

Potassamide Induced *In Situ* Alkylation of 5,6-Dihydroisoquinolines: Structure of Products

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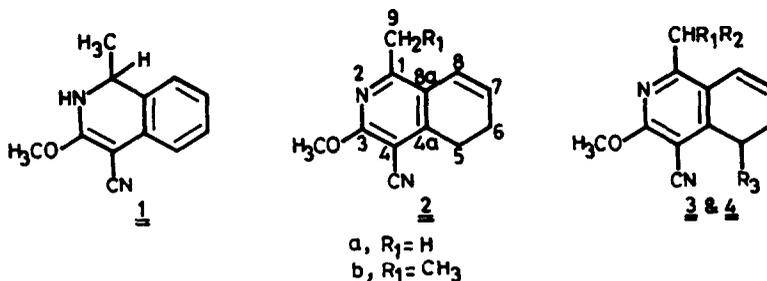
Abstract Potassamide induced *in situ* alkylation of 1-alkyl-4-cyano-3-methoxy-5,6-dihydroisoquinolines (**2a** & **2b**) with alkyl iodides (CH_3I , $\text{CH}_3\text{CH}_2\text{I}$ & cyclohexyl iodide) gave the 5-alkyl- and 5,9-dialkyl-5,6-dihydroisoquinolines (**4a-d** & **3a-e**), isoquinoline derivatives, (**5a-b**) and diastereomeric mixture of 4-alkyl-1,2,3,4-tetrahydroisoquinolin-3(2H)-ones (**6a-e** & **6'a-e**). Structures were assigned on the basis of spectral data [Mass, ^1H & ^{13}C NMR, 2D NOESY & HC-COLOC]. Amide induced *in situ* alkylation of compounds **3a** and **4a** with CH_3I gave in almost quantitative yield the dimethylated compounds **3d** and **3a** respectively. While $\text{KNH}_2/\text{liq NH}_3$ methylation of 1,2-dihydroisoquinoline, **1** with CH_3I gave the mixture of compounds, **6a** & **6'a** and the isoquinoline derivative **5a**, NaH /benzene reaction of **1** with CH_3I gave exclusively **5a**. N-methylation of the mixture of compounds **6a** & **6'a** with $\text{NaH}/\text{CH}_3\text{I}$ gave the methylated derivatives, **7** & **8**. A suitable mechanism has been proposed for the formation of products.

1,2-Dihydroisoquinolines are interesting species due to their chemical reaction¹ and their potential as building blocks in the synthesis of alkaloids² and medicinal agents^{3,4}. We have recently reported⁵ a novel and useful method for the synthesis of stable 1,2-dihydroisoquinoline derivatives. We planned to synthesize N-sugar derivatives⁶ of 1,2-dihydroisoquinoline compound, **1** by alkylation with appropriate halo sugars, as the presence of the ionic species, **9** is already visualised in the formation of 1,2-dihydroisoquinolines⁵. As model studies, the *in situ* alkylation of 1,2-dihydroisoquinoline compounds formed in the $\text{KNH}_2/\text{liq NH}_3$ reaction with alkyl iodides was initially undertaken. The results obtained in this study are discussed further.

Reaction of 4-cyano-3-methoxy-1-methyl-5,6-dihydroisoquinoline (**2a**) with $\text{KNH}_2/\text{liq NH}_3$ in the presence of a trace amount of ferric chloride was carried out as described earlier⁵. The reaction mixture was quenched by addition of CH_3I . Addition of NH_4Cl and workup of the reaction gave mixture of products, which was purified by column chromatography followed by PTLC. This resulted in the isolation of four compounds designated, A-D in the order of increasing polarity (TLC).

The least polar compound, A (30 %) analysing for $\text{C}_{14}\text{H}_{16}\text{N}_2\text{O}$ exhibited an IR absorption at 2220

cm^{-1} The signals at δ 1.18 (d, $J=7.2$ Hz, 3H), 1.26 (t, $J=7.5$ Hz, 3H), 2.24 (dd, $J=17.4$ & 7.2 Hz, 1H), 2.48-2.60 (m, 1H), 2.79 (q, $J=7.5$ Hz, 2H), 3.3 (qn, $J=7.5$ Hz, 1H) were seen in its ^1H NMR spectrum in addition to the methoxy and olefinic signals as in the case of 5,6-dihydroisoquinoline, **2a**. It is evident that the compound has been dimethylated. Irradiation of the proton signals at δ 3.3 (qn) and 2.79 (q) resulted in the collapse of the methyl doublet at δ 1.18 and the methyl triplet at δ 1.26 to two singlets respectively. Hence, methylation has occurred at C-5 & C-9 positions, and the structure **3a** was assigned to compound **A**.

