Reaction of diimines and azines with diphenylcyclopropenone

Mohsen Abdel-Motaal Gomaa

Chemistry Department, Faculty of Science, Minia University, 61519 El-Minia, Egypt

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1-Cyclohexyl-2-cyclohexylaminomethylene-4,5-diphenyl-1,2-dihydropyrrol-3-one **4a** and 1-aryl-2-arylaminomethylene-4,5-diphenyl-1,2-dihydropyrrol-3-ones **4b,c** as the *E*-form are synthesized by the reaction between N,N'-dicyclohexylethane-1,2-diylidenediamine **2a** and N,N'-diarylethane-1,2-diylidenediamines **2b,c** with diphenylcyclopropenone **1** through a formal [2 + 3] cycloaddition reaction. The structure assignment of **4a** is confirmed on the basis of an X-ray crystal-structure determination. Similarly, diaryl azines **8a–c** react with **1** through a formal [2 + 3] cycloaddition reaction to give the non-isolable product Δ^4 -pyrrolin-3-ones **10a–c** which undergo oxidative rearrangement to afford ultimately the indenone derivatives **9a–c**.

Introduction

The fascinating chemistry of cyclopropenones has attracted the attention of numerous researchers over the past three decades, ^{1,2} with special emphasis on the behaviour of diphenyl-cyclopropenone 1.³ Diphenylcyclopropenone has been found to react with a wide range of imines, amidines, guanidines and other compounds containing the C=N moiety to form azacyclopentenones (pyrrolinones) *via* a formal [2 + 3] cycloaddition reaction (Fig. 1).⁴⁻⁹ Similarly, 2-amino-1-azetines

$$R^{1}$$
 R^{2}
 R^{3}
 R^{3}
 R^{1}
 R^{1}
 R^{2}
 R^{3}

$$R^{1} = H, Ph, NR_{2}; R^{2} = Ar, NR_{2}; R^{3} = H, Alkyl$$

Fig. 1

reacted with 1 to afford azeto[1,2-a]pyrroles.¹⁰ Diphenylcyclopropenone 1 was used to synthesize the annelated heterocycles by its reaction with cyclic amidines to give rise to bi- and tricyclic pyrrolinones (Fig. 2).¹¹ Moreover, thiazolo[3,2-a]pyrimidones were synthesized from the reaction of diphenylcyclopropenone 1 with 2-aminothiazoles.¹² Similarly, tetrasubstituted 4(3*H*)-pyridones were produced from the reaction of 1 with 3-amino-2,2-dimethyl-2*H*-azirines (Fig. 3).¹³

As part of our programme to develop efficient procedures for the synthesis of heterocyclic compounds via the nucleophilic reactions of an amidine group (-HN-CR=N-) with π -deficient compounds ¹⁴⁻¹⁷ we have synthesized trisubstituted pyridin-2(1H)-ones from the reaction of N^1,N^2 -diarylacetamidines with π -deficient dibenzoylacetylene. ¹⁸ Recently we also prepared 1,4,5,6-tetrahydropyridines by reaction of N^1,N^2 -diarylacetamidines with benzylidenemalononitriles. ¹⁹

Results and discussion

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Herein I report the reaction of the diimines 2a-c and the aromatic azines 8a-c with the selected π -deficient diphenyl-cyclopropenone 1.

First we undertook to investigate the reaction of N,N'-dicyclohexyl- 2a and diaryl-ethane-1,2-diylidenediamines 2b,c with diphenylcyclopropenone 1. Thus diimines 2a-c and the

cyclopropenone 1 were heated in ethanol for 2–5 h to give the pyrrolinone derivatives 4a–c in 42–67% yield (Scheme 1). 13 C NMR spectra showed a characteristic peak at $\delta_{\rm C}$ 175.96–177.33 for the carbonyl groups, and the aliphatic carbon atoms (of 4a) of the cyclohexyl rings appeared as two peaks at $\delta_{\rm C}$ 56.05 and 56.33 for the cyclohexyl tertiary carbon atoms (CH) adjacent to the nitrogen atoms. A 13 C DEPT spectrum of 4a showed six peaks with a negative amplitude that were assigned for the cyclohexyl-CH₂ carbon atoms. 1 H NMR spectroscopy showed signals between δ 7.42–7.98 that were assigned to the NH group. IR spectroscopy showed absorptions at 3380–3410 cm⁻¹

