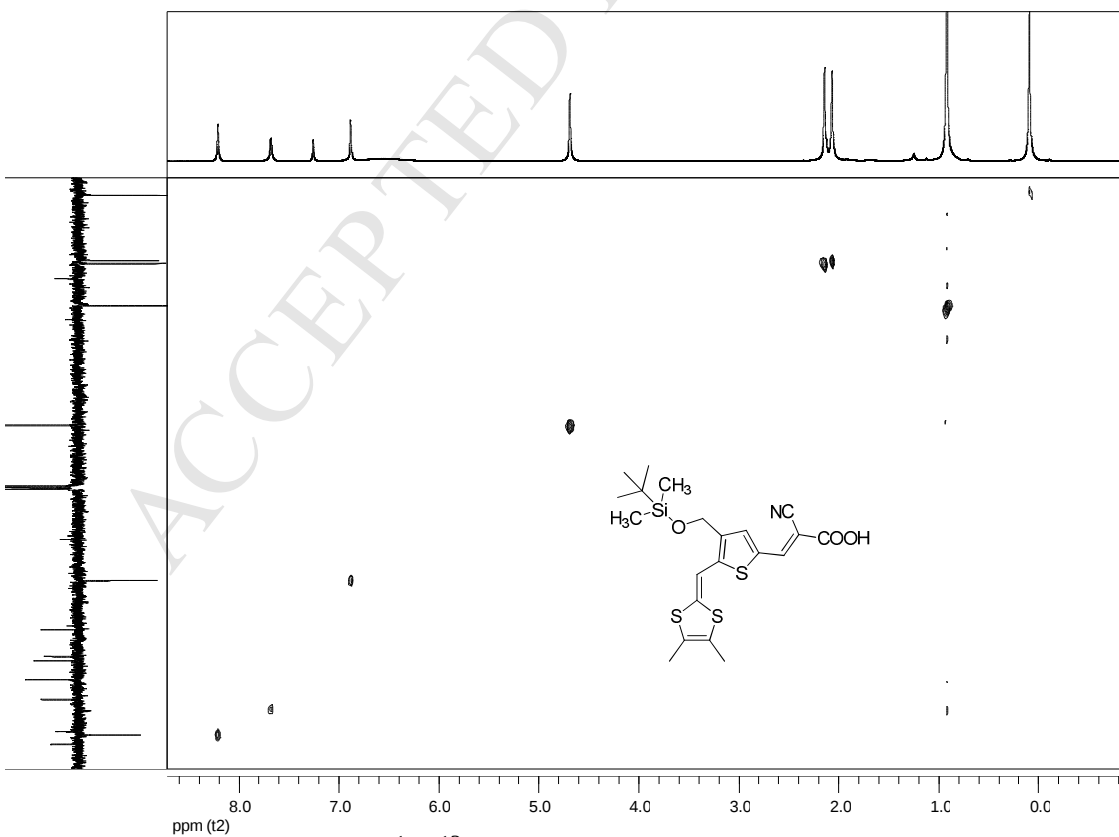


Figure S-13 ^1H - ^{13}C HSQC spectrum of compound **2**.

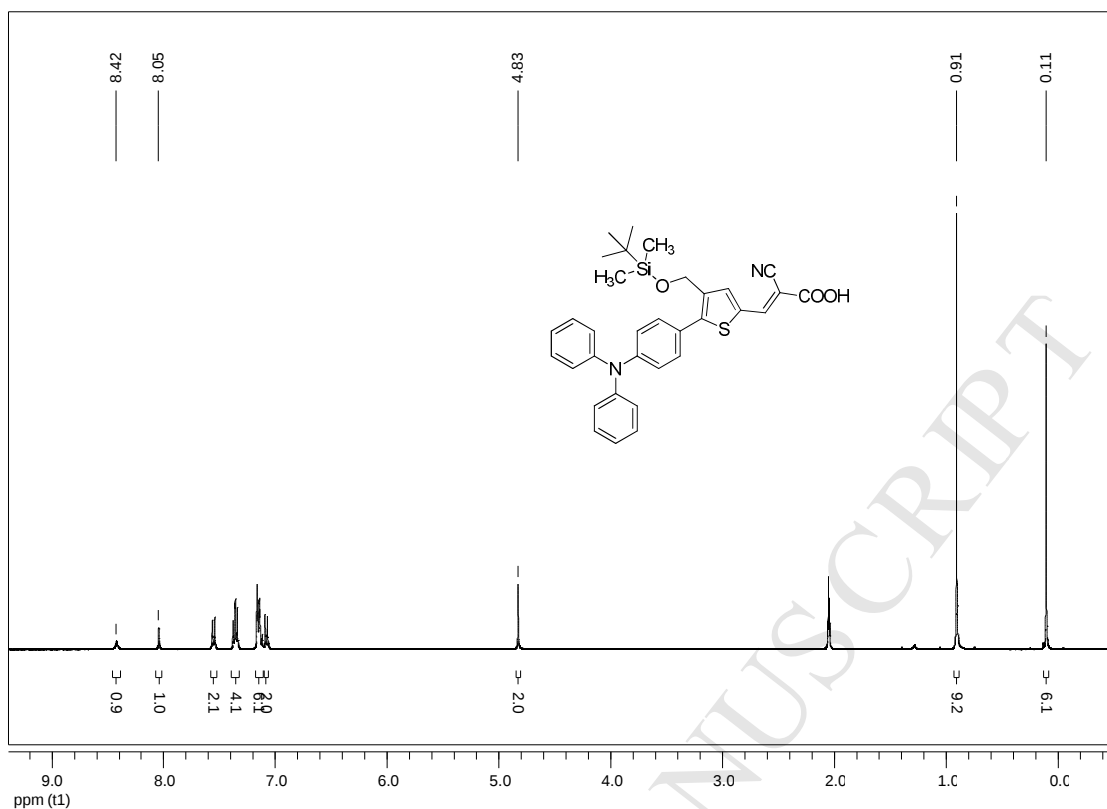


Figure S-14: ¹H NMR spectrum of compound **3** (400 MHz, acetone-d₆).

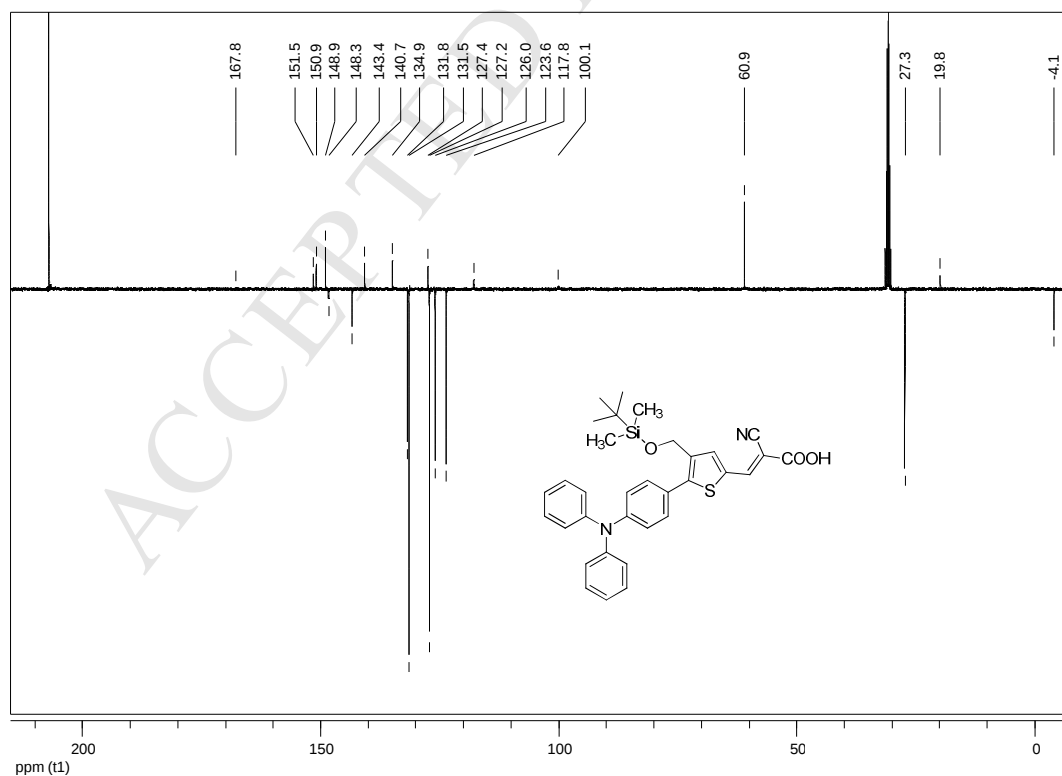


Figure S-15: ¹³C NMR (APT) spectrum of compound **3** (100 MHz, acetone-d₆).

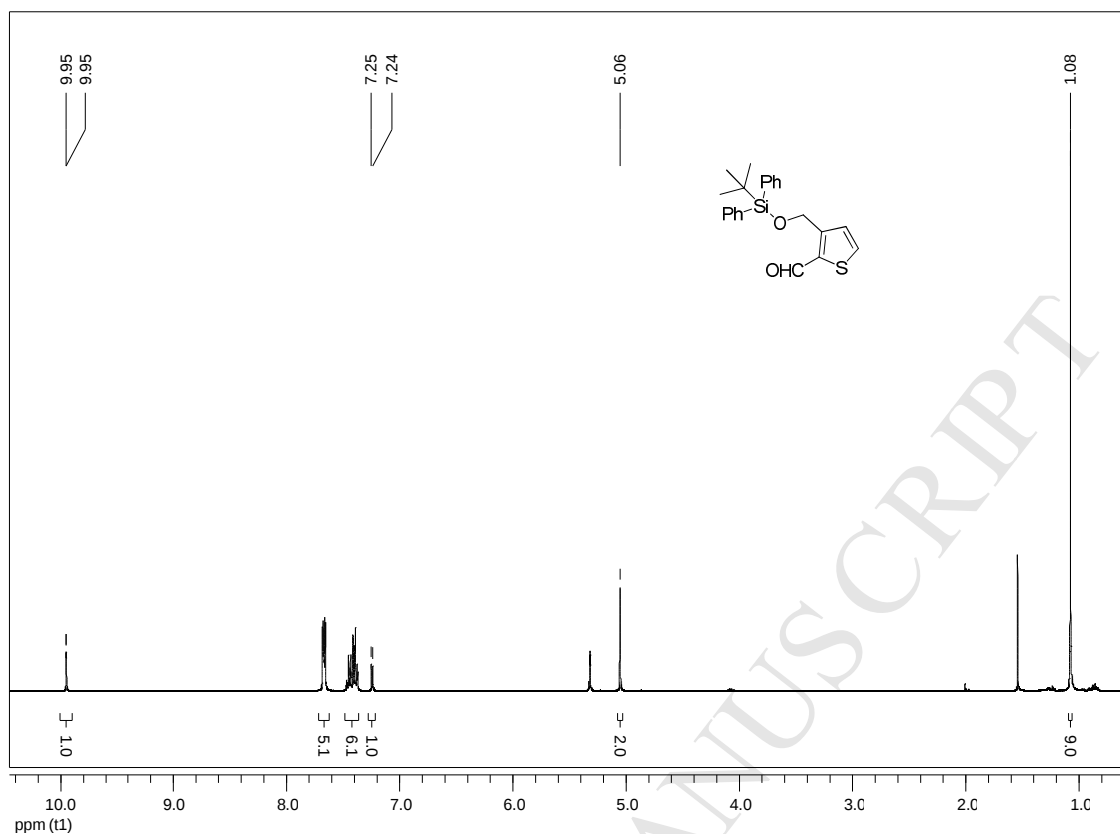


Figure S-16: ¹H NMR spectrum of compound **18** (400 MHz, CD₂Cl₂).

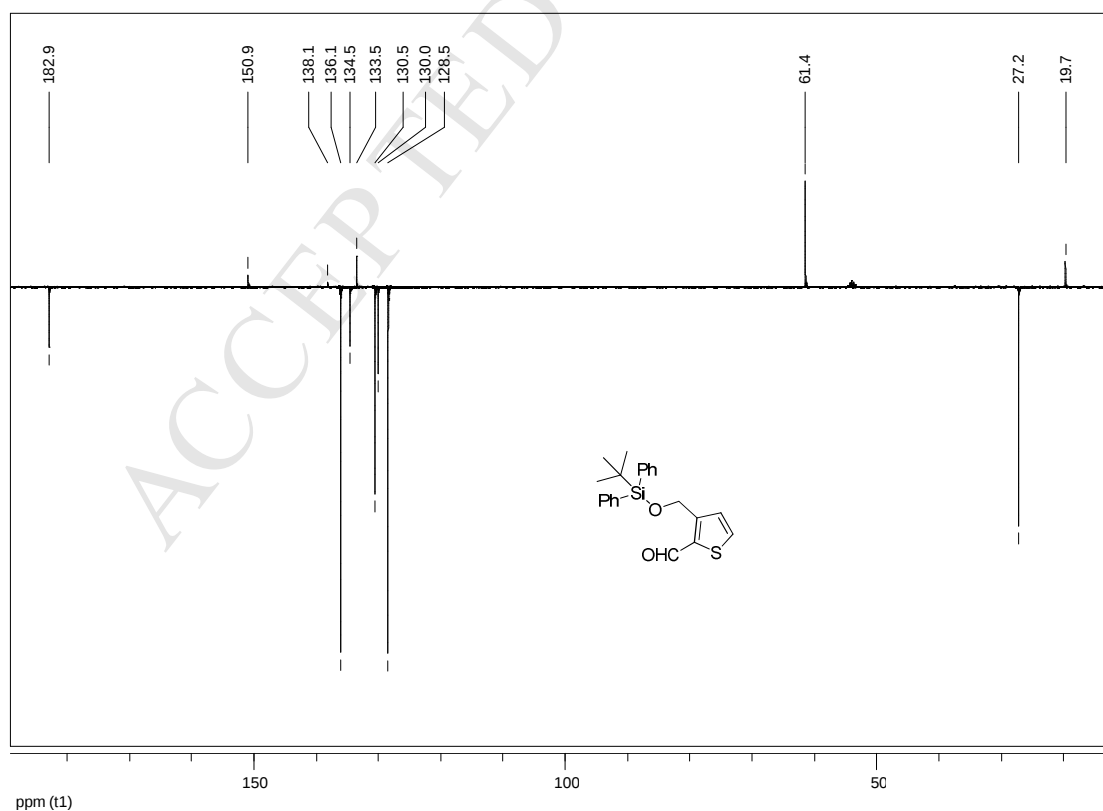


Figure S-17: ¹³C NMR (APT) spectrum of compound **18** (100 MHz, CD₂Cl₂).

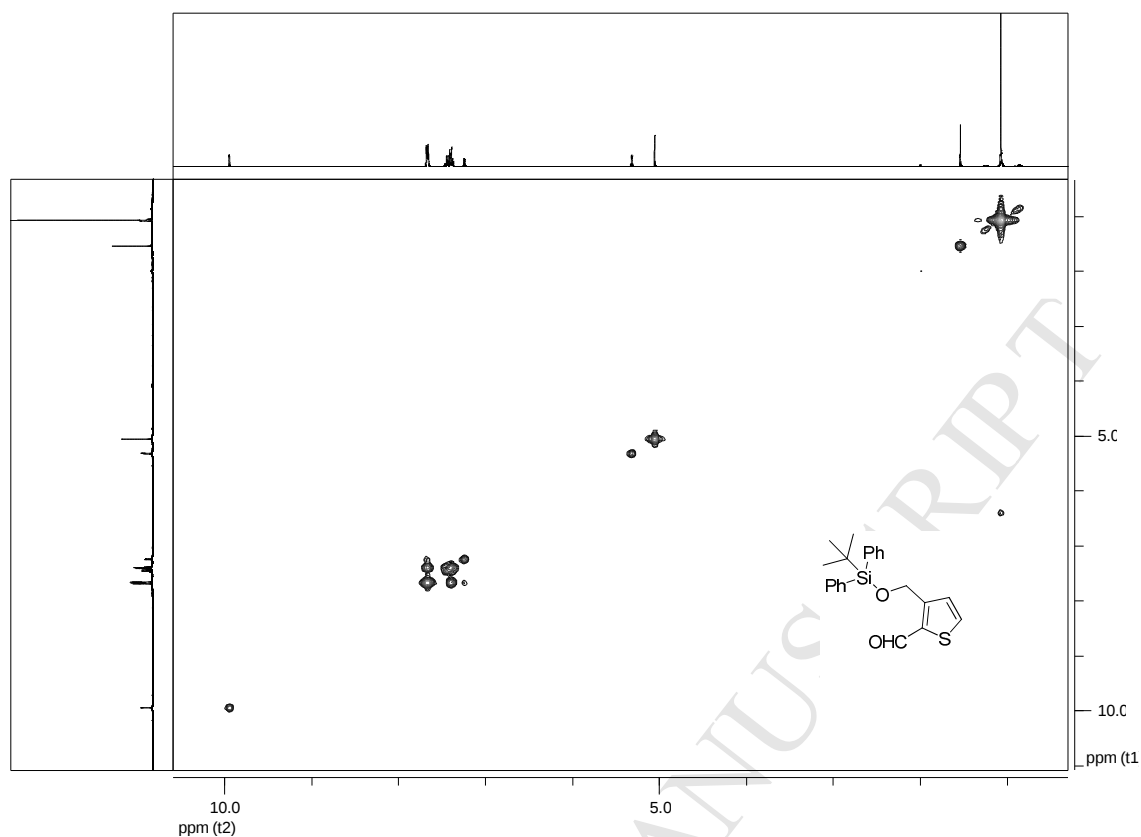


Figure S-18: ^1H - ^1H COSY spectrum of compound **18** (400 MHz, CD_2Cl_2).

