ISSN 1070-4280, Russian Journal of Organic Chemistry, 2009, Vol. 45, No. 12, pp. 1790–1794. © Pleiades Publishing, Ltd., 2009. Original Russian Text © N.V. Poplevina, A.P. Kriven'ko, O.A. Shchelochkova, A.G. Golikov, S.F. Solodovnikov, 2009, published in Zhurnal Organicheskoi Khimii, 2009, Vol. 45, No. 12, pp. 1796–1799.

Synthesis and Structure of (Thio)semicarbazonocyclohexanedicarboxylates. Crystalline and Molecular Structure of Diethyl 4-Hydroxy-4-methyl-2-phenyl-6-thiosemicarbazonocyclohexane-1,3-dicarboxylate

N. V. Poplevina^a, A. P. Kriven'ko^a, O. A. Shchelochkova^b, A. G. Golikov^a, and S. F. Solodovnikov^c

^a Chernyshevskii Saratov State University, ul. Astrakhanskaya 83-1, Saratov, 410012 Russia e-mail: PoplevinaNV@mail.ru

^b Saratov State Medical University, Saratov, Russia

^c Nikolaev Institute of Inorganic Chemistry, Siberian Division, Russian Academy of Sciences, Novosibirsk, Russia

Received April 2, 2009

Abstract—Diethyl 2-aryl-4-hydroxy-4-methyl-6-(thio)semicarbazonocyclohexane-1,3-dicarboxylates were synthesized, and their structure was determined by ¹H and ¹³C NMR spectroscopy, including the HSQC technique. The molecular and crystalline structure of diethyl 4-hydroxy-4-methyl-6-thiosemicarbazono-2-phenylcyclohexane-1,3-dicarboxylate was determined by X-ray analysis.

DOI: 10.1134/S1070428009120070

Polycarbonyl compounds of the dialkyl 2-aryl-4hydroxy-4-methyl-6-oxocyclohexane-1,3-dicarboxylate series (β -keto esters) are used as model structures for stereochemical studies and keto–enol transformations, as well as intermediate products in the synthesis of carbo- and heterocyclic systems [1–3]. Their reactions with nitrogen-centered nucleophiles have been well documented; depending on the nucleophile structure, these reactions lead to the formation of enamines, oximes, fused heterocycles, and heterocyclic spiro compounds whose structure was determined by spectral methods and X-ray analysis [4–7].

Reactions of such β -keto esters with thiosemicarbazides have been studied in the recent years [3, 8]; however, the available spectral data (IR, ¹H NMR) are clearly insufficient to unambiguously assign the struc-



Ia, IIa, IIb, Ar = Ph; Ib, IIc, Ar = 4-MeOC₆H₄; Ic, IId, Ar = 3-O₂NC₆H₄; Id, IIe, Ar = 4-HO-3-MeOC₆H₃; Ie, IIf, Ar = 2-thienyl; X = S (a), O (b-f).



 $\begin{array}{c}
2.0 \\
2.5 \\
3.0 \\
3.5 \\
4.0 \\
60 \\
55 \\
50 \\
45 \\
40 \\
35 \\
35 \\
40 \\
55 \\
50 \\
45 \\
40 \\
35 \\
30 \\
25 \\
20 \\
5_{\rm C}, ppm
\end{array}$

Fig. 1. A fragment of the HSQC spectrum of diethyl 4-hydroxy-4-methyl-2-phenyl-6-thiosemicarbazonocyclohexane-1,3-dicarboxylate (IIa).

ture of the reaction products (thiosemicarbazone A, thiosemicarbazide B, or spirane C). Therefore, additional studies using ¹³C NMR spectroscopy (including double resonance techniques) and X-ray analysis are necessary.

δ, ppm

1.0

1.5

In the present article we present new data on the structure of the reaction product of diethyl 4-hydroxy-4-methyl-6-oxo-2-phenylcyclohexane-1,3-dicarboxylate (**Ia**) with thiosemicarbazide, as well as on reactions of β -keto esters **I** with semicarbazide, which were not reported previously. Diethyl 4-hydroxy-4-methyl-6-thiosemicarbazono-2-phenylcyclohexane-1,3-dicarboxylate (**IIa**) was obtained in 79% yield by reaction of keto diester **Ia** with thiosemicarbazide according to the known procedure [8] (heating in boiling ethanol using 1.5 equiv of the nucleophile), and its structure was determined on the basis of the ¹H and ¹³C NMR, HSQC, and X-ray diffraction data.

