

Note

Organocatalytic aza-Michael Reaction to 3-Vinyl-1,2,4-Triazines as a Valuable Bifunctional Platform

Floris Buttard, Clément Berthonneau, Marie-Aude Hiebel, Jean-François Brière, and Franck Suzenet

J. Org. Chem., **Just Accepted Manuscript** • DOI: 10.1021/acs.joc.9b00141 • Publication Date (Web): 22 Feb 2019

Downloaded from <http://pubs.acs.org> on February 23, 2019

Just Accepted

“Just Accepted” manuscripts have been peer-reviewed and accepted for publication. They are posted online prior to technical editing, formatting for publication and author proofing. The American Chemical Society provides “Just Accepted” as a service to the research community to expedite the dissemination of scientific material as soon as possible after acceptance. “Just Accepted” manuscripts appear in full in PDF format accompanied by an HTML abstract. “Just Accepted” manuscripts have been fully peer reviewed, but should not be considered the official version of record. They are citable by the Digital Object Identifier (DOI®). “Just Accepted” is an optional service offered to authors. Therefore, the “Just Accepted” Web site may not include all articles that will be published in the journal. After a manuscript is technically edited and formatted, it will be removed from the “Just Accepted” Web site and published as an ASAP article. Note that technical editing may introduce minor changes to the manuscript text and/or graphics which could affect content, and all legal disclaimers and ethical guidelines that apply to the journal pertain. ACS cannot be held responsible for errors or consequences arising from the use of information contained in these “Just Accepted” manuscripts.

Organocatalytic aza-Michael Reaction to 3-Vinyl-1,2,4-Triazines as a Valuable Bifunctional Platform

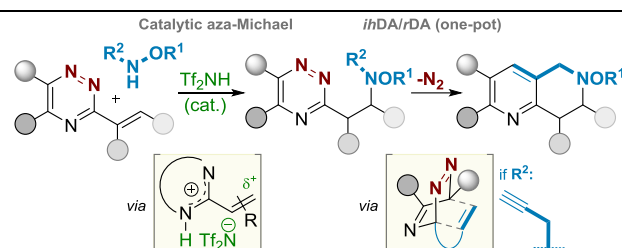
Floris Buttard,^a Clément Berthonneau,^b Marie-Aude Hiebel,^a Jean-François Brière,^{b*} and Franck Suzenet^{a*}

^a Université d'Orléans, CNRS, ICOA, UMR 7311, 45067 Orléans, France.

Phone: (+33)-2-3849-4580; E-mail: franck.suzenet@univ-orleans.fr

^b Normandie Univ, UNIROUEN, INSA Rouen, CNRS, COBRA, 76000 Rouen, France.

Phone: (+33)-2-3552-2464; e-mail: jean-francois.briere@insa-rouen.fr.



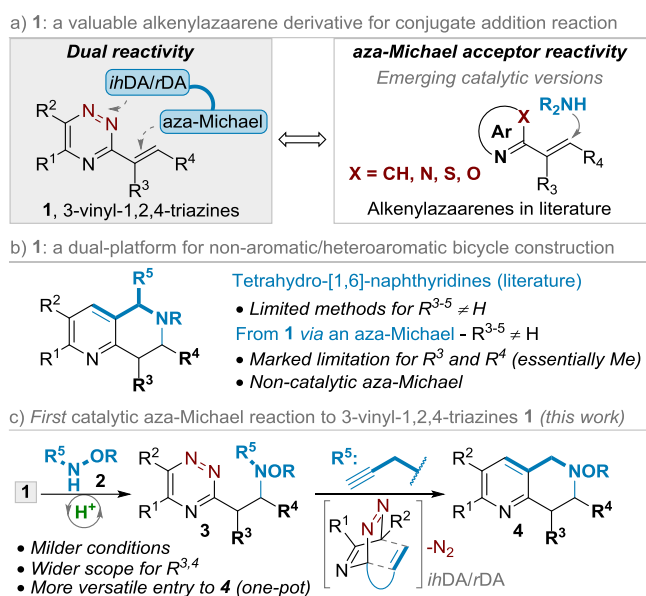
An unprecedented catalytic aza-Michael addition to substituted 3-vinyl-1,2,4-triazines, as original bifunctional platforms for domino conjugate addition-*ihDA/rDA* reaction, was achieved using the highly acidic triflimide as organocatalyst. Based on the use of alkoxyamine nucleophiles, this sequence not only highlights a rare example of catalytic aza-Michael reaction to alkenylazaarenes but proved to be useful for the elaboration of an array of biorelevant tetrahydro-[1,6]-naphthyridines.

Medicinal chemistry and chemical biology are currently seeking original heterocyclic scaffolds as biological tools. Accordingly, the development of new synthetic methodologies aiming at exploring chemical space, while increasing molecular diversity and complexity, is a current concern in organic synthesis.¹ The construction of fused non-aromatic/heteroaromatic bicycles has emerged as a promising strategy furnishing unprecedented potent bioactive

molecules.² The tuning of the selectivity and properties such as solubility was allowed by changing aromatic polycyclic drug candidates into partially saturated ones.³ Additionally, by escaping the flatland, the presence of sp^3 hybridized carbons and chiral stereocenters within bioactive compounds maximizes the on-target profile, which decreases the risk of toxicity issues in the clinic setting.⁴ However, the introduction of substituents on the saturated moiety of fused non-aromatic/heteroaromatic bicycles remains a challenging synthetic task.

We recently introduced 3-vinyl-1,2,4-triazines **1** behaving as (1) an original Michael acceptor belonging to the alkenylazaarene series,^{5,6} and (2) an aza-diene during the subsequent intramolecular domino *inverse-electron-demand hetero-Diels–Alder/retro-Diels–Alder* (*ihDA/rDA*) reaction (Scheme 1a).^{7,8} This versatile platform **1** gave access to non-aromatic/heteroaromatic bicycles⁸ displaying a privileged pyridine scaffold^{9,10,11} flanked by a saturated carba^{8c} and thia-ring.^{8a} In a recent aza-Michael based extension (Nu = N), platform **1** furnished a novel entry to tetrahydro-[1,6]-naphthyridines (Scheme 1b).^{8b}

Scheme 1. Context of the investigation



While this preliminary investigation afforded various derivatives functionalized on the pyridine ring, only a few substituents (*i.e.* methyl) were distributed on the saturated moiety.

Indeed, the 1,4-conjugate addition to **1** proved to be more challenging when the vinyl moiety is substituted. The reaction proceeded with high temperature or an excess of Lewis acid which poses compatibility issues with the subsequent (*ihDA/rDA*) reaction.^{8b} To overcome this limited reactivity, an improved catalytic aza-Michael reaction was sought which would also provide a new route to diversely substituted tetrahydro-[1,6]-naphthyridines. Importantly, despite the importance of this non-aromatic/heteroaromatic bicycles in medicinal chemistry, the reported syntheses addressed in some cases the substitution at C5 position (R^5),¹² but the strategies allowing the introduction of R^{3-4} remains rather limited.¹³

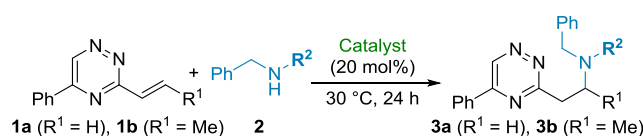
The conjugate addition reactions to alkenylazaarenes have elicited useful synthetic applications in organic synthesis,⁵ but the catalytic versions appeared only recently. Apart from metal-based catalysis,^{5a} a few elegant organocatalytic approaches,^{14,15,16} emerged to activate vinyl-derived pyridines ($X = CH$), diazole ($X = N$), -oxazole ($X = O$) and thiazole ($X = S$) compounds toward 1,4-conjugate addition reaction (Scheme 1a). Meanwhile, the catalytic aza-Michael reaction to alkenylazaarenes has remained elusive,^{5,17,18} until the recent asymmetric catalytic 1,4-conjugated addition of pyrazole-heterocycle derivatives to alkenylbenzimidazoles ($X = N$) described by the group of Terada.^{14c} Recently, Watson and colleagues reported an original catalytic and enantioselective aza-Michael-protonation reaction of aniline-derivatives to α -substituted vinyl-pyridines and derivatives ($R^3 = Ar$, $R^4 = H$, $X = CH, S, O$) in the presence of phosphoric acids.¹⁹

3-Vinyl-1,2,4-triazines **1** are novel platforms in this field of catalytic Michael reactions.²⁰ Herein, we are pleased to disclose an original Brønsted acid catalyzed aza-Michael reaction of alkoxyamine nucleophiles **2** to variously substituted 3-vinyl-1,2,4-triazines **1** (Scheme 1c). This organocatalytic approach not only overrides the previous limitation regarding the access of adducts **3** but also expands the scope for the subsequent construction of tetrahydro-[1,6]-

naphthyridines **4** diversely substituted on the saturated ring thanks to a straightforward *ihDA/rDA* sequence.

We previously observed the conjugate addition reaction of *N*-methylbenzylamine **2a** to unsubstituted 3-vinyl-1,2,4-triazine in MeOH at 30 °C to give adduct **3a** in quantitative yield (Table 1, entry 1).^{8b}

TABLE 1. Proof of principle^a

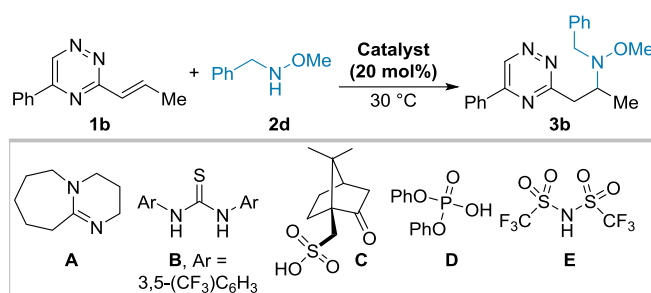


Entry	1 (R^1)	Amine (R^2)	Solvent	Cat.	Product (%) ^b
1	1a (H)	2a (Me)	MeOH	-	99% (3a)
2	1b (Me)	2a (Me)	MeOH	-	Traces
3	1b (Me)	2a (Me)	MeCN	CSA	0
4	1b (Me)	2b (H)	MeCN	CSA	0
5	1b (Me)	2c (Ph)	MeCN	CSA	0
6	1b (Me)	2d (OMe)	MeCN	CSA	72 ^c (3b)
7	1b (Me)	2d (OMe)	MeCN	-	0
8	1b (Me)	2e (Cbz)	MeCN	CSA	0
9	1b (Me)	2f (NMe ₂)	MeCN	CSA	0

^aReaction performed on 0.1 mmol of 3-vinyl-1,2,4-triazine **1a-b** (0.1 M in solvent), with amines **2** (1.5 equiv). ^b ¹H NMR yield with an internal standard. ^c 28% of starting material **1b** remained. CSA = (1S)-10-Camphorsulfonic acid.

However, this smooth reactivity was not adapted to more hindered 3-vinyl-1,2,4-triazines such as **1b** giving only traces of the corresponding adduct (entry 2). In line with the use of phosphoric acid in this field,^{14b-d,19} we foresaw that a more effective organocatalytic activation of the Michael acceptor **1b** would take place with a sufficiently strong acid catalyst through protonation of the triazine ring of **1b**. However, a Brønsted acid organocatalyst in the presence of a moderately basic triazine moiety may competitively lead to a non-productive protonation of either the amine **2** or the aza-Michael adduct **3**. First of all, we used camphor sulfonic acid (CSA, 20 mol%) as catalyst in acetonitrile with benzylamines **2a-d** bearing different *N*-functional groups (entries 3-6). While no transformation was observed with amines **2a-c** (entries 3-5), which possess carbon-based substituents, the methoxyamine

nucleophile **2d** (entry 6) led to the desired aza-Michael adduct **3b** in a promising 72% NMR yield. The CSA catalyst appeared to be essential for this transformation (entry 7), as well as the NHOMe motif since no aza-Michael reaction took place with NHCbz **2e** or hydrazine **2f** nucleophiles (entries 8-9). Although the exact mechanism deserves further investigations, it can be assumed that the moderate basicity of the methoxyamine **2d** minimizes its competitive (non-productive) protonation *versus* the protonative-activation of triazine **1b** while keeping a sufficiently nucleophilic character (in comparison to RNHCbz **2e** for instance) to secure the conjugate addition process.

TABLE 2. Catalyst screening^a

Entry	Catalyst (mol%)	Time (h)	Solvent	Product 3b (%) ^b
1	A (20)	24	MeCN	0
2	B (20)	24	MeCN	0
3	C (20)	24	MeCN	72
4	D (20)	24	MeCN	45
5	E (20)	24	MeCN	90 (81) ^c
6	E (20)	3	MeCN	54
7	E (20)	3	toluene	58
8	E (20)	3	THF	15
9	E (20)	3	CHCl_3	49
10	E (20)	3	Et_2O	58
11	E (20)	24	toluene	99
12	E (10)	24	toluene	89
13	E (5)	24	toluene	28

^aReaction performed on 0.1 mmol of 3-vinyl-1,2,4-triazine **1b** (0.1 M in solvent), with amine **2d** (1.5 equiv). ^b¹H NMR yield with an internal standard. ^cIsolated yield after silica gel column chromatography.

With this preliminary result in hand, different kind of organocatalysts (20 mol%) were subsequently evaluated with methoxybenzylamine nucleophile **2d** (Table 2, see supporting information for further details). It was demonstrated that neither Brønsted bases such as DBU **A** (entry 1), nor Schreiner's type catalyst **B** (entry 2), as a hydrogen bond donor species, were capable to provide any Michael-adduct. Brønsted acid catalysts, as expected, allowed the formation of adduct **3b** (entries 3-5). Interestingly, the efficiency of the reaction was directly correlated to the acidity of the catalyst. Lower activity was observed with phosphoric acid **D** (45%, entry 6, $\text{p}K_{\text{a}} \approx 3.8$ in DMSO) with regard to CSA **C** (72%, entry 5, $\text{p}K_{\text{a}} \approx 1.6$ in DMSO). Eventually, the highly acidic triflimide **E** proved to be the more potent organocatalyst (entry 5),^{21,22} allowing the formation of product **3b** with an excellent 90%

NMR yield after 24 hours reaction (81% isolated yield). Solvent optimization (entries 6-11) revealed that diethyl ether (entry 10), as a less polar solvent, gave the best conversions together with toluene in 3 hours (entry 7), which secured an excellent 99% NMR yield after 24 hours (entry 11). In these conditions, the triflimide catalyst **E** loading was decreased to 10 mol% with a minor impact on efficiency (entry 12, 89% yield) whereas a limit was reached with 5 mol% of **E** (entry 13, 28% yield after 24 hours).

We subsequently evaluated the scope of the optimal conditions (20 mol% of HNTf₂ **E** in toluene for 24 hours) by focusing on vinyl-triazines **1** with various substituents on the vinyl moiety (Figure 1). The precursors **1** were readily synthesized by means of a Liebeskind-Srogl-type coupling reaction²³ between alkenyl-boronic acid derivatives and 3-thiomethyl-functionalized triazines²⁴ in line with our previous developments.⁸ The conjugated addition reaction proceeded uneventfully with substrates possessing α - or β -substituted alkenes giving rise to adducts **3b** and **3c** with methyl group in 97% and 77% isolated yields, respectively. For comparison, we previously showed that the 2-propenyl platform **3c** required a large excess of BF₃.OEt₂ at refluxing THF to undergo the aza-Michael reaction.^{8b} Interestingly, an alcohol pendant was also tolerated with the catalytic conditions (**3d**, 85%). More hindered β -*n*-pentyl **3e** (79%), -cyclopropyl **3f** (46%) and -benzyl **3g** (39%) were also good substrates even if they required longer reaction times for completion (72-96 h). Due to instability issues on silica-gel column chromatography, **3g** was isolated in a moderate isolated 39% yield although 80% was estimated by ¹H NMR with an internal standard. Limitations were observed with more hindered cyclohexyl (**3h**) or phenyl (**3i**) derivatives. It was also proven that the reaction occurred efficiently with triazines flanked by various substituents such as 5-methyl (**3j**-73%), 5-CF₃ (**3k**-79%) or 5,6-diMe (**3l**-87%). The more challenging 5,6-unsubstituted triazine **1m**, as a less stable platform, was also transformed into **3m** with a promising 39% yield. The aza-Michael reaction was also extrapolated either to propargylamine nucleophiles **2g-h** and

primary BnONH₂ **2i** to furnish the corresponding adducts **3n-3p** in very good yields (81-91%), and up to 1.1 mmol scale for product **3n** (88% isolated yield).

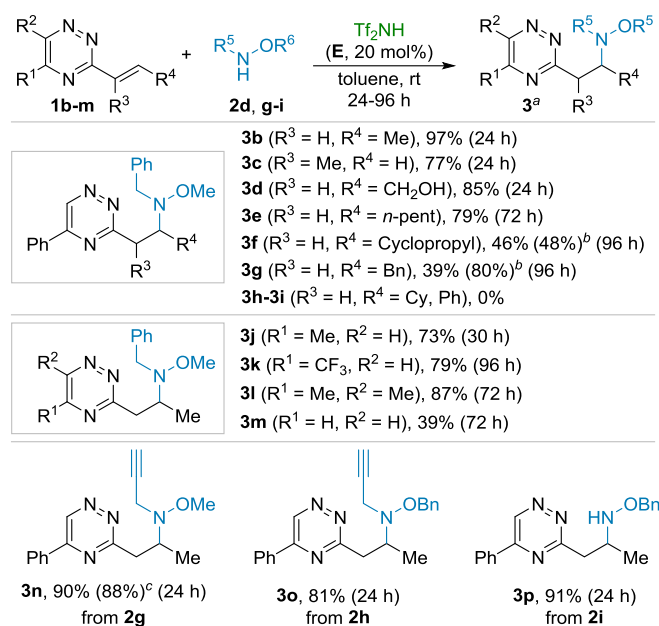


Figure 1. Scope and limitations of the aza-Michael reaction. ^aReaction performed on 0.1 mmol of 3-vinyl-1,2,4-triazines **1** (0.1 M), with amines **2** (1.5 equiv) in 24 hours. Isolated yield after silica gel column chromatography. ^b ¹H NMR yield determined by an internal standard into bracket. ^cCarried out on 1.1 mmol scale.

