

Letters to the Editor

Trifluoromethylation of the amide group*

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Amines containing an α -trifluoromethyl substituent constitute an important class of biologically active compounds.^{1–4} In particular, such a fragment is similar in biological properties to an amide fragment.



Replacement of the oxo fragment in a peptide drug by a trifluoromethyl group and a hydrogen atom can substantially enhance its therapeutic effect.^{1,2}

Known general routes to CF_3 -substituted amines involve nucleophilic addition to imines^{5–7} and fluorination of amino acids.^{8,9} Below we describe a first example of the straightforward transformation of amides into CF_3 -containing amines.

Earlier,^{5,10–13} in a search for efficient ways of trifluoromethylation of the $\text{C}=\text{N}$ bond, we have shown that

N,N -dialkyliminium cations react with Me_3SiCF_3 in the presence of a fluoride or acetate anion.¹³ Here we propose to use a cation generated in the *O*-alkylation of a carbamoyl group as an electrophile.^{14,15}

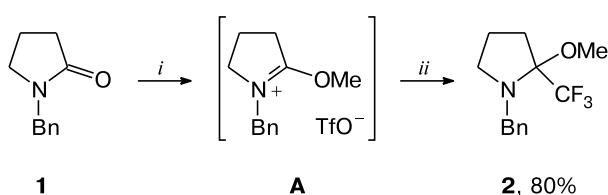
Methylation of *N*-benzylpyrrolidone (**1**) with methyl triflate in hot 1,2-dichloroethane produces cation **A**, which can be transformed into α -methoxy- α -trifluoromethyl amine **2** by a reaction with Me_3SiCF_3 and KF in DMF (Scheme 1).** Compound **2** should be stored at -25°C because of its low stability at room temperature.

Reduction of α -methoxy amine **2** with $\text{NaBH}_4-\text{BF}_3 \cdot \text{OEt}_2$ gives 2-trifluoromethylpyrrolidine **3** in 87% yield (Scheme 2). The methoxy group in amine **2** can also be replaced by an aliphatic residue or a cyano group in reactions with various C-nucleophiles or trimethylsilyl cyanide to give the corresponding products **4a–d** in good yields. It should be emphasized that compound **4c** contains adjacent quaternary C atoms,

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** With dimethyl sulfate used to generate iminium salt **A**, the reaction gives α -methoxy amine **2** in 58% yield and a by-product (supposedly, 1-benzyl-5-trifluoromethyl-2,3-dihydro-1*H*-pyrrole) in 15% yield.

Scheme 1



Me_3SiCF_3 ($332 \mu\text{L}$, 2.25 mmol) and KF (109 mg , 1.88 mmol) were added. The resulting suspension was stirred at 50°C for 1 h, cooled to room temperature, quenched with saturated aqueous Na_2CO_3 (0.5 mL), and stirred for 2 min. The mixture was diluted with water (6 mL) and extracted with hexane–ether ($1 : 1$, $3 \times 5 \text{ mL}$). The combined extracts were filtered through Na_2SO_4 and concentrated *in vacuo*. The residue was chromatographed on silica gel with a hexane–EtOAc system ($15 : 1$) containing 0.5% NEt_3 as an eluent. The yield of compound **2** was 311 mg (80%), a light yellow oil, $R_f 0.20$ (hexane–EtOAc, $6 : 1$). ^1H NMR (300 MHz), δ : 1.77 – 1.97 (m, 2 H); 2.18 – 2.32 (m, 2 H); 2.88 – 3.06 (m, 2 H) (CH_2) $_3$; 3.42 (s, 3 H , OMe); 4.02 (d, 1 H , NCH_AH_B , $J = 14.5 \text{ Hz}$); 4.11 (d, 1 H , PhCH_AH_B , $J = 14.5 \text{ Hz}$); 7.27 – 7.45 (m, 5 H , Ph). ^{13}C NMR (75 MHz), δ : 21.5 (NCH_2CH_2); 29.6 (CH_2); 49.7 (CH_2); 50.3 (q, CH_2 , $J = 1.8 \text{ Hz}$); 51.0 (CH_2); 95.2 (q, CCF_3 , $J = 28.9 \text{ Hz}$); 125.1 (q, CF_3 , $J = 289.9 \text{ Hz}$); 126.8 (CH); 127.8 (CH); 128.3 (CH); 139.5 (C_i). ^{19}F NMR (282 MHz), δ : -79.3 (s, 3 F).