The ¹H NMR spectrum of **Ha** (DMSO- d_6) contained singlets from three NH protons at δ 6.71, 8.20, and 10.4 ppm and hydroxy proton at δ 4.61 ppm; methylene protons on C⁵ resonated as doublets at δ 2.26 and 3.22 ppm with a coupling constant ²J of 14 Hz, and the 2-H signal was a doublet at δ 3.05 ppm with a coupling constant J of 12 Hz; and the 1-H and 3-H protons together with OCH₂ protons in the ester moiety give rise to a multiplet at δ 3.61–3.75 ppm. The presence of a signal from 1-H rules out one of the possible structures, thiosemicarbazide **B**, but we still cannot distinguish between thiosemicarbazone **A** and spiro structure **C**. Valuable information was obtained from the ¹³C NMR spectrum (DMSO-*d*₆) which contained 10 signals from *sp*³-hybridized carbon atoms and one signal assignable to *sp*²-carbon atom (C⁶, $\delta_{\rm C}$ 151 ppm); this spectral pattern is consistent with thiosemicarbazone structure **A** and excludes isomeric structures **B** and **C** where 9 and 11 *sp*³-carbon atoms are present, respectively. Detailed signal assignment was made on the basis of the two-dimensional HSQC spectrum (Fig. 1). It contained the following cross peaks: 1-H/C¹ (δ 3.63/ $\delta_{\rm C}$ 56.3 ppm), 2-H/C² (3.05/57.1), 3-H/C³ (3.73/44.6), 5-H_{ax}/C⁵ (2.26/41.0),



Fig. 2. Structure of the molecule of diethyl 4-hydroxy-4methyl-2-phenyl-6-thiosemicarbazonocyclohexane-1,3-dicarboxylate (**IIa**) according to the X-ray diffraction data.

and $5-H_{eq}/C^5$ (3.22/41.0). Thus the above data indicated that compound **IIa** in DMSO-*d*₆ has thiosemicarbazone structure **A**.

Thiosemicarbazone **IIa** crystallized to form welldefined crystals, and its molecular conformation was determined on the basis of the X-ray diffraction data (Fig. 2). The C⁵ atom (for atom numbering, see Fig. 2) has sp^2 -hybridization: the bond angle N¹C⁵C⁴ is 116°, and the C⁵–N¹ bond length is 1.28 Å. The thiosemicarbazone fragment is almost planar: the torsion angle C⁵N¹N²C²⁰ is –178.2°. The cyclohexane ring adopts a distorted *chair* conformation, and the torsion angles C⁶C¹C²C³ (-61.6°) and C¹C²C³C⁴ (61.4°) approach those in analogous cyclohexane conformer.

All substituents in the cyclohexane ring, except for the hydroxy group, occupy equatorial positions. The hydrogen atoms on C² and C³, as well as on C³ and C⁴, occupy axial positions and are arranged *trans* with respect to each others. This follows from the torsion angles between the corresponding ester fragment and phenyl substituent: C⁸C²C³C¹¹ –51.8°, C¹¹C³C⁴C¹⁷ 56.5°. Molecules **IIa** in crystal are characterized by weak intramolecular hydrogen bond O¹–H¹···O² (O²···H¹ 2.47, O¹···O² 3.09 Å) and intermolecular hydrogen bonds N²–H²···O^{2'} (O^{2'}···H² 2.25, N²···O^{2'} 3.08 Å) and N³–H³···O^{4'} (O^{4'}····H³ 2.23, N³···O^{4'} 3.03 Å). Intermolecular hydrogen bonds give rise to a three-dimensional network. Thus compound **IIa** has thiosemicarbazone structure **A** both in solution in DMSO-d₆ and in the crystalline state.

The reactions of β -keto esters **Ia–Ie** with semicarbazide were carried out following a modified procedure. Taking into account instability of semicarbazide as free base, it was used as hydrochloride, and the free base was generated by adding potassium hydroxide. As a result, we isolated the corresponding semicarbazones **IIb–IIe** in high yield (81–90%).

