

## Tetracyclic Terpenoids from *Dasyscyphus niveus*, Dasyscyphins D and E

Johannes C. Liermann,<sup>†</sup> Heinz Kolshorn,<sup>‡</sup> Heidrun Anke,<sup>§</sup> Eckhard Thines,<sup>§</sup> and Till Opatz<sup>\*,†</sup>

Institut für Organische Chemie, Universität Hamburg, Martin-Luther-King-Platz 6, D-20146 Hamburg, Germany, Institut für Organische Chemie, Universität Mainz, Duesbergweg 10-14, D-55128 Mainz, Germany, Institut für Biotechnologie und Wirkstoff-Forschung (IBWF), Erwin-Schrödinger-Strasse 56, D-67663 Kaiserslautern, Germany

Received June 16, 2008

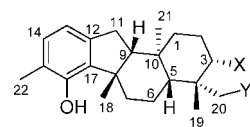
Cultures of the ascomycete *Dasyscyphus niveus* have yielded two new tetracyclic dasyscyphin-type terpenoids (**1** and **2**), and their structures were elucidated by NMR spectroscopy and X-ray crystallography. The absolute configuration of dasyscyphin D (**1**) was determined by synthesis and NMR spectroscopy of diastereomeric MTPA esters. Both compounds inhibited the germination of conidia of *Magnaporthe grisea* at 25 µg/mL.

The tiny, hairy, white, cup-shaped fruiting bodies of the ascomycete *Dasyscyphus niveus* (German: *Schneeweisses Haarbecherchen*) can be found growing on decaying oak or beech wood in European, North American, and New Zealandian forests.<sup>1</sup> Various compounds have recently been isolated from fermentation broths of this fungus. Structure elucidation of these compounds, the dasyscyphins, has been described along with their strong cytotoxic activity.<sup>2,3</sup> Here, we report on the isolation and structure elucidation of two novel structurally related terpenoids (**1** and **2**) from submerged cultures of *D. niveus* collected near Kaiserslautern.

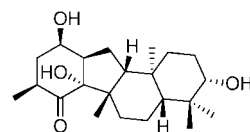
In a screening of extracts obtained from submerged cultures of ascomycetes, strain A79-88 exhibited antifungal activity, no antibacterial activity, and only very low cytotoxic activity. This selective biological activity prompted us to isolate the responsible compounds. The ascomycete was grown in YMG medium until the carbon source was consumed. The culture filtrate was extracted with ethyl acetate, and the resulting crude mixture was subjected to silica gel column chromatography (CC). Further purification by preparative HPLC yielded two nonpolar compounds, **1** (5.5 mg) as colorless crystals (mp 105–106 °C) and **2** (3.3 mg) as a yellow oil.

Analysis of <sup>13</sup>C NMR and DEPT data revealed the presence of 22 carbon atoms and 30 C–H protons in either case, suggesting that **1** and **2** were closely related structural isomers. The elemental composition for both compounds was determined to be C<sub>22</sub>H<sub>32</sub>O<sub>2</sub> by HRMS. This is in accordance with <sup>1</sup>H NMR data and proved the presence of 32 protons and thus two protons bound to heteroatoms. Furthermore, <sup>13</sup>C and DEPT spectra of compound **1** showed a methine carbon at δ 79.5, whereas the spectra of **2** show a methylene group at δ 65.3. This led to the conclusion that both compounds carry a hydroxy group, which is secondary in **1** and primary in **2**. The UV spectra of methanolic solutions of both compounds exhibited absorption maxima characteristic of a benzenoid system. In combination with a quaternary carbon atom resonating at δ 150.3 in both compounds, this was indicative for a substituted phenol. Tetrasubstitution of the aromatic ring was concluded from two ortho-coupled aromatic protons (AB spin system). The elemental composition of **1** and **2** requires seven double-bond equivalents, four of which are attributed to the benzene ring. As no further evidence for unsaturation was found, the molecule must contain three additional rings.

