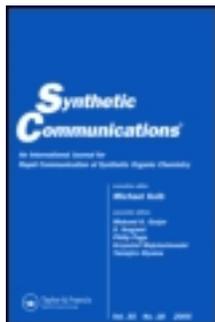


This article was downloaded by: [Moskow State Univ Bibliote]  
On: 17 October 2013, At: 05:41  
Publisher: Taylor & Francis  
Informa Ltd Registered in England and Wales Registered Number: 1072954  
Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH,  
UK



## Synthetic Communications: An International Journal for Rapid Communication of Synthetic Organic Chemistry

Publication details, including instructions for authors and subscription information:

<http://www.tandfonline.com/loi/lcyc20>

### Cyclohexenones Through Regioselective Addition of 1,3-Dicarbonyl Compounds to Terpenoid-Like Bischalcones

Esra Findik<sup>a</sup>, Yakup Budak<sup>a</sup> & Mustafa Ceylan<sup>a</sup>

<sup>a</sup> Department of Chemistry, Faculty of Arts and Sciences, Gaziosmanpasa University, Tokat, Turkey  
Published online: 08 Sep 2009.

To cite this article: Esra Findik, Yakup Budak & Mustafa Ceylan (2009) Cyclohexenones Through Regioselective Addition of 1,3-Dicarbonyl Compounds to Terpenoid-Like Bischalcones, *Synthetic Communications: An International Journal for Rapid Communication of Synthetic Organic Chemistry*, 39:20, 3647-3656, DOI: [10.1080/00397910902793877](https://doi.org/10.1080/00397910902793877)

To link to this article: <http://dx.doi.org/10.1080/00397910902793877>

PLEASE SCROLL DOWN FOR ARTICLE

Taylor & Francis makes every effort to ensure the accuracy of all the information (the "Content") contained in the publications on our platform. However, Taylor & Francis, our agents, and our licensors make no representations or warranties whatsoever as to the accuracy, completeness, or suitability for any purpose of the Content. Any opinions and views expressed in this publication are the opinions and views of the authors, and are not the views of or endorsed by Taylor & Francis. The accuracy of the Content should not be relied upon and should be independently verified with

primary sources of information. Taylor and Francis shall not be liable for any losses, actions, claims, proceedings, demands, costs, expenses, damages, and other liabilities whatsoever or howsoever caused arising directly or indirectly in connection with, in relation to or arising out of the use of the Content.

This article may be used for research, teaching, and private study purposes. Any substantial or systematic reproduction, redistribution, reselling, loan, sub-licensing, systematic supply, or distribution in any form to anyone is expressly forbidden. Terms & Conditions of access and use can be found at <http://www.tandfonline.com/page/terms-and-conditions>

## Cyclohexenones Through Regioselective Addition of 1,3-Dicarbonyl Compounds to Terpenoid-Like Bischalcones

Esra Findik, Yakup Budak, and Mustafa Ceylan

Department of Chemistry, Faculty of Arts and Sciences,  
Gaziosmanpasa University, Tokat, Turkey

**Abstract:** Terpenoid-like bischalcones (**3** and **4**) were synthesized from the reaction of  $\alpha$ - and  $\beta$ -ionones and benzaldehydes in excellent yields. The Michael addition of 1,3-dicarbonyl compounds to bischalcones (**3** and **4**) resulted in the formation of cyclohexenones derivatives (**10a–d** and **14a, b**) via regioselective addition of 1,3-dicarbonyls and then cyclization.

**Keywords:** Cyclization, cyclohexenones, regioselective addition, terpenoid-like bischalcone

1,5-Diaryl-1,4-pentadien-3-ones (bischalcones) are important intermediates and raw materials widely used as precursors to drugs,<sup>[1]</sup> nonlinear optical materials, organometallic materials,<sup>[2,3]</sup> and polymers.<sup>[4]</sup> The Michael reaction of bischalcones with active methylene compounds such as 1,3-dicarbonyl compounds has been the subject of many investigations.<sup>[5–9]</sup> It is known that a weak base or acid such as piperidine<sup>[10]</sup> or phosphorustrichloride<sup>[11]</sup> often affords open-chain adducts, whereas cyclic products have been obtained in the presence of sodium methoxide or sodium hydroxide.<sup>[12]</sup> Particularly, the products of these cyclic reactions are of interest in terms of their stereochemistry and as starting materials for the synthesis of compounds with possible biological activity.

Received December 23, 2008.

