

Asymmetric Induction of Three Consecutive Chiral Centers by Reactions of *N*-Enoylthioamides with Aldehydes**

Tadashi Kataoka,* Hironori Kinoshita, Sayaka Kinoshita, Takashi Osamura, Shin-ichi Watanabe, Tatsunori Iwamura, Osamu Muraoka, and Genzoh Tanabe

Recently we developed a chalcogenide/TiCl₄-mediated reaction, which consists of the Michael addition of a chloride ion generated from a chalcogenide/TiCl₄ complex to an enone, followed by an aldol reaction with an aldehyde to give an α -chloromethyl aldol, which can be transformed into an α -methylene aldol (the Morita–Baylis–Hillman adduct).^[1,2] This reaction is completed much faster than the Morita–Baylis–Hillman reaction^[3] and, therefore, can be utilized advantageously in several reactions which do not give good results under Morita–Baylis–Hillman reaction conditions.^[4–8] We further studied the tandem Michael–aldol reaction initiated by the intramolecular Michael cyclization of a chalcogenide group to an enone moiety.^[9–11] Goodman et al. reported a similar intermolecular example.^[12]

While investigating the chalcogenide catalyst, we found that a thioketone acted as a nucleophile toward an enone^[13] and a thiourea was also useful for the tandem Michael–aldol reaction.^[14] Recently, it was reported that a thiourea moiety added intramolecularly to an enone moiety to form a thiazine ring.^[15,16] These reports prompted us to examine the tandem Michael–aldol reaction of *N*-propenoyl cyclic thioamides with aldehydes. We now report the asymmetric tandem Michael–aldol reaction of chiral *N*-cinnamoyl-1,3-thiazolidine-2-thione and its 1,3-oxazolidine congener with aldehydes in the presence of BF₃·Et₂O.

The reaction of *N*-cinnamoyl-1,3-thiazolidine-2-thione (**1a**) with *p*-chlorobenzaldehyde (**2a**) was conducted for the first time [Eq. (1)]. The results are shown in Table 1. Various molar ratios of the starting compounds and the Lewis acid were tested, and the best result was obtained from the reaction of **1a** (2 equiv) with **2a** (1 equiv) in the presence of 3 equiv BF₃·Et₂O (Table 1, entry 1). This reaction was quite fast and was completed within 15 minutes at room temper-

[*] Prof. T. Kataoka, H. Kinoshita, S. Kinoshita, T. Osamura, Dr. S. Watanabe, Dr. T. Iwamura
Gifu Pharmaceutical University
6-1 Mitahora-higashi 5-chome, Gifu 502-8585 (Japan)
Fax: (+81) 58-237-5979
E-mail: kataoka@gifu-pu.ac.jp

Prof. O. Muraoka, Dr. G. Tanabe
Faculty of Pharmaceutical Sciences
Kinki University
3-4-1 Kowakae, Higashi-Osaka 577-8502 (Japan)

[**] This work was supported in part by a Grant-in-Aid (No. 3824) from the Japan Society for the Promotion of Science.



Supporting information for this article is available on the WWW under <http://www.angewandte.org> or from the author.

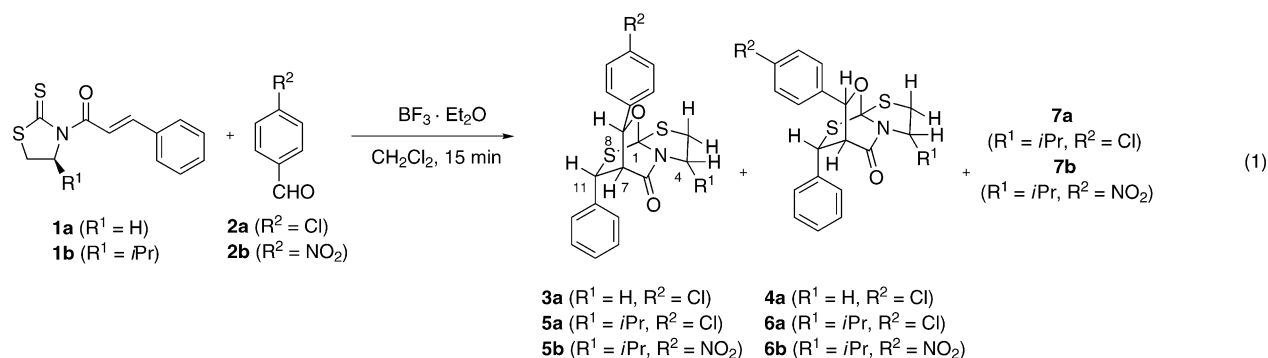


Table 1: Representative results for the screening of reaction conditions for the diastereoselective tandem Michael–aldol reactions of *N*-cinnamoyl-1,3-thiazolidine-2-thiones **1a,b** with aldehydes **2a,b**.