The spectral parameters of semicarbazones **IIb–IIe** were similar to those of thiosemicarbazone **IIa**. Their IR spectra contained absorption bands at 1703–1726 cm⁻¹, belonging to stretching vibrations of unconjugated ester carbonyl groups, two bands due to primary amino group (3082–3206 cm⁻¹), and a band corresponding to secondary amino group (3314–3364 cm⁻¹). In the ¹H NMR spectra of compounds **IIb–IIe** in DMSO-*d*₆ we observed doublets from axial and equatorial qprotons on C⁵ (δ 2.11–2.22 and 2.49–3.11 ppm, *J* = 14–15 Hz), signals from 2-H (δ 2.90–3.17 ppm), 1-H, and 3-H (δ 3.49–3.83 and 3.49–3.80 ppm), a singlet from the NH proton (δ 9.21–

9.25 ppm), a diffuse doublet from the NH₂ group (δ 5.53–6.28 ppm), and a singlet from the hydroxy proton (δ 4.34–4.55 ppm). Compound **IIb** displayed in the ¹³C NMR spectrum (DMSO-*d*₆) 10 signals from *sp*³-hybridized carbon atoms. The C⁶ atom resonated at $\delta_{\rm C}$ 147 ppm. The HSQC spectrum of **IIb** (DMSO-*d*₆) revealed the following cross peaks formed by directly coupled ¹H and ¹³C nuclei: 1-H/C¹ (δ 3.64/ $\delta_{\rm C}$ 56.1 ppm), 2-H/C² (3.00/57.2), 3-H/C³ (3.64/44.8), 5-H_{*ax*}/C⁵ (2.16/40.0), 5-H_{*eq*}/C⁵ (3.07/40.0).

We can conclude that products of the reactions of β -keto diesters I with thiosemicarbazide and semicarbazide have (thio)semicarbazone structure.

EXPERIMENTAL

The IR spectra were recorded in KBr on an FSM-1201 spectrometer with Fourier transform. The ¹H and ¹³C NMR spectra, and two-dimensional HSQC spectra were recorded on Bruker AM-200 (200 MHz), Bruker MSL-400 (400 MHz), and Varian 400 spectrometers using DMSO- d_6 as solvent and tetramethylsilane as internal reference.

The X-ray diffraction data for a $0.18 \times 0.14 \times 0.12$ mm single crystal of compound IIa were acquired at room temperature on a Bruker Nonius X8 Apex automatic diffractometer equipped with a two-dimensional CCD detector ($\lambda Mo K_{\alpha}$ irradiation, graphite monochromator, φ -scanning through a step of 0.5°, Bragg angle range $2.50^{\circ} \le \theta \le 27.50^{\circ}$). Total of 16745 reflection intensities were measured over one half of the reciprocal space sphere; among these, 5031 reflections were independent ($R_{int} = 0.0195$). Monoclinic crystal system; $C_{20}H_{27}N_3O_5S$; M 421.51; space group $P2_1/n$; unit cell parameters: a = 9.5131(4), b = 16.8601(6), c =12.1904(4) Å; $\beta = 101.748(1)^\circ$, V = 2195.78(14) Å³; Z = 4; $d_{calc} = 1.275$ g/cm³. No correction for absorption was introduced, taking into account small value of the linear absorption coefficient ($\mu = 0.182 \text{ mm}^{-1}$).

The structure was solved by the direct method using SHELXS-97 software [9] and was refined by full-matrix least-squares procedure in anisotropic approximation for non-hydrogen atoms using SHELXL-97 software package [10]. All hydrogen atoms were localized by the Fourier difference syntheses, and their positions were refined with isotropic thermal parameters. The final divergence factors were R = 0.0450, $wR_2 = 0.1195$ for 3857 reflections with $F \ge 4\sigma(F)$ (with respect to F_{hkl} ; 371 varied parameters) and R = 0.0636, $wR_2 = 0.1379$ for all independent reflections (5031); goodness of fit S = 1.058.

