

**SHORT  
COMMUNICATIONS**

## Reaction of 1-Methyl-3,4-dihydroisoquinolines with Alkyl 4-Aryl-2,4-dioxobutanoates

V. V. Khalturina<sup>a</sup>, Yu. V. Shklyaev<sup>a</sup>, and A. N. Maslivets<sup>b</sup>

<sup>a</sup> Institute of Technical Chemistry, Ural Division, Russian Academy of Sciences, Perm, Russia

<sup>b</sup> Perm State University, ul. Bukireva 15, Perm, 614990 Russia

e-mail: koh2@psu.ru

Received July 4, 2008

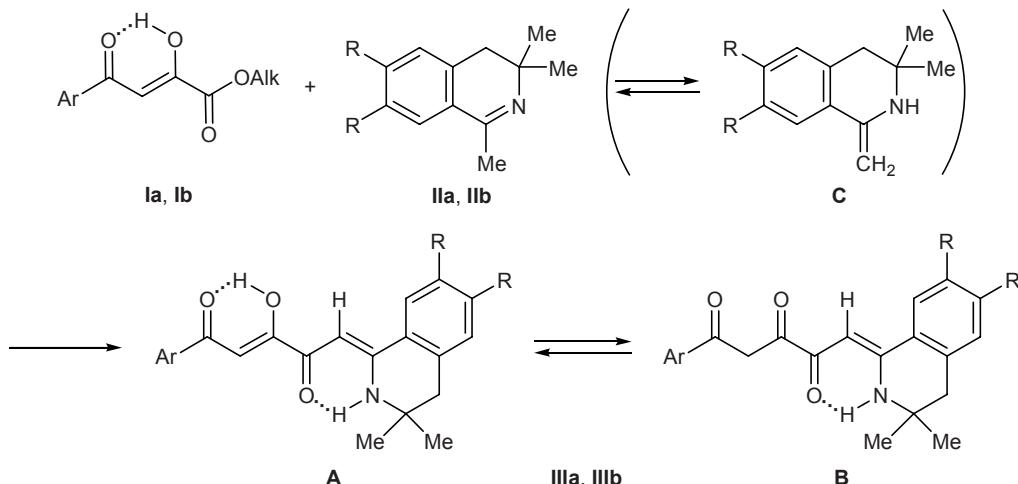
DOI: 10.1134/S107042800906030X

We previously reported on the acylation of enamino tautomers of substituted 1-methyl-3,4-dihydroisoquinolines at the exocyclic methylene group with 5-aryl-2,3-dihydrofuran-2,3-diones and obtained the corresponding (2Z,5Z)-1-aryl-3-hydroxy-5-[3,3-dimethyl-3,4-dihydroisoquinolin-1(2H)-ylidene]pent-2-ene-1,4-diones [1, 2]. In the present communication we describe the reaction of aroylpyruvic acid alkyl esters **I** with substituted 1-methyl-3,4-dihydroisoquinolines **II**. Our interest in this reaction originates from the fact that esters **I** (alcoholysis products of 5-aryl-2,3-dihydrofuran-2,3-diones [3]) usually differ in chemical behavior from parent furandiones [4]. It should also be noted that esters **I** may be prepared by the Claisen condensation of acetophenones with dialkyl oxalates [5] and that they are more accessible compounds than 5-aryl-2,3-dihydrofuran-2,3-diones.

By heating alkyl (Z)-4-aryl-2-hydroxy-4-oxobut-2-enoates (alkyl aroylpyruvates) **Ia** and **Ib** with isoquinolines **IIa** and **IIb** (molar ratio 1:1) in boiling anhydrous chloroform for 5–7 h we obtained the corresponding (2Z,5Z)-1-aryl-3-hydroxy-5-[3,3-dimethyl-3,4-dihydroisoquinolin-1(2H)-ylidene]pent-2-ene-1,4-diones **IIIa** and **IIIb** which were identified by comparing with authentic samples whose structure was confirmed by X-ray analysis.

The spectral parameters of compounds **IIIa** and **IIIb** in DMSO-*d*<sub>6</sub> solution indicated that they exist as mixtures of enol and ketone tautomers **A** and **B** at a ratio of ~9:1.

Presumably, the described transformation involves acylation of the exocyclic methylene group in enamino tautomer **C** of isoquinolines **II** with the ester group of



**I**, Alk = Me, Ar = Ph (**a**); Alk = Et, Ar = C<sub>6</sub>H<sub>4</sub>NO<sub>2</sub>-*p* (**b**); **II**, R = H (**a**), OMe (**b**); **III**, R = H, Ar = Ph (**a**); R = OMe, Ar = C<sub>6</sub>H<sub>4</sub>NO<sub>2</sub>-*p* (**b**).

compounds **I**. This process is not typical of esters **I**; they usually react with nucleophiles via replacement at the most electrophilic  $\alpha$ -carbon atom [4, 6].