Entry	Enone (equiv)	Aldehyde (equiv)	BF ₃ ·Et ₂ O [equiv]	T	Products ^[a] (% yield)
1	1a (2)	2a (1)	3	RT	3a (58), 4a (31)
2	1a (2)	2a (1)	2	RT	3a (50), 4a (33)
3	1a (1)	2a (2)	3	RT	3a (48), 4a (30)
4	1b (2)	2a (1)	3	0 °C	5a (54), 6a (28), 7a (4)
5	1b (2)	2b (1)	3	0 °C	5b (43), 6b (41), 7b (4)

[a] Minor products **7a** and **7b** have the same molecular weights and composition formulas, C₂₂H₂₂ClNO₂S₂ and C₂₂H₂₂N₂O₄S₂, as **5a**, **6a** and **5b**, **6b**, respectively, but the amounts of **7a** and **7b** are so small that their stereostructures could not yet be determined.

ature. The structures of products **3a** and **4a** were determined by comparing the their ¹H and ¹³C NMR data with those of **5a**, whose structure was elucidated by X-ray analysis (vide infra).

We obtained the three diastereomeric products **5–7** from the reactions of chiral thioamide **1b**, which has a 4*S* configuration, with aldehydes **2a,b**. X-ray analysis revealed that the major product **5a** ($R^1 = iPr$, $R^2 = Cl$) has a tricyclic ring system, in which a chiral bridgehead carbon is bound to four heteroatoms (Figure 1). The configuration of the four chiral centers is 1*R*,4*S*,7*R*,8*S*,11*R*.^[17] The structure of product **6a**, in which the ClC₆H₄ group is on the same side as H-11, was determined by comparing its ¹H and ¹³C NMR spectra with those of **5a**. The absolute configuration of the 8-position is 8*R*, opposite to that of **5a**.

Palomo et al. reported that a chiral thioamide can undergo an asymmetric Michael addition to an intramolecular enone moiety with high diastereoselectivity.^[16] Based on their findings and ours, we anticipated that the stereoselective formation of the four chiral centers involving three consecutive chiral carbons could be achieved if the aldol reaction of an enolate with an aldehyde could be stereocontrolled. Hence, we selected 4,5-disubstituted oxazolidine-2-thione as a chiral auxiliary and carried out reactions of *N*-cinnamoyl-4*S*-methyl-5*R*-phenyloxazolidinethione (**1c**) with various aldehydes [Eq. (2)] (Table 2).

The best molar ratio of enone **1** and aldehyde **2** from Table 1 was applied to the reaction of **1c** and **2b**, and the reaction temperature and time were examined. The reaction at –78 °C for 25 h gave **8b** in 27% yield (Table 2, entry 1). When the reaction was conducted at –40 °C, the

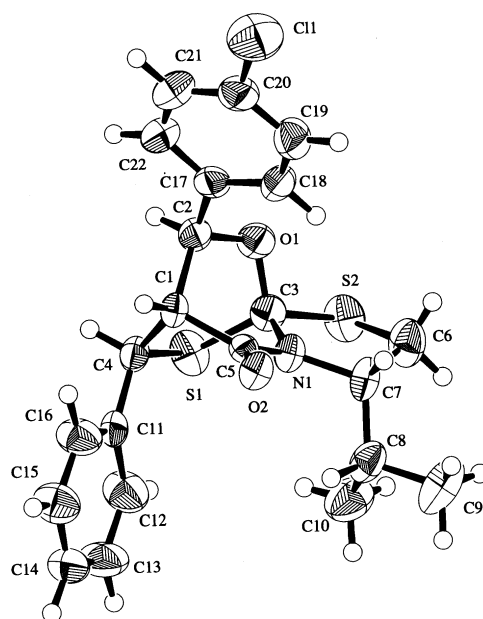
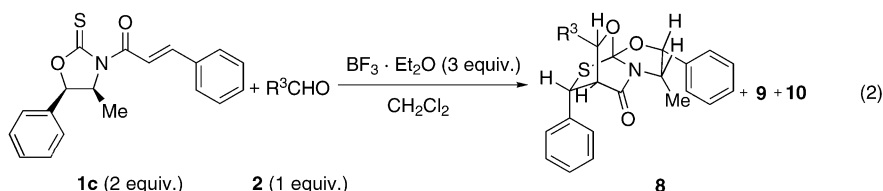


