ISSN 1070-3632, Russian Journal of General Chemistry, 2010, Vol. 80, No. 2, pp. 366–367. © Pleiades Publishing, Ltd., 2010. Original Russian Text © R.Kh. Bagautdinova, L.K. Kibardina, A.R. Burilov, M.A. Pudovik, 2010, published in Zhurnal Obshchei Khimii, 2010, Vol. 80, No. 2, pp. 339–340.

LETTERS TO THE EDITOR

Reaction of Salicylalkylenediimines with 2-Chloro-4,5-diphenyl-1,3,2-dioxaphospholene

R. Kh. Bagautdinova, L. K. Kibardina, A. R. Burilov, and M. A. Pudovik

Arbuzov Institute of Organic and Physical Chemistry, Kazan Research Center, Russian Academy of Sciences, ul. Arbuzova 8, Kazan, 420088 Tatarstan, Russia fax: (8432) 752253 e-mail: pudovik@iopc.knc.ru

Received July 9, 2009

DOI: 10.1134/S1070363210020313

Reaction of salicyl(cyclo)alkylenediimine **Ia–Ic** with 4,5-diphenyl-2-chloro-1,3,2-dioxaphospholene **II** was found to occur as a cascade cyclization giving rise

to tetracyclic derivatives IIIa–IIIc of the hexacoordinated phosphorus atom of the salt structure involving intramolecular transannular $N \rightarrow P$ bond.



I, **III**, **IV**: $R = (CH_2)_2$ (**a**); (CH₂)₃ (**b**); cyclohexyl (**c**).

The structure of compounds IIIa-IIIc was proved by the IR, ¹H, and ³¹P NMR spectroscopy, mass spectrometry, and composition, by the elemental analysis data. The IR spectra of the products IIIa-IIIc contain absorption bands at 1623-1627 (C=N) and $2590-2724 \text{ cm}^{-1}$ (broadened complex band, NH₂⁺). The chemical shift of the phosphorus atom at δ_P from -114to -122 ppm is characteristic of the hexacoordinated phosphorus derivatives. Notice that the ³¹P NMR spectra of the reaction mixtures of IIIa-IIIc contain signals indicative of the formation of diastereoisomers mixture. As a result of fractional crystallization compound IIIa was obtained as the two diastereoisomers mixture in the ratio 1:1. Compounds IIIb and IIIc were isolated as individual diastereomers. Due to their salt structure the ¹H NMR spectra of compounds IIIa**IIIc** contain poorly resolved signals, therefore only group analysis of the proton signals was performed on the basis of chemical shifts without considering the fine structure of the spectra. The transannular $N \rightarrow P$ bond (2.006–2.007 Å) is realized in the compounds **IIIa–IIIc**. The intermolecular bonding of similar type was earlier observed in the series of pentacoordinated phosphorus derivatives [1–5]. On treating with triethylamine, the product **IIIa** undergoes dehydro-chlorination to give phosphorate **IVa** of neutral structure (δ_P –105.87 ppm).

1,1-Diphenylethenedioxy-3,4,11,12-dibenzo-6-aza-9-ammonia-2,13-dioxa-1-phospha[**8.3.0**^{1,10}]**tridecatri-3,5,11-ene chloride (IIIa).** To a solution of 1.13 g of salicylethylenediimine Ia in 10 ml of anhydrous methylene chloride was added dropwise 1.17 g of chloroderivative **II.** After 24 h the precipitate was separated and washed with diethyl ether. Yield 1.88 g (82%), light brown powder, mp 165–167°C. IR spectrum, v, cm⁻¹: 1607 (Ph), 1627 (C=N), 2597, 2669 (NH₂⁺). ¹H NMR spectrum (CDCl₃), δ , ppm: 3.02 br.s (2H, NH₂⁺), 3.20–3.37 m (1H, CH₂), 3.60 d (1H, CH₂, ²J_{HH} 12.29 Hz), 3.77 d (1H, CH₂, ²J_{HH} 10.25 Hz), 4.10–4.50 m (1H, CH₂), 4.73 s (1H, PCH), 6.86–7.69 m (18H, Ph), 8.32 br.s (1H, N=CH). ³¹P NMR spectrum (DMSO-*d*₆), $\delta_{\rm P}$, ppm: from –115.98 to –122.60 (1:1). Mass spectrum (MALDI–TOF), *m*/z: 507 [*M*⁺ – HCl]. Found, %: C 66.44; H 4.92; Cl 6.36; N 5.36; P 5.46. C₃₀H₂₆ClN₂O₄P. Calculated, %: C 66.11; H 4.82; Cl 16.50; N 5.36; P 5.68.

