

Il Farmaco 57 (2002) 559-564

IL FARMACO

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## Synthesis and preliminary anticancer activity of new 1,4-dihydro-3-(3-hydroxy-2-naphthyl)-4-substituted-5*H*-1,2,4-triazoline-5-thiones

A. Duran, H.N. Dogan, Sevim Rollas\*

Department of Pharmaceutical Chemistry, Faculty of Pharmacy, Marmara University, 81010 Haydarpasa, Istanbul, Turkey

Received 22 January 2002; accepted 26 January 2002

### Abstract

1,4-Dihydro-3-(3-hydroxy-2-naphthyl)-4-substituted-5*H*-L2,4-triazolinc-5-thiones were synthesized. The structures of original eight compounds were confirmed by elemental analysis,<sup>1</sup>H NMR and mass spectral methods. One of the compounds (**3a**) was tested in vitro for its anticancer activity against 52 human tumor cell lines. © 2002 Éditions scientifiques et médicales Elsevier SAS. All rights reserved.

Keywords: 1,2,4-Triazoline-5-thione derivatives; Synthesis; Anticancer activity

### 1. Introduction

A series of 3,4-disubstituted-1,2,4-triazoline-5-thiones were synthesized by us [1]. In this study, these compounds were screened for in vitro anti-tumor activity by National Institutes of Health National Cancer Institute and one of them [1,4-dihydro-3-(3-acetyloxy-2-naph-thyl)-4-ethyl-5H-1,2,4-triazoline-5-thione (3a)] was found valuable to research. In this point of view, we synthesized new 1,2,4-triazoline-5-thiones and their acetylated and benzoylated derivatives. Thus, eight new compounds were obtained for anticancer drug development.

### 2. Chemistry

The compounds were synthesized in several steps (Fig. 1). First 1,2,4-triazoline-5-thiones  $2\mathbf{a}-\mathbf{f}$  were formed by the cyclization in alkaline medium of the thiosemicarbazides  $1\mathbf{a}-\mathbf{f}$ , which were obtained by the condensation of 3-hydroxy-2-naphthoic acid hydrazide with alkyl-aryl isothiocyanates [2,3]. Some of the physical properties of these compounds were shown in Table 1.

For obtaining benzoylated derivatives  $3a_1$  and  $3a_2$ , 2a was heated in benzoyl chloride and 2,6-dichlorobenzoyichloride [4], respectively, and 2b was heated in acetic anhydride to obtain acetylated derivative 3b [5].

The structures of the new compounds were confirmed using UV, <sup>1</sup>H NMR and mass (for only **2c** which was selected as a prototype) spectral data besides their elemental analysis.

According to the UV spectroscopic data (Table 1), strong bands, which observed at 229.8–233.4 nm, characterized to K band of triazole ring.

The <sup>1</sup>H NMR spectra of 2a-f displayed the OH and NH resonance at 10.15–10.55 and 13.87–14.20 ppm, respectively [6,7]. The signal of these protons were not observed because of removel of these protons by acetylation-benzoylation for **3b**, **3a**<sub>1</sub> and **3a**<sub>2</sub>. The other protons were shown in Table 2.

The mass spectrum of 2c revealed the molecular ion peak (100%). The important common fragments [5] were shown in Fig. 2.

### 3. Experimental

### 3.1. Chemistry

Chemicals used in the experiments were commercially available and were used without further purification.

<sup>\*</sup> Corresponding author

E-mail address: sevim@sevimrollas.com (S. Rollas).

Melting points were determined in glass capillary tubes on a Büchi 530 apparatus and unconnected. Elemental analyses were performed with Leco-932 elemental analyzer. UV spectra were obtained on a Shimadzu UV 2100S spectrophotometer (1 mg/100 ml in ethanol). The proton nuclear magnetic resonance (<sup>1</sup>H NMR) spectra were run on a Bruker AVANC DPX 400 spectrometer in DMSO- $d_6$  with tetramethylsilane as the internal reference: chemical shift values ( $\delta$ ) are expressed in ppm. Mass spectra were taken on a Fisons Instruments VG Platform II LS-MS spectrometer.

