Optically Active 6-Membered P-Heterocycles: 1-Phenyl-1,2-Dihydrophosphinine Oxide and 1-Phenyl-3-Diphenylphosphinoyl-1,2,3,6-Tetrahydrophosphinine Oxide

Viktória Ujj^{a,b}, Andrea Kerényi^a, Andrea Laki^a, Elemér Fogassy^a and György Keglevich^{*,a}

^aDepartment of Organic Chemistry and Technology, Budapest University of Technology and Economics, H-1521 Budapest, Hungary

^bResearch Group of the Hungarian Academy of Sciences at the Department of Organic Chemistry and Technology, Budapest University of Technology and Economics, H-1521 Budapest, Hungary

Received November 26, 2009: Revised January 07, 2010: Accepted January 12, 2010

Abstract: The antipodes of racemic 1-phenyl-1,2-dihydrophosphinine oxides **1A** and **1B** were separated by resolution *via* formation of a diastereomeric complex using (–)-(2R,3R)- α , α , α' , α' -tetraphenyl-1,4-dioxaspiro[4.5]decan-2,3-dimethanol (spiro-TADDOL) **2**. The (+)-antipode of 1-phenyl-3-diphenylphosphinoyl-1,2,3,6-tetrahydrophosphinine oxide **3** was synthesized by the diastereoselective addition of diphenylphosphine oxide to the α , β -double-bond of (–)-1,2-dihydrophosphinine oxide **1A**. The new chiral tetrahydrophosphinine oxide (+)-**3** served, after double deoxygenation, as a novel P-chiral bidentate ligand in the corresponding platinum complex.

Keywords: Resolution, dihydrophosphinine oxide, tetrahydrophosphinine oxide, Michael addition, platinum complex.

1. INTRODUCTION

Among the six-membered P-heterocycles, dihydro- and tetrahydrophosphinine oxides form a representative group [1,2]. Perhaps the most convenient method for the preparation of dihydrophosphinine oxides involves ring enlargement of easily available 2,5-dihydro-1H-phosphole oxides by the dichlorocarbene addition method [3,4]. Thermolysis of the 3-phosphabicyclo[3.1.0]hexane 3-oxides obtained by dichlorocyclopropanation furnished dihydrophosphinine oxides [5,6]. 1,2,3,6-Tetrahydrophosphinine-oxides with a P-function in position 3 have been introduced recently [7,8]. The related diphosphine behaved as a suitable P-ligand to form the corresponding *cis* chelate platinum complex [9].

We aimed at the synthesis of optically active 1-phenyl-3diphenylphosphinoyl-1,2,3,6-tetrahydrophosphinine oxide from one of the enantiomers of the 1-phenyl-1,2dihydrophosphinine oxide to make eventually available the corresponding bidentate P-ligand in an optically active form. Optically active P-ligands may be components of transition metal complexes used as enantioselective catalysts in homogeneous catalytic reactions [10]. Especially the homoand heterobidentate chiral P-ligands, such as BINAP, DIOP, DIPAMP, BINAPHOS are of practical importance [11].

2. RESULTS AND DISCUSSION

Recently, we have elaborated a simple method [12-15] for the resolution of a variety of 3-methyl-3-phospholene 1-

oxides *via* molecular complex formation with the chiral host, (-)-(4R,5R)-4,5-bis(diphenylhydroxymethyl)-2,2-dimethyl-dioxolane (TADDOL) or (-)-(2R,3R)- α , α , α ', α '-tetraphenyl-1,4-dioxaspiro[4.5]decan-2,3-dimethanol (spiro-TADDOL) **2** [16,17].

We could extend the method developed by us to the resolution of methyl-4-chloro-1-phenyl-1,2-dihydrophosphinine 1-oxide **1**. The resolution can be done *via* molecular complex formation using spiro-TADDOL (–)-**2**. In the course of the preparation of 1,2-dihydrophosphinine oxide **1**, a 3:1 mixture of two double-bond isomers (**A** and **B**) were obtained [5] that could not be separated by column chromatography. Therefore, the mixture of dihydrophosphinine oxides **1A** and **1B** was resolved using spiro-TADDOL (–)-**2** in a mixture of acetone and hexane.

