

Available online at www.sciencedirect.com



Bioorganic & Medicinal Chemistry Letters

Bioorganic & Medicinal Chemistry Letters 14 (2004) 4987-4990

Asymmetric synthesis of the stereoisomers of 11,12,15(S)-trihydroxyeicosa-5(Z),8(Z),13(E)-trienoic acid, a potent endothelium-derived vasodilator

J. R. Falck,^{a,*} Deb K. Barma,^a Suchismita Mohapatra,^a A. Bandyopadhyay,^a Komandla Malla Reddy,^a Jianjun Qi^a and William B. Campbell^b

^aDepartment of Biochemistry, University of Texas Southwestern Medical Center, Dallas, TX 75390-9038, USA ^bDepartment of Pharmacology and Toxicology, Medical College of Wisconsin, Milwaukee, WI 53226, USA

> Received 19 May 2004; accepted 7 July 2004 Available online 31 July 2004

Abstract—The four stereoisomers of the endothelial-derived vasorelaxant 11,12,15(*S*)-trihydroxyeicosatrienoic acid [1, 11,12,15(*S*)-THETA] were prepared by a triply convergent, asymmetric route that exploited the stereospecific, copper mediated cross-coupling of α , β -dialkoxystannanes with organic electrophiles and the utility of dialkylthionocarbamates as orthogonal alcohol protective groups. Only 11(*R*),12(*S*),15(*S*)-THETA was comparable to natural material by HPLC, GC/MS, and in vitro bioassay. © 2004 Elsevier Ltd. All rights reserved.

The endothelial lining of mammalian blood vessels generates¹ a variety of vasoactive substances, for example, prostacyclin (PGI₂),² nitric oxide,³ epoxyeicosatrienoic acids,⁴ and endothelin.⁵ As a part of the intense search for additional endogenous vasomodulators,⁶ our laboratories reported the isolation of triol 1, which is generated by rabbit aortic endothelium from arachidonic acid via the 15(S)-lipoxygenase pathway.⁷ The structure of 1 was assigned as 11,12,15(S)-trihydroxyeicosatrienoic acid [11,12,15(S)-THETA] based upon a combination of spectral and biochemical studies; however, the minute amounts available from natural sources precluded a complete stereochemical assignment.⁸ In the rabbit aorta, 1 mediates acetylcholine-induced vasodilation⁹ and its production is stimulated by interleukin-13 treatment.¹⁰ To unambiguously establish the absolute configuration of 1 as well as expedite additional pharmacological evaluations, we herein report a concise, asymmetric synthesis of all four stereoisomers of 11,12,15(S)-THETA (1) and their comparisons with natural material. Our triply convergent strategy is summarized in Figure 1 and highlights two synthetic procedures recently introduced by our laboratories,

0960-894X/\$ - see front matter @ 2004 Elsevier Ltd. All rights reserved. doi:10.1016/j.bmcl.2004.07.019



Figure 1. Retrosynthetic analysis.

viz., (i) the preparation and stereospecific, copper mediated cross-coupling of chiral α,β -dialkoxystannanes¹¹ with organic electrophiles and (ii) the utility of dialkylthionocarbamates as orthogonal protecting groups.¹²

Retron 3^{13} was conveniently accessed via diastereoselective addition¹¹ of zinc bis-(*n*-tributylstannane) to prochiral aldehyde (*R*)-2,3-*O*-isopropylidene-D-glyceraldehyde¹⁴ (5) (Scheme 1). Thionocarbamoylation¹² of the resultant *anti*-adduct **6** evolved **3** in good overall yield. The *syn*-adduct **8** was obtained from **6** by a conventional Mitsunobu inversion/saponification sequence via benzoate **7**.