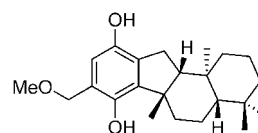
Compound **1** had five methyl groups, all of which appeared as singlets in the <sup>1</sup>H NMR spectrum. In contrast, compound **2** shows only four methyl singlets; thus, one of the methyl groups in **1** should



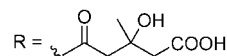
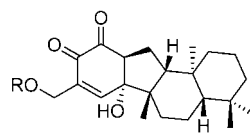
**1:** X = OH, Y = H, Dasyscyphin D  
**2:** X = H, Y = OH, Dasyscyphin E



Dasyscyphin A



Dasyscyphin B



Dasyscyphin C

be hydroxylated in **2**. Only compound **1** possessed two geminal methyl groups, whereas in both compounds one methyl group resonated at δ 2.20, characteristic for toluene-like aromatics. The substitution pattern of the phenol was established to be that of a 5,6-anellated ortho-cresole by HMBC and NOESY experiments. One of the aromatic protons in **1** showed NOE correlations to geminal methylene protons at δ 3.01 and 2.66, which was confirmed by a corresponding HMBC correlation. Both protons had COSY and NOESY correlations to the multiplet at δ 1.62 (<sup>1</sup>H), located on a tertiary center. An HMBC correlation of this proton as well as of the methyl protons at δ 1.23 with the ortho-position of the phenol suggested the presence of a five-membered ring with a methylated ring junction fused to the cresole moiety.

Analysis of further HMBC, COSY, and NOESY correlations revealed the presence of a fused trans-decalin system carrying an angular methyl group. In compound **1**, the carbinol center is bound to a quaternary carbon carrying the two remaining methyl groups. In its isomer **2**, the aliphatic OH group is instead located on one of the geminal methyl groups. The structures of **1** and **2** are based on a drimane-type sesquiterpene anellated to an indane ring system, a

\* To whom correspondence should be addressed. Tel: +49-(0)40-42838-4239. E-mail: opatz@chemie.uni-hamburg.de.

<sup>†</sup> Universität Hamburg.

<sup>‡</sup> Universität Mainz.

<sup>§</sup> Institut für Biotechnologie und Wirkstoff-Forschung (IBWF).



had reached the maximum and the glucose in the medium was consumed, the culture fluid (14.5 L) was separated from the mycelia and extracted with EtOAc (8 L). The mycelium containing no active compounds was discarded. The organic extract was dried with Na<sub>2</sub>SO<sub>4</sub> and concentrated to yield 4.2 g of crude product, which was further purified by silica gel chromatography (silica gel 60, Merck Darmstadt, 107 g, column size 17 cm × 4.5 cm). Elution with cyclohexane–EtOAc (7:3) yielded 98.1 mg of a product containing compounds **1** and **2**. The final purification by preparative HPLC (Merck LiChroSpher RP18, 5 μm, column 25 × 250 mm, flow 20 mL/min; isocratic mode in MeCN–H<sub>2</sub>O, 7:3 v/v) resulted in 5.5 mg of pure dasyscyphin D (**1**) and 3.3 mg of dasyscyphin E (**2**).

**Biological Assays.** Hela S3 (ATCC CCL 2.2 human cervix carcinoma) cell lines were grown in DMEM medium with 65 μg/mL penicillin G and 100 μg/mL streptomycin sulfate. The cells (10<sup>5</sup>/mL) were incubated in microtiter plates with the compounds at 37 °C in a humidified atmosphere containing 5% CO<sub>2</sub>. Viable cells were counted under the microscope after 24 and 48 h.

The antifungal activity was measured in a spore germination assay with *M. grisea* 70-15 conidia in water (5 × 10<sup>4</sup>/mL). Germination was evaluated after 18 h under the microscope and compared to the control containing no inhibitor. The number of ungerminated conidia or conidia with significantly shorter germ tubes (germ tube length <30% of the control) was assessed. A total of 3 × 100 conidia were counted. The antifungal spectrum of the compounds was evaluated using the conventional agar diffusion assay.