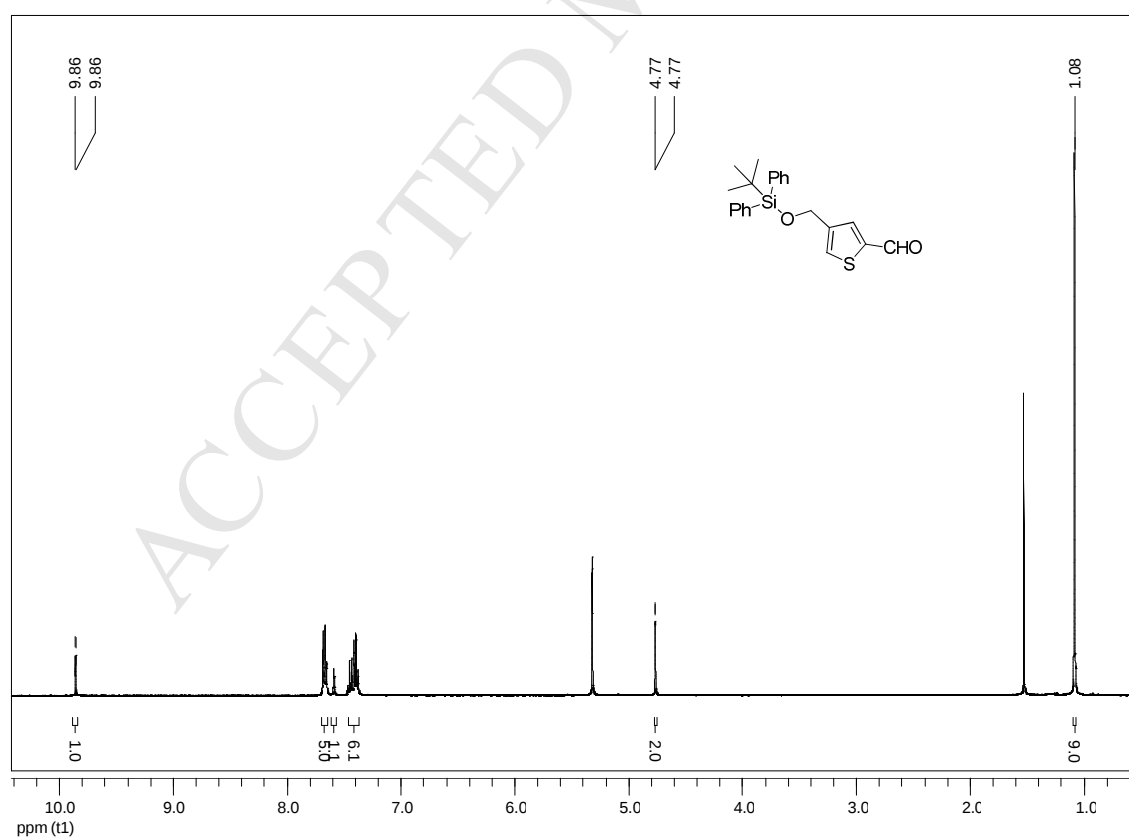


Figure S-19: ^1H NMR spectrum of compound **19** (400 MHz, CD_2Cl_2).

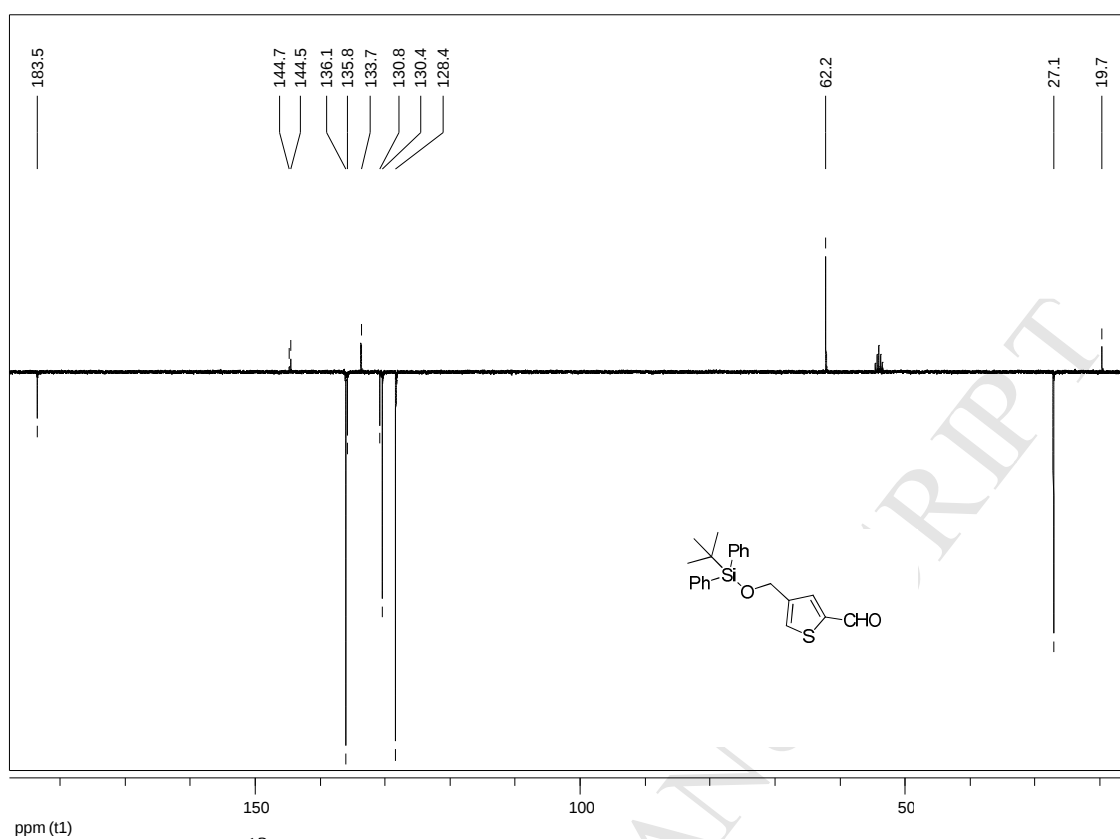


Figure S-20: ^{13}C NMR (APT) spectrum of compound **19** (100 MHz, CD_2Cl_2).

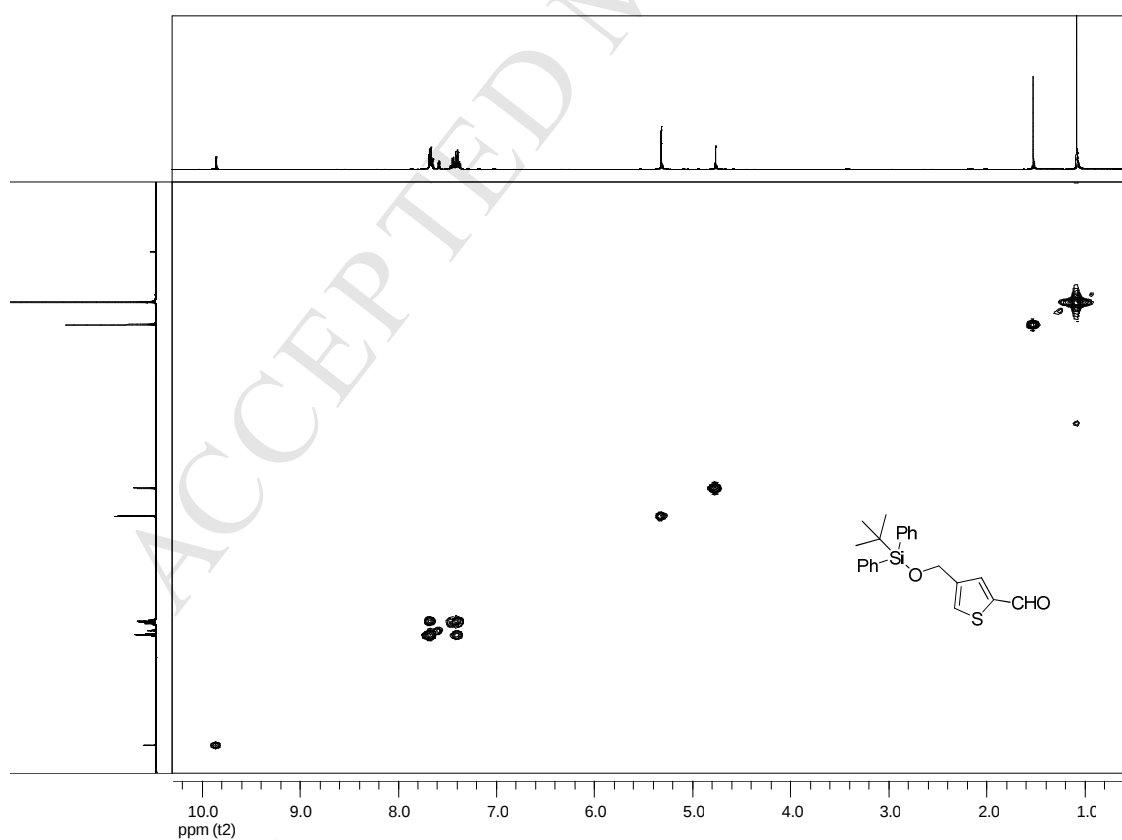


Figure S-21: ^1H - ^1H COSY spectrum of compound **19** (400 MHz, CD_2Cl_2).

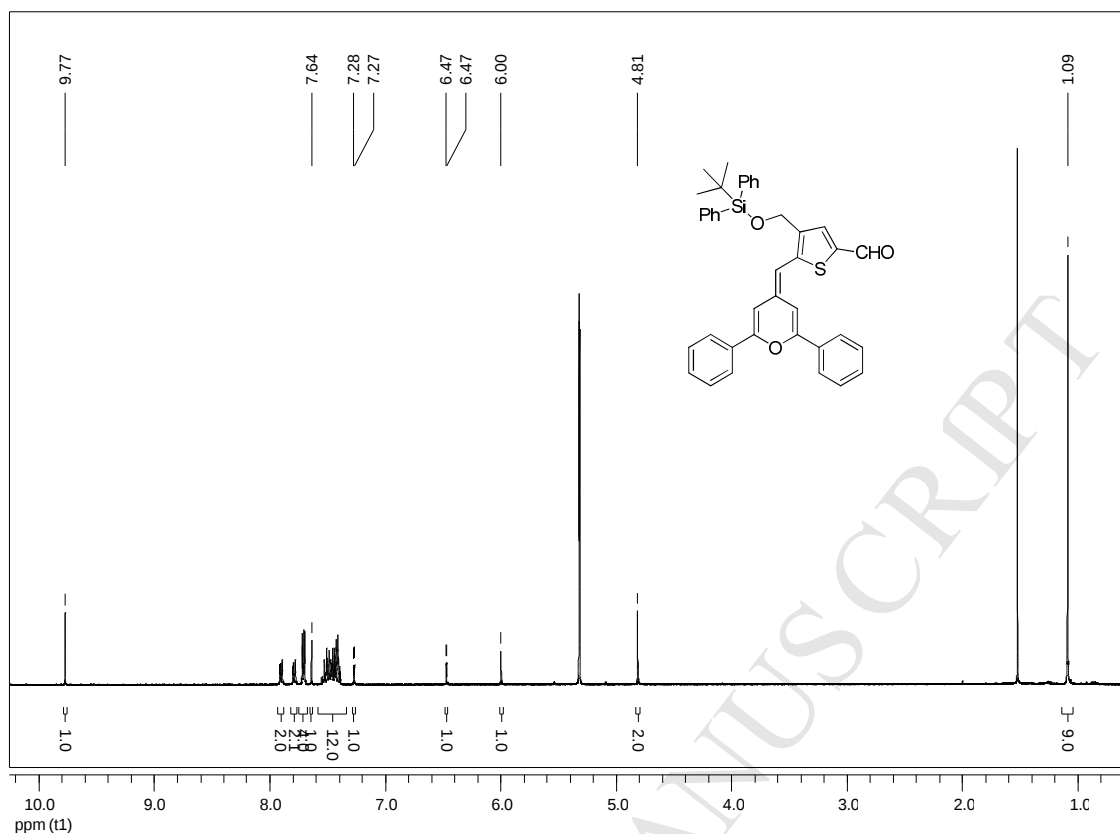


Figure S-22: ¹H NMR spectrum of compound **21** (400 MHz, CD₂Cl₂).

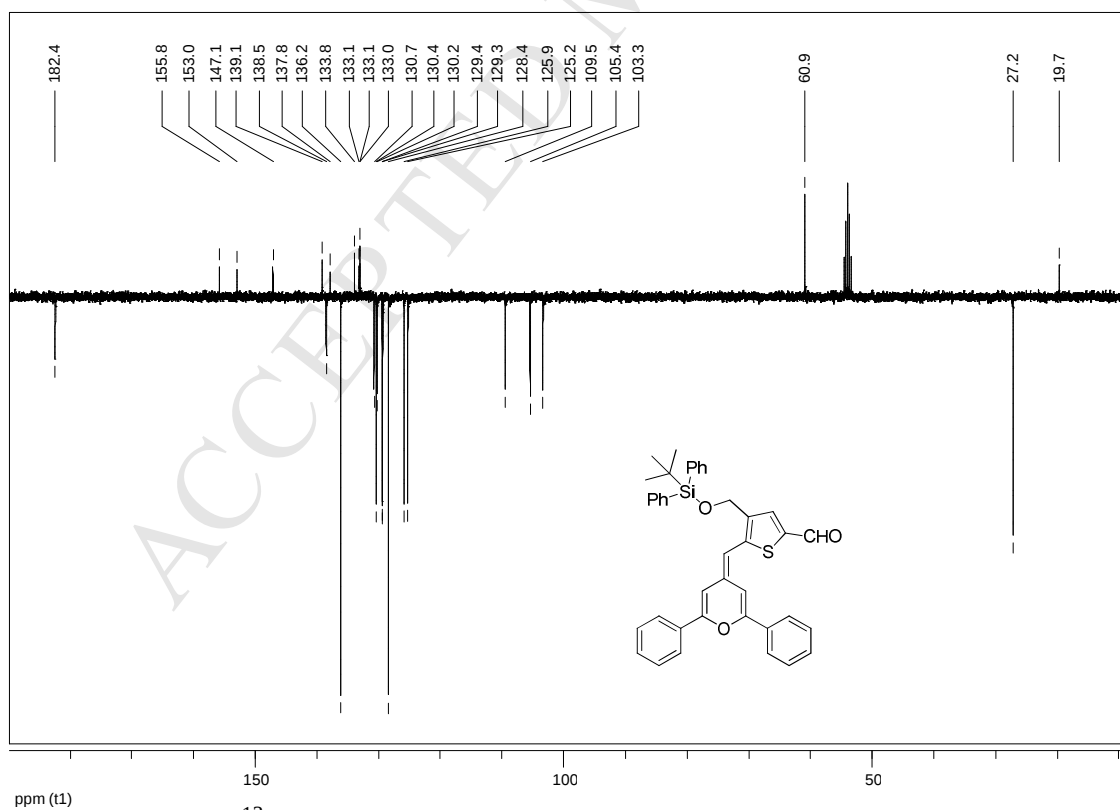


Figure S-23 ¹³C NMR (APT) spectrum of compound **21** (100 MHz, CD₂Cl₂).

