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A FACILE SYNTHESIS OF 2,3-DISUBSTITUTED PYRIDO[2,3-*h*]QUINAZOLIN-4(3H)-ONES

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ABSTRACT

Synthesis of 2-substituted-4H-Pyrido[2,3-*h*]benzoxazin-4-ones (2) from 5-Aminoquinoline-6-carboxylic acid (1) and their conversion to 2,3-disubstituted Pyrido[2,3-*h*]quinazolin-4(3H)-ones(3) is reported.

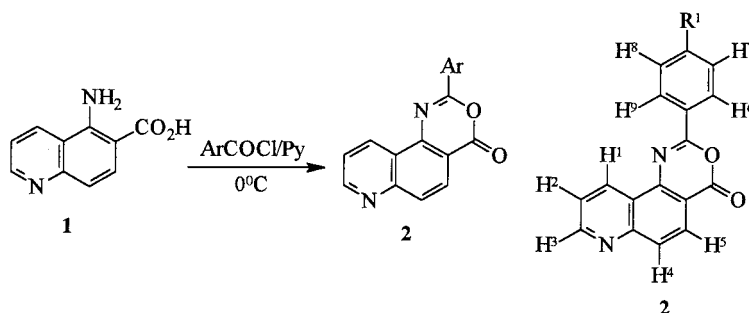
5-Aminoquinoline-6-carboxylic acid,¹ in spite of its wide potential has been scarcely used in the synthesis of heterocyclic synthesis. It has been successfully used by us² for the preparation of 2-methyl-4H-pyrido[2,3-*h*]quinazolin-4(3H)-one. Here in we report the synthesis of 2-aryl-4H-pyrido[2,3-*h*]benzoxazin-4-ones (2a–d) and 2,3-disubstituted pyrido[2,3-*h*]quinazolin-4(3H)-ones (3a–s).

The reaction of 5-aminoquinoline-6-carboxylic acid (1) with 4-methylbenzoyl chloride in pyridine at 0°C gave a crystalline compound in high yield. The mass spectrum of the compound revealed the molecular ion

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at m/z 288 indicating to be a 1:1 product formed by the elimination of hydrogen chloride and water. The IR spectrum (KBr) showed the absence of NH and presence of a δ -lactonic carbonyl at 1758 cm^{-1} . The $^1\text{H-NMR}$ spectrum (DMSO-d_6) exhibited signals at δ 2.5 (s, 3H, CH_3), 7.4 (d, 2H, Ar-H of H^7 , H^8), 7.7 (m, 1H, Ar-H of H^4), 8.2 (d, 1H, Ar-H of H^5), 8.4 (m, 3H, Ar-H of H^2 , H^6 , H^9), 9.2 (d, 1H, Ar-H of H^1), 9.4 (d, 1H, Ar-H of H^3). Based on the spectral and analytical data the compound has been characterised as 2-(4-methylphenyl)-4H-pyrido[2,3-*h*]3,1-benzoxazin-4-one (2c).



This reaction was extended to three other aryl chlorides and the products obtained were characterised as 2-aryl-4H-pyrido[2,3-*h*]3,1-benzoxazin-4-ones (2a–d) (Table 1) by analogy and on the basis of spectral and analytical data (Tables 1–3). Attempts to isolate *N*-acylamino acid in the reaction of 1 with aryl chlorides were not successful.

Compounds 2 exhibited intense molecular ions in their mass spectra. The mass spectral fragmentation is characterised by the presence of ions corresponding to the loss of Ar, CO_2 . In addition, Ar and ArCO fragment ions have been observed.

Compounds 2 possess a labile lactone ring which can readily undergo reaction with nucleophiles involving replacement of oxygen in the ring. Thus reaction of 2a with 4-methylaniline in acetic acid afforded a crystalline compound, identified as 2-phenyl-3-(4-methylphenyl)pyrido[2,3-*h*]quinoxalin-4(3H)-one (3c) on the basis of mass (M^+ at m/z 363), IR (KBr, $\nu\text{C=O}$ at 1670 cm^{-1}) and $^1\text{H-NMR}$ (DMSO-d_6) spectral data [δ 2.4 (s, 3H, CH_3), 7.0–7.4 (m, 9H, Ar-H of C-aryl and N-aryl), 7.6 (m, 1H, Ar-H of H^2), 8.1 (d, 1H, Ar-H of H^4), 8.5 (d, 1H, Ar-H of H^5), 9.1 (d, 1H, Ar-H of H^1) and 9.3 (d, 1H, Ar-H of H^3)]. The reaction of 2a was extended to aniline, 4-nitroaniline and 2-methylanilines and the corresponding 3 (Table 1) were obtained in good yields.