3.1.1. General procedure for the preparation of
1,4-dihydro-3-(3-hydroxy-2-naphthyl)-4-substituted5H-1,2,4-triazoline-5-thiones(2a-f)
A solution of 3-hydroxy-2-naphthoic acid hydrazide



Fig. 1. Synthesis of the compounds.

Table 1			
Physical	constants	of	2b-f

Composition	Yield (%)	M.p. (°C)	Molecular formula (mwt.)	UV( $\lambda_{max}$ ) $\varepsilon_{max}$	Analyses (cal	culated/found	d)
					C	Н	Ν
2b	87.5	233	C <sub>15</sub> H <sub>15</sub> N <sub>3</sub> OS 285.37		63.14 63.77	5.30 5.35	14.72 14.57
2c	94.9	298	$C_{18}H_{13}N_3OS$ 319.38	233.0 (13925)	67.69 66.96	4.10 4.17	13.15 12.78
2d	79.2	239–243	$C_{18}H_{12}FN_{3}OS. 1/2(C_{2}H_{5}OH \cdot H_{2}O)$ 369.42	232.7 (77430)	61.77 61.18	4.37 4.30	11.37 11.79
2e	89.2	301	C <sub>18</sub> H <sub>12</sub> FN <sub>3</sub> OS 337.38	232.7 (32591)	64.08 64.41	3.59 3.95	12.46 11.95
2f	96.3	290	$C_{19}H_{15}N_3OS$ 333.41	232.8 (13370)	68.45 68.42	4.53 4.79	12.60 12.51



		(1) = (1) + (1)
Comp.	R	<sup>1</sup> H-NMR (DMSO-d <sub>6</sub> ), δ ppm
2b	$c\ddot{H}_2 - c\ddot{H}_2 - c\ddot{H}_3$	13.87 (1H, s, NH), 10.55 (1H, s, OH), 8.01 (1H, s, C <sub>1</sub> H), 7.89 (1H, d, C <sub>8</sub> H), 7.79 (1H, d, C <sub>5</sub> H), 7.51 (1H, t, C <sub>6</sub> H), 7.36 (2H, t, C <sub>7</sub> H and C <sub>4</sub> H), 3.87 (2H, t, 2H <sub>a</sub> ), 1.51 (2H, m, 2H <sub>b</sub> ), 0.63 (3H, t, 3H <sub>c</sub> )
2c	a_b c a_b	$ \begin{array}{l} 14.06 \ (1H, \ s, \ NH), \ 10.15 \ (1H, \ s, \ OH), \ 8.02 \ (1H, \ s, \ C_1H), \ 7.79 \ (1H, \ d, \ C_8H), \ 7.65 \ (1H, \ d, \ C_5H), \ 7.44 \ (1H, \ t, \ C_6H), \ 7.36-7.28 \ (6H, \ m, \ C_7H, \ 2H_a, \ 2H_b \ and \ H_c), \ 7.05 \ (1H, \ s, \ C_4H) \end{array} $
2d	F d c b	$ \begin{array}{llllllllllllllllllllllllllllllllllll$
2e	<sup>a</sup> <sub>d</sub> <sup>F</sup> <sub>d</sub> <sup>c</sup>	14.14 (1H, s, NH), 10.28 (1H, s, OH), 8.03 (1H, s, C <sub>1</sub> H), 7.80 (1H, d, C <sub>8</sub> H), 7.67 (1H, d, C <sub>5</sub> H), 7.45 (1H, t, C <sub>6</sub> H), 7.41-7.32 (2H, m, C <sub>7</sub> H and H <sub>c</sub> ), 7.28 (1H, d, H <sub>d</sub> ), 7.21 (1H, t, H <sub>b</sub> ), 7.10 (1H, d, H <sub>a</sub> ), 7.07 (1H, s, C <sub>4</sub> H)
2f	$\xrightarrow{a \ b} CH_3$	$ \begin{array}{llllllllllllllllllllllllllllllllllll$

in ethanol was heated until dissolved, treated with equimolar appropriate isothiocyanate, refluxed for 3 h and precipitated thiosemicarbazide was recrystallized from ethanol [2].