^{*} Corresponding author. Tel.: +1-214-648-2406; fax: +1-214-648-6455; e-mail: j.falck@utsouthwestern.edu



Scheme 1. Reagents and conditions: (a) $Zn(Bu_3Sn)_2$, THF, $-78 \,^{\circ}C$, 4h; (b) 1,1'-Im₂C(S), DMAP, then pyrrolidine, CH₂Cl₂, 23 $^{\circ}C$, 8h; (c) DIAD, Ph₃P, PhCO₂H, C₆H₆, 23 $^{\circ}C$, 24h; (d) K₂CO₃, MeOH, 45 $^{\circ}C$, 6h.

Bromo-ester **4** was readily prepared (Scheme 2) from 2-butyn-1,4-diol (9) via mono-tosylation and coppercatalyzed alkynylation¹⁵ with commercial methyl 5hexynoate (**11**). Semi-hydrogenation of the resultant diyne **10**¹⁶ over P-2¹⁷ Ni/H₂ and exchange of the alcohol for bromide using PBr₃ completed the synthesis of the C(1)-C(10) moiety.

Copper mediated cross-coupling¹¹ of chiral stannane 3with bromide 4 smoothly generated ester 12 with retention of configuration at the C(11)-stereocenter (Scheme 3). Mild acidic hydrolysis of the acetonide and low temperature sodium periodate cleavage of the liberated diol transformed 12 into aldehyde 13. Subsequent Nozaki-Kishi addition¹⁸ of vinyl iodide 2^{19} provided a mixture (7:3) of diastereomeric triols 14 and 16, which was most conveniently separated chromatographically at a later stage.20 Concomitant cleavage of the thionocarbamate,¹² silyl ether, and methyl ester protecting groups by basic hydrogen peroxide and re-esterification with diazomethane afforded the methyl esters of 11(S), 12(R), 15(S)-THETA (15) and 11(S), 12(S), 15(S)-THE-TA (17), respectively, in good yield. The relative stereochemistries of 15 and 17 were confirmed as erythro and



Scheme 4. Synthesis of triols 19 and 20.

threo, respectively, by ${}^{1}H/{}^{13}C$ analysis of the corresponding 11,12-acetonides.²¹

Thionocarbamoylation¹² of **8** furnished chiral stannane **18**, the C(1)-epimer of **3**, which was then carried through the sequence in Scheme 3 in comparable yields to give methyl 11(R), 12(S), 15(S)-THETA (**19**) and 11(R), 12(R), 15(S)-THETA (**20**) (Scheme 4).²⁰

Saponification (LiOH, THF/H₂O, 23 °C, 6–10h, 90– 95%) of **15**, **17**, **19**, and **20** and extractive isolation furnished the corresponding free acids as colorless oils. Of the four stereoisomers, only 11(R), 12(S), 15(S)-THE-TA (**19**) was identical with natural material by normal and reverse-phase HPLC, GC/MS, and in vitro bioassay.⁸

In conclusion, we describe the first synthesis of all four stereoisomers of **1** and validate novel methodology for creating carbon–carbon bonds at heteroatom-substituted chiral centers and for the orthogonal protection/ deprotection of alcohols using thionocarbamate protective groups. The results of ongoing studies of the physiological significance of this new class of vasorelaxants will be reported elsewhere.



Scheme 2. Reagents and conditions: (a) TsCl, Py, CH₂Cl₂, 1 h, 0 °C, 85%; (b) 11, NaI, CuI, K₂CO₃, DMF, -20 °C, 2 h; 23 °C, 12 h, 89%; (c) Ni (P-2), H₂ (1 atm), EtOH, 23 °C, 12 h, 98%; (d) PBr₃, Et₂O, 0 °C, 1 h, 95%.



Scheme 3. Reagents and conditions: (a) CuCN (7mol%), THF, 50°C, 4h; (b) 80% aq HOAc, 23°C, 12h, 95%; (c) NaIO₄, THF/H₂O (3:1), -10°C, 1h, 85%; (d) 2, NiCl₂ (cat)/CrCl₂, DMF, 23°C, 12h, 82%; (e) H₂O₂, NaOH, MeOH, 23°C, 10h; CH₂N₂, 85%.