Fig. 3

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Ph NHR Ph NHR Ph NHR Ph NHR R S S 4a-c
$$\begin{array}{c} Ph & O \\ Ph & N \\ R & S \end{array}$$
 $\begin{array}{c} Ph & O \\ Ph & N \\ R & H \end{array}$ $\begin{array}{c} Ph & O \\ R & H \end{array}$ $\begin{array}{c} Ph & O \\ R & H \end{array}$ $\begin{array}{c} Ph & O \\ R & H \end{array}$ $\begin{array}{c} Ph & O \\ R & H \end{array}$ $\begin{array}{c} Ph & O \\ R & H \end{array}$ $\begin{array}{c} Ph & O \\ R & H \end{array}$ $\begin{array}{c} Ph & O \\ R & NR \end{array}$ $\begin{array}{c} Ph & O \\ R & NR \end{array}$ $\begin{array}{c} Aa-c \\ Aa-c \end{array}$ $\begin{array}{c} Aa-c \\$

Scheme 1

for the amino groups and at 1645–1650 cm⁻¹ for the C=O groups for structures **4**. We expected to obtain the structures **3a**–**c** but, in the ¹H NMR spectra, the H-2 methine hydrogen atom was not observed. To distinguish between arrangements of the tautomeric structure of **3a**–**c**, *i.e.* **4a**–**c**, and an alternative, *e.g.*, **5** an X-ray crystal-structure analysis of **4a** was carried out (Fig. 4). It clearly demonstrates the five-membered

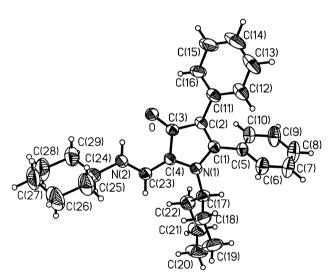


Fig. 4 Molecular structure of 4a in the crystal phase (the crystallographic numbering does not reflect the systematic numbering).

structure and shows that the adduct is a tautomer (enamine-like structure) of 3a-c in the E-configuration.

It is clear that the pyrrolinone structures **3a-c** tautomerize to the enaminone structures **4a-c** which are more stable, having extended conjugation and hydrogen bonding.

Formation of Δ^4 -pyrrolinones $\mathbf{4a-c}$ may be rationalized as an initial attack of one of the imino nitrogen atoms of $\mathbf{2a-c}$ on the electrophilic carbon C-2 or C-3 of cyclopropenone 1 giving the immoniumbetaine $\mathbf{6a-c}$, followed by ring opening to afford the ketene intermediate $\mathbf{7a-c}$. Subsequent cyclization by attack of the ketene onto the iminium function (electrocyclization) leads to imines $\mathbf{3a-c}$ and hence to isomeric enaminones $\mathbf{4a-c}$ (Scheme 2).

The reaction of dienophilic compounds with diaryl azines represents a valuable enlargement to synthetic heterocyclic chemistry. Aldazines derived from aromatic aldehydes undergo the so-called criss-cross addition with two mol equiv.

Scheme 2

of dienophilic compounds to yield 1,5-diazabicyclo[3.3.0]-octane derivatives (derivatives of pyrazolo[1,2-a]pyrazoles), while the addition of equimolar amounts of the same reagents leads to the formation of the corresponding azomethine imines and pyrazoles. Recently, aromatic azines have been shown to react with dibenzoylacetylene to afford unexpected [4 + 2] cycloadducts such as 4,5-dibenzoyl-3,6-diarylpyridazines.²²

From the above findings it may be appropriate to investigate the reactivity of diarylaldazines **8a-c** towards the diphenyl-cyclopropenone **1** to compare their reactivity with **2a-c**. Thus two mol equivalents of **1** together with one mol equivalent of an aldazine **8a-c** were kept in ethanol at reflux for 12 h to afford solid inden-1-one derivatives **9a-c** (Scheme 3). The structure of

Ar Ph Ph Ph Ar COAr

a, Ar = 4-CH
$$_{3}C_{6}H_{4}$$
 9a-c

b, Ar = 4-CH $_{3}OC_{6}H_{4}$

c, Ar = 4-(CH $_{3})_{2}NC_{6}H_{4}$

products **9a–c** was assigned on the basis of their spectral data. ¹H NMR spectra showed only one imine-proton (CH=N), at δ 7.97–8.03. ¹³C NMR spectra showed a signal between $\delta_{\rm C}$ 195.80–195.94 that was assigned for the carbonyl group of the indenone system, and another signal at higher field between $\delta_{\rm C}$ 170.44 and 170.72 was assigned to the amidic carbonyl group, in addition to a signal between $\delta_{\rm C}$ 144.22–145.28 that was attributed to the imine carbon atom.

