



Accepted Article

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This manuscript has been accepted after peer review and appears as an Accepted Article online prior to editing, proofing, and formal publication of the final Version of Record (VoR). This work is currently citable by using the Digital Object Identifier (DOI) given below. The VoR will be published online in Early View as soon as possible and may be different to this Accepted Article as a result of editing. Readers should obtain the VoR from the journal website shown below when it is published to ensure accuracy of information. The authors are responsible for the content of this Accepted Article.

To be cited as: Adv. Synth. Catal. 10.1002/adsc.202001254

Link to VoR: https://doi.org/10.1002/adsc.202001254

10.1002/adsc.202001254

FULL PAPER

DOI: 10.1002/adsc.2020######

Room Temperature Benzofused Lactam Synthesis Enabled by Cobalt(III)-Catalyzed $C(sp^2)$ -H Amidation

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Received: October ##, 2020; Revised: October ##, 2020; Published online: October ##, 2020

Supporting information for this article is available on the WWW under http://dx.doi.org/10.1002/adsc.2020######.

Abstract. Benzofused lactams, especially indolin-2-one and dihydroquinolin-2-one are popular structural motives in durgs and natural products. Herein, we developed a room temperature and robust synthesis of benzofused lactams throug cobalt(III)-catalyzed $C(sp^2)$ -H amidation. In this protocol, in-situ formation of Cp*Co(III)(ligand) catalyst from Cp*Co(CO)I₂ and ligand simplify the synthetic effort of cobalt complexes. Simple and readily synthesized 1,4,2-dioxazol-5 ones underwent room temperature intramolecular C-H amidation and afforded a wide variety of functionalized benzofused lactams in up to 86% yield. The scalability of the reaction is also be demonstrated.

Keywords: Cobalt catalysis; Benzofused lactams; Indolin-2-ones; Dihydroquinolin-2-ones; Intramolecular C-H amidation

Introduction

Benzofused lactams, especially five-membered indolin-2-one and six-membered dihydroquinolin-2one are valuable hetero-cyclic skeletons widely present in bioactive compounds and natural products.^[1] They are important targets because of their potent biological properties, including anticancer, antiviral, antibacterial, analgestic, anti-inflammatory, antihypertensive activities and NMDA antagonistic effect.^[2] (Figure 1). They also serve as synthetic blocks for the construction of highly complex heterocycles.^[3] Consequently, efficient and straightforward approaches for the synthesis of indolin-2-one and dihydroquinolin-2-one derivatives remain in demand.

For decades, organic chemists keep looking for new and efficient routes to prepare these benzofused lactams. Traditional synthetic routes for $C(sp^2)$ -N bond construction generally employ the transitionmetal-mediated cross-coupling methodologies of amines/amides with aryl halides,^[4] particularly Cucatalyzed Ullmann-type reactions^[5] and Pd-catalyzed Buchwald-Hartwig aminations^[6] (Scheme 1a). In 2020, Radosevich developed an organocatalytic method to construct oxindoles, which used a small-ring organophosphorus-based catalyst (4·[O]) and a through hydrosilane reductant а tandem intermolecular C-N coupling of nitroarenes and boronic acid, followed by intramolecular cyclization at 120 °C^[7] (Scheme 1b). Recently, significant progress has been made in the construction of lactam



Figure 1. Selected biological molecules bearing the scaffolds of indolin-2-one and dihydroquinolin-2-one.

by intramolecular C-H amination mediated by transition-metal catalysts, such as Ir and Ru complexes.^[8] Representative study by Chang reported a series of elegant approaches to form benzofused lactams via tailored Ir-catalyzed C-H amidation reaction using dioxazolone as the nitrene precursor and a number of well-designed Ir complexes as the catalysts^[9] (Scheme 1c). However, compared with Ir and Ru catalysts, there are few reports on Co-catalyzed C-H amidation and most of them employed organic azides.^[10] An advantage of Earth abundant cobalt catalysts over the precious metal iridium or rhodium catalysts is their lower cost. So far, no reports regarding Co-catalyzed synthesis of indolin-2-one and dihydro-quinolin-2-one derivatives through an intramolecular C-H amidation have been reported.

a. Ullmann-type couplings and Buchwald-Hartwig aminations.



Lautens, 2013



b. Organophosphorus-catalyzed cascade synthesis.



c. Ir-catalyzed intramolecular C-H amidation.

Chang, 2018, 2020



Liu and Li, 2020



d. This work.



Scheme 1. General strategies of indolin-2-one and dihydroquinolin-2-one derivatives.

Herein, we report an efficient and straightforward method for the synthesis of indolin-2-one and dihydroquinolin-2-one derivatives through $Cp*Co(CO)I_2$ catalyzed intramolecular C-H amidation for the first time. A readily available cobalt catalyst and 1,4,2-dioxazol-5-ones led to wide varieties indolin-2-ones and dihydroquinolin-2-ones in good yields (27 examples, up to 86%) under mild conditions. The scalability of the amidation was demonstrated. This approach would deliver an efficient and straightforward method for the synthesis of valuable benzofused lactams of biologically active compounds.

Results and Discussion

We selected 3-phenethyl-1,4,2-dioxazol-5-one 1a as the model substrate, which was easily derived from 2phenylacetic acids (95% yield, see Supporting Information). We began to study the cyclization reaction of 1,4,2-dioxazol-5-one 1a followed Chang's

Table 1. Optimization of Co-Catalyzed C-H Amidation. [a,b]



^[a]Reactions conducted on a 0.1 mmol scale and 0.1 M reaction concentration, using 5 mol% Co as a cobalt source, 10 mol% L as ligand, 10 mol% borate and 20 mol% silver salt. [b]NMR yield determined by ¹H NMR spectroscopy of the crude reaction mixture using CH₂Br₂ as an internal standard. [c]Isolated yield after chromatographic purification.



method^[9c] and replaced Ir complexes with the Co catalyst, using 5 mol% Cp*Co(CO)I₂ (Co1) as a cobalt source, 10 mol% 8-hydroxyquinoline (L1) as a ligand, mol% and 10 sodium tetrakis(3,5bis(trifluoromethyl)phenyl) borate $(NaBAr^{F_4})$ in hexafluoro-2-propanol (HFIP) at 60 °C. However, only trace amount of the product dihydroquinolin-2one 2a was detected (Table 1, entry 1). Then, reducing the reaction temperature to room temperature and 0 °C led to 28% and 20% NMR yield (Determined after reaction workup by ¹H NMR integration), respectively (entries 2 and 3). To our delight, with 20% Ag₂CO₃ additive, the cyclization product 2a was obtained in 74% at room temperature (entry 4). Switching to other silver salts, such as AgSbF₆, AgPF₆, AgOTf and AgNTf₂, however, resulted in lower NMR yield (3860%, entries 5–8). Replacing borate NaBAr^{F₄} with KBPh₄, NaBPh₄ and NaBF₄ furnished increased NMR yield with 73%, 76% and 83% (80% isolated yield), respectively (entries 9-11). We next studied the effect of Co catalysts and ligands. Switching the Cp* ring of $Cp*Co(CO)I_2$, a methyl group was missing (Co2) or a methyl group was replaced with a propyl group (Co3), caused the NMR yield to drop to 40% and 80%, respectively (entries 12 and 13). When replacing ligand 8-hydroxyquinoline (L1) with quinolone (L2), 8-hydroxyquinoline substituted (L3-L6),aminoquinoline (L7), no improvement of yield was observed (entries 14-19). Removing the Co catalyst from the system led to no reaction (entry 20). Thus, the optimal conditions for the intramolecular C-H amidation that will be used for the remainder of the study are listed in entry 11 of Table 1.