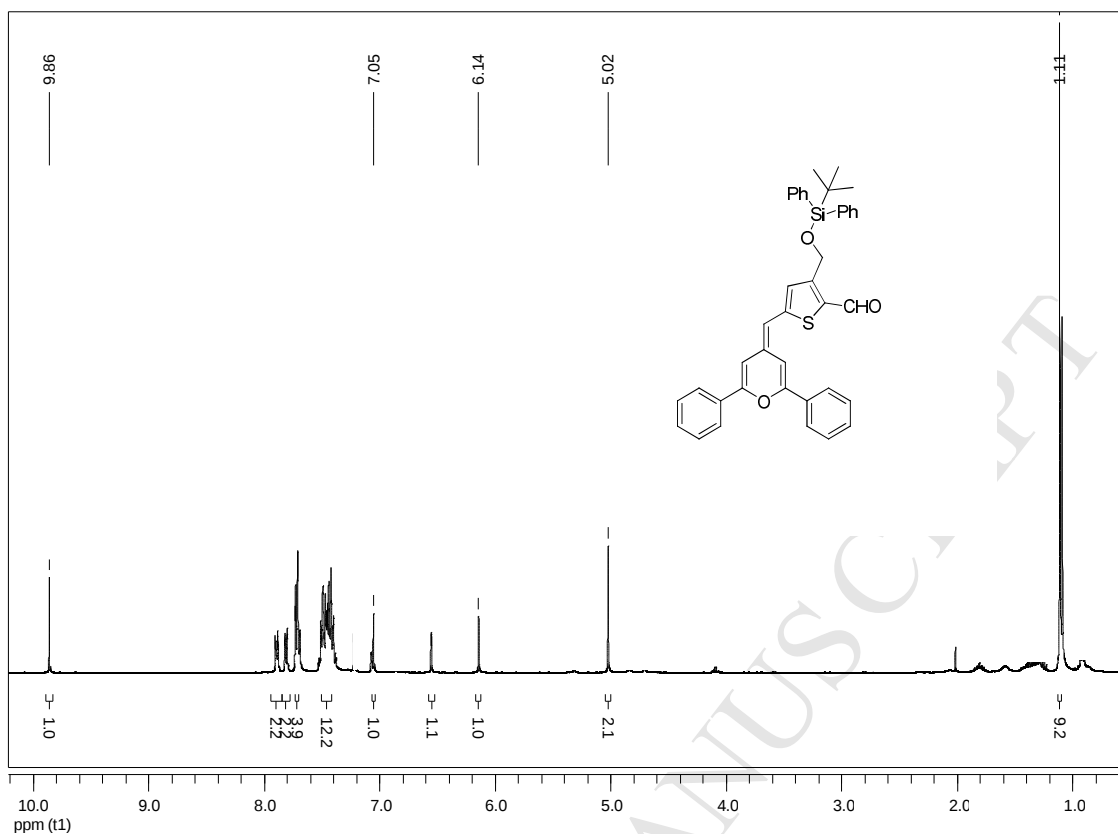


Figure S-24: ¹H NMR spectrum of compound **23** (400 MHz, CD₂Cl₂).

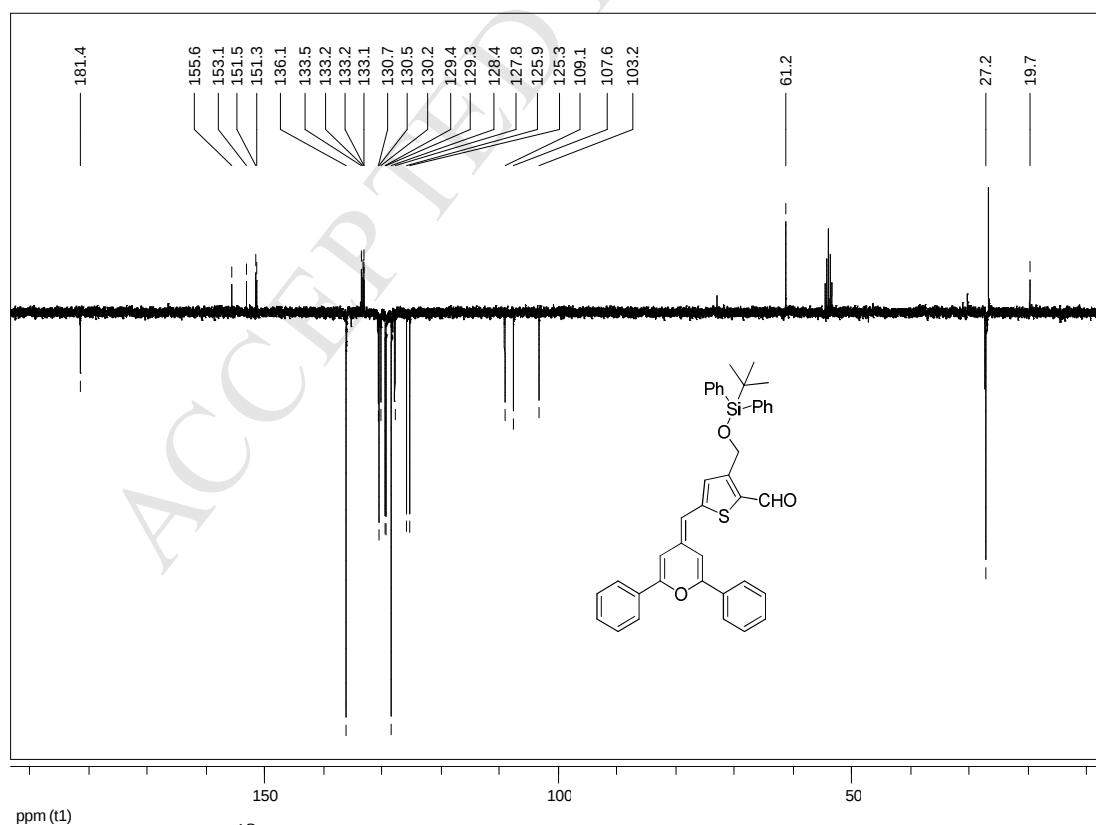


Figure S-25: ¹³C NMR (APT) spectrum of compound **23** (100 MHz, CD₂Cl₂).

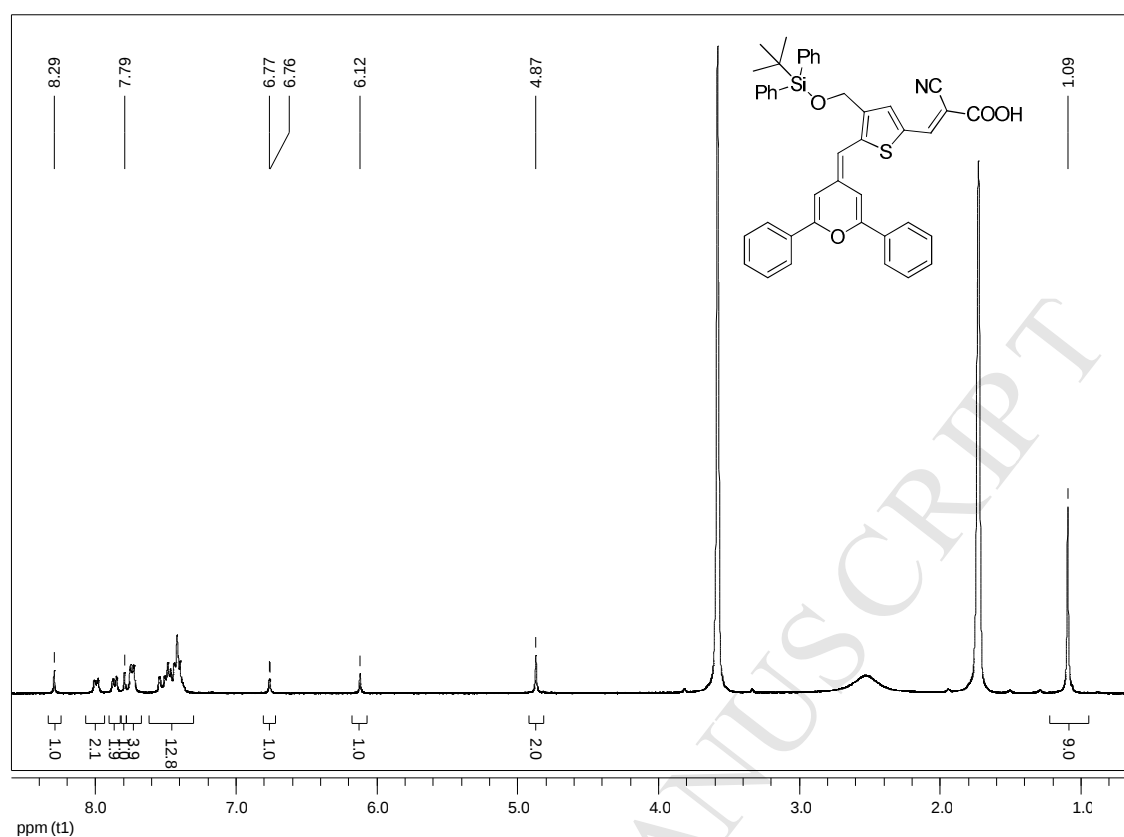


Figure S-26: ¹H NMR spectrum of compound 15 (300 MHz, THF-d₈).

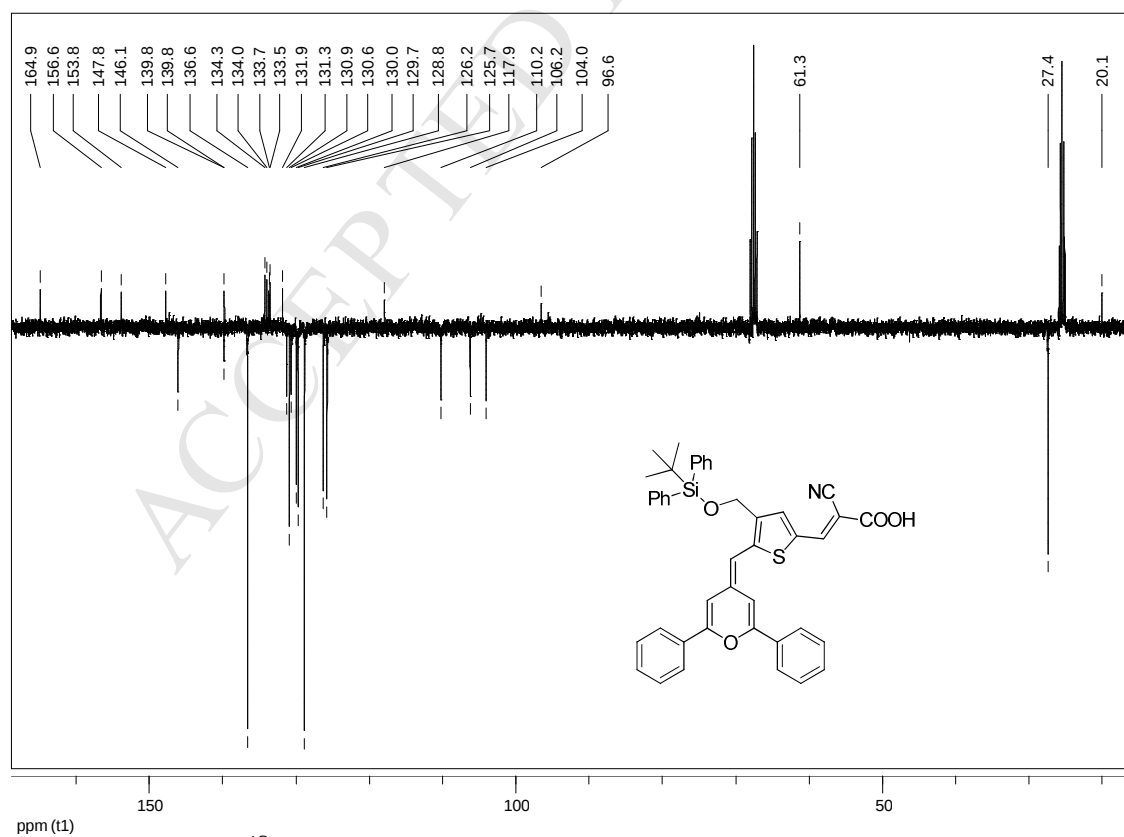


Figure S-27: ¹³C NMR (APT) spectrum of compound 15 (100 MHz, THF-d₈).

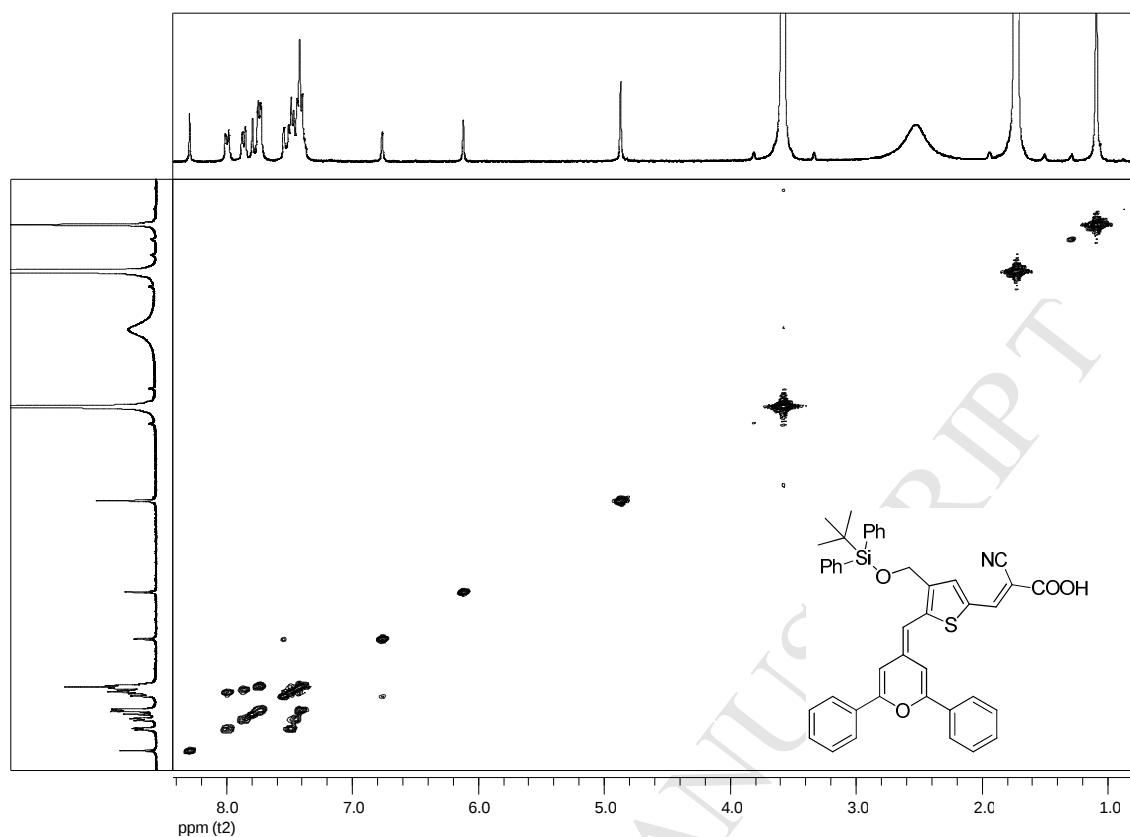


Figure S-28: ^1H - ^1H COSY spectrum of compound **15** (300 MHz, THF-d_8).