Figure 1. X-ray crystal structure of **5a** (ORTEP drawing).



chemical yield increased but a small amount of diastereomer **9b** formed (entry 2). The best result was obtained for the reaction at –40 °C for 24 h (entry 3). The stereostructure of

Table 2: Diastereoselective tandem Michael–aldol reactions of *N*-cinnamoyl-1,3-oxazolidine-2-thione **1c** with aldehydes **2a–e**.

Entry	Aldehyde (R ³)	Conditions	Yield [%] ^[a]	8 : 9 : 10 ^[b]
1	2b (<i>p</i> -NO ₂ C ₆ H ₄)	−78 °C, 25 h	27	100:0:0
2	2b (<i>p</i> -NO ₂ C ₆ H ₄)	−40 °C, 5 h	77	94:6:0
3	2b (<i>p</i> -NO ₂ C ₆ H ₄)	−40 °C, 24 h	93	95:5:0
4	2c (<i>m</i> -NO ₂ C ₆ H ₄)	−40 °C, 24 h	85	95:5:0
5	2a (<i>p</i> -ClC ₆ H ₄)	−40 °C, 24 h	71	86:7:7
6	2d (Ph)	0 °C, 1 h	69	69:3:28
7	2e (<i>p</i> -MeC ₆ H ₄)	0 °C, 1 h	59	92:0:8

[a] Mixture of diastereoisomers. [b] HRMS data indicate that products **9** and **10** have the same molecular formulas as product **8**, but their stereostructures could not be determined because of the small amounts.

the major product **8b** corresponds to that of **6b**, which has a 4-isopropylthiazolidine moiety. Diastereomer ratios of the products were determined from ¹H NMR spectra of the crude products. Reactions with *p*-chloro- (**2a**) and *m*-nitrobenzaldehyde (**2c**) were conducted similarly and gave products **8a** and **8c** in good yields together with isomers **9a,c** and **10a** (entries 4 and 5). The reaction with benzaldehyde was slow and was conducted at 0 °C, but the diastereomer ratio was decreased (entry 6). Reaction with *p*-tolualdehyde (**2e**) gave the products in a moderate chemical yield (entry 7). The reaction with *o*-nitrobenzaldehyde was very slow because of the steric hindrance, and the reaction with dihydrocinnamaldehyde gave a mixture of products with a low diastereomer ratio.

In conclusion, we have developed a novel tandem Michael–aldol reaction of chiral thioamide-enones with aldehydes, which induces four chiral centers simultaneously. This method is easy to use and gives unusual heterotricyclic compounds with three consecutive chiral centers and a chiral carbon center bound to four heteroatoms. If the products can be converted into polyfunctionalized chiral carboxylic acids, aldehydes, amides or alcohols, they can be widely utilized for organic synthesis. This is the subject of current investigation.

Experimental Section

General procedure: To a stirred solution of (4*S*,5*R*)-4-methyl-5-phenyl-3-[(*E*)-3-phenylprop-2-enoyl]-1,3-oxazolidine-2-thione (**1c**) (323 mg, 1.0 mmol) and *p*-nitrobenzaldehyde (**2b**) (76 mg, 0.5 mmol) in dry CH₂Cl₂ (1.6 mL) was added dropwise a solution of BF₃·Et₂O (190 μL, 1.5 mmol) at −40 °C. The mixture was stirred at the same temperature for 24 h, poured into NaHCO₃ solution, and extracted with CH₂Cl₂. The combined organic layers were dried over MgSO₄ and concentrated under reduced pressure. The residue was purified by recycling preparative HPLC, eluting with chloroform to give **8b** and **9b**.

Received: February 5, 2003 [Z51106]

Keywords: aldol reactions · asymmetric synthesis · C–C coupling · Michael addition · thiocarbonyl compounds

[2] T. Kataoka, H. Kinoshita, T. Iwama, S. Tsujiyama, T. Iwamura, S. Watanabe, O. Muraoka, G. Tanabe, *Tetrahedron* **2000**, *56*, 4725–4731.