With the optimal conditions in hand (Table 1, entry 11), we subsequently studied the substrate scope of the intramolecular C-H amidation using a variety of 3phenethyl-1,4,2-dioxazol-5-ones and 3-benzyl-1,4,2dioxazol-5-ones. As noted, 1,4,2-dioxazol-5-one readily synthesized derivatives were from commercially available carboxylic acids (56-95% vields).^[8a] For synthesis six-membered of dihydroquinolin-2-one derivatives, 3-phenethyl-1,4,2dioxazol-5-one 1a via intramolecular C-H amidation delivered product 2a in 80% yield. Moreover, 2a was characterized by X-ray crystallography, confirming its (CCDC 2033458, structure see Supporting Information for details). Phenylacetyl substrates bearing ortho-substituted groups were smoothly cyclized to obtain the corresponding dihydroquinolin-2-one. Ortho-alkyl (Me, 1b) and electro-donating (OMe, 1c) substituents furnished products 2b and 2c in 65% and 60%. Electro-withdrawing (Cl, 1d; Br, 1e; I, 1f) substituents, led to products 2d, 2e and 2f in 38%, 42% and 45%, respectively. Electron-deficient (CF₃, 1g and CO₂Me, 1h) substituents provided products 2g and 2h in 40% and 46%. Meta-substituted substrates afforded a mixture of isomeric desired products 2i/2i' and 2j/2j' in a total of 81% and 83% yields with 1.6:1 and 4:1 regioselectivities, probably owing to the steric hindrance of meta-position. Substrates bearing multiple substituent 3,4-dimethoxy (1k), polycyclic substrates benzo[1,3]dioxol (11) and naphthalen-1-yl (1m), were also suitable reactants in C–H amidation, giving products 2k, 2l and 2m in 86%, 58% and 61%, respectively. α -Substituted phenylacetyl substrates (1n-1p) underwent cyclization to obtain a series of functionalized lactam products 4-methyldihydroquinolin-2-one (2n). 4,4-dimethyland dihydroquinolin-2-one (20)4-phenyldihydroquinolin-2-one (2p) in 58%, 74% and 52%, respectively.





^[a]Reactions conducted on a 0.2 mmol scale and 0.1 M reaction concentration, using 5 mol% Cp*Co(CO)I₂ a cobalt source, 10 mol% 8-Hydroxyquinoline as ligand, 10 mol% NaBF₄ and 20 mol% Ag₂CO₃. ^[b]Yield of isolated product after chromatographic purification.

We next explored the method to synthesis of fivemembered indolin-2-one derivatives. Various substituted 3-benzyl-1,4,2-dioxazol-5-ones provided cyclization products in good yields. Benzyl substrate bearing ortho-substituted electro-donating (OMe, 1q) and electron-deficient (CF₃, 1r) groups furnished products 2q in 85% and 42 yields. Meta-substituted substrate (Cl, 1s) gave a C4-selective cyclization product (2s), suggesting that the steric hindrance may lead to this regioselectivity. Substrates bearing multiple substituents, such as 3,4-dimethoxy (1t), 3,5dimethoxy (1u), 3,5-dimethyl (1v) and 2-bromo-4methoxy (1w) groups, afforded products 2t, 2u, 2v and 2w in 83%, 64%, 58 and 61%, respectively. Polycyclic substrates benzo[1,3]dioxol (1x),2.3dihydrobenzofuran (1y) and naphthalen-1-yl (1z) led to products 2x, 2y and 2z in 53%, 67% and 62%, Naphthalen-2-yl respectively. substrate (1z')generated a mixture of isomeric desired products 2z'/2z'' in a total of 72% yield with 3.8:1 regioselectivity, and the regioselectivity presumably due to the electronic effects^[10a]. The structures of isomers were confirmed from literature report^[11].

Subsequently, we evaluated the scalability of our method. We conducted 1,4,2-dioxazol-5-ones **1a** and **1u** on a 1.2 g scale under the standard conditions of the intramolecular C–H amidation (Scheme 2). The desired dihydroquinolin-2-one **2a** and 5,6-dimethoxyindolin-2-one **2t** were isolated in 0.69 g (75% yield) and 0.75 g (78% yield). This means that the method has application value.



Scheme 2. Gram-scale synthesis.

Based on the relevant literatures on C-H amidation using Ir catalysts^[8a, 9c] and above study, we proposed the plausible mechanism for this intramolecular C-H amidation as shown in Scheme 3. First, a cationic Co(III) catalyst from Cp*Co(CO)I₂ undergoes the N,O-coordination of 8-hydroxyquinoline, forming a cobaltacycle intermediate I. Next, it further coordinates with dioxazolone of 1 to form intermediate II with releasing a molecule of CO₂. The electrons on the electron-rich aromatic ring transfer to the electrondeficient N atom to form intermediate III. Finally, it undergoes an elimination to obtain product 2a and regenerates the active cobalt complex I to continue the catalytic cycle. The reaction proceeds by an electrophilic aromatic substitution (SEAr), which provides support for Chang's proposed mechanistic pathways^[9c].



Scheme 3. Proposed reaction mechanism.

Conclusion

In summary, we have successfully synthesized a series of indolin-2-one and dihydroquinolin-2-one derivatives employing an Earth abundant cobalt catalyst and 1,4,2-dioxazol-5-ones for the first time. In this protocol, readily synthesized 1,4,2-dioxazol-5ones underwent intramolecular C-H amidation to provide a wide variety of functionalized benzofused lactams under mild conditions. Gram-scale reactions were demonstrate the scalability of the amidation and the competition and deuterium-labeling experiments were also probed. It is noteworthy that this method. not require a tailored transition-metal does complexes,^[9-10] which enables the synthesis of diverse array of benzofused lactams in a convenient and straightforward means. Furthermore, because 1,4,2-dioxazol-5-ones are easily synthesized from abundant carboxylic acids, we anticipated that the approach would deliver an efficient method for the synthesis of valuable benzofused lactams of biologically active compounds.

Experimental Section

General Methods

Unless otherwise stated, all reagents were commercially available and used as received without further purification. Chemicals were obtained from Sigma-Aldrich, Acros, TCI and Alfa-Aesar. TLC was performed with Merck TLC Silica gel60 F_{254} plates with detection under UV light at 254 nm. Silica gel (200-300 mesh, Qingdao) was used for flash chromatography. ¹H and ¹³C{¹H} NMR spectra were obtained using a Brüker DRX 600/400 spectrometer at 600/400 MHz and 150/100 MHz, respectively. Chemical shifts were reported in units of parts per million (ppm) downfield from tetramethylsilane (TMS), and all coupling constants were reported in hertz. The infrared (IR) spectra were measured on a Nicolet iS10 FTIR spectrometer with 4 cm⁻¹ resolution and 32 scans between wavenumber of 4000 cm⁻¹ and 400 cm⁻¹. High Resolution Mass spectra were

taken on AB QSTAR Pulsar mass spectrometer. Melting points were obtained on a XT-4 melting-point apparatus and were uncorrected.

