

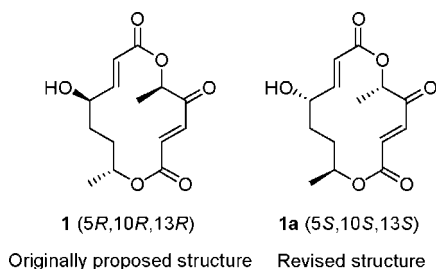
Asymmetric Total Synthesis and Revision of the Absolute Configuration of 4-Keto-Clonostachydiol

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Received February 18, 2009



The first total synthesis of the 14-membered nonsymmetric macrocyclic bislactone 4-keto-clonostachydiol, along with its enantiomer, has been accomplished in 13 steps with overall yields of 8.4% and 8.0%, respectively. The absolute configuration of 4-keto-clonostachydiol **1** has been revised as (5S,10S,13S).

4-Keto-clonostachydiol **1**, a 14-membered nonsymmetric macrocyclic bislactone, was first isolated from New Zealand marine alga-derived fungus *Gliocladium* sp.¹ 4-Keto-clonostachydiol **1** exhibits various biological properties, such as strong cytotoxicity against P388 cells (IC₅₀ 0.55 μ M) and significant activities against *Bacillus subtilis*, the fungi *trichophyton mentagrophytes*, and *cladosporium resinae*. The originally proposed stereochemistry of **1** was assigned as (5R,10R,13R) based on the comparison of NMR spectra and optical rotation of a known compound clonostachydiol **2**, which was the Luche reduction product of **1** (Figure 1).^{3,4} The structure of **1** has recently been patented for its excellent cytotoxic and antibacterial activities.² Syntheses toward clonostachydiol **2** have been

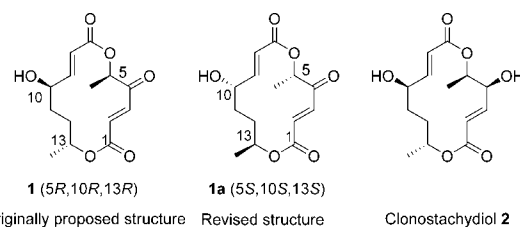
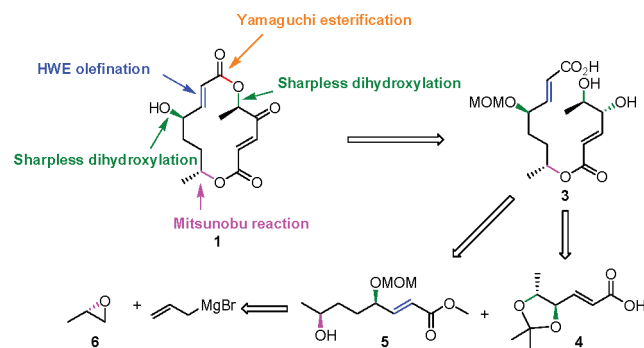
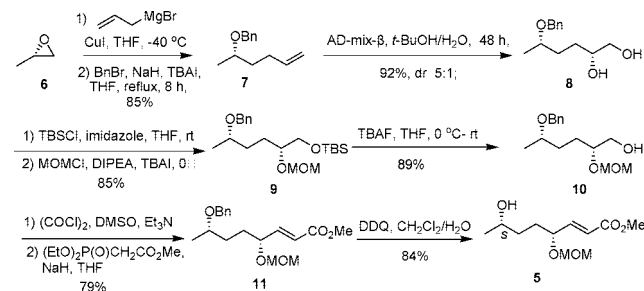


FIGURE 1. Structure of 4-Keto-Clonostachydiol (**1**) and Clonostachydiol (**2**).

SCHEME 1. Retrosynthetic Analysis of 4-Keto-Clonostachydiol 1



SCHEME 2. Synthesis of Enolate Alcohol 5



reported by two individual groups.^{5,6} Herein we describe the first asymmetric total synthesis of 4-keto-clonostachydiol from commercially available chiral methyloxirane along with the revision of the absolute configuration of **1** as (5S,10S,13S).

As outlined in Scheme 1, we envisioned that the lactone ring of 4-keto-clonostachydiol **1** could be closed by an intramolecular Yamaguchi lactonization reaction of dihydroxy acid **3**, which was assembled of carboxylic acid **4** and enolate alcohol **5** via Mitsunobu reaction. The two stereocenters in acid **4** could easily be established by Sharpless asymmetric dihydroxylation of the corresponding diene while fragment **5** was prepared from (S)-(+)-methyloxirane **6** and allyl magnesium bromide.