A solution of appropriate thiosemicarbazide 1a-f in 2 N sodium hydroxide was refluxed for 2 h. After cooling, cone, hydrochloric acid was added and precipitated product 2a-f was filtered, washed several times with distilled water, finally recrystallized from ethanol [1].

# 3.1.2. General procedure for the preparation of 1,4-dihydro-1-acetyl-3-(3-acetyloxy-2-naphthyl)-4-propyl-5H-l,2,4-triazoline-5-thione (**3b**)

A solution of **2b** in acetic anhydride was refluxed at 150-160 °C for 30 min. After cooling, distilled water was added and allowed to stand overnight. The resulting precipitate was filtered, washed with distilled water and recrystallized from methanol [5]. C<sub>19</sub>H<sub>19</sub>N<sub>3</sub>O<sub>3</sub>S, 86%, m.p. 159 °C. Elemental analysis (Calc./Found%): 61.77/62.33 (C), 5.18/5.59 (H), 11.38/11.46 (N), 8.68/8.42 (S).<sup>1</sup>H NMR ( $\delta$ , ppm): 0.67 (3H, s, CH<sub>3</sub>), 1.57 (2H, m, *CH*<sub>2</sub>-CH<sub>3</sub>), 2.24 (3H, s, OCOCH<sub>3</sub>), 2.65 (3H, s, NCOCH<sub>3</sub>), 3.94 (2H, t, N-CH<sub>2</sub>), 7.66 (1H, t, C<sub>6</sub>H of napht.), 7.73 (1H, t, C<sub>6</sub>H of napht.), 7.99 (1H, s, C<sub>4</sub>H of napht.), 8.06 (1H, d, C<sub>8</sub>H of napht.), 8.42 (1H, s, C<sub>1</sub>H of napht.).

### 3.1.3. General procedure for the preparation of 1,4-dihydro-3-(3-benzoyloxy-2-naphthyl)-4ethyl-5H-1,2,4-triazoline-5-thione (**3**a<sub>1</sub>)

Benzoyl chloride was added to a solution of 2a in acelone and refluxed at 120-140 °C for 6 h. After

cooling, the precipitated product was recrystallized from ethanol [4].  $C_{21}H_{17}N_3O_2S$ , 55%, m.p. 197–199 °C. Elemental analysis (Calc./Found%): 67.18/67.38 (C), 4.57/4.22 (H), 11.19/10,77 (N), 8.54/8.00 (S).<sup>1</sup>H NMR ( $\delta$ , ppm): 1.13 (3H, t, CH<sub>3</sub>), 4.00 (2H, q, CH<sub>2</sub>), 7.60 (2H, t, m-protons to carbonyl on benzoyl), 7.66–7.77 (3H, m, p-proton to carbonyl on benzoyl and C<sub>7</sub>H C<sub>6</sub>H of napht.), 7.99 (2H,d, o-protons to carbonyl on



Fig. 2. Common mass fragmentation patterns of 2c.

benzoyl), 8.07 (1H, d,  $C_5H$  of napht.), 8.13 (2H, d,  $C_4H$  and  $C_8H$  of napht.), 8.42 (1H, s,  $C_1H$  of napht.), 13.86 (1H, s, NH).