Racemic 1-phenyl-1,2-dihydrophosphinine oxide 1 was resolved by adding first half equivalent of spiro-TADDOL (-)-2 to the hot acetone solution of 1. The 1:1 crystalline complex $(-)-1 \cdot (-)-2$ was precipitated on the addition of hexane. The ratio of (-)-1 and resolving agent (-)-2 was determined by ¹H NMR spectoscopy. Complex $(-)-1 \cdot (-)-2$ was analyzed on a chiral HPLC column showing that the enantiomeric excess (ee) of major isomer (-)-1A was 40%. After two recrystallizations of the $(-)-1 \cdot (-)-2$ complex from a 1:2 mixture of acetone and hexane, complex $(-)-1A \cdot (-)-2$ was obtained with a 89% ee in 33 % yield. The complex could be further purified by a third recrystallization giving the complex with 96.5% ee, but the yield became rather low. After flash column chromatography on a silica gel, (-)-1A was recovered with an ee of 96.5%. Double-bond isomer 1B did not separate from major isomer **1A** during the process. The enantiomeric excess of minor isomer 1B was >99.9% after the three recrystallizations and the ratio of the doublebond isomers in the crystalline phase did not change

^{*}Address correspondence to this author at the Department of Organic Chemistry and Technology, Budapest University of Technology and Economics, H-1521 Budapest, Hungary; Tel: +36 1 4631111/5883; Fax: +36 1 4633648; E-mail: keglevich@mail.bme.hu



Scheme 1.

significantly (Scheme 1). Dihydrophosphinine oxide (+)-(1A and 1B) remained in solution. This is the first case that a 1,2-dihydrophosphinine oxide was obtained in an optically active form.

The resolution of 1-phenyl-1,2-dihydrophosphinine oxide 1 with half equivalent of (-)-2 was also accomplished from a 2:5 mixture of ethyl acetate and hexane, but in this case, the resolution sequence was less efficient than using a mixture of acetone and hexane. Crystallization of the corresponding supramolecular formation involving 1,2-dihydrophosphinine oxide 1 and TADDOL could not be accomplished.

We wished to synthesize optically active 1-phenyl-3diphenylphosphinoyl-1,2,3,6-tetrahydrophosphinine oxide **3** by the *Michael* addition of diphenylphosphine oxide on the electron-poor α,β -double-bond of the (–)-1-phenyl-1,2dihydrophosphinine oxide **1A** to obtain the precursor of an optically active bidentate P-ligand. The *ee* value of the starting (–)-1-phenyl-1,2-dihydrophosphinine oxide **1A** was 89%. To activate the diphenylphosphine oxide, it was first reacted with trimethylaluminium at 0°C in chloroform. Then, the corresponding ([Ph₂P(O)])⁻ anion so formed reacted easily with the α,β -double-bond of the (–)-dihydrophosphinine oxide **1A** added to the reaction mixture at 0°C furnishing the (+)-1-phenyl-3-diphenylphosphinoyl-1,2,3,6tetrahydrophosphinine oxide **3**. The unreactivity of isomer **1B** in the *Michael* addition is a consequence of steric hindrance due to the methyl group in position 5. Due to a selectivity observed earlier [7,8], only one of the two possible diastereomers was obtained. Column chromatography afforded tetrahydrophosphinine oxide (+)-**3** in 50% yield. The *ee* value of product (+)-**3** was determined by chiral HPLC. The structure of (+)-**3** was confirmed by ${}^{31}P$ and ${}^{1}H$ NMR spectral data (Scheme **2**).

A *twist-boat* conformer was established earlier for 1phenyl-3-diphenylphosphinoyl-1,2,3,6-tetrahydrophosphinine oxide by DFT calculations [8]. Single crystal X-ray analysis suggested a *twist-boat/cis* conformer containing the Ph₂P(O) moiety and the oxygen atom of the ring P=O moiety in the *cis* disposition [18].