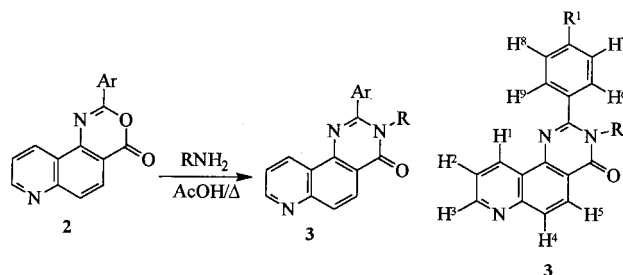


Table 1. Characterisation Data of 2-Substituted-4H-pyrido[2,3-*h*]-3,1-benzoxazin-4-ones (2) and 2,3-Disubstituted Pyrido[2,3-*h*]-quinazolin-4-(3H)-ones(3)

Compound	Ar	R	M.P. (°C)	Yield (%)
2a	C ₆ H ₅	—	204	80
2b	4-Cl-C ₆ H ₄	—	220	75
2c	4-CH ₃ -C ₆ H ₄	—	215	72
2d	4-OCH ₃ -C ₆ H ₄	—	258	80
3a	C ₆ H ₅	C ₆ H ₅	> 300	82
3b	C ₆ H ₅	4-NO ₂ -C ₆ H ₄	> 300	78
3c	C ₆ H ₅	4-CH ₃ -C ₆ H ₄	260	75
3d	C ₆ H ₅	2-CH ₃ -C ₆ H ₄	254	80
3e	4-Cl-C ₆ H ₄	C ₆ H ₅	> 300	80
3f	4-Cl-C ₆ H ₄	4-NO ₂ -C ₆ H ₄	278	75
3g	4-Cl-C ₆ H ₄	4-CH ₃ -C ₆ H ₄	> 300	78
3h	4-Cl-C ₆ H ₄	2-CH ₃ -C ₆ H ₄	288	75
3i	4-CH ₃ -C ₆ H ₄	C ₆ H ₅	> 300	82
3j	4-CH ₃ -C ₆ H ₄	4-NO ₂ -C ₆ H ₄	195	80
3k	4-CH ₃ -C ₆ H ₄	4-CH ₃ -C ₆ H ₄	> 300	78
3l	4-CH ₃ -C ₆ H ₄	2-CH ₃ -C ₆ H ₄	238	74
3m	4-OCH ₃ -C ₆ H ₄	C ₆ H ₅	> 300	76
3n	4-OCH ₃ -C ₆ H ₄	4-NO ₂ -C ₆ H ₄	188	80
3o	4-OCH ₃ -C ₆ H ₄	4-CH ₃ -C ₆ H ₄	140	72
3p	4-OCH ₃ -C ₆ H ₄	2-CH ₃ -C ₆ H ₄	196	82
3q	4-CH ₃ -C ₆ H ₄	H	> 300	76
3r	4-CH ₃ -C ₆ H ₄	CH ₂ Ph	228	75
3s	4-CH ₃ -C ₆ H ₄	CH ₃	230	80

Likewise the reaction of all other 2-aryl-4H-pyrido[2,3-*h*]-3,1-benzoxazin-4-ones (2b–d) with aromatic primary amines in acetic acid also afforded the corresponding 3 (Table 1) characterized by analogy and on the basis of spectral and analytical data (Tables 1–3). The above reaction constitutes



Table 2. ^1H NMR Data of 2 and 3 (DMSO- d_6)