1-Benzyl-2-trifluoromethylpyrrolidine (3). Sodium borohydride (29 mg , 0.75 mmol) and $\text{BF}_3 \cdot \text{OEt}_2$ ($71 \mu\text{L}$, 0.56 mmol) were successively added to a solution of α -methoxy amine **2** (130 mg , 0.5 mmol) in dioxane (1 mL). The resulting suspension was stirred at 50°C for 2 h, cooled to room temperature, quenched with saturated aqueous Na_2CO_3 (0.5 mL), and stirred for an additional 2 min. Then the mixture was diluted with water (6 mL) and extracted with hexane–ether ($1 : 1$, $3 \times 5 \text{ mL}$). The combined extracts were filtered through Na_2SO_4 and concentrated *in vacuo*. The residue was chromatographed on silica gel with hexane–EtOAc ($20 : 1$) as an eluent, $R_f 0.36$. The yield of compound **3** was 100 mg (87%), a colorless oil. Found (%): C, 62.74 ; H, 6.09 ; N, 6.05 . $\text{C}_{12}\text{H}_{14}\text{F}_3\text{N}$ (229.24). Calculated (%): C, 62.87 ; H, 6.16 ; N, 6.11 . ^1H NMR (300 MHz), δ : 1.70 – 2.08 (m, 4 H , (CH_2) $_2$); 2.41 (td, 1 H , NCH_AH_B , $J = 9.5 \text{ Hz}$, $J = 6.6 \text{ Hz}$); 3.00 (t, 1 H , NCH_AH_B , $J = 7.5 \text{ Hz}$); 3.30 (qt, 1 H , CHCF_3 , $J = 7.3 \text{ Hz}$); 3.63 (d, 1 H , PhCH_AH_B , $J = 13.3 \text{ Hz}$); 4.22 (d, 1 H , PhCH_AH_B , $J = 13.3 \text{ Hz}$); 7.23 – 7.42 (m, 5 H , Ph). ^{13}C NMR (75 MHz), δ : 24.3 (NCH_2CH_2); 26.4 (q, $\text{F}_3\text{CCH}_2\text{CH}_2$, $J = 1.8 \text{ Hz}$); 53.8 (NCH_2CH_2); 59.8 (q, CH_2Ph , $J = 1.1 \text{ Hz}$); 63.4 (q, CHCF_3 , $J = 28.3 \text{ Hz}$); 127.0 (CH); 127.1 (q, CF_3 , $J = 280.3 \text{ Hz}$); 128.3 (CH); 128.6 (CH); 138.9 (C_i). ^{19}F NMR (282 MHz), δ : -76.8 (d, 3 F , $J = 7.3 \text{ Hz}$).

Reactions of α -methoxy amine 2 with C-nucleophiles (general procedure). Boron trifluoride etherate ($95 \mu\text{L}$, 0.75 mmol) was added dropwise at -25°C to a solution of α -methoxy amine **2** (130 mg , 0.5 mmol) and an appropriate nucleophilic reagent (0.75 mmol) in dichloromethane (1 mL). The reaction mixture was stirred at -25°C for 10 min and then at room temperature for 1 h. Saturated aqueous Na_2CO_3 (0.5 mL) was added. The mixture was stirred for an additional 2 min, diluted with water (6 mL), and washed with hexane–ether ($1 : 1$, $3 \times 5 \text{ mL}$). The combined extracts were filtered through Na_2SO_4 and concentrated *in vacuo*. The residue was chromatographed on silica gel.