In order to address the synthesis of tetrahydro-[1,6]-naphthyridines such as **4a**, we investigated the *ihDA/rDA* sequence with the model substrate **3q** (Figure 2). The *hetero*-Diels-Alder reaction proceeded rapidly in 2 hours when 180 °C heating was carried out conveniently upon microwave (MW) irradiation. These conditions furnished **4a** in a complete conversion and high 95% yield up to 1.7 mmol scale. Importantly, it was observed that the pyridine product **4a**, or an intermediate derived thereof, were unstable upon longer heating time, as evidenced by a 79% yield obtained after 3 hours at 180 °C (Figure 2), showing the importance of mastering this cycloaddition step.

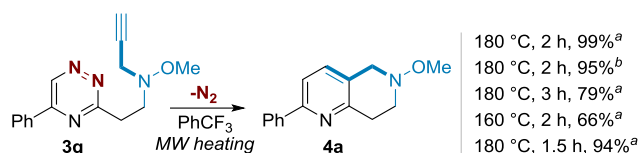


Figure 2. Optimization and prospective of the domino *ihDA/rDA* reaction. Conditions: starting material **3q** in CF₃Ph (0.1 M) was heated under microwave irradiation (MW). ^a ¹H NMR yield determined by an internal standard. ^b Isolated yield on 1.7 mmol scale after column chromatography.

Given this stability issue, a sequential aza-Michael reaction with a complete transformation followed by a well-controlled *ihDA/rDA* sequence upon heating in 2 hours was conducted in order to synthesize novel tetrahydro-[1,6]-naphthyridines **4** (Figure 3).^{8b}

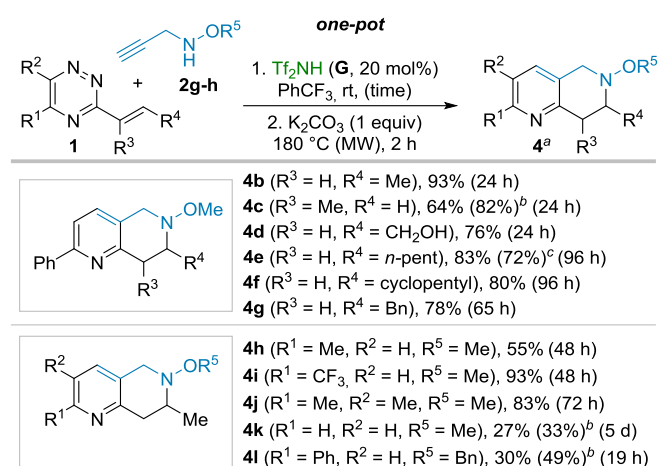


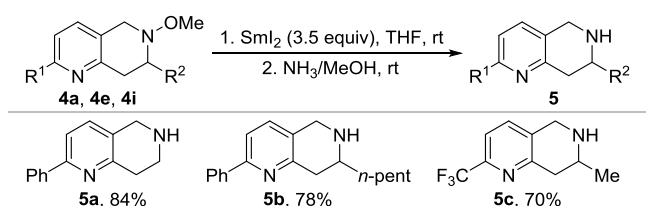
Figure 3. Scope and limitations of the domino sequential aza-Michael-*ihDA/rDA* reaction. ^aReaction performed on 0.3 mmol of 3-vinyl-1,2,4-triazines **1** (0.1 M), with propargylamines **2g-h** (1.5 equiv) in 24 hours. Isolated yield after silica gel column chromatography (over two steps). ^b ¹H NMR yield determined by an internal standard into bracket. ^cCarried out on 1 mmol scale for 90 hours.

This one-pot process was carried out in trifluorotoluene, as a microwave friendly solvent. Early attempts demonstrated that the presence of triflimide **E** led to major decomposition events during the cycloaddition process at high temperature. To prevent this acid promoted degradation, K₂CO₃ (1 equivalent) was added before the cycloaddition step allowing an efficient one-pot process (see experimental section) towards the formation of a series of novel

tetrahydro-[1,6]-naphthyridines **4b-j** flanked by various substituents as depicted in Figure 3, with good to excellent overall yields (55-93%). Even the more challenging 5,6-unsubstituted triazine starting material **1m** gave the corresponding product **4k** in 27% yield in two steps. Interestingly, the *N*-OBn derived product **4l** (via adduct **3n**) was also accessible albeit in a moderate 30% isolated yield, likely due to decomposition events.

Finally, in order to further exploit the new and valuable pyridine derivatives **4** in medicinal chemistry programs for example, we turned our attention to converting representative methoxyamine derived products **4** into the secondary amines **5**. A straightforward N-O bond cleavage took place in the presence of SmI₂, according to Brandi's procedure,²⁵ opening the access to *NH*-tetrahydro-[1,6]-naphthyridines **5a-c** with yields ranging from 70% to 84% (Scheme 2).

Scheme 2. N-O bond cleavage by SmI₂



In summary, the first catalytic aza-Michael reaction to substituted 3-vinyl-1,2,4-triazines **1** was achieved by means of the strong triflimide acid **E** as organocatalyst in the presence of alkoxyamines **2**. This approach shows a rare example of catalytic aza-Michael reaction to alkenylazaarene derivatives, and markedly broadens the scope of triazine aza-Michael adducts **3** accessible from various substituents on the vinyl moiety. Thanks to the unique dual reactivity of platform **1**, the subsequent *ihDA/rDA* reaction took place in a convenient one-pot fashion from the corresponding propargylamines **2** allowing the elaboration of various biorelevant tetrahydro-[1,6]-naphthyridine derivatives **4**. Investigations towards an enantioselective version are currently in progress.

Experimental Section

General Remarks

Unless otherwise specified, all reagent-grade chemicals and solvents commercially available were used without further purification. The reactions were monitored by thin-layer chromatography (TLC) using aluminum sheets coated with silica gel 60 F254. Flash column chromatography was carried out using silica gel 60 Å (0.04–0.06 mm). Solvents mentioned as dry were purified with a dry station GT S100 immediately prior to use. Microwave irradiation experiments were performed on a Monowave 300 single-mode microwave reactor (Anton Paar GmbH) using standard 10 or 30 mL Pyrex vessels (G10 or G30) equipped with Teflon-coated magnetic stir bar and closed with snap cap and silicone septum. The microwave apparatus has to be equipped with a safety pressure shutoff. Experiments carried out on 0.3 mmol scale of triazine in a 10 mL vial can generate 6–9 bars of pressure. NMR spectra were recorded with a 250 MHz (^1H : 250 MHz and ^{13}C : 63 MHz) or 400 MHz (^1H : 400 MHz and ^{13}C : 100.7 MHz) Bruker spectrometer. Unless noted the spectra were recorded at 25 °C. Chemical shifts are given in parts per million (ppm) from tetramethylsilane (TMS), calibrated to the residual solvent peak. Coupling constants “ J ” are expressed in hertz (multiplicity: s = singlet, br s = broad singlet, d = doublet, dd = double doublet, dt = double triplet, t = triplet, q = quartet, p = pentuplet, h = hexuplet, hept = heptuplet, m = multiplet ...). The ^{13}C NMR peak assignments have been confirmed using Heteronuclear Multiple Bond Correlation (HMBC) and Distortionless Enhancement by Polarization Transfer (DEPT-135) experiments. High-resolution accurate mass measurements (HRAM) were performed on a Bruker maXis mass spectrometer by the "Fédération de Recherche" ICOA/CBM (FR2708) platform using an electrospray ionization source (ESI) in positive mode and a time-of-flight (TOF) analyzer. Melting points were measured in open capillary tubes. The infrared spectra were recorded on a FT-IR Thermo Scientific Nicolet iS10 and maximum absorption wavenumbers ν_{max} are

given in cm^{-1} .

3-(methylthio)-1,2,4-triazine,²⁶ 3-(methylthio)-5-(trifluoromethyl)-1,2,4-triazine,^{10e} N-Benzyl-O-methylhydroxylamine **2d**,²⁷ benzyl N-benzylcarbamate **2e**²⁸ and 2-benzyl-1,1-dimethylhydrazine **2f**²⁹ were synthesized according to literature procedures. 3-(methylthio)-5-phenyl-1,2,4-triazine,⁶ 5,6-dimethyl-3-(methylthio)-1,2,4-triazine,^{8b} 3-vinyl-1,2,4-triazine **1a**,⁶ (E)-5-phenyl-3-(prop-1-en-1-yl)-1,2,4-triazine **1b**,^{8b} 5-phenyl-3-(prop-1-en-2-yl)-1,2,4-triazine **1c**^{8b} and (E)-3-(5-phenyl-1,2,4-triazin-3-yl)prop-2-en-1-ol **1d**^{8b} were synthesized as previously described.

5-methyl-3-(methylthio)-1,2,4-triazine

A solution of S-methylthiosemicarbazide iodohydrate (11.7 g, 50.2 mmol) in 50 mL of water at 0 °C was added to a stirred solution of pyruvaldehyde (40% in water, 9.2 mL, 61 mmol, 1.2 equiv) and sodium carbonate (5.8 g, 55 mmol, 1.1 equiv) at 0 °C. The mixture was then stirred at room temperature for 6 hours. The resulting aqueous media was extracted with dichloromethane (3 × 100 mL) and the combined organic extracts were dried over magnesium sulfate and concentrated on rotary evaporator. 3.86 g (55% yield) of the spectroscopically pure triazine were obtained after recrystallization from a 9:1 isopropanol/heptane mixture as a brown solid. ¹H NMR (250 MHz, chloroform-*d*): δ_{H} 8.81 (s, 1H), 2.67 (s, 3H), 2.49 (s, 3H). ¹³C{¹H} NMR (63 MHz, chloroform-*d*): δ_{C} 173.5 (C), 159.0 (C), 146.0 (CH), 21.8 (CH₃), 14.0 (CH₃). ESI⁺-HRAM: *m/z* calculated for [C₅H₈N₃S]⁺ ([M+H]⁺) 142.043345, found 142.043308. IR (neat): ν_{max} 3006, 2928, 1544, 1422, 1249. mp = 72-73 °C.

General procedure A for 3-vinyl-1,2,4-triazines (**1**) synthesis from boronic acid pinacol esters

3-methylthio-1,2,4-triazine, copper(I) 3-methylsalicylate (2.2 equiv) and vinylboronic acid pinacol ester (2.2 equiv) were dissolved in dry and degassed tetrahydrofuran (0.1 M of

triazine) under argon atmosphere. The solution was degassed by argon bubbling for 15 minutes before tetrakis(triphenylphosphine)palladium (5 mol%) was added. The mixture was subjected to argon bubbling for extra 15 minutes and subsequently heated up to reflux during 20 to 42 hours. Once cooled down to room temperature, the reaction mixture was filtered through a pad of diatomaceous earth. A 0.3 M aqueous solution of sodium bicarbonate was then added to the filtrate and the mixture was extracted with dichloromethane (three times). The combined organic layers were dried over magnesium sulfate and concentrated on rotary evaporator. The expected substituted 3-vinyl-1,2,4-triazine was finally isolated after silica gel column chromatography.

General procedure B for 3-vinyl-1,2,4-triazines (1) synthesis from boronic acids.

3-methylthio-5-phenyl-1,2,4-triazine, copper(I) 3-methylsalicylate (2.2 equiv) and vinylboronic acid (2.2 equiv) were dissolved in a degassed 9:1 tetrahydrofuran/water mixture (0.1 M of triazine) under argon atmosphere. The solution was degassed by argon bubbling for 15 minutes before tetrakis(triphenylphosphine)palladium (5 mol%) was added. The mixture was subjected to argon bubbling for extra 15 minutes and subsequently heated up to reflux during 20 to 48 hours. Once cooled down to room temperature, the reaction mixture was filtered through a pad of diatomaceous earth. A 0.3 M aqueous solution of sodium bicarbonate was then added to the filtrate and the mixture was extracted with dichloromethane (three times). The combined organic layers were dried over magnesium sulfate and concentrated on rotary evaporator. The attempted substituted 3-vinyl-1,2,4-triazine was finally isolated by silica gel column chromatography.

(*E*)-3-(hept-1-en-1-yl)-5-phenyl-1,2,4-triazine (1e).

Following the general procedure A from 3-methylthio-5-phenyl-1,2,4-triazine (500 mg, 2.46 mmol) and hept-1-en-1-ylboronic acid pinacol ester (1.22 g, 5.41 mmol, 2.2 equiv)³⁰ for 24 hours, the title compound (429 mg, 69% yield) was obtained as a yellow oil after purification

by column chromatography (eluent = 9:1 petroleum ether/ethyl acetate, R_f = 0.2).

^1H NMR (400 MHz, chloroform-*d*): δ_{H} 9.46 (s, 1H), 8.23 – 8.16 (m, 2H), 7.63 – 7.53 (m, 3H), 7.46 (dt, J = 15.6, 7.0 Hz, 1H), 6.82 (dt, J = 15.7, 1.5 Hz, 1H), 2.39 (qd, J = 7.2, 1.6 Hz, 2H), 1.65 – 1.53 (m, 2H), 1.43 – 1.34 (m, 4H), 0.98 – 0.85 (m, 3H). $^{13}\text{C}\{^1\text{H}\}$ NMR (101 MHz, chloroform-*d*): δ_{C} 163.7 (C), 155.0 (C), 144.9 (CH), 143.9 (CH), 134.0 (C), 132.4 (CH), 129.5 (2 CH), 127.7 (2 CH), 127.0 (CH), 33.1 (CH₂), 31.6 (CH₂), 28.4 (CH₂), 22.7 (CH₂), 14.2 (CH₃). ESI⁺-HRAM: m/z calculated for $[\text{C}_{16}\text{H}_{20}\text{N}_3]^+$ ($[\text{M}+\text{H}]^+$) 254.165174, found 254.165147. IR (neat): ν_{max} 2955, 2927, 2856, 1538, 1501, 1366, 765, 689.

(*E*)-3-(2-cyclopropylvinyl)-5-phenyl-1,2,4-triazine (1f).

Following the general procedure A from 3-methylthio-5-phenyl-1,2,4-triazine (456 mg, 2.25 mmol) and 2-cyclopropylvinylboronic acid pinacol ester (1.00 g, 4.95 mmol, 2.2 equiv) for 24 hours, the title compound (315 mg, 63% yield) was obtained as yellow solid after purification by silica gel column chromatography (eluent = 8:2 petroleum ether/ethyl acetate, R_f = 0.4).

^1H NMR (400 MHz, chloroform-*d*): δ_{H} 9.42 (s, 1H), 8.22 – 8.12 (m, 2H), 7.63 – 7.50 (m, 3H), 6.99 – 6.83 (m, 2H), 1.85 – 1.71 (m, 1H), 1.06 – 0.94 (m, 2H), 0.79 – 0.68 (m, 2H). $^{13}\text{C}\{^1\text{H}\}$ NMR (101 MHz, chloroform-*d*): δ_{C} 163.6 (C), 154.9 (C), 149.1 (CH), 143.6 (C), 134.0 (C), 132.3 (CH), 129.4 (2 CH), 127.6 (2 CH), 124.2 (C), 15.4 (CH), 8.8 (2 CH₂). ESI⁺-HRAM: m/z calculated for $[\text{C}_{14}\text{H}_{14}\text{N}_3]^+$ ($[\text{M}+\text{H}]^+$) 224.118224, found 224.117939. IR (neat): ν_{max} 3057, 3010, 2926, 1645, 1537, 1508, 1283, 944, 761, 687. mp = 65 °C.

(*E*)-5-phenyl-3-(3-phenylprop-1-en-1-yl)-1,2,4-triazine (1g)

Following the general procedure B from 3-methylthio-5-phenyl-1,2,4-triazine (553 mg, 2.72 mmol) and (3-phenylprop-1-en-1-yl)boronic acid (1.00 g, 5.99 mmol, 2.2 equiv) for 36 hours, the title compound (119 mg, 16% yield) was obtained as a yellow solid after purification by silica gel column chromatography (eluent = 9:1 pentane/ethyl acetate, R_f = 0.4).

^1H NMR (400 MHz, chloroform-*d*): δ_{H} 9.48 (s, 1H), 8.23 – 8.08 (m, 2H), 7.68 – 7.44 (m, 4H),

7.38 – 7.23 (m, 5H), 6.86 (dt, $J = 15.5, 1.6$ Hz, 1H), 3.73 (dd, $J = 7.0, 1.6$ Hz, 2H). $^{13}\text{C}\{^1\text{H}\}$ NMR (101 MHz, chloroform- d): δ_{C} 163.6 (C), 155.0 (C), 144.0 (CH), 142.6 (CH), 138.7 (C), 133.9 (C), 132.5 (CH), 129.5 (2 CH), 129.1 (2 CH), 128.8 (2 CH), 128.1 (CH), 127.7 (2 CH), 126.7 (CH), 39.4 (CH₂). ESI⁺-HRAM: m/z calculated for $[\text{C}_{18}\text{H}_{16}\text{N}_3]^+$ ($[\text{M}+\text{H}]^+$) 274.133874, found 274.133688. IR (neat): ν_{max} 3053, 2911, 1539, 1508, 1316, 974, 693. mp = 63–64 °C.