**Dasyscyphin D (1):** Colorless crystals (5.5 mg); mp 105–106 °C; [α]<sub>D</sub><sup>25</sup> +2.92 (c 0.49, CHCl<sub>3</sub>); UV (MeOH) λ<sub>max</sub> (log ε) 205 (4.53), 275 (2.91) nm; IR (KBr) ν<sub>max</sub> 3436, 2934, 1628, 1586, 1473, 1388, 1221, 1106, 1031, 798, 558 cm<sup>-1</sup>; <sup>1</sup>H NMR, COSY, NOESY (400 MHz, CDCl<sub>3</sub>) δ 6.87 (1H, d, *J* = 7.5 Hz, H-14), 6.62 (1H, d, *J* = 7.5 Hz, H-13), 4.53 (1H, s, OH-16), 3.26 (1H, dd, *J* = 11.3, 4.6 Hz, H-3), 3.01 (1H, dd, *J* = 16.6, 8.2 Hz, H-11a), 2.68 (1H, dt, *J* = 13.6, 6.2 Hz, H-7a), 2.66 (1H, br d, *J* = 16.6 Hz, H-11b), 2.20 (3H, s, H<sub>3</sub>-22), 1.78 (1H, mc, H-7b), 1.73 (1H, dd, *J* = 13.2, 3.5 Hz, H-1a), 1.71–1.58 (3H, m, H-2a, H-6a, H-9), 1.54 (1H, ddd, *J* = 12.8, 3.5, 1.5 Hz, H-2b), 1.42 (1H, dddd, *J* = 13.5, 11.0, 8.8, 4.8 Hz, H-6b), 1.31 (1H, br s, OH-3), 1.23 (3H, s, H<sub>3</sub>-18), 1.06 (1H, dd, *J* = 13.2, 4.3 Hz, H-1b), 1.01 (3H, s, H<sub>3</sub>-19), 0.99 (1H, dd, *J* = 11.0, 4.9 Hz, H-5), 0.78 (3H, s, H<sub>3</sub>-20), 0.52 (3H, s, H<sub>3</sub>-21); <sup>13</sup>C NMR, HSQC, HMBC (100.6 MHz, CDCl<sub>3</sub>) δ 150.3 (C, C-16), 143.8 (C, C-12), 136.0 (C, C-17), 129.1 (CH, C-14), 120.8 (C, C-15), 116.5 (CH, C-13), 79.5 (CH, C-3), 62.2 (CH, C-9), 51.2 (CH, C-5), 47.6 (C, C-8), 39.6 (CH<sub>2</sub>, C-1), 39.2 (C, C-4), 37.2 (C, C-10), 33.9 (CH<sub>2</sub>, C-7), 32.4 (CH<sub>2</sub>, C-11), 30.7 (CH<sub>3</sub>, C-18), 28.3 (CH<sub>3</sub>, C-19), 27.3 (CH<sub>2</sub>, C-2), 19.5 (CH<sub>2</sub>, C-6), 15.9 (CH<sub>3</sub>, C-21), 15.5 (CH<sub>3</sub>, C-20), 15.5 (CH<sub>3</sub>, C-22); APCIMS *m/z* pos. 311.2 [M – OH]<sup>+</sup> (100) neg. 309.2 [M – H<sub>3</sub>O]<sup>–</sup> (10), 327.2 [M – H]<sup>–</sup> (100); HREIMS *m/z* 328.2397 (calcd for C<sub>22</sub>H<sub>32</sub>O<sub>2</sub>, 328.2402).

**X-ray Crystallographic Data of 1·CH<sub>3</sub>OH.** Colorless single crystals of **1**·CH<sub>3</sub>OH were grown by slow evaporation of a methanolic solution of **1**. Data were obtained at 193 K on a Turbo CAD4 diffractometer with graphite-monochromated Cu Kα radiation. Formula C<sub>22</sub>H<sub>32</sub>O<sub>2</sub>·CH<sub>3</sub>OH, crystal size 0.032 × 0.064 × 0.265 mm<sup>3</sup>, orthorhombic, space group P2<sub>1</sub>2<sub>1</sub>2, *a* = 16.491(2) Å, *b* = 19.995(4) Å, *c* = 6.399(1) Å, *V* = 2110.0(6) Å<sup>3</sup>, *Z* = 4, *D* = 1.135 g cm<sup>-3</sup>, *R* = 0.0978, *R*<sub>w</sub> = 0.3326. CCDC-695149 contains the supplementary crystallographic data for this paper. These data can be obtained free of charge at [www.ccdc.cam.ac.uk/conts/retrieving.html](http://www.ccdc.cam.ac.uk/conts/retrieving.html) [or from the Cambridge Crystallographic Data Centre, 12 Union Road, Cambridge CB2 1EZ, UK; fax: (internat.) +44-1223/336-033; e-mail: [deposit@ccdc.cam.ac.uk](mailto:deposit@ccdc.cam.ac.uk)].