Synthesis of alcohol **5** began with commercially available (S)-methyloxirane **6** (Scheme 2). Regioselective ring-opening of epoxide **6** by allyl magnesium bromide in the presence of

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(1) Lang, G.; Mitova, M. I.; Ellis, G.; Sar, S.; Phipps, R. K.; Blunt, J. W.; Cummings, N. J.; Cole, A. L. J.; Munro, M. H. *J. Nat. Prod.* **2006**, 69, 621–624.

(2) Kano, C.; Adachi, K.; Shizusato, Y. Patent JP 5247737, 2005.

(3) Grabley, S.; Hammann, P.; Thiericke, R.; Wink, J.; Philipps, S.; Zeeck, A. *J. Antibiot.* **1993**, 46, 343–345.

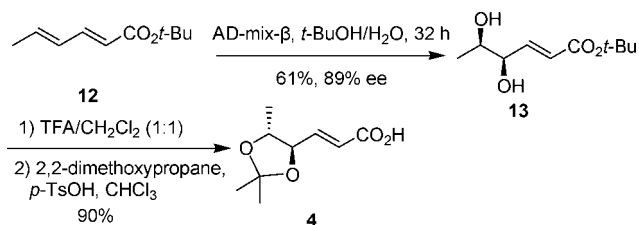
(4) Abate, D.; Abraham, W.; Meyer, H. *Phytochemistry* **1997**, 44, 1443–1448.

(5) (a) Rao, A. V. R.; Murthy, V. S.; Sharma, G. V. M. *Tetrahedron Lett.* **1995**, 36, 139–142. (b) Rao, A. V. R.; Murthy, V. S.; Sharma, G. V. M. *Tetrahedron Lett.* **1995**, 36, 143–146.

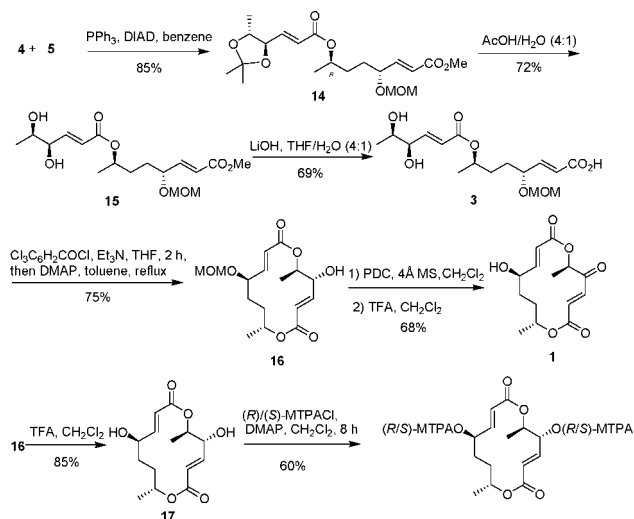
(6) Yadav, J. S.; Swamy, T.; Reddy, B. V. S. *Synlett* **2008**, 2773–2776.

(7) Bouz, S.; Cossy, J. *Tetrahedron Lett.* **2006**, 47, 901–904, and references cited therein.

SCHEME 3. Synthesis of Acid 4



SCHEME 4. Synthesis of 4-Keto-Colonostachydiol 1



CuI yielded the alcohol, which was protected as benzyl ether **7** (85% yield for two steps). Asymmetric dihydroxylation of the terminal olefin with AD-mix- β afforded diol **8** (dr 5:1).⁸ Protection of the primary and secondary hydroxyl group as TBS ether and MOM ether, respectively, resulted in compound **9**. Desilylation of **9** with treatment with TBAF generated primary alcohol **10**. After sequential Swern oxidation⁹ and Horner–Wadsworth–Emmons olefination,¹⁰ enoate **11** was obtained. Subsequently, enoate **11** was subjected to oxidative debenzoylation with DDQ¹¹ in aqueous CH_2Cl_2 to yield enoate alcohol **5**.