### 3.1.4. General procedure for the preparation of 1,4-dihydro-1-(2,6-dichlorobenzoyl)-3-(3-hydroxy-2-naphthyl)-4-ethyl-5H-1,2,4-triazoline-5-thione (3a<sub>2</sub>)

2,6-Dichlorobenzoyl chloride was added to a solution of **2a** in acelone and refluxed at 120–140 °C for 2 h. The precipitated product was filtered and recrystallized from ethanol [4].  $C_{21}H_{15}Cl_2N_5O_2S$ , 96%, m.p. 210 °C. Elemental analysis, (Calc./Found%): 56.77/56.77 (C), 3.40/3.31 (H), 9.46/9.33 (N), 7.21/7.22 (S).<sup>1</sup>H NMR ( $\delta$ , ppm): 1.15 (3H, t, CH<sub>3</sub>), 4.10 (2H q, CH<sub>2</sub>), 7.37 (2H, t, m-protons to chlorines on benzoyl and C<sub>4</sub>H of napht.), 7.53 (1H, t, C<sub>7</sub>H of napht.), 7.64–7.73 (3H, m, o-protons to chlorines on benzoyl and CgH of napht.), 7.82 (1H, d, C<sub>5</sub>H of naphl.), 7.94 (1H; d, C<sub>8</sub>H of napht.), 8.06 (1H, s, C<sub>1</sub>H of naphl.), 10.93 (1H, s, OH).

### 3.2. Anti-tumor activity screen

The NCl's in vitro anti-tumor screen consisted of 52 human tumor cell lines (Table 3) against which 3a was tested at a minimum of five concentrations at 10-fold dilutions. A 48 h continuous drug exposure protocol was used, and a sulforhodamine B (SRB) protein assay was used to estimate cell viability or growth. The measured effect of the 3a on a cell line was currently calculated according to one or the other of the following two expressions:

$$\begin{split} \text{If (Mean } \text{Od}_{\text{test}} - \text{Mean } \text{Od}_{\text{tzero}}) &\geq 0 \\ \text{PG} &= 100 \times (\text{Mean } \text{Od}_{\text{test}} - \text{Mean } \text{Od}_{\text{tzero}}) \\ &/(\text{Mean } \text{Od}_{\text{ctrl}} - \text{Mean } \text{Od}_{\text{tzero}}) \\ \text{If (Mean } \text{Od}_{\text{test}} - \text{Mean } \text{Od}_{\text{tzero}}) &< 0 \\ \text{PG} &= 100 \times (\text{Mean } \text{Od}_{\text{test}} - \text{Mean } \text{Od}_{\text{tzero}}) \\ \end{split}$$

/Mean Od<sub>tzero</sub>

Mean  $Od_{tzero}$  is the average of optical density measurements SRB-derived color just before exposure of cells to the test compound. Mean  $Od_{test}$  is the average of optical density measurements SRB-derived color after 48 h exposure of cells to the test compound. Mean $Od_{ctrl}$  is the average of optical density measurements SRB-derived color after 48 h with no exposure of cells to the test compound. PG is the percentage growth, the calculated measurement of effect.

#### 4. Results and discussion

Table 3 shows all tested human tumor cell lines. Table 4 represents the experimental data collected

Table :	,			
Tested	panels	and	cell	lines

Panel	Cell line
Leukemia	CCRF-CEM, HL-60(TB), K-562. MOLT-4, RPML-8226 SR
Non-small cell	EKVX, HOP-62, HOP-92, NCI-H226,
lung cancer Colon cancer	NCI-H23, NCI-H322M, NCI-H460, NCI-H522 HCC-2998, HCT-116, HCT-15, HT29, KM12,
CNS cancer	SW-620 SE-268 SE-295 SE-539 U251
Melanoma	LOX fMVI, M14, SK-MEL-2, SK-MEL-28,
Ovarian cancer	SK-MEL-5, UACC-257 IGROVI, OVCAR-3, OVCAR-4, OVCAR-5,
Panal concor	OVCAR-8, SK-OV-3
Kellai Cancei	TK-10, UO-31
Prostate cancer	PC-3, DU-145
Breast cancer	MCF7, NCI/ADR-RES; MDA-MB-231/ATCC, HS 578T, MDA-MB-435, MDA-N



Fig. 3. Dose-response curves of selected six cell lines.