The new chiral tetrahydrophosphinine oxide (+)-**3** may serve, after deoxygenation, as a novel P-chiral bidentate ligand in transition metal complexes. The bis(phosphine oxide) (+)-**3** was deoxygenated applying phenylsilane at 80 °C to give diphosphine **4**. The silane reductions are known to proceed with retention of configuration [19]. Diphosphine **4** was then reacted with one equivalent of dichlorodiben-zonitrile platinum(II) in benzene to afford *cis* chelate complex (+)-**5** (Scheme **3**). The structure of (+)-**5** was confirmed by ³¹P NMR spectral data [9].



Scheme 2.

Scheme 3.

3. CONCLUSION

In summary, the (–)-antipode of 1-phenyl-1,2dihydrophosphinine oxide **1** was made available by a novel resolution method applying a TADDOL derivative (–)-**2**. Diastereoselective hydrophosphinoxidation led to the corresponding (+)-Ph₂P(O)-1,2,3,6-tetrahydrophosphinine oxide **3** that, after double deoxygenation, was a suitable optically active bidentate P-ligand in a *cis* chelate platinum complex (+)-**5**.

4. EXPERIMENTAL

4.1. General

The ¹H and ³¹P NMR spectra were recorded on a Bruker DRX-500 spectrometer operating at 500 and 202.4 MHz, respectively. Chemical shifts are downfield relative to TMS and 85% H₃PO₄. The coupling constants are given in Hertz. Optical rotations were determined on a Perkin-Elmer 241 polarimeter.

The enantiomeric excesses were determined by chiral HPLC. The *ee* of **1** was determined by chiral HPLC (Daicel Chem. Ind., having a Chiralpack AD-H column 250*4.6 mm ID, using hexane/isopropanol 85/15 as the eluent with a flow rate of 0.8 ml/min, T=20 °C, UV detector λ =254 nm). The *ee* of **3** was determined by chiral HPLC (Daicel Chem. Ind., having a Chiralcel OD column 250*4.6 mm ID, using hexane/isopropanol 75/25 as the eluent with a flow rate of 0.8 ml/min, T=20 °C, UV detector λ =254 nm).

The starting 1-phenyl-1,2-dihydrophosphinine oxide [5] **1** and (-)-(2R,3R)- $\alpha,\alpha,\alpha',\alpha'$ -tetraphenyl-1,4-dioxaspiro [4.5] decan-2,3-dimethanol [16] (-)-**2** were synthesized as described earlier. Trimethylaluminium, 2.0 M sol. in heptane and *cis*-PtCl₂(PhCN)₂ were purchased from Aldrich Chemical Co.

4.2. Resolution of methyl-4-chloro-1-phenyl-1,2-dihydrophosphinine 1-oxide 1 with spiro-TADDOL (-)-2

To 0.43 g (1.80 mmol) of racemic 3- and 5-methyl-4chloro-1-phenyl-1,2-dihydrophosphinine 1-oxide 1A (75%) and 1B (25%) and 0.46 g (0.90 mmol) of spiro-TADDOL (-)-2 in 2.1 mL of hot acetone was added 4.3 mL of hexane. After the addition, crystals of the complex started to appear immediately. After 1 h, the crystals were separated by filtration to give 0.54 g (80%) of complex consisting of $(-)-1A \cdot (-)-2$ (75%) with an *ee* of 40% and 1B \cdot 2 (25%) with an ee of 55%. The complex was further purified by two recrystallizations from acetone-hexane (2.1 mL/4.3 mL) to afford 0.22 g (33%) of complex (-)- $1A^{\bullet}(-)-2$ with an *ee* of 89% and **1B**•2 with an *ee* of 95%, respectively, $[\alpha]_D^{25} = -$ 90.2 (c 1, CHCl₃). Column chromatography (silica gel, 3% methanol in chloroform) of the complex regenerated 0.07 g (0.3 mmol, 31%) of (-)-3- and 5-methyl-4-chloro-1-phenyl-1,2-dihydrophosphinine 1-oxides 1A and 1B, 1A (δ_P 16.1 $[\delta_{P}^{\text{lit}} 15.3] [5]$ with an ee of 89%, **1B** ($\delta_{P} 15.0 [\delta_{P}^{\text{lit}} 14.2] [5]$) with an ee of 95%), $[\alpha]_D^{25} = -169.1$ (*c* 0.5, CHCl₃). Based on ³¹P NMR, the ratio of **1A** and **1B** remained ca. 3:1. Retention times: 13.5 and 15.9 min for 1A, 15.0 and 18.1 min for 1B.