General Procedure for the Preparation and Characterization of Indolin-2-ones and Dihydroquinolin-2-ones (2a–2z'').

To an oven-dried reaction flask were added Cp*Co(CO)I₂ (5 mg, 5.0 mol%), 8-Hydroxyquinoline (3 mg, 10 mol%), Ag₂CO₃ (11 mg, 20 mol%), NaBF₄ (2 mg, 10 mol%), and hexafluoro-2-propanol (2 mL) under argon atmosphere and stirred for 1 min. To the reaction flask was added 1,4,2-dioxazol-5-one (1, 0.2 mmol) and then sealed. The reaction mixture was vigorously stirred for 3 hours at room temperature. Filtered through a pad of celite with DCM (10 mL × 2) and concentrated under reduced pressure. Desired product **2** was obtained by silica chromatography (eluent: Petroleum ether/EtOAc, 3:1 ~ 1:1).

3,4-Dihydroquinolin-2(1*H*)-one (2a).

Isolated from flash chromatography on silica gel (eluted with petroleum ether : ethyl acetate = 3:1, $R_f = 0.30$); White solid (24.1 mg, 82%); m.p. 159 – 161 °C ¹H NMR (400 MHz, Chloroform-*d*) δ 9.84 (s, 1H), 7.19 – 7.12 (m, 2H), 6.97 (td, J = 7.6, 1.2 Hz, 1H), 6.89 (d, J = 7.6 Hz, 1H), 2.95 (t, J = 8.0 Hz, 2H), 2.64 (t, J = 8.0 Hz, 2H) ppm. ¹³C NMR (100 MHz, Chloroform-*d*) δ 172.7, 137.4, 127.9, 127.5, 123.6, 123.1, 115.8, 30.7, 25.3 ppm. IR (thin film): 3050, 1683, 1591, 1489, 1383, 1280, 1245, 940, 813, 745, 681 cm⁻¹; HRMS (ESI-TOF) m/z: [M+H]⁺ Calcd for C₉H₁₀NO⁺ 148.0757; Found 148.0761.

5-Methyl-3,4-dihydroquinolin-2(1*H*)-one (2b).

Isolated from flash chromatography on silica gel (eluted with petroleum ether : ethyl acetate = 3:1, $R_f = 0.28$); White solid (20.8 mg, 65%); m.p. 160 – 162 °C ¹H NMR (400 MHz, Chloroform-*d*) δ 9.69 (s, 1H), 7.06 (t, *J* = 7.6 Hz, 1H), 6.86 (d, *J* = 7.6 Hz, 1H), 6.75 (d, *J* = 8.0 Hz, 1H), 2.91 (t, *J* = 7.6 Hz, 2H), 2.64 (t, *J* = 7.6 Hz, 2H), 2.28 (s, 3H) ppm. ¹³C NMR (100 MHz, 172.4, 137.4, 136.0, 127.1, 125.0, 122.0, 113.8, 30.4, 22.0, 19.4 ppm. IR (thin film): 3204, 3076, 2948, 1676, 1591, 1478, 1391, 1217, 1198, 813, 769, 540, 452 cm⁻¹; HRMS (ESI-TOF) m/z: [M+H]⁺ Calcd for C₁₀H₁₂NO⁺ 162.0913; Found 162.0914.

5-Methoxy-3,4-dihydroquinolin-2(1*H*)-one (2c).

Isolated from flash chromatography on silica gel (eluted with petroleum ether : ethyl acetate = 2:1, R_f = 0.25); White solid (21.2 mg, 60%); m.p. 169 – 171 °C ¹H NMR (400 MHz, Chloroform-*d*) δ 7.86 (s, 1H), 6.91 (t, *J* = 8.0 Hz, 1H), 6.74 (d, *J* = 8.0 Hz, 2H), 3.83 (s, 3H), 2.93 (t, *J* = 7.6 Hz, 2H), 2.59(t, *J* = 7.2 Hz, 2H) ppm. ¹³C{¹H} NMR (100 MHz, Chloroform-*d*) δ : 170.4, 145.8, 126.5, 124.0, 122.7, 120.0, 109.0, 55.8, 30.7, 25.4 ppm. IR (thin film): 3416, 3239,

2940, 2840, 1682, 1618, 1497, 1449, 1301, 1260, 1092, 792, 733, 713, 535 cm⁻¹; HRMS (ESI-TOF) m/z: $[M+H]^+$ Calcd for $C_{10}H_{12}NO_2^+$ 178.0863; Found 178.0870.

5-Chloro-3,4-dihydroquinolin-2(1H)-one (2d)

Isolated from flash chromatography on silica gel (eluted with petroleum ether : ethyl acetate = 3:1, $R_f = 0.31$); White solid (14.0 mg, 38%); m.p. 109 – 111 °C ¹H NMR (400 MHz, Chloroform-*d*) δ 7.88 (s, 1H), 7.22 (dd, *J* = 8.0, 1.2 Hz, 1H), 7.09 – 7.04 (m, 1H), 6.91 (t, *J* = 8.0 Hz, 1H), 3.00 – 2.95 (m, 2H), 2.66 – 2.60 (m, 2H) ppm. ¹³C NMR (100 MHz, Chloroform-*d*) δ 170.5, 134.1, 127.8, 126.5, 125.5, 123.3, 119.7, 30.6, 25.8 ppm. IR (thin film): 3379, 3041, 2945, 1764, 1695, 1530, 1358, 1320, 1178, 1136, 813, 755, 681 cm⁻¹; HRMS (ESI-TOF) m/z: [M+H]⁺ Calcd for C₉H₉CINO⁺ 182.0367; Found 182.0367.

5-Bromo-3,4-dihydroquinolin-2(1H)-one (2e).

Isolated from flash chromatography on silica gel (eluted with petroleum ether : ethyl acetate = 3:1, $R_f = 0.32$); White solid (18.9 mg, 42%); m.p. 102 – 104 °C ¹H NMR (400 MHz, Chloroform-*d*) δ 7.81 (s, 1H), 7.39 (dd, *J* = 8.0, 1.2 Hz, 1H), 7.11 (d, *J* = 7.6 Hz, 1H), 6.86 (t, *J* = 7.6 Hz, 1H), 2.99 (t, *J* = 7.6 Hz, 2H), 2.64 (dd, *J* = 8.4, 6.4 Hz, 2H) ppm. ¹³C NMR (100 MHz, Chloroform-*d*) δ 170.7, 135.3, 131.0, 127.3, 125.7, 123.9, 109.7, 30.8, 26.1 ppm. IR (thin film): 3382, 3241, 2948, 1784, 1698, 1481, 1362, 1322, 1190, 1139, 803, 764, 597, 522 cm⁻¹; HRMS (ESI-TOF) m/z: [M+H]⁺ Calca for C₉H₉BrNO⁺ 225.9862; Found 225.9869.