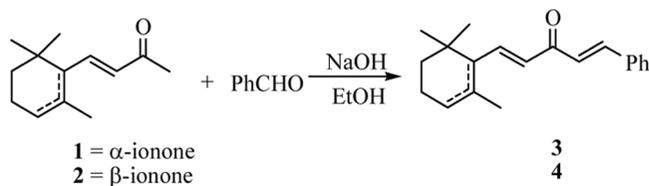
Address correspondence to Esra Findik, Department of Chemistry, Faculty of Arts and Sciences, Gaziosmanpasa University, 60250 Tokat, Turkey. E-mail: mceylan@gop.edu.tr

Reddy et al.<sup>[13]</sup> have reported that the synthesis of 1,1-disubstituted-2,6-diarylcyclohexane-4-ones was carried out in moderate to good yields via double Michael addition of bis-chalcone with active methylene compounds in the presence of sodium methoxide. Addition, the Michael reaction of chalcones with 1,3-dicarbonyl compounds may be exploited with the view of obtaining highly functionalized cyclohexene derivatives but is more commonly used for the preparation of 3,5-diaryl-6-substituted cyclohexenones. The mentioned cyclohexenones are efficient synthons in benzoselenadiazoles and benzothiadiazoles,<sup>[14]</sup> benzopyrazoles and benzisoxazoles,<sup>[15,16]</sup> and carbazole derivatives.<sup>[17]</sup>

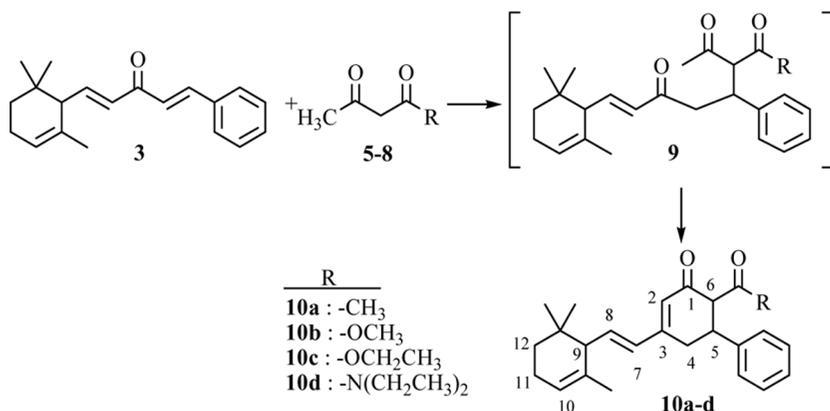
In this study, we report synthesis of cyclohexenone derivatives by regioselective Michael addition of 1,3-dicarbonyl compounds to bischalcones (**3** and **4**), the latter which were obtained by intramolecular Knoevenagel condensation of the methyl group in 1,3-dicarbonyl compounds. The formation of these products was confirmed by infrared (IR) and <sup>1</sup>H NMR spectral studies.

The terpenoid-like bischalcones **3** and **4** were synthesized from the reaction of  $\alpha$ - and  $\beta$ -ionones with the benzaldehyde in yields of 90% and 89%, respectively (Scheme 1). The reactions were performed in the presence of aqueous NaOH in ethanol (EtOH) at room temperature for 3 h. The compounds **3** and **4** are best known in the literature.<sup>[18–22]</sup>

The reaction of chalcones and bischalcones with 1,3-dicarbonyl compounds is known to lead to three structurally diverse types of compounds, depending on the experimental conditions employed: pyrylium salts,<sup>[23]</sup> Michael addition products,<sup>[24]</sup> and cyclohexenone derivatives. The catalyst plays a key role in directing the reaction to different end products. The basic catalyst would turn the intermediate Michael addition product into cyclohexenones through the intramolecular cyclocondensation of the methyl group originating from 1,3-dicarbonyl compounds and the ketone function of the initial  $\alpha,\beta$ -unsaturated system.<sup>[25]</sup> In this research, acetylacetone (**5**), methyl acetoacetate (**6**), ethyl acetoacetate (**7**), and N,N-diethyl-3-oxobutanamide (**8**) were chosen as 1,3-dicarbonyl compounds. The reaction of **3** with 1,3-dicarbonyl



*Scheme 1.* Synthesis of bischalcones.