Compound	^1H NMR Data (δ)
2b	7.5 (d, 2H, Ar-H of H^7 , H^8), 7.7 (m, 1H, Ar-H of H^4), 8.2 (d, 1H, Ar-H of H^5), 8.4 (m, 3H, Ar-H of H^2 , H^6 , H^9), 9.1 (d, 1H, Ar-H of H^1), 9.4 (d, 1H, Ar-H of H^3).
2c	2.5 (s, 3H, CH_3), 7.4 (d, 2H, Ar-H of H^7 , H^8), 7.7 (m, 1H, Ar-H of H^4), 8.2 (d, 1H, Ar-H of H^5), 8.4 (m, 3H, Ar-H of H^2 , H^6 , H^9), 9.2 (d, 1H, Ar-H of H^1), 9.4 (d, 1H, Ar-H of H^3).
3a	7.1–7.5 (m, 10H, Ar-H of C-aryl and N-aryl), 7.6 (m, 1H, Ar-H of H^2), 8.1 (d, 1H, Ar-H of H^4), 8.5 (1H, Ar-H of H^5), 9.1 (d, 1H, Ar-H of H^1), 9.4 (d, 1H, Ar-H of H^3).
3b	7.2–7.6 (m, 10H, Ar-H of C-aryl & N-aryl and H^2), 8.2 (d, 1H, Ar-H of H^4), 8.5 (d, 1H, Ar-H of H^5), 9.1 (d, 1H, Ar-H of H^1), 9.4 (d, 1H, Ar-H of H^3).
3c	2.4 (s, 3H, CH_3), 7.0–7.4 (m, 9H, Ar-H of C-aryl & N-aryl), 7.6 (m, 1H, Ar-H of H^2), 8.1 (d, 1H, Ar-H of H^4), 8.5 (d, 1H, Ar-H of H^5), 9.1 (d, 1H, Ar-H of H^1), 9.3 (d, 1H, Ar-H of H^3).
3d	2.1 (s, 3H, CH_3), 7.1–7.5 (m, 9H, Ar-H of C-aryl & N-aryl), 7.7 (m, 1H, Ar-H of H^2), 8.1 (d, 1H, Ar-H of H^4), 8.4 (d, 1H, Ar-H of H^5), 9.1 (d, 1H, Ar-H of H^1), 9.4 (d, 1H, Ar-H of H^3).
3h	2.1 (s, 3H, CH_3), 7.1–7.4 (m, 8H, Ar-H of C-aryl & N-aryl), 7.6 (m, 1H, Ar-H of H^2), 8.1 (d, 1H, Ar-H of H^4), 8.5 (d, 1H, Ar-H of H^5), 9.1 (d, 1H, Ar-H of H^1), 9.4 (d, 1H, Ar-H of H^3).
3l	2.1 (s, 3H, ortho CH_3), 2.3 (s, 3H, para CH_3), 7.0–7.4 (m, 8H, Ar-H of C-aryl & N-aryl), 7.7 (m, 1H, Ar-H of H^2), 8.1 (d, 1H, Ar-H of H^4), 8.5 (d, 1H, Ar-H of H^5), 9.1 (d, 1H, Ar-H of H^1), 9.4 (d, 1H, Ar-H of H^3).
3m	3.9 (s, 3H, OCH_3), 7.0–7.6 (m, 10H, Ar-H of C-aryl & N-aryl and H^2), 8.2 (d, 1H, Ar-H of H^4), 8.8 (d, 1H, Ar-H of H^5), 9.1 (d, 1H, Ar-H of H^1), 9.3 (d, 1H, Ar-H of H^3).
3o	2.5 (s, 3H, CH_3), 3.9 (s, 3H, OCH_3), 7.1–7.3 (m, 8H, Ar-H of C-aryl & N-aryl), 7.7 (m, 1H, Ar-H of H^2), 8.1 (d, 1H, Ar-H of H^4), 8.3 (d, 1H, Ar-H of H^5), 9.1 (d, 1H, Ar-H of H^1), 9.3 (d, 1H, Ar-H of H^3).

a simple one-step synthesis for a variety of pyrido[2,3-*h*]quinazolin-4-(3H)-ones (3). Pyrolysis of 2 with aromatic primary amines also resulted in corresponding 3. However, the yields were relatively low. The formation of 3 from 2 probably involves nucleophilic attack by anilino nitrogen on carbonyl carbon of the latter, resulting in an open chain intermediate. Subsequent loss of elements of water from the tautomeric form of the intermediate yields 3. All the 3 exhibited intense molecular ions in their mass spectra indicating the stability of ring system. Generally M^+ or $\text{M}^+ - 1$ is observed as the base peak.