Diethyl 4-hydroxy-4-methyl-2-phenyl-6-thiosemicarbazonocyclohexane-1,3-dicarboxylate (IIa) was synthesized according to the procedure described in [8]. Yield 79%, colorless crystals, mp 160–161°C [8]. Single crystals suitable for X-ray analysis were obtained by slowly cooling a solution of **Ha** in ethanol. ¹H NMR spectrum, δ , ppm: 0.78 t and 0.90 t (3H each, CH_2CH_3 , J = 7.0 Hz), 1.23 s (3H, 4-CH₃), 2.26 d (1H, 5-H_{ax}, J = 14 Hz), 3.05 d (1H, 2-H, J = 12 Hz), 3.22 d $(1H, 5-H_{eq}, J = 14 \text{ Hz}), 3.61-3.75 \text{ m} (4H, 1-H, 3-H)$ OCH₂), 3.79–3.87 m (2H, OCH₂), 4.61 s (1H, OH), 6.71 s (1H, NH), 7.12-7.26 m (5H, Ph), 8.20 s (1H, NH), 10.4 s (1H, NH). ¹³C NMR spectrum, δ_C , ppm: 13.6 and 13.8 (CH₂CH₃), 28.3 (4-CH₃), 44.6 (C³), 55.6 (C^{1}) , 56.6 (C^{2}) , 59.2 and 59.5 (OCH_{2}) , 71.1 (C^{4}) ; 127, 128, 129, 140 (C_{arom}); 151 (C⁶), 169 and 170 (C=O), 179 (C=S). HSQC spectrum, δ , ppm/ δ_{C} , ppm: 0.78/14.0 (CH₂CH₃), 0.90/14.5 (CH₂CH₃), 1.23/28.6 $(4-CH_3)$, 2.26/41.0 $(5-H_{ax}/C^5)$, 3.05/57.1 $(2-H/C^2)$, 3.22/41.0 (5-H_{ea}/C⁵), 3.63/44.6 (3-H/C³), 3.73/56.3 $(H^{1}/C^{1}), 3.63/59.9 (OCH_{2}), 3.73/60.1 (OCH_{2});$ 7.13/127, 7.20/128, 7.23/129 (CH_{arom}).

Diethyl 4-hydroxy-4-methyl-2-phenyl-6-semicarbazonocyclohexane-1,3-dicarboxylate (IIb). A solution of 0.97 g (8.7 mmol) of semicarbazide hydrochloride and 0.49 g (8.7 mmol) of potassium hydroxide in 2 ml of water was added to a solution of 2 g (5.8 mmol) of compound Ia in 50 ml of ethanol, and the mixture was heated for 1 h under reflux. After cooling, the precipitate was filtered off, washed with water, propan-2-ol, and diethyl ether, and dried under reduced pressure. Yield 1.97 g (85%), colorless crystals, mp 201-203°C (from EtOH). IR spectrum, v, cm⁻¹: 3501 (OH); 3353 (NH); 3088, 3183 (NH₂); 1713, 1719 (C=O, ester); 1676 (C=O, amide); 1560 (C=N). ¹H NMR spectrum, δ , ppm: 0.83 t and 0.97 t (3H each, CH_2CH_3 , J = 6.8 Hz), 1.25 s (3H, 4-CH₃), 2.16 d (1H, 5-H_{ax}, J = 14 Hz), 2.99–3.04 m (1H, 2-H), 3.10 d (1H, 5-H_{eq}, J = 15 Hz), 3.62–3.68 m (2H, 1-H, 3-H), 3.75 q $(2H, OCH_2, J = 6.8 Hz), 3.82-3.92 m (2H, OCH_2),$ 4.36 s (1H, OH), 5.53-6.19 br.d (NH₂), 7.13-7.25 m (5H, Ph), 9.25 s (1H, NH). ¹³C NMR spectrum, δ_c , ppm: 13.7 and 14.0 (CH₂CH₃), 28.5 (4-CH₃), 44.4 (\hat{C}^3) , 55.8 (C^1), 56.8 (C^2), 59.3 and 59.5 (OCH₂), 71.0 (C⁴); 127, 128, 129, 141 (C_{arom}); 147 (C⁶), 157 (C=O, amide), 169 and 171 (C=O, ester). HSQC spectrum, δ, ppm/ $\delta_{\rm C}$, ppm: 0.80/14.1 (CH₂CH₃), 0.93/14.5 (CH₂CH₃), 1.20/29.2 (4-CH₃), 2.16/40.0 (5-H_{ax}/C⁵), 3.00/57.2 (2-H/C²), 3.07/40.0 (5-H_{ea}/C⁵), 3.64/44.8 $(3-H/C^3)$, 3.64/56.1 $(1-H/C^1)$, 3.81/59.4 (OCH_2) , 3.82/60.0 (CH₂); 7.16/127, 7.17/128, 7.20/129 (CH_{arom}). Found, %: C 59.38; H 6.72; N 9.76. C₂₀H₂₇N₃O₆. Calculated, %: C59.25; H 6.71; N 10.36.