Synthesis of acid **4** started from (2*E*,4*E*)-*tert*-butyl hexa-2,4-dienoate **12**. Regioselective Sharpless asymmetric dihydroxylation of ester **12** with AD-mix- β (89% ee)^{12,13} afforded the corresponding *syn* diol **13**, which was converted to key fragment **4** through acidic hydrolysis and acetonide protection (Scheme 3).¹⁴

With fragments **4** and **5** in hand (Scheme 4), the Mitsunobu reaction¹⁵ was employed to construct dienoate **14** with the inversion of the C-13 configuration. Deprotection of **14**,¹⁶ followed by

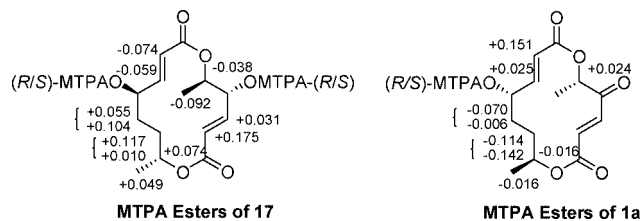
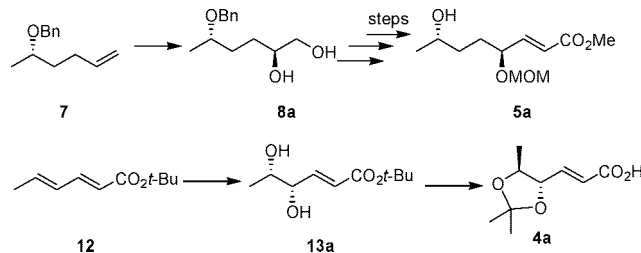


FIGURE 2. $\Delta\delta$ values for the MTPA esters of **17** and **1a** ($\Delta\delta = \delta_S - \delta_R$).

SCHEME 5. Synthesis of Fragments 5a and 4a



hydrolysis of the methyl ester group with LiOH in THF/ H_2O gave dihydroxy acid **3**, which underwent intramolecular ring closure by using Yamaguchi's protocol¹⁷ to afford 14-membered macrolide **16**. Then PDC oxidation of **16**, followed by deprotection of MOM ether with TFA,¹⁸ gave the 4-keto-clonostachydiol **1**, whose spectral data (^1H and ^{13}C NMR) were in good agreement with the literature (see the Supporting Information). The specific optical rotation of synthetic 4-keto-clonostachydiol ($[\alpha]_D^{20} -76$ (c 0.1, CH_3OH)) is quite different from that of the natural one ($[\alpha]_D^{20} +49$ (c 0.33, CH_3OH)). To confirm our result and further determine the absolute configuration, the modified Mosher's method¹⁹ was adopted. Removal of the MOM group in **16** with TFA afforded 4-*epi*-clonostachydiol **17**. Upon treatment of **17** with (*R*)- and (*S*)-MTPACl [α -methoxy- α -(trifluoromethyl)phenylacetyl chloride], (*S*)- and (*R*)-MTPA esters were obtained, respectively. ^1H – ^1H COSY data enabled assignment of the proton signals for MTPA ester. Analysis of the $\Delta\delta_{S-R}$ values (Figure 2) for the two diastereomeric esters confirmed the absolute configuration of **17** at C-4 as *R* and that at C-10 as *R*, but the optical rotation symbols of synthetic product and natural product were opposite, indicating that our synthetic 4-keto-clonostachydiol **1** was actually the enantiomer of the natural product, which prompted our further synthesis toward its antipode.

The antipode of **1** was then prepared from precursors **7** and **12** through similar routes. Treatment of benzyl ether **7** with AD-mix- α produced diol **8a**, which was converted into **5a** according to the above procedures (see the Supporting Information). Fragment **4a** was prepared from dienoate **12** following the same pathway of **4** except the AD-mix- β reagent was replaced by AD-mix- α (Scheme 5).

Condensation of fragments **4a** and **5a** under Yamaguchi condition provided enoate **14a** in a yield of 76%, while Keck protocol only yielded 35%.²⁰ Consecutive manipulation, including deprotection, hydrolysis, esterification, oxidation, and deprotection, finally led to (+)-4-keto-clonostachydiol **1a** (Scheme 6) (see the Supporting Information). The spectral data (^1H and ^{13}C NMR) and optical rotation of synthetic (+)-4-keto-clonostachydiol **1a** were both in accord with the reported ones

(8) Kolb, H. C.; VanNieuwenhze, M. S.; Sharpless, K. B. *Chem. Rev.* **1994**, *94*, 2483–2547.

(9) Huang, S. L.; Omura, K.; Swern, D. J. *Org. Chem.* **1978**, *43*, 4537–4538.

(10) (a) Horner, L.; Hoffmann, H.; Wippel, H. G. *Chem. Ber.* **1958**, *91*, 61–63. (b) Wadsworth, W. S.; Emmons, W. E. *J. Am. Chem. Soc.* **1961**, *83*, 1733–1738.