Table 3. Analytical Data of 2 and 3

Compound	Molecular Formula	Found % (Calc.)		
		C	H	N
2a	C ₁₇ H ₁₀ N ₂ O ₂	74.42 (74.45)	3.62 (3.65)	10.20 (10.22)
2b	C ₁₇ H ₉ N ₂ O ₂ Cl	66.10 (66.13)	2.89 (2.92)	9.06 (9.08)
2c	C ₁₈ H ₁₂ N ₂ O ₂	75.03 (75.00)	4.16 (4.17)	9.73 (9.72)
2d	C ₁₈ H ₁₂ N ₂ O ₃	71.02 (71.05)	3.97 (3.95)	9.23 (9.21)
3a	C ₂₃ H ₁₅ N ₃ O	79.10 (79.08)	4.28 (4.30)	12.00 (12.03)
3b	C ₂₃ H ₁₄ N ₄ O ₃	70.07 (70.05)	3.57 (3.55)	14.19 (14.21)
3c	C ₂₄ H ₁₇ N ₃ O	79.30 (79.34)	4.66 (4.68)	11.57 (11.57)
3d	C ₂₄ H ₁₇ N ₃ O	79.37 (79.34)	4.69 (4.68)	11.55 (11.57)
3e	C ₂₃ H ₁₄ N ₃ OCl	71.97 (71.97)	3.66 (3.65)	10.98 (10.95)
3f	C ₂₃ H ₁₃ N ₄ O ₃ Cl	64.39 (64.41)	3.06 (3.03)	13.09 (13.07)
3g	C ₂₄ H ₁₆ N ₃ OCl	72.44 (72.45)	4.05 (4.03)	10.59 (10.57)
3h	C ₂₄ H ₁₆ N ₃ OCl	72.48 (72.45)	4.01 (4.03)	10.56 (10.57)
3i	C ₂₄ H ₁₇ N ₃ O	79.32 (79.34)	4.66 (4.68)	11.56 (11.57)
3j	C ₂₄ H ₁₆ N ₄ O ₃	70.56 (70.59)	3.94 (3.92)	13.72 (13.73)
3k	C ₂₅ H ₁₉ N ₃ O	79.54 (79.58)	5.06 (5.04)	11.11 (11.14)
3l	C ₂₅ H ₁₉ N ₃ O	79.57 (79.58)	5.06 (5.04)	11.12 (11.14)
3m	C ₂₄ H ₁₇ N ₃ O ₂	75.98 (75.99)	4.46 (4.49)	11.06 (11.08)
3n	C ₂₄ H ₁₆ N ₄ O ₄	67.93 (67.92)	3.79 (3.77)	13.18 (13.21)
3o	C ₂₅ H ₁₉ N ₃ O ₂	76.31 (76.34)	4.85 (4.83)	10.70 (10.69)
3p	C ₂₅ H ₁₉ N ₃ O ₂	76.31 (76.34)	4.80 (4.83)	10.71 (10.69)
3q	C ₁₈ H ₁₃ N ₃ O	75.23 (75.26)	4.51 (4.53)	14.61 (14.63)
3r	C ₂₅ H ₁₉ N ₃ O	79.56 (79.58)	5.01 (5.04)	11.11 (11.14)
3s	C ₁₉ H ₁₅ N ₃ O	75.72 (75.75)	4.96 (4.98)	13.92 (13.95)

Condensation of 2c with ammonia, methylamine and benzylamine afforded the corresponding pyridoquinazolin-4-(3H)-ones (3q–s). Compound 3s on treatment with aniline in acetic acid resulted in 3i. The ready replacement of the methylamino moiety by phenylamino component in the above reaction may be attributed to the enhanced conjugation in the resulting product. The formation of 3i from 3s probably involves nucleophilic attack by anilino nitrogen on carbonyl carbon of the latter, resulting in an open chain intermediate. Subsequent loss of elements of methylamine from the tautomeric form of the intermediate yields 3i. The 3-unsubstituted (3q) and 3-benzyl (3r) 2-(4-methylphenyl) pyrido[2,3-*h*]quinazolin-4(3H)-ones proved to be exceptional in their reaction with aniline. In both cases replacement of nitrogen function did not occur. Thus ring NH and NCH₂Ph are found to be resistant to replacement by NPh, which may be attributed to relatively less electrophilic nature of carbonyl carbon in



the former (3q) and due to tautomerism in N-H free compound and to steric factors in 3r.

EXPERIMENTAL SECTION

Melting points are uncorrected. IR spectra were recorded in KBr on a Perkin-Elmer 1605 spectrophotometer, ^1H -NMR (200 MHz) spectra on Varian Gemini-200 spectrometer using TMS as internal standard and mass spectra on Micromass 70-70H instrument.

2-Aryl-4H-pyrido[2,3-*h*]benzoxazin-4-ones (2): 5-Aminoquinoline-6-carboxylic acid (1, 1 mmol) was dissolved in pyridine (5 ml) and cooled at 0°C . To this the appropriate aroyl chloride (1 mmol) was added dropwise with stirring and stirred for 2 h. The mixture was poured into ice-cold water (25 ml) and the compound separated was filtered, washed with small volumes of cold water and few drops of ethanol and dried. The characterisation data of crystalline compounds 2 thus obtained are included in Tables 1–3.

Condensation of 2 with Primary Amines: A mixture of appropriate 2 (1 mmol), primary amine (1 mmol) and glacial acetic acid (5 ml) was heated on steam bath for 5 h. The reaction mixture was cooled and poured into water (25 ml) with stirring. The compound separated was filtered, washed with small volumes of cold water and few drops of ethanol and dried. The characterisation data 2,3-disubstituted pyrido[2,3-*h*]-quinazolin-4-(3H)-ones(3) thus obtained are given in Tables 1–3.

IR spectra of 2 showed $\nu\text{C=O}$ around 1750 cm^{-1} while those of 3 exhibited around 1670 cm^{-1} .

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