Compounds **IIc–IIf** were synthesized in a similar way.

Diethyl 4-hydroxy-2-(4-methoxyphenyl)-4-methyl-6-semicarbazonocyclohexane-1,3-dicarboxylate (IIc). Yield 90%, colorless crystals, mp 211–212°C (from EtOH). IR spectrum (KBr), v, cm⁻¹: 3497 (OH); 3363 (NH); 3088, 3182 (NH₂); 1715 (C=O, ester); 1672 (C=O, amide); 1561 (C=N). ¹H NMR spectrum, δ , ppm: 0.87 t and 1.00 t (3H each, CH₂CH₃, *J* = 6.8 Hz), 1.23 s (3H, 4-CH₃), 2.13 d (1H, 5-H_{ax}, *J* = 14 Hz), 2.90–2.95 m (1H, 2-H), 3.08 d (1H, 5-H_{eq}, *J* = 14 Hz), 3.51–3.58 m (2H, 1-H, 3-H), 3.69 s (3H, OCH₃), 3.76 q (2H, OCH₂, *J* = 7.2Hz), 3.82–3.90 m (2H, OCH₂), 4.34 s (1H, OH), 5.66–6.22 br.d (NH₂), 6.75 d and 7.15 d (2H each, C₆H₄), 9.22 s (1H, NH). Found, %: C 58.60; H 6.74; N 9.77. C₂₁H₂₉N₃O₇. Calculated, %: C 58.49; H 7.26; N 9.34.

Diethyl 4-hydroxy-4-methyl-2-(3-nitrophenyl)-6semicarbazonocyclohexane-1,3-dicarboxylate (IId). Yield 89%, colorless crystals, mp 207–210°C (from EtOH). IR spectrum, v, cm⁻¹: 3477 (OH); 3314 (NH); 3150, 3206 (NH₂); 1715–1730 (C=O); 153 (C=N). ¹H NMR spectrum, δ , ppm: 0.85 t (3H, CH₂CH₃, *J* = 6.8 Hz), 0.98 t (3H, CH₂CH₃, *J* = 7.4 Hz), 1.27 s (3H, 4-CH₃), 2.22 d (1H, 5-H_{ax}, *J* = 14 Hz), 3.11 d (1H, 5-H_{eq}, *J* = 14 Hz), 3.14–3.17 m (1H, 2-H), 3.75–3.83 m (3H, 1-H, OCH₂), 3.85–3.93 m (3H, 3-H, OCH₂), 4.55 s (1H, OH), 5.62–6.16 br.d (NH₂); 7.55 t, 7.73 d, 8.04 d, and 8.20 s (4H, C₆H₄); 9.25 s (1H, NH). Found, %: C 53.84; H 5.72; N 12.32. C₂₀H₂₆N₄O₈. Calculated, %: C 55.05; H 5.69; N 12.84.