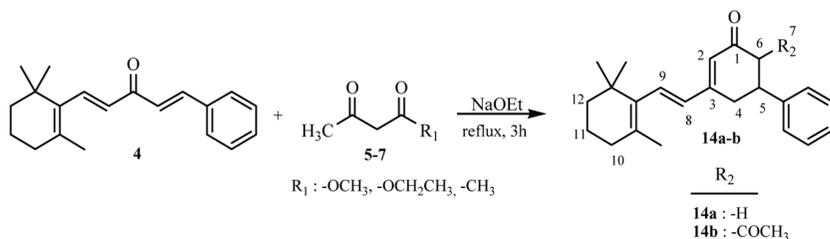
**Scheme 2.** Addition of 1,3-dicarbonyls to bischalcone (3).

compounds 5-8 in basic medium resulted in the formation 3,5,6-trisubstituted cyclohexenones derivatives (10a-d) by means of an intermediate Michael adduct, as outlined in Scheme 2.

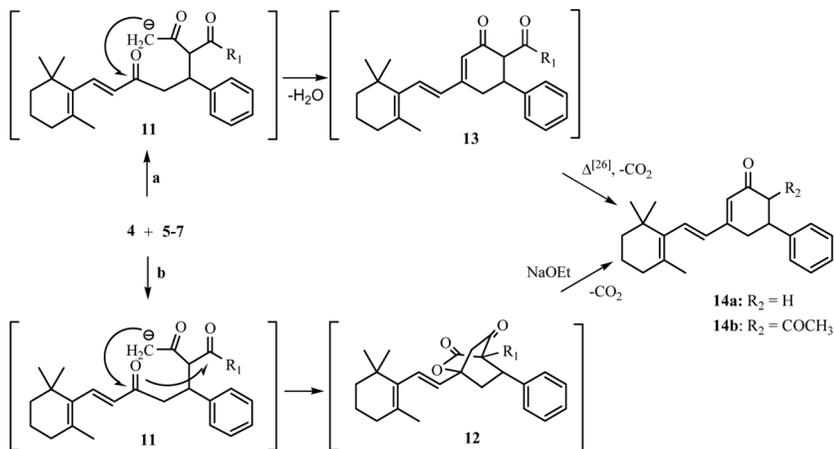
The reaction of 4 with acetylacetone (5), methyl acetoacetate (6), and ethyl acetoacetate (7) in the same conditions gave the 3,5-disubstituted cyclohexenone 14a and 3,5,6-trisubstituted cyclohexenone 14b, different from that of 3.

Control of selectivity, for example, chemo- and regioselectivity, is among the most important objectives in organic chemistry. In this reaction, 1,3-dicarbonyl compounds may be added to two sides (phenyl side or cyclohexenyl side) of bischalcones 3 and 4. It was observed that the 1,3-dicarbonyl compounds were added as regioselective from the phenyl side of bischalcones 3 and 4 (Scheme 3).

The formation of compound 14a may be explained by two different mechanisms, depicted in Scheme 4. Path a: Compound 13 converted to compound 14a by elimination of CO<sub>2</sub>, and EtOH, and/or MeOH effect



**Scheme 3.** Addition of 1,3-dicarbonyls to bischalcone (4).



**Scheme 4.** Formation mechanism of cyclohexanone derivatives (**14a–b**).

of base and temperature.<sup>[26]</sup> Path b: Lactone compound **12** may be occurring by the intramolecular esterification of intermediate **11** and elimination of CO<sub>2</sub> from **12** (acting base) to give **14a**.

The structure of cyclohexenone derivatives **10a–d** and **14a, b** were elucidated by spectroscopic studies (IR, <sup>1</sup>H and <sup>13</sup>C NMR, elemental analysis) and comparison with literature data.

In conclusion, the reaction of bischalcones with 1,3-dicarbonyl compounds in the presence of sodium ethoxide has been shown to afford cyclohexenone derivatives via Michael addition followed by intramolecular cyclocondensation.

## EXPERIMENTAL

### Instruments

IR spectrums (KBr or liquid) were recorded on a Jasco Fourier Transform (FT) IR-430 spectrometer. <sup>1</sup>H and <sup>13</sup>C NMR spectra were recorded on a Bruker Avance III instrument. Internal standards were tetramethylsilane (TMS, δ 0.00) for <sup>1</sup>H NMR and CDCl<sub>3</sub> (δ 77.0) for <sup>13</sup>C NMR spectroscopy. *J* values are given in hertz. The multiplicities of the signals in the <sup>1</sup>H NMR spectra are abbreviated by s (singlet), d (doublet), t (triplet), q (quartet), m (multiplet), br (broad), and combinations thereof. Elemental analyses were obtained from a LECO CHNS 932 elemental analyzer.

Reagents  $\alpha$ -ionone,  $\beta$ -ionone, benzaldehyde, and 1,3-dicarbonyl compounds were commercial products of highest reagent grade.

