Contents lists available at ScienceDirect

Tetrahedron Letters

journal homepage: www.elsevier.com/locate/tetlet



Synthesis of 1-deoxy-L-gulonojirimycin and 1-deoxy-L-talonojirimycin

Annalisa Guaragna, Daniele D'Alonzo*, Concetta Paolella, Giovanni Palumbo

Dipartimento di Chimica Organica e Biochimica, Università di Napoli Federico II, via Cinthia, 4 I-80126 Napoli, Italy

ARTICLE INFO

Article history: Received 16 December 2008 Revised 5 February 2009 Accepted 10 February 2009 Available online 21 February 2009

Keywords: L-Iminosugars Polyhydroxylated piperidines Glycosidase inhibitors Dihydroxylation

ABSTRACT

De novo synthesis of noncompetitive glycosidase inhibitors L-gulo-DNJ and L-talo-DNJ has been achieved in 9-10 steps starting from Garner's aldehyde. Key to the success of this procedure was the construction of the 2,3-unsaturated piperidine 14, which syn dihydroxylation under Kishi's and Donohoe's conditions led to the desired iminosugars.

© 2009 Elsevier Ltd. All rights reserved.

Iminosugars (sugar analogues having the endocyclic oxygen replaced with nitrogen) undoubtedly represent one of the most attractive classes of carbohydrate mimetics.¹ The great deal of attention developed around iminosugars lies in their powerful inhibitory aptitude towards carbohydrate processing enzymes, that is, glycosidases² and glycosyltransferases.³ As these enzymes are involved in a plethora of key biochemical events, such as digestion, lysosomal catabolism of glycoconjugates and post-translational glycoprotein processing, the significant inhibitory properties of iminosugars, such as deoxynojirimycin (DNJ, 1, Fig. 1) and its derivatives, make them excellent candidates for medical intervention, ranging from antidiabetics⁴ and antivirals⁵ to agents devoted to the treatment of genetic disorders.⁶ In search for new, more efficient and selective inhibitors, L-iminosugars currently represent a significant breakthrough,⁷ especially regarding glycosidase inhibition. Deep interference by L-iminosugars has been found against L-glycosidases⁸ (i.e., fucosidases and rhamnosidases), this behaviour being related to their structural similarity with the natural substrates (L-fucose and L-rhamnose) of the corresponding enzymes. Remarkably, activity has also been extended to glycosidases belonging to p-series, often displaying considerably selective as well as potent inhibition.⁹ Recent studies into the action mechanism of p-glycosidase inhibition have revealed that some L-iminosugars, especially pyrrolidines, are able to mimic the conformation of natural D-hexose substrates, by virtue of their high structural flexibility.¹⁰ On the other hand, activity of the more rigid L-piperidines has been justified invoking a noncompetitive mode of action.^{7,11} Driven by the intriguing therapeutic potential

E-mail address: dandalonzo@unina.it (D. D'Alonzo).

of such molecules, considerable efforts have been devoted to the synthesis of L-DNJ (ent-1) (Fig. 1) and its congeners.

In spite of the great amount of routes leading to one or some Lpiperidines, just a few of them can claim to be applied to the construction of most L-epimers.¹² In this context, in a previous report we had developed a general procedure¹³ for the synthesis of 1deoxy-L-iminopyranoses by a non-carbohydrate-based route; as proof of it, iminosugars belonging to L-manno-, L-altro- and L-alloconfiguration (2-4, Fig. 2) were prepared in high yields and stereoselective fashion. In order to widen this strategy, access to L-guloDNJ (5) and L-taloDNJ (6) (Fig. 2) has been examined in this Letter.

We began our synthesis with the coupling reaction of enol thioether¹⁴ **7** with the Garner's aldehyde **8** in the presence of BuLi at -78 °C (Scheme 1), to achieve a mixture of diastereomers **9** (*syn*/ anti, dr = 1:9) in good overall yield (72%). As already noticed,¹³ a preference for the *anti*-adduct was found in anhydrous Et₂O, as a consequence of the poorly ionised nature of the organolithium intermediate.¹⁵ On the other hand, no reversal stereoselection was observed in other solvents (such as THF) or after addition of several chelating¹⁶ catalysts [ZnBr₂,Ti(O-*i*-Pr)₄, Cp₂TiCl₂]. Thus, in

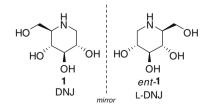


Figure 1. Deoxynojirimycin (1) and its enantiomer L-DNJ (2).

^{*} Corresponding author. Tel./fax: +39 081 674119.