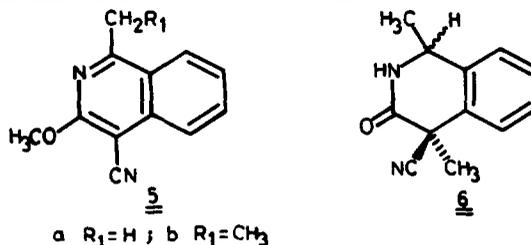


- 3** (a) $R_1 = \text{CH}_3, R_2 = \text{H}, R_3 = \text{CH}_3$, (b) $R_1 = \text{Et}, R_2 = \text{H}, R_3 = \text{Et}$; (c) $R_1 = \text{---} \bigcirc \text{---}, R_2 = \text{H}, R_3 = \text{---} \bigcirc \text{---}$,
 (d) $R_1 = R_2 = R_3 = \text{CH}_3$, (e) $R_1 = \text{Et}, R_2 = \text{CH}_3, R_3 = \text{Et}$
4 (a) $R_1 = \text{H}, R_2 = \text{H}, R_3 = \text{CH}_3$, (b) $R_1 = \text{H}, R_2 = \text{H}, R_3 = \text{Et}$, (c) $R_1 = \text{H}, R_2 = \text{H}, R_3 = \text{---} \bigcirc \text{---}$,
 (d) $R_1 = \text{H}, R_2 = \text{CH}_3, R_3 = \text{Et}$

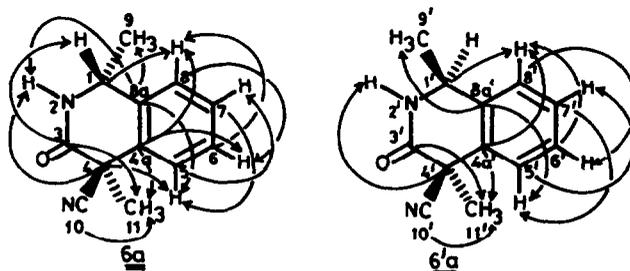
Compound **B** (25 %) analysing for $\text{C}_{13}\text{H}_{14}\text{N}_2\text{O}$ and showing spectral characteristics (IR & ^1H NMR) similar to those of 5,6-dihydroisoquinoline **2a**, also exhibited ^1H NMR signals at δ 1.16 (d, $J=7.2$ Hz, 3H) and 3.29 (qn, $J=7.3$ Hz, 1H). On the basis of double irradiation experiments, (irradiation of the signal at δ 3.29 (qn) resulted in the collapse of methyl doublet at δ 1.16 to a singlet) compound **B** was assigned structure **4a**.

The medium polar Compound, **C** (8 %) was shown to be the already reported⁵ isoquinoline derivative **5a**.

The most polar Compound, **D** (20 %) analysing for $\text{C}_{12}\text{H}_{12}\text{N}_2\text{O}$ exhibited IR absorption frequencies at 3250 (-NH or -OH), 2260 (-CN) & 1680 cm^{-1} (-CONH). ^1H NMR spectrum of this compound showed two sets of signals for each type of proton in the ratio of 3:1 viz, doublets at δ 1.59 ($J=6.7$ Hz) & 1.68 ($J=6.7$ Hz) (3H), singlets at δ 1.99 & 1.9 (3H), a quartet at δ 4.75 ($J=6.7$ Hz) & a doublet of quartet at δ 4.84 ($J=6.6$ Hz & 3 Hz) collapsing to a quartet on D_2O exchange (1H), sets of aromatic protons at δ 7.22-7.63 (4H) and broad singlets at δ 8.36 & 8.43 (1H). It is evident that demethylation as well as alkylation has occurred (^{13}C NMR spectrum also indicated 23 well resolved signals). The mixture of compounds **7** could be the diastereomeric forms of structure **6**.



2D HC-COLOC^{8,9} & NOESY¹⁰ experiments were undertaken in order to assign the correct structure as well as the relative configuration. A COrelation LOngrange Coupling spectrum (¹H-¹³C, COLOC) of compound D revealed long range coupling (³J) of the C-4 quaternary carbon with the N₂-H proton for both the compounds. Of the two downfield carbons(-C=O, δ 167.62 & 167.83), only one (δ, 167.62) showed long range coupling (³J) to the methine C₁-H. Further, the long range coupling (³J) of C-10 carbon(-CN, δ 119.98) with the C₁₁-3H protons for both the compounds is also observed [Tables-1 & 2]. The mixture of compounds could thus be assigned the diastereomeric structures 6a & 6'a

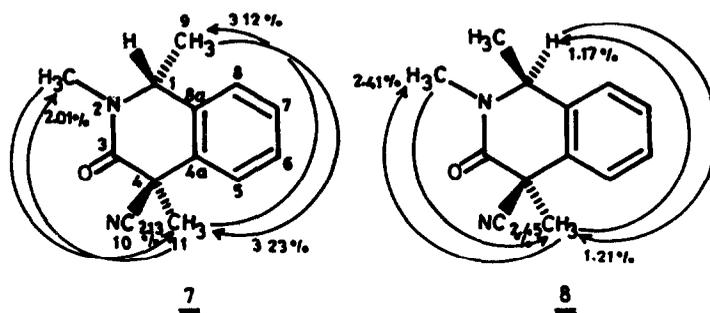


HC-COLOC Correlations of 6a & 6'a

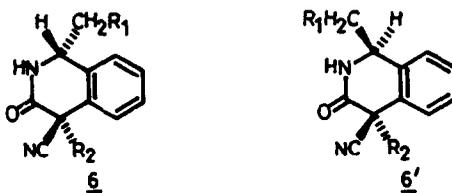
NOESY spectrum revealed correlation between C₉-3H (δ 1.59) & C₁₁-3H (δ 1.99) protons for the major diastereomer. However, there is no NOESY correlation between the same for the minor diastereomer [Tables-1 & 2]. On this basis, the major diastereomer was assigned structure 6a with C₁-Me & C₄-Me in *syn* configuration, while the minor diastereomer structure 6'a with C₁-Me & C₄-Me in *anti* configuration. This was further substantiated by N-methylation studies.