(E)-3-(2-cyclohexylvinyl)-5-phenyl-1,2,4-triazine (1h)

Following the general procedure A from 3-methylthio-5-phenyl-1,2,4-triazine (781 mg, 3.84 mmol) and (2-cyclohexylvinyl)boronic acid pinacol ester (2.00 g, 8.45 mmol, 2.2 equiv)³⁰ for 24 hours, the title compound (164 mg, 16% yield) was obtained as an orange oil after purification by silica gel column chromatography (eluent = dichloromethane, $R_f = 0.2$).

^1H NMR (400 MHz, chloroform- d): δ_{H} 9.46 (s, 1H), 8.26 – 8.04 (m, 2H), 7.61 – 7.52 (m, 3H), 7.41 (dd, $J = 16.0, 6.9$ Hz, 1H), 6.77 (dd, $J = 16.0, 1.5$ Hz, 1H), 2.39 – 2.26 (m, 1H), 1.95 – 1.86 (m, 2H), 1.86 – 1.77 (m, 2H), 1.76 – 1.68 (m, 1H), 1.40 – 1.20 (m, 5H). $^{13}\text{C}\{^1\text{H}\}$ NMR (101 MHz, chloroform- d): δ_{C} 164.0 (C), 155.1 (C), 149.8 (CH), 143.8 (CH), 134.0 (C), 132.4 (CH), 129.4 (2 CH), 127.7 (2 CH), 124.7 (CH), 41.3 (CH), 32.3 (2 CH₂), 26.2 (CH₂), 26.1 (2 CH₂). ESI⁺-HRAM: m/z calculated for $[\text{C}_{17}\text{H}_{20}\text{N}_3]^+$ ($[\text{M}+\text{H}]^+$) 266.165174, found 266.165352. IR (neat): ν_{max} 2923, 2850, 1537, 1503, 1360, 1282, 977, 689.

(E)-5-phenyl-3-styryl-1,2,4-triazine (1i).

Following the general procedure B from 3-methylthio-5-phenyl-1,2,4-triazine (203 mg, 1.00 mmol) and styrylboronic acid (326 mg, 2.2 mmol, 2.2 equiv) for 48 hours, the title compound (71 mg, 27% yield) was obtained as a yellow oil after purification by silica gel column chromatography (eluent = 9:1 petroleum ether/ethyl acetate, $R_f = 0.2$).

^1H NMR (250 MHz, chloroform- d): δ_{H} 9.50 (s, 1H), 8.29 – 8.19 (m, 3H), 7.73 – 7.67 (m, 2H), 7.65 – 7.56 (m, 3H), 7.49 (d, $J = 16.1$ Hz, 1H), 7.49 – 7.37 (m, 3H). $^{13}\text{C}\{^1\text{H}\}$ NMR (63 MHz, chloroform- d): δ_{C} 164.1 (C), 155.1 (C), 143.9 (CH), 140.1 (CH), 135.8 (C), 134.0 (C), 132.5

(CH), 129.8 (CH), 129.5 (2 CH), 129.1 (2 CH), 128.1 (2 CH), 127.8 (2 CH), 124.7 (CH).
ESI⁺-HRAM: m/z calculated for [C₁₇H₁₄N₃]⁺ ([M+H]⁺) 260.118224, found 260.118214. IR
(neat): $\nu_{\max}$ 3056, 1636, 1537, 1505, 1451, 973, 767, 681.

(E)-5-methyl-3-(prop-1-en-1-yl)-1,2,4-triazine (1j)

Following the general procedure A from 5-methyl-3-methylthio-1,2,4-triazine (370 mg, 2.62 mmol) and trans-1-propenylboronic acid pinacol ester (1.00 g, 5.77 mmol, 2.2 equiv) for 24 hours, the title compound (57 mg, 16% yield) was obtained as a light-yellow solid after purification by silica gel column chromatography (eluent = gradient from dichloromethane to 9:1 dichloromethane/ethyl acetate, R_f (9:1 CH₂Cl₂/EtOAc) = 0.4).

¹H NMR (400 MHz, chloroform-*d*): δ_H 8.89 (s, 1H), 7.31 (dq, J = 15.6, 6.9 Hz, 1H), 6.72 (dq, J = 15.6, 1.7 Hz, 1H), 2.52 (s, 3H), 2.01 (dd, J = 6.9, 1.7 Hz, 3H). ¹³C{¹H} NMR (101 MHz, chloroform-*d*): δ_C 163.3 (C), 159.0 (C), 147.7 (CH), 139.4 (CH), 128.2 (CH), 22.0 (CH₃), 18.7 (CH₃). ESI⁺-HRAM: m/z calculated for [C₇H₁₀N₃]⁺ ([M+H]⁺) 136.086924, found 136.086793. IR (neat): $\nu_{\max}$ 3043, 2917, 1655, 1548, 1356, 1283, 965. mp = 77–79 °C.

(E)-3-(prop-1-en-1-yl)-5-(trifluoromethyl)-1,2,4-triazine (1k).

Following the general procedure A from 3-methylthio-5-trifluoromethyl-1,2,4-triazine (512 mg, 2.62 mmol) and trans-1-propenylboronic acid pinacol ester (1.00 g, 5.77 mmol, 2.2 equiv) for 24 hours, the title compound (349 mg, 67% yield) was obtained as volatile bright-yellow liquid after purification by column chromatography (dry loading, eluent = gradient from pentane to 95:5 pentane/diethyl ether, R_f (95:5 petr. eth./Et₂O) = 0.5). N.B.: the solution obtained after column chromatography should not be heated on rotary evaporator for concentration (volatile compound).

¹H NMR (400 MHz, chloroform-*d*): δ_H 9.37 (s, 1H), 7.52 (dq, J = 15.6, 7.0 Hz, 1H), 6.89 (dq, J = 15.6, 1.8 Hz, 1H), 2.09 (dd, J = 7.0, 1.8 Hz, 3H). ¹³C{¹H} NMR (63 MHz, chloroform-*d*): δ_C 163.9 (C), 147.3 (q, ² J_{C-F} = 37.6 Hz, C), 143.1 (CH), 142.1 (q, ³ J_{C-F} = 2.5 Hz, CH), 126.8

(CH), 120.1 (q, $^1J_{\text{C-F}} = 276.2$ Hz, CF_3), 18.8 (CH_3). ESI⁺-HRAM: m/z calculated for $[\text{C}_7\text{H}_7\text{F}_3\text{N}_3]^+$ ($[\text{M}+\text{H}]^+$) 190.058658, found 190.058495. IR (neat): ν_{max} 3050, 2973, 2858, 1418, 1199, 1144.

(E)-5,6-dimethyl-3-(prop-1-en-1-yl)-1,2,4-triazine (1l).

Following the general procedure A from 5,6-dimethyl-3-methylthio-1,2,4-triazine (407 mg, 2.62 mmol) and trans-1-propenylboronic acid pinacol ester (1.00 g, 5.77 mmol, 2.2 equiv) for 20 hours, the title compound (373 mg, 95% yield) was obtained as a beige solid after purification by silica gel column chromatography (eluent = 8:2 pentane/ethyl acetate, $R_f = 0.3$).

^1H NMR (250 MHz, chloroform-*d*): δ_{H} 7.23 (dq, $J = 15.6, 7.0$ Hz, 1H), 6.69 (dq, $J = 15.6, 1.7$ Hz, 1H), 2.63 (s, 3H), 2.49 (s, 3H), 1.99 (dd, $J = 7.0, 1.7$ Hz, 3H). $^{13}\text{C}\{^1\text{H}\}$ NMR (63 MHz, chloroform-*d*): δ_{C} 162.2 (C), 158.3 (C), 155.1 (C), 137.8 (CH), 128.2 (CH), 22.0 (CH_3), 19.6 (CH_3), 18.6 (CH_3). ESI⁺-HRAM: m/z calculated for $[\text{C}_8\text{H}_{12}\text{N}_3]^+$ ($[\text{M}+\text{H}]^+$) 150.102574, found 150.102505. IR (neat): ν_{max} 3012, 2935, 2858, 1653, 1541, 1517, 995. mp = 41-42 °C.

(E)-3-(prop-1-en-1-yl)-1,2,4-triazine (1m).

Following the general procedure A from 3-methylthio-1,2,4-triazine (333 mg, 2.62 mmol) and trans-1-propenylboronic acid pinacol ester (1.00 g, 5.77 mmol, 2.2 equiv) for 20 hours, the title compound (172 mg, 54% yield) was obtained as yellow crystals after purification by silica gel column chromatography (eluent = 8:2 pentane/ethyl acetate, $R_f = 0.4$).

^1H NMR (250 MHz, chloroform-*d*): δ_{H} 9.01 (d, $J = 2.4$ Hz, 1H), 8.50 (d, $J = 2.4$ Hz, 1H), 7.34 (dq, $J = 15.6, 6.9$ Hz, 1H), 6.77 (dq, $J = 15.6, 1.7$ Hz, 1H), 2.03 (dd, $J = 6.9, 1.7$ Hz, 3H). $^{13}\text{C}\{^1\text{H}\}$ NMR (63 MHz, chloroform-*d*): δ_{C} 164.4 (C), 148.6 (CH), 147.2 (CH), 140.1 (CH), 128.1 (CH), 18.8 (CH_3). ESI⁺-HRAM: m/z calculated for $[\text{C}_6\text{H}_8\text{N}_3]^+$ ($[\text{M}+\text{H}]^+$) 122.071274, found 122.071296. IR (neat): ν_{max} 3050, 3034, 2966, 2943, 2915, 1655, 1550, 1527, 1432, 1409, 1553. mp = 67 °C.

***N*-methoxypropargylamine (2g).**

240 mL of a 5 M hydrochloric acid solution (1.2 mol, 10 equiv) in a water/methanol mixture, obtained by mixing 100 mL of a 37% aqueous hydrochloric acid solution and 140 mL of methanol, were added to stirred *N*-(tert-butoxycarbonyl)-*N*-(propargyl)methoxyamine (21.5 g, 116 mmol, prepared according to literature procedure).³¹ The mixture was subsequently refluxed for 3 hours (TLC monitoring using ethyl acetate as eluent). Once cooled down to room temperature, reactional mixture was diluted with 150 mL of distilled water and 150 mL of dichloromethane. Layers were separated, aqueous layer was washed with dichloromethane (150 mL) and ethyl acetate (150 mL) and finally evaporated on rotatory evaporator and dried under vacuum. The title ammonium salt was obtained as a white crystalline solid (8.71 g, 61% yield).

N-methoxypropargylamine hydrochloride (4.95 g, 40.7 mmol) was introduced in a 250 mL round bottom flask with 125 mL of diethyl ether. 90 mL (2.2 equiv) of a 1 M sodium hydroxide aqueous solution were then added and the biphasic mixture was vigorously stirred for 15 minutes. Layers were separated, aqueous phase was extracted with diethyl ether (2 × 50 mL). Gathered organic extracts were dried over magnesium sulfate and concentrated on rotary evaporator in a 0-5 °C cold bath. 3.28 g (95% yield) of the expected *N*-methoxypropargylamine were obtained as a colorless liquid.

¹H NMR (250 MHz, chloroform-*d*): δ_H 5.56 (t, *J* = 6.8 Hz, 1H), 3.67 (dd, *J* = 6.8, 2.5 Hz, 2H), 3.58 (s, 3H), 2.25 (t, *J* = 2.5 Hz, 1H). ¹³C NMR (63 MHz, chloroform-*d*): δ_C 80.3 (C), 71.9 (CH), 62.1 (CH₃), 41.1 (CH₂). ESI⁺-HRAM: the compound is too volatile to be detected.

***O*-benzyl-*N*-(prop-2-yn-1-yl)hydroxylamine (2h).**

A 5 M hydrochloric acid solution in a water/methanol mixture (obtained by diluting 3.2 mL of 37% aqueous hydrochloric acid in 4.4 mL of methanol, 7.6 mL, 38 mmol, 10 equiv) was slowly added to tert-butyl (benzyloxy)(prop-2-yn-1-yl)carbamate (1.00 g, 3.83 mmol,

prepared according to literature procedure)³¹ and the mixture was stirred 24 hours at room temperature (TLC monitoring, eluent = 9:1 petroleum ether/EtOAc, R_f = 0.5). 15 mL of dichloromethane were then added into the flask. After 15 minutes of stirring, phases were separated, and aqueous layer was washed with 25 mL of dichloromethane. 25 mL of sodium bicarbonate saturated aqueous solution were added to the aqueous phase which was subsequently extracted with dichloromethane (2 × 25 mL). The combined organic extracts were dried over magnesium sulfate and concentrated on rotary evaporator at room temperature, affording the spectroscopically pure propargylamine **2h** (455 mg, 74% yield).

¹H NMR (250 MHz, chloroform-*d*): δ_H 7.44 – 7.25 (m, 5H), 5.52 (br s, 1H), 4.77 (s, 2H), 3.66 (s), 2.25 (t, J = 2.5 Hz, 1H). ¹³C{¹H} NMR (63 MHz, chloroform-*d*): δ_C 137.8 (C), 128.6 (2 CH), 128.61 (2 CH), 128.55 (CH), 80.3 (C), 76.5 (CH₂), 72.0 (CH), 41.5 (CH₂). ESI⁺-HRAM: m/z calculated for [C₁₀H₁₂NO]⁺ ([M+H]⁺) 162.091340, found 162.091115. IR (neat): ν_{max} 3288, 2915, 2859, 1545, 697, 636.

General procedure C for the aza-Michael reaction to 3-vinyl-1,2,4-triazine. The alkoxyamine derivative **2** (1.5 equiv) was added to a solution of 3-vinyl-1,2,4-triazines **1** (0.1-1 mmol, 0.1 M) and trifluoromethanesulfonimide **E** (20 mol%) in toluene. The mixture was stirred at room temperature until a complete conversion of starting material **1** was observed by TLC. A saturated aqueous solution of sodium bicarbonate was then added and the organic phase was separated. The aqueous layer was extracted by dichloromethane (twice). The combined organic layers were dried over anhydrous magnesium sulfate and concentrated on rotary evaporator. The crude product was purified by silica gel column chromatography to yield the desired adduct **3**. If required the yield was also estimated on the crude product by ¹H NMR thanks to dibenzylether (0.25 equiv.) as an internal standard.

***N*-benzyl-*N*-methyl-2-(5-phenyl-1,2,4-triazin-3-yl)ethan-1-amine (3a).**

N-methybenzylamine (97%, 20 μ L, 18.8 mg, 0.15 mmol, 1.5 equiv) was added into a solution

of 5-phenyl-3-vinyl-1,2,4-triazine (18.3 mg, 0.10 mmol) in 1 mL of methanol. The mixture was stirred at 30 °C for 24 hours, concentrated on rotary evaporator and purified by column chromatography (eluent = 1:4 petroleum ether/ethyl acetate, R_f = 0.3) to afford the title product (31 mg, 0.10 mmol, quantitative yield) as a yellow oil.

^1H NMR (400 MHz, chloroform-*d*): δ_{H} 9.55 (s, 1H), 8.17 – 8.10 (m, 2H), 7.63 – 7.51 (m, 3H), 7.24 – 7.15 (m, 5H), 3.59 (s, 2H), 3.40 (t, J = 7.2 Hz, 2H), 3.06 (t, J = 7.2 Hz, 2H), 2.32 (s, 3H). $^{13}\text{C}\{^1\text{H}\}$ NMR (100.7 MHz, chloroform-*d*) δ_{C} 169.0 (C), 155.2 (C), 144.3 (CH), 139.1 (C), 133.8 (C), 132.5 (CH), 129.5 (2 CH), 129.1 (2 CH), 128.3 (2 CH), 127.8 (2 CH), 127.0 (CH), 62.2 (CH₂), 55.7 (CH₂), 42.2 (CH₃), 35.5 (CH₂). ESI⁺-HRAM: m/z calculated for [C₁₉H₂₁N₄]⁺ ([M+H]⁺) 305.176073, found 305.176150. IR (neat): ν_{max} 3058, 3027, 2788, 1601, 1544, 1508, 1495, 1445, 1319, 1075, 1046, 738, 690.

***N*-Benzyl-*O*-methyl-*N*-(1-(5-phenyl-1,2,4-triazin-3-yl)propan-2-yl)hydroxylamine (3b).**

Following the general procedure C from 5-phenyl-3-(prop-1-en-1-yl)-1,2,4-triazine **1b** (19.7 mg, 0.10 mmol) and *N*-methoxybenzylamine **2d** (19.8 μL , 0.15 mmol, 1.5 equiv) in 24 hours, the title compound (32.4 mg, 97% yield) was obtained as a yellow oil after purification by column chromatography (9:1 pentane/AcOEt, R_f = 0.3). ^1H NMR (400 MHz, chloroform-*d*): δ_{H} 9.55 (s, 1H), 8.16–8.08 (m, 2H), 7.64–7.51 (m, 3H), 7.26–7.16 (m, 5H), 3.98 (d, J = 13.0 Hz, 1H), 3.83 (d, J = 13.0 Hz, 1H), 3.76 (h, J = 6.8 Hz, 1H), 3.61 (dd, J = 6.8, 13.8 Hz, 1H), 3.28 (s, 3H), 3.22 (dd, J = 6.8, 13.8 Hz, 1H), 1.27 (d, J = 6.8 Hz, 3H). $^{13}\text{C}\{^1\text{H}\}$ NMR (100.7 MHz, chloroform-*d*): δ_{C} 169.0 (C), 155.0 (C), 144.1 (CH), 138.2 (C), 133.8 (C), 132.5 (CH), 129.6 (2 CH), 129.5 (2 CH), 128.2 (2 CH), 127.7 (2 CH), 127.2 (CH), 61.7 (CH₃), 59.3 (CH), 58.0 (CH₂), 41.5 (CH₂), 14.6 (CH₃). ESI⁺-HRAM: m/z calculated for [C₂₀H₂₃N₄O]⁺ ([M+H]⁺) 335.186638, found 335.186339. IR (neat): ν_{max} 2968, 2929, 2853, 1729, 1544, 1508, 1445, 1317, 1049, 762, 732, 691.