4.3. Preparation of (+)-4-chloro-5-methyl-1-phenyl-3diphenylphosphinoyl-1,2,3,6-tetrahydrophosphinine 1oxide 3

Tetrahydrophosphinine 1-oxide (+)-**3** was prepared from (-)-**1** as described earlier for the racemic case [7]. *Ee* of (+)-**3**: 89%; yield: 50%. ³¹P NMR (CDCl₃), δ 33.0 (d, ³*J*_{P-P} = 13.8, P1), 34.0 (d, ³*J*_{P-P} = 13.8, P2), [δ_P^{lit} 34.0 (d, ³*J*_{P-P} = 13.8, P1), 34.8 (d, ³*J*_{P-P} = 13.8, P2)] [7], [α]_D²⁵ = +257.6 (*c* 0.5, CHCl₃). Retention times: 26.2 min for (-)-**3** and 36.5 min for (+)-**3**.

4.4. Deoxygenation of the bis(phosphine oxide) precursor (+)-3 and complexation of the bisphosphine 4 by PtCl₂(PhCN)₂

23.0 mg (0.05 mmol, ee 89%) of 1-Phenyl-3diphenylphosphinoyl-1,2,3,6-tetrahydrophosphinine-1-oxide (+)-**3** and 32 µL (0.26 mmol) of phenylsilane was kept at 80 °C under nitrogen for 3 days. Then, the mixture was taken up in 4 mL of degassed benzene, the bisphosphine **4** so obtained was immediately reacted with 25.0 mg (0.05 mmol) of dichlorodibenzonitrile platinum(II) and the mixture was stirred at room temperature for 24 h under nitrogen. After crystallization from the benzene solution, 32.0 mg (+)-**5** (93%) was obtained, ³¹P NMR (CDCl₃), $\delta_{\rm P}$ 14.4 ($J_{\rm Pt-P}$ = 3496, $J_{\rm P-P}$ = 7.4, P1), 47.6 ($J_{\rm Pt-P}$ = 3678, $J_{\rm P-P}$ = 7.4, C3-P), [$\delta_{\rm P}^{\rm lit}$ 17.7 ($J_{\rm Pt-P}$ = 3496, $J_{\rm P-P}$ = 6.7, P1), 50.3 ($J_{\rm Pt-P}$ = 3681, $J_{\rm P-P}$ = 6.7, C3-P] [9], [α]_D²⁵ = +47.7 (*c* 0.5, CHCl₃).

ACKNOWLEDGEMENTS

Authors are grateful for the financial support from the Hungarian Scientific Research Fund (OTKA, Grants No. T075236, T067679).

