

Catalytic Asymmetric Access to α,β Unsaturated δ -Lactones through a Vinylogous Aldol Reaction: Application to the Total Synthesis of the Prelog-Djerassi Lactone

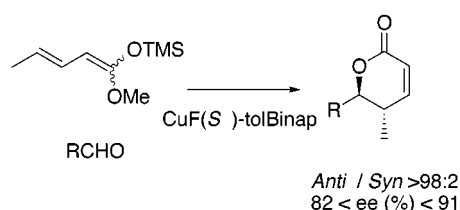
Guillaume Bluet, Belén Bazán-Tejeda, and Jean-Marc Campagne*

Institut de Chimie des Substances Naturelles 91198 Gif sur Yvette, France

jean-marc.campagne@icsn.cnrs-gif.fr

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ABSTRACT



A one-step catalytic asymmetric access to α,β unsaturated δ -lactones is described, using a vinylogous Mukaiyama-aldol reaction between a γ -substituted dienolate and various aldehydes in the presence of Carreira catalyst CuF-(S)-tolBinap. This methodology has been further applied to a straightforward access to the Prelog-Djerassi lactone.

The α,β unsaturated and saturated δ -lactones are found in an impressive number of natural and unnatural products possessing interesting biological activities.^{1–9} These com-

pounds are also useful chiral building blocks, such as for example the Prelog-Djerassi lactone.^{10,11} Efficient asymmetric syntheses of such lactones have been described but required the use of a stoichiometric amount of chiral auxiliary and/or a multiple-step sequence.^{12–18}

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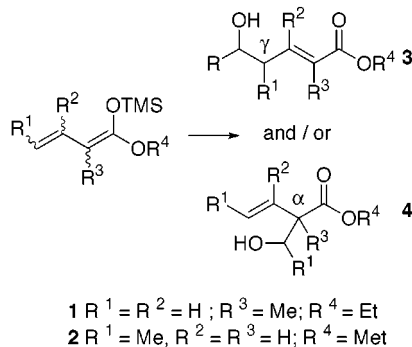
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We wish to report herein an efficient catalytic asymmetric one-step protocol to access α,β unsaturated δ -lactones using a vinylogous aldol reaction.¹⁹

Recently, we have described the efficient formation of vinylogous aldol product using silyl dienolate **1** in good yields, excellent γ : α regioselectivity, and moderate to good enantioselectivity (Scheme 1).²⁰

Scheme 1. Vinylogous Mukaiyama-Aldol Reactions

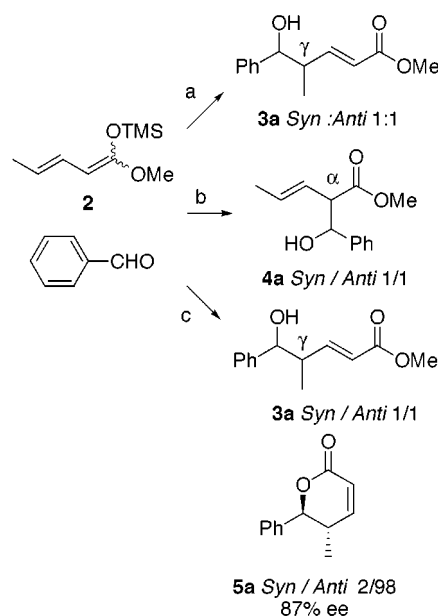


To observe the influence of a γ substituent on the course (α/γ and *syn/anti* ratio) of the reaction, we envisioned the reaction of γ -substituted silyl-dienolate **2**²¹ with benzaldehyde. Unexpected results were observed depending on the nature of the dienolate activation (Scheme 2).

Using 10% of tetrabutylammonium triphenyldifluorosilicate TBAT as a racemic nonhygroscopic source of fluoride,²² the expected vinylogous aldol product **3a** was isolated in 45% yield in a disappointing 1:1 *syn/anti* ratio. Changing the fluoride source to a chiral nonracemic ammonium fluoride,^{20c} we were surprised to isolate only the α aldol product in 68% yield, in a 1:1 *syn/anti* mixture.

Moving to the Carreira catalyst $CuF \cdot (S)\text{-tolBinap}$,²³ a 14:86 mixture of the vinylogous aldol product **3a** and the

Scheme 2^a



^a (a) TBAT 10%, THF, rt, 60%; (b) *N*-benzyl cinchodinium fluoride 10%, THF, rt, 60%; (c) $CuF \cdot (S)\text{-tolBinap}$, 10%, rt, 85% (**3a/5a** 16/84).

lactone **5a** was isolated in 85% yield (Table 1). The vinylogous aldol product **3a** was obtained with no *syn/anti* diastereoselectivity and very poor enantioselectivities (<5% ee for both *syn* and *anti* products). On the other hand, the α,β unsaturated lactone **5a** was found to be highly *anti* selective (*syn/anti* > 2:98) in 87% ee, suggesting that a more organized transition state had occurred.

Table 1. Vinylogous Mukaiyama Reactions of Dienolate **2** with Various Aldehydes in the Presence of 10% of $CuF \cdot (S)\text{-tolBinap}$

entry	aldehyde	yield % (4 + 5)	ratio ^a 5/4	no.	lactones 5 <i>anti/syn</i> ^a	ee
1	benzaldehyde	85	86/14	5a	>98/2	87 ^b , 98 ^c
2	2-naphthaldehyde	95	80/20	5b	>98/2	85 ^d
3	2,3-dimethoxy benzaldehyde	87	81/19	5c	>98/2	91 ^e , 98 ^c
4	2-furaldehyde	60	50/50	5d	>98/2	86 ^f
5	(<i>E</i>)-cinnamal- dehyde	60	70/30	5e	>98/2	82 ^g
6	isobutyraldehyde	95	64/36	5f	>98/2	91 ^h

^a Determined by ¹H NMR on the crude product. ^b HPLC DAICEL-OD, hexane/2-propanol 95/5. ^c After recrystallization (heptane). ^d HPLC DAICEL-OJ, hexane/2-propanol 82/18. ^e HPLC DAICEL-OJ, hexane/2-propanol 95/5. ^f HPLC DAICEL-OJ, hexane/2-propanol 95/5. ^g HPLC DAICEL-OJ, hexane/2-propanol 90/10. ^h HPLC ChiralPAK AD, hexanes/ethanol 99/1.