5-Iodo-3,4-dihydroquinolin-2(1H)-one (2f)

Isolated from flash chromatography on silica gel (eluted with petroleum ether : ethyl acetate = 3:1, R_f = 0.33); White solid (24.6 mg, 45%); m.p. 115 – 117 °C ¹H NMR (400 MHz Chloroform-*d*) δ 7.69 (s, 1H), 7.60 (dd, *J* = 8.0, 1.2 Hz, 1H), 7.12 (d, *J* = 7.2 Hz, 1H), 6.71 (t, *J* = 7.6 Hz, 1H), 2.94 (dd, *J* = 8.4, 6.4 Hz, 2H), 2.64 – 2.57 (m, 2H) ppm. ¹³C NMR (100 MHz, Chloroform-*d*) δ 170.9 , 137.8 , 137.4 , 128.3 , 125.3 , 124.5 , 84.9 , 30.9 , 26.3 ppm. IR (thin film): 3388, 3265, 2966, 1795, 1496, 1377, 1335, 1212, 1151, 786, 588 cm⁻¹; HRMS (ESI-TOF) m/z: [M+H]⁺ Calcd for C₉H₉INO⁺ 273.9723; Found 273.9719.

5-(Trifluoromethyl)-3,4-dihydroquinolin-2(1*H*)-one (2g)

Isolated from flash chromatography on silica gel (eluted with petroleum ether : ethyl acetate = 3:1, $R_f = 0.30$); White solid (17.2 mg, 40%); m.p. 132 – 134 °C. ¹H NMR (400 MHz, Chloroform-*d*) δ 7.68 (d, *J* = 7.6 Hz, 1H), 7.54 (t, *J* = 7.6 Hz, 1H), 7.40 (t, *J* = 7.6 Hz, 1H), 7.34 (d, *J* = 7.6 Hz, 1H), 3.22 (t, *J* = 8.0 Hz, 2H), 2.99 – 2.92 (t, *J* = 8.0 Hz, 2H) ppm.¹³C NMR (100 MHz, Chloroform-*d*) δ 165.6, 154.0, 136.8, 132.5, 131.1, 128.8 (q, ²J_{C-F} = 29.8 Hz), 127.6, 126.7 (q, ³J_C. *F* = 5.6 Hz), 124.5 (q, ¹J_{C-F} = 272.1 Hz), 27.8, 26.8 ppm. IR

(thin film): 3088, 2953, 17108, 1585, 1432, 1255, 1210, 985, 802, 734, 551 cm⁻¹; HRMS (ESI-TOF) m/z: $[M+H]^+$ Calcd for $C_{10}H_9F_3NO^+$ 216.0631; Found 216.0630.

Methyl 2-oxo-1,2,3,4-tetrahydroquinoline-5carboxylate (2h)

Isolated from flash chromatography on silica gel (eluted with petroleum ether : ethyl acetate = 3:1, $R_f = 0.28$); White solid (18.9 mg, 46%); m.p. 145 – 157 °C ¹H NMR (400 MHz, Chloroform-*d*) δ 9.57 (s, 1H), 7.58 (dd, *J* = 8.0, 1.2 Hz, 1H), 7.22 (t, *J* = 8.0 Hz, 1H), 7.03 (dd, *J* = 8.0, 1.2 Hz, 1H), 3.90 (s, 3H), 3.39 (t, *J* = 8.0 Hz, 2H), 2.61 (t, *J* = 8.0 Hz, 2H) ppm. ¹³C NMR (100 MHz, Chloroform-*d*) δ 172.5 , 167.6 , 138.7 , 129.7 , 127.3 , 125.7 , 125.4 , 119.6 , 52.3 , 30.2 , 22.9 ppm. IR (thin film): 3088, 2953, 17108, 1585, 1432, 1255, 1210, 985, 802, 734, 551 cm⁻¹; HRMS (ESI-TOF) m/z: [M+H]⁺ Calcd for C₁₁H₁₁NO₃⁺ 206.0812; Found 206.0815.

6-Methyl-3,4-dihydroquinolin-2(1H)-one/8-Methyl-3,4-

dihydroquinolin-2(1H)-one (2i)/(2i')

Isolated from flash chromatography on silica gel (eluted with petroleum ether : ethyl acetate = 3:1, $R_f = 0.29$); White solid (26.1 mg, 81 %); Obtained as mixture of two isomers (1.6 : 1). Major isomer: 6-Methyl-3,4-dihydroquinolin-2(1*H*)-one ¹H NMR (600 MHz, Chloroform-*d*) δ 9.02 (s, 1H), 6.98 - 6.97 (m, 2H), 6.75 (d, J = 7.8 Hz, 1H), 2.93 (t, J = 6.0 Hz, 2H), 2.64(t, J = 6.0 Hz, 2H), 2.29 (s, 3H) ppm. ¹³C NMR (150 MHz, Chloroform-*d*) δ 172.4, 134.9, 132.7, 128.6, 128.0, 116.4, 115.6, 30.9, 25.4, 20.8 ppm. Minor isomer: 8-Methyl-3,4-dihydroquinolin-2(1*H*)-one ^{1}H NMR (600 MHz, Chloroform-*d*) δ 9.05 (s, 1H), 7.03 (d, *J* = 7.8 Hz, 1H), 6.80 (d, J = 4.2 Hz, 1H), 6.67 (s, 1H), 2.93 (t, J = 6.0 Hz, 2H), 2.64 (t, J = 6.0 Hz, 2H), 2.30 (s, 3H) ppm. 13 C NMR (150 MHz, Chloroform-*d*) δ 172.7, 137.6, 137.3, 127.8, 123.9, 123.6, 120.7, 31.0, 25.0, 21.1 ppm. HRMS (ESI-TOF) m/z: [M+H]⁺ Calcd for C₁₀H₁₂NO⁺ 162.0913; Found 162.0912.

6-Methoxy-3,4-dihydroquinolin-2(1H)-one (2j).

Isolated from flash chromatography on silica gel (eluted with petroleum ether : ethyl acetate = 2:1, R_f = 0.22); White solid (23.5 mg, 66%); m.p. 169 – 171 °C ¹H NMR (400 MHz, Chloroform-*d*) δ 10.01 (s, 1H), 6.85 – 6.76 (m, 1H), 6.66 (d, J = 7.6 Hz, 2H), 3.72 (s, 3H), 2.88 (t, J = 7.6 Hz, 2H), 2.63 – 2.54 (m, 2H) ppm. ¹³C NMR (100 MHz, Chloroform-*d*) δ 172.3, 155.5, 131.0, 124.9, 116.5, 113.7, 112.3, 55.5, 30.5, 25.6 ppm. IR (thin film): 3064, 2970, 2935, 1976, 1770, 1661, 1497, 1463, 1418, 1294, 1240, 1120, 792, 764, 600 cm⁻¹; HRMS (ESI-TOF) m/z: [M+H]⁺ Calcd for C₁₀H₁₂NO₂⁺ 178.0863; Found 178.0865.

8-Methoxy-3,4-dihydroquinolin-2(1H)-one (2j').