Table 1 High Resolution ¹ H and ¹³ C NMR Assignment of 6a in CDCl ₃ Solution (δ)					Table 2 High Resolution ¹ H and ¹³ C NMR Assignment of 6'a in CDCl ₃ Solution (δ)				
Position	¹³ C	¹ H	HC-COLOC	NOESY	Position	¹³ C	¹ H	HC-COLOC	NOESY
1	51.15	4.84	C1 → H ₉ ,H ₈ ,H ₂	H1 → H ₉ ,H ₈ ,H ₂	1'	49.58	4.75	C1' → H _{9'} ,H _{8'}	H1' → H _{11'} ,H _{9'} ,H _{8'} ,H _{2'}
2	—	8.43	—	H2 → H ₁₁ ,H ₉ ,H ₁	2'	—	8.36	—	H2' → H _{9'} ,H _{1'}
3	167.62	—	C3 → H ₁₁ ,H ₁	—	3'	167.83	—	C3' → H _{11'}	—
4	43.81	—	C4 → H ₁₁ ,H ₅ ,H ₂	—	4'	44.79	—	C4' → H _{11'} ,H _{2'}	—
4a	131.98	—	C4a → H ₁₁ ,H ₈ ,H ₆	—	4a'	132.48	—	C4a' → H _{11'} ,H _{8'}	—
5	126.77	7.57	C5 → H ₇ ,H ₅	H5 → H ₁₁ ,H ₆	5'	126.35	7.63	C5' → H _{7'} ,H _{5'}	H5' → H _{11'} ,H _{6'}
6	128.49	7.43	C6 → H ₈ ,H ₆	H6 → H ₇ ,H ₅	6'	128.58	7.48	C6' → H _{8'} ,H _{6'}	H6' → H _{7'} ,H _{5'}
7	128.79	7.39	C7 → H ₇ ,H ₅	H7 → H ₈ ,H ₆	7'	128.86	7.42	C7' → H _{5'} ,H _{7'}	H7' → H _{8'} ,H _{6'}
8	126.13	7.22	C8 → H ₈ ,H ₆	H8 → H ₉ ,H ₇ ,H ₁	8'	125.22	7.25	C8' → H _{6'} ,H _{8'}	H8' → H _{9'} ,H _{7'} ,H _{1'}
8a	134.18	—	C8a → H ₉ ,H ₅ ,H ₂ ,H ₁	—	8a'	134.38	—	C8a' → H _{9'} ,H _{5'}	—
9	25.38	1.59	C9 → H ₁	H9 → H ₁₁ ,H ₈ ,H ₂ ,H ₁	9'	21.74	1.68	—	H9' → H _{8'} ,H _{2'} ,H _{1'}
10	119.98	—	C10 → H ₁₁	—	10'	119.98	—	C10' → H _{11'}	—
11	29.56	1.99	—	H ₁₁ → H ₉ ,H ₅ ,H ₂	11'	26.65	1.90	—	H _{11'} → H _{5'} ,H _{1'}

Methylation [NaH/CH₃I] of the diastereomeric mixture (6a & 6'a) gave two separable [PTLC, 8 l, hexane EtOAc] N-methylated compounds, 7 (major) & 8 (minor), corresponding to the major and minor isomers of 6. NOE experiments of the methylated compounds further confirmed the assigned structures.



In order to explore the generality of this reaction, we repeated the reaction of alkyl iodides with different 5,6-dihydroisoquinolines, (2a & 2b) which could be prepared by the reaction of the corresponding β -diketones with cyanoacetamide by the general method⁵ Reaction of 1-alkyl-4-cyano-3-methoxy-5,6-dihydroisoquinolines (2a & 2b) in $\text{KNH}_2/\text{liq NH}_3$ followed by quenching with different alkyl iodides (CH_3I , $\text{CH}_3\text{CH}_2\text{I}$ & cyclohexyl iodide) gave the expected sets of compounds, 3a-e, 4a-d, 5a-b, 6b-e & 6'b-e characterised by spectral data