***N*-benzyl-*O*-methyl-*N*-(2-(5-phenyl-1,2,4-triazin-3-yl)propyl)hydroxylamine (3c).**

Following the general procedure A from 5-phenyl-3-(prop-1-en-2-yl)-1,2,4-triazine **1c** (19.7 g, 0.10 mmol) and *N*-methoxybenzylamine **2d** (19.8 μ L, 0.15 mmol, 1.5 equiv) in 24 hours, the title compound (25.9 g, 77% yield) was obtained as a yellow oil after purification by column chromatography (eluent = pentane/EtOAc, R_f = 0.3). ^1H NMR (400 MHz, chloroform-*d*): δ_{H} 9.55 (s, 1H), 8.14 – 8.01 (m, 2H), 7.64 – 7.48 (m, 3H), 7.19 – 7.14 (m, 5H), 3.93 (d, J = 12.8 Hz, 1H), 3.84 – 3.75 (m, 1H), 3.77 (d, 12.8 Hz, 1H), 3.37 (dd, J = 12.9, 9.5 Hz, 1H), 3.33 (s, 3H), 2.93 (dd, J = 12.7, 5.1 Hz, 1H), 1.41 (d, J = 7.0 Hz, 3H). $^{13}\text{C}\{^1\text{H}\}$ NMR (101 MHz, chloroform-*d*): δ_{C} 172.7 (C), 155.2 (C), 144.3 (CH), 137.4 (C), 134.0 (C), 132.4 (CH), 129.5 (2 CH), 129.4 (2 CH), 128.2 (2 CH), 127.7 (2 CH), 127.3 (CH), 63.3 (CH₂), 62.9 (CH₂), 61.0 (CH₃), 40.0 (CH), 18.6 (CH₃). ESI⁺-HRAM: m/z calculated for [C₂₀H₂₃N₄O]⁺ ([M+H]⁺) 335.186638, found 335.186299. IR (neat): ν_{max} 3060, 2971, 2934, 1542, 1508, 1319, 1049, 769, 692.

2-(benzyl(methoxy)amino)-3-(5-phenyl-1,2,4-triazin-3-yl)propan-1-ol (3d). Following the general procedure C from 3-(5-phenyl-1,2,4-triazin-3-yl)prop-2-en-1-ol **1d** (21.3 mg, 0.10 mmol) and *N*-methoxybenzylamine **2d** (19.8 μ L, 0.15 mmol, 1.5 equiv) in 24 hours, the title compound (29.8 mg, 85% yield) was obtained as a dark-yellow oil after purification by column chromatography (eluent = 1:1 pentane/EtOAc, R_f = 0.4). ^1H NMR (400 MHz, chloroform-*d*): δ_{H} 9.56 (s, 1H), 8.16 – 8.07 (m, 2H), 7.65 – 7.51 (m, 3H), 7.32 – 7.20 (m, 5H), 4.11 (d, J = 13 Hz, 1H), 4.02 (d, J = 13 Hz, 1H), 3.91 – 3.78 (m, 2H), 3.71 (dd, J = 14, 5.5 Hz, 1H), 3.68 – 3.60 (m, 1H), 3.37 (s, 3H), 3.29 (dd, J = 14, 7 Hz, 1H), 2.91 (br s, 1H). $^{13}\text{C}\{^1\text{H}\}$ NMR (101 MHz, chloroform-*d*): δ_{C} 168.6 (C), 155.2 (C), 144.2 (CH), 137.2 (C), 133.5 (C), 132.7 (CH), 129.7 (2 CH), 129.5 (2 CH), 128.4 (2 CH), 127.7 (2 CH), 127.6 (CH), 64.6 (CH), 62.3 (CH₂), 62.2 (CH₃), 58.9 (CH₂), 34.7 (CH₂). ESI⁺-HRAM: m/z calculated for [C₂₀H₂₃N₄O₂]⁺ ([M+H]⁺) 351.181552, found 351.181459. IR (neat): ν_{max} 3381, 3061, 3030, 2934, 2891, 1545, 1509, 1445, 1319, 1039, 1002, 907, 734, 691.

***N*-benzyl-*O*-methyl-*N*-(1-(5-phenyl-1,2,4-triazin-3-yl)heptan-2-yl)hydroxylamine (3e).**

Following the general procedure C from 3-(hept-1-en-1-yl)-5-phenyl-1,2,4-triazine **1e** (25.3 mg, 0.10 mmol) *N*-methoxybenzylamine **2d** (19.8 μ L, 0.15 mmol, 1.5 equiv) in 72 hours, the title compound (31 mg, 79% yield) was obtained as a yellow oil after purification by column chromatography (eluent = 9:1 pentane/EtOAc, R_f (8:2 petroleum ether/EtOAc) = 0.6). ^1H NMR (400 MHz, chloroform-*d*): δ_{H} 9.54 (s, 1H), 8.15 – 8.11 (m, 2H), 7.63 – 7.51 (m, 3H), 7.27 – 7.15 (m, 5H), 3.96 (d, J = 13.2 Hz, 1H), 3.91 (d, J = 13.2 Hz, 1H), 3.66 – 3.51 (m, 2H), 3.27 (dd, J = 12.8, 5.2 Hz, 1H), 3.24 (s, 3H), 1.89 – 1.74 (m, 1H), 1.59 – 1.47 (m, 2H), 1.47 – 1.38 (m, 1H), 1.28 (m, 4H), 0.87 (t, J = 6.8 Hz, 3H). $^{13}\text{C}\{^1\text{H}\}$ NMR (101 MHz, chloroform-*d*): δ_{C} 169.6 (C), 154.9 (C), 143.9 (CH), 138.5 (C), 133.9 (C), 132.4 (CH), 129.5 (2 CH), 129.4 (2 CH), 128.1 (2 CH), 127.7 (2 CH), 127.1 (CH), 63.7 (CH), 61.3 (CH₃), 57.4 (CH₂), 38.3 (CH₂), 32.0 (CH₂), 30.4 (CH₂), 26.7 (CH₂), 22.7 (CH₂), 14.2 (CH₃). ESI⁺-HRAM: m/z calculated for [C₂₄H₃₁N₄O]⁺ ([M+H]⁺) 391.249238, found 391.249002. IR (neat): ν_{max} 2930, 2857, 1544, 1508, 1496, 1318, 1047, 1001, 907, 758, 731, 690.

***N*-benzyl-*N*-(1-cyclopropyl-2-(5-phenyl-1,2,4-triazin-3-yl)ethyl)-*O*-**

methylhydroxylamine (3f). Following the general procedure A from 3-(2-cyclopropylvinyl)-5-phenyl-1,2,4-triazine **1f** (22 mg, 0.10 mmol) and *N*-methoxybenzylamine **2d** (19.8 μ L, 0.15 mmol, 1.5 equiv) in 96 hours, the title compound (17 mg, 46% yield) was obtained as a yellow oil after purification by silica gel column chromatography (eluent = 95:5 CH₂Cl₂/EtOAc, R_f = 0.4). ^1H NMR (400 MHz, chloroform-*d*): δ_{H} 9.55 (s, 1H), 8.23 – 8.03 (m, 2H), 7.63 – 7.52 (m, 3H), 7.25 – 7.15 (m, 5H), 4.13 (d, J = 13.1 Hz, 1H), 4.01 (d, J = 13.1 Hz, 1H), 3.71 (dd, J = 13.7, 8.3 Hz, 1H), 3.44 (dd, J = 13.7, 6.1 Hz, 1H), 3.24 (s, 3H), 2.90 (ddd, J = 9.7, 8.3, 6.1 Hz, 1H), 1.28 – 1.15 (m, 1H), 0.69 (m, 1H), 0.48 (m, 1H), 0.41 (m, 1H), -0.05 – -0.14 (m, 1H). $^{13}\text{C}\{^1\text{H}\}$ NMR (101 MHz, chloroform-*d*): δ_{C} 169.2 (C), 154.8 (C), 143.9 (CH), 138.5 (C), 133.9 (C), 132.4 (CH), 129.44 (2 CH), 129.39 (2 CH), 128.1 (2 CH), 127.7 (2 CH),

127.0 (CH), 69.4 (CH), 61.6 (CH₃), 58.5 (CH₂), 40.3 (CH₂), 11.6 (CH), 5.8 (CH₂), 2.8 (CH₂).

ESI⁺-HRAM: *m/z* calculated for [C₂₂H₂₅N₄O]⁺ ([M+H]⁺) 361.202288, found 361.202518.

IR (neat): $\nu_{\max}$ 3030, 2939, 2895, 1543, 1507, 1319, 1027, 691.

***N*-benzyl-*O*-methyl-*N*-(1-phenyl-3-(5-phenyl-1,2,4-triazin-3-yl)propan-2-yl)-**

hydroxylamine (3g). Following the general procedure C from 5-phenyl-3-(3-phenylprop-1-en-1-yl)-1,2,4-triazine **1g** (27 mg, 0.10 mmol) and *N*-methoxybenzylamine **2d** (19.8 μ L, 0.15 mmol, 1.5 equiv) in 96 hours, the title compound (15.9 mg, 39% yield) was obtained as a yellow oil after purification by silica gel column chromatography (eluent = 95:5 CH₂Cl₂/EtOAc, *R_f* = 0.3). ¹H NMR (400 MHz, chloroform-*d*): δ_{H} 9.47 (s, 1H), 8.08 – 8.01 (m, 2H), 7.62 – 7.49 (m, 3H), 7.25 – 7.18 (m, 4H), 7.15 (s, 5H), 7.15 – 7.10 (m, 1H), 4.06 (d, *J* = 13.1 Hz, 1H), 4.02 – 3.92 (m, 1H), 3.95 (d, *J* = 13.1 Hz, 1H), 3.55 (dd, *J* = 14.2, 8.3 Hz, 1H), 3.34 (dd, *J* = 13.7, 4.9 Hz, 1H), 3.30 (s, 3H), 3.27 (dd, *J* = 14.2, 5.8 Hz, 1H), 2.85 (dd, *J* = 13.7, 9.1 Hz, 1H). ¹³C{¹H} NMR (101 MHz, chloroform-*d*): δ_{C} 169.1 (C), 154.7 (C), 143.9 (CH), 139.8 (C), 138.0 (C), 133.8 (C), 132.4 (CH), 129.44 (CH), 129.42 (CH), 129.37 (CH), 128.5 (CH), 128.2 (CH), 127.7 (2 CH), 127.2 (2 CH), 126.1 (2 CH), 65.1 (CH), 61.4 (CH₃), 57.7 (CH₂), 38.7 (CH₂), 35.7 (CH₂). ESI⁺-HRAM: *m/z* calculated for [C₂₆H₂₇N₄O]⁺ ([M+H]⁺) 411.217938, found 411.218212. IR (neat): $\nu_{\max}$ 3028, 2939, 2179, 1544, 1508, 691.

***N*-benzyl-*O*-methyl-*N*-(1-(5-methyl-1,2,4-triazin-3-yl)propan-2-yl)hydroxylamine (3j).**

Following the general procedure C from 5-methyl-3-(prop-1-en-1-yl)-1,2,4-triazine **1j** (13.5 mg, 0.10 mmol) and *N*-methoxybenzylamine **2d** (19.8 μ L, 0.15 mmol, 1.5 equiv) in 30 hours, the title compound (19.9 mg, 73% yield) was obtained as a yellow oil after purification by column chromatography (eluent = 1:1 pentane/EtOAc, *R_f* = 0.4). N.B.: the mixture turned from yellow to dark brown after 1–2 hours. ¹H NMR (400 MHz, chloroform-*d*): δ_{H} 8.98 (s, 1H), 7.29 – 7.19 (m, 5H), 3.94 (d, *J* = 13.0 Hz, 1H), 3.79 (d, *J* = 13.0 Hz, 1H), 3.66 (h, *J* = 7.0 Hz, 1H), 3.50 (dd, *J* = 13.8, 7.0 Hz, 1H), 3.24 (s, 3H), 3.10 (dd, *J* = 13.8, 7.0 Hz, 1H), 2.50

(s, 3H), 1.21 (d, $J = 7.0$ Hz, 3H). $^{13}\text{C}\{^1\text{H}\}$ NMR (101 MHz, chloroform- d): δ_{C} 168.6 (C), 159.0 (C), 147.9 (CH), 138.3 (C), 129.6 (2 CH), 128.1 (2 CH), 127.1 (CH), 61.7 (CH₃), 59.4 (CH), 58.0 (CH₂), 41.3 (CH₂), 21.9 (CH₃), 14.4 (CH₃). ESI⁺-HRAM: m/z calculated for $[\text{C}_{15}\text{H}_{21}\text{N}_4\text{O}]^+$ ($[\text{M}+\text{H}]^+$) 273.170988, found 273.171000. IR (neat): ν_{max} 2971, 2937, 2895, 1604, 1548, 1530, 1359, 1035.

***N*-benzyl-*O*-methyl-*N*-(1-(5-(trifluoromethyl)-1,2,4-triazin-3-yl)propan-2-yl)-**

hydroxylamine (3k). Following the general procedure C from 5-trifluoromethyl-3-(prop-1-en-1-yl)-1,2,4-triazine **1k** (18.9 mg, 0.10 mmol) and *N*-methoxybenzylamine **2d** (19.8 μL , 0.15 mmol, 1.5 equiv) in 96 hours, the title compound (26 mg, 79% yield) was obtained as a bright-yellow oil after purification by silica gel column chromatography (eluent = 95:5 pentane/Et₂O, $R_f = 0.3$). ^1H NMR (400 MHz, chloroform- d): δ_{H} 9.47 (s, 1H), 7.25 – 7.12 (m, 5H), 3.95 (d, $J = 13.0$ Hz, 1H), 3.76 (d, $J = 13.0$ Hz, 1H), 3.76 – 3.60 (m, 2H), 3.35 – 3.25 (m, 1H), 3.19 (s, 3H), 1.28 (d, $J = 6.2$ Hz, 3H). $^{13}\text{C}\{^1\text{H}\}$ NMR (101 MHz, chloroform- d): δ_{C} 170.1 (C), 147.3 (q, $^2J_{\text{C-F}} = 37.7$ Hz, C), 142.7 (q, $^3J_{\text{C-F}} = 2.6$ Hz, CH), 137.8 (C), 129.4 (2 CH), 128.2 (2 CH), 127.3 (CH), 120.2 (q, $^1J_{\text{C-F}} = 276.0$ Hz, CF₃), 61.3 (CH₃), 59.3 (CH), 57.8 (CH₂), 41.4 (CH₂), 14.4 (CH₃). ESI⁺-HRAM: m/z calculated for $[\text{C}_{15}\text{H}_{18}\text{F}_3\text{N}_4\text{O}]^+$ ($[\text{M}+\text{H}]^+$) 327.142722, found 327.142697. IR (neat): ν_{max} 2976, 2937, 1334, 1199, 1143, 1036, 697.

***N*-benzyl-*N*-(1-(5,6-dimethyl-1,2,4-triazin-3-yl)propan-2-yl)-*O*-methylhydroxylamine**

(3l). Following the general procedure C from 5,6-dimethyl-3-(prop-1-en-1-yl)-1,2,4-triazine **1l** (14.9 mg, 0.10 mmol) and *N*-methoxybenzylamine **2d** (19.8 μL , 0.15 mmol, 1.5 equiv) in 72 hours, the title compound (25 mg, 87% yield) was obtained as a yellow oil after purification by column chromatography (eluent = 4:1 CH₂Cl₂/EtOAc, $R_f = 0.3$). ^1H NMR (400 MHz, chloroform- d): δ_{H} 7.28 – 7.21 (m, 5H), 3.94 (d, $J = 13.0$ Hz, 1H), 3.80 (d, $J = 13$ Hz, 1H), 3.63 (h, $J = 7$ Hz, 1H), 3.46 (dd, $J = 14, 7$ Hz, 1H), 3.26 (s, 3H), 3.05 (dd, $J = 14, 7$ Hz, 1H), 2.66 (s, 3H), 2.47 (s, 3H), 1.20 (d, $J = 7$ Hz, 3H). $^{13}\text{C}\{^1\text{H}\}$ NMR (101 MHz, chloroform-

d): δ_{C} 166.8 (C), 158.4 (C), 155.3 (CH), 138.4 (C), 129.6 (2 CH), 128.1 (2 CH), 127.1 (CH), 61.8 (CH₃), 59.4 (CH), 58.1 (CH₂), 40.8 (CH₂), 21.8 (CH₃), 19.5 (CH₃), 14.5 (CH₃). ESI⁺-HRAM: *m/z* calculated for [C₁₆H₂₃N₄O]⁺ ([M+H]⁺) 287.186638, found 287.186558. IR (neat): ν_{max} 3297, 2972, 2935, 2899, 1398, 1039, 735, 698.