### Synthesis of Bischalcones 3 and 4

A solution of NaOH (0.26 g, 6.5 mmol) in water was added to a stirred solution of ionone (0.5 g, 2.6 mmol) and benzaldehyde (2.6 mmol) in ethanol (10 ml) and stirred at room temperature for 3 h. The reaction was complete; the mixture was diluted with  $\text{CHCl}_3$ , treated with HCl solution (10%), and washed with water. The organic layer was dried over anhydrous  $\text{Na}_2\text{SO}_4$ . The solvent was evaporated. The crude product was chromatographed on a silica-gel column, eluting with  $\text{CHCl}_3$ /petroleum ether (2:8).

#### 1-Phenyl-5-(2,6,6-trimethylcyclohex-2-enyl)penta-1,4-dien-3-one (3)

Viscous oil;  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  = 7.66 (d,  $J$  = 16 Hz, 1H, H1, A part of AB system), 7.55–7.54 (m, 2H, ArH), 7.35–7.34 (m, 3H, ArH), 6.99 (d,  $J$  = 16 Hz, 1H, H2, B part of AB system), 6.83 (dd,  $J$  = 15.6, 9.6 Hz, 1H, H5, A part of AB system), 6.39 (d,  $J$  = 15.6 Hz, 1H B part of AB system, H4), 5.48 (m, as if brs, 1H), 2.32 (d,  $J$  = 9.6 Hz, 1H), 1.58 (s, 3H), 1.54–2.45 (m, 2H), 1.24–1.18 (m, 2H), 0.93 (s, 3H), 0.87 (s, 3H).  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ ):  $\delta$  = 188.76, 148.74, 143.0, 134.81, 131.96, 130.54, 130.34, 128.88 (2C), 128.30 (2C), 124.77, 122.63, 54.52, 32.62, 31.18, 27.89, 26.85, 23.06, 22.91. IR (liquid): 2955, 2915, 1655, 1623, 1598, 1338, 989, 794,  $757\text{ cm}^{-1}$ . Anal. calcd. for  $\text{C}_{20}\text{H}_{24}\text{O}$ : C, 85.67; H, 8.63. Found: C, 85.42; H, 8.38.

#### 1-Phenyl-5-(2,6,6-trimethylcyclohex-1-enyl)penta-1,4-dien-3-one (4)

Viscous oil;  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ ):  $\delta$  = 7.64 (d,  $J$  = 16 Hz, 1H, H1, A part of AB system), 7.55–7.52 (m, 3H, 2ArH and H5), 7.34–7.33 (m, 3H, ArH), 6.99 (d,  $J$  = 16 Hz, 1H, H2, B part of AB system), 6.47 (d,  $J$  = 16 Hz, 1H, B part of AB system, H4), 2.06–2.04 (m, 2H), 1.80 (s, 3H), 1.61–1.59 (m, 2H), 1.48–1.46 (m, 2H), 1.09 (s, 6H).  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ ):  $\delta$  = 188.21, 142.92, 142.63, 136.61, 136.48, 134.89, 130.28, 129.56, 128.88 (2C), 128.27 (2C), 125.66, 39.69, 34.14, 33.69, 28.87 (2C), 21.89, 19.08. IR (liquid): 2958, 2930, 2865, 1671, 1653, 1612, 1449, 1333, 1088, 982, 786,  $758\text{ cm}^{-1}$ . Anal. calcd. for  $\text{C}_{20}\text{H}_{24}\text{O}$ : C, 85.67; H, 8.63. Found: C, 85.74; H, 8.47.

### General Procedure for Synthesis of Cyclohexenone Derivatives (10a–d and 14a, b)

Bischalcone (**3–4**) (1.78 mmol) and 1,3-dicarbonyl (**5–8**) (1.78 mmol) were refluxed 3 h in 10–15 mL ethanol in the presence of sodium ethoxide (0.1 mmol). Water was added to the mixture and extracted with EtOAc (3 × 20 mL), dried over anhydrous Na<sub>2</sub>SO<sub>4</sub>, and evaporated. Crude products were purified by column chromatography on silica gel or preparative thin-layer chromatography (TLC; 20 × 20 cm plates, 2 mm thick) using *n*-hexane/EtOAc (9:1) as eluent.