[3] For reviews of the Morita–Baylis–Hillman reaction: a) S. E. Drewes, G. H. P. Roos, *Tetrahedron* **1988**, *44*, 4653–4670; b) D. Basavaiah, P. D. Rao, R. S. Hyma, *Tetrahedron* **1996**, *52*, 8001–8062; c) E. Ciganek in *Organic Reactions*, Vol. 51 (Ed.: L. A. Paquette), Wiley, New York, **1997**, pp. 201–350; d) P. Langer, *Angew. Chem.* **2000**, *112*, 3177–3180; *Angew.*

Chem. Int. Ed. **2000**, *39*, 3049–3052; e) D. Basavaiah, A. J. Rao, T. Satyanarayana, *Chem. Rev.* **2003**, *103*, 811–891.

[4] a) T. Kataoka, T. Iwama, H. Kinoshita, S. Tsujiyama, Y. Tsurukami, T. Iwamura, S. Watanabe, *Synlett* **1999**, 197–198; b) T. Kataoka, T. Iwama, H. Kinoshita, Y. Tsurukami, S. Tsujiyama, M. Fujita, E. Honda, T. Iwamura, S. Watanabe, *J. Organomet. Chem.* **2000**, *611*, 455–462.

[5] D. Basavaiah, K. Muthukumaran, B. Sreenivasulu, *Synlett* **1999**, 1249–1250.

[6] T. Bauer, J. Tarasiuk, *Tetrahedron: Asymmetry* **2001**, *12*, 1741–1745.

[7] R. Pathak, A. K. Shaw, A. P. Bhaduri, *Tetrahedron* **2002**, *58*, 3535–3541.

[8] T. Kataoka, H. Kinoshita, S. Kinoshita, T. Iwamura, S. Watanabe, *Angew. Chem.* **2000**, *112*, 2448–2450; *Angew. Chem. Int. Ed.* **2000**, *39*, 2358–2360.

[9] T. Kataoka, S. Kinoshita, H. Kinoshita, M. Fujita, T. Iwamura, S. Watanabe, *Chem. Commun.* **2001**, 1958–1959.

[10] T. Kataoka, H. Kinoshita, S. Kinoshita, T. Iwamura, *J. Chem. Soc. Perkin Trans. 1* **2002**, 2043–2045.

[11] T. Kataoka, H. Kinoshita, S. Kinoshita, T. Iwamura, *Tetrahedron Lett.* **2002**, *43*, 7039–7041.

[12] L. M. Walsh, C. L. Winn, J. M. Goodman, *Tetrahedron Lett.* **2002**, *43*, 8219–8222.

[13] T. Iwama, H. Kinoshita, T. Kataoka, *Tetrahedron Lett.* **1999**, *40*, 3741–3744.

[14] S. Kinoshita, H. Kinoshita, T. Iwamura, S. Watanabe, T. Kataoka, *Chem. Eur. J.* **2003**, *9*, 1496–1502.

[15] A. Hari, B. L. Miller, *Org. Lett.* **2000**, *2*, 3667–3670.

[16] a) C. Palomo, M. Oiarbide, F. Dias, A. Ortiz, A. Linden, *J. Am. Chem. Soc.* **2001**, *123*, 5602–5603; b) A. Ortiz, L. Quintero, H. Hernández, S. Maldonado, G. Mendoza, S. Bernès, *Tetrahedron Lett.* **2003**, *44*, 1129–1132.

[17] Crystal structure data for **5a**: C₂₂H₂₂ClNO₂S₂, *M*_r = 431.99, prismatic, space group *P*2₁2₁2₁, *a* = 11.636(2), *b* = 18.035(2), *c* = 9.947(2) Å, *V* = 2087.4(5) Å³, *T* = 296 K, *Z* = 4, ρ_{calcd} = 1.375 g cm^{−3}, μ(MoKα) = 4.01 cm^{−1}, *R* = 0.0243, *R*_w = 0.061. CCDC-200264 contains the supplementary crystallographic data for this paper. These data can be obtained free of charge via www.ccdc.cam.ac.uk/conts/retrieving.html (or from the Cambridge Crystallographic Data Centre, 12 Union Road, Cambridge CB2 1EZ, UK; fax: (+44) 1223-336-033; or deposit@ccdc.cam.ac.uk).

[1] a) T. Kataoka, T. Iwama, S. Tsujiyama, *Chem. Commun.* **1998**, 197–198; b) T. Kataoka, T. Iwama, S. Tsujiyama, T. Iwamura, S. Watanabe, *Tetrahedron* **1998**, *54*, 11813–11824.