

Published on Web 03/08/2003

Highly Enantioselective Conjugate Addition of Dialkylzinc Reagents to Acyclic Nitroalkenes: A Catalytic Route to β^2 -Amino Acids, Aldehydes, and Alcohols

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In recent years, considerable effort has been devoted to catalytic asymmetric conjugate additions of dialkylzinc reagents to nitroalkenes.¹ Significant progress has been made using Cu(I) catalysts with phosphorus-based chiral ligands, cumulating enantioselectivities up to 96% for cyclic nitroalkenes as reported by Hoveyda.² Acyclic nitroalkenes, however, constitute a challenging class of substrates, because of the much lower selectivities obtained so far. Furthermore, almost exclusively, Et₂Zn as an organometallic reagent has been used.¹ Methodologies that provide high enantioselectivities in this 1,4-addition are highly warranted because the produced nitroalkanes have a wide range of applications.³ This is due to the versatility of the nitro group, sometimes entitled a "chemical chameleon",4 that can be transformed into a range of other functional groups including amine, aldehyde, or acid moieties.⁵ We envisioned that enantioselective 1,4-addition to acetal substituted nitroalkenes⁶ could provide an attractive route to β^2 -amino acids and derivatives, which are important building blocks in the synthesis of natural products, β -peptides, and pharmaceuticals (Scheme 1).⁷

We wish to report here that by using phosphoramidite ligand **L1**, developed in our laboratory,⁸ for the first time, enantioselectivities up to the 98% level for acyclic nitroalkenes are obtained. Furthermore, the use of acyclic substrates with different alkylzinc reagents provides a catalytic enantioselective route to (functionalized) β^2 -amino aldehydes, acids, and alcohols.

Encouraged by the results obtained with our one-pot multisubstrate screening procedure for copper-phosphoramidite catalysts,^{1e} various acetal substituted nitropropenes were examined. The synthesis is based on a transacetalization of dimethoxynitropropene (**1a**), prepared via a Henry reaction on multigram scale starting with commercially available dimethoxyacetaldehyde (Scheme 2). Nitroalkenes **1b** and **1c** were obtained in good yield; the (nonoptimized) low yield for **1d** can be explained by the low solubility of the diol under the reaction conditions. In preliminary experiments, we screened four different copper-phosphoramidite catalysts, based on ligands **L1–L4**, with respect to their ability to induce enantioselectivity in the 1,4-addition reaction of diethylzinc with nitropropene acetals **1a–d** (Table 1).

As can be seen from entries 1-3 in Table 1, bulky chiral substituents at the amine moiety of the ligand are necessary to reach high ee, and comparison of entries 1,4, 5,6, and 7,8 shows that **L1**, with (*S*)-BINOL as the diol part, gives in all cases better results than bisphenol-based ligand **L4**.⁹ It is also found that **L1** is able to generate high ee values (>90%) for nitroalkanes with acetals based on methanol (**2a**) and 2,2-dimethylpropanol (**2c**). The use of acetals based on pinacol (**2b**) and 2,2-diphenylpropanol (**2d**), the latter being very successful in the conjugate addition to cyclopentene-3,5-dione,¹⁰ leads to lower selectivities (<90%). Under optimized reaction conditions, that is, at -55 °C, and slower addition of the dialkylzinc reagent (over 1 min), nitropropene acetals **1a** and **1c**

Scheme 1. Proposed Route to β^2 -Amino Acids and Derivatives



Scheme 2. Synthesis of Nitropropene Acetals





Table 1. Diethylzinc Additions to Nitropropene Acetals^a

^{*a*} Conditions: 1.0 mmol of **1a**–**d**, 1.2 equiv of Et₂Zn in 2 mL of toluene; all reactions went to completion in 3 h. ^{*b*} Isolated yield. ^{*c*} Nonoptimized conditions; see Table 2 for optimized yield. ^{*d*} Determined by chiral GC. ^{*e*} Determined by chiral HPLC.

were reacted with various alkylzinc reagents using 1 mol % of the copper-phosphoramidite catalyst based on ligand L1 (Table 2).

Under these optimized conditions, the readily accessible nitropropene acetal **1a** proved to be the most suitable substrate resulting in very high enantioselectivities with simple aliphatic dialkylzinc reagents (entries 1,3,5). A functionalized zinc reagent can also be used, albeit with slightly lower selectivity (entry 6). We were **Table 2.** Conjugate Addition of Dialkylzinc Reagents to 1a and $1c^a$



^{*a*} Conditions: 1.0 mmol of **1a,1c**, 1.2 equiv of R_2Zn in 2 mL of toluene; all reactions went to completion (18 h for entries 3,4; 3 h for all others). ^{*b*} Isolated yield. ^{*c*} Determined by chiral GC.



delighted to isolate the Me₂Zn 1,4-addition product (*R*)-2e¹¹ with the excellent ee of 98%, as it contains a stereogenic center bearing a methyl substituent, which is a prominent feature in many natural products.¹² A major advantage of this asymmetric synthesis is that the obtained nitroalkanes can be easily converted into (protected) β^2 -amino aldehydes, alcohols, and acids (Scheme 3). Raney-nickelcatalyzed reduction of nitroalkane 2e, followed by Boc-protection, gives amino-acetal 3 which was deprotected to give β^2 -aminoaldehyde 4, a building block used in the total synthesis of cyclamenol A.^{13a} Because of the intermediate oxidation state of amino-acetal 3, oxidation under acidic conditions (H₅IO₆, 1% CrO₃) gives in a single step the corresponding N-Boc-protected β^2 -amino acid 6, used in the total synthesis of cryptophycins.¹⁴

The corresponding free amino acid has been isolated from human urine and *Iris tingitana*.¹⁵ Furthermore, β -amino-alcohol **5**, a starting material in the synthesis of β -methyl carbapenem antibiotics,¹⁶ was obtained by reduction of aldehyde **4**. Independent ee determinations of **3–6** confirmed that no racemization had occurred and all products were isolated with 98% ee.

To the best of our knowledge, this is the first example of a catalytic enantioselective route to these versatile building blocks and an important addition to existing routes, which make use of the chiral pool,¹⁷ (enzymatic) resolution,¹⁸ or chiral auxiliaries.¹¹

The practicality of this new catalytic route is demonstrated by (i) the synthesis of **2e** on gram scale by starting with 10 mmol of **1a**, resulting in yields ranging from 86 to 91%,¹⁹ and (ii) the few efficient steps that are needed to obtain the corresponding β^2 -amino compounds (Scheme 3), in particular, the aldehydes which are usually obtained via consecutive reduction and oxidation of the acids.¹⁴ Together with the rhodium-catalyzed asymmetric hydrogenation of β -dehydroamino acids using phosphoramidite ligands reported recently by our group,²⁰ both kinds of β -amino acids (β^2 -and β^3 -substituted) can be obtained using the same class of monodentate phosphoramidite ligands.

Acknowledgment. This project was funded by the National Research School Combination Catalysis (NRSCC).

Supporting Information Available: Experimental procedures and spectral and analytical data for reaction products (PDF). This material is available free of charge via the Internet at http://pubs.acs.org.

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JA029817D