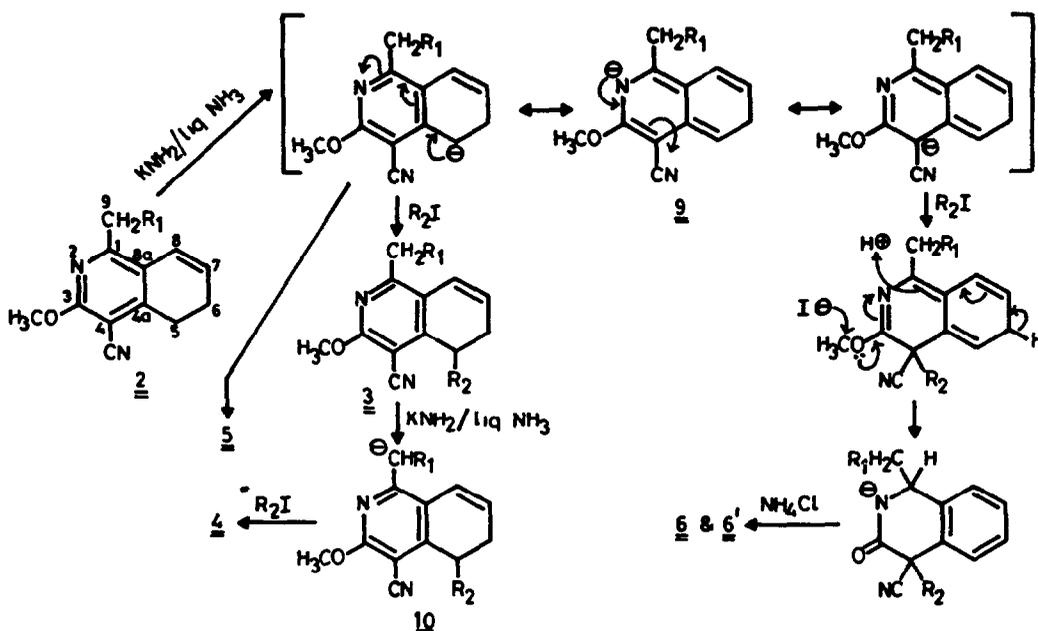


(b) $\text{R}_1 = \text{H}$, $\text{R}_2 = \text{Et}$, (c) $\text{R}_1 = \text{H}$, $\text{R}_2 = -\text{O}$
 (d) $\text{R}_1 = \text{R}_2 = \text{CH}_3$, (e) $\text{R}_1 = \text{CH}_3$, $\text{R}_2 = \text{Et}$

MECHANISM OF FORMATION OF PRODUCTS

The formation of products in the above alkylation can be visualised as in Scheme - 1 The initially formed C-5 benzylic anion can lead to 5-alkyl-5,6-dihydroisoquinoline, 4a-e and isoquinoline, 5a-b derivatives The 5,9-dialkyl derivative, 3a-e is formed from 10 by the generation of C-1Me anion followed by alkylation This is confirmed by the methylation of 3a & 4a leading to the formation of dialkyl compounds 3d & 3a respectively Alkylation at C-4 position of the isomeric C-4 anion followed by demethylation gives diastereomeric compounds, 6a-e & 6'a-e

In order to get the N-alkyl products, alkylation of 1,2-dihydroisoquinoline derivative, 1 in $\text{KNH}_2/\text{liq NH}_3$ followed by quenching with CH_3I was attempted This also led to the formation of 6a & 6'a along with the isoquinoline derivative, 5a When the alkylation of 1,2-dihydroisoquinoline, 1 was attempted with $\text{NaH}/\text{CH}_3\text{I}$, only 5a was obtained. It is strange that the N-alkylation does not take place under these reaction conditions



Scheme-1

EXPERIMENTAL

All melting points are uncorrected. UV (nm) and IR (cm^{-1}) spectra were recorded on HITACHI Model 557 Double wave length/Double beam and HITACHI 270-50 Infrared spectrophotometers respectively. NMR spectra were recorded on Jeol FX-90Q, 22 49MHz (^{13}C), Bruker ACF200, Bruker WH-270 and Bruker AMX400, 100 61MHz (^{13}C) spectrometers with Me_4Si as internal standard ($\delta = 0$ ppm). Mass spectra (70eV) were recorded on a Jeol MS-DX 303 spectrometer fitted with a built-in direct inlet system. Analytical and preparative TLC were carried out using silica gel. Column chromatography was carried out using silica gel. All organic extracts were dried over anhydrous Na_2SO_4 .

Potassamide reaction followed by *in situ* alkylation of 5,6-dihydroisoquinolines (2a-b): General Procedure

Freshly cut potassium (600 mg) was added to distilled ammonia (200 ml) and a pinch of ferric chloride was added and the solution stirred vigorously for about 45 minutes, after which a solution of the 5,6-dihydroisoquinoline (4 mmol) in dry THF (5 ml) was added in one lot. Stirring was continued for another hour after which alkyl iodide (8 mmol) was added. Solid NH_4Cl was added after 5 minutes to quench the reaction. The ammonia was allowed to evaporate and the residue, after dissolving in water was extracted with CHCl_3 . The organic layer was washed with water, dried and solvent removed. The mixture of products were separated by column chromatography followed by PTLC [hexane EtOAc, (8:1)].

Reaction of 4-cyano-3-methoxy-1-methyl-5,6-dihydroisoquinoline (2a):

