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Stereoselective synthesis of C12–C21 common fragment of thermolides 1–5

Vavilapalli Satyanarayana, Gavireddy Chaithanya Kumar, Katta Muralikrishna, Jhillu Singh Yadav*

Centre for Semiochemicals, CSIR-Indian Institute of Chemical Technology, Hyderabad 500007, India

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Introduction

Thermolides are PKS-NRPS hybrid metabolites, which were isolated in 2012 by Niu et al., from thermophilic fungus *Thermomyces thermophilus* YM 3–4 (1–6) collected from tengchong hotspring of Yunnan, China [1,2] (Fig. 1). Thermolides **1** and **2** exhibits potent inhibitory activity against a wide range of nematodes including economically destructive rootknot nematode (*Meloidogyne incognita*) and pine-wood nematode (*Bursaphelenchus xylophilus*) [3]. Compounds **3** and **4** also show a moderate nematicidal activity against above two nematodes.

From structural side, all natural products containing a common 13-membered lactam-bearing macrolactone. Thermolides 1-5 have a common C₉ polypropionate side chain with three methyl groups and three hydroxyls, in which two hydroxyls at C16 and C18 are *syn* to each other and third hydroxyl at C20 is *anti*-relation-ship with C16 and C18. These unusual structural features, twelve stereogenic centers, sensitive functionalities and nematicidal activities make thermolides 1-5 attractive for synthesis [1,2].

For the last two decades, our group has explored the potentiality of desymmetrization strategy for the synthesis of biologically active natural and designed molecules [4]. This strategy offers the generation of contiguous chiral centers from a single bicyclic precursor by using desymmetrization strategy.

Herein, we describe stereoselective synthesis of **C12–C21** common fragment of thermolides **1–5** by using desymmetrization

* Corresponding author. *E-mail address:* yadavpub@gmail.com (J. Singh Yadav).

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ABSTRACT

A highly stereoselective synthesis of **C12–C21** common fragment of thermolides **1–5** has been described. The salient features of the synthesis are the utilization of desymmetrization protocol, Barton-McCombie reaction, Brown's asymmetric allylation and Wacker oxidation.

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Fig. 1. Structures of thermolides 1–5.

protocol, Brown's asymmetric allylation and Wacker oxidation as key reactions.

The retrosynthetic analysis of **C12–C21** fragment of thermolides **1–5** is outlined in Scheme 1. We envisioned that fragment **2** could be synthesized from Brown's allylated compound **3** by using Wacker oxidation and anti-reduction. The allyl compound **3** could be obtained from triol **4**, which in turn could be easily synthesized from known bicyclic olefine **5** [5].

Results and discussion

Our synthesis was commenced from reductive cleavage of bicyclic lactone **6**, which was synthesized by desymmetrization



Scheme 1. Retrosynthetic analysis of thermolides.

of the bicyclic olefine **5**. Compound **6** was treated with LiAlH_4 in dry THF to afford the triol **4** in 90% yield [6] (Scheme 2).



The 1,3-diol group of compound **4** was protected as an acetonide [6] using 2,2-dimethoxypropane (2,2-DMP) and catalytic amount of CSA in CH₂Cl₂ and the free primary hydroxyl group of acetonide was protected as its TBS ether to furnish **7**. The TBS ether **7** was treated with CuCl₂·2H₂O in acetonitrile followed by the selective silylation of the primary hydroxy group as its TBDPS ether **8** was achieved in 90% yield under the classical conditions (TBDPSCl, imidazole in CH₂Cl₂). The free secondary hydroxyl group of **8** was converted into the corresponding xanthate ester **9** using LiHMDS, CS₂ and methyl iodide at -78 °C with 86% yield [6a,7]. Later the compound **9** was deoxygenated under Barton-McCombie conditions using tri-*n*-butyltin hydride and catalytic amount of



Scheme 3. Synthesis of C12-C21 fragment 2.

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AIBN as radical initiator, followed by selective deprotection of TBS group by using CSA in CH₂Cl₂/MeOH (3:1) to afford compound **10** in 92% yield (Scheme 2).

The compound **10** was oxidized to aldehyde using Dess-Martin periodinane in CH₂Cl₂, which on subsequent asymmetric allylboration using Brown's conditions [8] to afford the homoallylic alcohol **3** with desired (*R*)-configuration in a diastereomeric ratio of 95:5. This asymmetric reaction fixed the chiral center at C-18 of the target molecule (Scheme 3). The allyl compound 3 was subjected to Wacker oxidation [8a,9] by using oxygen, PdCl₂ and CuCl in DMF and water to afford methyl ketone 11 in 70% yield. The methyl ketone **11** was converted into *anti* diol **12** by the treatment with tetramethylammonium triacetoxyborohydride [10] in CH₃CN: AcOH (1:1) at -40 °C in an 85% yield (96:4 dr). The 1,3-anti diol was protected as its benzylidene acetal [11] with benzaldehyde dimethyl acetal in the presence of a catalytic amount of CSA to afford compound 2 in 89% vield (Scheme 3). The synthesis of C12-C21 fragment of thermolides 1-5 involved 13 steps starting from *exo*-methylated bicyclic lactone **6** with a 20.1% overall yield. Further efforts towards the completion of the total synthesis of thermolides 1-5 are currently underway.

Conclusion

In conclusion, a concise and highly stereoselective approach for **C12–C21** fragment of thermolides **1–5** were achieved by employing desymmetrization strategy, Barton-McCombie reaction, Brown's asymmetric allylation, Wacker oxidation and *anti*-reduction as key steps. The synthesis involved 13 steps starting from bicylic lactone **6** with a 20.1% overall yield. Further work towards the total synthesis of thermolides **1–5** is ongoing.