against six cell lines for anti-tumor activity screen. Out of 52 cell lines studied, only six were presented here. The first column describes the subpanel and cell line involved. The second column lists the Mean Od<sub>tzero</sub> and Mean Od<sub>ctrl.</sub> The third column lists the mean optical densities (Mean Od<sub>test</sub>) for each of five different concentrations and their calculated percent growth in paranthesis. Each concentration was expressed as the  $\log_{10}$  (molar or  $\mu g/ml$ ). The next column lists the response parameters GI50 (%50 growth inhibition), TGI (total growth inhibition) and LC50 (lethal concentration), these parameters were interpolated values representing the concentrations at which the PG was +50, 0and -50, respectively. In the case which these response parameters can not be obtained by interpolation, the value given for each response parameter is the highest concentration tested and is preceded by a '>' sign. The last column lists the  $\log_{10}$  of the values in fourth column and these values were calculated from the curves in Fig. 3. The values extending to the smaller

Table 4 In vitro testing	results of <b>3a</b> fo	or selected six o	cell lines out o	f 52 lines								
Cell line	Mean Od <sub>tzero</sub> (Od <sub>ctrl</sub> )	Mean Od <sub>test</sub>	(percent growt	h)			Response pai	rameters		log <sub>10</sub> GI50	$\log_{10} \mathrm{TGI}$	log <sub>10</sub> LC50
		-8.0	- 7.0	-6.0	-5.0	-4.0	GI50	TGI	LC50			
CCRF-CEM	0.207 (0.836)	0.587 (60)	0.581 (59)	0.533 (52)	0.350 (23)	0.166(-20)	1.16E - 0.6	3.42E - 0.5	> 1.00E - 0.4	- 5.94	-4.47	> -4.00
HL-60(TB)	0.737 (2.117)	1.704 (70)	1.398 (48)	1.356 (45)	1.062 (24)	0.314(-57)	8.05E - 08	1.95E - 05	8.10E - 05	-7.09	-4.71	-4.09
SR	0.695 (1.995)	2.005(101)	1.984(99)	1.180 (37)	0.573 (-18)	0.414(-41)	6.23E - 0.7	4.79E - 0.6	> 1.00E - 0.4	-6.21	-5.32	> -4.00
(Leukemia)												
NCI-H460 (lung)	0.280 (1.357)	1.305 (95)	1.337 (98)	1.302 (95)	1.289 (94)	0.232 (-17)	2.47E - 0.5	6.98E - 0.5	>1.00E-0.4	-4.61	-4.16	>4.00
SF-268 (CNS)	0.469 (1.294)	1.293 (100)	1.280 (98)	1.185 (87)	1.136 (81)	0.495(3)	2.49E - 0.5	> 1.00E - 0.4	> 1.00E - 0.4	4.60	>4.00	> -4.00
MCF7	0.273 (1.219)	1.268 (105)	1.164 (94)	1.174 (95)	1.165 (94)	0.307 (4)	3.08E - 0.5	> 1.00E - 0.4	> 1.00E - 0.4	-4.51	>4.00	> 4.00
(breast)												
									Mean of 52	-4.68	-4.17	-4.01
									lines:			

values from the mean value of 52 cell lines represent sensitivity of cell line to the test agent in excess of the average sensitivity of all tested cell lines. The values to the higher values represent sensitivity less than the mean.

The values at either limit were also calculated in the mean used for the  $\log_{10}$  Gl50,  $\log_{10}$  TGI and  $\log_{10}$  LCSO. Therefore, the mean was given in the Table 4 may not be the actual mean of the G150 for instance. For this reason, we referred to this value as the mean (-4.68). According to mean values, the principal response parameters lay about (-7.09) to (-4.00). The values within > -4 are uneffective [8].

The dose-response curves (Fig. 3 which was drawn for only six cell lines) were created by plotting the percent growths against the  $\log_{10}$  of the corresponding concentrations for 52 cell lines. Horizontal lines were provided at the PG values of + 50, 0 and - 50. The concentrations corresponding to these points where the curves cross these lines were the GI50, TGI and LC50.

### Acknowledgements

This research was supported by The Research Fund of Marmara University, project number SA/1999-132. We thank Dr. Edward Sausville from National Institutes of Health, National Cancer Institute for the in vitro anticancer activity.

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