The reaction with other aromatic aldehydes (Table 1, entries 2–4) was also efficient, leading to lactones with excellent diastereoselectivities (>2:98 *syn/anti*) and high enantioselectivities (85–91% ee and even 98% ee for the

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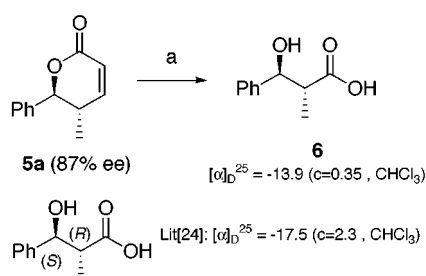
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recrystallized lactones **5a** and **5c**). Reaction with 2-furaldehyde proved to be less selective, leading to the lactone with a high ee (86%) but generally in a 1:1 ratio of linear and lactone products. The reactions with unsaturated (entry 5) and aliphatic (entry 6) aldehydes were also efficient in terms of *anti/syn* ratio (>98/2) and ee (respectively, 82%, 91%), but the lactone/linear product ratios were somewhat lower (respectively, 70/30, 64/36) compared to aromatic aldehydes.

To determine the absolute configuration of the lactones, lactone **5a** was oxidized to the previously described enantiomerically pure compound **6**.²⁴ A rotation of -13.9 (for 87% ee) was found, in agreement with the reported rotation of -17.5 described for the enantiomerically pure *anti* (2*R*,3*S*) compound.

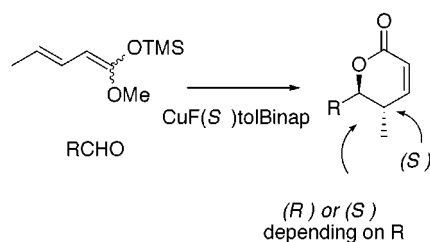
Scheme 3^a



^a (a) RuCl_3 6%, NaIO_4 4.2 equiv, $\text{CH}_3\text{CN}/\text{CCl}_4/\text{H}_2\text{O}$ (1/1/1.5), 50 °C, 24 h, 35%.

Consequently, absolute configurations of lactones **5**, obtained with $\text{CuF}\cdot(\text{S})\text{-tolBinap}$, were tentatively assigned as shown in Scheme 4.

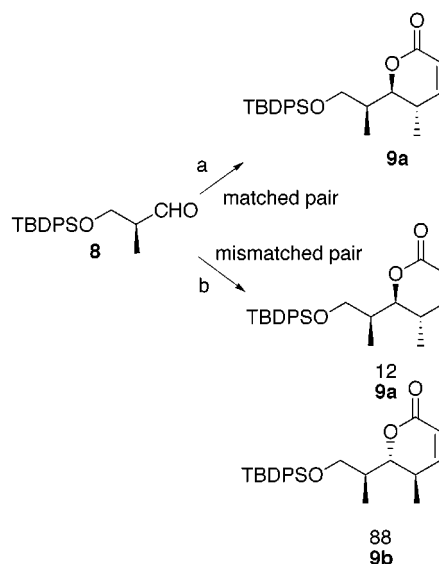
Scheme 4



This methodology was then applied to chiral aldehyde **8**.²⁵ The reaction of chiral aldehyde **8** with the (S)-tolBinap ligand (matched pair) led predominantly to the *syn/anti* lactone **9a**.^{11d,26} The other *anti/anti* lactone **9b**²⁶ could not be observed by ^1H NMR of the crude product, and linear

products were found to be less than 10% of the mixture. After flash chromatography, lactone **9a** was isolated in a gratifying 60% yield. Using the (R)-tolBinap ligand with aldehyde **8** (mismatched pair), an inversion of the diastereoselectivity could be observed: a 9/1 mixture of lactones **9b**²⁷ and **9a** was obtained (the amount of linear products was again found to be less than 10%). After purification by flash chromatography, the mixture of lactones was isolated in 55% yield.

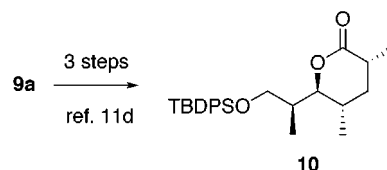
Scheme 5^a



^a (a) $\text{CuF}\cdot(\text{S})\text{-tolBinap}$, 10%, rt, 60%; (b) $\text{CuF}\cdot(\text{R})\text{-tolBinap}$, 10%, rt, 55%.

According to the procedures described by Cossy,^{11d} the lactone **9a** was further transformed in three steps to the Prelog-Djerassi lactone **10**. This procedure constitutes a straightforward (four steps from aldehyde **8**) catalytic asymmetric access to the Prelog-Djerassi lactone (Scheme 6).

Scheme 6



In conclusion, the catalytic asymmetric vinylogous aldol of γ -substituted dienolate constitutes a valuable one-step route to α,β unsaturated lactones, as illustrated by the synthesis of the Prelog-Djerassi lactone. Further optimization

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and applications of this reaction to the synthesis of natural products are currently under investigation.

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Supporting Information Available: Experimental procedures and characterization data for all new compounds. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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