Acknowledgements

The authors express their appreciation to Dr. Raymond E. Conrow for critically reviewing the manuscript. Nancy Spitzbarth and Erik Edwards are thanked for expert technical assistance. Financial support provided by the Robert A. Welch Foundation, NIH (GM 31278, DK38226, HL-37981).

References and notes

- 1. Lind, L. Lipids 2002, 37, 1-15.
- (a) Moncada, S.; Gryglewski, R.; Bunting, S.; Vane, J. R. *Nature* 1976, 263, 663–665; (b) Raz, A.; Isakson, P. C.; Minkes, M. S.; Needleman, P. J. Biol. Chem. 1977, 252, 1123–1126.
- (a) Furchgott, R. F.; Zawadzki, J. V. Nature 1980, 288, 373–376;
 (b) Ignarro, L. J.; Buga, G. M.; Wood, K. S.; Byrns, R. E. Proc. Natl. Acad. Sci. U.S.A. 1987, 84, 9265–9269.
- Campbell, W. B.; Gebremedhin, D.; Pratt, P. F.; Harder, D. R. Circ. Res. 1996, 78, 415–423.
- Yanagisawa, M.; Kuihara, H.; Kimura, S.; Tomobe, Y.; Kobayashi, M.; Mitsui, Y.; Yazaki, Y.; Goto, Y. K.; Masaki, T. *Nature* **1988**, *332*, 411–415.
- (a) Pace-Asciak, C. R.; Reynaud, D.; Demin, P.; Nigam, S. *Adv. Exp. Med. Biol.* **1999**, 447, 123–132; (6) Narumiya, S.; Salmon, J. A.; Cottee, F. H.; Weatherley, B. C.; Flower, R. J. *J. Biol. Chem.* **1981**, 256, 9583–9592.
- Pfister, S. L.; Spitzbarth, N.; Zeldin, D. C.; Lafite, P.; Mansuy, D.; Campbell, W. B. *Arch. Biochem. Biophys.* 2003, 420, 142–152.
- Pfister, S. L.; Spitzbarth, N.; Nithipatikom, K.; Edgemond, W. S.; Falck, J. R.; Campbell, W. B. J. Biol. Chem. 1998, 273, 30879–30887.
- Campbell, W. B.; Spitzbarth, N.; Gauthier, K. M.; Pfister, S. L. Am. J. Physiol. 2003, 285, H2648–H2656.
- Tang, X.; Spitzbarth, N.; Kuhn, H.; Chaitidis, P.; Campbell, W. B. Arterioscler. Throm. Vasc. Biol. 2003, 23, 1768–1774.
- Mohapatra, S.; Bandyopadhyay, A.; Barma, D. K.; Capdevila, J. H.; Falck, J. R. Org. Lett. 2003, 5, 4759–4762.
- 12. Barma, D. K.; Bandyopadhyay, A.; Capdevila, J. H.; Falck, J. R. Org. Lett. 2003, 5, 4755–4757.
- 13. Spectral and physical data for 3: ¹H NMR (CDCl₃, 400 MHz) δ 0.80-1.00 (m, 15H), 1.27-1.35 (m, 9H), 1.42 (s, 3H), 1.48-1.56 (m, 6H), 1.90-2.01 (m, 4H), 3.47-3.64 (m, 2H), 3.70–3.74 (m, 3H), 3.90–3.94 (m, 1H), 4.38–4.42 (m, 1H), 5.73–5.79 (m, 1H); $^{13}\mathrm{C}$ NMR (CDCl₃, 75 MHz) δ 10.87, 13.78, 24.66, 25.20, 25.84, 26.64, 27.52, 29.10, 47.77, 52.13, 65.08, 76.65, 79.10, 108.13, 185.51; HRMS (CI, CH₄) calcd for $C_{19}H_{37}NO_3SSn: 479.1516 (M+H-Bu)^+$ found 479.1512. 4: ¹H NMR (CDCl₃, 300 MHz) δ 1.60-1.77 (m, 2H), 2.02–2.18 (m, 2H), 2.34 (t, 2H, J = 7.8 Hz), 2.88 (apparent t, 2H, J = 6.4 Hz), 3.67 (s, 3H), 4.02 (d, 2H, J = 9.0 Hz), 5.40–5.82 (m, 4H); Anal. Calcd for C₁₁H₁₇BrO₂: C, 50.59; H, 6.56. Found: C, 50.68; H, 6.62. **6**: ¹H NMR (CDCl₃, 300 MHz) δ 0.87–0.96 (m, 15H), 1.28-1.35 (m, 6H), 1.39 (s, 3H), 1.43 (s, 3H), 1.48-1.54 (m, 6H), 1.99-2.01 (m, 1H), 3.83-3.88 (m, 1H), 3.92-3.97 (m, 1H), 4.19–4.21 (m, 1H), 4.30–4.36 (m, 1H); $[\alpha]_{D}^{23}$ 11.9 (c 1.65, CHCl₃). 7: ¹H NMR (CDCl₃, 400 MHz) δ 0.85–0.98 (m, 15H), 1.21–1.62 (m, 18H), 3.77 (t, 1H, J = 7.9 Hz), 3.97 (t, 1H, J = 6.4 Hz), 4.41-4.49 (m, 1H), 5.12-5.21 (m, 1H),7.40-7.58 (m, 3H), 8.04 (d, 2H, J = 7.6 Hz); Anal. Calcd