REFERENCES

- Hewitt, D.G. Comprehensive Heterocyclic Chemistry II; Katritzky, A.R.; Rees, C.W.; Scriven, E.F.V. Eds.; Pergamon: New York, 1996; Vol. 5, pp. 639-668.
- [2] Keglevich, G. Phosphorus Heterocycles I.; Bansal, R.K. Ed.; Springer: Berlin 2009; Vol. 20, pp. 65-99.
- [3] Keglevich, G. 3-Phosphabicyclo[3.1.0]hexane 3-oxides: Useful intermediates for dihydro-, tetrahydro- and hexahydrophosphinines and phosphinines. *Rev. Heteroatom Chem.*, **1996**, *14*, 119-136.
- [4] Keglevich, G. Synthesis of 6-membered and 7-membered Pheterocycles by ring enlargement. Synthesis, 1993, 931-942.
- [5] Keglevich, G.; Androsits, B.; Tőke, L. Synthesis of dihydrophosphorins by the thermal transformation of phosphole dichlorocarbene adducts. J. Org. Chem., 1988, 53, 4106-4108.
- [6] Keglevich, G.; Brlik, J.; Janke, F.; Tőke, L. Convenient synthesis of 1-alkoxy-1,2-dihydrophosphinine 1-oxides by ring enlargement. *Heteroatom Chem.*, **1990**, *1*, 419-424.
- [7] Keglevich, G.; Sipos, M.; Imre, T.; Ludányi, K.; Szieberth, D.; Tőke, L. Diastereoselective synthesis of 1,2,3,6-tetrahydrophosphinine 1-oxides with an exocyclic P-function by a Michael type addition. *Tetrahedron Lett.*, **2002**, *43*, 8515-8518.
- [8] Keglevich, G.; Sipos, M.; Szieberth, D.; Nyulászi, L.; Imre, T.; Ludányi, K.; Tőke, L. Weak intramolecular interactions as controlling factors in the diastereoselective formation of 3phosphinoxido- and 3-phosphono-1,2,3,6-tetrahydrophosphinine 1oxides. *Tetrahedron*, 2004, 60, 6619-6627.
- [9] Keglevich, G.; Sipos, M.; Szieberth, D.; Petőcz, G.; Kollár, L. 4-Chloro-5-methyl-3-diphenylphosphino-1-phenyl-1,2,3,6-tetrahydro-

phosphinine as a bidentate P-ligand in a cis chelate Pt(II) complex. J. Organomet. Chem., **2004**, 689, 3158-3162.

- [10] Brunner, H.; Zettlmeier, W. Handbook of Enantioselective Catalysis with Transition Metal Compounds, VCH: Weinheim, 1993.
- [11] Noyori, R. Asymmetric Catalysis in Organic Synthesis; John Wiley & Sons: New York, 1994.
- [12] Novák, T.; Schindler, J.; Ujj, V.; Czugler, M.; Fogassy, E.; Keglevich, G. Resolution of 3-methyl-3-phospholene 1-oxides by molecular complex formation with TADDOL derivatives. *Tetrahedron: Asymmetry*, **2006**, *17*, 2599-2602.
- [13] Novák, T.; Ujj, V.; Schindler, J.; Czugler, M.; Kubinyi, M.; Mayer, Z.A.; Fogassy, E.; Keglevich, G. Resolution of 1-substituted-3methyl-3-phospholene 1-oxides by molecular complex formation with TADDOL derivatives. *Tetrahedron: Asymmetry*, 2007, 18, 2965-2972.
- [14] Novák, T.; Schindler, J.; Ujj, V.; Czugler, M.; Fogassy, E.; Keglevich, G. Chiral P-heterocycles: Efficient method for the resolution of 3-methyl-3-phospholene 1-oxides. *Phosphorus, Sulfur, Silicon*, 2008, 183, 543-546.

- [15] Fogassy, E.; Keglevich, G.; Novák, T.; Schindler, J.; Ujj, V. In Hung. Pat. HU0700278, 2007.
- [16] Beck, A.K.; Bastani, B.; Plattner, D.A.; Petter, W.; Seebach, D.; Braunschweiger, H.; Gysi, P.; Lavecchia, L. Large-scale preparation of alpha,alpha,alpha,alpha-tetraaryl-1,3-dioxolane-4,5dimethanols (TADDOLE) - versatile auxiliaries for epc synthesis and its solid-state structure. *Chimia*, **1991**, *45*, 238-244.
- [17] Seebach, D.; Beck, A.; Heckel, A. TADDOLs, their derivatives, and TADDOL analogues: versatile chiral auxiliaries. *Angew. Chem. Int. Ed. Engl.*, **2001**, *40*, 92-138.
- [18] Czugler, M.; Körtvélyesi, T.; Fábián, L.; Sipos, M.; Keglevich, G. Intra- and intermolecular interactions and water pincer in the crystal structure of a 3-P(O)Ph-2 substituted 1,2,3,6tetrahydrophosphinine oxide hydrate. *Cryst. Eng. Commun.*, 2007, 9, 561-565.
- [19] Quin, L.D. A guide to Organophosphorus Chemistry; John Wiley & Sons: New York, 2000.