**Dasyscyphin E (2):** Yellow oil (3.3 mg); [α]<sub>D</sub><sup>25</sup> +1.07 (c 0.33, CHCl<sub>3</sub>); UV (MeOH) λ<sub>max</sub> (log ε) = 204 (4.59), 271 (2.99) nm; IR (KBr) ν<sub>max</sub> 3436, 2927, 1714, 1632, 1473, 1263, 1223, 1012, 797, 602 cm<sup>-1</sup>. <sup>1</sup>H NMR, COSY, NOESY (400 MHz, CDCl<sub>3</sub>) δ 6.86 (1H, d, *J* = 7.5 Hz, H-14), 6.62 (1H, d, *J* = 7.5 Hz, H-13), 4.50 (1H, s, OH-16), 3.73 (1H, br d, *J* = 10.9 Hz, H-20a), 3.42 (1H, d, *J* = 10.9 Hz, H-20b), 2.99 (1H, dd, *J* = 16.5, 7.8 Hz, H-11a), 2.79 (1H, m, H-7a), 2.63 (1H, d, *J* = 16.5 Hz, H-11b), 2.20 (3H, s, H<sub>3</sub>-22), 1.83 (1H, ddd, *J* = 13.7, 4.7, 3.1 Hz, H-3a), 1.76–1.55 (4H, m, H-1a, H-6a, H-7b, H-9), 1.48 (1H, m, H-2a), 1.41 (1H, m, H-2b), 1.34 (1H, m, H-6b), 1.23 (3H, s, H<sub>3</sub>-18), 1.15 (1H, dd, *J* = 11.6, 3.7 Hz, H-5), 1.04 (1H, br s, OH-20), 0.99 (3H, s, H<sub>3</sub>-19), 0.95 (1H, m, H-3b), 0.93 (1H, m, H-1b), 0.45

(3H, s, H<sub>3</sub>-21); <sup>13</sup>C NMR, HSQC, HMBC (100.6 MHz, CDCl<sub>3</sub>) δ 150.3 (C, C-16), 144.1 (C, C-12), 135.6 (C, C-17), 129.1 (CH, C-14), 120.8 (C, C-15), 116.5 (CH, C-13), 65.3 (CH<sub>2</sub>, C-20), 62.5 (CH, C-9), 53.4 (CH, C-5), 47.7 (C, C-8), 41.4 (CH<sub>2</sub>, C-1), 38.6 (C, C-4), 37.3 (C, C-10), 35.7 (CH<sub>2</sub>, C-3), 34.8 (CH<sub>2</sub>, C-7), 32.5 (CH<sub>2</sub>, C-11), 31.0 (CH<sub>3</sub>, C-18), 26.7 (CH<sub>3</sub>, C-19), 19.7 (CH<sub>2</sub>, C-6), 18.3 (CH<sub>2</sub>, C-2), 16.7 (CH<sub>3</sub>, C-21), 15.5 (CH<sub>3</sub>, C-22); APCIMS *m/z* pos. 311.2 [M – OH]<sup>+</sup> (100), neg. 309.2 [M – H<sub>3</sub>O]<sup>–</sup> (6), 327.2 [M – H]<sup>–</sup> (100); HREIMS *m/z* 328.2413 (calcd for C<sub>22</sub>H<sub>32</sub>O<sub>2</sub>, 328.2402).

**3,16-Bis(α-methoxy-α-(trifluoromethyl)phenylacetyl)dasyscyphin D (3).** Dasyscyphin D (**1**) (1.0 mg, 3.0 μmol) was dissolved in dry CH<sub>2</sub>Cl<sub>2</sub> (1 mL) under argon atmosphere. MTPA chloride (5.0 mg, 19.8 μmol), DMAP (1.0 mg, 8.2 μmol), and dry triethylamine (10 μL, 7.2 mg, 6.9 μmol) were added. The solution was stirred at room temperature for 1 h. Ether (5 mL) was added, and the resulting solution was washed with aqueous 1 M HCl (5 mL) and saturated aqueous Na<sub>2</sub>CO<sub>3</sub> (5 mL). The organic phase was dried over Na<sub>2</sub>SO<sub>4</sub> and the solvent was removed in vacuo to furnish the diesters as yellow oils.

**Compound 3a.** Prepared from (+)-MTPA chloride. Yield: 1.0 mg (44%). Characteristic <sup>1</sup>H NMR shifts: COSY, NOESY (500 MHz, CDCl<sub>3</sub>) δ 4.72 (H-3), 1.73 (H<sub>α</sub>-2), 1.65 (H<sub>β</sub>-2), 1.11 (H-1), 0.94 (H-5), 0.78 (H<sub>3</sub>-19), 0.73 (H<sub>3</sub>-20), 0.43 (H<sub>3</sub>-21).

