Palladium(II)-Catalyzed Highly Stereoselective Sandwich-Type Triple Cyclization Reaction of 1,5-Bisallenes and 2,3-Allenoic Acids

Xiongdong Lian^[a] and Shengming Ma^{*[a, b]}

Transition-metal-catalyzed cyclization of functionalized allenes has become one of the most efficient and powerful tools for the synthesis of carbo- and heterocyclic systems.^[1,2] In this area, the metal-catalyzed homodimerization reaction of 1,2-allenvl ketones was first reported by Hashmi et al., which led to the formation of furan derivatives.^[3] In 2006, Alcaide and co-workers also reported the cross-coupling cyclization reaction of allenols in the presence of 2,3-allenyl carboxylates.^[4] Since 2002,^[5] we have demonstrated the cyclization of 2,3-allenoic acids in the presence of 1,2-allenvl ketones^[5] or 2.3-allenols.^[6] affording differently substituted 2(5H)-furanones. We also studied the cyclization reaction of 2,3-allenoic acids in the presence of simple monoallenes, which yielded the stereodefined 4-(bromo-2(*E*)-alken-2-yl)-2(5H) furanones (Scheme 1).^[7] Based on this, we reasoned that the cyclization of 2,3-allenoic acids in the presence of 1,5-bisallenes may involve further cyclization to form 3-type of products based on the regioselectivity of carbopalladation of allenes.^[8] Herein, we wish to disclose our recent unexpected observation of Pd^{II}-catalyzed sandwich-type triple cyclization of 2,3-allenoic acids in the presence of 1,5-bisallenes affording 4-type tricyclic products (Scheme 2). By applying the optically active 2,3-allenoic acids, an efficient chirality transfer has also been achieved.

At first, we conducted the reaction of 2,4-dimethyl-2,3pentadienoic acid (2a) in the presence of bis(2,3-butadienyl) tosylamide (1a) under the conditions described in Ref. [7].

[a] X. Lian, Prof. Dr. S. Ma State Key Laboratory of Organometallic Chemistry Shanghai Institute of Organic Chemistry Chinese Academy of Sciences 345 Linglin Lu, Shanghai 200032 (P.R. China)
[b] Prof. Dr. S. Ma

Shanghai Key Laboratory of Green Chemistry and Chemical Process Department of Chemistry, East China Normal University 3663 North Zhongshan Lu, Shanghai 200062 (P.R. China) Fax: (+86)21-64167510 E-mail: masm@mail.sioc.ac.cn

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Scheme 1. Palladium-catalyzed cyclizations of 2,3-allenoic acids in the presence of monoallenes or 1,5-bisallenes (BQ=benzoquinone).

The expected product **3a** was not formed, instead, unknown products were formed as an inseparable mixture. Interestingly, when the reaction was conducted in the absence of LiBr by using $[PdCl_2(PhCN)_2]$ as the catalyst in DMSO at 60 °C for 2 h, two tricyclic products *cis*-**4a** (32%) and **5a** (10%) were afforded, which were confirmed by spectroscopic analysis: two molecules of allenoic acids formed the two butenolide moieties connected to a central five- or ninemembered ring formed from the cyclization of the bisallene **1a** (Scheme 2).

Optimization of reaction conditions with the purpose of improving the selectivity of cis-4a/5a was then conducted with some of the representative results being listed in Table 1. First, the solvent effect was examined and the reaction in MeCN afforded cis-4a and 5a in 45% (41% isolated) and 15% (10% isolated) yield, respectively (Table 1, entry 5). Fortunately, cis-4a was afforded exclusively and the yield was improved significantly when the reaction was carried out in MeCN by adding 1,4-bis(diphenylphosphino)-butane (dppb; 5 mol%) as the ligand (Table 1, entry 9). Then some bidentate phosphine ligands were screened and 1,2-bis(diphenylphosphino)ethane (dppe) was found to be the best, affording cis-4a in 79% ¹H NMR yield (Table 1, entry 10).