(a) **Quenching with CH_3I :** Treatment of the 5,6-dihydroisoquinoline (2a) (800mg) with $\text{KNH}_2/\text{liq NH}_3$ and quenching with CH_3I gave (i) 4-cyano-1-ethyl-3-methoxy-5-methyl-5,6-dihydroisoquinoline, 3a (least polar, 270mg, 30%) viscous liquid, UV λ_{max} 324 (4040), 270 (12995), 242 (9333), IR (neat) 2220, 1625 cm^{-1} ; ^1H NMR (270MHz, CDCl_3) δ 1.18 (d, $J=7$ Hz, 3H), 1.26 (t, $J=7$ Hz, 3H), 2.24 (dd, $J=17.4$ & 7.2 Hz, 1H, $\text{C}_6\text{-H}$), 2.48-2.60 (m, 1H, $\text{C}_6\text{-H}$), 2.79 (q, $J=7.5$ Hz, 2H), 3.3 (qn, $J=7.5$ Hz, 1H, $\text{C}_5\text{-H}$), 4.02 (s, 3H, -OMe), 5.9-6.0 (m, 1H, $\text{C}_7\text{-H}$) and 6.55 (dd, $J=10$ & 3 Hz, $\text{C}_8\text{-H}$), ^{13}C NMR (22 49MHz, CDCl_3) δ 12.62(q), 19.50(q), 27.84(t), 28.92(t), 30.87(d), 54.01(q), 92.58(s), 114.47(s), 119.72(s), 120.80(d), 126.06(d), 156.94(s), 160.03(s), 162.19(s), MS m/e (relative intensity) 228 (M^+),

100 %), 227 (46), 213 (76), 199 (31), 185 (13), 170 (10), HRMS calcd for $C_{14}H_{16}N_2O$ 228.1263 Found, 228.1250, (ii) 4-cyano-3-methoxy-1,5-dimethyl-5,6-dihydroisoquinoline, **4a** (medium polar, 210mg, 25 %) m p 92-93°C, UV λ_{max} 324 (4193), 270 (11872), 244 (8866), IR (Nujol) 2220, 1625 cm^{-1} , 1H NMR (270MHz, $CDCl_3$) δ 1.16 (d, J=7.2Hz, 3H), 2.24 (dd, J=17.8 & 6.5Hz, 1H, C_6-H), 2.47 (s, 3H, ArMe), 2.47-2.60 (m, 1H, C_6-H), 3.29 (qn, J=7.3Hz, 1H, C_5-H), 4.0 (s, 3H, -OMe), 5.7-6.06 (m, 1H, C_7-H) and 6.48 (dd, J=10 & 2Hz, C_8-H), ^{13}C NMR (22.49MHz, $CDCl_3$) δ 19.19(q), 21.57(q), 28.62(t), 30.46(d), 53.87(q), 92.46(s), 114.14(s), 120.10(s), 120.87(d), 125.85(d), 155.22(s), 156.42(s), 161.73(s), MS m/e (relative intensity) 214 (M^+ , 100 %), 213 (42), 199 (81), 185 (22), 184 (20), 169 (10), Analysis calcd for $C_{13}H_{14}N_2O$, C, 72.90, H, 6.54, N, 13.08, Found, C, 72.97, H, 6.52, N, 13.09, (iii) 4-cyano-3-methoxy-1-methyl-5,6-dihydroisoquinoline, **5a** (60mg, 8 %, reported 5 m p 89-90°C), (iv) (1R*, 4R*)-4-cyano-1,4-dimethyl-1,2,3,4-tetrahydroisoquinolin-3(2H)-one, **6a** & (1S*, 4R*)-4-cyano-1,4-dimethyl-1,2,3,4-tetrahydroisoquinolin-3(2H)-one, **6'a** (most polar, 155mg, 20 %) m p 148-150°C, UV λ_{max} 260 (247), 257 (240), IR (nujol) 3290, 2260, 1680 cm^{-1} , 1H NMR (200MHz, $CDCl_3$) δ 1.59 (d, J=7Hz, C_1-Me), 1.68 (d, J=7Hz, $C_1'-Me$), 1.90 (s, C_4-Me), 1.99 (s, $C_4'-Me$), 4.75 (q, J=7.5Hz, C_1-H), 4.84 (dq, J=7.5 & 3Hz, C_1-H), 7.22-7.63 (m, ArH) 8.36 (bs, N_2-H , D_2O exchangeable), 8.43 (bs, N_2-H , D_2O exchangeable), ^{13}C NMR (100.61MHz, $CDCl_3$) δ 21.74(q), 25.38(q), 26.65(q), 29.56(q), 43.81(s), 44.79(s), 49.58(d), 51.15(d), 119.98 (2 x s), 125.22(d), 126.13(d), 126.35(d), 126.77(d), 128.49(d), 128.58(d), 128.79(d), 128.86(d), 131.95(s), 132.48(s), 134.18(s), 134.38(s), 167.62(s), 167.83(s), MS m/e (relative intensity) 200 (M^+ , 7 %), 185 (25), 157 (100), 142 (10), 130 (15), 115 (13), 82 (73), HRMS calcd for $C_{14}H_{16}N_2O$ 200.1233 Found, 200.1230