for C₂₅H₄₂O₄Sn: C, 57.16; H, 8.06. Found: C, 57.30; H, 8.29. 8: ¹H NMR (CDCl₃, 400 MHz) δ 0.89–0.97 (m, 15H), 1.30-1.36 (m, 6H), 1.37 (s, 3H), 1.44 (s, 3H), 1.48-1.55 (m, 6H), 1.82 (d, 1H, J = 6.7 Hz), 3.75 (t, 1H, J = 7.30 Hz), 3.91-3.96 (m, 2H), 4.39 (q, 1H, J = 6.4 Hz); ¹³C NMR (CDCl₃, 75 MHz) δ 9.11, 13.84, 25.82, 26.55, 27.59, 29.26, 65.36, 65.58, 81.18, 108.89. 12: ¹H NMR (CDCl₃, 400 MHz) δ 1.36 (s, 3H), 1.46 (s, 3H), 1.61–1.73 (m, 2H), 1.90–1.98 (m, 4H), 2.05–2.12 (m, 2H), 2.28–2.34 (m, 2H), 2.38–2.61 (m, 2H), 2.79 (apparent t, 2H, J = 6.4 Hz), 3.49-3.63 (m, 2H), 3.62 (s, 3H), 3.66-3.67 (m, 3H), 3.70-3.75 (m, 2H), 3.90-4.20 (m, 2H), 4.26-4.36 (m, 1H), 5.01-5.15 (m, 1H), 5.34-5.50 (m, 2H), 5.60-5.72 (m, 1H), 5.81–5.83 (m, 1H); Anal. Calcd for C₂₂H₃₅NO₅S: C, 62.09; H, 8.29. Found: C, 62.16; H, 8.10. 13: ¹H NMR (CDCl₃, 400 MHz) & 1.60-1.70 (m, 2H), 1.84-2.00 (m, 4H), 2.01-2.10 (m, 2H), 2.20-2.32 (m, 2H), 2.50-2.65 (m, 2H), 2.78 (apparent t, 2H, J = 6.4 Hz), 3.47 (t, 2H, J = 6.9 Hz), 3.55– 3.62 (m, 2H), 3.81 (s, 3H), 4.04 (m, 1H), 5.33-5.55 (m, 4H), 9.67 (d, 1H, J = 1.5Hz); HRMS (CI, CH₄) calcd for $C_{18}H_{28}NO_4S$: 354.1739 (M+H)⁺, found 354.1734. 15: TLC: $R_f \sim 0.25$ (80% EtOAc/hexane); ¹H NMR (CDCl₃, 400 MHz) δ 0.89 (t, 3H, J = 6.0 Hz), 1.20–1.42 (m, 6H), 1.43-1.60 (m, 2H), 1.61-1.80 (m, 2H), 2.00-2.40 (m, 6H), 2.70-2.90 (m, 2H), 3.61 (s, 3H), 3.62-3.70 (m, 1H), 4.05-4.14 (m, 2H), 5.26-5.42 (m, 3H), 5.43-5.52 (m, 1H), 5.77 (dd, J = 6, 16Hz, 1H), 5.85 (dd, J = 4.8, 16Hz, 1H); ¹³C NMR (CDCl₃, 75 MHz) δ 14.25, 22,81, 24.92, 25.36, 26.03, 26.78, 31.79, 31.94, 33.61, 37.38, 51.82, 72.59, 74.16, 74.70, 125.50, 128.52, 128.75, 129.30, 131.37, 136.86, 174.56; $[\alpha]_D^{23}$ 2.66 (*c* 0.45, CHCl₃); HRMS (CI, CH₄) calcd for $\bar{C}_{21}H_{37}O_5$: 369.2641 (M+H)⁺, found 369.2639. 