Isolated from flash chromatography on silica gel (eluted with petroleum ether : ethyl acetate = 2:1, R_f = 0.23); White solid (6 mg, 17%); m.p. 165 – 167 °C ¹H NMR (400 MHz, Chloroform-*d*) δ 7.86 (s, 1H), 6.94 (t, *J* = 8.0 Hz, 1H), 6.76 (d, *J* = 7.6 Hz, 2H), 3.86 (d, *J* = 2.0 Hz, 3H), 2.96 (t, *J* = 7.6 Hz, 2H), 2.65 – 2.60 (m, 2H) ppm. ¹³C NMR (100 MHz, Chloroform-*d*) δ 172.3, 155.5, 131.0, 124.9, 116.5, 113.7, 112.3, 55.5, 30.5, 25.6 ppm. IR (thin film): 3058, 2972, 1775, 1680, 1489, 1447, 1382, 1274, 1239, 1148, 1036, 879, 764, 554 cm⁻¹; HRMS (ESI-TOF) m/z: [M+H]⁺ Calcd for C₁₀H₁₂NO₂⁺ 178.0863; Found 178.0869.

6,7-Dimethoxy-3,4-dihydroquinolin-2(1H)-one (2k).

Isolated from flash chromatography on silica gel (eluted with petroleum ether : ethyl acetate = 1:1, R_f = 0.22); White solid (35.6 mg, 86%); m.p. 180 – 182 °C ¹H NMR (400 MHz, Chloroform-*d*) δ 9.26 (s, 1H), 6.67 (s, 1H), 6.41 (s, 1H), 3.85 (s, 3H), 3.84 (s, 3H), 2.89 (t, *J* = 7.2 Hz, 2H), 2.66 – 2.56 (m, 2H) ppm. ¹³C NMR (100 MHz, Chloroform-*d*) δ 172.3, 148.5, 144.8, 130.8, 114.9, 111.7, 100.5, 56.5, 56.3, 31.1, 25.2 ppm. IR (thin film): 3218, 2938, 2844, 1675, 1522, 1395, 1233, 1201, 1132, 1029, 933, 857, 772, 585 cm⁻¹; HRMS (ESI-TOF) m/z: [M+H]⁺ Calcd for C₁₁H₁₄NO₃⁺ 208.0968; Found 208.0971.

7,8-Dihydro-[1,3]dioxolo[4,5-g]quinolin-6(5H)-one (2l).

Isolated from flash chromatography on silica gel (eluted with petroleum ether : ethyl acetate = 2:1, $R_f = 0.25$); White solid (22.2 mg, 58%); m.p. 189 – 191 °C. ¹H NMR (400 MHz, DMSO-*d*₆) δ 9.86 (s, 1H), 6.77 (s, 1H), 6.46 (s, 1H), 5.92 (s, 2H), 2.75 (t, *J* = 7.6 Hz, 2H), 2.37 (t, *J* = 7.6 Hz, 2H) ppm. ¹³C NMR (100 MHz, DMSO-*d*₆) δ 170.4, 146.4, 142.5, 132.8, 116.3, 108.5, 101.2, 97.6, 30.9, 25.1 ppm. IR (thin film): 3148, 2781, 1705, 1635, 1503, 1358, 1290, 1192, 1034, 940, 850, 683 cm⁻¹; HRMS (ESI-TOF) m/z: [M+Na]⁺ Calcd for C₁₀H₉NaNO₃⁺ 214.0475; Found 214.0480.

1,4-Dihydrobenzo[f]quinolin-3(2H)-one (2m).

Isolated from flash chromatography on silica gel (eluted with petroleum ether : ethyl acetate = 3:1, $R_f = 0.33$); White solid (24.0 mg, 61%); m.p. 192 – 194 °C ¹H NMR (400 MHz, DMSO- d_6) δ 10.22 (s, 1H), 7.94 (d, J = 8.4 Hz, 1H), 7.83 (d, 1H), 7.75 (d, J = 8.8 Hz, 1H), 7.52 (ddd, J = 8.4, 6.8, 1.6 Hz, 1H), 7.37 (ddd, J = 8.0, 6.8, 1.2 Hz, 1H), 7.14 (d, J = 8.8 Hz, 1H), 3.25 (t, J = 7.6 Hz, 2H), 2.59 (t, J = 7.6 Hz, 2H) ppm. ¹³C NMR (100 MHz, DMSO- d_6) δ 170.1, 135.7, 131.2, 129.9, 128.4, 127.5, 126.7, 123.7, 122.7, 116.8, 115.5, 30.0, 20.5 ppm. IR (thin film): 2905, 2783, 1863, 1835, 1637, 1505, 1492, 1448, 1344, 1302, 1149, 773, 632 cm⁻¹; HRMS (ESI-TOF) m/z: [M+Na]⁺ Calcd for C₁₃H₁₁NaNO⁺ 220.0733; Found 220.0728.

4-Methyl-3,4-dihydroquinolin-2(1*H*)-one (2n).

Isolated from flash chromatography on silica gel (eluted with petroleum ether : ethyl acetate = 3:1, $R_f = 0.28$); White solid (18.7 mg, 58%); m.p. 166 – 168 °C ¹H NMR (400 MHz, Chloroform-*d*) δ 9.33 (s, 1H), 7.18 (ddd, J = 9.2, 7.6, 1.6 Hz, 2H), 7.02 (td, J = 7.6, 1.2 Hz, 1H), 6.90 – 6.83 (m, 1H), 3.13 (m, 1H), 2.74 (dd, J = 16.0, 6.0 Hz, 1H), 2.44 (dd, J = 16.0, 7.2 Hz, 1H), 1.31 (d, J = 7.2 Hz, 3H) ppm. ¹³C NMR (100 MHz, Chloroform-*d*) δ 171.9, 136.6, 128.8, 127.6, 126.6, 123.4, 115.9, 38.5, 30.8, 19.9 ppm. IR (thin film): 3204, 3076, 2948, 1676, 1591, 1478, 1391, 1217, 1198, 813, 769, 540 cm⁻¹; HRMS (ESI-TOF) m/z: [M+H]⁺ Calcd for C₁₀H₁₂NO⁺ 162.0913; Found 162.0916.

4,4-Dimethyl-3,4-dihydroquinolin-2(1H)-one (20).

Isolated from flash chromatography on silica gel (eluted with petroleum ether : ethyl acetate = 3:1, $R_f = 0.27$); White solid (25.9 mg, 74%); m.p. 170 – 172 °C. ¹H NMR (400 MHz, DMSO- d_6) δ 10.12 (s, 1H), 7.28 (dd, J = 7.6, 1.6 Hz, 1H), 7.13 (td, J = 7.6, 1.6 Hz, 1H), 6.96 (td, J = 7.6, 1.2 Hz, 1H), 6.86 (dd, J = 8.0, 1.2 Hz, 1H), 2.34 (s, 2H), 1.21 (s, 6H) ppm. ¹³C NMR (100 MHz, DMSO- d_6) δ 169.3, 136.9, 132.1, 127.1, 124.2, 122.4, 115.4, 44.9, 33.5, 27.3 ppm. IR (thin film): 3204, 3076, 2948, 1676, 1591, 1478, 1391, 1217, 1198, 813, 769, 540 cm⁻¹; HRMS (ESI-TOF) m/z: [M+Na]⁺ Calcd for C₁₁H₁₃NaNO⁺ 198.0889; Found 198.0892.

