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## Functionally Substituted Isocyanurates with Heteroatoms (O, S) in the N-Alkyl Chain

S. G. Fattakhov, M. M. Shulaeva, and V. S. Reznik

Arbuzov Institute of Organic and Physical Chemistry, Kazan Research Center, Russian Academy of Sciences, Kazan, Tatarstan, Russia

Received May 4, 2000

**Abstract** — Alkylation of disodium benzyl and disodium methyl isocyanurates with  $\beta$ , $\beta$ '-dichlorodiethyl and  $\beta$ , $\beta$ '-dibromodiethyl ethers was used to prepare 1-benzyl and 1-methyl 3,5-bis[2-(2-haloethoxy)ethyl] isocyanurates whose reaction with thiourea followed by hydrolysis gave the corresponding 1-alkyl 3,5-bis[2-(2-mercaptoethoxy)ethyl] isocyanurates. 1-Benzyl and 1-methyl 3,5-bis[2-(2-chloroethyl)ethyl] isocyanurates were reacted with NaSCH<sub>2</sub>CH<sub>2</sub>OH to obtain the corresponding 1-alkyl 3,5-bis[2-(2-hydroxyethylthio)ethyl] isocyanurates. Treatment of the latter two esters with thionyl chloride gave 1-benzyl and 1-methyl 3,5-bis[2-(2-chloroethylthio)ethyl] isocyanurates, respectively.

Earlier [1] we reported the synthesis of a series of 1-alkyl 3,5-bis( $\omega$ -haloalkyl)-1,3,5-triazine-2,4,6-(1*H*,3*H*,5*H*)-triones whose reaction with thiourea followed by hydrolysis of the resulting isothiuronium salts gave rise to 1-alkyl 3,5-bis( $\omega$ -mercaptoalkyl) isocyanurates [2]. Proceeding with these studies, we developed in the present work methods for synthesis of a series of  $\omega$ -functionally substituted isocyanurates with heteroatoms (O, S) in the *N*-alkyl chain, which are of interest as starting materials for subsequent syntheses and as building blocks for constructing macrorings with isocyanurate fragments [3].

1-Benzyl and 1-methyl 3,5-bis[2-(2-haloethoxy)-ethyl] isocyanurates **I–IV** in yields of 46–71% were prepared similarly to bis( $\omega$ -haloalkyl) isocyanurates [1] by alkylation of anhydrous disodium alkyl isocyanurates with excess  $\beta$ , $\beta$ '-dichlorodiethyl or  $\beta$ , $\beta$ '-dibromodiethyl ethers (Chlorex and Bromex) (Table 1).

The disodium salts could also be alkylated with

$$O \longrightarrow N(CH_2)_2X(CH_2)_2Y$$

$$O \longrightarrow N \longrightarrow O$$

$$(CH_2)_2X(CH_2)_2Y$$

I–X

 $R = C_6H_5CH_2, CH_3; X = O, S; Y = Cl, Br, OH, SH (for R, X, Y, see Table 1).$ 

bis(2-chloroethyl) sulfide (yperite) [4], but we considered this approach hardly practicable because of the high toxicity of yperite. Therefore, 1-alkyl 3,5-bis-[2-(2-chloethylthio)ethyl] isocyanurates **V** and **VI** were obtained by reaction of 1-alkyl 3,5-bis(2-chloroethyl) isocyanurates **XI** and **XII** with NaSCH<sub>2</sub>CH<sub>2</sub>OH with subsequent treatment of the resulting 1-alkyl 3,5-bis[2-(2-hydroxyethylthio)ethyl] isocyanurates **VII** and **VIII** with thionyl chloride.