(11) Yadav, J. S.; Chandrashekar, S.; Sumitra, G.; Rajashekar, K. *Tetrahedron Lett.* **1996**, *37*, 6603–6606.

(12) Xu, D.; Crispino, G. A.; Sharpless, K. B. *J. Am. Chem. Soc.* **1992**, *114*, 7570–7571.

(13) (a) Sunazuka, T.; Hirose, T.; Harigaya, Y.; Takamatsu, S.; Hayashi, M.; Komiyama, K.; Omura, S.; Sprengeler, P. A.; Smith, A. B. *J. Am. Chem. Soc.* **1997**, *119*, 10247–10248. (b) Sunazuka, T.; Hirose, T.; Chikaraishi, N.; Harigaya, Y.; Hayashi, M.; Komiyama, K.; Sprengeler, P. A.; Smith, A. B., III; Omura, S. *Tetrahedron* **2005**, *61*, 3789–3803.

(14) Fletcher, S.; Jorgensen, M. R.; Miller, A. D. *Org. Lett.* **2004**, *6*, 4245–4248.

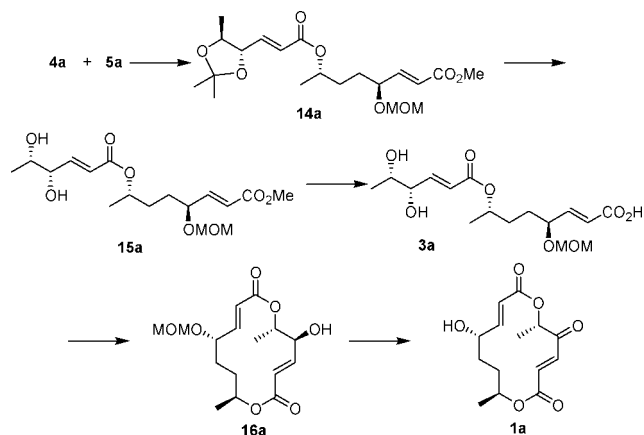
(15) Mitsunobu, O. *Synthesis* **1981**, 1–28.

(16) Sasai, H.; Shibasaki, M. *Tetrahedron Lett.* **1987**, *28*, 333–336.

(17) Inanaga, J.; Hirata, K.; Saeki, H.; Katsuki, T.; Yamaguchi, M. *Bull. Chem. Soc. Jpn.* **1979**, *52*, 1989–1993.

(18) Matsuya, Y.; Kawaguchi, T.; Nemoto, H. *Org. Lett.* **2003**, *5*, 2939–2941.

(19) Ohtani, I.; Kusumi, T.; Kashman, Y.; Kakisawa, H. *J. Am. Chem. Soc.* **1991**, *113*, 4092–4096.

SCHEME 6. Synthesis of (+)-4-Keto-Colonostachydiol **1a**

of the natural product. Upon treatment of **1a** with (*R*)- and (*S*)-MTPACl, (*S*)- and (*R*)-MTPA esters were obtained, respectively. Analysis of the $\Delta\delta_{S-R}$ values for the two diastereomeric esters demonstrated that the absolute configuration of **1a** at C-10 was *S* (Figure 2). Therefore, we corrected the absolute configuration of the natural product 4-keto-clonostachydiol **1** as (5*S*,10*S*,13*S*).

In summary, we have finished the total synthesis of both (+)- and (–)-4-keto-clonostachydiol through a highly stereoselective route, and the result was further confirmed by Mosher's method. The absolute configuration of 4-keto-clonostachydiol was revised as (5*S*,10*S*,13*S*) on the basis of our current synthesis.