^{0040-4039/\$ -} see front matter © 2009 Elsevier Ltd. All rights reserved. doi:10.1016/j.tetlet.2009.02.111

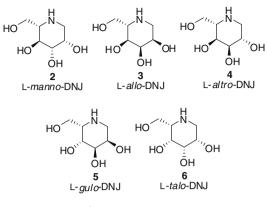
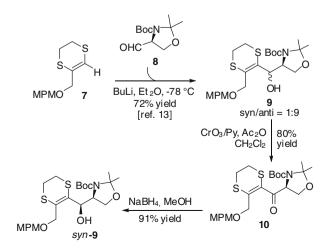


Figure 2. L-Iminosugars.

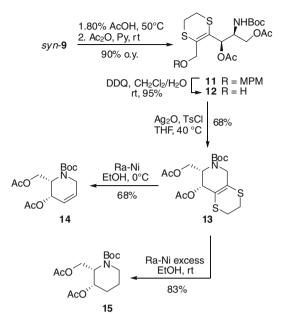
order to selectively obtain the adduct *syn*-**9** (which is the suitable precursor for L-gulo and L-taloDNJ synthesis), an oxidation/reduction procedure¹⁷ was preferred. *sec*-Alcohols **9** were treated with in situ-generated PDC and Ac₂O in CH₂Cl₂ at room temperature, affording ketone **10** (Scheme 1). Then, reduction of **10** with sodium borohydride proceeded with full stereoselectivity, giving alcohol *syn*-**9** as the only diastereomer (73% yield over two steps).

Next, alcohol *syn*-**9** was converted into its diacetate **11** by oxazolidine ring opening (80% aq AcOH) and subsequent acetylation (90% o.y.) (Scheme 2). MPM group removal (DDQ) of **11** furnished the primary alcohol **12** in very good yield (95%). Tosylation of primary hydroxyl group (Ag₂O/TsCl) followed by in situ intramolecular attack on tosylate intermediate¹⁸ by the amino group gave piperidine **13** (68% yield). Finally, removal of dithioethylene bridge of **13** (Raney-Ni) led to olefin **14** (68% yield). As previously reported for similar substrates,^{13,19} when the desulfurisation reaction was carried out with a Raney-Ni excess (or for prolonged reaction times), the over-reduction product was obtained in a satisfying 83% yield, affording the 1,2,3-trideoxy iminosugar derivative **15** (Scheme 2).

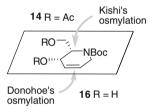
With the key unsaturated piperidine **14** in our hand, access to desired iminosugars was planned by an appropriate choice of the conditions for double-bond *syn* dihydroxylation (Scheme 3). Particularly, osmylation under Kishi's conditions²⁰ (OsO₄/NMO) (which is driven by steric factors) is expected to occur from the less hindered face of the olefin,²¹ leading to the iminosugar with L-gulo-configuration. On the other hand, since access to L-taloDNJ is hampered by the difficulty to carry out osmylation from the same side of the allylic acetoxy group, synthesis of the latter was envisaged,



Scheme 1. Synthesis of sec-alcohol syn-9.



Scheme 2. Preparation of unsaturated L-piperidine 14.

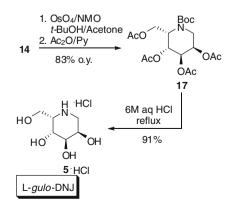


Scheme 3. Prevision of the stereoselective outcome of dihydroxylation reaction under Kishi's or Donohoe's conditions.

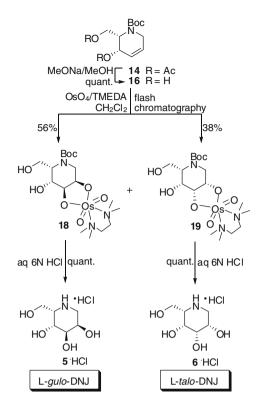
starting from the deprotected allylic alcohol **16**, using the Donohoe's conditions²² (OsO₄/TMEDA), which usually take place as a consequence of the hydrogen bond formation between the allylic OH group of **16** and the incoming OsO₄ reagent (Scheme 3).

Thus, treatment of the diacetate **14** with OsO_4/NMO in *t*-BuOH/ acetone followed by acetylation of the crude residue afforded exclusively the L-guloDNJ derivative **17**. Exposure of **17** to refluxing aq 6 N HCl furnished the pure 1-deoxy-L-gulonojirimycin (**5**)²³ in an excellent 91% yield (Scheme 4).