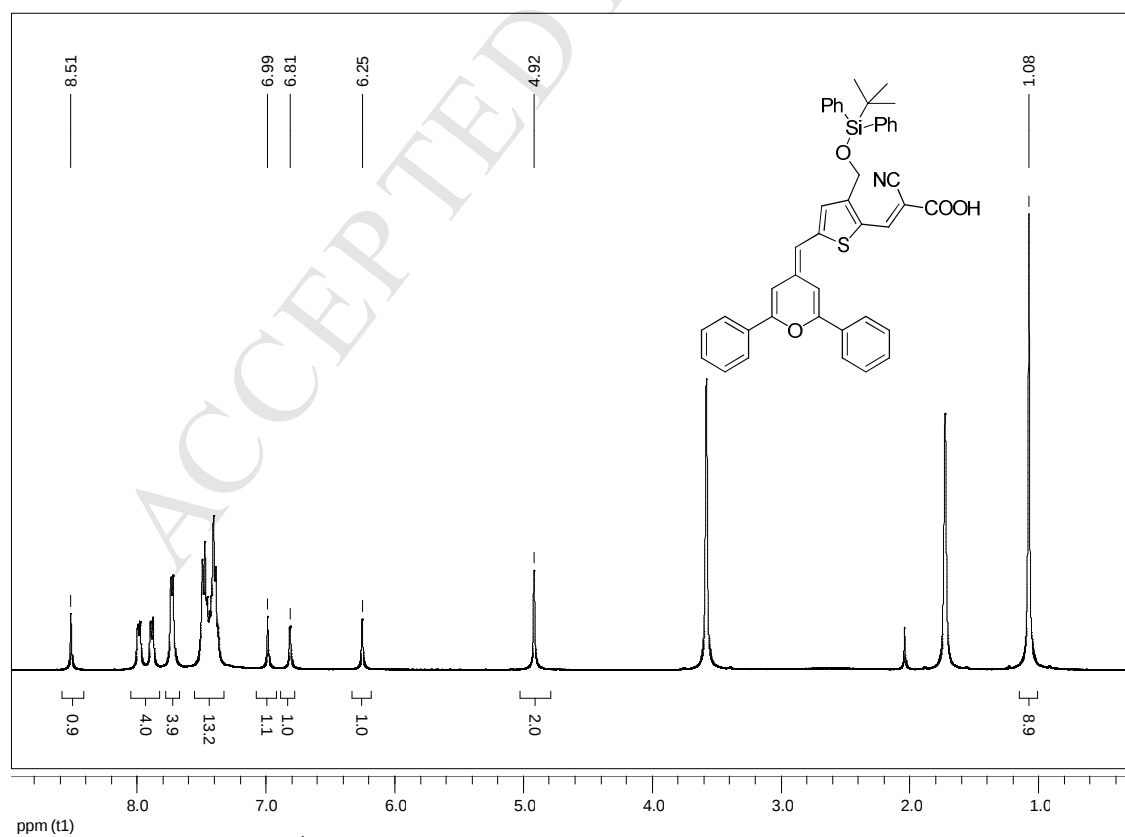
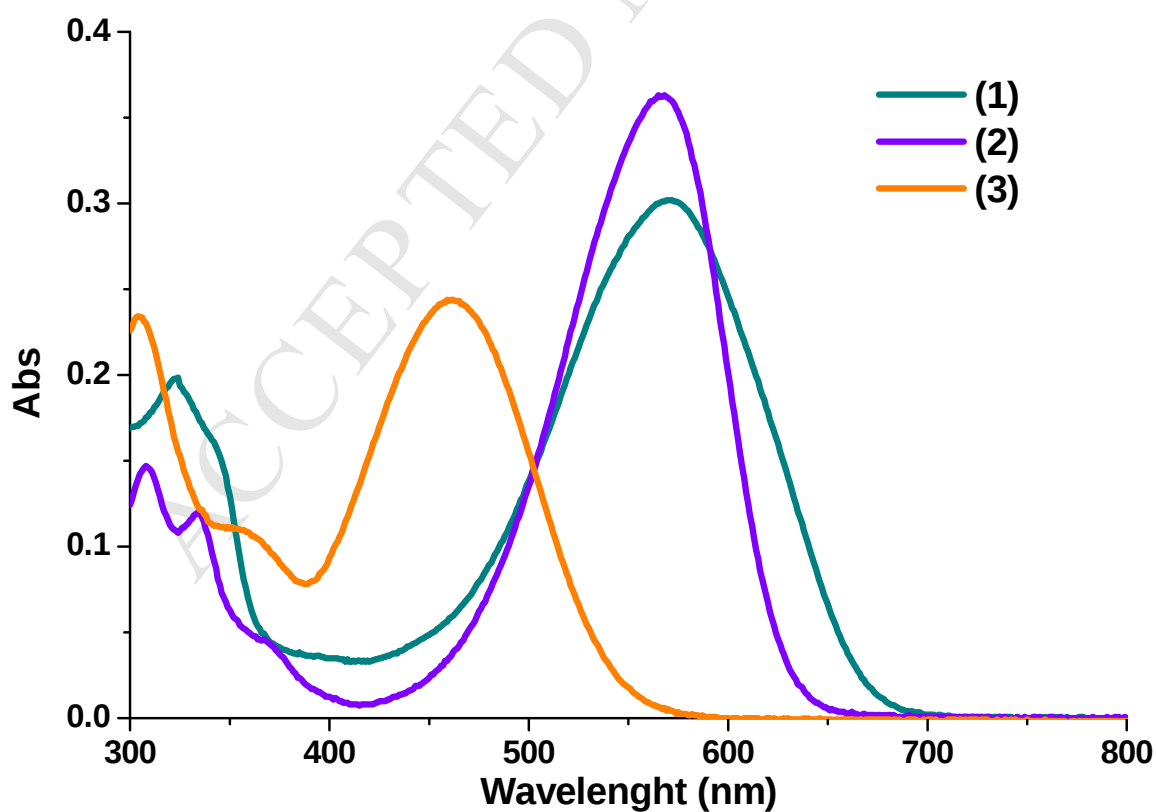
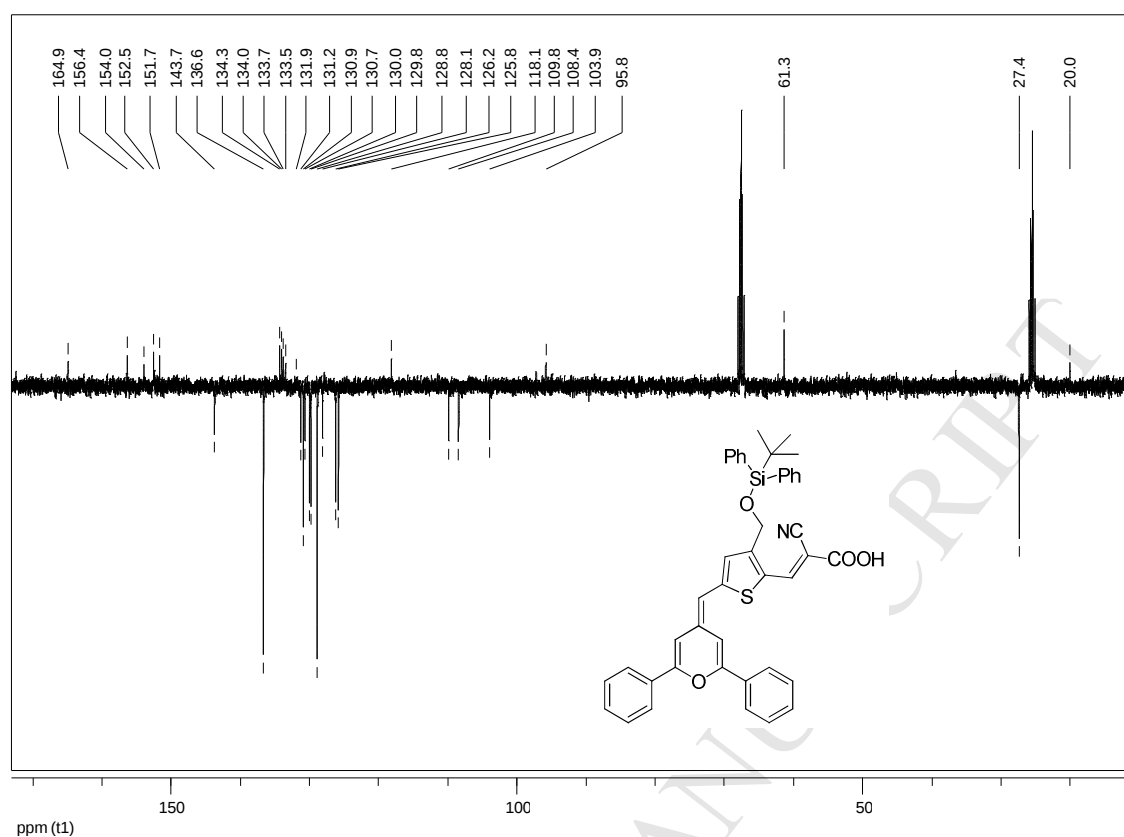


Figure S-29: ^1H NMR spectrum of compound **16** (400 MHz, THF-d_8).



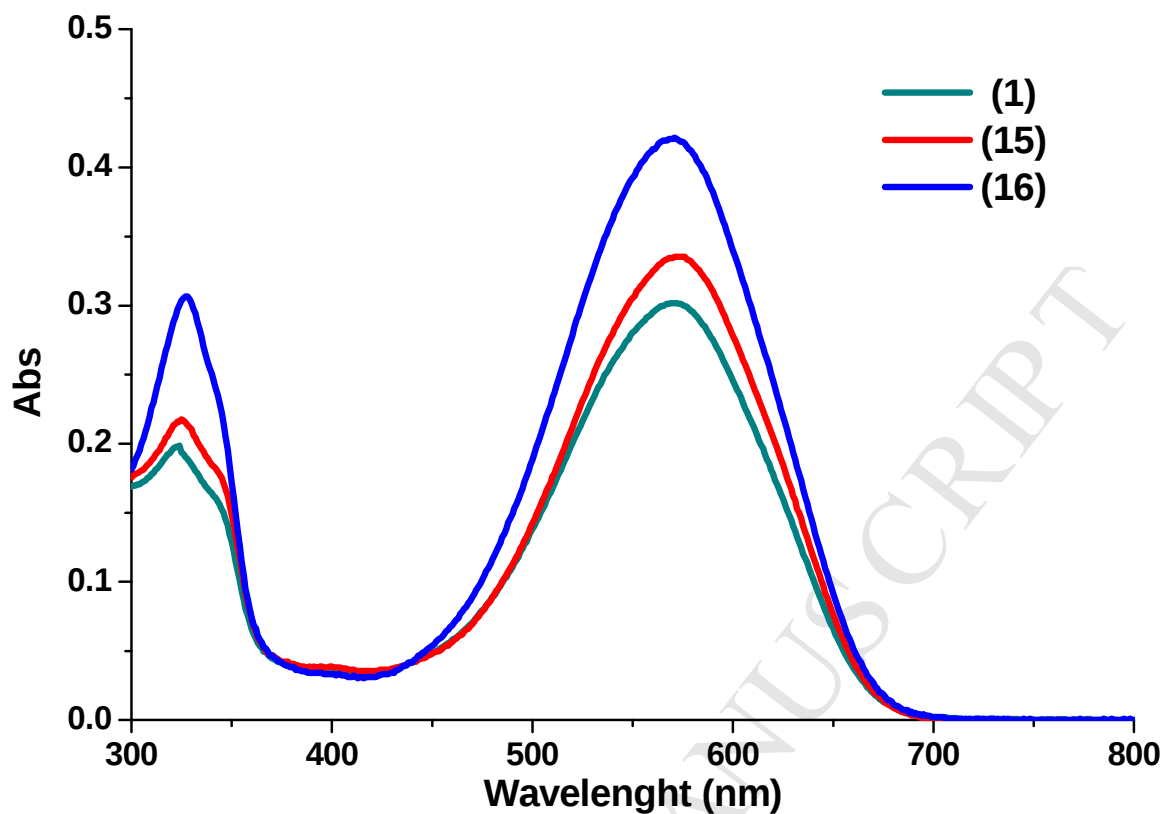


Figure S-32: Normalized UV-vis absorption of compounds (1, 15 and 16)

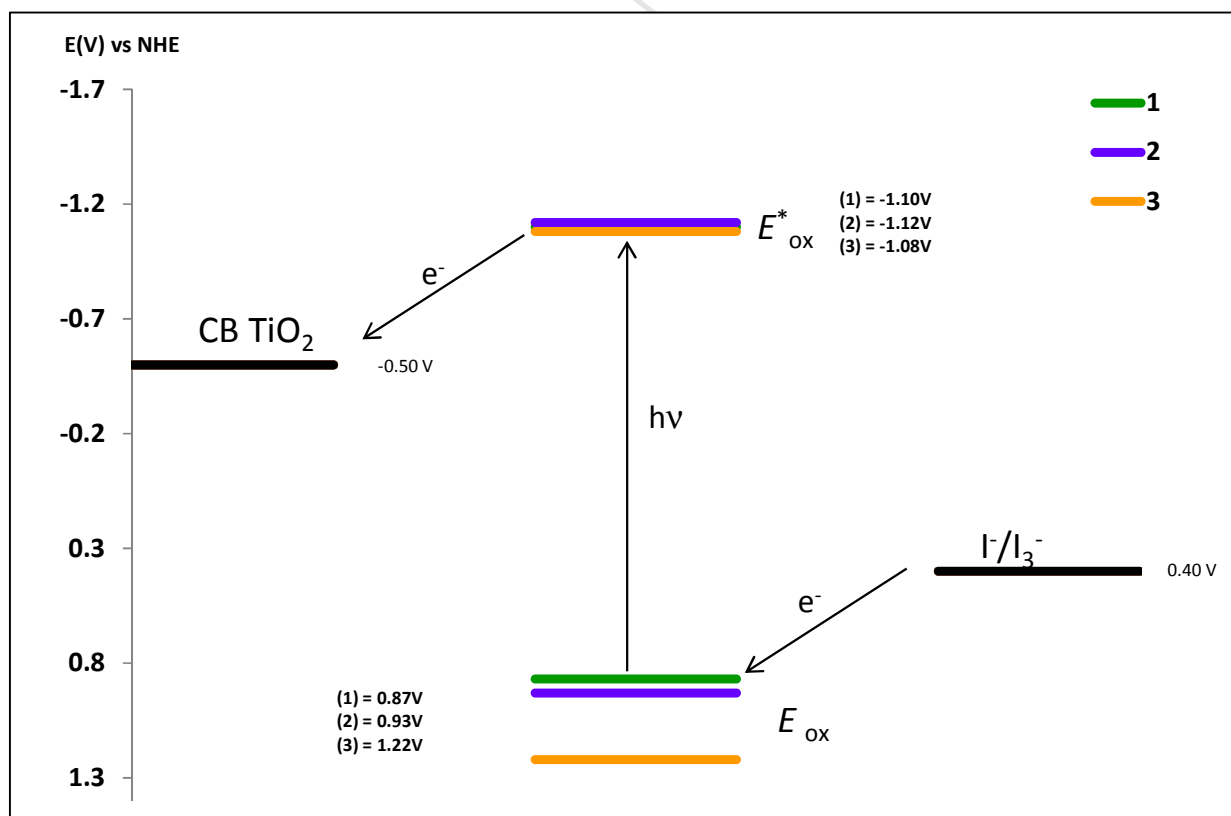


Figure S-33: Energy diagram of compounds (1-3)

