J. CHEM. SOC., CHEM. COMMUN., 1991

A Novel Intramolecular $[2\pi + 2\pi]$ Photocycloaddition of Di- and Tetra-allylsilanes

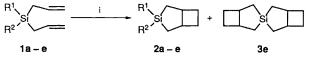
Kazuhisa Nakanishi, Kazuhiko Mizuno* and Yoshio Otsuji*

Department of Applied Chemistry, College of Engineering, University of Osaka Prefecture, Sakai, Osaka 591, Japan

Irradiation of di- and tetra-allylsilanes in the presence of 1,4-dicyanonaphthalene (1,4-DCN) in benzene afforded the intramolecular [$2\pi + 2\pi$] photocycloadducts in good yields *via* (1,4-DCN–C₆H₆–Si-compound) triplexes.

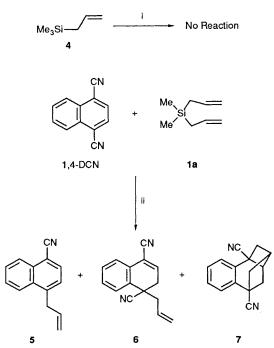
Allylic silanes are versatile compounds in organic synthesis.¹ Photochemically, they can be used as allylating agents of electron-deficient molecules such as iminium salts,² aromatic nitriles,^{3,4} quinones,⁵ electron-deficient alkenes⁶ and aromatic imides,⁷ via photoinduced electron transfer. However, little is

known about the photochemical reactivity of the alkenic moiety in allylic silanes.⁸ We now report the intramolecular $[2\pi + 2\pi]$ photocycloaddition of di- and tetra-allylsilanes in the presence of 1,4-dicyanonaphthalene (1,4-DCN) that probably proceeds *via* triplexes.



a, $R^1 = R^2 = Me$; **b**, R^1 , $R^2 = -(CH_2)_4$ -; **c**, $R^1 = R^2 = Ph$; **d**, $R^1 = Me$, $R^2 = Ph; e, R^1 = R^2 = CH_2 = CHCH_2 -$

Scheme 1 Reagents and conditions: i, hv, 1,4-DCN, C₆H₆

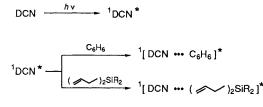


Scheme 2 Reagents and conditions: i, hv, 1,4-DCN, C₆H₆; ii, hv, MeCN

Irradiation of a benzene solution of diallyldimethylsilane 1a in the presence of a catalytic amount of 1,4-DCN through Pyrex by a high-pressure mercury lamp ($\lambda > 280$ nm) under nitrogen atmosphere for 48 h afforded 3,3-dimethyl-3-silabicyclo[3.2.0]heptane 2a in 75% yield (Scheme 1). Irradiation of the silanes 1b-e in a similar manner gave the corresponding $[2\pi + 2\pi]$ cycloadducts **2b-e** and **3e** in 54, 69, 47 and 65% yields, respectively. It was found that 2e is converted to 3e by irradiation in the presence of 1,4-DCN. These cycloadducts were isolated by column chromatography on silica gel. The structures were determined by their spectral properties (1H NMR, ¹³C NMR, IR and mass)[†] and their elemental analyses.

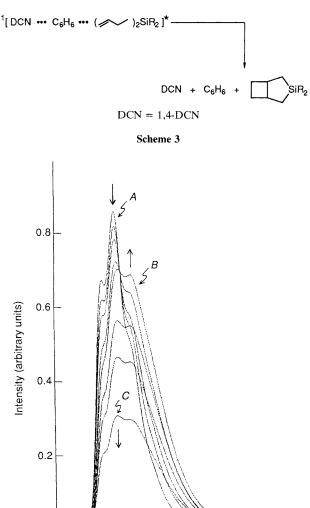
The photoreaction of allyltrimethylsilane 4 in the presence of 1,4-DCN in benzene afforded no cycloadducts even after prolonged irradiation. However, when the photoreaction was carried out in acetonitrile, the allylation of 1,4-DCN efficiently occurred to give 5, 6, 7 as previously reported (Scheme 2).4

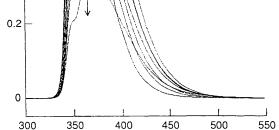
To gain mechanistic information, the effects of solvents, sensitizers and additives on the photoreactions were examined. The intramolecular photocycloaddition of 1a to give 2a did occur efficiently in aromatic solvents such as benzene, toluene and xylene, but did not occur in cyclohexane, diethyl ether, dichloromethane and ethyl acetate. However, this photocycloaddition began to proceed, though with less