 $\mathbf{V}$ 

VI

VII

VIII

IX

X

<b>Table</b>	1.	Yields,	$R_f$	values,	and	elemental	analyses	of	compounds	I-X	
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Comp.			Y	Yield,	$R_f$ (solvent		Found, %					
no.	R	X			for crystallization)	С	Н	і Н	llg	N	S	
I	C <sub>6</sub> H <sub>5</sub> CH <sub>2</sub>	О	Cl	70	0.33 (benzene–ethyl acetate,	10:1)	49.84	5.2	4 15	.91	9.66	
II	CH <sub>3</sub>	О	Cl	49	0.20 (benzene-ethyl acetate,		40.29	5.0	4 11	.47	19.68	_
III	$C_6H_5CH_2$	О	Br	71	0.33 (benzene-ethyl acetate,	-	41.47	4.4	7 30	.63	8.03	_
IV	CH <sub>3</sub>	O	Br	46	0.39 (benzene-ethyl acetate,	5:1)	32.86	4.4	3 35	.64	9.24	_
${f V}$	$C_6H_5CH_2$	S	Cl	61	0.50 (benzene-ethyl acetate,	46.29	5.0	1 14	.85	9.37	14.02	
VI	CH <sub>3</sub>	S	Cl	65	0.35 (diethyl ether-hexane, 1	:1)	37.45	5.0	0 18	.18	10.85	16.77
VII	$C_6H_5CH_2$	S	ОН	60	0.38 (benzene–methanol, 5:1) 0.29 (ethyl acetate)	49.79	6.3	3	_	10.16	15.25	
VIII	CH <sub>3</sub>	S	ОН	85	0.39 (benzene–methanol, 5:1	40.79	6.0	3	_	12.23	18.46	
IX	$C_6H_5CH_2$	O	SH	50	0.31 (benzene-ethyl acetate,	5:1)	50.00	6.3	2	_	9.37	14.98
X	CH <sub>3</sub>	O	SH	57	0.31 (diethyl ether–hexane, 2 0.71 (diethyl ether)	:1)	40.36	6.1	0	_	11.26	18.58
Compound no.			F			Calculated, %						
			Formula -				]	Н			N	S
					$_{18}\text{H}_{23}\text{Cl}_2\text{N}_3\text{O}_5$ 50.0			36	16.40		9.66	_
П				$C_{12}H_{19}Cl_2N_3O_5$				38	11.18		19.91	-
III				$C_{18}H_{23}Br_2N_3O_5$				45	30.66		8.06	_
	IV		$C_{12}H_{19}Br_2N_3O_5$			32.3	8 4.	30	35.90		9.44	_

$$\mathbf{XI} + 2 \text{NaSCH}_2 \text{CH}_2 \text{OH} \xrightarrow{\text{MeOH, 6 h}} \mathbf{VII} + O \xrightarrow{\text{N}} O \\ (\text{CH}_2)_2 \text{S}(\text{CH}_2)_2 \text{OH}$$

C<sub>18</sub>H<sub>23</sub>Cl<sub>2</sub>N<sub>3</sub>O<sub>5</sub>S<sub>2</sub> C<sub>12</sub>H<sub>19</sub>Cl<sub>2</sub>N<sub>3</sub>O<sub>5</sub>S<sub>2</sub>

 $C_{18}H_{25}N_3O_5S_2$ 

 $C_{12}^{10}H_{21}^{23}N_3O_5S_2$ 

 $C_{18}^{12}H_{25}^{21}N_3O_5S_2$  $C_{12}H_{21}N_3O_5S_2$ 

 $R = C_6H_5CH_2$  (XI),  $CH_3$  (XII).

The reaction of 1-benzyl 3,5-bis(2-chloroethyl) isocyanurate (**XI**) with NaSCH<sub>2</sub>CH<sub>2</sub>OH in methanol (boiling, 6 h) gave, along with compound **VII**, a monosubstitution product, 1-benzyl 3-(2-chloroethyl) 5-[2-(2-hydroxyethylthio)ethyl] isocyanurate (**XIII**) in 19% yield. The same reaction in DMF resulted in exclusive formation of a disubstitution product (Table 1, compound **VIII**). Treatment of compound **XIII** with thionyl chloride provided 1-benzyl 3-(2-

$$\mathbf{XIII} \xrightarrow{SOCl_2} \begin{array}{c} C_6H_5CH_2N & N(CH_2)_2Cl \\ O & N & O \\ (CH_2)_2S(CH_2)_2Cl \\ \mathbf{XIV} \end{array}$$

chloroethyl) 5-[2-(2-chloroethylthio)ethyl] isocyanurate (**XIV**).