***N*-(1-(1,2,4-triazin-3-yl)propan-2-yl)-*N*-benzyl-*O*-methylhydroxylamine (3m).** Following the general procedure C from 3-(prop-1-en-1-yl)-1,2,4-triazine **1m** (12.1 mg, 0.10 mmol) and *N*-methoxybenzylamine **2d** (19.8 μ L, 0.15 mmol, 1.5 equiv) in 72 hours, the title compound (10 mg, 39% yield) was obtained as a light-yellow oil after purification by silica gel column chromatography (eluent = 4:1 pentane/EtOAc, *R_f* = 0.3). ¹H NMR (400 MHz, chloroform-*d*): δ_{H} 9.11 (d, *J* = 2.4 Hz, 1H), 8.53 (d, *J* = 2.4 Hz, 1H), 7.27 – 7.22 (m, 5H), 3.95 (d, *J* = 13.1 Hz, 1H), 3.79 (d, *J* = 13.1 Hz, 1H), 3.68 (h, *J* = 7 Hz, 1H), 3.56 (dd, *J* = 13.8, 7 Hz, 1H), 3.23 (s, 3H), 3.19 (dd, *J* = 13.8, 7 Hz, 1H), 1.23 (d, *J* = 7 Hz, 3H). ¹³C{¹H} NMR (101 MHz, chloroform-*d*): δ_{C} 169.9 (C), 148.5 (C), 147.6 (CH), 138.2 (C), 129.6 (2 CH), 128.2 (2 CH), 127.2 (CH), 61.7 (CH₃), 59.6 (CH), 58.1 (CH₂), 41.4 (CH₂), 14.7 (CH₃). ESI⁺-HRAM: *m/z* calculated for [C₁₄H₁₉N₄O]⁺ ([M+H]⁺) 259.155338, found 259.155307. IR (neat): ν_{max} 3406, 2971, 2935, 2899, 1410, 1045, 735, 698.

***O*-methyl-*N*-(1-(5-phenyl-1,2,4-triazin-3-yl)propan-2-yl)-*N*-(prop-2-yn-1-yl)-hydroxylamine (3n).** Following the general procedure C from 5-phenyl-3-(prop-1-en-1-yl)-1,2,4-triazine **1b** (99 mg, 0.5 mmol) and *N*-methoxypropargylamine **2g** (67 μ L, 0.75 mmol, 1.5 equiv) in 24 hours, the title compound (127 mg, 90% yield) was obtained as a yellow oil after purification by silica gel column chromatography (eluent = 4:1 pentane/EtOAc, *R_f* = 0.25). From 5-phenyl-3-(prop-1-en-1-yl)-1,2,4-triazine **1b** (215 mg, 1.09 mmol) and *N*-methoxypropargylamine **2g** (150 μ L, 1.69 mmol, 1.5 equiv) in 24 hours, the title compound (270 mg, 88% yield) was obtained as a yellow oil after purification by silica gel column chromatography. ¹H NMR (400 MHz, chloroform-*d*): δ_{H} 9.54 (s, 1H), 8.21 – 8.17 (m, 2H),

7.62 – 7.51 (m, 3H), 3.88 (h, $J = 7$ Hz, 1H), 3.77 (dd, $J = 17.0, 2.5$ Hz, 1H), 3.62 (dd, $J = 17.0, 2.5$ Hz, 1H), 3.57 (s, 3H), 3.57 (dd, $J = 14.0, 7.0$ Hz, 1H), 3.21 (dd, $J = 14.0, 7.0$ Hz, 1H), 2.20 (t, $J = 2.5$ Hz, 1H), 1.19 (d, $J = 7.0$ Hz, 3H). $^{13}\text{C}\{^1\text{H}\}$ NMR (101 MHz, chloroform- d): δ_{C} 168.5 (C), 155.0 (C), 144.2 (CH), 133.8 (C), 132.5 (CH), 129.5 (2 CH), 127.7 (2 CH), 79.1 (C), 72.4 (CH), 61.6 (CH₃), 59.1 (CH), 43.2 (CH₂), 41.6 (CH₂), 15.7 (CH₃). ESI⁺-HRAM: m/z calculated for $[\text{C}_{16}\text{H}_{19}\text{N}_4\text{O}]^+$ ($[\text{M}+\text{H}]^+$) 283.155338, found 283.155508. IR (neat): ν_{max} 3291, 2935, 1508, 1041, 690.

***O*-benzyl-*N*-(1-(5-phenyl-1,2,4-triazin-3-yl)propan-2-yl)-*N*-(prop-2-yn-1-yl)-**

hydroxylamine (3o). Following the general procedure C from 5-phenyl-3-(prop-1-en-1-yl)-1,2,4-triazine **1b** (19.7 mg, 0.10 mmol) and *O*-benzyl-*N*-propargylhydroxylamine **2h** (24 mg, 0.15 mmol, 1.5 equiv) in 24 hours, the title compound (29.2 mg, 81% yield) was obtained as a yellow oil after purification by silica gel column chromatography (eluent = 4:1 pentane/EtOAc, $R_f = 0.3$). ^1H NMR (400 MHz, chloroform- d): δ_{H} 9.51 (s, 1H), 8.17 (dd, $J = 8.1, 1.7$ Hz, 2H), 7.64 – 7.49 (m, 3H), 7.38 – 7.24 (m, 5H), 4.83 (d, $J = 10.7$ Hz, 1H), 4.78 (d, $J = 10.7$ Hz, 1H), 3.94 (h, $J = 7$ Hz, 1H), 3.76 (dd, $J = 16.7, 2.5$ Hz, 1H), 3.65 (dd, $J = 16.7, 2.5$ Hz, 1H), 3.60 (dd, $J = 14.0, 7$ Hz, 1H), 3.25 (dd, $J = 14.0, 7$ Hz, 1H), 2.20 (t, $J = 2.5$ Hz, 1H), 1.23 (d, $J = 7$ Hz, 3H). $^{13}\text{C}\{^1\text{H}\}$ NMR (101 MHz, chloroform- d): δ_{C} 168.6 (C), 155.0 (C), 144.1 (CH), 137.3 (C), 133.8 (C), 132.5 (CH), 129.4 (2 CH), 128.9 (2 CH), 128.4 (2 CH), 128.0 (CH), 127.7 (2 CH), 79.5 (C), 76.3 (CH₂), 72.4 (CH), 59.5 (CH), 43.8 (CH₂), 41.4 (CH₂), 15.7 (CH₃). ESI⁺-HRAM: m/z calculated for $[\text{C}_{22}\text{H}_{23}\text{N}_4\text{O}]^+$ ($[\text{M}+\text{H}]^+$) 359.186638, found 359.186831. IR (neat): ν_{max} 3293, 2973, 2875, 1601, 1544, 1508, 691.

***O*-benzyl-*N*-(1-(5-phenyl-1,2,4-triazin-3-yl)propan-2-yl)hydroxylamine (3p).** Following the general procedure A from 5-phenyl-3-(prop-1-en-1-yl)-1,2,4-triazine **1b** (19.7 mg, 0.10 mmol) and *O*-benzylhydroxylamine **2i** (16 μL , 0.15 mmol, 1.5 equiv) in 24 hours, the title compound (29.1 mg, 91% yield) was obtained as a yellow oil after purification by silica gel

column chromatography (eluent = 65:35 petroleum ether/EtOAc, R_f = 0.3). ^1H NMR (400 MHz, chloroform- d): δ_{H} 9.52 (s, 1H), 8.22 – 8.09 (m, 2H), 7.65 – 7.50 (m, 3H), 7.39 – 7.20 (m, 5H), 4.71 (s, 2H), 3.87 – 3.74 (m, 1H), 3.43 (dd, J = 14.5, 7.3 Hz, 1H), 3.24 (dd, J = 14.5, 5.5 Hz, 1H), 1.24 (d, J = 6.5 Hz, 3H). $^{13}\text{C}\{^1\text{H}\}$ NMR (101 MHz, chloroform- d): δ_{C} 168.3 (C), 155.1 (C), 144.3 (CH), 138.0 (C), 133.7 (C), 132.6 (CH), 129.5 (2 CH), 128.40 (2 CH), 128.39 (2 CH), 127.8 (CH), 127.7 (2 CH), 76.7 (CH₂), 55.5 (CH), 41.8 (CH₂), 18.5 (CH₃). ESI⁺-HRAM: m/z calculated for $[\text{C}_{19}\text{H}_{21}\text{N}_4\text{O}]^+$ ($[\text{M}+\text{H}]^+$) 321.170988, found 321.171098. IR (neat): ν_{max} 2968, 2910, 1601, 1544, 1508.

***O*-methyl-*N*-(2-(5-phenyl-1,2,4-triazin-3-yl)ethyl)-*N*-(prop-2-yn-1-yl)hydroxylamine (3q).**

N-methoxypropargylamine hydrochloride **2g.HCl** (845 mg, 6.96 mmol, 1.5 equiv) was added into a mixture of 5-phenyl-3-vinyl-1,2,4-triazine **1a** (850 mg, 4.64 mmol) and sodium bicarbonate (585 mg, 6.96 mmol, 1.5 equiv) in 50 mL of tetrahydrofuran. The mixture was stirred at room temperature for 24 hours (it turned brown after a few hours). 50 mL of a 0.3 M sodium bicarbonate aqueous solution were then added, and the mixture was extracted with dichloromethane (3 × 50 mL). The combined organic extracts were dried over magnesium sulfate and concentrated on rotary evaporator. Purification by silica gel column chromatography (eluent = 8:2 petroleum ether/EtOAc, R_f = 0.3) afforded the title compound (960 mg, 3.58 mmol, 77% yield) as a yellow solid. ^1H NMR (400 MHz, chloroform- d): δ_{H} 9.55 (s, 1H), 8.21 – 8.16 (m, 2H), 7.63 – 7.52 (m, 3H), 3.64 (d, J = 2.5 Hz, 2H), 3.56 (s, 3H), 3.48 – 3.44 (m, 4H), 2.25 (t, J = 2.5 Hz, 1H). $^{13}\text{C}\{^1\text{H}\}$ NMR (101 MHz, chloroform- d): δ_{C} 168.4 (C), 155.0 (C), 144.2 (CH), 133.6 (C), 132.4 (CH), 129.4 (2 CH), 127.6 (2 CH), 78.4 (C), 73.0(CH), 61.1 (CH₃), 55.4 (CH₂), 46.5 (CH₂), 35.3 (CH₂). ESI⁺-HRAM: m/z calcd for $[\text{C}_{15}\text{H}_{17}\text{N}_4\text{O}]^+$ ($[\text{M}+\text{H}]^+$) 269.139688, found 269.139649 ; m/z calcd for $[\text{C}_{15}\text{H}_{16}\text{N}_4\text{NaO}]^+$ ($[\text{M}+\text{Na}]^+$) 291.121632, found 291.121608. IR (neat): ν_{max} 3263, 2931, 2859, 1552, 1517, 1319, 1044, 1032, 1000, 686, 667. mp = 57-59 °C.

6-methoxy-2-phenyl-5,6,7,8-tetrahydro-1,6-naphthyridine (4a). A solution of triazine **3q** (440 mg, 1.64 mmol) in 16 mL of α,α,α -trifluorotoluene was heated at 180 °C under microwaves irradiation for 2 hours. The mixture was then concentrated on rotary evaporator and the product was purified by silica gel column chromatography (eluent = 4:1 petroleum ether/EtOAc, R_f = 0.4) to afford the expected tetrahydro-[1,6]-naphthyridine (320 mg, 81% yield) as a beige solid. ^1H NMR (250 MHz, dimethylsulfoxide- d_6 , 80 °C): δ_{H} 8.08 – 7.96 (m, 2H), 7.69 (d, J = 8.0 Hz, 1H), 7.56 (d, J = 8.0 Hz, 1H), 7.54 – 7.35 (m, 3H), 4.10 (s, 2H), 3.55 (d, J = 0.4 Hz, 3H), 3.30 (t, J = 6.2 Hz, 2H), 3.04 (t, J = 6.2 Hz, 2H). $^{13}\text{C}\{^1\text{H}\}$ NMR (101 MHz, dimethylsulfoxide- d_6): δ_{C} 153.9 (C), 153.5 (C), 138.6 (C), 135.9 (CH), 128.7 (CH), 128.6 (2 CH), 127.4 (C), 126.4 (2 CH), 117.7 (CH), 58.7 (CH₃), 55.4 (CH₂), 51.4 (CH₂), 30.1 (CH₂). ESI⁺-HRAM: m/z calculated for $[\text{C}_{15}\text{H}_{17}\text{N}_2\text{O}]^+$ ($[\text{M}+\text{H}]^+$) 241.133540, found 241.133561. IR (neat): ν_{max} 2969, 2933, 2888, 2831, 2809, 1568, 1458, 1038, 774, 735, 694. mp = 62-64 °C.

General procedure D for the one-pot synthesis of *N*-methoxytetrahydro-[1,6]-naphthyridines (4). *N*-methoxy propargylamines **2g-h** (1.5 equiv) were added to a solution of 3-vinyl-1,2,4-triazines **1a-l** (0.3 mmol, 0.1 M) and trifluoromethanesulfonimide **E** (20 mol%) in α,α,α -trifluorotoluene in a microwaves vial. The mixture was stirred at room temperature until TLC monitoring revealed a complete transformation of starting material **1**. Potassium carbonate (1 equiv) was then added, the vial was sealed and heated at 180 °C under microwaves irradiation (gradient temperature from room temperature to 180 °C over 5 minutes) for 2 hours. At room temperature, the reaction mixture was subsequently filtered through cotton and purified by silica gel column chromatography to afford the expected tetrahydronaphthyridine **4**. Remark: NMR spectra for tetrahydro-[1,6]-naphthyridines were carried out at 80 °C in dimethylsulfoxide- d_6 in order to overcome the broad signal issues observed for the aliphatic cycle. We considered the difference of chemical shifts values for

tetramethylsilane between 25 and 80 °C insignificant.³² Chemical shifts were consequently calibrated on dimethylsulfoxide signals ($\delta = 2.50$ ppm for ^1H and $\delta = 39.52$ ppm for ^{13}C).

6-Methoxy-7-methyl-2-phenyl-5,6,7,8-tetrahydro-1,6-naphthyridine (4b). Following the general procedure D from 5-phenyl-3-(prop-1-en-1-yl)-1,2,4-triazine **1b** (59 mg, 0.30 mmol) and *N*-methoxypropargylamine **2g** (40 μL , 0.45 mmol, 1.5 equiv.) for 24 hours at room temperature (TLC monitoring in 8:2 petroleum ether/AcOEt), the title compound was obtained as a yellow solid (71 mg, 93%) after column chromatography (8:2 pentane/AcOEt, $R_f = 0.3$). ^1H NMR (250 MHz, dimethylsulfoxide- d_6 , 80 °C): δ_{H} 8.05–7.96 (m, 2H), 7.68 (d, $J = 8.0$ Hz, 1H), 7.56 (d, $J = 8.0$ Hz, 1H), 7.51–7.39 (m, 3H), 4.32 (d, $J = 16.0$ Hz, 1H), 4.00 (d, $J = 16.0$ Hz, 1H), 3.55 (s, 3H), 3.25 (dq, $J = 9.6, 6.3, 4.9$ Hz, 1H), 2.97 (dd, $J = 17.2, 4.9$ Hz, 1H), 2.82 (dd, $J = 17.2, 9.6$ Hz, 1H), 1.28 (d, $J = 6.3$ Hz, 3H). $^{13}\text{C}\{^1\text{H}\}$ NMR (63 MHz, dimethylsulfoxide- d_6 , 80 °C): δ_{C} 153.7 (C), 153.4 (C), 138.5 (C), 135.0 (CH), 128.1 (CH), 128.1 (2 CH), 126.9 (C), 126.0 (2 CH), 117.2 (CH), 59.8 (CH₃), 56.2 (CH), 53.8 (CH₂), 36.4 (CH₂), 18.0 (CH₃). ESI⁺-HRAM: m/z calculated for $[\text{C}_{16}\text{H}_{19}\text{N}_2\text{O}]^+$ ($[\text{M}+\text{H}]^+$) 255.149190, found 255.149401. IR (neat): ν_{max} 2981, 2941, 2843, 1463, 1449, 1028, 776, 695. mp = 65–66 °C.

6-methoxy-8-methyl-2-phenyl-5,6,7,8-tetrahydro-1,6-naphthyridine (4c). Following the general procedure D from 5-phenyl-3-(prop-1-en-2-yl)-1,2,4-triazine **1c** (59 mg, 0.30 mmol) and *N*-methoxypropargylamine **2g** (40 μL , 0.45 mmol, 1.5 equiv) for 24 hours at room temperature (TLC monitoring in 4:1 petroleum ether/EtOAc), the title compound was obtained as a brown oil (49 mg, 64% yield) after silica gel column chromatography (eluent = 4:1 petroleum ether/Et₂O, $R_f = 0.2$). ^1H NMR (250 MHz, dimethylsulfoxide- d_6 , 80 °C): δ_{H} 8.12 – 7.99 (m, 2H), 7.69 (d, $J = 8.0$ Hz, 1H), 7.55 (d, $J = 8.0$ Hz 1H), 7.55 – 7.35 (m, 3H), 4.13 (d, $J = 14.8$ Hz, 1H), 4.04 (d, $J = 14.8$ Hz, 1H), 3.55 (s, 3H), 3.43 (dd, $J = 11.0, 5.5$ Hz, 1H), 3.31 – 3.10 (m, 1H), 2.90 (dd, $J = 11.0, 7.7$ Hz, 1H), 1.43 (d, $J = 7.0$ Hz, 3H). $^{13}\text{C}\{^1\text{H}\}$

NMR (63 MHz, dimethylsulfoxide- d_6 , 80 °C): δ_C 157.0 (C), 153.8 (C), 138.5 (C), 135.3 (CH), 128.2 (CH), 128.1 (2 CH), 126.4 (CH), 125.9 (2 CH), 117.2 (CH), 59.0 (CH₂), 58.2 (CH₃), 56.03 (CH₂), 34.5 (CH), 18.1 (CH₃). ESI⁺-HRAM: m/z calculated for [C₁₆H₁₉N₂O]⁺ ([M+H]⁺) 255.149190, found 255.149032. IR (neat): $\nu_{\max}$ 2931, 2809, 1458, 1048, 762, 692.