Scheme 3

The formation of products 9a-c was rationalized as a formal [2+3] cycloaddition reaction between 1 and 8a-c to give the Δ^4 -pyrrolin-3-ones 10a-c, which undergo an oxidative rearrangement to afford ultimately the indenone derivatives 9a-c

Scheme 4

c *via* the intermediates 11a-c-13a-c (Scheme 4). Eicher *et al.* have isolated indenones as a result of oxidative processes during the reaction of benzylidene azines with $1.^{23}$ Thus the aldazines 8a-c react with the dienophiles like 1 as do imines *via* a [2 + 3] cycloaddition reaction but not as azines *via* the criss-cross addition reaction.

Conclusions

The present investigation clearly indicates that diaryl azines react with diphenylcyclopropenone through a formal [3 + 2] cycloaddition reaction giving the non-isolable Δ^4 -pyrrolin-3-ones 10a-c products, which undergo oxidative processes to afford ultimately the indenone derivatives 9a-c. Similarly, diimines react with diphenylcyclopropenone to afford the Δ^4 -pyrrolin-3-one derivatives which are more stable against air oxidation due to high conjugation.

Experimental

Uncorrected melting points were determined on a Griffin & George apparatus. Elemental analyses were carried out by the Microanalysis Center at Cairo University. IR spectra (KBr) were recorded on a Shimadzu 470 spectrophotometer. 400 MHz ¹H NMR and 100 MHz ¹³C NMR spectra were observed on a Bruker AM 400 spectrometer. Mass spectra (70 eV, electron-impact mode) were recorded on a JEOL JMS600 instrument. Preparative thin-layer chromatography (PLC) used air-dried 1.0 mm thick layers of slurry-applied silica gel Merck PF₂₅₄ on 48 cm wide and 20 cm high glass plates and toluene–ethyl acetate (2 : 1) as developing solvent. Zones were detected by the colour or by quenching of indicator fluorescence upon exposure to 254 nm light and eluted with acetone.

Starting materials

N,N'-Diarylethane-1,2-diylidenediamines **2b,c** and N,N'-dicyclohexylethane-1,2-diylidenediamine **2a**^{24,25} and diaryl azines **8a**- \mathbf{c}^{26} were prepared as reported.

Reaction of N,N'-dicyclohexylethane-1,2-diylidenediamine 2a and N,N'-diarylethane-1,2-diylidenediamines 2b,c with diphenyl-cyclopropenone 1 (general procedure)

A solution of 1 (206 mg, 1.0 mmol) in ethanol (10 cm³) was added to solutions of a diimine 2a–c (1.0 mmol) in ethanol (20 cm³). The mixture was heated to reflux temperature for 2–5 h. After this period, orange crystals of the corresponding enaminone 4a–c precipitated, which were filtered off and recrystallized from ethanol and identified as follows.

(*E*)-1-Cyclohexyl-2-cyclohexylaminomethylene-4,5-diphenyl-1,2-dihydropyrrol-3-one [(*E*)-4a]. Yield 180 mg (42%), mp 259–260 °C as orange crystals (from ethanol); ^1H NMR (d₆-DMSO) δ 0.95–1.92 (m, 20 H, cyclohexyl-CH₂), 3.52–3.54 (m, 2 H, cyclohexyl-CH), 6.99 (s, 1 H, CH), several multiplets at 7.03, 7.13, 7.27 and 7.27 (10 H, ArH), 7.98 (s, 1 H, NH); ^{13}C NMR (d₆-DMSO) δ 24.10, 24.35, 24.97, 26.10, 31.63, 33.97 (cyclohexyl-CH₂), 56.05 and 56.33 (cyclohexyl-CH), 111.42, 114.05, 124.04, 127.34, 128.24, 127.66, 128.96, 129.64 (all Ar–CH and vinylic CH), 129.18, 132.56, 134.20, 140.99, 148.82, 176.39; IR (KBr) 3380 (NH), 1645 (CO) cm⁻¹; MS (70 eV) *mlz* (%) 426 (M⁺, 100), 425 (15), 409 (5), 383 (7), 369 (5), 344 (51), 327 (57), 261 (19), 248 (14), 178 (15), 55 (24), 41 (14), [Calc. for C₂₉H₃₄N₂O (426.6): C, 81.65; H, 8.03; N, 6.57. Found: C, 81.50; H, 7.95; N, 6.51%].