46.55

37.12

50.57

41.01

50.57

41.01

XIII

4.99

4.93

5.89

6.02

5.89

6.02

15.28

18.26

9.05

10.82

9.83

11.96

9.83

11.96

13.81

16.51

15.00

18.25

15.00

18.25

Table 2. IR and <sup>1</sup>H NMR spectra of compounds I-X

Commond	IR spectrum <sup>a</sup> (thin layer), v, cm <sup>-1</sup>							
Compound no.	triazine ring [5]	C=O	СО	ОН	SH			
I	753 s	1740 m, 1675 s	1115 s	=				
II	755 s	1740 s, 1675 s	1110 s	_	_			
III	750 s	1740 m, 1675 s	1114 s	_	_			
IV	750 s	1750 m, 1686 s	1114 s	_	_			
${f V}$	750 s	1690 s	_	_	_			
VI	745 s	1740 m, 1675 s	_	_	_			
VII	745 s	1664 s	1033 s	3340 s	_			
VIII	750 s	1740 m, 1673 s	1040 s	3350 s	_			
IX	750 s	1675 s	1100 s	_	2557			
X	745 s	1748 m, 1686 s	1100 s	_	2557			

Comm	<sup>1</sup> H NMR spectrum (CCl <sub>4</sub> ), δ, ppm <sup>b</sup>											
Comp. no.	CH <sub>3</sub>	C <sub>6</sub> H <sub>5</sub>	$C_6H_5CH_2$	$ \begin{array}{c c} \text{NCH}_2\\ (^3J_{\text{HH}} \ 6 \ \text{Hz}) \end{array} $	OCH <sub>2</sub>	CH <sub>2</sub> Hlg	SCH <sub>2</sub>	SH ( <sup>3</sup> J <sub>HH</sub> 8 Hz)				
I	_	7.16 m	4.92 s	4.00 t		3.30–3.72 m	_	_				
II	3.30 s	_	_	4.02 t	3	3.26–3.75 m	_	_				
III	_	7.21 m	4.95 s	4.03 t	3.58-3.80 m	$3.26 \text{ t} (^{3}J_{HH} 6 \text{ Hz})$	_	_				
IV	3.27 s	_	_	4.04 t	3.53-3.83 m	$3.26 \text{ t} (^3J_{HH} 6 \text{ Hz})$	_	_				
$\mathbf{V}$	_	7.21 m	4.95 s	4.00 t	=	$3.74 \text{ t} (^3J_{HH} 8 \text{ Hz})$	2.95 m	_				
VI	3.30 s	_	_	4.00 t	=	$3.60 \text{ t} (^3J_{HH} 8 \text{ Hz})$	3.00 m	_				
VII	_	7.28 m	5.03 s	4.08 t	3.73 t	=	2.65–2.95 m	_				
VIII	3.37 s	_	_	4.14 t	$(^{3}J_{\text{HH}} 5.5 \text{ Hz})$ 3.81 t $(^{3}J_{\text{HH}} 5.5 \text{ Hz})$	_	2.68–2.95 m	_				
IX	_	7.27 m	4.95 s	4.00 t	3.37–3.83 m	_	2.50 q ( <sup>4</sup> J <sub>HH</sub> 7 Hz)	1.38 t				
X	3.28 s	_	_	4.00 t	3.43–3.68 m	_	2.40–2.83 m	1.40 t				

<sup>&</sup>lt;sup>a</sup> Along with the listed bands, the spectra of compounds **I–X** contain strong bands at, cm<sup>-1</sup>, 2960–2860 [v(CH)] and 1450 [ $\delta$ (CH<sub>2</sub>)], as well as 3030 m, 1600 w, 1570 w, 694 s (benzene ring) or 1370–1360 s [ $\delta$ (CH<sub>3</sub>)].

Compounds **III** and **IV** were reacted with thiourea, and the resulting isothiuronium salts were hydrolyzed to obtain thiols **IX** and **X** in 50–60% yields (Table 1). The reactions of chloro derivatives **I** and **II** with thiorea (boiling in isopropanol, 16–18 h) gave no isothiuronium salts.

## **EXPERIMENTAL**

The IR spectra were recorded on a Specord IR-75 instrument in thin film or in suspensions in mineral oil. The <sup>1</sup>H NMR spectra were measured on a Varian T-60 at 60 MHz, internal reference tetramethylsilane.

The <sup>1</sup>H NMR and IR spectra of compounds **I–X** are given in Table 2.