1,1-Diphenylethenedioxy-3,4,11,12-dibenzo-6-aza-10-ammonia-2,14-dioxa-1-phospha[9.3.0^{1,11}]tridecatri-3,5,12-ene chloride (IIIb). To a solution of 1.23 g of salicylpropylenediimine Ib in 10 ml of anhydrous methylene chloride was added dropwise 1.21 g of chloroderivative II. After 2 days to this reaction mixture was added 5 ml of anhydrous diethyl ether. The precipitate was separated and washed with diethyl ether. Yield 2.0 g (82%), light brown powder, mp 149– 151°C. IR spectrum, v, cm⁻¹: 1626 (C=N), 2724 (NH₂⁺). ³¹P NMR spectrum (CDCl₃), δ_P , ppm: –120.98. Mass spectrum (MALDI–TOF), *m*/z: 521 [*M*⁺ – HCl]. Found, %: C 66.41; H 5.44; Cl 6.54; N 5.12; P 5.46. C₃₁H₂₈ClN₂O₄P. Calculated, %: C 66.59; H 5.66; Cl 6.34; N 5.01; P 5.54.

1,1-Diphenylethenedioxy-3,4,11,12-dibenzo-6-aza-7,8-tetramethyleno-9-ammonia-2,13-dioxa-1-phospha-[8.3.0^{1,10}]tridecatri-3,5,11-ene chloride (IIIc) was prepared similarly from 1.0 g of salicylcyclohexyldiimine Ic and 0.86 g of chloroderivative II. Yield 1.4 g (75%), light brown powder, mp 153.5–154°C. IR spectrum, v, cm⁻¹: 1623 (C=N), 2667 (NH₂⁺). ³¹P NMR spectrum (CDCl₃), δ_P , ppm: –105.87. Mass spectrum (MALDI-TOF), *m*/z: 561 [*M*⁺ – HCl]. Found, %: C 67.96; H 5.36; Cl 5.93; N 4.76; P 5.06. C₃₄H₃₂ClN₂O₄P. Calculated, %: C 68.16; H 5.39; Cl 5.92; N 4.68; P 5.17.

1,1-Diphenylethenedioxy-3,4,11,12-dibenzo-6-aza-9-ammonia-2,13-dioxa-1-phospha[8.3.0^{1,10}]tridecatri-**3.5.11-ene (IV).** To a suspension of 0.6 g of compound Ia in 5 ml of anhydrous methylene chloride was added dropwise 0.2 g of triethylamine. After 5 days the reaction mixture was concentrated and mixed with a small amount of anhydrous diethyl ether. The precipitate was filtered off and washed with water to remove triethylamine hydrochloride. Yield 0.43 g (74%), mp 196-198°C. IR spectrum, v, cm⁻¹: 1625 (C=N), 3398 (NH). ¹H NMR spectrum (CDCl₃), δ, ppm: 3.11-4.51 m (4H, NCH₂), 6.94–7.28 m (18H, Ph), 7.57 d. d (1H, PCH, ${}^{2}J_{PH} = {}^{2}J_{HH}$ 7.62 Hz), 8.20 d (1H, NCH, ${}^{2}J_{HH}$ 4.12 Hz). ³¹P NMR spectrum (CDCl₃), δ_P , ppm: -105.87. Found, %: N 5.73; P 5.97. C₃₀H₂₅N₂O₄P. Calculated, %: N 5.51; P 6.10.

The IR spectra were recorded on a Vector-22 (Bruker) spectrometer in the range of 400–3600 cm⁻¹ from mulls in mineral oil. The ¹H NMR spectra were registered on a Avance-600 instrument (600.13 MHz) relative to signals of the residual protons of deuterated solvent (chloroform-*d*). The ³¹P NMR spectra were taken on a Bruker MSL-400 NMR-Fourier spectrometer (100.62 MHz). The mass spectra MALDI–TOF were obtained on a ULTRAFLEX III instrument (matrix *p*-nitroaniline).

ACKNOWLEDGMENTS

This work was financially supported by the Russian Foundation for Basic Research (grant no. 08-03-00029).

REFERENCES

- 1. John, K.P. and Schmutzler, R.J., J. Chem. Soc., 1974, no. 22, p. 2466.
- John, K.P., Schmutzler, R.J., and Sheldrick, W.S., J. Chem. Soc., 1974, no. 17, p. 1841.
- Wong, Ch.Y., McDonald, R., and Cavell, R.G., *Inorg. Chem.*, 1996, vol. 35, p. 325.
- 4. Pudovik, M.A., Terentyeva, S.A., and Pudovik, A.N., *Zh. Obshch. Khim.*, 1982, vol. 52, p. 491.
- Said, M., Pulm, M., Herbert-Irmer, R., Swamy, K, and Kumara, C., J. Am. Chem. Soc., 1996, vol. 118, p. 9841.