

Bio-inspired polyene cyclization: aziridinyll polyene cyclization catalyzed by InBr_3 [†]

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This communication describes a highly efficient aziridinyll polyolefin cyclization catalyzed by InBr_3 to synthesize chiral terpenoid bearing a 3-amino group in the A ring; both good yields and excellent asymmetric induction were achieved.

The efficiency exhibited by epoxy-squalene cyclase has continued to intrigue chemists over the decades.¹ Therefore, there have been many efforts directed toward the mechanistic understanding as well as development of new methods which can mimic this fascinating biological process.² Inspired by the epoxy-squalene cyclase, we are interested in the analogous aziridine initiated polyene cyclization. If successful, this aziridine initiated polyene cyclization will provide easy access to a wide variety of biologically interesting aza-terpenes.³ However, despite recent advances in the development of new bio-inspired polyene cyclizations using different initiating groups,^{4,5} to the best of our knowledge, there is no report on the aziridine initiated polyene cyclization. In addition, cationic biomimetic polyene cyclizations promoted by a catalytic amount of promoter are still rare.⁷ In this communication, we document a catalytic bio-inspired polyene cyclization using aziridinyll polyolefins as the starting precursors catalyzed by InBr_3 . Both the tricyclic and tetracyclic terpenoids were obtained in good yields with excellent diastereoselectivities.

Our initial efforts were focused on finding proper reaction conditions to promote the cyclization of aziridinyll olefin^{8,9} **1**, and the results are summarized in Table 1.[†] We found that 0.2 equivalents of InBr_3 successfully catalyzed the cyclization of **1** to afford tricyclic compound **2** in 69% yield as a single isomer (Table 1, entry 3). The relative stereochemistry of the cyclization product was elucidated by a single-crystal X-ray analysis as depicted in Fig. 1 (ESI[†]). When less InBr_3 (<0.2 equiv.) was used, the cyclization products were obtained in lower yield. TfOH (2.0 equivalents) was also found to promote the cyclization smoothly to furnish **2** in 62% yield (Table 1, entry 1), but the yield decreased to 30% when the amount of TfOH was reduced to 0.2 equivalents. On the other hand, other indium salts catalyzed the cyclization reaction to afford the desired product in lower yields (entries 6 and 7). Other Lewis acids,

such as $\text{La}(\text{OTf})_3$, were found to be less effective, affording the cyclization product in even lower yield (entry 8).

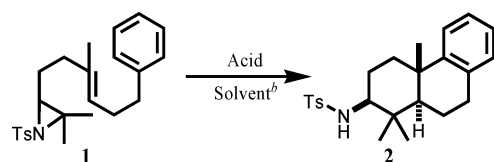
With the optimized reaction conditions in hand, various aziridinyll olefin substrates were carried out to provide a wide variety of 3-amino substituted tricyclic terpenoids,³ and the results are summarized in Table 2. In all cases, good yields and excellent diastereoselectivities (single isomer) were achieved.

Based on the encouraging results from the tricyclic formation of aziridinyll olefin **1** catalyzed by InBr_3 , we further explored the scope of aziridinyll olefin substrates **3**, and the results are summarized in Table 3. In all cases, moderate to good yields with good diastereoselectivities were achieved for tetracyclic cyclization products (Table 3, entries 1 to 5).

The relative stereochemistry of cyclization product **4d** was again elucidated by a single-crystal X-ray analysis as depicted in Fig. 2 (ESI[†]).

Next, we explored the asymmetric version of aziridinyll olefin cyclization with enantiomerically-enriched substrate **8**^{8,10} (97% ee) as shown in Scheme 1. Cbz group was used instead of Ts group to facilitate the removal of the nitrogen protecting group at a later stage. Compound **8** was synthesized from known epoxide^{5d} **5** over three steps in 25% yield without loss of enantiomeric purity. To this end, the aziridinyll olefin **8** was subjected to 0.2 equivalents of InBr_3 for 2 hours at room temperature in CH_2Cl_2 , and the cyclization product **9** was obtained in 67% yield with 97% ee indicating retention of the enantiomeric purity of the cyclization starting material in the

Table 1 Aziridinyll olefin cyclization promoted by various catalysts^a



Entry	Acid	Equiv.	T/°C	Solvent	Yield (%) ^c
1	TfOH	2.0	rt	CH_2Cl_2	62
2	InBr_3	2.0	rt	CH_2Cl_2	65
3	InBr_3	0.2	rt	CH_2Cl_2	69
4	InBr_3	0.2	−20	CH_2Cl_2	Trace ^d
5	InBr_3	0.2	rt	PhMe	58
6	InCl_3	0.2	rt	CH_2Cl_2	28 ^d
7	$\text{In}(\text{OTf})_3$	0.2	rt	CH_2Cl_2	33 ^e
8	$\text{La}(\text{OTf})_3$	0.2	rt	CH_2Cl_2	Trace ^d

^a Conditions: acid and substrate **1** (1.0 equiv., 0.025 M, solvent) were mixed and stirred at room temperature for 2 hours. ^b CH_2Cl_2 was used as solvent except for entry 5. ^c Isolated yield. ^d Starting material was recovered after reaction. ^e Unknown side product was isolated.