1-Benzyl and 1-methyl 3,5-bis[2-(2-chloroethoxy)ethyl]-1,3,5-triazine-2,4,6(1*H*,3*H*,5*H*)-triones (I, II). Benzyl isocyanurate VII or methyl isocyanurate VIII, 0.05 mol, was added with stirring to sodium butylate obtained from 0.1 mol of sodium in 300 ml of absolute butanol, and the mixture was boiled for 8 h. The butanol was distilled off in a vacuum, the residual solvent was removed by azeotropic distillation with toluene, and the residual toluene was distilled off in a vacuum. The dry disodium salt that remained was treated with 300 ml of absolute DMF and 0.5 mol (5-fold excess) of Cl(CH<sub>2</sub>)<sub>2</sub>O(CH<sub>2</sub>)<sub>2</sub>Cl, the mixture was stirred at 85–90°C for 4–5 h (to pH 7), and the precipitate was filtered off. The DMF and Chlorex were removed in a

b The <sup>1</sup>H NMR spectra of compounds VII and VIII were obtained in CDCl<sub>3</sub>.

vacuum, and the residue was subjected to column chromatography on silica gel, eluent benzene-ethyl acetate, 10:1. Compounds **I** and **II** are transparent yellowish thick oils.

1-Benzyl and 1-methyl 3,5-bis[2-(2-bromoethoxy)ethyl]-1,3,5-triazine-2,4,6(1*H*,3*H*,5*H*)-triones (III, IV) were obtained in a similar way. The reactions were performed at 45–50°C. Compounds III and IV are transparent yellowish thick oils.

1-Benzyl and 1-methyl 3,5-bis[2-(2-chloroethyl-thio)ethyl]-1,3,5-triazine-2,4,6(1*H*,3*H*,5*H*)-triones (V, VI). Thionyl chloride, a 5-fold molar excess, was slowly added at 10–15°C to a solution of compound VII or VIII in absolute 1,2-dichloroethane, and the mixture was boiled for 3 h. After cooling, it was treated with water, the organic layer was dried with MgSO<sub>4</sub>, the solvent was removed, and the residue was subjected to column chromatography on silica gel, eluent benzene–ethyl acetate (first 10:1 and then 5:1). Compounds V and VI were isolated as transparent oils, and compound V crystallized on standing.

**1-Benzyl-3-(2-chloroethyl)-5-[2-(2-chloroethyl-thio)ethyl]-1,3,5-triazine-2,4,6(1H,3H,5H)-trione (<b>XIV**) was obtained in a similar way by treatment with thionyl chloride of compound **XIII**. The eluent for chromatography was benzene. Yield 65%, transparent tar.  $R_f$  0.33 (ether–hexane, 1:1). IR spectrum (thin film), v, cm<sup>-1</sup>: 3030 m, 1580 w, 1600 w, 700 s, 680 s (benzene ring), 2960 s, 2920 s (CH), 1690 s (C=O), 1464 s [δ(CH<sub>2</sub>)], 747 m (triazine ring). <sup>1</sup>H NMR spectrum (CCl<sub>4</sub>), δ, ppm: 2.63–2.95 m (4H, SCH<sub>2</sub>), 3.42–3.73 m (4H, CH<sub>2</sub>Cl), 4.00 t (4H, NCH<sub>2</sub>,  $^3J_{\rm HH}$  6 Hz), 4.95 s (2H, C<sub>6</sub>H<sub>5</sub>C $H_2$ ), 7.22 m (5H, C<sub>6</sub>H<sub>5</sub>). Found, %: C 47.68; H 4.81; Cl 17.36; N 10.18; S 8.08. C<sub>16</sub>H<sub>19</sub>Cl<sub>2</sub>N<sub>3</sub>O<sub>3</sub>S. Calculated, %: C 47.53; H 4.74; Cl 17.54; N 10.39; S 7.93.