Experimental Section

(2*R*,5*S*)-5-(Benzyloxy)hexane-1,2-diol (8). A mixture of AD-mix- β (7.0 g, 5 mmol) in 50 mL of *t*-BuOH/H₂O (1:1 v:v) was stirred at rt for 15 min, and then cooled to 0 °C. To this solution was added benzyl ether **7** (0.95 g, 5 mmol). The reaction mixture was stirred at 0 °C for 48 h and then quenched with Na₂SO₃ (7.5 g) at 0 °C within 0.5 h. EtOAc was added to the reaction mixture, and the aqueous layer was further extracted with EtOAc twice. The combined organic layers were dried over Na₂SO₄ and the solvents were evaporated. The crude product was purified by column chromatography on silica gel (hexanes/EtOAc 1:1) to give 1.03 g of the corresponding diol **8** (92%) as a colorless oil: $[\alpha]_D^{20} +29$ (*c* 1.18, CHCl₃); ¹H NMR (300 MHz, CDCl₃) δ 7.26–7.34 (m, 5 H), 4.59 (d, *J* = 11.4 Hz, 1 H), 4.20 (d, *J* = 11.4 Hz, 1 H), 3.52–3.62 (m, 3 H), 3.52–3.62 (m, 2 H), 2.79 (s, 1 H), 1.59–1.70 (m, 2 H), 1.47–1.58 (m, 2 H), 1.21 (d, *J* = 6.3 Hz, 3 H); ¹³C NMR (75 MHz, CDCl₃) δ 138.3, 128.4, 127.8, 127.6, 74.8, 72.1, 70.3, 66.7, 32.7, 29.0, 19.3; IR (film) ν_{\max} 3384, 2930, 1066, 738, 698 cm^{–1}; HRMS (ESIMS) calcd for C₁₃H₂₁O₃ [M + H]⁺ 225.1485, found 225.1492.

Dienoate 14. To a solution of **4** (400 mg, 1.6 mmol) and triphenylphosphine (852 mg, 3.2 mmol) in benzene (20 mL) was added a solution of DIAD (657 mg, 3.2 mmol) and acid **5** (358 mg, 1.9 mmol) in benzene (20 mL) over 30 min at rt. The reaction mixture was stirred overnight, and then quenched with H₂O (40 mL). This mixture was extracted with CHCl₃, washed with brine, and dried by Na₂SO₄, then the solvents were evaporated. The crude product was purified by column chromatography on silica gel (hexanes/EtOAc 10:1) to give 572 mg of **15** (85%) as a colorless oil: $[\alpha]_D^{20} +19$ (*c* 1.0, CHCl₃); ¹H NMR (300 MHz, CDCl₃) δ 6.81 (m, 2 H), 6.09 (d, *J* = 15.3 Hz, 1 H), 5.98 (d, *J* = 15.9 Hz, 1 H), 4.97 (m, 1 H), 4.58 (dd, *J* = 14.7, 6.3 Hz, 2 H), 4.18 (m, 1 H), 4.05 (m, 1 H), 3.83 (m, 1 H), 3.73 (s, 3 H), 3.35 (s, 3 H), 1.64 (br, 4 H), 1.46 (s, 4 H), 1.30 (d, *J* = 6 Hz, 3 H), 1.23 (d, *J* = 6 Hz, 3 H); ¹³C NMR (75 MHz, CDCl₃) δ 166.5, 165.4, 147.5, 143.3, 123.1,

121.7, 109.2, 94.6, 81.6, 76.4, 74.8, 70.9, 55.6, 51.6, 31.4, 30.7, 27.2, 26.6, 19.9, 16.6; IR (film) ν_{\max} 3365, 2983, 1721, 1272, 1095, 1036, 984, 919 cm^{–1}; HRMS (ESIMS) calcd for C₂₀H₃₆NO₈ [M + NH₄]⁺ 418.2435, found 418.2426.

Macrolide 16. To a solution of *seco*-acid **3** (40 mg, 0.12 mmol) in anhydrous THF (1 mL) under argon was added Et₃N (63 μ L, 0.46 mmol), after 30 min, followed by trichlorobenzoyl chloride (20 μ L, 0.47 mmol). The resulting heterogeneous mixture was stirred at room temperature for 18 h. The reaction mixture was diluted with dry toluene (100 mL) and the supernatant solution was transferred to a flask. The resulting solution was slowly added (5 mL/h) to a refluxing solution of DMAP (107 mg, 0.88 mmol) in toluene (50 mL) over 20 h. The mixture was refluxed for a further 11 h, and then diluted with diethyl ether. The organic layer was washed with brine and dried by MgSO₄, then the solvents were evaporated. The crude product was purified by column chromatography on silica gel (hexanes/EtOAc 2:1) to give 28 mg of **17** (75%) as a colorless oil: $[\alpha]_D^{20} +22$ (*c* 2.7, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ 6.78 (m, 2 H), 6.15 (d, *J* = 15.6 Hz, 1 H), 5.89 (d, *J* = 15.6 Hz, 1 H), 5.32 (m, 1 H), 5.19 (m, 1 H), 4.60 (m, 2 H), 4.50 (m, 1 H), 4.45 (m, 1 H), 3.36 (m, 1 H), 2.03–2.09 (m, 1H), 1.53–1.76 (m, 3 H), 1.44 (d, *J* = 6.3 Hz, 3 H), 1.20 (d, *J* = 6 Hz, 3 H); ¹³C NMR (100 MHz, CDCl₃) δ 165.8, 165.7, 150.0, 146.8, 122.2, 121.5, 94.6, 74.0, 73.2, 71.1, 69.4, 55.5, 27.1, 26.1, 17.4, 15.9; IR (film) ν_{\max} 3502, 2984, 1713, 1361, 1223, 1043, 513 cm^{–1}; HRMS (ESIMS) calcd for C₁₆H₂₈NO₇ [M + NH₄]⁺ 346.1860, found 346.1870.