4-Phenyl-3,4-dihydroquinolin-2(1*H*)-one (2p).

Isolated from flash chromatography on silica gel (eluted with petroleum ether : ethyl acetate = 3:1, $R_f = 0.28$); White solid (21.2 mg, 60%); m.p. 188 – 190 °C ¹H NMR (400 MHz, DMSO-*d*₆) δ 10.28 (s, 1H), 7.32 (t, *J* = 7.5 Hz, 2H), 7.28 – 7.14 (m, 4H), 6.94 (m, 3H), 4.32 (t, *J* = 6.4 Hz, 1H), 2.84 (d, *J* = 6.4 Hz, 1H), 2.81 – 2.72 (m, 1H) ppm. ¹³C NMR (100 MHz, DMSO-*d*₆) δ 169.2, 142.5, 138.1, 128.6, 128.1, 127.7, 127.5, 126.8, 126.3, 122.2, 115.5, 40.9, 37.9 ppm. IR (thin film): 3060, 2890, 1675, 1612, 1488, 1387, 1265, 1188, 1030, 944, 759, 715, 508 cm⁻¹; HRMS (ESI-TOF) m/z: [M+H]⁺ Calcd for C₁₅H₁₄NO⁺ 224.1070; Found 224.1065.

4-Methoxyindolin-2-one (2q).

Isolated from flash chromatography on silica gel (eluted with petroleum ether : ethyl acetate = 2:1, $R_f = 0.22$); White solid (31.0 mg, 85%); m.p. 175 – 177 °C. ¹H NMR (400 MHz, Chloroform-*d*) δ 9.30 (s, 1H), 6.95 (t, *J* = 8.0 Hz, 1H), 6.81 (m, 2H), 3.86 (s, 3H), 3.54 (s, 2H) ppm. ¹³C NMR (100 MHz, Chloroform-*d*) δ 177.6, 143.9, 131.6, 126.2, 122.7, 116.9, 110.2, 55.7, 36.9 ppm. IR (thin film): 3160, 3073, 2938, 1698, 1628, 1495, 1465, 1328, 1259, 1211, 1078, 761, 721, 653, 569 cm⁻¹; HRMS (ESI-TOF) m/z: [M+H]⁺ Calcd for C₉H₁₀NO₂⁺ 164.0706; Found 164.0709.

4-Methoxyindolin-2-one (2r)

Isolated from flash chromatography on silica gel (eluted with petroleum ether : ethyl acetate = 3:1, $R_f = 0.28$); White

solid (16.9 mg, 42%); m.p. 172 – 174 °C ¹H NMR (400 MHz, DMSO- d_6) δ 10.72 (s, 1H), 7.38 (t, J = 8.0 Hz, 1H), 7.21 (d, J = 8.0 Hz, 1H), 7.07 (d, J = 8.0 Hz, 1H), 3.62 (s, 2H) ppm. ¹³C NMR (100 MHz, DMSO- d_6) δ 175.4 , 145.0 , 128.6 , 125.0 (q, ${}^2J_{C\cdot F} = 32.0$ Hz), 124.0 (q, ${}^1J_{C\cdot F} = 271.2$ Hz), 123.6 (q, ${}^3J_{C\cdot F} = 3.0$ Hz), 117.3 (q, ${}^3J_{C\cdot F} = 4.4$ Hz), 112.8 , 35.1 ppm.IR (thin film): 3083, 2958, 1702, 1622, 1484, 1425, 1365, 1257, 1218, 1089, 721, 653 cm⁻¹; HRMS (ESI-TOF) m/z: [M+H]⁺ Calcd for C₉H₇F₃NO⁺ 202.0474; Found 202.0478.

5-Chloroindolin-2-one (2s).

Isolated from flash chromatography on silica gel (eluted with petroleum ether : ethyl acetate = 3:1, R_f = 0.30); White solid (20.0 mg, 60%); m.p. 158 – 160 °C. ¹H NMR (400 MHz, DMSO- d_6) δ 10.47 (s, 1H), 7.25 (s, 1H), 7.20 (d, *J* = 8.4 Hz, 1H), 6.79 (d, *J* = 8.4 Hz, 1H), 3.49 (s, 2H) ppm. ¹³C NMR (100 MHz, DMSO- d_6) δ 176.0, 142.6, 128.0, 127.2, 125.1, 124.5, 110.3, 35.8 ppm. IR (thin film): 3159, 2961, 2582, 1704, 1618, 1476, 1316, 1244, 1024, 869, 798, 656, 561 cm⁻¹; HRMS (ESI-TOF) m/z: [M+H]⁺ Calcd for C₈H₇CINO⁺ 168.0211; Found 168.0217.

5,6-Dimethoxyindolin-2-one (2t).

Isolated from flash chromatography on silica gel (eluted with petroleum ether : ethyl acetate = 1:1, $R_f = 0.20$); White solid (32.0 mg, 83%); m.p. 184 – 186 °C ¹H NMR (400 MHz, Chloroform-*d*) δ 9.17 (s, 1H), 6.83 (s, 1H), 6.55 (s, 1H), 3.86 (s, 3H), 3.84 (s, 3H), 3.49 (s, 2H) ppm. ¹³C NMR (100 MHz, Chloroform-*d*) δ 179.4, 149.4, 145.1, 136.3, 116.0, 109.7, 95.9, 56.9, 56.4, 37.1 ppm. IR (thin film): 3218, 2937, 1675, 1521, 1394, 1284, 1260, 1234, 1131, 1012, 933, 859, 585 cm⁻¹; HRMS (ESI-TOF) m/z: [M+Na]⁺ Calcd for C₁₀H₁₁NaNO₃⁺216.0631; Found 216.0632.

5,7-Dimethoxyindolin-2-one (2u).

Isolated from flash chromatography on silica gel (eluted with petroleum ether : ethyl acetate = 1:1, $R_f = 0.21$); White solid (24.7 mg, 53%); m.p. 183 – 185 °C. ¹H NMR (600 MHz, DMSO- d_6) δ 10.16 (s, 1H), 6.49 (s, 1H), 6.49 (s, 1H), 3.79 (s, 3H), 3.71 (s, 3H), 3.43 (s, 2H) ppm. ¹³C NMR (150 MHz, DMSO- d_6) δ 176.4, 156.0, 144.3, 127.3, 126.0, 102.9, 98.9, 56.2, 56.1, 37.2 ppm. IR (thin film): 3065, 2948, 1695, 1642, 1480, 1340, 1257, 1223, 1075, 763, 653, 568 cm⁻¹; HRMS (ESI-TOF) m/z: [M+Na]⁺ Calcd for C₁₀H₁₁NaNO₃⁺ 216.0631; Found 216.0636.

5,7-Dimethylindolin-2-one (2v).