#### 6-Acetyl-5-phenyl-3-(2-(2,6,6-trimethylcyclohex-2-enyl)vinyl)cyclohex-2-enone (**10a**)

Yield: 78%; viscous oil. IR  $\nu$  (KBr): 3085, 3061, 3025, 2957, 2918, 2867, 1660, 1624, 1453, 1382, 1363, 1292, 1252, 1216, 970, 755, 700, 667 cm<sup>-1</sup>; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  = 7.39–7.25 (m, 5H), 6.26 (d, *J* = 15.6 Hz, 1H, H7), 6.09–6.02 (m, 2H, H8 and H2), 5.47 (m, as if brs, 1H, H10), 3.35–3.32 (m, 2H, H4), 2.89–2.81 (m, 1H, H5), 2.72–2.68 (m, 1H, H6), 2.26 (d, *J* = 9.2 Hz, 1H, H2), 2.02 (s, 3H, -CH<sub>3</sub>) 1.60 (s, 3H, -CH<sub>3</sub>), 1.42–1.37 (m, 2H, H11), 1.23–1.16 (m, 2H, H12), 0.94 (s, 3H, -CH<sub>3</sub>), 0.83 (s, 3H, -CH<sub>3</sub>); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>):  $\delta$  = 199.66, 156.32, 143.48, 140.39, 132.03, 128.97, 128.89, 128.81, 127.31, 126.95, 126.82, 126.53, 122.68, 55.11, 44.53, 40.74, 38.99, 33.52, 32.61, 31.33, 27.90, 26.88, 23.03. Anal. calcd. for C<sub>25</sub>H<sub>30</sub>O<sub>2</sub>: C, 82.83; H, 8.34; O, 8.83. Found: C, 82.75; H, 8.29; O, 8.74.

#### Methyl 2-Oxo-6-phenyl-4-(2-(2,6,6-trimethylcyclohex-2-enyl)vinyl)cyclohex-3-ene-carboxylate (**10b**)

Yield: 81%; yellowish liquid. IR  $\nu$  (KBr): 3086, 3062, 2954, 2916, 1668, 1624, 1455, 1374, 1240, 1115, 1096, 971, 760, 732, 700 cm<sup>-1</sup>; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  = 7.37–7.26 (m, 5H), 6.24 (d, *J* = 15.6 Hz, 1H, H7), 6.07–6.0 (m, 2H, H8 and H2), 5.46 (m, as if brs, 1H, H10), 3.75–3.68 (m, 2H, H4), 3.57 (s, 3H, -OCH<sub>3</sub>), 2.89–2.84 (m, 1H, H5), 2.65–2.56 (m, 1H, H6), 2.26 (d, *J* = 9.6 Hz, 1H, H9), 1.57 (s, 3H, -CH<sub>3</sub>), 1.44–1.36 (m, 2H, H11), 1.22–1.17 (m, 2H, H12), 0.92 (s, 3H, -CH<sub>3</sub>), 0.82 (s, 3H, -CH<sub>3</sub>); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>):  $\delta$  = 194.24, 169.87, 156.22, 141.36, 141.24, 132.66, 131.45, 128.87, 127.49, 127.23, 125.20, 122.14, 59.96, 59.89, 55.13, 52.01, 43.57, 33.28, 32.64, 31.28, 27.96, 26.79, 23.00. Anal. calcd. for C<sub>25</sub>H<sub>30</sub>O<sub>3</sub>: C, 79.33; H, 7.99; O, 12.68. Found: C, 79.35; H, 7.91; O, 12.63.

Ethyl-2-oxo-6-phenyl-4-(2-(2,6,6-trimethylcyclohex-2-enyl)vinyl)cyclohex-2-enone (**10c**)

Yield: 70%; yellowish liquid. IR  $\nu$  (KBr): 3086, 3062, 2958, 2916, 1666, 1625, 1455, 1381, 1336, 1291, 1258, 970, 757, 700  $\text{cm}^{-1}$ ;  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ ):  $\delta$  = 7.33–7.29 (m, 5H), 6.24 (d,  $J$  = 15.6 Hz, 1H, H7), 6.07–6.03 (m, 2H, H8 and H2), 5.45 (m, as if brs, 1H, H10), 4.00 (q,  $J$  = 7.1 Hz, 2H,  $\text{CH}_2\text{CH}_3$ ), 3.73–3.64 (m, 2H, H4), 2.89–2.83 (m, 1H, H5), 2.67–2.56 (m, 1H, H6), 2.26 (d,  $J$  = 9.85 Hz, 1H, H9), 1.59 (s, 3H,  $-\text{CH}_3$ ); 1.44–1.37 (m, 2H, H11), 1.21–1.17 (m, 2H, H12), 1.01 (t,  $J$  = 7.1 Hz, 3H,  $-\text{CH}_2\text{CH}_3$ ), 0.92 (s, 3H,  $-\text{CH}_3$ ), 0.81 (s, 3H,  $-\text{CH}_3$ );  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ ):  $\delta$  = 194.20, 169.24, 157.00, 141.24, 141.10, 132.63, 131.56, 128.84, 128.76, 127.43, 127.37, 125.35, 122.12, 55.13, 55.08, 43.74, 33.32, 32.63, 32.60, 32.32, 27.95, 26.84, 23.01, 22.93. Anal. calcd. for  $\text{C}_{26}\text{H}_{32}\text{O}_2$ : C, 82.94; H, 8.57; O, 8.50. Found: C, 82.98; H, 8.61; O, 8.41.