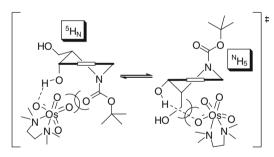
Subsequently, acetyl groups of olefin **14** were removed under common Zemplén conditions, affording diol **16**. The latter was then treated with stoichiometric OsO_4 and TMEDA in CH_2Cl_2 at -78 °C,



Scheme 4. Osmylation of 14 under Kishi's conditions.



Scheme 5. Osmylation of 14 under Donohoe's conditions.



Scheme 6. Proposed ^NH₅ and ⁵H_N conformers for olefin 16.

affording a mixture of L-gulo- and L-taloDNJ derivatives **18** and **19** (dr = 6:4), which can be separated by flash chromatography (CHCl₃/MeOH, 8:2). Hydrolysis of osmate esters **18** and **19** along with removal of *N*-Boc protection using aq 6 N HCl²⁴ furnished deoxy-L-gulonojirimycin (**5**) and deoxy-L-talonojirimycin (**6**)²⁵ in a very good 93% overall yield (Scheme 5).

In our opinion, the low level of selectivity observed above could be due to the difficulty in the access of the incoming $OsO_4/TMEDA$ complex to the *syn* face of allylic alcohol **16**. Indeed, even though NMR data suggest that olefin **16** exists as a mixture of conformers (presumably corresponding to $^{\rm N}H_5$ and $^{5}H_{\rm N}$),²⁶ it can be conjectured that entry of OsO₄/TMEDA is hampered by the presence of *t*-butoxycarbonyl group in the ${}^{5}H_{N}$ conformation, and by the axially oriented *C*-6 methylene group when **16** adopts the ${}^{N}H_{5}$ conformation (Scheme 6).

In summary, in this Letter, we have outlined a synthetic path for the preparation of non-naturally occurring L-guloDNJ (**5**) and L-taloDNJ (**6**). Studies aimed to obtain the remaining deoxyiminopyranoses belonging to L-series by *anti*-dihydroxylation reactions of olefins **14** and **16** are ongoing and will be published in due course.