17: TLC: $R_f \sim 0.27$ (80% EtOAc/hexane); ¹H NMR (CDCl₃, 400 MHz) δ 0.89 (t, 3H, J = 8.8 Hz), 1.20-1.42 (m, 6H), 1.43-1.60 (m, 2H), 1.61-1.80 (m, 2H), 2.00-2.18 (m, 2H), 2.20-2.40 (m, 4H), 2.80 (apparent t, 2H, J = 6.4 Hz), 3.53 (br s, 1H), 3.67 (s, 3H), 4.02 (br s, 1H), 4.10–4.20 (m, 1H), 5.30–5.60 (m, 4H), 5.73 (dd, J = 6, 16Hz, 1H), 5.85 (dd, J = 6, 16Hz, 1H); ¹³C NMR (CDCl₃, 75 MHz) & 14.25, 22.81, 24.91, 25.30, 26.01, 26.03, 31.42, 31.95, 33.60, 37.43, 51.81, 72.32, 74.32, 74.79, 125.25, 128.78, 129.33, 129.67, 131.56, 136.71, 175.89; $[\alpha]_D^{23}$ -8.5 (c 0.2, CHCl₃); HRMS (CI, CH₄) calcd for $C_{21}H_{37}O_5 (M+H)^+ m/z$ 369.2641, found 369.2641. 18: ¹H NMR (CDCl₃, 400 MHz) δ 0.80–1.00 (m, 15H), 1.27-1.35 (m, 9H), 1.42 (s, 3H), 1.48-1.56 (m, 6H), 1.90-2.01 (m, 4H), 3.47-3.64 (m, 2H), 3.70-3.74 (m, 3H), 3.90-3.94 (m, 1H), 4.38-4.42 (m, 1H), 5.73-5.79 (m, 1H); ¹³C NMR (CDCl₃, 75 MHz) δ 10.87, 13.78, 24.66, 25.20, 25.84, 26.64, 27.52, 29.10, 47.77, 52.13, 65.08, 76.65, 79.10, 108.13, 185.51; HRMS (CI, CH₄) calcd for C₁₉H₃₇NO₃SSn: 479.1516 (M+H-Bu)⁺, found 479.1520. **19**: TLC: $R_{f} \sim 0.20$ (80% EtOAc/hexane); ¹H NMR $(CDCl_3, 400 \text{ MHz}) \delta 0.89 \text{ (t, 3H, } J = 6.3 \text{ Hz}), 1.21-1.41$ (m, 4H), 1.45–1.60 (m, 2H), 1.62–1.78 (m, 2H), 2.02–2.18 (m, 2H), 2.20-2.40 (m, 6H), 2.72-2.85 (m, 2H), 3.68 (s, 3H), 3.69–3.74 (m, 1H), 4.12–4.20 (m, 2H), 5.32–5.60 (m, 4H), 5.74 (dd, *J* = 5.6, 16 Hz, 1H), 5.82 (dd, *J* = 5.2, 16 Hz, 1H); ¹³C NMR (CDCl₃, 75 MHz) δ 14.23, 22,79, 24.89, 25.33, 26.00, 26.73, 30.39, 31.95, 33.57, 37.35, 51.80, 72.18, 74.28, 74.84, 126.62, 128.26, 128.81, 129.23, 131.11, 136.79, 174.57; $[\alpha]_D^{23}$ 7.24 (*c* 1.68, CHCl₃); HRMS (CI, CH₄) calcd for C₂₁H₃₇O₅: 369.2641 (M+H)⁺, found 369.2643. **20**: TLC: R_f~0.23 (80% EtOAc/hexane); ¹H NMR (CDCl₃, 400 MHz) δ 0.89 (t, 3H, J = 8.80 Hz), 1.20-1.41 (m, 6H), 1.42-1.60 (m, 2H), 1.61-1.80 (m, 2H), 2.02-2.18 (m, 2H), 2.23-2.40 (m, 4H), 2.80 (t, 2H, J = 6.4 Hz), 3.55 (m, 1H), 3.68 (s, 3H), 4.02 (br s, 1H), 4.10–4.20 (m, 1H), 5.32–5.60 (m, 4H), 5.73 (dd, J = 6,