7960

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Scheme 2. Pd-catalyzed cyclization reaction of 1,5-bisallene 1a and 2,3-allenoic acid 2a.

Table 1. Pd-catalyzed cyclization reaction of 2,3-allenoic acid 2a in the presence of 1,5-bisallene 1a: Optimization of reaction conditions.^[a]



[a] The reaction was carried out under argon in a Schlenk tube by using **1a** (0.2 mmol), **2a** (0.6 mmol), BQ (0.25 mmol), and the palladium complex (5 mol%) in a solvent (3 mL) at 70 °C. [b] Determined by ¹H NMR analysis with 1,3,5-trimethyl benzene as the internal standard (n.d. = not determined).

The optimized reaction conditions were then applied to a range of different 1,5-bisallenes and 2,3-allenoic acids (Table 2). For the 2,3-allenoic acids, the fully alkyl-, 2-allyl-, disubstituted, and monosubstituted acids may all react with differently tethered 1,5-bisallenes (Z=CH₂, NTs, O) to afford the tricyclic products *cis*-**4** in moderate to good yields with excellent stereoselectivity. In all these cases, only the *cis* product was formed. The structure of the product *cis*-**4c** was further determined by its X-ray diffraction study (Figure 1 a).^[9]

Note that the diallyl-substituted *cis* products 4c and 4g may be easily converted to fused tetracyclic products 6a and 6b with a twelve-membered ring that has three Z-C=C



Scheme 3. The RCM reaction to construct the fused twelve-membered rings (Grubbs 2nd=Grubbs second-generation Ru catalyst).

bonds^[10] through a ring-closing

However, the reaction of racemic allenoic acid 2e with axial chirality gave diastereomeric mixtures.^[13] Interestingly, when the corresponding optically active (R)-2,3-allenoic acid 2e^[14] was treated with 1a under the standard reaction conditions, the optically active product (R,S,R,R)-4i was isolated (Scheme 4a). To determine the absolute configurations in the product, bisallene 1d with a bromine atom was prepared. (S)-2,3-Allenoic acids **2f** and 2g were able to couple with 1d under the similar conditions yielding the products (S,R,S,S)-4j and (S,R,S,S)-4k, respectively, in acceptable yields (Scheme 4b). The absolute configurations of the products were then unambiguously assigned by the X-ray diffraction study of (S,R,S,S)-4k, indicating the excellent chirality transfer from the axial chirality of the optically active allenoic acid to the product (Figure 1 c).[15]

With the explicit absolute stereochemical information for the starting materials and the products, a possible mechanism for this transformation is shown in Scheme 5 by using the Pd-catalyzed triple cyclization of (S)-2g with 1,5-bisallene 1d af-

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Table 2. Pd^{II} -catalyzed cyclization reaction of bimolecular 2,3-allenoic acids **2** in the presence of 1,5-bisallene **1**: The substrate scope.^[a]

a)



				cis-4 O
Entry	\mathbb{R}^1	\mathbb{R}^2	Z	Yield of <i>cis</i> -4 [%] ^[b]
1	Me	Me (2a)	TsN (1a)	73 (cis- 4a)
2	Н	Me (2b)	TsN (1a)	81 (cis-4b)
3	allyl	Me (2c)	TsN (1a)	59 (cis- 4c)
4	Н	$-(CH_2)_5-(2d)$	TsN (1a)	59 (cis-4d)
5	Me	Me (2a)	CH ₂ (1b)	68 (cis- 4e)
6	Н	Me (2b)	CH ₂ (1b)	79 (cis- 4 f)
7	allyl	Me (2c)	CH ₂ (1b)	52 (cis- 4 g)
8	Н	Me (2a)	O (1c)	54 (<i>cis</i> - 4h)





Scheme 4. Pd^{II} -catalyzed cyclization reaction of optically active 2,3-allenoic acids **2e-g** in the presence of 1,5-bisallenes a) **1a** and b) **1d**.