References and notes

- Compain, P.; Martin, O. R. Iminosugars—From Synthesis to Therapeutic Applications; John Wiley & Sons Ltd: West Sussex England, 2007.
- Lillelund, V. H.; Jensen, H. H.; Liang, X.; Bols, M. Chem. Rev. 2002, 102, 515–553.
 (a) Compain, P.; Martin, O. R. Bioorg. Med. Chem. 2001, 9, 3077–3092; (b) Compain, P.; Martin, O. R. Curr. Top. Med. Chem. 2003, 3, 541–560.
- 4. Asano, N. *Glycobiology* **2003**, 13, 93R–104R.
- Pavlovic, D.; Neville, D. C.; Argaud, O.; Blumberg, B.; Dwek, R. A.; Fischer, W. B.; Zitzmann, N. Proc. Natl. Acad. Sci. U.S.A. 2003, 100, 6104–6108.
- (a) Butters, T. D.; Dwek, R. A.; Platt, F. M. Glycobiology 2005, 15, 43R–52R; (b) Butters, T. D. Curr. Opin. Chem. Biol. 2007, 11, 412–418.
- 7. D'Alonzo, D.; Guaragna, A.; Palumbo, G. Curr. Med. Chem. 2009, 16, 473-505.
- (a) Wu, C.-Y.; Chang, C.-F.; Chen, J. S.-Y.; Wong, C.-H.; Lin, C.-H. Angew. Chem., Int. Ed. 2003, 42, 4661–4664; (b) Chang, C.-F.; Ho, C.-W.; Wu, C.-Y.; Chao, T.-A.; Wong, C.-H.; Lin, C.-H. Chem. Biol. 2004, 11, 1301–1306.
- See, for example: Yu, C.-Y.; Asano, N.; Ikeda, K.; Wang, M.-X.; Butters, T. D.; Wormald, M. R.; Dwek, R. A.; Winters, A. L.; Nash, R. J.; Fleet, G. W. J. Chem. Commun. 2004, 1936–1937.
- Carmona, A. T.; Popowycz, F.; Gerber-Lemaire, S.; Rodríguez-García, E.; Schütz, C.; Vogel, P.; Robina, I. *Bioorg. Med. Chem.* 2003, *11*, 4897–4911.
- Asano, N.; Ikeda, K.; Yu, L.; Kato, A.; Takebayashi, K.; Adachi, I.; Kato, I.; Ouchi, H.; Takahata, H.; Fleet, G. W. J. *Tetrahedron: Asymmetry* **2005**, *16*, 223–229.
- 12. Kato, A.; Kato, N.; Kano, E.; Adachi, I.; Ikeda, K.; Yu, L.; Okamoto, T.; Banba, Y.; Ouchi, H.; Takahata, H.; Asano, N. *J. Med. Chem.* **2005**, *48*, 2036–2044.
- 13. Guaragna, A.; D'Errico, S.; D'Alonzo, D.; Pedatella, S.; Palumbo, G. Org. Lett. **2007**, 9, 3473–3476.
- Guaragna, A.; Pedatella, S.; Palumbo, G. In *e-Encyclopedia of Reagents for Organic Synthesis (e-EROS)*; Paquette, L. A., Ed.; John Wiley & Sons: New York, US, 2008.
- 15. Maercker, A.; Roberts, J. D. J. Am. Chem. Soc. **1966**, 88, 1742–1759.
- Liang, X.; Andersch, J.; Bols, M. J. Chem. Soc., Perkin Trans. 1 2001, 2136–2157.
 Okamoto, N.; Hara, O.; Makino, K.; Hamada, Y. J. Org. Chem. 2002, 67, 9210– 9215.
- The presence of tosylate intermediate has been ascertained as it can be easily isolated by common chromatographic purification techniques.
- D'Alonzo, D.; Guaragna, A.; Napolitano, C.; Palumbo, G. J. Org. Chem. 2008, 73, 5636–5639.
- 20. Cha, J. K.; No-Soo, K. Chem. Rev. 1995, 95, 1761-1795.
- See, for example: Guaragna, A.; Napolitano, C.; D'Alonzo, D.; Pedatella, S.; Palumbo, G. Org. Lett. 2006, 8, 4863–4866.
- 22. Donohoe, T. J. Synlett 2002, 1223-1232.
- 23. Data for compound **5**·HCl: $[\alpha]_{\rm D} 2.5$ (c 0.5 MeOH); ¹H NMR (500 MHz, D₂O) δ 3.13 (t, *J* = 12.2 Hz, 1H), 3.31 (dd, *J* = 4.8, 12.2 Hz, 1H), 3.55-3.59 (ddd, *J* = 1.5, 4.4, 9.3 Hz, 1H), 3.82 (dd, *J* = 9.3, 12.2 Hz, 1H), 3.91 (dd, *J* = 4.4, 12.2 Hz, 1H), 4.07 (dd, *J* = 3.0, 4.8 Hz, 1H), 4.16 (dd, *J* = 1.5, 4.8 Hz, 1H), 4.26 (ddd, *J* = 3.0, 4.9, 11.7 Hz, 1H). ¹³C NMR (125 MHz, D₂O) δ 42.4, 55.5, 59.0, 62.6, 67.2, 68.5. Anal. Calcd for C₆H₁₃NO₄: C, 44.16; H, 8.03; N, 8.58. Found: C, 43.89; H, 7.87; N, 8.70.
- Donohoe, T. J.; Blades, K.; Moore, P. R.; Waring, M. J.; Winter, J. J. G.; Helliwell, M.; Newcombe, N. J.; Stemp, G. J. Org. Chem. 2002, 67, 7946–7956.
- 25. Data for compound **6**+*I*C1: $[\alpha]_D$ +22.0 (c 0.5 MeOH); ¹H NMR (400 MHz, D₂O) δ 3.10 (dd, *J* = 1.6, 13.6 Hz, 1H), 3.24 (dt, *J* = 1.8, 6.7 Hz, 1H), 3.35 (dd, *J* = 2.8, 13.6 Hz, 1H), 3.65–3.74 (m, 3H), 3.99–4.04 (m, 1H), 4.02–4.08 (m, 1H), ¹³C NMR (100 MHz, D₂O) δ 50.4, 61.1, 62.5, 68.7, 69.2, 69.9. Anal. Calcd for C₆H₁₃NO₄: C, 44.16; H, 8.03; N, 8.58. Found: C, 44.45; H, 8.10; N, 8.39.
- 26. ¹H NMR (500 MHz, CD₃OD) δ 1.47 (s, 4.5H), 1.48 (s, 4.5H), 3.47–3.60 (m, 1H), 3.51 (br t, *J* = 10.8 Hz, 1H), 3.81 (br dd, *J* = 11.2, 3.3 Hz, 1H), 4.09–4.12 (m, 0.5H), 4.15–4.18 (m, 0.5H), 4.40 (br s, 1H), 4.54 (br s, 1H), 5.63 (br d, *J* = 10.2 Hz, 1H), 5.64–5.75 (m, 1H).