2-Allyl-1-benzyl-2-trifluoromethylpyrrolidine (4a), an oil, $R_f 0.15$ (hexane). Found (%): C, 66.87 ; H, 6.80 ; N, 5.15 . $\text{C}_{15}\text{H}_{18}\text{F}_3\text{N}$ (269.31). Calculated (%): C, 66.90 ; H, 6.74 ; N, 5.20 . ^1H NMR (300 MHz), δ : 1.61 – 1.88 (m, 2 H), 1.94 – 2.18 (m, 2 H) ((CH_2) $_2$); 2.45 (dd, 1 H , $=\text{CCH}_A\text{H}_B$, $J = 14.5 \text{ Hz}$, $J = 8.1 \text{ Hz}$); 2.66 (dd, 1 H , $=\text{CCH}_A\text{H}_B$, $J = 14.5 \text{ Hz}$, $J = 6.1 \text{ Hz}$); 2.70 – 2.82 (m, 2 H , NCH_2); 3.87 (d, 1 H , PhCH_AH_B , $J = 13.8 \text{ Hz}$); 4.00 (d, 1 H , PhCH_AH_B , $J = 13.8 \text{ Hz}$); 5.13 – 5.32 (m, 2 H , $\text{CH}_2=$); 5.85 – 6.02 (m, 1 H , $-\text{CH}=$); 7.20 – 7.41 (m, 5 H , Ph).

which is evidence for the efficiency of this method of C–C bond formation.

^1H , ^{13}C , and ^{19}F NMR spectra were recorded on Bruker AM-300 and Bruker AC-200 instruments in CDCl_3 . Dichloroethane and dichloromethane were distilled over CaH_2 immediately before use; DMF was distilled *in vacuo* over P_2O_5 and kept over molecular sieves (4 \AA). *N*-Benzylpyrrolidone was prepared according to a known procedure.¹⁶ Boron trifluoride etherate was purified by distillation *in vacuo*. The other commercial reagents were used as purchased.

1-Benzyl-2-methoxy-2-trifluoromethylpyrrolidine (2). Methyl triflate ($176 \mu\text{L}$, 1.8 mmol) was added to a solution of *N*-benzylpyrrolidone (263 mg , 1.5 mmol) in 1,2-dichloroethane (1.5 mL). The mixture was stirred at 80°C for 15 min and concentrated *in vacuo*. The residue was dissolved in DMF (2.5 mL) and then

¹³C NMR (75 MHz), δ : 22.1 (NCH_2CH_2); 31.2 (CH_2); 35.7 (CH_2); 51.6 (CH_2); 52.2 (CH_2); 67.2 (q, CCF_3 , $J = 23.8$ Hz); 119.0 ($\text{H}_2\text{C}=\text{}$); 126.8 (CH); 128.0 (CH); 128.2 (CH); 128.6 (q, CF_3 , $J = 291.4$ Hz); 132.8 ($\text{CH}=\text{CH}_2$); 140.2 (C_i). ¹⁹F NMR (282 MHz), δ : -75.0 (s, 3 F).

1-Benzyl-2-phenacyl-2-trifluoromethylpyrrolidine (4b), an oil, R_f 0.31 (hexane-EtOAc, 25 : 1). Found (%): C, 69.21; H, 5.76; N, 3.99. $\text{C}_{20}\text{H}_{20}\text{F}_3\text{NO}$ (347.37). Calculated (%): C, 69.15; H, 5.80; N, 4.03. ¹H NMR (300 MHz), δ : 1.78–1.93 (m, 2 H); 2.23 (ddd, 1 H, $J = 13.2$ Hz, $J = 7.0$ Hz, $J = 4.9$ Hz); 2.49 (ddt, 1 H, $J = 18.3$ Hz, $J = 9.1$ Hz, $J = 0.8$ Hz); 2.80–2.90 (m, 1 H); 2.95 (q, 1 H, $J = 7.7$ Hz) ((CH_2)₃); 3.42–3.57 (m, 2 H, CH_2CO); 3.75 (d, 1 H, PhCH_AH_B , $J = 13.6$ Hz); 3.99 (d, 1 H, PhCH_AH_B , $J = 13.6$ Hz); 7.17–7.32 (m, 5 H, $\underline{\text{PhCH}}_2$); 7.46–7.55 (m, 2 H, PhCO); 7.61 (tt, 1 H, PhCO, $J = 7.3$ Hz, $J = 2.4$ Hz); 7.98–8.04 (m, 2 H, PhCO). ¹³C NMR (75 MHz), δ : 22.7 (NCH_2CH_2); 32.1 (CH_2); 37.4 (CH_2); 52.1 (CH_2); 53.0 (CH_2); 67.4 (q, CCF_3 , $J = 26.0$ Hz); 126.7 (CH); 127.9 (CH); 128.0 (CH); 128.0 (q, CF_3 , $J = 288.6$ Hz); 128.2 (CH); 128.7 (CH); 133.3 (CH); 137.7 (C_i); 140.0 (C_i); 196.9 (C=O). ¹⁹F NMR (282 MHz), δ : -78.2 (s, 3 F).