(6-methoxy-2-phenyl-5,6,7,8-tetrahydro-1,6-naphthyridin-7-yl)methanol (4d). Following the general procedure D from 3-(5-phenyl-1,2,4-triazin-3-yl)prop-2-en-1-ol **1d** (64 mg, 0.30 mmol) and *N*-methoxypropargylamine **2g** (40 μ L, 0.45 mmol, 1.5 equiv) for 24 hours at room temperature (TLC monitoring in 1:1 petroleum ether/ EtOAc, N.B.: the mixture initially heterogeneous became limpid after 4 hours), the title compound was obtained as a brown solid (62 mg, 76% yield) after silica gel column chromatography (eluent = 1:4 petroleum ether/EtOAc, R_f = 0.2). ¹H NMR (250 MHz, dimethylsulfoxide- d_6 , 80 °C): δ_H 8.09 – 7.96 (m, 2H), 7.68 (d, J = 8.0 Hz, 1H), 7.55 (dt, J = 8.0, 0.8 Hz, 1H), 7.51 – 7.35 (m, 3H), 4.35 (br s, 1H), 4.33 (d, J = 16.4 Hz, 1H), 4.04 (dd, J = 16.4, 0.8 Hz, 1H), 3.80 (dd, J = 10.7, 5.3 Hz, 1H), 3.65 – 3.53 (m, 1H), 3.53 (s, 3H), 3.27 – 3.14 (m, 1H), 3.09 – 2.82 (m, 2H). ¹³C{¹H} NMR (63 MHz, dimethylsulfoxide- d_6 , 80 °C): δ_C 153.7 (C), 153.6 (C), 138.6 (C), 135.0 (CH), 128.1 (CH), 128.1 (2 CH), 127.0 (C), 126.0 (2 CH), 117.2 (CH), 62.5 (CH), 62.1 (CH₂), 59.5 (CH₃), 54.0 (CH₂), 31.1 (CH₂). ESI⁺-HRAM: m/z calculated for [C₁₆H₁₉N₂O₂]⁺ ([M+H]⁺) 271.144104, found 271.144071. IR (neat): $\nu_{\max}$ 3250, 2937, 2360, 1459, 1071, 1037, 770, 687. mp = 109-111 °C.

6-methoxy-7-pentyl-2-phenyl-5,6,7,8-tetrahydro-1,6-naphthyridine (4e). Following the general procedure D from 3-(hept-1-en-1-yl)-5-phenyl-1,2,4-triazine **1e** (76 mg, 0.30 mmol) and *N*-methoxypropargylamine **2g** (40 μ L, 0.45 mmol, 1.5 equiv) for 96 hours at room temperature (TLC monitoring in 9:1 petroleum ether/EtOAc), the title compound was obtained as a light-brown solid (77 mg, 83% yield) after silica gel column chromatography (eluent = 95:5 petroleum ether /EtOAc, R_f = 0.3). From 3-(hept-1-en-1-yl)-5-phenyl-1,2,4-

triazine **1e** (253 mg, 1.0 mmol) and *N*-methoxypropargylamine **2g** (133 μ L, 1.5 mmol, 1.5 equiv) for 96 hours at room temperature, the title compound was obtained as a light-brown solid (223 mg, 0.72 mmol, 72% yield) after silica gel column chromatography. ^1H NMR (250 MHz, dimethylsulfoxide- d_6 , 80 $^\circ\text{C}$): δ_{H} 8.08 – 7.96 (m, 2H), 7.68 (d, J = 7.99, 1H), 7.55 (dt, J = 8.0, 1.0 Hz, 1H), 7.54 – 7.32 (m, 3H), 4.32 (d, J = 16.8 Hz, 1H), 4.05 (dd, J = 16.8, 1.0 Hz, 1H), 3.52 (s, 3H), 3.16 – 3.04 (m, 2H), 3.00 – 2.69 (m, 2H), 1.86 – 1.70 (m, 1H), 1.50 (m, 2H), 1.45 – 1.21 (m, 4H), 0.98 – 0.83 (m, 3H). $^{13}\text{C}\{^1\text{H}\}$ NMR (63 MHz, dimethylsulfoxide- d_6 , 80 $^\circ\text{C}$): δ_{C} 153.7 (C), 153.6 (C), 138.6 (C), 134.9 (CH), 128.1 (CH), 128.0 (2 CH), 127.0 (C), 125.9 (2 CH), 117.2 (CH), 60.3 (CH), 59.4 (CH₃), 53.6 (CH₂), 33.3 (CH₂), 32.1 (CH₂), 31.0 (CH₂), 24.8 (CH₂), 21.6 (CH₂), 13.3 (CH₃). ESI⁺-HRAM: m/z calculated for $[\text{C}_{20}\text{H}_{27}\text{N}_2\text{O}]^+$ ($[\text{M}+\text{H}]^+$) 311.211790, found 311.211662. IR (neat): ν_{max} 2931, 2809, 2359, 1586, 1458, 1048, 762, 692. mp = 45-46 $^\circ\text{C}$.

7-cyclopropyl-6-methoxy-2-phenyl-5,6,7,8-tetrahydro-1,6-naphthyridine (4f). Following the general procedure D from 3-(2-cyclopropylvinyl)-5-phenyl-1,2,4-triazine **1e** (67 mg, 0.30 mmol) and *N*-methoxypropargylamine **2g** (40 μ L, 0.45 mmol, 1.5 equiv) for 96 hours at room temperature (TLC monitoring in 4:1 petroleum ether/EtOAc), the title compound was obtained as a light-brown solid (67 mg, 80% yield) after silica gel column chromatography (eluent = 9:1 pentane/EtOAc, R_f = 0.3). ^1H NMR (250 MHz, dimethylsulfoxide- d_6 , 80 $^\circ\text{C}$): δ_{H} 8.02 – 7.94 (m, 2H), 7.63 (d, J = 7.9 Hz, 1H), 7.50 (dt, J = 8.1, 0.8 Hz, 1H), 7.47 – 7.30 (m, 3H), 4.30 (d, J = 16.5 Hz, 1H), 3.97 (dd, J = 16.5, 0.8 Hz, 1H), 3.55 (s, 3H), 2.93 (d, J = 7.2 Hz, 2H), 2.54 – 2.37 (m, 1H), 1.06 – 0.91 (m, 1H), 0.60 – 0.45 (m, 2H), 0.45 – 0.34 (m, 1H), 0.28 – 0.16 (m, 1H). $^{13}\text{C}\{^1\text{H}\}$ NMR (63 MHz, dimethylsulfoxide- d_6 , 80 $^\circ\text{C}$): δ_{C} 153.7 (C), 153.5 (C), 138.5 (C), 134.9 (CH), 128.12 (CH), 128.07 (2 CH), 127.0 (C), 125.9 (2 CH), 117.2 (CH), 64.7 (CH), 59.6 (CH₃), 53.9 (CH₂), 33.4 (CH₂), 13.6 (CH), 3.6 (CH₂), 1.7 (CH₂). ESI⁺-HRAM: m/z calculated for $[\text{C}_{18}\text{H}_{21}\text{N}_2\text{O}]^+$ ($[\text{M}+\text{H}]^+$) 281.164840, found 281.164805.

IR (neat): $\nu_{\max}$ 3068, 2926, 2906, 1567, 1456, 1423, 1036, 769, 689. mp = 85-86 °C.

7-benzyl-6-methoxy-2-phenyl-5,6,7,8-tetrahydro-1,6-naphthyridine (4g). Following the general procedure D from 5-phenyl-3-(3-phenylprop-1-en-1-yl)-1,2,4-triazine **1g** (82 mg, 0.30 mmol) and *N*-methoxypropargylamine **2g** (40 μ L, 0.45 mmol, 1.5 equiv) for 65 hours at room temperature (TLC monitoring in 4:1 petroleum ether/EtOAc), the title compound was obtained as a brown solid (77 mg, 78% yield) after silica gel column chromatography (eluent = 9:1 pentane/EtOAc, R_f = 0.5). ^1H NMR (250 MHz, dimethylsulfoxide- d_6 , 80 °C): δ_{H} 8.03 – 7.95 (m, 2H), 7.68 (d, J = 8.0, 1H), 7.57 (d, J = 8.0, 1H), 7.50 – 7.35 (m, 3H), 7.35 – 7.29 (m, 4H), 7.28 – 7.14 (m, 1H), 4.37 (d, J = 16.5 Hz, 1H), 4.09 (d, J = 16.5 Hz, 1H), 3.58 (s, 3H), 3.47 (ddt, J = 9.9, 8.3, 5.5 Hz, 1H), 3.22 (dd, J = 13.5, 5.5 Hz, 1H), 2.91 (dd, J = 17.0, 9.9 Hz, 1H), 2.76 (dd, J = 13.5, 8.3 Hz, 1H), 2.72 (dd, J = 17.0, 5.5 Hz, 1H). $^{13}\text{C}\{^1\text{H}\}$ NMR (63 MHz, dimethylsulfoxide- d_6 , 80 °C): δ_{C} 153.7 (C), 153.4 (C), 138.8 (CH), 138.5 (CH), 135.0 (CH), 128.7 (2 CH), 128.11 (CH), 128.06 (2 CH), 127.8 (2 CH), 126.8 (C), 126.0 (2 CH), 125.5 (CH), 117.3 (CH), 61.8 (CH), 59.5 (CH₃), 53.6 (CH₂), 38.4 (CH₂), 33.1 (CH₂). ESI⁺-HRAM: m/z calculated for $[\text{C}_{22}\text{H}_{23}\text{N}_2\text{O}]^+$ ($[\text{M}+\text{H}]^+$) 331.180490, found 331.181164. IR (neat): $\nu_{\max}$ 2926, 1716, 1586, 1570, 1457, 995. mp = 113-115 °C.

6-methoxy-2,7-dimethyl-5,6,7,8-tetrahydro-1,6-naphthyridine (4h). Following the general procedure D from 5-methyl-3-(prop-1-en-1-yl)-1,2,4-triazine **1h** (28 mg, 0.2 mmol) and *N*-methoxypropargylamine **2g** (27 μ L, 0.31 mmol, 1.5 equiv) for 48 hours at room temperature (TLC monitoring in 7:3 petroleum ether/EtOAc), the title compound was obtained as a brown oil (21 mg, 55% yield) after silica gel column chromatography (eluent = 1:1 petroleum ether/EtOAc, R_f = 0.2). ^1H NMR (250 MHz, dimethylsulfoxide- d_6 , 80 °C): δ_{H} 7.34 (d, J = 7.8 Hz, 1H), 6.99 (d, J = 7.8 Hz, 1H), 4.22 (d, J = 15.7 Hz, 1H), 3.89 (d, J = 15.7 Hz, 1H), 3.52 (s, 3H), 3.27 – 3.02 (m, 1H), 2.84 (dd, J = 17.3, 5.0 Hz, 1H), 2.68 (dd, J = 17.3, 9.5 Hz, 1H), 2.39 (s, 3H), 1.23 (d, J = 6.3 Hz, 3H). $^{13}\text{C}\{^1\text{H}\}$ NMR (63 MHz, dimethylsulfoxide- d_6 , 80 °C):

δ (ppm) = 155.0 (C), 152.6 (C), 134.2 (CH), 124.8 (C), 120.0 (CH), 59.8 (CH₃), 56.2 (CH₂), 53.8 (CH), 36.4 (CH₂), 23.2 (CH₃), 17.9 (CH₃). ESI⁺-HRAM: m/z calculated for [C₁₁H₁₇N₂O]⁺ ([M+H]⁺) 193.133540, found 193.133203. IR (neat): $\nu_{\max}$ 2935, 2810, 2359, 1470, 1033.

6-methoxy-7-methyl-2-(trifluoromethyl)-5,6,7,8-tetrahydro-1,6-naphthyridine (4i).

Following the general procedure D from 3-(prop-1-en-1-yl)-5-(trifluoromethyl)-1,2,4-triazine **1i** (57 mg, 0.30 mmol) and *N*-methoxypropargylamine **2g** (40 μ L, 38 mg, 0.45 mmol, 1.5 equiv) for 48 hours at room temperature (TLC monitoring in 95:5 petroleum ether/EtOAc), the title compound was obtained as a brown oil (69 mg, 93% yield) after silica gel column chromatography (eluent = gradient from 9:1 to 8:2 pentane/EtOAc, R_f (8:2 petr. eth./EtOAc) = 0.5). ¹H NMR (250 MHz, dimethylsulfoxide-*d*₆, 80 °C): δ_H 7.78 (d, J = 7.7 Hz, 1H), 7.62 (d, J = 7.9 Hz, 1H), 4.39 (d, J = 16.8 Hz, 1H), 4.08 (d, J = 16.8 Hz, 1H), 3.53 (s, 3H), 3.29 (dq, J = 9.7, 6.4, 5.1 Hz, 1H), 2.95 (dd, J = 17.5, 5.1 Hz, 1H), 2.80 (dd, J = 17.5, 9.7 Hz, 1H), 1.26 (d, J = 6.4 Hz, 3H). ¹³C{¹H} NMR (63 MHz, dimethylsulfoxide-*d*₆, 80 °C): δ_C 155.0 (C), 144.13 (q, ² J_{C-F} = 33.6 Hz, C), 135.6 (CH), 132.5 (q, ⁵ J_{C-F} = 1.2 Hz, C), 121.3 (q, ¹ J_{C-F} = 274.0 Hz, CF₃), 117.5 (q, ³ J_{C-F} = 2.9 Hz, CH), 60.0 (CH₃), 55.6 (CH), 53.6 (CH₂), 35.1 (CH₂), 18.0 (CH₃). ESI⁺-HRAM: m/z calculated for [C₁₁H₁₄F₃N₂O]⁺ ([M+H]⁺) 247.105274, found 247.105466. IR (neat): $\nu_{\max}$ 3187, 3085, 2974, 2359, 1686, 1340, 1127.

6-methoxy-2,3,7-trimethyl-5,6,7,8-tetrahydro-1,6-naphthyridine (4j). Following the general procedure D from 5,6-dimethyl-3-(prop-1-en-1-yl)-1,2,4-triazine **1j** (45 mg, 0.30 mmol) and *N*-methoxypropargylamine **2g** (40 μ L, 38 mg, 0.45 mmol, 1.5 equiv) for 72 hours at room temperature (TLC monitoring in 1:1 petroleum ether/EtOAc), the title compound was obtained as a brown solid (52 mg, 83% yield) after silica gel column chromatography (eluent = 1:1 petroleum ether/EtOAc, R_f = 0.2). ¹H NMR (250 MHz, dimethylsulfoxide-*d*₆, 80 °C): δ_H 7.16 (s, 1H), 4.18 (d, J = 15.6 Hz, 1H), 3.86 (d, J = 15.6 Hz, 1H), 3.52 (s, 3H), 3.14 (dq, J =

9.5, 6.3, 5.0 Hz, 1H), 2.80 (dd, $J = 17.2, 5.0$ Hz, 1H), 2.64 (dd, $J = 17.2, 9.5$ Hz, 1H), 2.35 (s, 3H), 2.19 (s, 3H), 1.22 (d, $J = 6.3$ Hz, 3H). $^{13}\text{C}\{^1\text{H}\}$ NMR (63 MHz, dimethylsulfoxide- d_6 , 80 °C): δ_{C} 153.7 (C), 149.6 (C), 134.7 (CH), 127.8 (C), 125.1 (C), 59.8 (CH₃), 56.4 (CH), 53.6 (CH₂), 36.0 (CH₂), 21.3 (CH₃), 17.9 (CH₃), 17.6 (CH₃). ESI⁺-HRAM: m/z calculated for $[\text{C}_{12}\text{H}_{19}\text{N}_2\text{O}]^+$ ($[\text{M}+\text{H}]^+$) 207.149190, found 207.148912. IR (neat): ν_{max} 2971, 2931, 2897, 1443, 1195, 1042. mp = 64-65 °C.

6-methoxy-7-methyl-5,6,7,8-tetrahydro-1,6-naphthyridine (4k). Following the general procedure D from 3-(prop-1-en-1-yl)-1,2,4-triazine **1m** (36 mg, 0.30 mmol) and *N*-methoxypropargylamine **2g** (40 μL , 0.45 mmol, 1.5 equiv) for 5 days at room temperature (TLC monitoring in 7:3 petroleum ether/EtOAc), the title compound was obtained as a light-yellow oil (15 mg, 0.08 mmol, 27% yield) after column chromatography (eluent = 1:1 pentane/EtOAc, $R_f = 0.3$). ^1H NMR (250 MHz, dimethylsulfoxide- d_6 , 80 °C): δ_{H} 8.38 – 8.28 (m, 2H), 7.53 – 7.41 (m, 1H), 7.20 – 7.07 (m, 1H), 4.27 (d, $J = 16.0$ Hz, 1H), 3.96 (d, $J = 16.0$ Hz, 1H), 3.53 (s, 3H), 3.20 (ddd, $J = 9.6, 6.3, 5.0$ Hz, 1H), 2.89 (dd, $J = 17.4, 5.0$ Hz, 1H), 2.74 (dd, $J = 17.4, 9.6$ Hz, 1H), 1.25 (d, $J = 6.3$ Hz, 3H). $^{13}\text{C}\{^1\text{H}\}$ NMR (63 MHz, dimethylsulfoxide- d_6 , 80 °C): δ_{C} 153.6 (C), 146.9 (CH), 133.8 (CH), 128.0 (C), 120.7 (CH), 59.8 (CH₃), 54.0 (CH), 18.0 (CH₂), no signal for C⁸ (CH₂). ESI⁺-HRAM: m/z calculated for $[\text{C}_{10}\text{H}_{15}\text{N}_2\text{O}]^+$ ($[\text{M}+\text{H}]^+$) 179.117890, found 179.117741. IR (neat): ν_{max} 2969, 2934, 2896, 2360, 1446, 1032.