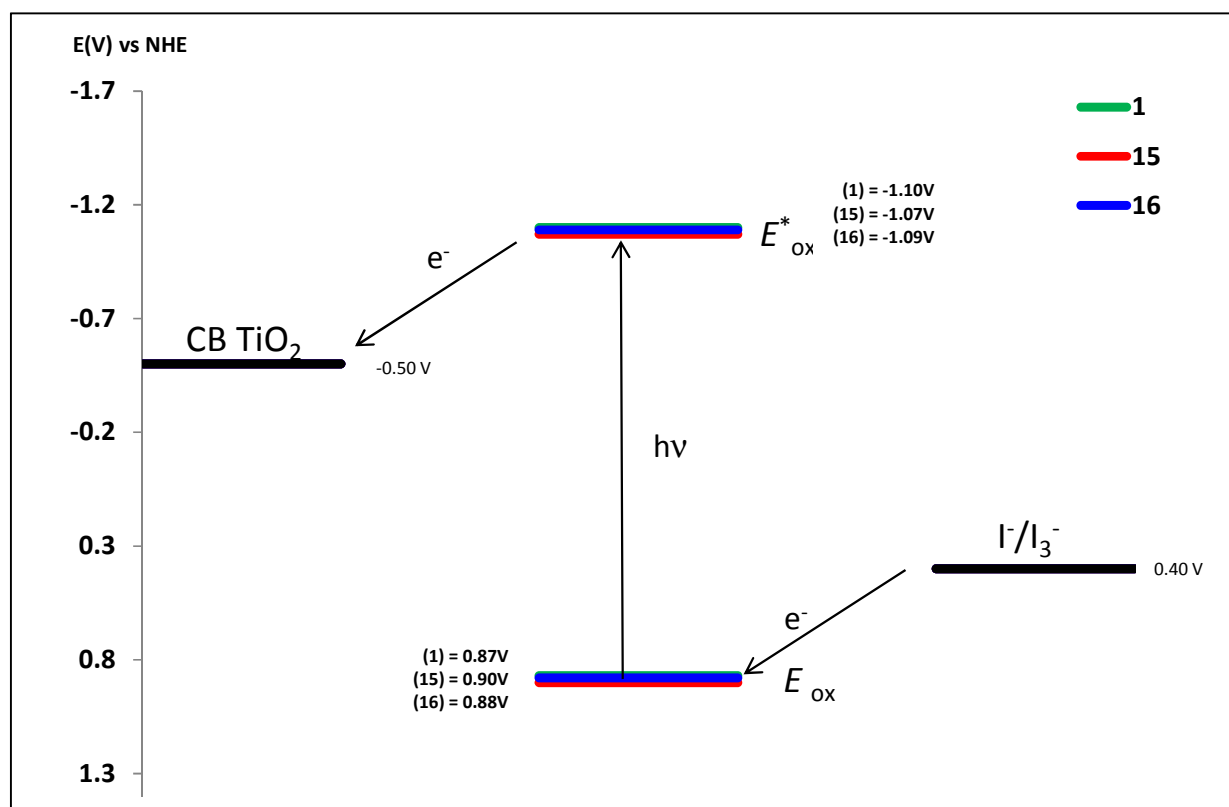


Figure S-34: Energy diagram of compounds (1, 15 and 16)

Calculated molecular geometries and energies (M06-2x/6-31G* PCM-Dichloromethane)**Compound 1: Ground State.**

Standard orientation:

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	3.829068	0.342082	0.207179
2	6	0	2.478089	0.326360	0.247164
3	6	0	1.731942	-0.892423	0.028851
4	6	0	2.564467	-2.070627	-0.092789
5	6	0	3.912057	-1.990169	-0.096131
6	1	0	1.981432	1.251855	0.495987
7	1	0	2.094994	-3.038354	-0.226174
8	8	0	4.551757	-0.789206	0.006538
9	6	0	0.365498	-1.000661	-0.060482
10	6	0	-0.629703	0.027659	-0.076143
11	6	0	-2.010315	-0.207906	-0.063898
12	16	0	-0.302215	1.731147	-0.194028
13	6	0	-2.763090	0.966586	-0.132055
14	6	0	-2.002098	2.128615	-0.204331
15	1	0	-3.847412	0.976890	-0.123754
16	1	0	-0.029902	-2.008728	-0.161306
17	6	0	-2.544154	3.435840	-0.260948
18	6	0	-1.913927	4.650039	-0.317747
19	1	0	-3.631334	3.469993	-0.256994
20	6	0	-2.612789	-1.585528	0.019983
21	1	0	-2.278136	-2.188277	-0.839224
22	1	0	-2.253256	-2.092671	0.928396
23	6	0	-0.492120	4.790478	-0.332263
24	7	0	0.663384	4.891731	-0.343831
25	6	0	-2.657749	5.922795	-0.367085
26	8	0	-2.134075	7.014452	-0.420847
27	8	0	-3.990355	5.753046	-0.346417
28	1	0	-4.392111	6.639242	-0.380817
29	8	0	-4.018277	-1.479167	0.034402
30	14	0	-4.999634	-2.846555	0.090944
31	6	0	-4.816614	-3.818593	-1.504009
32	1	0	-3.778480	-4.130953	-1.660826
33	1	0	-5.429549	-4.725896	-1.474552
34	1	0	-5.125731	-3.228056	-2.372478
35	6	0	-4.487610	-3.907903	1.553904
36	1	0	-4.423236	-3.313236	2.471208
37	1	0	-5.213668	-4.710786	1.722829
38	1	0	-3.512502	-4.377580	1.386236
39	6	0	-6.744843	-2.154285	0.295155
40	6	0	-6.866241	-1.467502	1.662589
41	6	0	-7.025867	-1.129039	-0.812023
42	6	0	-7.770280	-3.293237	0.203372
43	1	0	-6.712658	-2.176170	2.484736
44	1	0	-6.134951	-0.658233	1.770875
45	1	0	-7.868193	-1.032724	1.781113
46	1	0	-6.952751	-1.578621	-1.809370
47	1	0	-8.042076	-0.725656	-0.703841
48	1	0	-6.323787	-0.289436	-0.767002
49	1	0	-8.785384	-2.897665	0.344165
50	1	0	-7.741206	-3.787721	-0.774706
51	1	0	-7.605519	-4.056479	0.973437
52	6	0	4.855745	-3.111812	-0.242315
53	6	0	6.163844	-2.874641	-0.682069
54	6	0	4.459138	-4.421588	0.055315
55	6	0	7.052623	-3.933559	-0.835810
56	1	0	6.477081	-1.861900	-0.912125
57	6	0	5.350763	-5.475813	-0.101663
58	1	0	3.460276	-4.615887	0.433294
59	6	0	6.649344	-5.235704	-0.549181
60	1	0	8.062636	-3.739914	-1.182854
61	1	0	5.034268	-6.486085	0.137019
62	1	0	7.345593	-6.059797	-0.668074
63	6	0	4.675973	1.532870	0.392938
64	6	0	6.022594	1.390435	0.748475
65	6	0	4.143206	2.817146	0.216961
66	6	0	6.816772	2.515999	0.940871

67	1	0	6.440660	0.398578	0.883721
68	6	0	4.940125	3.937364	0.415601
69	1	0	3.111633	2.948415	-0.096955
70	6	0	6.278417	3.790793	0.779189
71	1	0	7.858306	2.395931	1.221487
72	1	0	4.515437	4.926174	0.275007
73	1	0	6.900173	4.667618	0.930565

E(RM062X) = -2320.01579137

Compound 1: First Excited State.

Standard orientation:

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	3.836092	0.290195	0.276223
2	6	0	2.472512	0.282149	0.330487
3	6	0	1.736599	-0.910680	0.091749
4	6	0	2.513623	-2.089577	-0.101965
5	6	0	3.877057	-2.045331	-0.106981
6	1	0	1.979051	1.201536	0.609828
7	1	0	2.015515	-3.036345	-0.274620
8	8	0	4.525935	-0.857127	0.044060
9	6	0	0.315085	-0.997692	0.046313
10	6	0	-0.626250	0.015848	-0.019158
11	6	0	-2.048843	-0.177854	-0.018474
12	16	0	-0.254543	1.735927	-0.182879
13	6	0	-2.760202	0.990737	-0.123002
14	6	0	-1.958884	2.161669	-0.216274
15	1	0	-3.843023	1.031773	-0.132816
16	1	0	-0.082820	-2.010871	0.032661
17	6	0	-2.454264	3.464019	-0.313147
18	6	0	-1.759528	4.673012	-0.395421
19	1	0	-3.538013	3.544080	-0.322958
20	6	0	-2.674385	-1.538972	0.093305
21	1	0	-2.320850	-2.169321	-0.738804
22	1	0	-2.337091	-2.019978	1.024816
23	6	0	-0.341497	4.768637	-0.392845
24	7	0	0.819216	4.851907	-0.390109
25	6	0	-2.447120	5.960517	-0.485148
26	8	0	-1.894690	7.043534	-0.559455
27	8	0	-3.792795	5.844001	-0.478304
28	1	0	-4.149143	6.747423	-0.538939
29	8	0	-4.075626	-1.419182	0.071879
30	14	0	-5.074803	-2.775302	0.118099
31	6	0	-4.886190	-3.747060	-1.476130
32	1	0	-3.848869	-4.065009	-1.627093
33	1	0	-5.504219	-4.651040	-1.451337
34	1	0	-5.186965	-3.153914	-2.345720
35	6	0	-4.586712	-3.841472	1.585513
36	1	0	-4.527406	-3.248093	2.504002
37	1	0	-5.320523	-4.638933	1.746175
38	1	0	-3.613478	-4.318463	1.427430
39	6	0	-6.812645	-2.061290	0.305066
40	6	0	-6.941275	-1.379382	1.674346
41	6	0	-7.066952	-1.027400	-0.800615
42	6	0	-7.851078	-3.187103	0.196000
43	1	0	-6.806775	-2.093817	2.494840
44	1	0	-6.201191	-0.580035	1.795197
45	1	0	-7.938965	-0.932467	1.783028
46	1	0	-6.989611	-1.473813	-1.799042
47	1	0	-8.078356	-0.610079	-0.701277
48	1	0	-6.353160	-0.198483	-0.744111
49	1	0	-8.862656	-2.779769	0.327954
50	1	0	-7.817288	-3.676743	-0.784355
51	1	0	-7.703952	-3.956239	0.963822
52	6	0	4.788257	-3.173490	-0.296833
53	6	0	6.137887	-2.944796	-0.607073
54	6	0	4.331022	-4.495002	-0.168961
55	6	0	7.003349	-4.015653	-0.796135
56	1	0	6.500318	-1.928144	-0.712033

57	6	0	5.201485	-5.558708	-0.360083
58	1	0	3.299775	-4.695979	0.102024
59	6	0	6.540412	-5.324548	-0.675482
60	1	0	8.043541	-3.825831	-1.040705
61	1	0	4.835610	-6.574733	-0.253511
62	1	0	7.219481	-6.158311	-0.822326
63	6	0	4.688931	1.460290	0.469502
64	6	0	6.053946	1.303679	0.755247
65	6	0	4.152552	2.755126	0.365923
66	6	0	6.858532	2.419784	0.949465
67	1	0	6.474300	0.307304	0.840287
68	6	0	4.963130	3.863634	0.564319
69	1	0	3.108758	2.905490	0.104830
70	6	0	6.317602	3.701451	0.858171
71	1	0	7.911561	2.287942	1.176611
72	1	0	4.536527	4.857697	0.476696
73	1	0	6.949264	4.570959	1.010030

E(TD-HF/TD-KS) = -2319.93927608

Compound 1: Radical-Cation.

Standard orientation:

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	-3.844499	0.307557	-0.323411
2	6	0	-2.479457	0.291155	-0.435395
3	6	0	-1.761967	-0.908539	-0.213348
4	6	0	-2.526790	-2.078722	0.019185
5	6	0	-3.893090	-2.011443	0.088762
6	1	0	-1.992378	1.202245	-0.748441
7	1	0	-2.032388	-3.027062	0.188082
8	8	0	-4.510870	-0.824664	-0.053489
9	6	0	-0.338670	-1.011889	-0.209182
10	6	0	0.608967	0.001030	-0.134891
11	6	0	2.017138	-0.223569	-0.145388
12	16	0	0.262852	1.704529	0.075109
13	6	0	2.744651	0.945671	0.000564
14	6	0	1.953456	2.092265	0.131188
15	1	0	3.827731	0.973318	0.015204
16	1	0	0.053272	-2.025763	-0.217475
17	6	0	2.490633	3.406670	0.289483
18	6	0	1.830488	4.590731	0.411477
19	1	0	3.576066	3.455413	0.313371
20	6	0	2.646003	-1.580333	-0.298536
21	1	0	2.291378	-2.234324	0.513912
22	1	0	2.312237	-2.029260	-1.246666
23	6	0	0.403578	4.703881	0.393401
24	7	0	-0.752468	4.778623	0.373699
25	6	0	2.547257	5.886858	0.570840
26	8	0	1.978262	6.946049	0.681406
27	8	0	3.874578	5.739312	0.576486
28	1	0	4.273412	6.622230	0.681552
29	8	0	4.040339	-1.441273	-0.265091
30	14	0	5.075781	-2.777760	-0.268082
31	6	0	4.593929	-3.914577	1.145637
32	1	0	3.641183	-4.418794	0.952024
33	1	0	5.351574	-4.693056	1.287251
34	1	0	4.500872	-3.361580	2.086201
35	6	0	4.922033	-3.676624	-1.906721
36	1	0	5.215853	-3.038781	-2.746372
37	1	0	5.559782	-4.567014	-1.917113
38	1	0	3.893239	-4.010813	-2.079965
39	6	0	6.786901	-2.019348	-0.026422
40	6	0	7.028263	-0.939942	-1.091032
41	6	0	6.874752	-1.383336	1.368003
42	6	0	7.858289	-3.111761	-0.158076
43	1	0	6.978043	-1.351778	-2.105863
44	1	0	6.291557	-0.132702	-1.016134
45	1	0	8.026197	-0.499737	-0.961074
46	1	0	6.749079	-2.129571	2.160917

47	1	0	7.857517	-0.912856	1.507334
48	1	0	6.110558	-0.609585	1.505232
49	1	0	8.855910	-2.680935	0.001463
50	1	0	7.722539	-3.909260	0.582259
51	1	0	7.851679	-3.568965	-1.154478
52	6	0	-4.814736	-3.119341	0.344774
53	6	0	-6.114154	-2.858032	0.801288
54	6	0	-4.401945	-4.442265	0.135009
55	6	0	-6.983202	-3.911609	1.053637
56	1	0	-6.433859	-1.835630	0.971723
57	6	0	-5.278438	-5.489161	0.387273
58	1	0	-3.410153	-4.656314	-0.249840
59	6	0	-6.567837	-5.226384	0.848553
60	1	0	-7.985422	-3.706407	1.414497
61	1	0	-4.957329	-6.511009	0.216000
62	1	0	-7.250255	-6.046885	1.045434
63	6	0	-4.704569	1.479539	-0.486655
64	6	0	-6.066674	1.319504	-0.775666
65	6	0	-4.163793	2.766589	-0.352800
66	6	0	-6.872729	2.438272	-0.941623
67	1	0	-6.484570	0.324459	-0.886028
68	6	0	-4.977618	3.878638	-0.519733
69	1	0	-3.119355	2.908062	-0.090656
70	6	0	-6.330667	3.716920	-0.816102
71	1	0	-7.925029	2.312279	-1.172835
72	1	0	-4.555388	4.871559	-0.407921
73	1	0	-6.964344	4.588511	-0.945140

E(UM062X) = -2319.82537035 A.U.

Compound 2: Ground State.

Standard orientation:

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	1.426071	-1.571114	-0.022272
2	6	0	0.711921	-0.328655	-0.025472
3	6	0	-0.681025	-0.222364	-0.039683
4	16	0	1.460585	1.238451	-0.010271
5	6	0	-1.117116	1.104975	-0.038003
6	6	0	-0.089779	2.040866	-0.023001
7	1	0	-2.164696	1.384772	-0.046952
8	1	0	0.799835	-2.459782	-0.030767
9	6	0	-0.302107	3.444352	-0.018660
10	6	0	0.596543	4.473113	-0.002133
11	1	0	-1.350089	3.734676	-0.030105
12	6	0	-1.607567	-1.410134	-0.052109
13	6	0	2.010941	4.270792	0.015196
14	7	0	3.157349	4.097411	0.029225
15	6	0	0.176179	5.888404	-0.000155
16	8	0	0.945564	6.823941	0.015778
17	8	0	-1.158206	6.038083	-0.017922
18	1	0	-1.339809	6.994649	-0.015071
19	8	0	-2.941178	-0.957986	-0.087731
20	14	0	-4.233903	-2.036519	-0.011307
21	6	0	-5.758459	-0.924935	0.036404
22	6	0	-5.779506	-0.138449	1.354840
23	6	0	-5.717196	0.061447	-1.139171
24	6	0	-7.028140	-1.782392	-0.065349
25	1	0	-5.854584	-0.804909	2.221817
26	1	0	-4.876762	0.471860	1.472905
27	1	0	-6.646140	0.536309	1.378863
28	1	0	-5.710755	-0.458075	-2.104596
29	1	0	-6.603676	0.710096	-1.118720
30	1	0	-4.829465	0.701245	-1.093266
31	1	0	-7.919832	-1.143425	-0.009092
32	1	0	-7.072952	-2.329038	-1.014330
33	1	0	-7.095518	-2.512687	0.750007
34	1	0	-1.390079	-2.041124	-0.927562
35	1	0	-1.429915	-2.026374	0.842899
36	6	0	2.770362	-1.783043	-0.009761
37	6	0	5.399790	-1.652791	0.015415
38	6	0	5.124736	-2.964653	0.009282
39	16	0	3.419890	-3.415068	-0.007517
40	16	0	4.015137	-0.551172	0.005602

41	6	0	6.093490	-4.110816	0.014888
42	1	0	5.958392	-4.734837	-0.873769
43	1	0	7.123289	-3.753503	0.027620
44	1	0	5.939199	-4.742485	0.894972
45	6	0	6.742121	-0.983506	0.029590
46	1	0	6.849881	-0.351124	0.916132
47	1	0	7.546028	-1.719844	0.036322
48	1	0	6.867229	-0.348235	-0.852590
49	6	0	-4.074756	-3.074099	1.545733
50	1	0	-3.893162	-2.444601	2.423049
51	1	0	-4.992319	-3.645309	1.724776
52	1	0	-3.252212	-3.792837	1.468171
53	6	0	-4.219131	-3.143530	-1.526925
54	1	0	-4.324564	-2.564679	-2.450109
55	1	0	-3.286261	-3.713942	-1.591336
56	1	0	-5.041352	-3.865907	-1.483483

E(RM062X) = -2580.41952258 A.U.