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[†] Electronic supplementary information (ESI) available: Detailed experimental procedures, ¹H NMR, ¹³C NMR and analytical data for all compounds. CCDC 715056 and 715057 for **2** and **4d**. For ESI and crystallographic data in CIF or other electronic format see DOI: 10.1039/b903696b

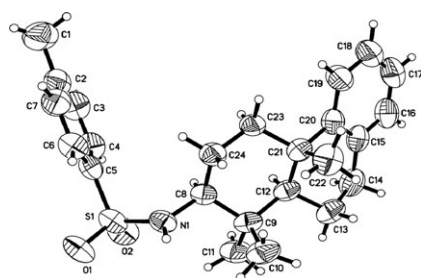
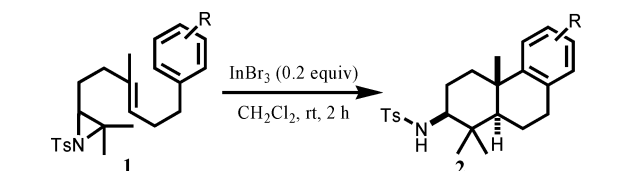


Fig. 1 Single-crystal X-ray structure (50% ellipsoids) of cyclization product **2** (ESI†).

Table 2 Synthesis of tricyclic terpenoids using InBr_3 catalyzed aziridinyll olefin cyclization^a

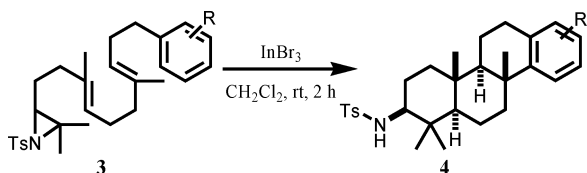


Entry	R	Substrate	Product	Yield (%) ^b
1	—	1	2	69
2	4-Me	1a	2a	70
3	4-OMe	1b	2b	75
4	4- <i>i</i> -Pr	1c	2c	74
5	2-Me	1d	2d	70

^a Conditions: InBr_3 (0.2 equiv.) and substrate **1** (1.0 equiv., 0.025 M, CH_2Cl_2) were mixed and stirred at room temperature for 2 hours.

^b Isolated yield.

Table 3 Synthesis of tetracyclic terpenoids using InBr_3 catalyzed aziridinyll polyolefin cyclization^a



Entry	R	Substrate	Product ^b	Equiv.	Yield (%) ^c	Ratio ^d
1	—	3	4	0.2	63	82 : 18
2	4-Me	3a	4a	0.2	51	84 : 16
3	4-OMe	3b	4b	0.2	55	87 : 13
4	3-Me	3c	4c	0.2	52	87 : 13
5	2-Me	3d	4d	0.2	65	86 : 14
6	2-Me	3d	4d	2.0	62	88 : 12

^a Conditions: InBr_3 (0.2 equiv.) and substrate **3** (1.0 equiv., 0.025 M, CH_2Cl_2) were mixed and stirred at room temperature for 2 hours.

^b Bicycled isomers were obtained. See ESI†. ^c Combined yields of desired product and bicycled isomers. ^d Desired tetracyclic product (major) and bicycled isomers (minor) ratios were determined by ^1H NMR and ^{13}C NMR. The ratio was not improved after treating with TFOH at room temperature for 12 hours.

final cyclization product. In addition, the cyclization product **9** was further functionalized into primary amine **10**, which provides a practical approach to the synthesis of biologically