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References

- 1. Guo Ji-Peng, Zhang Chu-Ping, Chu Yan-Sheng, et al. J. Am. Chem. Soc.. 2012;134:20306-20309.
- 2. Niu Xuemei, Chen Li, Yue Qun, et al. Org. Lett.. 2014;16:3744–3747.
- Taniwiryonoc Wiratno D, Van den Bergb H, Riksen JAG, et al. Open Nat. Prod. J.. 2009;2:77–85.

 (a) Rama Rao AV, Yadav JS, Vidyasagar V. J. Chem. Soc., Chem. Commun., 1985;55;

(b) Yadav JS, Srinivas Rao C, Chandrasekhar S, Rama Rao AV. *Tetrahedron Lett.*. 1995;36:7717;

- (c) Yadav JS, Rao KVR, Ravinder K, Reddy BVS. Eur. J. Org. Chem. 2011;58–61;
- (d) Yadav JS, Pratap TV, Rajender V. J. Org. Chem.. 2007;72:5882. 5. (a) Yadav JS, Sathaiah K, Srinivas R. Tetrahedron. 2009;65:3545-3552;
- (b) Yadav JS, Hossain S, Hossain SK, Madhu M, Mohapatra DK. J. Org. Chem.. 2009;74:8822-8825.
- (a) Satyanarayana V, Muralikrishna K, Kumar GC, Kumar AS, Babu NJ, Yadav JS. Chem. Select. 2018;3:1000;
- (b) Yadav JS, Gyanchander E, Saibal Das. Tetrahedron Lett.. 2014;55:3996–3998.
 7. a) Andrel T, Nicolas L, Miinkemer J, et al. Angew. Chem. Int. Ed.. 2011;123:972–975;
- b) Chandra B, Fu D, Nelson SG. Angew. Chem. Int. Ed., 2010;122:2645–2648.
 8. (a) Satyanarayna V, Kumar AS, Muralikrishna K, Kumar GC, Kumar RS, Kumar
- JS, Yadav JS. Chem. Select. 2018;3:1024;
 (b) Kuntiyong P, Akkarasamiyo S, Piboonsrinakara N, Hemmara C, Songthammawat P. Tetrahedron.. 2011;67:8034–8040;
- (c) Wender PA, Koehler MFT, Sendzik M. Org. Lett.. 2003;5:4549–4552;
- (d) Tanabe Y, Sato E, Nakajima N, Ohkubo A, Ohno O, Suenaga K. Org. Lett..
- 2014;16:2858–2861; (e) Fuwa H, Saito A, Naito S, Konoki K, Yotsu-Yamashita M, Sasaki M. *Chem. Eur.* J. 2009;15:12807–12818;
- (f) Wender PA, Horan JC, Verma VA. *Org. Lett.*. 2006;8:5299–5302;
- (g) Srinivas E, Palash DB, Ganganna Alghamdi AA, Yadav JS. Synthesis. 2016;48:1561–1567;
- (h) Nicolaou KC, Nold AL, Milburn RR, Schindler CS. Angew. Chem. Int. Ed., 2006;45:6527-6532;
- (i) Moretti JD, Wang X, Curran DP. J. Am. Chem. Soc.. 2012;134:7963-7970;
- (j) Alberto Marco J, García-Pla J, Carda M, Murga J, Falomir E, Trigili C, Notararigo S, Fernando Díaz J, Barasoain I. Eur. J. Med. Chem.. 2011;46:1630–1637.
- 9. (a) White JD, Hanselmann R, Jackson RW, et al. J. Org. Chem.. 2001;66:5217-5231;
 (b) Trost BM, O'Boyle BM. J. Am. Chem. Soc.. 2008;130:16190-16192;
 (c) Wickens ZK, Skakuj K, Morandi B, Grubbs RH. J. Am. Chem. Soc.. 2014;136:890-893;
 (d) Park PK, O'Malley SJ, Schmidt DR, Leighton JL. J. Am. Chem. Soc.. 2006;128:2796-2797;
 (e) Li D, Zhang D, Sun C, et al. Chem. Eur. J.. 2006;12:1185-1204;
 (f) Dias LC, Kuroishi PK, Polo EC, de Lucca Jr EC. Tetrahedron Lett.. 2013;54:980-982;
 (g) Shing TKM, Cheng HM. Org. Biomol. Chem.. 2015;13:4795-4802;
 - (h) Paterson I, Anderson EA, Dalby SM. Synthesis. 2005;19:3225–3228.
- (a) Sabitha G, Srinivas S, Ramamohan Reddy T, Yadagiri K, Yadav JS. *Tetrahedron: Asymmetry*. 2011;22:2124;
 (b) Holmes M, Kwon D, Taron M, Britton R. *Org. Lett.*. 2015;17:3868;
 - (c) Sunil Kumar S, Prasad K. R. Tetrahedron. 2014;70:2096;
- (d) Evans DA, Chapman KT, Carreira EM. J. Am. Chem. Soc.. 1988;110:3560.
 (a) Tao Hu, Takenaka N, James Panek S. J. Am. Chem. Soc.. 2002;124:12806;
 (b) Jed Hubbs L, Heathcock Clayton H. J. Am. Chem. Soc.. 2003;125:12836;
 (c) Kim Woo Han, Jung Jae Hoon, Sung Lee Taek, Lim Sang Min, Lee Eun. Org. Lett.. 2005;7:1085.