**(2Z,5Z)-5-[3,3-Dimethyl-3,4-dihydroisoquinolin-1(2H)-ylidene]-3-hydroxy-1-phenylpent-2-ene-1,4-dione (IIIa).** A solution of 5.0 mmol of ester **Ia** and 5.0 mmol of isoquinoline **IIa** in 20 ml of anhydrous chloroform was heated for 5 h under reflux (the progress of the reaction was monitored by TLC). The mixture was cooled, and the precipitate was filtered off. Yield 1.06 g (89%), mp 146–148°C (decomp.; from ethyl acetate–ethanol, 2:1). IR spectrum,  $\nu$ ,  $\text{cm}^{-1}$ : 3150 br (NH, OH, assoc.), 1593 br (C=O, assoc.).  $^1\text{H}$  NMR spectrum,  $\delta$ , ppm: **A**: 1.32 s (6H, Me), 2.99 s (2H, 4'-H), 6.51 s (1H, 5-H), 7.16 s (1H, 2-H), 7.36–8.07 m (9H, H<sub>arom</sub>), 12.06 s (1H, NH), 15.95 br.s (1H, OH); **B**: 1.28 s (6H, Me), 2.95 s (2H, 4'-H), 4.51 s (2H, 2-H), 6.30 s (1H, 5-H), 7.34–8.07 m (9H, H<sub>arom</sub>), 11.57 s (1H, NH).  $^{13}\text{C}$  NMR spectrum,  $\delta_{\text{C}}$ , ppm: 27.38 (Me), 50.05 (C<sup>4'</sup>), 85.09 (C<sup>3'</sup>), 93.58 (C<sup>5</sup>), 112.05 (C<sup>2</sup>), 119.78–135.86 (C<sub>arom</sub>), 159.37 (C<sup>3</sup>), 180.00 (C<sup>1'</sup>), 180.82 (C<sup>4</sup>), 188.17 (C<sup>1</sup>). Found, %: C 63.59; H 5.41; N 6.03.  $\text{C}_{24}\text{H}_{24}\text{N}_2\text{O}_7$ . Calculated, %: C 63.71; H 5.35; N 6.19.

**(2Z,5Z)-5-[6,7-Dimethoxy-3,3-dimethyl-3,4-dihydroisoquinolin-1(2H)-ylidene]-3-hydroxy-1-(4-nitrophenyl)pent-2-ene-1,4-dione (IIIb)** was synthesized in a similar way. Yield 86%, mp 148–150°C (decomp.; from ethanol). IR spectrum,  $\nu$ ,  $\text{cm}^{-1}$ : 3190 br (NH, OH, assoc.), 1603 br (C=O, assoc.).  $^1\text{H}$  NMR spectrum,  $\delta$ , ppm: **A**: 1.32 s (6H, Me), 2.91 s (2H, 4'-H), 3.86 s (3H, OMe), 3.87 s (3H, OMe), 6.44 s (1H, 5-H), 6.98 s (1H, H<sub>arom</sub>), 7.22 s (1H, 2-H), 7.31 s (1H, H<sub>arom</sub>), 7.08–8.38 m (4H, C<sub>6</sub>H<sub>4</sub>), 12.07 s (1H, NH), 16.20 br.s (1H, OH); **B**: 1.27 s (6H, Me), 2.87 s

(2H, 4'-H), 3.85 s (3H, OMe), 3.86 s (3H, OMe), 4.55 s (2H, 2-H), 6.22 s (1H, 5-H), 6.96 s (1H, H<sub>arom</sub>), 7.26 s (1H, H<sub>arom</sub>), 7.06–8.23 m (4H, C<sub>6</sub>H<sub>4</sub>), 11.56 s (1H, NH). Found, %: C 63.59; H 5.41; N 6.03.  $\text{C}_{24}\text{H}_{24}\text{N}_2\text{O}_7$ . Calculated, %: C 63.71; H 5.35; N 6.19.

The IR spectra were recorded on an FMS-1201 spectrometer from samples dispersed in mineral oil. The  $^1\text{H}$  NMR spectra were obtained on a Bruker WP-400 spectrometer using DMSO-*d*<sub>6</sub> as solvent and TMS as internal reference. The purity of the products was checked by TLC on Silufol plates; eluent ethyl acetate–benzene, 1:5; development with iodine vapor.

This study was performed under financial support by the Russian Foundation for Basic Research (project nos. 08-03-01032, 07-03-96036, 07-03-00001).

## REFERENCES

- Shklyaev, Yu.V. and Maslivets, A.N., *Russ. J. Org. Chem.*, 1996, vol. 32, p. 302.
- Khalturina, V.V., Shklyaev, Yu.V., and Maslivets, A.N., *Doklady IV Vserossiiskoi konferentsii "Enaminy v organiceskem sinteze"* (Proc. IVth All-Russian Conf. "Enamines in Organic Synthesis"), Perm, 2007, p. 307.
- Andreichikov, Yu.S., Tendryakova, S.P., Nalimova, Yu.A., and Plakhina, G.D., *Khim. Geterotsikl. Soedin.*, 1977, p. 1030.
- Andreichikov, Yu.S., Gein, V.L., Zalesov, V.V., Kozlov, A.P., Kollenz, G., Maslivets, A.N., Pimenova, E.V., and Shurov, S.N., *Khimiya pyatichlennykh 2,3-diokso-geterotsiklov* (Chemistry of Five-Membered 2,3-Dioxo Heterocycles), Perm: Perm. Gos. Univ., 1994, p. 5.
- Maurin, C., Bailly, F., and Cotelle, Ph., *Tetrahedron*, 2004, vol. 60, p. 6479.
- Andreichikov, Yu.S., Kozlov, A.P., Tendryakova, S.P., and Nalimova, Yu.A., *Zh. Org. Khim.*, 1977, vol. 13, p. 2559.