16Hz, 1H), 5.85 (dd, J = 6, 16Hz, 1H); ¹³C NMR (CDCl₃, 75MHz) δ 14.25, 22,81, 24.93, 25.34, 25.99, 26.77, 31.37, 31.94, 33.61, 37.31, 51.81, 72.59, 74.22, 74.57, 125.33, 128.80, 129.28, 130.40, 131.36, 136.39, 174.54; $[\alpha]_{22}^{D}$ 21.9 (c0.8, CHCl₃); HRMS (CI, CH₄) calcd for C₂₁H₃₇O₅: 369.2641 (M+H)⁺, found 369.2643.

- Ermolenko, L.; Sasaki, N. A.; Potier, P. Synlett 2001, 1565–1566.
- Sonogashira, K. In Metal-Catalyzed Cross-Coupling Reactions; Diederich, F., Stang, P. J., Eds.; Wiley-VCH: New York, 1998; pp 203–229.
- Dasse, O.; Mahadevan, A.; Han, L.; Martin, B. R.; Marzo, V. D.; Razdan, R. K. *Tetrahedron* 2000, 56, 9195–9202.
- 17. Brown, C. A.; Ahuja, V. K. Chem. Commun. 1973, 15, 553–554.

- Jin, H.; Uenishi, J.; Christ, W. J.; Kishi, Y. J. Am. Chem. Soc. 1986, 108, 5644–5646.
- Davies, J.; Roberts, S. M.; Reynolds, D. P.; Newton, R. F. J. Chem. Soc., Perkin Trans. 1 1981, 1317–1320.
- 20. In practice, the Nozaki-Kishi adducts were most conveniently separated after removal of the alcohol protecting groups by NP-HPLC using Microsorb silica $(10 \times 250 \text{ mm})$, 6% i PrOH/0.1% HOAc/hexane, UV 215 nm: R_f of **15** = 17 min, R_f of **17** = 19 min, R_f of **19** = 25 min, R_f of **20** = 27 min.
- 21. The spectral values for the acetonide of **15** (¹H NMR: $J_{11,12} = 6.0$ Hz, ¹³C NMR: δ 78.4, 79.0) and the acetonide of **17** (¹H NMR: $J_{11,12} = 7.9$ Hz, ¹³C NMR: δ 80.7, 81.6) are consistent with literature *erythro* and *threo*-diol data, respectively. See: Kato, T.; Hirukawa, T.; Yano, M. *Bull. Chem. Soc. Jpn.* **1994**, 67, 839–842.