fording (S,R,S,S)-4k as the typical example. Firstly, the highly stereoselective cyclic *anti*-oxypalladation^[2g] of (S)-2g would form intermediate M1 with a center of chirality. Subsequent carbopalladation of one allene moiety in bisallene 1d with M1 would form the π -allylic intermediate M2-a or M2-b.^[8] Owing to the steric interaction between the pseudoaxial proton and the 2(5*H*)-furanonyl vinyl group in M2b, the reaction proceeds via the intermediate M2-a to generate *cis*-M3.^[8,16] A second molecule of 2,3-allenoic acid (*S*)-2g would undergo sequential coordination and *anti*-oxypalladation^[2g] with the vinyl palladium *cis*-M3 to generate *cis*-



Figure 1. ORTEP representations of a) cis-**4c**, b) cis-**6a**, and c) (S,R,S,S)-**4k** shown with ellipsoids at the 30% probability level.

M4, which upon reductive elimination would release the final tricyclic product (S,R,S,S)-**4k** and Pd⁰. The Pd^{II} catalyst would finally be regenerated by the reaction of Pd⁰ with BQ and H⁺.^[17]

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Scheme 5. Plausible catalytic cycle for the formation of (S,R,S,S)-4k from (S)-2g and 1d.

In summary, we have observed an unexpected sandwichtype triple cyclization of two molecules of 2,3-allenoic acid in the presence of one molecule of 1,5-bisallene with excellent stereoselectivity. The axial chiralities of the optically active 2,3-allenoic acids can be transferred to the products efficiently. Based on the careful X-ray diffraction studies of some of the products, a possible mechanism involving sequential highly stereoselective *anti*-oxypalladation, intermolecular carbopalladation, cyclic *cis*-carbopalladation, *anti*oxypalladation, and reductive elimination to afford the tricyclic products have been proposed. Further study in this area is being carried out in our laboratory.

Experimental Section

Typical procedure: 1,5-Bisallene 1a (55 mg, 0.20 mmol), 2,3-allenoic acid 2a (74 mg, 0.59 mmol), [PdCl₂(PhCN)₂] (4 mg, 0.010 mmol), dppe (5 mg, 0.013 mmol), BQ (29 mg, 0.27 mmol), and MeCN (3 mL) were sequentially added to a dried Schlenk tube under an Ar atmosphere. The resulting mixture was stirred at 70 °C under an Ar atmosphere and after completion of the reaction (monitored by TLC), the resulting mixture was diluted with CH₂Cl₂ (50 mL) and was washed sequentially with aqueous K₂CO₃ (10%, 30 mL) and water (30 mL). Drying over Na₂SO₄, filtration, evaporation, and purification by column chromatography on silica gel (eluent: CH_2Cl_2 /diethyl ether=10:1) afforded the product cis-4a as a white solid (76 mg, 73 % yield). M.p. 197-199 °C (acetone/ethyl ether = 1:10); ¹H NMR (300 MHz, CDCl₃): $\delta = 7.73$ (d, J = 7.8 Hz, 2H), 7.36 (d, J=7.8 Hz, 2H), 5.45 (s, 2H), 5.22 (s, 2H), 3.64-3.54 (m, 2H), 3.50-3.40 (m, 2H), 3.05-2.94 (m, 2H), 2.45 (s, 3H), 1.76 (s, 6H), 1.40 (s, 6H), 1.38 ppm (s, 6H); 13 C NMR (CDCl₃, 75.4 MHz): $\delta = 172.1$, 165.1, 144.0, 137.5, 133.5, 129.8, 127.1, 124.9, 120.7, 85.4, 51.2, 43.7, 25.9, 25.8, 21.3, 10.0 ppm; IR (neat): v=1749, 1667, 1621, 1596, 1487, 1465, 1367, 1331, 1279, 1220, 1198, 1163, 1132, 1091, 1046, 1015, 972, 955, 926, 903, 810, 763, 723, 705, 661, 623 cm⁻¹; EIMS: m/z (%): 525 [M⁺] (4.38), 480 [M⁺ $-3CH_3$] (6.24), 370 [M⁺-Ts] (68.99), 91 [CH₃C₆H₄⁺] (100); elemental analysis calcd (%) for C29H35NO6S: C 66.26, H 6.71, N 2.66; found: C 66.29, H 6.68, N 2.58.