Isolated from flash chromatography on silica gel (eluted with petroleum ether : ethyl acetate = 3:1, R_f = 0.31); White solid (18.8 mg, 58%); m.p. 168 – 170 °C. ¹H NMR (400 MHz, DMSO-*d*₆) δ 10.30 (s, 1H), 6.81 (s, 1H), 6.77 (s, 1H), 3.40 (s, 2H), 2.19 (s, 3H), 2.14 (s, 3H) ppm. ¹³C NMR (100 MHz, DMSO-*d*₆) δ 176.7, 139.8, 129.8, 129.0, 125.4, 122.3, 118.1,

36.1, 20.6, 16.4 ppm. IR (thin film): 3028, 2924, 2869, 1824, 1634, 1492, 1352, 1262, 1148, 956, 764, 632 cm⁻¹; HRMS (ESI-TOF) m/z: $[M+H]^+$ Calcd for $C_{10}H_{12}NO^+$ 162.0913; Found 162.0912.

4-Bromo-6-methoxyindolin-2-one (2w).

Isolated from flash chromatography on silica gel (eluted with petroleum ether : ethyl acetate = 2:1, R_f = 0.27); White solid (29.3 mg, 61%); m.p. 178 – 180 °C. ¹H NMR (400 MHz, Chloroform-*d*) δ 7.94 (s, 1H), 6.88 (s, 1H), 6.79 (s, 1H), 3.77 (s, 3H), 3.61 (s, 2H) ppm. ¹³C NMR (100 MHz, Chloroform-*d*) δ 175.5, 156.3, 135.5, 127.4, 115.2, 111.7, 102.3, 56.2, 37.8 ppm. IR (thin film): 3147, 2926, 2852, 1683, 1579, 1457, 1387, 1286, 1172, 1120, 877, 700, 638, 560 cm⁻¹; HRMS (ESI-TOF) m/z: [M+H]⁺ Calcd for C₉H₉BrNO₂⁺ 241.9811; Found 241.9813.

5,7-Dihydro-6*H*-[1,3]dioxolo[4,5-f]indol-6-one (2x).

Isolated from flash chromatography on silica gel (eluted with petroleum ether : ethyl acetate = 2:1, $R_f = 0.24$); White solid (18.8 mg, 53%); m.p. 175 – 177 °C. ¹H NMR (400 MHz, Chloroform-*d*) δ 8.36 (s, 1H), 6.74 (s, 1H), 6.48 (s, 1H), 5.92 (s, 2H), 3.46 (s, 2H) ppm. ¹³C NMR (100 MHz, Chloroform-*d*) δ 178.0, 147.2, 143.3, 136.3, 116.8, 106.2, 101.1, 93.5, 36.6 ppm. IR (thin film): 3148, 2781, 1704, 1635, 1476, 1358, 1319, 1192, 1152, 1035, 938, 848, 652 cm⁻¹; HRMS (ESI-TOF) m/z: [M+H]⁺ Calcd for C₉H₈NO₃⁺ 178.0499; Found 178.0499.

2,3,5,7-Tetrahydro-6H-furo[3,2-f]indol-6-one (2y).

Isolated from flash chromatography on silica gel (eluted with petroleum ether : ethyl acetate = 2:1, $R_f = 0.28$); White solid (23.5 mg, 67%); m.p. 172 – 174 °C. ¹H NMR (400 MHz, DMSO-*d*₆) δ 10.13 (s, 1H), 6.69 (s, 1H), 6.66 (s, 1H), 4.45 (t, *J* = 8.8 Hz, 2H), 3.37 (s, 2H), 3.12 (t, *J* = 8.8 Hz, 2H) ppm. ¹³C NMR (100 MHz, DMSO-*d*₆) δ 176.3, 154.6, 136.7, 125.4, 124.8, 106.2, 70.6, 36.3, 29.6 ppm. IR (thin film): 3133, 2919, 2580, 1697, 1668, 1479, 1399, 1297, 1259, 1182, 1014, 590, 854, 665, 539 cm⁻¹; HRMS (ESI-TOF) m/z: [M+H]⁺ Calcd for C₁₀H₁₀NO₂⁺ 176.0706; Found 176.0710.

1,3-Dihydro-2H-benzo[e]indol-2-one (2z).

Isolated from flash chromatography on silica gel (eluted with petroleum ether : ethyl acetate = 3:1, $R_f = 0.32$); White solid (22.7 mg, 62%); m.p. 175 – 177 °C. ¹H NMR (400 MHz, DMSO- d_6) δ 11.17 (s, 1H), 8.07 (d, J = 8.0 Hz, 1H), 7.93 – 7.86 (m, 1H), 7.56 – 7.40 (m, 4H), 3.69 (s, 2H) ppm. ¹³C NMR (100 MHz, DMSO- d_6) δ 177.6, 139.4, 132.7, 128.3, 125.6, 125.5, 122.6, 121.9, 120.8, 120.3, 119.1, 36.9 ppm. IR (thin film): 3039, 2850, 1679, 1576, 1528, 1375, 1318, 1254, 1198, 928, 805, 774, 661, 565 cm⁻¹; HRMS (ESI-TOF) m/z: [M+H]⁺ Calcd for C₁₂H₁₀NO⁺ 184.0757; Found 184.0759.

1,3-Dihydro-2*H*-benzo[g]indol-2-one (2z').

Isolated from flash chromatography on silica gel (eluted with petroleum ether : ethyl acetate = 3:1, $R_f = 0.32$); White solid (20.9 mg, 57%); m.p. 176 – 178 °C ¹H NMR (400 MHz, DMSO- d_6) δ 11.13 (s, 1H), 8.04 (d, J = 7.9 Hz, 1H), 7.87 (d, J = 6.4 Hz, 1H), 7.51 (d, J = 8.2 Hz, 1H), 7.49 – 7.35 (m, 3H), 3.65 (s, 2H) ppm. ¹³C NMR (100 MHz, DMSO- d_6) δ 177.6, 139.4, 132.7, 128.3, 125.5, 125.5, 122.6, 121.9, 120.8, 120.3, 119.1, 36.9 ppm. IR (thin film): 3148, 2929, 1675, 1558, 1462, 1318, 1254, 1198, 805, 661, 564, 532 cm⁻¹; HRMS (ESI-TOF) m/z: [M+H]⁺ Calcd for C₁₂H₁₀NO⁺ 184.0757; Found 184.0759.

1,3-Dihydro-2H-benzo[f]indol-2-one (2z'').

Isolated from flash chromatography on silica gel (eluted with petroleum ether : ethyl acetate = 3:1, R_f = 0.31); White solid (5.5 mg, 15%); m.p. 173 – 175 °C ¹H NMR (400 MHz, Chloroform-*d*) δ 10.44 (s, 1H), 7.88 (dd, *J* = 8.0, 2.0 Hz, 2H), 7.56 (t, *J* = 7.6 Hz, 2H), 7.49 (t, *J* = 7.6 Hz, 1H), 7.40 (d, *J* = 8.4 Hz, 1H), 3.80 (s, 2H) ppm. ¹³C NMR (100 MHz, Chloroform-*d*) δ 179.7, 138.7, 133.4, 128.8, 126.4, 126.1, 122.4, 122.3, 121.4, 120.1, 119.8, 37.8 ppm. IR (thin film): 3120, 2928, 1676, 1557, 1438, 1325, 1268, 1098, 798, 656, 558 cm⁻¹; HRMS (ESI-TOF) m/z: [M+H]⁺ Calcd for C₁₂H₁₀NO⁺ 184.0757; Found 184.0759.