N,N-Diethyl-2-oxo-6-phenyl-4-(2-(2,6,6-trimethylcyclohex-2-enyl)vinyl)cyclohex-3-enecarboxamide (**10d**)

Yield: 75%; yellowish liquid. IR  $\nu$  (KBr): 3086, 3062, 2965, 2930, 2896, 1661, 1626, 1454, 1381, 1362, 1280, 1211, 1131, 1099, 1078, 971, 756, 701, 666  $\text{cm}^{-1}$ ;  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ ):  $\delta$  = 7.34–7.20 (m, 5H), 6.22 (d,  $J$  = 15.6 Hz, 1H, H7), 6.06–6.00 (m, 2H, H8 and H2), 5.43 (m, as if brs, 1H, H10), 3.93–3.78 (m, 2H, H4), 3.38–3.32 (m, 4H,  $-\text{NCH}_2\text{CH}_3$ ), 2.91–2.85 (m, 1H, H5), 2.73–2.65 (m, 1H, H6), 2.25 (d,  $J$  = 9.4 Hz, 1H, H9), 1.56 (s, 3H,  $-\text{CH}_3$ ), 1.42–1.37 (m, 2H, H11), 1.19–1.16 (m, 2H, H12), 0.89 (s, 3H,  $-\text{CH}_3$ ), 0.84 (t,  $J$  = 6.8 Hz, 6H,  $-\text{NCH}_2\text{CH}_3$ ), 0.79 (s, 3H,  $-\text{CH}_3$ );  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ ):  $\delta$  = 195.63, 168.19, 156.02, 142.06, 140.79, 132.73, 131.75, 128.49, 127.72, 127.08, 125.93, 122.03, 56.36, 55.08, 43.49, 42.77, 40.59, 32.57, 32.08, 31.34, 27.90, 26.82, 22.91, 22.88, 14.48, 12.71. Anal. calcd. for  $\text{C}_{28}\text{H}_{37}\text{NO}_2$ : C, 80.15; H, 8.89; O, 7.63; N, 3.34. Found: C, 80.11; H, 8.78; O, 7.68; N, 3.29.

5-Phenyl-3-(2-(2,6,6-trimethylcyclohex-1-enyl)vinyl)cyclohex-2-enone (**14a**)

Yield: 68%; yellowish liquid. IR  $\nu$  (KBr): 3086, 3061, 2956, 2928, 2864, 1660, 1607, 1454, 1379, 1360, 1287, 1257, 970, 757, 699  $\text{cm}^{-1}$ ;  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ ):  $\delta$  = 7.38–7.32 (m, 5H), 6.66 (d,  $J$  = 16.2 Hz, 1H, H9

or H8), 6.26 (d,  $J=16.2$  Hz, 1H, H9 or H8), 6.06 (s, 1H, H2), 3.76–3.71 (m, 2H, H4), 2.98–1.93 (m, 1H, H5), 2.71–2.64 (m, 2H, H6), 1.74 (s, 3H,  $-\text{CH}_3$ ), 1.63–1.59 (m, 4H, H10 and H11), 1.48–1.45 (m, 2H, H12), 1.02 (s, 6H,  $2 \times -\text{CH}_3$ );  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ ):  $\delta=194.19, 169.36, 156.50, 141.30, 136.97, 136.24, 133.31, 132.67, 128.79, 127.42, 125.39, 44.08, 39.50, 34.15, 33.21, 32.97, 28.95, 28.91, 21.81, 19.01$ . Anal. calcd. for  $\text{C}_{23}\text{H}_{28}\text{O}$ : C, 86.20; H, 8.81; O, 4.99. Found: C, 86.24; H, 8.87; O, 4.89.