(*E*)-1-(4-Methylphenyl)-2-(4-methylphenylaminomethylene)-4,5-diphenyl-1,2-dihydropyrrol-3-one [(*E*)-4b]. Yield 264 mg (60%), mp 208–210 °C as orange crystals (from ethanol); ^1H NMR (d₆-DMSO) δ 2.22 (s, 3 H, CH₃), 2.25 (s, 3 H, CH₃), 6.95 (s, 1 H, CH), several multiplets at 7.07, 7.13, 7.17 and 7.23 (18 H, ArH), 7.46 (s, 1 H, NH); ^{13}C NMR (d₆-DMSO) δ 20.33, 20.63 (2 × CH₃), 116.40, 125.18, 127.72, 128.39, 128.46, 129.06, 129.92, 130.14, 130.28 and 131.0 (all Ar–CH and vinylic CH), 113.39, 120.86, 128.5, 131.15, 132.71, 134.20, 137.17, 137.95, 152.14, 177.33; IR (KBr) 3400 (NH), 1650 (CO) cm⁻¹; MS (70 eV) mlz (%) 442 (M⁺, 100), 336 (7), 221 (6), 212 (4), 178 (5), 118 (9), 91 (11), 65 (4) [Calc. for C₃₁H₂₆N₂O (442.6): C, 84.13; H, 5.92; N, 6.33. Found: C, 84.31; H, 5.83; N, 6.30%].

Reaction of diarylaldazines 8a-c with diphenylcyclopropenone 1 (general procedure)

A solution of 1 (206 mg, 1.0 mmol) in ethanol (10 cm³) was added to a solution of an aldazines 8a–c (0.5 mmol) in ethanol (20 cm³) or toluene (30 cm³). The mixture was heated to reflux temperature for 12 h before being concentrated and the residue was subjected to PLC using toluene–ethyl acetate (2:1) as the developing solvent to give one or two main zones. The faster

moving one contained **9a**, **9b** or **9c** respectively, while the more slowly moving one contained the unchanged cyclopropenone **1**. The zonal components were extracted, crystallized, and identified as follows.

3-[1'-(4-Methylbenzoyl)-2'-(4-methylbenzylidene)hydrazino]- 2-phenylinden-1-one 9a. Yield 120 mg (52%), mp 118–119 °C as yellow crystals (from cyclohexane); 1 H NMR (CDCl₃) δ 2.34 (s, 3 H, CH₃), 2.41 (s, 3 H, CH₃), several multiplets at 7.11, 7.16, 7.24, 7.49, 7.53 and 7.61 (17 H, ArH), 8.03 (s, 1 H, CH=N); 13 C NMR (CDCl₃) δ 21.12, 21.35, 91.75, 108.75, 125.59, 126.07, 126.87, 127.84, 127.91, 129.07, 129.14, 129.36, 129.81, 130.52, 130.60, 131.60, 132.17, 138.69, 139.51, 144.27, 170.65, 195.96; IR (KBr) 1690 (CO), 1645 (CO) cm⁻¹; MS (70 eV) mlz (%) 457 (1), 456 (M⁺, 48), 440 (14), 429 (4), 412 (2), 341 (72), 337 (58), 325 (44), 310 (65), 222 (68), 191 (28), 177 (82), 164 (18), 120 (37), 118 (100), 105 (37), 91 (87) [Calc. for C₃₁H₂₄N₂O₂ (456.6): C, 81.55; H, 5.30; N, 6.14. Found: C, 81.42; H, 5.18; N, 6.10%].