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Keywords: allenes • chirality • cyclization • palladium • ringclosing metathesis reaction

- For the most recent monograph on the chemistry of allenes, see: Modern Allene Chemistry, Vol. 1 (Eds.: N. Krause, A. S. K. Hashmi), Wiley-VCH, Weinheim, 2004; Modern Allene Chemistry, Vol. 2 (Eds.: N. Krause, A. S. K. Hashmi), Wiley-VCH, Weinheim, 2004.
- [2] For selected reviews on the synthesis of cyclic compounds based on the chemistry of allenes, see: a) Y. Yamamoto, U. Radhakrishnan, Chem. Soc. Rev. 1999, 28, 199; b) R. C. Larock, J. Organomet. Chem. 1999, 576, 111; c) R. Grigg, V. Sridharan, J. Organomet. Chem. 1999, 576, 65; d) R. Zimmer, C. U. Dinesh, E. Nandanan, F. A. Khan, Chem. Rev. 2000, 100, 3067; e) A. S. K. Hashmi, Angew. Chem. 2000, 112, 3737; Angew. Chem. Int. Ed. 2000, 39, 3590; f) S. Ma, Acc. Chem. Res. 2003, 36, 701; g) S. Ma in Palladium in Organic Synthesis (Ed.: J. Tsuji), Springer, Berlin, 2005, pp. 183–210; h) S. Ma, Chem. Rev. 2005, 105, 2829; i) S. Ma, Aldrichimica Acta 2007, 40, 91; j) H. Kim, L. J. Williams, Curr. Opin. Drug Discovery Dev. 2008, 11, 870.
- [3] a) A. S. K. Hashmi, Angew. Chem. 1995, 107, 1749; Angew. Chem. Int. Ed. Engl. 1995, 34, 1581; b) A. S. K. Hashmi, L. Schwarz, Chem. Ber./Recl. 1997, 130, 1449; c) A. S. K. Hashmi, T. L. Ruppert, T. Knöfel, J. W. Bats, J. Org. Chem. 1997, 62, 7295; d) A. S. K. Hashmi, J.-H. Choi, J. W. Bats, J. Prakt. Chem. 1999, 341, 342; e) A. S. K. Hashmi, L. Schwarz, J. W. Bats, J. Prakt. Chem. 2000, 342, 40; f) A. S. K. Hashmi, L. Schwarz, J. H. Choi, T. M. Frost, Angew. Chem. 2000, 112, 2382; Angew. Chem. Int. Ed. 2000, 39, 2285; g) A. S. K. Hashmi, M. C. Blanco, D. Fischer, J. W. Bats, Eur. J. Org. Chem. 2006, 1387.
- [4] B. Alcaide, P. Almendros, T. Martínez del Campo, Angew. Chem. 2006, 118, 4613; Angew. Chem. Int. Ed. 2006, 45, 4501.
- [5] a) S. Ma, Z. Yu, Angew. Chem. 2002, 114, 1853; Angew. Chem. Int. Ed. 2002, 41, 1775; b) S. Ma, Z. Yu, Chem. Eur. J. 2004, 10, 2078.
- [6] a) S. Ma, Z. Gu, J. Am. Chem. Soc. 2005, 127, 6182; b) Y. Deng, J. Li, S. Ma, Chem. Eur. J. 2008, 14, 4263.
- [7] Z. Gu, X. Wang, W. Shu, S. Ma, J. Am. Chem. Soc. 2007, 129, 10948.
 [8] For a comprehensive review on the regioselectivity of carbopalladation of allenes, see: T. Bai, S. Ma, G. Jia, Coord. Chem. Rev. 2009, 253, 423.
- [9] Crystal data for *cis*-4c: Recrystallization from acetone/diethyl ether (1:10); $C_{33}H_{39}NO_6S$; $M_W = 577.71$; monoclinic; P21/c; a=14.215(6), b=8.829(4), c=25.699(10) Å; a=90, $\beta=100.439(10)$, $\gamma=90^{\circ}$; V=3168(2) Å³; Z=4; T=293(2) K; MO_{Ka} ; $\lambda=0.71073$ Å; final *R* indices $[I>2\sigma(I)]$, R1=0.0744, wR2=0.1781; *R* indices (all data) R1=0.1374, wR2=0.2053; reflections collected/unique=16275/5899 ($R_{int}=0.1565$); number of observations $[I>2\sigma(I)]=2836$; parameters=375. CCDC-763152 contains the supplementary crystallographic data for this paper. These data can be obtained free of charge from These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/data_request/cif.
- [10] For a review on the synthesis of medium-sized rings by using the RCM strategy, see: M. E. Maier, Angew. Chem. 2000, 112, 2153; Angew. Chem. Int. Ed. 2000, 39, 2073.
- [11] a) Handbook of Metathesis (Ed.: R. H. Grubbs), Wiley-VCH, Weinheim, **2003**; b) R. H. Grubbs, *Tetrahedron* **2004**, 60, 7117; c) R. H. Grubbs, Angew. Chem. **2006**, 118, 3845; Angew. Chem. Int. Ed. **2006**, 45, 3760.