6-Acetyl-5-phenyl-3-(2-(2,6,6-trimethylcyclohex-1-enyl)vinyl)cyclohex-2-enone (**14b**)

Yield: 73%; yellowish liquid. IR  $\nu$  (KBr): 3088, 3064, 2966, 2932, 1653, 1622, 1455, 1379, 1295, 1258, 975, 760, 699  $\text{cm}^{-1}$ ;  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ ):  $\delta=7.42\text{--}7.36$  (m, 2H), 7.34–7.30 (m, 3H), 6.66 (d,  $J=16.2$  Hz, 1H, H9 or H8), 6.26 (d,  $J=16.2$  Hz, 1H, H9 or H8), 6.03 (s, 1H, H2), 3.47–3.33 (m, 2H, H4), 2.94–2.89 (m, 1H, H5), 2.60–2.56 (m, 1H, H6), 2.05 (s, 3H,  $-\text{CH}_3$ ), 1.74 (s, 3H,  $-\text{CH}_3$ ), 1.64–1.56 (m, 4H, H10 and H11), 1.49–1.46 (m, 2H, H12), 1.03 (s, 3H,  $-\text{CH}_3$ ), 1.02 (s, 3H,  $-\text{CH}_3$ );  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ ):  $\delta=199.54, 156.68, 143.58, 137.06, 135.41, 133.27, 132.74, 128.83, 128.72, 127.05, 126.87, 126.48, 44.54, 40.74, 39.49, 34.22, 33.21, 33.14, 28.94, 28.91, 21.78, 19.01$ . Anal. calcd. for  $\text{C}_{25}\text{H}_{30}\text{O}_2$ : C, 82.83; H, 8.34; O, 8.83. Found: C, 82.60; H, 8.40; O, 8.78.

## ACKNOWLEDGMENTS

The authors are indebted to the Department of Chemistry (Gaziosmapasa University) and the Scientific and Technical Research Council of Turkey (Grants TUBITAK-TBAG 106T103 and TUBITAK-BİDEP) for financial support.

## REFERENCES

1. Anto, R. J.; Sukumaran, K.; Kuttan, G.; Rao, M. N. A.; Subbaraju, V.; Kuttan, R. Anticancer and antioxidant activity of synthetic chalcones and related compounds. *Cancer Lett.* **1995**, *97* (1), 33–37.
2. Turowska, I. T. Monitoring initial structural changes in a crystal during photo-induced disappearance of its diffracting properties. *Chem. Phys.* **2003**, *288*, 241–247.
3. Burrows, A. D.; Choi, N.; McPartlin, M.; Mingos, D. M. P.; Tarlton, S. V.; Vilar, R. Syntheses and structural characterisation of the compounds

- [Pd<sub>2</sub>(dba)<sub>3</sub>] (where L = PBz<sub>3</sub> and PPh<sub>2</sub>Np) and of the novel dimer [Pd<sub>2</sub>(μ-dba)(μ-SO<sub>2</sub>)(PBz<sub>3</sub>)<sub>2</sub>]. *J. Organomet. Chem.* **1999**, *573*, 313.
- Insuasty, B.; Rodriguez, R.; Quiroga, J.; Abonia, R.; Martinez, R.; Toscano, A.; Angeles, E. Synthesis of 1-benzyl-6-(4-chlorophenyl)-2-(4-R-phenyl)-4-(4-R-styryl)-2,3-dihydropyrazolo[3,4-b][1,4]-diazepines. *Molecules* **2001**, *6* (8), 710–715.
  - Helmkamp, R. W.; Tanghe, L. J.; John, J.; Plati, J. T. Some Michael condensations involving benzyl cyanide. *J. Am. Chem. Soc.* **1940**, *62* (11), 3215–3219.
  - Kohler, E. P.; Helmkamp, R. W. The addition reactions of certain pentadienones. *J. Am. Chem. Soc.* **1924**, *46*, 1018–1024.
  - Kohler, E. P.; Dewey, C. S. The addition reactions of certain pentadienones, II: Addition of malonic esters. *J. Am. Chem. Soc.* **1924**, *46*, 1267.
  - Hoeve, W. T.; Wynberg, H. Chiral spiranes. Optical activity and nuclear magnetic resonance spectroscopy as a proof for stable twist conformation. *J. Org. Chem.* **1979**, *44* (9), 1508–1514.
  - Rowland, A. T.; Filla, S. A.; Coutlangus, M. L. On the stereochemistry of diaryl-substituted cyclohexanone formed by Michael reactions: Trans to cis isomerization of their ketals under basic conditions. *J. Org. Chem.* **1998**, *63* (13), 4359–4365.
  - Borsche, W. Über die Verteilung der Affinität in ungesättigten organischen Verbindungen. *Liebigs. Ann. chem.* **1910**, *375*, 145–180.
  - Canant, J. B.; Bump, A. H.; Holt, H. S. Addition reactions of phosphorus halides, III: The reaction with dibenzal-acetone and cinnamylideneacetophenone. *J. Am. Chem. Soc.* **1921**, *43* (7), 1677–1684.
  - Loganathan, D.; Verghese, T.; Trivedi, G. K. Phase-transfer catalysed single-pot preparation of 7,11-bisarylspiro (5,5)-undecane-1,9-dione. *Org. Prep. Proced. Intl.* **1984**, *16*, 115–119.
  - Reddy, D. B.; Padmavathi, V.; Reddy, M. M. Double Michael addition reactions of bischalcones. *Indian J. Chem.* **1992**, *31B*, 407.
  - Tabba, H. D.; Yousef, N. M.; Al-Arab, M. M. Michael–Michael aldol reaction of chalcones with cyanoacetylurea and cyanoacetyl piperidine. *Coll. Czech. Chem. Commun.* **1995**, *60* (4), 594–604.
  - Reddy, D. B.; Reddy, A. S.; Padmavathi, V. Synthesis of annelated 1,2,3-selena- or thia-diazoles. *J. Chem. Res., Synop.* **1998**, 784–785.
  - Padmavathi, V.; Mohan Reddy, B. J.; Balaiah, A.; Venugopal Reddy, K.; Bhaskar Reddy, D. Synthesis of some fused pyrazoles and isoxazoles. *Molecules* **2000**, *5*, 1281–1286.
  - Padmavathi, V.; Sharmila, K.; Balaiah, A.; Reddy, A. S.; Reddy, D. B. Cyclohexenone carboxylate: A versatile source for fused isoxazoles and pyrazoles. *Synth. Commun.* **2001**, *31*, 2119–2126.
  - Padmavathi, V.; Sharmila, K.; Padmaja, A.; Reddy, D. B. An efficient synthesis of 6,8-diaryl-carbazoles via Fischer indole cyclizations. *Heterocycl. Commun.* **1999**, *5*, 451–456.
  - Gandhi, R. P.; Kumar, S.; Aryan, R. C.; Ishar, M. P. S. Regioselective photoelectro-cyclization of (E/E)-arylidene-β-ionones: Synthesis of some