3-[1'-(4-Methoxybenzoyl)-2'-(4-methoxybenzylidene)-hydrazino]-2-phenylinden-1-one 9b. Yield 118 mg (48%), mp 85–89 °C as yellow crystals (from cyclohexane); ¹H NMR (CDCl₃) δ 3.75 (s, 3 H, OCH₃), 3.77 (s, 3 H, OCH₃), several multiplets at 6.75, 6.88, 7.09, 7.11, 7.24, 7.35, 7.49, 7.53, 7.58 and 7.85 (17 H, ArH), 7.99 (s, 1 H, CH=N); ¹³C NMR (CDCl₃) δ 55.26, 55.27, 91.52, 108.55, 113.89, 114.54, 126.15, 126.48, 127.12, 127.63, 127.92, 127.99, 128.43, 129.37, 130.58, 131.53, 132.73, 144.22, 158.96, 160.67, 170.72, 195.80; IR (KBr) 1690 (CO), 1645 (CO) cm⁻¹; MS (70 eV) m/z (%) 489 (1), 488 (M⁺, 20), 467 (14), 354 (18), 340 (44), 326 (65), 223 (68), 209 (12), 196 (20), 152 (82), 150 (18), 137 (100), 123 (70), 121 (48), 105 (33), 91 (77) [Calc. for C₃₁H₂₄N₂O₄ (488.6): C, 76.21; H, 4.95; N, 5.74. Found: C, 76.11; H, 4.89; N, 5.63%].

3-{1'-[4-(Dimethylamino)benzoyl]-2'-[4-(dimethylamino)benzylidene]hydrazino}-2-phenylinden-1-one 9c. Yield 130 mg (50%), mp 176–178 °C as orange crystals (from cyclohexane); ¹H NMR (CDCl₃) δ 2.91 (s, 6H, NCH₃), 2.92 (s, 6H, NCH₃), several multiplets at 6.52, 6.55, 6.67, 6.69, 7.11, 7.22, 7.26, 7.35, 7.44, 7.58, 7.60 (17 H, ArH), 7.97 (s, 1 H, CH=N); ¹³C NMR (CDCl₃) δ 40.21, 40.33 (4 × NCH₃), 91.88, 107.81, 111.67, 112.70, 121.70, 123.07, 125.81, 126.68, 127.83, 127.88, 128.34, 129.08, 129.86, 130.30, 130.68, 131.02, 145.28, 150.62, 151.23, 170.44, 195.94; IR (KBr) 1690 (CO), 1645 (CO) cm⁻¹; MS (70 eV) m/z (%) 514 (M+, 20), 499 (2), 484 (2), 471 (8), 425 (11), 365 (80), 338 (65), 322 (18), 294 (18), 282 (11), 264 (11), 263 (27), 193 (13), 178 (25), 164 (32), 149 (100), 134 (55), 121 (32), 105 (50), 91 (17), 84 (54), 77 (49) [Calc. for $C_{33}H_{30}N_4O_2$ (514.6): C, 77.02; H, 5.88; N, 10.89. Found: C, 76.91; H, 5.79; N, 10.73%].

Crystal structure determination of compound 4a

Crystal data. C₂₉H₃₄N₂O, M = 426.60. Trigonal, a = 32.814(2), c = 12.309(1) Å, $a = 90.00^\circ$, T = 293 K, V = 11 478.1(14) ų, Mo-Kα radiation ($\lambda = 0.710$ 73 Å), space group R-3, Z = 18, $D_x = 1.111$ Mg m $^{-3}$. Dark orange prisms. Crystal dimensions: $0.50 \times 0.43 \times 0.25$ mm, μ (Mo-Kα) = 0.067 mm $^{-1}$, F(000) = 4140.

Data collection and processing. Nonius-Kappa CCD diffractometer, graphite-monochromator Mo-Kα (radiation; 6293 reflections measured ($\theta = 24.18^{\circ}$), 4039 unique) merging at R = 0.0317, linear and approx. isotropic crystal decay, ca.23% corrected during processing. Refinements on F^2 , R1 = 0.0769 for 2411 reflections with $I > 2\sigma(I)$, 291 variables, wR2 = 0.2018 for all unique data, SHELXL program package.

Atomic coordinates, bond lengths and angles, and parameters have been deposited at the Cambridge Crystallographic Data Centre. †

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