(b) Quenching with CH_3CH_2I : Treatment of the 5,6-dihydroisoquinoline (800mg) with $KNH_2/liq NH_3$ and quenching with CH_3CH_2I gave (i) 4-cyano-5-ethyl-3-methoxy-1-propyl-5,6-dihydroisoquinoline, **3b** (least polar, 250mg, 25 %) viscous liquid, UV λ_{max} 321 (2676), 297 (9018), 283 (10124), 245 (11287), IR (neat) 2220, 1625 cm^{-1} , 1H NMR (270MHz, $CDCl_3$) δ 0.92-1.01 (m, 6H), 1.5-1.75 (m, 4H), 2.4-2.47 (m, 2H, C_6-CH_2), 2.73 (t, J=6Hz, 2H, C_1-CH_2), 2.98-3.07 (m, 1H, C_5-H), 4.01 (s, 3H, -OMe), 5.88-5.95 (m, 1H, C_7-H) and 6.54 (dd, J=9 & 2Hz, 1H, C_8-H), Analysis calcd for $C_{16}H_{20}N_2O$, C, 75.00, H, 7.81; N, 10.94 Found, C, 75.08, H, 7.78, N, 10.91, (ii) 4-cyano-5-ethyl-1-methyl-3-methoxy-5,6-dihydroisoquinoline, **4b** (medium polar, 200mg, 22 %) viscous liquid, UV λ_{max} 322 (4013), 299 (5107), 272 (13133), 244 (11066), IR (neat) 2220, 1625 cm^{-1} , 1H NMR (270MHz, $CDCl_3$) δ 0.95 (t, J=7Hz, 3H), 1.55 (q, J=7Hz, 2H) 2.3-2.5 (m, 2H, C_6-CH_2), 2.47 (s, 3H, C_1-Me), 2.8-3.2 (m, 1H, C_5-H), 4.0 (s, 3H, -OMe), 5.9-5.96 (m, 1H, C_7-H) and 6.51 (dd, J=10 & 1Hz, C_8-H), MS m/e (relative intensity) 228 (M^+ , 63 %), 215 (12), 213 (15), 200 (20), 119 (100), 185 (13), 184 (34), 169 (15), HRMS calcd for $C_{14}H_{16}N_2O$, 228.1263 Found, 228.1251, (iii) 4-cyano-3-methoxy-1-methyl-5,6-dihydroisoquinoline, **5a** (60mg, 7 %, reported 5 m p 89-90°C), (iv) (1R*, 4R*)-4-cyano-4-ethyl-1-methyl-1,2,3,4-tetrahydroisoquinolin-3(2H)-one, **6b** & (1S*, 4R*)-4-cyano-4-ethyl-1-methyl-1,2,3,4-tetrahydroisoquinolin-3(2H)-one, **6'b** (most polar, 130mg, 15 %) m p 114-115°C, IR (nujol) 3320, 2240, 1670 cm^{-1} , 1H NMR (90MHz, $CDCl_3$) δ 1.07 (t, J=7.5Hz), 1.15 (t, J=7.5Hz), 1.56 (d, J=7Hz), 1.64 (d, J=7Hz), 2.05-2.38 (m), 4.69-4.86 (m, C_1-H), 7.16-7.60 (m, ArH), Analysis calcd for $C_{13}H_{14}N_2O$, C, 72.90, H, 6.54, N, 13.08 Found, C, 72.87, H, 6.51, N, 13.05

(c) Quenching with cyclohexyl iodide Treatment of 5,6-dihydroisoquinoline (800mg) with $KNH_2/liq NH_3$ and quenching with cyclohexyl iodide gave (i) 4-cyano-5,9-bis(cyclohexyl)-3-methoxy-5,6-dihydroisoquinoline, **3c** (least polar, 280mg, 20 %) viscous liquid, IR (nujol) 2220, 1630 cm^{-1} , 1H NMR (90MHz, $CDCl_3$) δ 0.92-1.42 (m, 11H), 1.45-1.91 (m, 11H), 2.31-2.96 (m, 5H), 4.0 (s, 3H, -OMe), 5.76-6.42 (m, 1H), 6.49 (dd, J=10 & 3Hz, 1H), MS m/e (relative intensity) 364 (M^+ , 15 %), 351 (10), 298 (10), 282 (100), 255 (25), 199 (40), 83 (45), 55 (50), HRMS calcd for $C_{24}H_{32}N_2O$ 364.2514 Found, 364.2496; (ii) 4-cyano-5-cyclohexyl-3-methoxy-1-methyl-5,6-dihydroisoquinoline, **4c** (medium polar, 180mg, 16 %) m p 96-97°C, IR (nujol) 2220, 1630 cm^{-1} , 1H NMR (90MHz, $CDCl_3$) δ 1.04-1.86 (m, 11H), 2.36-2.44 (m, 2H), 2.49 (s, 3H), 2.76-2.98 (m, 1H), 4.02 (s, 3H, -OMe), 5.82-6.06 (m, 1H), 6.49 (dd, J=10 & 3Hz, 1H); MS m/e (relative intensity) 282 (M^+ , 45 %), 267 (10), 253 (10), 225 (10), 199 (100), 184 (15), 169 (10), 81 (20), 55 (35), HRMS calcd for $C_{18}H_{22}N_2O$, 282.1732 Found, 282.1721, (iii)

4-cyano-3-methoxy-1-methyl-5,6-dihydroisoquinoline, **5a** (85mg, 10%, reported ⁵ m p. 89-90°C), (iv) (1R*, 4R*)-4-cyano-4-cyclohexyl-1-methyl-1,2,3,4-tetrahydroisoquinolin-3(2H)-one, **6c** & (1S*, 4R*)-4-cyano-4-cyclohexyl-1-methyl-1,2,3,4-tetrahydroisoquinolin-3(2H)-one, **6'c** (highly polar, 140mg, 13 %) m p. 112-113°C; IR (nujol) 3280, 2240, 1660 cm⁻¹, ¹H NMR (90MHz, CDCl₃) δ 1.36 (d, J=7.2Hz), 1.46-1.96 (m), 2.22-2.43 (m), 4.64-4.86 (m, C₁-H), 7.18-8.22 (m, ArH), Analysis calcd for C₁₇H₂₀N₂O C, 76.12, H, 7.46; N, 10.45. Found: C, 76.09; H, 7.44, N, 10.41.