- 1,7,7-trimethyl-3-(*e*-2'-arylethenyl)-2-oxabicyclo [4.4.0] deca-3,5-dienes. *Synth. Commun.* **1989**, *19* (9–10), 1759–1762.
20. Gandhi, R. P.; Aryan, R. C. Photochemical reorganisation of 1,7,7-trimethyl-3-(*E*-2'-arylethenyl)-2-oxabicyclo[4.4.0]deca-3,5-dienes. *J. Chem. Soc., Chem. Commun.* **1988**, *15*, 1024–1025.
21. Ter-Sarkisyan, G. S.; Mikhailov, B. M. Reactions of 1-(2,6,6-trimethyl-2-cyclohexenyl)-3-(ethylthio)-1,3-butadiene with thioacetals. *Zhurnal Organicheskoi Khimii* **1965**, *1* (7), 1239–1241.
22. Ishar, M. P. S.; Rai, T.; Agrawal, S. K.; Saxena, A. K.; Singh, L.; S̄nggh, R.; Bhella, S. S. Synthesis and cytotoxic activity of some novel polycyclic  $\gamma$ -butyrolactones. *Bioorg. Med. Chem. Lett.* **2008**, *18* (17), 4809–4812.
23. Ishar, M. P. S.; Singh, R.; Kumar, K.; Singh, G.; Velmurugan, D.; Subbiah, H. A.; Sundra, R. S. S.; Fun, H. K. Photochemistry of arylidene- $\beta$ -ionones: A highly efficient route to novel tricyclic ketones through intramolecular, exo-selective photochemical (4 + 2) cycloadditions, occurring only in an aqueous organic solvent. *J. Org. Chem.* **2002**, *67* (7), 2234–2240.
24. Davey, W.; Gwilt, J. R. Chalcones and related compounds, part II: Addition of thiols and esters to the chalcone system. *J. Chem. Soc.* **1957**, 1015–1017.
25. Van Allan, J. A.; Reynolds, G. A. Preparation of certain pyrylium salts by using chalcone and boron trifluoride etherate. *J. Org. Chem.* **1968**, *33*, 1102–1105.
26. Roman, G. Cyclohexenones through addition of ethyl acetoacetate to chalcones derived from 2-acetylthiophene. *Acta Chim. Slov.* **2004**, *51*, 537–544.
27. Padmavathi, V.; Reddy, B. J. M.; Balaiah, A.; Reddy, V. K.; Reddy, D. B. Synthesis of some fused pyrazoles and isoxazoles. *Molecules* **2000**, *5*, 1281–1286.