Methyl 2-(1-benzyl-2-trifluoromethylpyrrolidin-2-yl)-2-methylpropionate (4c), R_f 0.10 (hexane-EtOAc, 30 : 1), m.p. 38–40 °C. Found (%): C, 62.15; H, 6.65; N, 4.07. $\text{C}_{17}\text{H}_{22}\text{F}_3\text{NO}_2$ (329.36). Calculated (%): C, 61.99; H, 6.73; N, 4.25. ¹H NMR (300 MHz), δ : 1.43 (s, 3 H, CMe_AMe_B); 1.46 (s, 3 H, CMe_AMe_B); 1.59–1.81 (m, 2 H); 1.98–2.14 (m, 1 H); 2.45–2.59 (m, 1 H); 2.72 (dd, 1 H, $J = 13.9$ Hz, $J = 7.7$ Hz); 2.95–3.06 (m, 1 H) ((CH_2)₃); 3.64–3.75 (m, 4 H, PhCH_AH_B + CO_2Me); 4.51 (d, 1 H, PhCH_AH_B , $J = 14.7$ Hz); 7.21–7.45 (m, 5 H, Ph). ¹³C NMR (75 MHz), δ : 22.7 (q, CMe_AMe_B , $J = 2.4$ Hz); 23.3 (NCH_2CH_2); 23.5 (q, CMe_AMe_B , $J = 1.5$ Hz); 32.5 (q, CH_2 , $J = 2.2$ Hz); 49.3 (CH_2); 52.0 (CH_2); 53.9 (CH_2); 55.0 (q, CH_2 , $J = 2.4$ Hz); 72.4 (q, CCF_3 , $J = 22.3$ Hz); 126.8 (CH); 127.5 (CH); 128.4 (CH); 129.0 (q, CF_3 , $J = 294.5$ Hz); 140.4 (q, C_i , $J = 1.4$ Hz); 175.9 (C=O). ¹⁹F NMR (282 MHz), δ : -65.3 (s, 3 F).

1-Benzyl-2-trifluoromethylpyrrolidine-2-carbonitrile (4d), an oil, R_f 0.25 (hexane-EtOAc, 15 : 1). Found (%): C, 61.27; H, 5.14; N, 10.87. $\text{C}_{13}\text{H}_{13}\text{F}_3\text{N}_2$ (254.25). Calculated (%): C, 61.41; H, 5.15; N, 11.02. ¹H NMR (300 MHz), δ : 1.83–2.03 (m, 2 H); 2.43–2.65 (m, 3 H); 3.03–3.16 (m, 1 H) ((CH_2)₃); 3.68 (d, 1 H, PhCH_AH_B , $J = 13.2$ Hz); 4.35 (d, 1 H, PhCH_AH_B , $J = 13.2$ Hz); 7.28–7.44 (m, 5 H, Ph). ¹³C NMR (75 MHz), δ : 23.4 (NCH_2CH_2); 34.8 (q, CH_2CCF_3 , $J = 1.0$ Hz); 52.7 (CH_2); 56.2 (q, CH_2); 66.3 (q, CCF_3 , $J = 31.3$ Hz); 114.8 (CN); 123.6 (q, $J = 283.1$ Hz); 127.5 (CH); 128.2 (CH); 128.5 (CH); 137.6 (C_i). ¹⁹F NMR (282 MHz), δ : -77.2 (s, 3 F).

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