6-(benzyloxy)-7-methyl-2-phenyl-5,6,7,8-tetrahydro-1,6-naphthyridine (4l). Following the general procedure D from 5-phenyl-3-(prop-1-en-1-yl)-1,2,4-triazine **1b** (59 mg, 0.30 mmol) and *O*-benzyl-*N*-propargylhydroxylamine **2h** (73 mg, 0.45 mmol, 1.5 equiv.) for 19 hours at room temperature (TLC monitoring in 4:1 petroleum ether/EtOAc), the title compound was obtained as a beige solid (30 mg, 30% yield) after silica gel column chromatography (dry loading, eluent = 9:1 pentane/EtOAc, $R_f = 0.3$). ^1H NMR (250 MHz, dimethylsulfoxide- d_6 , 80

°C): δ_{H} 8.08 – 7.95 (m, 2H), 7.68 (d, $J = 8.0$ Hz, 1H), 7.55 (dt, $J = 8.0, 0.9$ Hz, 1H), 7.51 – 7.24 (m, 8H), 4.78 (s, 2H), 4.31 (d, $J = 16.2$ Hz, 1H), 4.05 (dd, $J = 16.2, 0.9$ Hz, 1H), 3.40 – 3.21 (m, 1H), 3.00 (dd, $J = 17.5, 5.1$ Hz, 1H), 2.84 (dd, $J = 17.5, 9.5$ Hz, 1H), 1.29 (d, $J = 6.4$ Hz, 3H). $^{13}\text{C}\{^1\text{H}\}$ NMR (63 MHz, dimethylsulfoxide- d_6 , 80 °C): δ_{C} 153.8 (C), 153.3 (C), 138.5 (C), 137.4 (C), 134.9 (CH), 128.2 (C), 128.10 (2 CH), 128.08 (2 CH), 127.7 (2 CH), 127.2 (CH), 126.9 (CH), 126.0 (2 CH), 117.3 (CH), 74.5 (CH_2), 56.4 (CH), 54.5 (CH_2), 38.5 (CH_2), 18.0 (CH_3). ESI⁺-HRAM: m/z calculated for $[\text{C}_{22}\text{H}_{23}\text{N}_2\text{O}]^+$ ($[\text{M}+\text{H}]^+$) 331.180490, found 331.180870. IR (neat): ν_{max} 3060, 2933, 2872, 1451, 1023, 694. mp = 89-90 °C.

General procedure E, Samarium iodide mediated N-O bond cleavage of 6-methoxytetrahydro-1,6-naphthyridines (5).²⁵ Samarium iodide (0.1 M solution in THF, 3.5 equiv) was added to the 6-methoxytetrahydro-1,6-naphthyridines **4** under argon atmosphere at room temperature. The dark-blue mixture was stirred at room temperature until complete conversion of starting material was observed by TLC. Then, an ammonia solution in methanol (1 M, 20 equiv) was added. After 30 minutes of stirring, the reaction medium was diluted with water and saturated with solid sodium thiosulfate. The mixture was subsequently extracted with diethyl ether, the combined organic layers were dried over magnesium sulfate and concentrated on rotary evaporator. The expected tetrahydro-1,6-naphthyridine **5** was purified by silica gel column chromatography.

2-Phenyl-5,6,7,8-tetrahydro-1,6-naphthyridine (5a). Following the general procedure E from 6-methoxy-2-phenyl-5,6,7,8-tetrahydro-1,6-naphthyridine **4a** (120 mg, 0.5 mmol) for 4 hours, the title compound was obtained as a light-yellow oil (88 mg, 84%) after column chromatography (4:1 AcOEt/EtOH, $R_f = 0.2$). ^1H NMR (400 MHz, chloroform- d): δ_{H} 8.04–7.81 (m, 2H), 7.53–7.34 (m, 5H), 4.05 (s, 2H), 3.27 (t, $J = 6.0$ Hz, 2H), 3.03 (t, $J = 6.0$ Hz, 2H), 2.28 (br s, 1H); $^{13}\text{C}\{^1\text{H}\}$ NMR (100.7 MHz, chloroform- d): δ_{C} 155.5 (C), 155.1 (C), 139.8 (C), 134.8 (CH), 129.8 (C), 128.8 (2 CH), 128.7 (CH), 127.0 (2 CH), 118.3 (CH), 47.6

(CH₂), 44.2 (CH₂), 33.0 (CH₂). ESI⁺-HRAM: *m/z* calculated for [C₁₄H₁₅N₂]⁺ ([M+H]⁺) 211.122975, found 211.122710. IR (neat): $\nu_{\max}$ 3367, 3036, 2926, 1636, 1588, 1253, 1025, 831, 773, 738.

7-pentyl-2-phenyl-5,6,7,8-tetrahydro-1,6-naphthyridine (5b). Following the general procedure E from 6-methoxy-7-pentyl-2-phenyl-5,6,7,8-tetrahydro-1,6-naphthyridine **4e** (93 mg, 0.5 mmol) for 10 hours, the title compound was obtained as a light-orange oil (65 mg, 78% yield) after silica gel column chromatography (eluent = EtOAc, *R_f* = 0.2). ¹H NMR (400 MHz, chloroform-*d*): δ_{H} 7.99 – 7.90 (m, 2H), 7.51 – 7.40 (m, 3H), 7.42 – 7.33 (m, 2H), 4.10 (d, *J* = 16.5 Hz, 1H), 4.06 (d, *J* = 16.5 Hz, 1H) 3.09 (dd, *J* = 16.9, 4.0 Hz, 1H), 2.98 (ddd, *J* = 10.5, 6.6, 4.0 Hz, 1H), 2.69 (dd, *J* = 16.9, 10.5 Hz, 1H), 1.83 (s, 1H), 1.64 – 1.54 (m, 2H), 1.48 (m, 2H), 1.35 (m, 4H), 0.97 – 0.87 (m, 3H). ¹³C{¹H} NMR (101 MHz, chloroform-*d*): δ_{C} 155.5 (C), 155.4 (C), 139.8 (C), 134.5 (CH), 129.4 (C), 128.8 (2 CH), 128.7 (CH), 127.0 (2 CH), 118.2 (CH), 54.2 (CH), 47.6 (CH₂), 39.3 (CH₂), 36.8 (CH₂), 32.1 (CH₂), 25.7 (CH₂), 22.8 (CH₂), 14.2 (CH₃). ESI⁺-HRAM: *m/z* calculated for [C₁₉H₂₅N₂]⁺ ([M+H]⁺) 281.201225, found 281.201334. IR (neat): $\nu_{\max}$ 2954, 2925, 2855, 2360, 1587, 1458, 692.

7-methyl-2-(trifluoromethyl)-5,6,7,8-tetrahydro-1,6-naphthyridine (5c). Following the general procedure E from 6-methoxy-7-methyl-2-(trifluoromethyl)-5,6,7,8-tetrahydro-1,6-naphthyridine **4i** (62 mg, 0.25 mmol) for 8 hours, the title compound was obtained as a yellow oil (38 mg, 70% yield) after silica gel column chromatography (eluent = 4:1 EtOAc/EtOH, *R_f* = 0.2). ¹H NMR (400 MHz, chloroform-*d*): δ_{H} 7.48 (d, *J* = 8.0 Hz, 1H), 7.43 (d, *J* = 7.9 Hz, 1H), 4.16 (d, *J* = 17.5 Hz, 1H), 4.08 (d, *J* = 17.5 Hz, 1H), 3.17 – 3.08 (m, 1H), 3.05 (dd, *J* = 17.4, 3.9 Hz, 1H), 2.67 (dd, *J* = 17.4, 10.6 Hz, 1H), 1.80 (s, 1H), 1.28 (d, *J* = 6.3 Hz, 3H). ¹³C{¹H} NMR (101 MHz, chloroform-*d*): δ_{C} 156.6 (C), 146.1 (q, ²*J*_{C-F} = 34.3 Hz, C), 134.9 (CH), 134.1 (C), 121.7 (q, ¹*J*_{C-F} = 274.0 Hz, CF₃), 117.8 (q, ³*J*_{C-F} = 2.8 Hz, CH), 49.4 (CH), 47.6 (CH₂), 40.5 (CH₂), 22.3 (CH₃). ESI⁺-HRAM: *m/z* calculated for [C₁₀H₁₂F₃N₂]⁺ ([M+H]⁺)

217.094709, found 217.094581. IR (neat): ν_{max} 3400, 3297, 2965, 2932, 1340, 1128, 1110.

Acknowledgment

This work was partially supported by INSA Rouen, Rouen University, Orléans University, CNRS, EFRD and Labex SynOrg (ANR-11-LABX-0029), Labex IRON (ANR-11-LABX-0018-01) région Normandie (CRUNCH network) and région Centre-Val de Loire.

Supporting Information Available

Supporting Information Available: general experimental information, copies of NMR spectra for all newly formed products and X-Ray crystal data for compounds **9a**, **9b** and **9d**. This material is available free of charge via the Internet at <http://pubs.acs.org>.

References

- (1) Blakemore, D. C.; Castro, L.; Churcher, I.; Rees, D. C.; Thomas, A. W.; Wilson, D. M.; Wood, A. Organic synthesis provides opportunities to transform drug discovery. *Nat. Chem.* **2018**, *10*, 383.
- (2) (a) Taylor, R. D.; MacCoss, M.; Lawson, A. D. G. Rings in Drugs. *J. Med. Chem.* **2014**, *57*, 5845. (b) Ritchie, T. J.; Macdonald, S. J. F.; Peace, S.; Pickett, S. D.; Luscombe, C. N. The developability of heteroaromatic and heteroaliphatic rings - do some have a better pedigree as potential drug molecules than others? *MedChemComm* **2012**, *3*, 1062. (c) Ritchie, T. J.; Macdonald, S. J. F.; Young, R. J.; Pickett, S. D. The impact of aromatic ring count on compound developability: further insights by examining carbo- and hetero-aromatic and -aliphatic ring types. *Drug Discovery Today* **2011**, *16*, 164.
- (3) (a) Hoegenauer, K.; Soldermann, N.; Zécari, F.; Strang, R. S.; Graveleau, N.; Wolf, R. M.; Cooke, N. G.; Smith, A. B.; Hollingworth, G. J.; Blanz, J.; Gutmann, S.; Rummel, G.; Littlewood-Evans, A.; Burkhardt, C. Discovery of CDZ173 (Leniolisib), Representing a Structurally Novel Class of PI3K Delta-Selective Inhibitors. *ACS Med. Chem. Lett.* **2017**, *8*, 975. (b) Clohessy, T. A.; Roberts, A.; Manas, E. S.; Patel, V. K.; Anderson, N. A.; Watson, A. J. B. Chemoselective One-Pot Synthesis of Functionalized Amino-azaheterocycles Enabled by COWare. *Org. Lett.* **2017**, *19*, 6368. (c) Ishikawa, M.; Hashimoto, Y. Improvement in Aqueous Solubility in Small Molecule Drug Discovery Programs by Disruption of Molecular Planarity and Symmetry. *J. Med. Chem.* **2011**, *54*, 1539.
- (4) (a) Schneider, P.; Schneider, G. Privileged Structures Revisited. *Angew. Chem. Int. Ed.* **2017**, *56*, 7971. (b) Lovering, F.; Bikker, J.; Humblet, C. Escape from Flatland: Increasing Saturation as an Approach to Improving Clinical Success. *J. Med. Chem.* **2009**, *52*, 6752. (c) Lovering, F. Escape from Flatland 2: complexity and promiscuity. *MedChemComm* **2013**, *4*, 515.
- (5) (a) Best, D.; Lam, H. W. C=N-Containing Azaarenes as Activating Groups in Enantioselective Catalysis. *J. Org. Chem.* **2014**, *79*, 831. (b) For reviews on conjugated

addition reaction to alkenylazaarenes, see: Klumpp, D. A. Conjugate Additions to Vinyl-Substituted Aromatic N-Heterocycles. *Synlett* **2012**, 23, 1590.

(6) To the best of our knowledge only one study briefly mentioned the use of 3-vinyl-1,2,4-triazine in an aza-Michael reaction: Badarau, E.; Bugno, R.; Suzenet, F.; Bojarski, A. J.; Finaru, A.-L.; Guillaumet, G. SAR studies on new bis-aryls 5-HT₇ ligands: Synthesis and molecular modeling. *Bioorg. Med. Chem.* **2010**, 18, 1958.

(7) (a) Raw, S. A.; Taylor, R. J. K. In *Advances in Heterocyclic Chemistry*; Alan, R. K., Ed.; Academic Press: 2010; Vol. Volume 100, p 75. (b) Foster, R. A. A.; Willis, M. C. Tandem inverse-electron-demand hetero-/retro-Diels-Alder reactions for aromatic nitrogen heterocycle synthesis. *Chem. Soc. Rev.* **2013**, 42, 63. (c) Boger, D. L. Diels-alder reactions of azadienes. *Tetrahedron* **1983**, 39, 2869. (d) Duret, G.; Le Fouler, V.; Bissere, P.; Bizet, V.; Blanchard, N. Diels–Alder and Formal Diels–Alder Cycloaddition Reactions of Ynamines and Ynamides. *Eur. J. Org. Chem.* **2017**, 2017, 6816.

(8) (a) Berthonneau, C.; Buttard, F.; Hiebel, M.-A.; Suzenet, F.; Brière, J.-F. Organocatalyzed Thia-Michael Addition and Sequential Inverse Electron Demanding Diels-Alder Reaction to 3-Vinyl-1,2,4- triazine Platforms. *Adv. Synth. Catal.* **2017**, 359, 4106. (b) Jouha, J.; Buttard, F.; Lorion, M.; Berthonneau, C.; Khouili, M.; Hiebel, M. A.; Guillaumet, G.; Brière, J. F.; Suzenet, F. Domino Aza-Michael-ih-Diels-Alder Reaction to Various 3-Vinyl-1,2,4-triazines: Access to Polysubstituted Tetrahydro-1,6-naphthyridines. *Org. Lett.* **2017**, 19, 4770. (c) Lorion, M.; Guillaumet, G.; Brière, J.-F.; Suzenet, F. Sequential Michael Addition and Enamine-Promoted Inverse Electron Demanding Diels–Alder Reaction upon 3-Vinyl-1,2,4-triazine Platforms. *Org. Lett.* **2015**, 17, 3154.

(9) (a) Vitaku, E.; Smith, D. T.; Njardarson, J. T. Analysis of the Structural Diversity, Substitution Patterns, and Frequency of Nitrogen Heterocycles among U.S. FDA Approved Pharmaceuticals. *J. Med. Chem.* **2014**, 57, 10257. (b) Hill, M. D. Recent Strategies for the Synthesis of Pyridine Derivatives. *Chem.-Eur. J.* **2010**, 16, 12052. (c) Allais, C.; Grassot, J.-M.; Rodriguez, J.; Constantieux, T. Metal-Free Multicomponent Syntheses of Pyridines. *Chem. Rev.* **2014**, 114, 10829.