Gram scale synthesis

To an oven-dried 100 mL reaction flask were adde Cp*Co(CO)I₂ (0.157 g, 5.0 mol%), 8-hydroxyquinoline (0.095 g, 10 mol%), Ag₂CO₃ (0.347 g, 20 mol%), NaBF4 (0.07 g, 10 mol%), and hexafluoro-2-propanol (50 mL) under argon atmosphere and stirred for 1 min. To the reaction flask was added **1a** or **1t** and then sealed. The reaction mixture was vigorously stirred for 3 hours at room temperature. Filtered through a pad of celite with DCM (50 mL × 2) and concentrated under reduced pressure. Desired product **2a** was obtained by silica chromatography (eluent: Petroleum ether/EtOAc, 3:1) with 0.69 g, 75% yield. Desired product **2t** was obtained by silica chromatography (eluent: Petroleum ether/EtOAc, 1:1) with 0.75 g, 78% yield.

Acknowledgements

This work was supported by grants from the National Key R&D Program of China (2019YFE0109200), NSFC (21662043 and U1702286), Program for Changjiang Scholars and Innovative Research Teams in Universities (IRT17R94) and IRTSTYN, Ling-Jun Scholars (202005AB160003) and NSF (2019FY003010) of Yunnan Province, the DongLu Scholar and the YunLing Scholar Programs.

References

- [1] F.-X. Felpin, J. Coste, C. Zakri, E. Fouquet, *Chem. Eur. J.*, **2009**, *15*, 7238-7245.
- [2] a) R. Bergmann, R. Gericke, J. Med. Chem., 1990, 33, 492-504; b) J. J.-W. Duan, L. Chen, Z. R. Wasserman, Z. Lu, R.-Q. Liu, M. B. Covington, M. Qian, K. D. Hardman, R. L. Magolda, R. C. Newton, D. D. Christ, R. R. Wexler, C. P. Decicco, J. Med. Chem., 2002, 45, 4954-4957; c) R. H. Feling, G. O. Buchanan, T. J. Mincer, C. A. Kauffman, P. R. Jensen, W. Fenical, Angew. Chem. Int. Ed., 2003, 42(3), 355-357; d) A. Millemaggi, R. J. K. Taylor, Eur. J. Org. Chem., 2010, 4527-4547; e) V. V. Vintonyak, K. Warburg, H. Kruse, S. Grimme, K. Hübel, D. Rauh, H. Waldmann, Angew. Chem. Int. Ed., 2010, 49, 5902-5905.
- [3] P. Gandeepan, P. Rajamalli, C.-H. Cheng, Synthesis, 2016, 48, 1872-1879.
- [4] a) B. Egle, J. Muñoz, N. Alonso, W. De Borggraeve, A. de la Hoz, A. Díaz-Ortiz, J. Alcázar, J. Flow Chem., 2015, 4, 22-25; b) Y. Motoyama, K. Kamo, H. Nagashima, Org. Lett., 2009, 11, 1345-1348; c) L. Yang, L. Shi, Q. Xing, K. W. Huang, C. Xia, F. Li, ACS Catal., 2018, 8, 10340-10348; d) T. Yoshino, S. Matsunaga, Adv. Synth. Catal., 2017, 359, 1245-1262; e) W. H. Li, C. Li, L. Dong, Asian J. Org. Chem., 2018, 7, 2448-2451; f) C. Rajitha, P. K. Dubey, V. Sunku, V. R. Veeramaneni, M. Pal, J. Heterocyclic Chem., 2013, 50, 630-637. g) S.-B. Wang, Q. Gu, S.-L. You, Organometallics, 2017, 36, 4359-4362.
- [5] a) B. Lu, D. Ma, Org. Lett., 2006, 8, 6115-6118; b) F. Monnier, M. Taillefer, Angew. Chem. Int. Ed., 2009, 48, 6954-6971; c) C. Sambiagio, S. P. Marsden, A. J. Blacker, P. C. McGowan, Chem. Soc. Rev, 2014, 43, 3525-3550.

- [6] a) C.-H. Cheng, P. Gandeepan, P. Rajamalli, Synthesis, **2016**, 48, 1872-1879; b) B. Gao, G. Zhang, X. Zhou, H. Huang, Chem. Sci., **2018**, 9, 380-386; c) Z. Xu, K. Li, R. Zhai, T. Liang, X. Gui, R. Zhang, RSC Adv., **2017**, 7, 51972-51977; d) L. Zhang, L. Sonaglia, J. Stacey, M. Lautens, Org. Lett., **2013**, 15, 2128–2131.
- [7] T. V. Nykaza, G. Li, J. Yang, M. R. Luzung, A. T. Radosevich, Angew. Chem. Int. Ed., 2020, 59, 4505-4510.
- [8] a) J. Liu, W. Ye, S. Wang, J. Zheng, W. Tang, X. Li, J. Org. Chem., 2020, 85, 4430-4440; b) Y. Park, S. Chang, Nat. Catal., 2019, 2, 219-227; c) H. Wang, Y. Park, Z. Bai, S. Chang, G. He, G. Chen, J. Am. Chem. Soc., 2019, 141, 7194-7201; d) Q. Xing, C. M. Chan, Y. W. Yeung, W. Y. Yu, J. Am. Chem. Soc., 2019, 141, 3849-3853; e) Z. Zhou, S. Chen, Y. Hong, E. Winterling, Y. Tan, M. Hemming, K. Harms, K. N. Houk, E. Meggers, J. Am. Chem. Soc., 2019, 141, 19048-19057.
- [9] a) S. Y. Hong, Y. Park, Y. Hwang, Y. B. Kim, M.-H Baik, S. Chang, *Science*, **2018**, *359*, 1016–1021; b) Y. Hwang, H. Jung, E. Lee, D. Kim, S. Chang, *J. Am. Chem. Soc.*, **2020**, *142*, 8880-8889; c) Y. Hwang, Y. Park, Y. B. Kim, D. Kim, S. Chang, *Angew. Chem. Int. Ed.*, **2018**, *57*, 13565-13569.
- [10] a) J. Lee, J. Lee, H. Jung, D. Kim, J. Park, S. Chang, J. Am. Chem. Soc., 2020, 142, 12324-12332; b) V. Lyaskovskyy, A. I. Suarez, H. Lu, H. Jiang, X. P. Zhang, B. de Bruin, J. Am. Chem. Soc., 2011, 133, 12264-12273; c) O. Villanueva, N. M. Weldy, S. B. Blakey, C. E. MacBeth, Chem. Sci., 2015, 6, 6672-6675. d) S.-B. Wang, Q. Gu, S.-L. You, J. Catal., 2018, 361, 393-397.
- [11] A. Ganguly, J.-F Zhu, T. L. Kelly, J. Phys. Chem. C, 2017, 121, 9110–9119.

FULL PAPER

Room Temperature Benzofused Lactam Synthesis Enabled by Cobalt(III)-Catalyzed C(sp²)-H Amidation

Adv. Synth. Catal. 2020, 362, Page – Page

Minyan Li* and Xiaodong Yang*

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