Reaction of 4-cyano-1-ethyl-3-methoxy-5,6-dihydroisoquinoline (**2b**):

(a) **Quenching with CH₃I**: Treatment of the 5,6-dihydroisoquinoline (856mg) with KNH₂/liq NH₃ and quenching with CH₃I gave (i) 4-cyano-1-isopropyl-3-methoxy-5-methyl-5,6-dihydroisoquinoline, **3d** (least polar, 150mg, 16 %) viscous liquid, IR (nujol) 2220, 1625 cm⁻¹, ¹H NMR (200MHz, CDCl₃) δ 1.18 (d, J=6.7Hz, 3H), 1.23 (d, J=6.7Hz, 3H), 1.35 (d, J=7.2Hz, 3H), 2.43-2.83 (m, 2H, C₆-H), 3.11-3.29 (m, 2H) 4.02 (s, 3H, -OMe), 5.89-5.99 (m, 1H, C₇-H), 6.56 (dd, J=10 & 3Hz, C₈-H), MS m/e (relative intensity) 242 (M⁺, 60 %), 227 (100), 213 (65), 199 (60), 185 (39), 169 (15), 115 (20), 77 (20), HRMS calcd for C₁₅H₁₈N₂O 242.1623 Found, 242.1541, (ii) 4-cyano-1-ethyl-3-methoxy-5-methyl-5,6-dihydroisoquinoline, **3a** (medium polar, 150mg, 16 %) (iii) 4-cyano-1-ethyl-3-methoxy isoquinoline, **5b** (70mg, 8 %, reported ⁵ m p. 96-98°C), (iv) (1R*, 4R*)-4-cyano-1-ethyl-4-methyl-1,2,3,4-tetrahydroisoquinolin-3(2H)-one, **6d** & (1S*, 4R*)-4-cyano-1-ethyl-4-methyl-1,2,3,4-tetrahydroisoquinolin-3(2H)-one, **6'd** (highly polar, 100mg, 12 %) m p. 172-176°C, IR (nujol) 3250, 2240, 1670 cm⁻¹, ¹H NMR (60MHz, CDCl₃) δ 1.0 (t, J=7Hz), 1.97 (s), 2.02 (s), 1.57-1.9 (m), 4.4-4.75 (m, C₁-H), 7.1-7.67 (m, ArH) and 8.0 (bs, -CONH, D₂O exchangeable), Analysis calcd for C₁₃H₁₄N₂O C, 72.90, H, 6.54, N, 13.08 Found C, 72.83, H, 6.51, N, 13.01

(b) **Quenching with CH₃CH₂I**: Treatment of 5,6-dihydroisoquinoline(856mg) with KNH₂/liq NH₃ and quenching with CH₃CH₂I gave (i) 1-(2'-butyl)-4-cyano-5-ethyl-3-methoxy-5,6-dihydroisoquinoline, **3e** (least polar, 200mg, 19 %) viscous liquid, UV λ_{max} 325 (5815), 269 (16061), 244 (12254), IR (neat) 2230, 1630 cm⁻¹, ¹H NMR (270MHz, CDCl₃) δ 0.77-1.25 (m, 9H), 1.49-1.87 (m, 4H), 2.40-2.43 (m, 2H, C₆-H), 2.98-3.11 (m, 2H, C₅-H & C₁-H), 4.01 (s, 3H, -OMe), 5.88-5.95 (m, 1H, C₇-H), 6.61 (dd, J=10 & 2.5Hz, C₈-H), MS m/e (relative intensity) 270 (M⁺, 75 %), 271 (40), 269 (20), 255 (50), 242 (100), 228 (20), 213 (30), 185(30), HRMS calcd for C₁₇H₂₂N₂O 270.1732 Found, 270.1776, (ii) 4-cyano-1,5-diethyl-3-methoxy-5,6-dihydroisoquinoline, **4d** (medium polar, 450mg, 47 %) viscous liquid, UV λ_{max} 324 (4974), 270 (13550), 244 (10037), IR (neat) 2230, 1630, 1570 cm⁻¹, ¹H NMR (60MHz, CDCl₃) δ 0.95 (t, 7.5Hz, 3H), 1.25 (t, J=7Hz, 3H), 1.46-1.61 (m, 2H), 2.40-2.41 (m, 2H, C₆-2H), 2.79 (q, J=7Hz, 2H), 2.99-3.07 (m, 1H, C₅-H), 4.03 (s, 3H, -OMe), 5.89-5.96 (m, 1H, C₇-H), 6.55 (d, J=10Hz, C₈-H), ¹³C NMR (22.49MHz, CDCl₃) δ 11.51, 12.16, 25.38, 25.60, 27.33, 37.31, 53.57, 92.80, 114.37, 119.69, 120.77, 125.86, 155.78, 159.47, 161.74, MS m/e, (relative intensity) 242 (M⁺, 80 %), 243 (20), 241 (18), 227 (20), 213 (100), 198 (20), 185(30), 170(15), 145(15), HRMS calcd for C₁₅H₁₈N₂O 242.1419 Found, 242.1427, (iii) 4-cyano-1-ethyl-3-methoxyisoquinoline, **5b** (65mg, 8 %, reported ⁵ m p. 96-98°C), (iv) (1R*, 4R*)-4-cyano-1,4-diethyl-1,2,3,4-tetrahydroisoquinolin-3(2H)-one, **6e** & (1S*, 4R*)-4-cyano-1,4-diethyl-1,2,3,4-tetrahydroisoquinolin-3(2H)-one, **6'e** (highly polar, 100mg, 12 %) m p. 161-162°C, UV λ_{max} 261 (249), IR (nujol) 3300, 2260, 1680 cm⁻¹, ¹H NMR (270MHz, CDCl₃) δ 1.09 (t, J=7Hz), 1.11 (t, J=7Hz), 1.67-1.79 (m), 1.87-1.97 (m), 2.08-2.34 (m), 4.46-4.52 (m, C₁-H), 6.73 (bs, -CONH, D₂O exchangeable) 7.2-7.57 (m, ArH), MS m/e (relative intensity) 228 (M⁺, 5 %), 226(5), 200(14), 199(100), 172(11), 171(87), 143(11), 142(11), Analysis calcd for C₁₄H₁₆N₂O C, 73.68, H, 7.02, N, 12.28, Found C, 73.64, H, 7.06, N, 12.25