1-Benzyl-3,5-bis[2-(2-hydroxyethylthio)ethyl]-1,3,5-triazine-2,4,6(1H,3H,5H)-trione (VII). 2-Mercaptoethanol, 2 g, and 4.3 g of 1-benzyl-3,5-bis-(2-chloroethyl)-1,3,5-triazine-2,4,6(1*H*,31*H*,51*H*)trione (XI) were added to sodium methylate obtained from 0.57 g of sodium in 50 ml of absolute methanol, and the mixture was boiled for 6 h. The precipitate was filtered off, and the filtrate was evaporated in a vacuum. The residue was subjected to column chromatography on silica gel, eluents benzene-ethyl acetate, 5:1, and benzene-methanol, 5:1, in succession. From the benzene-ethyl acetate fraction we obtained 1 g (19%) of 1-benzyl-3-(2-chloroethyl)-5-[2-(2-hydroxyethylthio)ethyl]-1,3,5-triazine-2,4,6(1H, **3H,5H)-trione** (XIII) as a transparent oil.  $R_f$  0.45 (benzene-ethyl acetate, 1:1). IR spectrum (thin film), v, cm<sup>-1</sup>: 3425 m (OH), 3030 m, 1580 w, 1600 w, 694 s, 670 s (benzene ring), 2906 s, 2940 s (CH), 1740 m, 1680 s (C=O), 1440 s [ $\delta$ (CH<sub>2</sub>)], 1040 m (CO), 750 m (triazine ring). <sup>1</sup>H NMR spectrum (CDCl<sub>3</sub>),  $\delta$ , ppm: 2.53–2.83 m (4H, SCH<sub>2</sub>), 3.53–3.73 m (4H, CH<sub>2</sub>Cl and CH<sub>2</sub>O), 4.00 t (4H, NCH<sub>2</sub>, <sup>3</sup> $J_{HH}$  6 Hz), 4.93 s (2H, C<sub>6</sub>H<sub>5</sub>CH<sub>2</sub>), 7.18 m (5H, C<sub>6</sub>H<sub>5</sub>). Found, %: C 50.01; H 5.13; Cl 8.95; N 10.45; S 8.41. C<sub>16</sub>H<sub>20</sub>ClN<sub>3</sub>O<sub>4</sub>S. Calculated, %: C 49.80; H 5.22; Cl 9.19; N 10.89; S 8.31.

From the benzene-methanol fraction we isolated 3.15 g (60%) of compound **VII** as an oil that crystallized on standing.

1-Methyl-3,5-bis[2-(2-hydroxyethylthio)ethyl]-1,3,5-triazine-2,4,6(1*H*,3*H*,5*H*)-trione (VIII). 2-Mercaptoethanol, 1.1 g, was added to sodium methylate obtained from 0.31 g of sodium in 40 ml of absolute methanol, and the mixture was stirred for 15 min. The methanol was removed in a vacuum, and the residue was treated with 30 ml of absolute DMF and 1.8 g of 1-methyl-3,5-bis(2-chloroethyl)-1,3,5-triazine-2,4,6(1*H*,3*H*,5*H*)-trione (**XII**). The mixture was stirred for 1 h at 30°C, cooled, the precipitate was filtered off, the DMF was removed in a vacuum, and the residue was subjected to column chromatography in the above-described conditions to isolate 2 g (85%) of compound **VIII** as a transparent thick oil.

1-Benzyl and 1-methyl-3,5-bis[2-(2-mercaptoethoxy)ethyl]-1,3,5-triazine-2,4,6(1H,3H,5H)-triones (IX, X). Equimolar amounts of compound III or IV and thiourea were boiled in isopropanol for 4–12 h. The alcohol was removed in a vacuum, and the residue was treated with aqueous K<sub>2</sub>CO<sub>3</sub>, stirred for 90°C (on a water bath) for 2 h, and extracted with methylene chloride. The aqueous layer was acidified with HCl, and extracted with methylene chloride. The extract was dried with MgSO<sub>4</sub> and purified on a column of silica gel. Compound IX: eluent benzene–ethyl acetate, 10:1. Yield 50%, transparent yellowish oil. Compound X: eluent benzene–ethyl acetate, 7.5:1. Yield 57%, transparent yellowish oil, crystallizes on standing, mp 103–105°C (from benzene).

**1-Methyl-3,5-bis(2-chloroethyl)-1,3,5-triazine-2,4,6(1***H***,3***H***,5***H***)-trione (XII) was obtained by treatment with thionyl chloride of 1-methyl 3,5-bis-(2-hydroxyethyl) isocyanurate similarly to compound XIII [1]. Yield 56%, mp 114–115°C (from benzene). R\_f 0.25 (ether–hexane, 1:1). IR spectrum (Vaseline), v, cm<sup>-1</sup>: 1750 m, 1665 v.s (C=O), 747 m (triazine ring). <sup>1</sup>H NMR spectrum (CDCl<sub>3</sub>), δ, ppm: 3.32 s (3H, CH<sub>3</sub>), 3.68 t (4H, CH<sub>2</sub>Cl, ^3J\_{\rm HH} 7 Hz), 4.21 t (4H, NCH<sub>2</sub>, ^3J\_{\rm HH} 6.5 Hz). Found, %: C 35.80; H 4.06; Cl 25.15; N 15.52. C\_8H\_{11}Cl\_2N\_3O\_3. Calculated, %: C 35.84; H 4.14; Cl 25.45; N 15.67.** 

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