$$[DCN \cdots C_{6}H_{6}]^{*} + ())_{2}SiR_{2}$$

$$[DCN \cdots C_6H_6 \cdots (///)_2SiR_2]^*$$





Wavelength / nm

Fig. 1 Fluorescence spectra of 1,4-DCN in cyclohexane: A, 1,4-DCN $([1,4-DCN] = 1.0 \times 10^{-4} \text{ mol dm}^{-3}); B, \text{ exciplex emission between } 1,4-DCN and toluene ([1,4-DCN] = 1.0 \times 10^{-4} \text{ mol dm}^{-3}, [toluene]$ = 0.2 mol dm⁻³); C, exciplex emission between 1,4-DCN and toluene quenched by 1e ([1,4-DCN] = 1.0×10^{-4} mol dm⁻³, [toluene] = 0.2 mol dm⁻³, [1e] = 3.3×10^{-2} mol dm⁻³)

efficiency, upon adding 10 vol% of benzene, toluene or xylene into cyclohexane or dichloromethane solutions of 1a. In acetonitrile, 5-7 were produced, accompanying the disappearance of 1,4-DCN.

[†] Spectroscopic data for 3e: ¹H NMR (270 MH, CDCl₃) δ 0.64 (dd, J 3.0 and 15.5 Hz, CH₂), 0.83 (dd, J 3.0 and 15.5 Hz, CH₂), 0.92 (dd, J 9.4 and 16.9 Hz, CH₂), 1.09 (dd, J 9.5 and 15.5 Hz, CH₂), 1.40–1.59 (m, CH₂), 2.11–2.22 (m, CH₂) and 2.73–2.95 (m, CH); ¹³C NMR (270 MHz, CDCl₃) δ 18.8 (CH₂), 19.8 (CH₂), 27.2 (CH₂), 29.8 (CH₂), 38.2 (CH) and 39.0 (CH).

Fluorescence of 1,4-DCN, 9,10-dicyanoanthracene (DCA), 2,3-dicyanonaphthalene (2,3-DCN) and 1,4-dicyanobenzene (DCB) was efficiently quenched in benzene by 1a-e. However, the only one of these aromatic nitriles to sensitize the photocycloaddition was 1,4-DCN. All the results strongly suggest that 1,4-DCN and benzene or its methylated compounds are indispensable for this intramolecular photocycloaddition. The formation of exciplexes between 1,4-DCN and benzene derivatives,9 and the formation of triplexes among 1,4-DCN, benzene derivatives, and 1,2-diarylcyclobutane or hexamethyldewarbenene¹⁰ have been reported. On the basis of these experimental results, we propose the mechanism shown in Scheme 3 for the photocycloaddition. Termolecular interaction among 1,4-DCN, toluene and 1e in the excited state was suggested from the fluorescence quench-ing experiments (Fig. 1). Fluorescence of 1,4-DCN in cyclohexane was quenched by toluene and 1a-e. In the 1,4-DCN-toluene system, an exciplex emission was observed, showing an isoemissive point at 368 nm. This exciplex emission was efficiently quenched by 1e, but a triplex emission was not observed at longer wavelength.

Received, 16th October 1990; Com. 0/04652C

J. CHEM. SOC., CHEM. COMMUN., 1991

References

- E. W. Colvin, Silicon in Organic Synthesis, Butterworth, London, 1981; W. P. Weber, Silicon Reagents for Organic Synthesis, Springer-Verlag, Berlin, 1983; S. Patai and Z. Rappoport, The Chemistry of Organosilicon Compounds, Wiley, New York, 1989.
- 2 K. Ohga and P. S. Mariano, J. Am. Chem. Soc., 1982, 104, 617;
 K. Ohga, U. C. Yoon and P. S. Mariano, J. Org. Chem., 1984, 49, 213.
- 3 K. Mizuno, M. Ikeda and Y. Otsuji, *Tetrahedron Lett.*, 1985, 26, 461; K. Mizuno, K. Nakanishi and Y. Otsuji, *Chem. Lett.*, 1988, 1833.
- 4 K. Mizuno, K. Terasaka, M. Ikeda and Y. Otsuji, *Tetrahedron Lett.*, 1985, 26, 5819; K. Mizuno, K. Terasaka, M. Yasueda and Y. Otsuji, *Chem. Lett.*, 1988, 145.
- 5 K. Maruyama, H. Imahori, A. Osuka, A. Takuwa and H. Tagawa, *Chem. Lett.*, 1986, 1719.
- 6 K. Mizuno, M. Ikeda and Y. Otsuji, Chem. Lett., 1988, 1507.
- 7 Y. Kubo, T. Imaoka, T. Shiragami and T. Araki, Chem. Lett., 1986, 1749.
- 8 M. Ochiai, M. Arimoto and E. Fujita, J. Chem. Soc., Chem. Commun., 1981, 481.
- 9 H. F. Davis, S. K. Chattopadhyay and P. K. Das, J. Phys. Chem., 1984, 88, 2798.
- 10 Y. Masaki, S. Yanagida and C. Pac, *Chem. Lett.*, 1988, 1305; Y. Masaki, Y. Uehara, S. Yanagida and C. Pac, *Chem. Lett.*, 1990, 1339.