(10) (a) Platform **1** affords a new manifold in the field of the intramolecular ihDA/rDA reaction of 1,2,4-triazines towards the construction of original fused-pyridine derivatives; for representative examples, see: Seitz, G.; Dietrich, S. Intramolekulare[4+2]-Cycloadditionen substituierter Triazine mit inversem Elektronenbedarf

Intramolecular[4+2]-Cycloadditions of Substituted Triazines with Inverse Electron Demand. *Archiv der Pharmazie* **1984**, 317, 379. (b) Taylor, E. C.; Macor, J. E. Novel intramolecular diels-alder reactions with alkynylthio derivatives of 1,2,4-triazines. New routes to thieno[2,3-b]pyridines and thieno[2,3-c]pyridines. *Tetrahedron Lett.* **1985**, 26, 2419. (c) Taylor, E. C.; Pont, J. L. Further intramolecular diels-alder reactions of 1,2,4-triazines. Synthesis of dihydropyrrolo[2,3-b]pyridines. *Tetrahedron Lett.* **1987**, 28, 379. (d) Taylor, E. C.; Macor, J. E.; French, L. G. Intramolecular Diels-Alder reactions of 1,2,4-triazines. Synthesis of 2,3-cyclopentenopyridines and 5,6,7,8-tetrahydroquinolines. *J. Org. Chem.* **1991**, 56, 1807. (e) Haenel, F.; John, R.; Seitz, G. Trifluormethyl-substituierte, heterocyclisch anellierte Pyridine durch intramolekulare Diels-Alder-Cycloaddition mit inversem Elektronenbedarf. *Archiv der Pharmazie* **1992**, 325, 349. (f) Branowska, D.; Ostrowski, S.; Rykowski, A. Tandem Vicarious Nucleophilic Substitution of Hydrogen/Intramolecular Diels–Alder Reaction of 1,2,4-Triazines into Functionalized Cycloalkenopyridines. *Chem. Pharm. Bull.* **2002**, 50, 463. (g) Diring, S.; Retailleau, P.; Ziesel, R. A Rational Protocol for the Synthesis of Arylated Bipyridine Ligands via a Cycloaddition Pathway. *J. Org. Chem.* **2007**, 72, 10181. (h) Hajbi, Y.; Suzenet, F.; Khouili, M.; Lazar, S.; Guillaumet, G. Polysubstituted 2,3-

dihydrofuro[2,3-b]pyridines and 3,4-dihydro-2H-pyrano[2,3-b]pyridines via microwave-activated inverse electron demand Diels–Alder reactions. *Tetrahedron* **2007**, 63, 8286. (i) Akritopoulou-Zanze, I.; Wang, Y.; Zhao, H.; Djuric, S. W. Synthesis of substituted fused pyridines, pyrazines and pyrimidines by sequential Ugi/inverse electron demand Diels–Alder transformations. *Tetrahedron Lett.* **2009**, 50, 5773. (j) Hajbi, Y.; Suzenet, F.; Khouili, M.; Lazar, S.; Guillaumet, G. A Straightforward Approach to Substituted 2-(Hydroxymethyl)-2,3-dihydrofuro[2,3-b]pyridines and 3-Hydroxy-3,4-dihydro-2H-pyrano[2,3-b]pyridines. *Synthesis* **2010**, 1349. (k) Ma, Z.; Ni, F.; Woo, G. H. C.; Lo, S. M.; Roveto, P. M.; Schaus, S. E.; Snyder, J. K. An intramolecular inverse electron demand Diels–Alder approach to annulated alpha-carbolines. *Beilstein J Org Chem* **2012**, 8, 829. (l) Dinh Ngoc, T.; Moons, N.; Kim, Y.; De Borggraeve, W.; Mashentseva, A.; Andrei, G.; Snoeck, R.; Balzarini, J.; Dehaen, W. Synthesis of triterpenoid triazine derivatives from allobetulone and betulonic acid with biological activities. *Bioorg. Med. Chem.* **2014**, 22, 3292. (m) Bachollet, S. P. J. T.; Vivat, J. F.; Cocker, D. C.; Adams, H.; Harrity, J. P. A. Development of a Mild and Versatile Directed Cycloaddition Approach to Pyridines. *Chem.-Eur. J.* **2014**, 20, 12889. (n) Bachollet, S. P. J. T.; Volz, D.; Fiser, B.; Münch, S.; Röncke, F.; Carrillo, J.; Adams, H.; Schepers, U.; Gómez-Bengoa, E.; Bräse, S.; Harrity, J. P. A. A Modular Class of Fluorescent Difluoroboranes: Synthesis, Structure, Optical Properties, Theoretical Calculations and Applications for Biological Imaging. *Chem.-Eur. J.* **2016**, 22, 12430.

(11) For a recent alternative approach to fused-pyridines, see: Xie, L.-G.; Shaaban, S.; Chen, X.; Maulide, N. Metal-Free Synthesis of Highly Substituted Pyridines by Formal [2+2+2] Cycloaddition under Mild Conditions. *Angew. Chem. Int. Ed.* **2016**, 55, 12864 and references cited therein.

(12) For representative examples, see: (a) Shah, U.; Lankin, C. M.; Boyle, C. D.; Chackalamannil, S.; Greenlee, W. J.; Neustadt, B. R.; Cohen-Williams, M. E.; Higgins, G. A.; Ng, K.; Varty, G. B.; Zhang, H.; Lachowicz, J. E. Design, synthesis, and evaluation of fused heterocyclic analogs of SCH 58261 as adenosine A2A receptor antagonists. *Bioorg. Med. Chem. Lett.* **2008**, 18, 4204. (b) Kono, M.; Ochida, A.; Oda, T.; Imada, T.; Banno, Y.; Taya, N.; Masada, S.; Kawamoto, T.; Yonemori, K.; Nara, Y.; Fukase, Y.; Yukawa, T.; Tokuhara, H.; Skene, R.; Sang, B.-C.; Hoffman, I. D.; Snell, G. P.; Uga, K.; Shibata, A.; Igaki, K.; Nakamura, Y.; Nakagawa, H.; Tsuchimori, N.; Yamasaki, M.; Shirai, J.; Yamamoto, S. Discovery of [cis-3-((5R)-5-[(7-Fluoro-1,1-dimethyl-2,3-dihydro-1H-inden-5-yl)carbamoyl]-2-methoxy-7,8-dihydro-1,6-naphthyridin-6(5H)-yl)carbonyl]cyclobutyl]acetic Acid (TAK-828F) as a Potent, Selective, and Orally Available Novel Retinoic Acid Receptor-Related Orphan Receptor γ t Inverse Agonist. *J. Med. Chem.* **2018**, 61, 2973. (c) Johnson, R. J.; O'Mahony, D. J. R.; Edwards, W. T.; Duncton, M. A. J. A concise one-pot synthesis of trifluoromethyl-containing 2,6-disubstituted 5,6,7,8-tetrahydroquinolines and 5,6,7,8-tetrahydronaphthyridines. *Org. Biomol. Chem.* **2013**, 11, 1358. (d) Jackl, M. K.; Kreituss, I.; Bode, J. W. Synthesis of Tetrahydronaphthyridines from Aldehydes and HARP Reagents via Radical Pictet–Spengler Reactions. *Org. Lett.* **2016**, 18, 1713.

(13) (a) Tahirovic, Y. A.; Truax, V. M.; Wilson, R. J.; Jecs, E.; Nguyen, H. H.; Miller, E. J.; Kim, M. B.; Kuo, K. M.; Wang, T.; Sum, C. S.; Cvijic, M. E.; Schroeder, G. M.; Wilson, L. J.; Liotta, D. C. Discovery of N-Alkyl Piperazine Side Chain Based CXCR4 Antagonists with Improved Drug-like Properties. *ACS Med. Chem. Lett.* **2018**, 9, 446. (b) Yoshiizumi, K.; Yamamoto, M.; Miyasaka, T.; Ito, Y.; Kumihara, H.; Sawa, M.; Kiyoi, T.; Yamamoto, T.; Nakajima, F.; Hirayama, R.; Kondo, H.; Ishibushi, E.; Ohmoto, H.; Inoue, Y.; Yoshino, K. Synthesis and structure–Activity relationships of 5,6,7,8-Tetrahydropyrido[3,4-b]pyrazine-based hydroxamic acids as HB-EGF shedding inhibitors. *Bioorg. Med. Chem.* **2003**, 11, 433. (c) Luckhurst, C. A.; Stein, L. A.; Furber, M.; Webb, N.; Ratcliffe, M. J.;

Allenby, G.; Botterell, S.; Tomlinson, W.; Martin, B.; Walding, A. Discovery of isoindoline and tetrahydroisoquinoline derivatives as potent, selective PPAR δ agonists. *Bioorg. Med. Chem. Lett.* **2011**, *21*, 492. (d) For representative examples, see: Shiozawa, A.; Ichikawa, Y.; Komuro, C.; Kurashige, S.; Miyazaki, H.; Yamanaka, H.; Sakamoto, T. Antivertigo Agents. III. Synthesis of 5, 6, 7, 8-Tetrahydro-1, 6-naphthyridine Methyl Homologs. *Chem. Pharm. Bull.* **1984**, *32*, 2522. (e) Cruz, D.; Wang, Z.; Kibbie, J.; Modlin, R.; Kwon, O. Diversity through phosphine catalysis identifies octahydro-1,6-naphthyridin-4-ones as activators of endothelium-driven immunity. *Proc. Natl. Acad. Sci.* **2011**, *108*, 6769. (f) Harling, J. D.; Harrington, F. P.; Thompson, M. A facile synthesis of the 3-amino-5,6,7,8-tetrahydro[1,6]naphthyridine system and some alkylated and polycyclic homologues. *Synth. Commun.* **2001**, *31*, 787. (g) Vanlaer, S.; Voet, A.; Gielens, C.; De Maeyer, M.; Compennolle, F. Bridged 5,6,7,8-Tetrahydro-1,6-naphthyridines, Analogues of Huperzine A: Synthesis, Modelling Studies and Evaluation as Inhibitors of Acetylcholinesterase. *Eur. J. Org. Chem.* **2009**, *2009*, 643.

(14) For leading examples of organocatalyzed conjugated addition reactions to alkenylazaarene, see: (a) Wang, S.; Li, X.; Liu, H.; Xu, L.; Zhuang, J.; Li, J.; Li, H.; Wang, W. Organocatalytic Enantioselective Direct Additions of Aldehydes to 4-Vinylpyridines and Electron-Deficient Vinylarenes and Their Synthetic Applications. *J. Am. Chem. Soc.* **2015**, *137*, 2303. (b) Hepburn, H. B.; Melchiorre, P. Brønsted acid-catalysed conjugate addition of photochemically generated [small alpha]-amino radicals to alkenylpyridines. *Chem. Commun.* **2016**, *52*, 3520. (c) Wang, Y.-Y.; Kanomata, K.; Korenaga, T.; Terada, M. Enantioselective Aza Michael-Type Addition to Alkenyl Benzimidazoles Catalyzed by a Chiral Phosphoric Acid. *Angew. Chem. Int. Ed.* **2016**, *55*, 927. (d) Yin, Y.; Dai, Y.; Jia, H.; Li, J.; Bu, L.; Qiao, B.; Zhao, X.; Jiang, Z. Conjugate Addition–Enantioselective Protonation of N-Aryl Glycines to α -Branched 2-Vinylazaarenes via Cooperative Photoredox and Asymmetric Catalysis. *J. Am. Chem. Soc.* **2018**, *140*, 6083.

(15) For the alternative use of styrylisoxazoles as alkenylazaarene homologues, see: (a) Baschieri, A.; Bernardi, L.; Ricci, A.; Suresh, S.; Adamo, M. F. A. Catalytic Asymmetric Conjugate Addition of Nitroalkanes to 4-Nitro-5-styrylisoxazoles. *Angew. Chem. Int. Ed.* **2009**, *48*, 9342. (b) Sun, H.-W.; Liao, Y.-H.; Wu, Z.-J.; Wang, H.-Y.; Zhang, X.-M.; Yuan, W.-C. Enantioselective 1,6-Michael addition of anthrone to 3-methyl-4-nitro-5-alkenyl-isoxazoles catalyzed by bifunctional thiourea-tertiary amines. *Tetrahedron* **2011**, *67*, 3991. (c) Pei, Q.-L.; Sun, H.-W.; Wu, Z.-J.; Du, X.-L.; Zhang, X.-M.; Yuan, W.-C. Catalytic Asymmetric 1,6-Michael Addition of Arylthiols to 3-Methyl-4-nitro-5-alkenyl-isoxazoles with Bifunctional Catalysts. *J. Org. Chem.* **2011**, *76*, 7849.

(16) For a recent catalytic sulfa-Michael addition to alkenyl benzimidazoles, see: Formica, M.; Sorin, G.; Farley, A. J. M.; Diaz, J.; Paton, R. S.; Dixon, D. J. Bifunctional iminophosphorane catalysed enantioselective sulfa-Michael addition of alkyl thiols to alkenyl benzimidazoles. *Chem Sci* **2018**, *9*, 6969.

(17) (a) Kennedy, S. H.; Klumpp, D. A. Michael Additions Involving Amino Acid Esters with Alkenyl N-Heterocycles. *J. Org. Chem.* **2017**, *82*, 10219. (b) For an insightful pioneer investigation of aza-Michael reaction to vinylpyridine mentioning the use of metal or base/acid promoters, see: Reich, H. E.; Levine, R. The Pyridylethylation of Active Hydrogen Compounds. IV. The Acid-catalyzed Pyridylethylation of Primary Amines. *J. Am. Chem. Soc.* **1955**, *77*, 5434.

(18) (a) Han, X.; Civiello, R. L.; Mercer, S. E.; Macor, J. E.; Dubowchik, G. M. Syntheses of aza and fluorine-substituted 3-(piperidin-4-yl)-4,5-dihydro-1H-benzo[d][1,3]diazepin-2(3H)-ones. *Tetrahedron Lett.* **2009**, *50*, 386. (b) Boy, K. M.; Guernon, J. M. Synthesis of 2-aminoethyl-5-carbomethoxythiazoles utilizing a Michael-like addition

strategy. *Tetrahedron Lett.* **2005**, 46, 2251. (c) For representative examples of aza-Michael reaction to azaarenes, see: Cliffe, I. A.; Ifill, A. D.; Mansell, H. L.; Todd, R. S.; White, A. C. 7-amino-5,6,7,8-tetrahydroquinolines. Preparation from 5,6-dihydroquinoline and nitrogen nucleophiles. *Tetrahedron Lett.* **1991**, 32, 6789. (d) Macor, J. E.; Ordway, T.; Smith, R. L.; Verhoest, P. R.; Mack, R. A. Synthesis and Use of 5-Vinyl-1,2,4-oxadiazoles as Michael Acceptors. A Rapid Synthesis of the Potent Muscarinic Agonist L-670,548. *J. Org. Chem.* **1996**, 61, 3228.

(19) Xu, C.; Muir, C. W.; Leach, A. G.; Kennedy, A. R.; Watson, A. J. B. Catalytic Enantioselective Synthesis of α -Chiral Azaheteroaryl Ethylamines by Asymmetric Protonation. *Angew. Chem. Int. Ed.* **2018**, 57, 11374.

(20) (a) Hui, C.; Pu, F.; Xu, J. Metal-Catalyzed Asymmetric Michael Addition in Natural Product Synthesis. *Chem.-Eur. J.* **2017**, 23, 4023. (b) Amara, Z.; Caron, J.; Joseph, D. Recent contributions from the asymmetric aza-Michael reaction to alkaloids total synthesis. *Nat. Prod. Rep.* **2013**, 30, 1211. (c) Howell, G. P. Asymmetric and Diastereoselective Conjugate Addition Reactions: C–C Bond Formation at Large Scale. *Org. Proc. Res. Dev.* **2012**, 16, 1258. (d) For selected reviews, see: Zhang, Y.; Wang, W. Recent advances in organocatalytic asymmetric Michael reactions. *Catal. Sci. Technol.* **2012**, 2, 42.

(21) Zhao, W.; Sun, J. Triflimide (HNTf₂) in Organic Synthesis. *Chem. Rev.* **2018**, 118, 10349.

(22) As first hypothesis, from either a thermodynamic point of view or for the ability to activate the vinyl moiety, the protonation of N² of the vinyltriazines might be favored: M^o, O.; De Paz, J. L. G.; Y^añez, M. Protonation of azines: An ab initio molecular orbital study. *Journal of Molecular Structure: THEOCHEM* **1987**, 150, 135.

(23) Liebeskind, L. S.; Srogl, J. Thiol Ester–Boronic Acid Coupling. A Mechanistically Unprecedented and General Ketone Synthesis. *J. Am. Chem. Soc.* **2000**, 122, 11260.

(24) (a) Alphonse, F.-A.; Suzenet, F.; Keromnes, A.; Lebre, B.; Guillaumet, G. Palladium-catalyzed 3-Thiomethyltriazine-boronic Acid Cross Coupling: Easy Access to 3-Substituted-1,2,4-triazines. *Synlett* **2002**, 447. (b) Alphonse, F.-A.; Suzenet, F.; Keromnes, A.; Lebre, B.; Guillaumet, G. Copper(I)-Promoted Palladium-Catalyzed Cross-Coupling of Unsaturated Tri-n-butylstannane with Heteroaromatic Thioether. *Org. Lett.* **2003**, 5, 803.

(25) Revuelta, J.; Cicchi, S.; Brandi, A. Samarium(II) iodide reduction of isoxazolidines. *Tetrahedron Lett.* **2004**, 45, 8375.

(26) Shi, D.-H.; Harjani, J. R.; Gable, R. W.; Baell, J. B. Synthesis of 3-(Alkylamino)-, 3-(Alkoxy)-, 3-(Aryloxy)-, 3-(Alkylthio)-, and 3-(Arylthio)-1,2,4-triazines by Using a Unified Route with 3-(Methylsulfonyl)-1,2,4-triazine. *Eur. J. Org. Chem.* **2016**, 2016, 2842.

(27) Gantt, R. W.; Goff, R. D.; Williams, G. J.; Thorson, J. S. Probing the Aglycon Promiscuity of an Engineered Glycosyltransferase. *Angew. Chem. Int. Ed.* **2008**, 47, 8889.

(28) Shrikhande, J. J.; Gawande, M. B.; Jayaram, R. V. A catalyst-free N-benzyloxycarbonylation of amines in aqueous micellar media at room temperature. *Tetrahedron Lett.* **2008**, 49, 4799.

(29) Gros, G.; Martinez, L.; Gimenez, A. S.; Adler, P.; Maurin, P.; Wolkowicz, R.; Falson, P.; Hasserodt, J. Modular construction of quaternary hemiaminal-based inhibitor candidates and their in cellulo assessment with HIV-1 protease. *Bioorg. Med. Chem.* **2013**, 21, 5407.

(30) Wang, Y. D.; Kimball, G.; Prashad, A. S.; Wang, Y. Zr-Mediated hydroboration: stereoselective synthesis of vinyl boronic esters. *Tetrahedron Lett.* **2005**, 46, 8777.

(31) Miyabe, H.; Asada, R.; Takemoto, Y. Cascade radical reaction of substrates with a carbon–carbon triple bond as a radical acceptor. *Beilstein J. Org. Chem.* **2013**, 9, 1148.

(32) Cross, B. P.; Schleich, T. Temperature dependence of the chemical shifts of commonly employed proton n.m.r. reference compounds. *Org. Magn. Reson.* **1977**, *10*, 82.