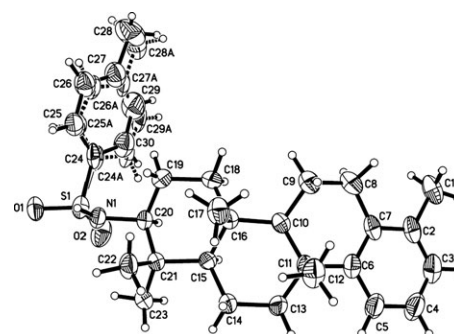
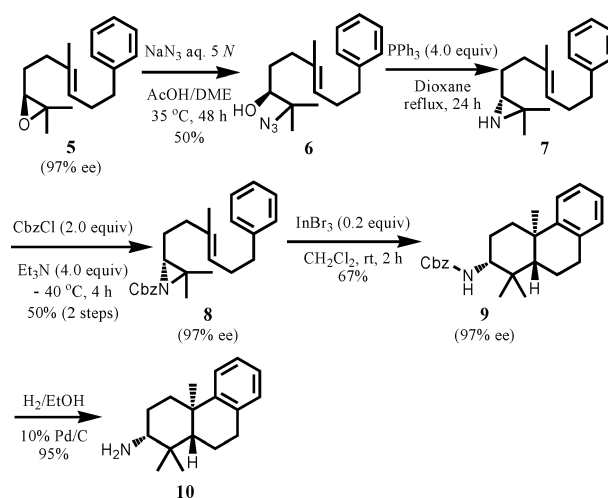


Fig. 2 Single-crystal X-ray structure (50% ellipsoids) of tetracyclic cyclization product **4d** (ESI†).



Scheme 1 InBr_3 catalyzed asymmetrical aziridinyll olefin cyclization.

active chiral terpenoid compounds bearing a 3-amino group in the A ring.³

In conclusion, we have demonstrated an aziridinyll polyolefin cyclization promoted by catalytic amounts of InBr_3 . In contrast to epoxide initiated polyene cyclization, it is surprising to note that this aziridine initiated polyene cyclization works well with a catalytic amount of Lewis acid. Both good yields and good diastereoselectivities were obtained for the A ring 3-amino substituted tricyclic and tetracyclic terpenoid compounds. When an enantiomerically-enriched cyclization substrate was used, retention of the enantiomeric purity of the cyclization starting material was achieved. We believe that this process would allow rapid and efficient access to chiral terpenoid compounds bearing a 3-amino group in the A ring. The applications of this method to the synthesis of diverse compounds as well as total synthesis are currently in progress in our group.

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Notes and references

† Representative procedure: to a 10 mL round-bottom flask equipped with a magnetic stirring bar was added InBr_3 (7 mg, 0.02 mmol, 0.20 equiv.). CH_2Cl_2 (1.5 mL) was added via syringe. Aziridinyll olefin substrate **1** (40 mg, 0.1 mmol, 1.0 equiv.) was added as CH_2Cl_2

solution *via* syringe. The reaction mixture was stirred at room temperature for 2 hours. The reaction mixture was quenched by aqueous NaHCO₃ (5 mL) saturated solution. The mixture was extracted with CH₂Cl₂ (30 mL × 3). The combined organic extracts were washed with brine (20 mL), dried over anhydrous sodium sulfate, filtered and concentrated *in vacuo*. The residual crude product was purified by column chromatography to afford the desired amine product as a colorless solid in 69% yield. Mp: 223.5–225 °C. *R*_f: 0.50 (hexane–ethyl acetate = 4 : 1). ¹H NMR (400 MHz, CDCl₃): 7.78–7.72 (m, 2H), 7.31–7.26 (m, 2H), 7.16–6.98 (m, 4H), 4.27 (d, *J* = 9.33 Hz, 1H), 2.97–2.89 (m, 2H), 2.89–2.78 (m, 1H), 2.43 (s, 3H), 2.19 (dt, *J* = 13.10, 3.10 Hz, 1H), 1.86 (dd, *J* = 13.27, 7.39 Hz, 1H), 1.75–1.65 (m, 1H), 1.60–1.53 (m, 1H), 1.51–1.45 (m, 1H), 1.44–1.36 (m, 1H), 1.35–1.28 (m, 1H), 1.11 (s, 3H), 0.90 (s, 3H), 0.82 (s, 3H). ¹³C NMR (100 MHz, CDCl₃): 148.7, 143.2, 138.1, 134.8, 129.6, 128.9, 127.1, 125.7, 125.5, 124.3, 62.0, 50.8, 38.2, 37.6, 37.2, 30.5, 28.1, 26.7, 24.7, 21.5, 19.2, 16.2. HRMS (ESI): *m/z* calculated for C₂₄H₃₂NO₂S [M + H]⁺: 398.2159, found: 398.2154. FTIR (NaCl): ν 3462 (br), 2957, 2928, 1700 (br), 1645 (br), 1450, 115 cm⁻¹.

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