Diethyl 4-hydroxy-2-(4-hydroxy-3-methoxyphenyl)-4-methyl-6-semicarbazonocyclohexane-1,3dicarboxylate (IIe). Yield 90%, colorless crystals, mp 186–187°C (from EtOH). IR spectrum, v, cm^{-1} : 3480 (OH); 3441 (OH_{arom}); 3320 (NH); 3150, 3206 (NH₂); 1726 (C=O), 1545 (C=N). ¹H NMR spectrum, δ , ppm: 0.89 t and 1.02 t (3H each, CH₂CH₃, J = 6.8 Hz), 1.23 s (3H, 4-CH₃), 2.11 d (1H, 5-H_{ax}, J =15 Hz), 2.95 d (1H, 2-H, J = 10 Hz), 3.07 d (1H, 5-H_{eq}, J = 15 Hz), 3.49–3.59 m (2H, 1-H, 3-H), 3.72 s (3H, OCH₃), 3.79 q (2H, OCH₂, J = 7.2 Hz), 3.86–3.92 m (2H, OCH₂), 4.30 s (1H, OH), 5.58–6.28 br.d (NH₂), 6.56-6.60 m and 6.78 s (3H, H_{arom}), 8.60 s (1H, OH_{arom}), 9.21 s (1H, NH). Found, %: C 55.41; H 6.58; N 9.61. C₂₁H₂₉N₃O₈. Calculated, %: C 55.88; H 6.43; N 9.31.

Diethyl 4-hydroxy-4-methyl-6-semicarbazono-2-(2-thienyl)cyclohexane-1,3-dicarboxylate (IIf). Yield 81%, colorless crystals, mp 194°C (from EtOH). IR spectrum, v, cm⁻¹: 3502 (OH); 3352 (NH); 3082, 3182 (NH₂); 1703, 1722 (C=O, ester); 1680 (C=O, amide); 1557 (C=N). ¹H NMR spectrum, δ , ppm: 0.96 t and 1.06 t (3H each, CH₂CH₃, J = 6.8 Hz), 1.24 s (3H, 4-CH₃), 2.15 d (1H, 5-H_{ax}, J = 14 Hz), 2.95 d (1H, 2-H, J = 12Hz), 3.08 d (1H, 5-H_{eq}, J = 15 Hz), 3.55 d (1H, 1-H, J = 12 Hz), 3.86 q (2H, OCH₂, J = 7.2 Hz), 3.92–3.98 m (3H, 3-H, OCH₂), 4.45 s (1H, OH), 5.58– 6.27 br.d (NH₂), 6.86–6.88 m and 7.24 d (3H, C₄H₃S), 9.24 s (1H, NH). Found, %: C 52.55; H 6.08; N 10.22. C₁₈H₂₅N₃O₆S. Calculated, %: C 52.70; H 5.72; N 10.58.

REFERENCES

- Kriven'ko, A.P. and Sorokin, V.V., Russ. J. Org. Chem., 1999, vol. 35, p. 1097.
- Smirnova, N.S., Plotnikov, O.V., Vinogradova, N.A., Sorokin, V.V., and Kriven'ko, A.P., *Khim.-Farm. Zh.*, 1995, vol. 1, p. 44.

- 3. Zorina, A.A., Cand. Sci. (Chem.) Dissertation, Perm, 2006.
- Sorokin, V.V., Grigor'ev, A.V., Ramazanov, A.K., and Kriven'ko, A.P., *Russ. J. Org. Chem.*, 2000, vol. 36, p. 781.
- Grigor'eva, E.A., Kriven'ko, A.P., Sorokin, V.V., Ramazanov, A.K., and Inozemtseva, O.A., *Izv. Vyssh. Uchebn. Zaved., Ser. Khim. Khim. Tekhnol.*, 2004, vol. 50, p. 131.
- Sorokin, V.V., Grigor'ev, A.V., Ramazanov, A.K., and Kriven'ko, A.P., *Khim. Geterotsikl. Soedin.*, 1999, p. 757.
- Sorokin, V.V., Suponitskii, K.Yu., and Kriven'ko, A.P., *Zh. Strukt. Khim.*, 2006, vol. 47, p. 598.
- Shchelochkova, O.A., Grigor'eva, E.A., and Kriven'ko, A.P., *Izv. Vyssh. Uchebn. Zaved., Ser. Khim. Khim. Tekhnol.*, 2006, vol. 49, p. 139.
- 9. Sheldrick, G.M., Acta Crystallogr., Sect. A, 1990, vol. 46, suppl., p. 467.
- Sheldrick, G.M., SHELX97. Programs for Crystal Structure Analysis (Release 97-2), Göttingen: Univ. of Göttingen, 1997.