4-Keto-Clonostachydiol 1. Under Ar atmosphere, pyridinium dichromate (PDC) (515 mg, 2.88 mmol) was added in portions to a stirred solution of the alcohol **17** (75 mg, 0.23 mmol) and molecular sieves 4Å (450 mg) in CH₂Cl₂ (10 mL) at 0 °C. After continuous stirring for 1 h at rt, the reaction mixture was diluted with ether (50 mL) and filtered through Celite. The solvents were evaporated. The crude product was purified by column chromatography on silica gel (hexanes/EtOAc 5:1) to give 59 mg of the corresponding ketone (80%) as a colorless oil: $[\alpha]_D^{20} -44$ (*c* 1.3, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ 7.31 (d, *J* = 16 Hz, 1 H), 7.02 (dd, *J* = 15.6, 4 Hz, 1 H), 6.48 (d, *J* = 16 Hz, 1 H), 6.14 (dd, *J* = 15.6, 1.2 Hz, 1 H), 5.31 (q, *J* = 7.2 Hz, 1 H), 5.05 (m, 1 H), 4.64 (m, 2 H), 4.44 (m, 1 H), 3.38 (s, 3 H), 1.81–1.92 (m, 3 H), 1.60 (m, 1 H), 1.56 (d, *J* = 6.4 Hz, 3 H), 1.28 (d, *J* = 6.4 Hz, 3 H); ¹³C NMR (100 MHz, CDCl₃) δ 199.5, 165.7, 164.1, 150.1, 135.3, 131.1, 120.6, 94.7, 75.6, 74.0, 72.0, 55.7, 28.3, 28.1, 18.5, 16.4; IR (film) ν_{\max} 3420, 2935, 1722, 1263, 1042, 981 cm^{–1}; HRMS (ESIMS) calcd for C₁₆H₂₆NO₇ [M + NH₄]⁺ 344.1704, found 344.1710.

Trifluoroacetic acid (1.5 mL) was added to a solution of the ketone (13 mg, 0.04 mmol) in CH₂Cl₂ (0.5 mL) at 0 °C under Ar atmosphere. After continuous stirring for 2 h at rt the solvents were evaporated. The crude product was purified by column chromatography on silica gel (hexanes/EtOAc 2:1) to give 8 mg of 4-keto-clonostachydiol **1** (85%) as a colorless oil: $[\alpha]_D^{20} -76$ (*c* 0.1, CH₃OH); ¹H NMR (400 MHz, DMSO-*d*₆) δ 7.37 (d, *J* = 16 Hz, 1 H), 7.20 (dd, *J* = 15.6, 3.2 Hz, 1 H), 6.39 (d, *J* = 16 Hz, 1 H), 5.93 (dd, *J* = 15.6, 2 Hz, 1 H), 5.21 (d, *J* = 4.4 Hz, 1 H), 5.16 (q, *J* = 7.2 Hz, 1 H), 4.92 (m, 1 H), 4.36 (br, 1 H), 1.89 (m, 1 H), 1.68 (m, 1 H), 1.56 (m, 1 H), 1.49 (d, *J* = 7.2 Hz, 4 H), 1.20 (d, *J* = 6.4 Hz, 3 H); ¹³C NMR (100 MHz, DMSO-*d*₆) δ 200.3, 165.8, 164.0, 154.5, 136.8, 129.9, 118.1, 75.4, 72.0, 68.6, 30.4, 27.6, 18.6, 15.9; IR (film) ν_{\max} 3375, 2990, 1764, 1243, 1058 cm^{–1}; HRMS (ESIMS) calcd for C₁₄H₂₂NO₆ [M + NH₄]⁺ 300.1442, found 300.1444.

Acknowledgment. We are grateful for the generous financial support by the NSFC (QT program, 20872054, 20732002), NCET-05-0879.

Supporting Information Available: General experimental details and spectral data for all compounds. This material is available free of charge via the Internet at <http://pubs.acs.org>.