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- [12] Crystal data for **6a**·¹/₂Et₂O: Recrystallization from acetone/diethyl ether (1:10); C₃₃H₄₀NO_{6.5}S; M_W =586.72; monoclinic; P21/c; ; a= 10.4161(13), b=12.9402(17), c=13.4539(18) Å; a=111.791(2), β = 92.116(3), γ =109.239(3)°; V=1563.3(4) Å³; Z=2; T=293(2) K; MO_{Ka}; λ =0.71073 Å; final *R* indices [$I > 2\sigma(I)$], R1=0.0775, wR2= 0.2097; *R* indices (all data): R1=0.1126, wR2=0.2334 reflections collected/unique=8581/6010 (R_{int} =0.1109); number of observations [$I > 2\sigma(I)$]=3494; parameters 367. CCDC-763154 contains the supplementary crystallographic data for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/data_request/cif.
- [13] See the Supporting Information for the crude ¹H NMR spectra.
- [14] K. Kobayashi, H. Naka, A. E. H. Wheatley, Y. Kondo, Org. Lett. 2008, 10, 3375.
- [15] Crystal data for (*S*,*R*,*S*,*S*)-**4k**: Recrystallization from acetone/diethyl ether (1:10); C₂₆H₂₈BrNO₆S; M_W =562.46; monoclinic; *P*21/*c*; *a*= 9.893(4), *b*=7.771(3), *c*=17.251(7) Å; *a*=90, *β*=98.223(7), *γ*=90°;

V=1312.5(9) Å³; Z=2; T=293(2) K; Mo_{Ka}; $\lambda=0.71073$ Å; final *R* indices $[I>2\sigma(I)]$, R1=0.0631, wR2=0.1425; *R* indices (all data): R1=0.1083, wR2=0.1582; reflections collected/unique=6708/4596 ($R_{iat}=0.1090$); number of observation $[I>2\sigma(I)]=2439$; parameters 320. CCDC-763153 contains the supplementary crystallographic data for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via www.ccdc.cam. ac.uk/data_request/cif.

- [16] For the reports on the formation of vinylic palladium intermediates from the carbopalladation of allenes, see: a) W. Oppolzer, A. Pimm, B. Stammen, W. E. Hume, *Helv. Chim. Acta* 1997, *80*, 623; b) J. Böhmer, R. Grigg, J. D. Marchbank, *Chem. Commun.* 2002, 768; c) G. Zhu, Z. Zhang, *Org. Lett.* 2004, *6*, 4041.
- [17] For a seminal paper, see: H. Grennberg, A. Gogoll, J.-E. Bäckvall, Organometallics 1993, 12, 1790.

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7964 -