Methylation of mixture of compounds, **6a** & **6'a** with NaH/CH₃I:

NaH (15 mg, 0.5 mmol), mixture of compound, **6a** & **6'a** (40 mg, 0.2 mmol) and CH₃I (40 % molar excess) were taken in dry benzene (20 ml) and refluxed for 6 hrs. The reaction mixture was acidified with AcOH and the benzene layer was separated. The aqueous layer was extracted with two 5 ml portions of benzene and the combined benzene extract was washed with water, dried and the solvent evaporated. Separation of the crude mixture by PTLC (silica gel) gave (i) (1R*, 4R*)-4-cyano-1,2,4-trimethyl-1,2,3,4-

tetrahydroisoquinolin-3-one, **7** (24 mg, 60 %, semisolid), IR (nujol) 3330, 2240, 1675 cm^{-1} ; ^1H NMR (90MHz, CDCl_3) δ 1.64 (d, $J=7\text{Hz}$, 3H), 2.07 (s, 3H), 3.17 (s, 3H), 4.54 (q, $J=7\text{Hz}$, 1H), 7.12-7.53 (m, 4H), MS m/e (relative intensity) 214 (M^+ , 10 %), 199 (100), 184 (55), 171 (25), 157 (94), 130 (85), 115 (50); HRMS calcd. for $\text{C}_{13}\text{H}_{14}\text{N}_2\text{O}$ 214.1106. Found 214.1139; (ii) (1S*, 4R*)-4-cyano-1,2,4-trimethyl-1,2,3,4-tetrahydroisoquinolin-3-one, **8** (8 mg, 30 %, semisolid); IR (nujol) 3330, 2240, 1675 cm^{-1} ; ^1H NMR (90MHz, CDCl_3) δ 1.53 (d, $J=7\text{Hz}$, 3H), 1.90 (s, 3H), 3.16 (s, 3H), 4.57 (q, $J=7\text{Hz}$, 1H), 7.1-7.66 (m, 4H); MS m/e (relative intensity) 214 (M^+ , 10 %), 199 (100), 184 (55), 171 (25), 157 (94), 130 (85), 115 (50); HRMS calcd for $\text{C}_{13}\text{H}_{14}\text{N}_2\text{O}$ 214.1106, Found 214.1129.

Methylation of 4-cyano-3-methoxy-1-methyl-1,2-dihydroisoquinoline (**1**):

[a] with $\text{KNH}_2/\text{liq. NH}_3$: Treatment of 1,2-dihydroisoquinoline (400mg, 2 mmol) with $\text{KNH}_2/\text{liq. NH}_3$ followed by quenching with CH_3I gave the isoquinoline derivative, **5a** (100mg, 25 %) and the diastereomeric mixture of compounds, **6a** & **6'a** (200mg, 45 %).

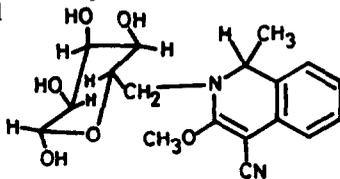
[b] with $\text{NaH}/\text{benzene}$: A mixture of 1,2-dihydroisoquinoline (50 mg, 0.25 mmol), NaH (10 mg, 0.35 mmol) and CH_3I (40 % molar excess) in dry benzene (10 ml) was refluxed for 6 hrs. The reaction mixture was acidified with AcOH and the benzene layer was separated, washed with water, dried and the solvent removed. The crude product obtained was purified by column chromatography (silica gel) to give exclusively compound, **5a**.

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References and Notes

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- 6 This was undertaken as part of a Department of Science and Technology, New Delhi, project, Synthesis of **11** was envisaged



11

- 7 The highly polar compound **D** could not be separated by TLC. However, methylation resulted in the separation of the two compounds
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