Compound 2: First Excited State.

Standard orientation:

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	1.389727	-1.532117	-0.023849
2	6	0	0.713844	-0.312847	-0.026747
3	6	0	-0.709313	-0.175246	-0.044129
4	16	0	1.485606	1.271611	-0.007967
5	6	0	-1.130702	1.132643	-0.042974
6	6	0	-0.079211	2.088298	-0.024602
7	1	0	-2.173851	1.425073	-0.054182
8	1	0	0.756721	-2.417562	-0.034009
9	6	0	-0.261717	3.471900	-0.019795
10	6	0	0.693637	4.495984	-0.001339
11	1	0	-1.298292	3.796699	-0.031841
12	6	0	-1.636833	-1.357858	-0.059072
13	6	0	2.094186	4.265414	0.016025
14	7	0	3.244376	4.086297	0.030200
15	6	0	0.318486	5.906457	0.001104
16	8	0	1.101064	6.841227	0.016851
17	8	0	-1.019994	6.099903	-0.016201
18	1	0	-1.158107	7.063035	-0.013037
19	8	0	-2.970451	-0.914380	-0.094808
20	14	0	-4.259388	-1.997600	-0.009832
21	6	0	-5.786752	-0.890529	0.036483
22	6	0	-5.808223	-0.100396	1.352774
23	6	0	-5.749423	0.092621	-1.141987
24	6	0	-7.053972	-1.752106	-0.061321
25	1	0	-5.880555	-0.764592	2.221733
26	1	0	-4.907065	0.512771	1.467943
27	1	0	-6.676649	0.572035	1.375794
28	1	0	-5.742406	-0.429639	-2.105943
29	1	0	-6.637903	0.738506	-1.122436
30	1	0	-4.863649	0.735255	-1.098652
31	1	0	-7.947409	-1.115607	-0.005307
32	1	0	-7.098558	-2.301166	-1.008921
33	1	0	-7.118055	-2.480605	0.755928
34	1	0	-1.412076	-1.986749	-0.934910
35	1	0	-1.452373	-1.974345	0.835142
36	6	0	2.763034	-1.796113	-0.010148
37	6	0	5.363997	-1.762249	0.015204
38	6	0	5.031403	-3.074676	0.008910
39	16	0	3.325952	-3.437114	-0.007570
40	16	0	4.037902	-0.622532	0.003689
41	6	0	5.959399	-4.253978	0.017524
42	1	0	5.772796	-4.895648	-0.848411
43	1	0	7.000325	-3.933566	-0.013737
44	1	0	5.810559	-4.853245	0.920547
45	6	0	6.739154	-1.163399	0.030250
46	1	0	6.869953	-0.524525	0.908493
47	1	0	7.503633	-1.939395	0.055673
48	1	0	6.900794	-0.549372	-0.860589
49	6	0	-4.090865	-3.026322	1.552038
50	1	0	-3.905561	-2.391446	2.424697

51	1	0	-5.007143	-3.597033	1.739104
52	1	0	-3.268345	-3.745130	1.474659
53	6	0	-4.243039	-3.111305	-1.520412
54	1	0	-4.352764	-2.536791	-2.445793
55	1	0	-3.308117	-3.678446	-1.584418
56	1	0	-5.062336	-3.836684	-1.472066

E(TD-HF/TD-KS) = -2580.33519498

Compound 2: Radical-Cation.

Standard orientation:

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	1.434745	-1.526387	-0.045285
2	6	0	0.729910	-0.317292	-0.043559
3	6	0	-0.684074	-0.233605	-0.061785
4	16	0	1.455360	1.268302	-0.018197
5	6	0	-1.133821	1.081451	-0.057469
6	6	0	-0.108057	2.025469	-0.035199
7	1	0	-2.183118	1.350989	-0.069270
8	1	0	0.826009	-2.427095	-0.066110
9	6	0	-0.347769	3.438235	-0.028236
10	6	0	0.546845	4.458820	-0.002524
11	1	0	-1.397858	3.717051	-0.046100
12	6	0	-1.600897	-1.425962	-0.077576
13	6	0	1.964450	4.261601	0.024512
14	7	0	3.108953	4.085108	0.046112
15	6	0	0.122246	5.887938	0.000304
16	8	0	0.904310	6.806908	0.024423
17	8	0	-1.205582	6.025345	-0.026776
18	1	0	-1.408498	6.978566	-0.023251
19	8	0	-2.929251	-0.980886	-0.115080
20	14	0	-4.224838	-2.064435	-0.002578
21	6	0	-5.744502	-0.949554	0.044307
22	6	0	-5.748140	-0.140277	1.349185
23	6	0	-5.717974	0.015311	-1.149562
24	6	0	-7.015169	-1.809307	-0.025969
25	1	0	-5.813200	-0.791237	2.228564
26	1	0	-4.843862	0.471732	1.445558
27	1	0	-6.613950	0.535231	1.371615
28	1	0	-5.723690	-0.521250	-2.105524
29	1	0	-6.604529	0.663482	-1.129095
30	1	0	-4.830647	0.657059	-1.126485
31	1	0	-7.905607	-1.168972	0.030278
32	1	0	-7.072258	-2.372667	-0.964376
33	1	0	-7.072234	-2.524587	0.803268
34	1	0	-1.368548	-2.050683	-0.953697
35	1	0	-1.410458	-2.036262	0.818993
36	6	0	2.817292	-1.742757	-0.023340
37	6	0	5.392648	-1.615272	0.034400
38	6	0	5.100742	-2.944162	0.011147
39	16	0	3.424706	-3.350647	-0.029410
40	16	0	4.039080	-0.536769	0.016769
41	6	0	6.125805	-4.040936	0.026903
42	1	0	5.669500	-5.021471	-0.115634
43	1	0	6.855769	-3.880385	-0.770907
44	1	0	6.655528	-4.044118	0.984268
45	6	0	6.778862	-1.042547	0.052883
46	1	0	6.775691	0.009741	0.341026
47	1	0	7.398763	-1.593300	0.764851
48	1	0	7.233761	-1.127335	-0.938925
49	6	0	-4.031850	-3.066573	1.572237
50	1	0	-3.832844	-2.418301	2.431920
51	1	0	-4.946948	-3.631185	1.781912
52	1	0	-3.213616	-3.790216	1.494695
53	6	0	-4.215204	-3.193251	-1.500129
54	1	0	-4.329022	-2.629054	-2.431245
55	1	0	-3.282193	-3.763748	-1.562331
56	1	0	-5.035554	-3.916415	-1.439292

E(UM062X) = -2580.22366026 A.U.

Compound 3: Ground State.

Standard orientation:

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	0.925769	-0.944905	0.261375
2	6	0	1.923397	0.004844	0.398190
3	16	0	1.575201	-2.548047	0.150145
4	6	0	3.209647	-0.575905	0.397114
5	6	0	3.207690	-1.951553	0.265228
6	1	0	4.122339	0.006035	0.464269
7	6	0	4.402672	-2.734928	0.229696
8	6	0	4.562616	-4.078900	0.100391
9	1	0	5.317668	-2.154693	0.322422
10	6	0	1.714312	1.495480	0.480880
11	6	0	3.464826	-4.987564	-0.028644
12	7	0	2.572131	-5.719066	-0.131938
13	6	0	5.898194	-4.719873	0.082678
14	8	0	6.069834	-5.912338	-0.027903
15	8	0	6.900302	-3.837555	0.201818
16	1	0	7.733946	-4.340659	0.179960
17	8	0	2.873811	2.144414	0.006013
18	14	0	2.966249	3.825584	-0.065986
19	6	0	4.766248	4.172780	-0.515505
20	6	0	5.672341	3.805607	0.668150
21	6	0	5.165513	3.332655	-1.736784
22	6	0	4.938127	5.662531	-0.845460
23	1	0	5.455321	4.420481	1.549195
24	1	0	5.556256	2.753125	0.952043
25	1	0	6.726101	3.966419	0.402382
26	1	0	4.543327	3.564287	-2.609394
27	1	0	6.209387	3.537986	-2.010996
28	1	0	5.073560	2.260989	-1.531113
29	1	0	5.990533	5.879400	-1.073444
30	1	0	4.343618	5.954889	-1.718614
31	1	0	4.644916	6.304434	-0.005835
32	1	0	0.829402	1.775670	-0.109146
33	1	0	1.519695	1.793299	1.522219
34	6	0	-0.524459	-0.747416	0.198444
35	6	0	-1.331030	-1.547293	-0.626088
36	6	0	-1.153454	0.241112	0.969903
37	6	0	-2.701070	-1.358456	-0.693243
38	1	0	-0.873336	-2.307060	-1.253688
39	6	0	-2.525385	0.430776	0.915925
40	1	0	-0.568973	0.841662	1.659998
41	6	0	-3.323552	-0.362656	0.077423
42	1	0	-3.298253	-1.974997	-1.356369
43	1	0	-2.989120	1.188983	1.537712
44	7	0	-4.711628	-0.167362	0.012081
45	6	0	-5.583155	-1.259161	-0.248861
46	6	0	-6.639823	-1.109742	-1.152527
47	6	0	-5.402127	-2.482572	0.404461
48	6	0	-7.505371	-2.171734	-1.393928
49	1	0	-6.777079	-0.158945	-1.658479
50	6	0	-6.261190	-3.545121	0.142966
51	1	0	-4.587226	-2.593881	1.113395
52	6	0	-7.318471	-3.395157	-0.752870
53	1	0	-8.322674	-2.043741	-2.096791
54	1	0	-6.109985	-4.490526	0.654540
55	1	0	-7.991157	-4.223720	-0.948538
56	6	0	-5.270770	1.124691	0.202414
57	6	0	-4.691321	2.245276	-0.402367
58	6	0	-6.418959	1.281374	0.985417
59	6	0	-5.247659	3.505774	-0.210886
60	1	0	-3.806256	2.121520	-1.019036
61	6	0	-6.979507	2.543079	1.155948
62	1	0	-6.866821	0.410293	1.453875
63	6	0	-6.395403	3.661866	0.564535
64	1	0	-4.788057	4.367932	-0.684102

65	1	0	-7.871511	2.651557	1.765030
66	1	0	-6.831167	4.645531	0.705215
67	6	0	2.512934	4.538197	1.611434
68	1	0	1.442082	4.434249	1.816014
69	1	0	3.061919	4.037416	2.415806
70	1	0	2.753185	5.606308	1.654541
71	6	0	1.783804	4.463395	-1.376720
72	1	0	2.059032	4.107788	-2.374841
73	1	0	0.757045	4.139781	-1.174785
74	1	0	1.784185	5.558586	-1.394529

E(RM062X) = -2300.17371142 A.U.

Compound 3: First Excited State.

Standard orientation:

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	-0.797953	0.864027	-0.000994
2	6	0	-1.918820	-0.020630	0.098354
3	16	0	-1.360879	2.537309	-0.144039
4	6	0	-3.128739	0.635088	0.064069
5	6	0	-3.042983	2.047078	-0.062855
6	1	0	-4.080942	0.121732	0.116251
7	6	0	-4.145173	2.900988	-0.117365
8	6	0	-4.188293	4.296482	-0.236267
9	1	0	-5.110220	2.405643	-0.058968
10	6	0	-1.848652	-1.524089	0.193785
11	6	0	-3.030560	5.113552	-0.314177
12	7	0	-2.088265	5.794062	-0.377223
13	6	0	-5.446841	5.031769	-0.284527
14	8	0	-5.551543	6.242676	-0.384284
15	8	0	-6.534245	4.230222	-0.206489
16	1	0	-7.309515	4.816652	-0.246296
17	8	0	-3.143833	-2.062787	0.096727
18	14	0	-3.413082	-3.723320	0.206764
19	6	0	-5.293079	-3.882601	0.214813
20	6	0	-5.852874	-3.279557	1.511071
21	6	0	-5.880859	-3.129659	-0.987110
22	6	0	-5.688276	-5.363771	0.128933
23	1	0	-5.496721	-3.820390	2.395463
24	1	0	-5.568368	-2.226580	1.619425
25	1	0	-6.950167	-3.332191	1.509527
26	1	0	-5.512292	-3.532514	-1.937842
27	1	0	-6.975565	-3.222072	-0.991040
28	1	0	-5.631967	-2.063665	-0.952060
29	1	0	-6.781507	-5.464726	0.163174
30	1	0	-5.343467	-5.821624	-0.805366
31	1	0	-5.278370	-5.945815	0.963156
32	1	0	-1.205189	-1.914422	-0.609238
33	1	0	-1.390660	-1.812895	1.152717
34	6	0	0.592186	0.610750	-0.007353
35	6	0	1.525493	1.664653	-0.289491
36	6	0	1.166000	-0.671864	0.274549
37	6	0	2.877933	1.463402	-0.296449
38	1	0	1.155990	2.652754	-0.543509
39	6	0	2.518464	-0.881158	0.261615
40	1	0	0.534982	-1.503844	0.553282
41	6	0	3.421309	0.179731	-0.023890
42	1	0	3.543567	2.282104	-0.545905
43	1	0	2.912109	-1.859046	0.515353
44	7	0	4.782220	-0.029224	-0.025009
45	6	0	5.682335	1.031069	0.275173
46	6	0	6.829582	1.208907	-0.505471
47	6	0	5.426351	1.879083	1.358488
48	6	0	7.711121	2.239221	-0.202339
49	1	0	7.015055	0.546949	-1.345372
50	6	0	6.312648	2.909853	1.647086
51	1	0	4.545162	1.716324	1.971196
52	6	0	7.455830	3.093965	0.870167
53	1	0	8.596860	2.380223	-0.812914
54	1	0	6.114524	3.562835	2.490639
55	1	0	8.146587	3.898016	1.101013
56	6	0	5.326717	-1.308948	-0.321439

57	6	0	4.821563	-2.054911	-1.392437
58	6	0	6.380541	-1.808630	0.451927
59	6	0	5.364572	-3.302647	-1.675698
60	1	0	4.020874	-1.646268	-2.001058
61	6	0	6.918823	-3.054106	0.153922
62	1	0	6.760933	-1.222695	1.282500
63	6	0	6.412828	-3.806357	-0.906410
64	1	0	4.975068	-3.876414	-2.510144
65	1	0	7.731533	-3.442304	0.758929
66	1	0	6.836046	-4.779010	-1.133890
67	6	0	-2.646263	-4.363497	1.796825
68	1	0	-1.552347	-4.366351	1.743312
69	1	0	-2.942160	-3.749769	2.654034
70	1	0	-2.967167	-5.392552	1.992612
71	6	0	-2.640022	-4.582281	-1.272187
72	1	0	-3.087489	-4.245748	-2.212861
73	1	0	-1.562819	-4.390151	-1.323350
74	1	0	-2.775341	-5.666888	-1.200771

E(TD-HF/TD-KS) = -2300.07905494

Compound 3: Radical-Cation.