**Compound 3b.** Prepared from (–)-MTPA chloride. Yield: 1.0 mg (44%). ESIMS: *m/z* 761 [M + H]<sup>+</sup> (46), 783 [M + Na]<sup>+</sup> (86), 799 [M + K]<sup>+</sup> (88). Characteristic <sup>1</sup>H NMR shifts: COSY, NOESY (500 MHz, CDCl<sub>3</sub>) δ 4.69 (H-3), 1.69 (H<sub>α</sub>-2), 1.55 (H<sub>β</sub>-2), 1.20 (H-5), 1.10 (H-1), 0.88 (H<sub>3</sub>-19), 0.75 (H<sub>3</sub>-20), 0.42 (H<sub>3</sub>-21).

**Acknowledgment.** We thank Dr. S. Franke (University of Hamburg) for the HREIMS data, Dr. D. Schollmeyer (University of Mainz) for the X-ray crystallographic analysis of dasyscyphin D (**1**), and the Kompetenzzentrum der integrierten Naturstoff-Forschung. The expert technical assistance of A. Spohn (IBWF) is gratefully acknowledged.

**Supporting Information Available:** <sup>1</sup>H, <sup>13</sup>C, and 2D NMR spectra of compounds **1** and **2** and CIF of **1**. This material is available free of charge via the Internet at <http://pubs.acs.org>.

## References and Notes

- (1) Saccardo, P. A. *Syll. Fung.* **1889**, 8, 437.
- (2) Rojas de la Parra, V.; Mierau, V.; Anke, T.; Sterner, O. *Tetrahedron* **2006**, 62, 1828–1832.
- (3) Mierau, V.; Rojas de la Parra, V.; Sterner, O.; Anke, T. *J. Antibiot.* **2006**, 59, 53–56.
- (4) Mukku, V. J. R. V.; Edrada, R. A.; Schmitz, F. J.; Shanks, M. K.; Chaudhuri, B.; Fabbro, D. *J. Nat. Prod.* **2003**, 66, 686–689.
- (5) Kwak, J. H.; Schmitz, F. J.; Kelly, M. J. *Nat. Prod.* **2000**, 63, 1153–1156.
- (6) Goclik, E.; Koenig, G. M.; Wright, A. D.; Kaminsky, R. *J. Nat. Prod.* **2000**, 63, 1150–1152.
- (7) Andersen, R.; Nodwell, M.; Mui, A. (University of British Columbia, Can.) PCT Int. Appl. 2007147251, 2007.
- (8) Andersen, R.; Nodwell, M.; Mui, A. (University of British Columbia, Can.) PCT Int. Appl. 2007147252, 2007.
- (9) Dale, J. A.; Dull, D. L.; Mosher, H. S. *J. Org. Chem.* **1969**, 34, 2543–2549.
- (10) Dale, J. A.; Mosher, H. S. *J. Am. Chem. Soc.* **1973**, 95, 512–519.
- (11) Ohtani, I.; Kusumi, T.; Kashman, Y.; Kakisawa, H. *J. Am. Chem. Soc.* **1991**, 113, 4092–4096.
- (12) Ono, M.; Sawamura, H.; Ito, Y.; Mizuki, K.; Nohara, T. *Phytochemistry* **2000**, 55, 873–877.
- (13) Oku, N.; Matsunaga, S.; Wada, S.; Watabe, S.; Fusetani, N. *J. Nat. Prod.* **2000**, 63, 205–209.
- (14) Orihara, Y.; Yang, J. W.; Komiya, N.; Koge, K.; Yoshikawa, T. *Phytochemistry* **2002**, 59, 385–389.
- (15) Sunazuka, T.; Handa, M.; Nagai, K.; Shirahata, T.; Harigaya, Y.; Otoguro, K.; Kuwajima, I.; Omura, S. *Tetrahedron* **2004**, 60, 7845–7859.
- (16) Gottlieb, H. E.; Kotlyar, V.; Nudelman, A. *J. Org. Chem.* **1997**, 62, 7512–7515.
- (17) Smith, P. M.; Thomas, E. J. *J. Chem. Soc., Perkin Trans. 1* **1998**, 3541–3556.
- (18) Breitenbach, J.; Kränzlin, F. *Pilze der Schweiz*; Mykologia: Luzern, 1984; Vol. 1, Ascomyceten.

NP800355A