Standard orientation:

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	0.868945	-0.932596	0.206689
2	6	0	1.883486	0.004093	0.339982
3	16	0	1.491000	-2.546760	0.070216
4	6	0	3.157925	-0.600968	0.321433
5	6	0	3.124909	-1.975192	0.186217
6	1	0	4.080716	-0.036223	0.387404
7	6	0	4.310364	-2.786667	0.158314
8	6	0	4.433632	-4.130680	0.036854
9	1	0	5.237018	-2.225646	0.249667
10	6	0	1.720088	1.501336	0.433476
11	6	0	3.311534	-5.011274	-0.091420
12	7	0	2.396370	-5.713616	-0.194554
13	6	0	5.756507	-4.810063	0.026252
14	8	0	5.888250	-6.006182	-0.080019
15	8	0	6.776259	-3.953044	0.146386
16	1	0	7.600981	-4.471576	0.129962
17	8	0	2.915642	2.111005	0.012969
18	14	0	3.094685	3.790511	0.006290
19	6	0	4.921345	4.056197	-0.381717
20	6	0	5.772705	3.603047	0.812886
21	6	0	5.312250	3.238182	-1.620706
22	6	0	5.177685	5.545578	-0.654444
23	1	0	5.561942	4.198618	1.708510
24	1	0	5.595704	2.548971	1.056857
25	1	0	6.840266	3.718108	0.581314
26	1	0	4.727550	3.530302	-2.501008
27	1	0	6.372237	3.399817	-1.859436
28	1	0	5.162306	2.165988	-1.455423
29	1	0	6.245920	5.714941	-0.845289
30	1	0	4.625020	5.897244	-1.533171
31	1	0	4.892875	6.173347	0.198612
32	1	0	0.869257	1.821851	-0.185594
33	1	0	1.496085	1.785265	1.473306
34	6	0	-0.570894	-0.715926	0.153241
35	6	0	-1.401337	-1.574572	-0.598912
36	6	0	-1.169039	0.346671	0.861538
37	6	0	-2.763342	-1.381731	-0.652563
38	1	0	-0.961194	-2.377565	-1.181112
39	6	0	-2.530026	0.559323	0.809395
40	1	0	-0.567102	0.976595	1.505732
41	6	0	-3.347629	-0.304404	0.049626
42	1	0	-3.381973	-2.021463	-1.271768
43	1	0	-2.980971	1.352267	1.395278
44	7	0	-4.721654	-0.099541	-0.001629
45	6	0	-5.605476	-1.194909	-0.140361

46	6	0	-6.726812	-1.074052	-0.975192
47	6	0	-5.358161	-2.381300	0.567066
48	6	0	-7.589998	-2.151004	-1.106213
49	1	0	-6.890739	-0.156348	-1.530014
50	6	0	-6.235563	-3.446337	0.428235
51	1	0	-4.507628	-2.444467	1.237713
52	6	0	-7.348182	-3.335537	-0.407300
53	1	0	-8.449320	-2.070537	-1.762508
54	1	0	-6.058235	-4.360535	0.983563
55	1	0	-8.030011	-4.172599	-0.512073
56	6	0	-5.255365	1.207153	0.085611
57	6	0	-4.621903	2.262899	-0.587686
58	6	0	-6.420900	1.426765	0.835329
59	6	0	-5.158686	3.538752	-0.500574
60	1	0	-3.743272	2.068189	-1.193712
61	6	0	-6.939989	2.709841	0.916001
62	1	0	-6.885154	0.602098	1.365575
63	6	0	-6.313290	3.765409	0.250348
64	1	0	-4.682847	4.356101	-1.030777
65	1	0	-7.830916	2.889827	1.507188
66	1	0	-6.727382	4.765872	0.314657
67	6	0	2.627353	4.457244	1.697814
68	1	0	1.547365	4.399340	1.870011
69	1	0	3.129271	3.900184	2.495873
70	1	0	2.917568	5.509769	1.789301
71	6	0	1.980352	4.527698	-1.310992
72	1	0	2.263411	4.189975	-2.313080
73	1	0	0.933368	4.251902	-1.144779
74	1	0	2.037151	5.621402	-1.293060

E(UM062X) = -2299.97188393 A.U.

Compound 15: Ground State.

Standard orientation:

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	-4.674455	-0.199336	-0.213047
2	6	0	-3.371845	0.145756	-0.103460
3	6	0	-2.358469	-0.833897	0.216520
4	6	0	-2.860025	-2.189491	0.290046
5	6	0	-4.172128	-2.470769	0.144806
6	1	0	-3.115398	1.173603	-0.311376
7	1	0	-2.169309	-2.996929	0.503332
8	8	0	-5.089269	-1.483297	-0.067510
9	6	0	-1.025346	-0.584664	0.436844
10	6	0	-0.327588	0.661524	0.508944
11	6	0	1.062806	0.779645	0.639258
12	16	0	-1.076809	2.230274	0.519961
13	6	0	1.489688	2.106787	0.730148
14	6	0	0.460695	3.041331	0.678921
15	1	0	2.532553	2.388729	0.827819
16	1	0	-0.399430	-1.457907	0.604113
17	6	0	0.653298	4.443379	0.738410
18	6	0	-0.261379	5.460233	0.685483
19	1	0	1.691989	4.748655	0.841677
20	6	0	1.990786	-0.405955	0.662887
21	1	0	1.726059	-1.082151	1.488292
22	1	0	1.873825	-0.982036	-0.268853
23	6	0	-1.666299	5.237443	0.550598
24	7	0	-2.805033	5.045577	0.441233
25	6	0	0.132001	6.879588	0.762047
26	8	0	-0.649314	7.804939	0.716882
27	8	0	1.458718	7.049753	0.888374
28	1	0	1.621270	8.008733	0.930596
29	8	0	3.322510	0.038854	0.826444
30	14	0	4.603130	-0.785269	0.124669
31	6	0	6.147076	-0.007454	0.881498
32	6	0	6.110975	1.505497	0.617994
33	6	0	6.175894	-0.259069	2.395497
34	6	0	7.406070	-0.608839	0.241765

35	1	0	6.077334	1.727825	-0.455012
36	1	0	5.239447	1.972140	1.090128
37	1	0	7.011901	1.979426	1.030421
38	1	0	6.255398	-1.327335	2.625638
39	1	0	7.046039	0.243128	2.839418
40	1	0	5.276997	0.131294	2.886086
41	1	0	8.302781	-0.192962	0.720279
42	1	0	7.442686	-1.699095	0.359740
43	1	0	7.465491	-0.377760	-0.827593
44	6	0	-4.796211	-3.802431	0.222542
45	6	0	-6.163276	-3.921379	0.502257
46	6	0	-4.034809	-4.960255	0.018535
47	6	0	-6.752130	-5.178550	0.591471
48	1	0	-6.758821	-3.028571	0.659111
49	6	0	-4.628029	-6.213471	0.110494
50	1	0	-2.982389	-4.883190	-0.235251
51	6	0	-5.987623	-6.327061	0.398774
52	1	0	-7.811132	-5.259757	0.814217
53	1	0	-4.029212	-7.103311	-0.054160
54	1	0	-6.449338	-7.306844	0.466883
55	6	0	-5.781096	0.723115	-0.518910
56	6	0	-6.988919	0.227869	-1.025500
57	6	0	-5.636669	2.101120	-0.308495
58	6	0	-8.028056	1.100584	-1.331463
59	1	0	-7.105546	-0.838174	-1.188407
60	6	0	-6.676998	2.967894	-0.619394
61	1	0	-4.721110	2.500849	0.117334
62	6	0	-7.874809	2.471023	-1.132543
63	1	0	-8.958536	0.708322	-1.728951
64	1	0	-6.554051	4.032699	-0.450010
65	1	0	-8.687261	3.149920	-1.370980
66	6	0	4.523765	-0.496270	-1.729396
67	6	0	5.398512	-1.113805	-2.639369
68	6	0	3.582547	0.411280	-2.240370
69	6	0	5.327710	-0.846002	-4.004569
70	1	0	6.154020	-1.811751	-2.284214
71	6	0	3.505870	0.682160	-3.605387
72	1	0	2.901350	0.920070	-1.560483
73	6	0	4.377670	0.051138	-4.489821
74	1	0	6.013993	-1.335514	-4.688846
75	1	0	2.767820	1.386282	-3.977519
76	1	0	4.320221	0.260183	-5.553700
77	6	0	4.397509	-2.601592	0.559971
78	6	0	4.028100	-2.934287	1.875356
79	6	0	4.568574	-3.649074	-0.356049
80	6	0	3.852055	-4.259660	2.263953
81	1	0	3.868084	-2.143541	2.606399
82	6	0	4.398413	-4.978629	0.027632
83	1	0	4.823782	-3.430547	-1.389568
84	6	0	4.042588	-5.285314	1.338818
85	1	0	3.564764	-4.493006	3.284704
86	1	0	4.537296	-5.773661	-0.698711
87	1	0	3.906982	-6.320140	1.638361

E(RM062X) = -2703.34040424 A.U.

Compound 15: First Excited State.

Standard orientation:

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	-4.657042	-0.232123	-0.342067
2	6	0	-3.336126	0.095355	-0.239489
3	6	0	-2.369149	-0.875053	0.141789
4	6	0	-2.856000	-2.203534	0.313153
5	6	0	-4.180371	-2.494992	0.167142
6	1	0	-3.047749	1.101264	-0.508830
7	1	0	-2.169237	-2.993493	0.593252
8	8	0	-5.071864	-1.507508	-0.125210
9	6	0	-0.983697	-0.610758	0.344922
10	6	0	-0.330364	0.603043	0.476173

11	6	0	1.088030	0.762265	0.632787
12	16	0	-1.124174	2.180280	0.546638
13	6	0	1.478185	2.069640	0.782790
14	6	0	0.410140	3.008574	0.758307
15	1	0	2.510386	2.374497	0.911219
16	1	0	-0.356751	-1.495247	0.440763
17	6	0	0.561640	4.391669	0.881409
18	6	0	-0.408219	5.395553	0.846092
19	1	0	1.584238	4.731384	1.021575
20	6	0	2.030711	-0.406727	0.614450
21	1	0	1.754762	-1.124475	1.400927
22	1	0	1.927586	-0.936529	-0.346550
23	6	0	-1.793723	5.145407	0.649971
24	7	0	-2.928795	4.947328	0.487690
25	6	0	-0.068487	6.809508	0.999885
26	8	0	-0.868638	7.727191	0.970570
27	8	0	1.253142	7.019530	1.183610
28	1	0	1.370686	7.981422	1.272807
29	8	0	3.353762	0.038014	0.818894
30	14	0	4.649253	-0.752233	0.101830
31	6	0	6.180161	0.008609	0.899886
32	6	0	6.139076	1.529086	0.684359
33	6	0	6.197049	-0.291719	2.405128
34	6	0	7.447422	-0.566319	0.251827
35	1	0	6.116806	1.785309	-0.381323
36	1	0	5.259946	1.976762	1.160695
37	1	0	7.032907	1.993614	1.122087
38	1	0	6.280119	-1.366472	2.601377
39	1	0	7.060881	0.200005	2.872408
40	1	0	5.292189	0.078361	2.900407
41	1	0	8.338037	-0.160276	0.749698
42	1	0	7.489342	-1.659417	0.336913
43	1	0	7.513617	-0.302555	-0.809578
44	6	0	-4.811481	-3.804426	0.326235
45	6	0	-6.203652	-3.900126	0.478362
46	6	0	-4.040847	-4.978418	0.328790
47	6	0	-6.805799	-5.142024	0.641695
48	1	0	-6.807016	-2.999276	0.482420
49	6	0	-4.649936	-6.214620	0.493030
50	1	0	-2.967186	-4.930714	0.179907
51	6	0	-6.033675	-6.302092	0.651503
52	1	0	-7.882261	-5.202966	0.764495
53	1	0	-4.043798	-7.114556	0.488369
54	1	0	-6.506580	-7.270625	0.777665
55	6	0	-5.738198	0.682697	-0.698825
56	6	0	-6.976677	0.181201	-1.128693
57	6	0	-5.552174	2.072872	-0.610860
58	6	0	-7.999477	1.054468	-1.478336
59	1	0	-7.125878	-0.890625	-1.201596
60	6	0	-6.578784	2.937235	-0.964065
61	1	0	-4.617385	2.485171	-0.242164
62	6	0	-7.804739	2.432763	-1.400200
63	1	0	-8.951175	0.657397	-1.816297
64	1	0	-6.423808	4.008526	-0.887139
65	1	0	-8.606144	3.111944	-1.672422
66	6	0	4.585305	-0.396098	-1.740535
67	6	0	5.462378	-0.985725	-2.666476
68	6	0	3.656117	0.539226	-2.222904
69	6	0	5.404680	-0.665080	-4.020875
70	1	0	6.209904	-1.702344	-2.332165
71	6	0	3.592875	0.862984	-3.577057
72	1	0	2.975298	1.028819	-1.528455
73	6	0	4.466065	0.258327	-4.478414
74	1	0	6.092389	-1.133734	-4.718191
75	1	0	2.864495	1.588054	-3.927380
76	1	0	4.419023	0.508668	-5.533867
77	6	0	4.445971	-2.582722	0.474078
78	6	0	4.065244	-2.957813	1.774801
79	6	0	4.626695	-3.600131	-0.473441
80	6	0	3.888083	-4.295123	2.119402
81	1	0	3.897004	-2.191056	2.529220
82	6	0	4.455305	-4.941462	-0.133819
83	1	0	4.890514	-3.348682	-1.497227
84	6	0	4.088529	-5.290406	1.163739
85	1	0	3.592239	-4.561275	3.129629
86	1	0	4.601776	-5.712692	-0.883943
87	1	0	3.952161	-6.334482	1.428821

$$E(\text{TD-HF/TD-KS}) = -2703.26408981$$

Compound 15: Radical-Cation.

Standard orientation:

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	-4.680240	-0.209540	-0.353717
2	6	0	-3.356279	0.131598	-0.266250
3	6	0	-2.393995	-0.841930	0.094933
4	6	0	-2.858511	-2.169215	0.270433
5	6	0	-4.191086	-2.459835	0.145926
6	1	0	-3.079784	1.139695	-0.536399
7	1	0	-2.166602	-2.955180	0.545356
8	8	0	-5.064432	-1.476263	-0.136246
9	6	0	-1.003378	-0.581370	0.282807
10	6	0	-0.347503	0.635671	0.414654
11	6	0	1.065155	0.761705	0.568813
12	16	0	-1.114833	2.206936	0.494279
13	6	0	1.468328	2.076800	0.728599
14	6	0	0.414589	2.998283	0.709902
15	1	0	2.503416	2.370763	0.857006
16	1	0	-0.376732	-1.464816	0.371671
17	6	0	0.601681	4.407760	0.849526
18	6	0	-0.329372	5.399743	0.824722
19	1	0	1.633511	4.718008	0.990449
20	6	0	2.009429	-0.406821	0.543133
21	1	0	1.728047	-1.131070	1.321408
22	1	0	1.912010	-0.922993	-0.425317
23	6	0	-1.728913	5.162203	0.640627
24	7	0	-2.858301	4.952658	0.489339
25	6	0	0.035408	6.835253	0.983119
26	8	0	-0.776594	7.728478	0.959325
27	8	0	1.348318	7.013131	1.151929
28	1	0	1.510655	7.969327	1.247656
29	8	0	3.321413	0.051453	0.758341
30	14	0	4.640879	-0.770032	0.108349
31	6	0	6.144875	0.016207	0.929457
32	6	0	6.109656	1.529775	0.668474
33	6	0	6.120732	-0.240640	2.442691
34	6	0	7.429424	-0.576364	0.332510
35	1	0	6.113515	1.755456	-0.404270
36	1	0	5.220252	1.991567	1.111290
37	1	0	6.992969	2.006241	1.114181
38	1	0	6.200908	-1.308888	2.672186
39	1	0	6.970652	0.266671	2.918335
40	1	0	5.202435	0.141604	2.903065
41	1	0	8.305656	-0.153573	0.841447
42	1	0	7.470379	-1.666147	0.452334
43	1	0	7.523621	-0.344874	-0.734251
44	6	0	-4.824020	-3.767671	0.319492
45	6	0	-6.200179	-3.851709	0.574342
46	6	0	-4.057719	-4.938224	0.232225
47	6	0	-6.796167	-5.093951	0.749662
48	1	0	-6.793812	-2.947215	0.649908
49	6	0	-4.662582	-6.175899	0.405830
50	1	0	-2.998311	-4.888375	0.003498
51	6	0	-6.030196	-6.255910	0.666723
52	1	0	-7.859523	-5.155082	0.954411
53	1	0	-4.067471	-7.079387	0.329848
54	1	0	-6.499370	-7.224829	0.802568
55	6	0	-5.780597	0.692785	-0.692381
56	6	0	-6.994191	0.174706	-1.165714
57	6	0	-5.621063	2.078172	-0.544382
58	6	0	-8.030167	1.037986	-1.498235
59	1	0	-7.115525	-0.896343	-1.286957
60	6	0	-6.662920	2.933010	-0.877893
61	1	0	-4.700595	2.492582	-0.144613
62	6	0	-7.866293	2.415493	-1.356566
63	1	0	-8.965503	0.635727	-1.872062

64	1	0	-6.537340	4.003340	-0.755179
65	1	0	-8.678521	3.086590	-1.615982
66	6	0	4.627967	-0.469252	-1.743989
67	6	0	5.523704	-1.097553	-2.625706
68	6	0	3.725196	0.461367	-2.281859
69	6	0	5.508342	-0.818287	-3.990270
70	1	0	6.252620	-1.811912	-2.248332
71	6	0	3.703823	0.743819	-3.646368
72	1	0	3.032678	0.982276	-1.622940
73	6	0	4.594604	0.101059	-4.502935
74	1	0	6.209894	-1.315901	-4.652694
75	1	0	2.995385	1.466486	-4.039728
76	1	0	4.580899	0.319027	-5.566408
77	6	0	4.407075	-2.583802	0.533088
78	6	0	4.004093	-2.918006	1.838396
79	6	0	4.585209	-3.628445	-0.384959
80	6	0	3.802553	-4.242942	2.215614
81	1	0	3.838775	-2.129721	2.571015
82	6	0	4.389954	-4.957497	-0.011943
83	1	0	4.866551	-3.408248	-1.411190
84	6	0	4.001046	-5.266070	1.289319
85	1	0	3.490521	-4.477920	3.228629
86	1	0	4.535793	-5.750580	-0.738923
87	1	0	3.846632	-6.300609	1.580389

E(UM062X) = -2703.14956783 A.U.

Compound 16: Ground State.

Standard orientation:

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	5.284291	0.427806	0.107187
2	6	0	3.936706	0.325973	0.071430
3	6	0	3.284547	-0.934926	-0.202134
4	6	0	4.198478	-2.051184	-0.320348
5	6	0	5.535988	-1.885508	-0.248561
6	1	0	3.369544	1.215398	0.301262
7	1	0	3.800042	-3.041684	-0.506181
8	8	0	6.088585	-0.650496	-0.073035
9	6	0	1.933977	-1.139129	-0.343967
10	6	0	0.861497	-0.194476	-0.352412
11	6	0	-0.487941	-0.544564	-0.436346
12	16	0	1.034191	1.532919	-0.325271
13	6	0	-1.357867	0.547280	-0.467461
14	6	0	-0.696152	1.779125	-0.413192
15	1	0	1.608612	-2.166310	-0.490497
16	6	0	-1.332998	3.041165	-0.422101
17	6	0	-0.799624	4.302648	-0.365171
18	1	0	-2.418427	3.008366	-0.484981
19	6	0	-2.857477	0.405040	-0.527060
20	6	0	0.603178	4.557978	-0.280325
21	7	0	1.743379	4.760865	-0.212909
22	6	0	-1.641297	5.512591	-0.390584
23	8	0	-1.208725	6.644004	-0.344361
24	8	0	-2.954700	5.237840	-0.471472
25	1	0	-3.423962	6.090722	-0.484249
26	8	0	-3.197523	-0.953141	-0.693495
27	14	0	-4.671345	-1.554852	-0.157432
28	6	0	-4.625847	-3.383058	-0.628326
29	6	0	-4.226830	-3.573025	-2.099748
30	6	0	-3.561371	-4.066797	0.247722
31	6	0	-5.988726	-4.033948	-0.351913
32	1	0	-4.947252	-3.132248	-2.795828
33	1	0	-3.248360	-3.126298	-2.303426
34	1	0	-4.164082	-4.645333	-2.328960
35	1	0	-3.816457	-4.015464	1.311401
36	1	0	-3.480604	-5.126174	-0.030088
37	1	0	-2.573935	-3.609423	0.113138
38	1	0	-5.937355	-5.111606	-0.556449
39	1	0	-6.285205	-3.910435	0.697088

40	1	0	-6.783419	-3.612933	-0.976993
41	6	0	6.557754	-2.938647	-0.375447
42	6	0	7.870535	-2.602705	-0.729277
43	6	0	6.232049	-4.281260	-0.144950
44	6	0	8.834521	-3.596502	-0.865005
45	1	0	8.129575	-1.564580	-0.907019
46	6	0	7.198724	-5.269928	-0.283635
47	1	0	5.228452	-4.553714	0.165767
48	6	0	8.502181	-4.931411	-0.645599
49	1	0	9.847510	-3.325940	-1.145006
50	1	0	6.936025	-6.306282	-0.097921
51	1	0	9.256117	-5.705033	-0.750154
52	6	0	6.043252	1.663877	0.363086
53	6	0	7.371411	1.594414	0.801123
54	6	0	5.445911	2.917417	0.174282
55	6	0	8.082954	2.761038	1.061432
56	1	0	7.838419	0.626354	0.947152
57	6	0	6.160661	4.078915	0.439588
58	1	0	4.429073	2.992479	-0.199322
59	6	0	7.480165	4.004640	0.884825
60	1	0	9.110001	2.697231	1.406048
61	1	0	5.687653	5.043752	0.288005
62	1	0	8.037448	4.913493	1.087954
63	1	0	-0.823173	-1.573710	-0.478926
64	1	0	-3.264322	0.995913	-1.358582
65	1	0	-3.288474	0.809298	0.402375
66	6	0	-4.694040	-1.333841	1.707077
67	6	0	-5.889196	-1.320212	2.441401
68	6	0	-3.484296	-1.237125	2.412706
69	6	0	-5.879023	-1.221350	3.831484
70	1	0	-6.843738	-1.377682	1.921085
71	6	0	-3.467325	-1.130648	3.801678
72	1	0	-2.542052	-1.242509	1.865529
73	6	0	-4.666621	-1.125342	4.512276
74	1	0	-6.814558	-1.213688	4.382542
75	1	0	-2.521709	-1.051188	4.329448
76	1	0	-4.656397	-1.043363	5.594956
77	6	0	-6.084910	-0.547661	-0.883574
78	6	0	-6.794996	-0.921073	-2.035503
79	6	0	-6.423407	0.676303	-0.280529
80	6	0	-7.799288	-0.111915	-2.562654
81	1	0	-6.571717	-1.859432	-2.534653
82	6	0	-7.421226	1.494166	-0.805730
83	1	0	-5.907651	0.993770	0.623830
84	6	0	-8.112510	1.098950	-1.948997
85	1	0	-8.336845	-0.427034	-3.451730
86	1	0	-7.662235	2.435120	-0.320642
87	1	0	-8.894388	1.731109	-2.358790

E(RM062X) = -2703.33896712 A.U.

Compound 16: First Excited State.

Standard orientation:

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	5.282313	0.409001	0.220923
2	6	0	3.925573	0.267248	0.196362
3	6	0	3.325920	-0.984220	-0.117837
4	6	0	4.225869	-2.072918	-0.310418
5	6	0	5.576127	-1.894163	-0.244172
6	1	0	3.328137	1.123601	0.474081
7	1	0	3.833405	-3.055525	-0.543023
8	8	0	6.093138	-0.655839	-0.011779
9	6	0	1.926762	-1.210394	-0.240231
10	6	0	0.887654	-0.295902	-0.318258
11	6	0	-0.492240	-0.643953	-0.408367
12	16	0	1.066908	1.458407	-0.393468
13	6	0	-1.343768	0.425810	-0.517443
14	6	0	-0.670682	1.689505	-0.519506
15	1	0	1.623122	-2.253036	-0.312955

16	6	0	-1.298749	2.933307	-0.605734
17	6	0	-0.736228	4.213289	-0.596141
18	1	0	-2.381839	2.910956	-0.691448
19	6	0	-2.838417	0.312890	-0.620010
20	6	0	0.657046	4.464412	-0.473696
21	7	0	1.795878	4.680910	-0.371590
22	6	0	-1.555361	5.419471	-0.703046
23	8	0	-1.125275	6.559009	-0.699051
24	8	0	-2.876042	5.155137	-0.810489
25	1	0	-3.325986	6.015532	-0.875322
26	8	0	-3.209949	-1.047614	-0.626451
27	14	0	-4.701216	-1.553095	-0.044465
28	6	0	-4.666554	-3.430141	-0.255113
29	6	0	-4.232974	-3.826373	-1.674810
30	6	0	-3.632328	-3.994466	0.734892
31	6	0	-6.043048	-4.026129	0.072991
32	1	0	-4.929409	-3.479308	-2.444450
33	1	0	-3.243720	-3.421732	-1.911934
34	1	0	-4.178764	-4.920694	-1.751749
35	1	0	-3.913056	-3.793773	1.774143
36	1	0	-3.557377	-5.082825	0.608775
37	1	0	-2.636465	-3.568360	0.564316
38	1	0	-5.997719	-5.122266	0.024646
39	1	0	-6.365501	-3.753219	1.085465
40	1	0	-6.816709	-3.692905	-0.626972
41	6	0	6.605242	-2.915841	-0.433978
42	6	0	7.938408	-2.538251	-0.658432
43	6	0	6.278522	-4.281063	-0.394398
44	6	0	8.917022	-3.506495	-0.849657
45	1	0	8.200238	-1.486760	-0.696004
46	6	0	7.261507	-5.241571	-0.587299
47	1	0	5.260652	-4.596783	-0.190902
48	6	0	8.583832	-4.859211	-0.816825
49	1	0	9.943332	-3.202310	-1.027730
50	1	0	6.996827	-6.293173	-0.549765
51	1	0	9.350140	-5.613167	-0.964954
52	6	0	6.007490	1.644913	0.506086
53	6	0	7.364705	1.602561	0.860868
54	6	0	5.356682	2.887665	0.424610
55	6	0	8.049067	2.778488	1.143621
56	1	0	7.872479	0.646621	0.928949
57	6	0	6.047643	4.056668	0.711018
58	1	0	4.317918	2.950978	0.113875
59	6	0	7.394565	4.007369	1.072671
60	1	0	9.096516	2.734745	1.423806
61	1	0	5.534219	5.010099	0.640477
62	1	0	7.932238	4.923887	1.293040
63	1	0	-0.827063	-1.675041	-0.392823
64	1	0	-3.193398	0.803173	-1.538032
65	1	0	-3.296921	0.847587	0.227933
66	6	0	-4.761408	-1.075695	1.770438
67	6	0	-5.970803	-0.951933	2.470239
68	6	0	-3.564982	-0.892832	2.481667
69	6	0	-5.987182	-0.664263	3.833878
70	1	0	-6.915260	-1.071791	1.942167
71	6	0	-3.574487	-0.597657	3.843082
72	1	0	-2.612511	-0.979127	1.959440
73	6	0	-4.787665	-0.486154	4.520801
74	1	0	-6.933378	-0.573667	4.358782
75	1	0	-2.638821	-0.454222	4.375067
76	1	0	-4.798145	-0.257386	5.582217
77	6	0	-6.088119	-0.648165	-0.937261
78	6	0	-6.773627	-1.175892	-2.042640
79	6	0	-6.427628	0.651417	-0.522467
80	6	0	-7.755326	-0.442382	-2.705431
81	1	0	-6.547837	-2.177085	-2.397698
82	6	0	-7.403033	1.393829	-1.184345
83	1	0	-5.930134	1.090269	0.340451
84	6	0	-8.069949	0.845665	-2.277919
85	1	0	-8.273824	-0.876429	-3.554858
86	1	0	-7.645395	2.396034	-0.844226
87	1	0	-8.833979	1.419320	-2.793779

$$E(\text{TD-HF/TD-KS}) = -2703.26276188$$

Compound 16: Radical-Cation.

Standard orientation:

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	5.329078	0.405998	0.207354
2	6	0	3.963247	0.300673	0.207992
3	6	0	3.346578	-0.945772	-0.060415
4	6	0	4.204065	-2.058860	-0.246372
5	6	0	5.563929	-1.901975	-0.208743
6	1	0	3.393785	1.179561	0.470161
7	1	0	3.787109	-3.035584	-0.455845
8	8	0	6.087880	-0.679523	-0.007771
9	6	0	1.939181	-1.158160	-0.148537
10	6	0	0.906447	-0.229033	-0.211102
11	6	0	-0.461378	-0.591893	-0.279425
12	16	0	1.076423	1.507687	-0.281820
13	6	0	-1.322534	0.486086	-0.370692
14	6	0	-0.644348	1.723004	-0.382485
15	1	0	1.621376	-2.195882	-0.206227
16	6	0	-1.297518	2.988316	-0.464733
17	6	0	-0.753355	4.237258	-0.469952
18	1	0	-2.381708	2.951015	-0.530146
19	6	0	-2.821293	0.346697	-0.439948
20	6	0	0.652889	4.488979	-0.388842
21	7	0	1.793785	4.679953	-0.322496
22	6	0	-1.591715	5.465017	-0.559913
23	8	0	-1.129781	6.580542	-0.566033
24	8	0	-2.896817	5.188889	-0.632206
25	1	0	-3.378268	6.034116	-0.691435
26	8	0	-3.148745	-1.019876	-0.462697
27	14	0	-4.693872	-1.556931	-0.055825
28	6	0	-4.600488	-3.427964	-0.285130
29	6	0	-4.021578	-3.797269	-1.659538
30	6	0	-3.660130	-3.988296	0.796544
31	6	0	-5.992991	-4.049237	-0.102667
32	1	0	-4.643969	-3.450601	-2.490252
33	1	0	-3.020255	-3.374710	-1.792406
34	1	0	-3.941667	-4.889312	-1.743694
35	1	0	-4.046012	-3.806119	1.804984
36	1	0	-3.554886	-5.073442	0.665905
37	1	0	-2.659356	-3.544816	0.730625
38	1	0	-5.924559	-5.143552	-0.160040
39	1	0	-6.418977	-3.795519	0.875873
40	1	0	-6.698501	-3.718296	-0.872131
41	6	0	6.573385	-2.944376	-0.397336
42	6	0	7.886316	-2.593906	-0.742833
43	6	0	6.231577	-4.293996	-0.233394
44	6	0	8.839935	-3.586061	-0.930908
45	1	0	8.151351	-1.550926	-0.877734
46	6	0	7.192158	-5.278879	-0.420672
47	1	0	5.227669	-4.577005	0.065386
48	6	0	8.495457	-4.927552	-0.771291
49	1	0	9.852817	-3.311916	-1.205439
50	1	0	6.925338	-6.321342	-0.285293
51	1	0	9.243450	-5.699964	-0.917159
52	6	0	6.099652	1.626499	0.446387
53	6	0	7.441119	1.542787	0.844883
54	6	0	5.496571	2.881419	0.279096
55	6	0	8.163822	2.704339	1.084894
56	1	0	7.906458	0.572539	0.981366
57	6	0	6.227532	4.036957	0.520143
58	1	0	4.469766	2.963948	-0.063742
59	6	0	7.559329	3.950635	0.924822
60	1	0	9.199131	2.636919	1.401101
61	1	0	5.758414	5.005281	0.383368
62	1	0	8.127566	4.855827	1.112253
63	1	0	-0.792973	-1.623119	-0.261680
64	1	0	-3.204744	0.847753	-1.339707
65	1	0	-3.267058	0.849120	0.432923
66	6	0	-4.955368	-1.097749	1.744864
67	6	0	-6.236634	-0.964157	2.300035
68	6	0	-3.848079	-0.936503	2.592629
69	6	0	-6.408350	-0.686781	3.654944

70	1	0	-7.114407	-1.068995	1.664897
71	6	0	-4.012590	-0.651638	3.946225
72	1	0	-2.842269	-1.035308	2.185882
73	6	0	-5.295182	-0.529040	4.478524
74	1	0	-7.408108	-0.588729	4.066794
75	1	0	-3.143875	-0.525498	4.585248
76	1	0	-5.426679	-0.308380	5.533464
77	6	0	-5.973172	-0.648110	-1.090134
78	6	0	-6.543902	-1.172080	-2.260603
79	6	0	-6.347516	0.652729	-0.709855
80	6	0	-7.449056	-0.432848	-3.019034
81	1	0	-6.288864	-2.174400	-2.591762
82	6	0	-7.246416	1.400357	-1.466883
83	1	0	-5.940739	1.087879	0.201282
84	6	0	-7.799904	0.856194	-2.623952
85	1	0	-7.880422	-0.863654	-3.917315
86	1	0	-7.518930	2.402789	-1.151325
87	1	0	-8.504969	1.433579	-3.214090

E(UM062X) = -2703.14753888 A.U.