

Available online at www.sciencedirect.com



Tetrahedron: Asymmetry 15 (2004) 2875-2880

Tetrahedron: Asymmetry

# Advanced procedure for the enzymatic ring opening of unsaturated alicyclic β-lactams

Enikő Forró and Ferenc Fülöp\*

Institute of Pharmaceutical Chemistry, University of Szeged, H-6701 Szeged, PO Box 121, Hungary

Received 29 April 2004; accepted 18 May 2004 Available online 29 July 2004

Abstract—Enantiopure  $\beta$ -amino acids 1a–4a and  $\beta$ -lactams 1b–4b were prepared simultaneously through the lipolase-catalysed enantioselective ring opening of unsaturated racemic  $\beta$ -lactams (±)-1-(±)-4. High enantioselectivities (E > 200) were observed when the reactions were performed with 1 equiv of water in *i*Pr<sub>2</sub>O at 70 °C. The resolved (1*R*,2*S*)-amino acids (yield  $\geq 45\%$ ) and (1*S*,5*R*)-, (1*S*,6*R*)- and (1*S*,8*R*)-lactams (yield  $\geq 47\%$ ) could be easily separated. The ring opening of lactam enantiomers 1b–4b with 18% HCl afforded the corresponding  $\beta$ -amino acid hydrochlorides 1c·HCl–4c·HCl (ee >95%).

© 2004 Elsevier Ltd. All rights reserved.

### 1. Introduction

Interest in cyclic  $\beta$ -amino acids and  $\beta$ -lactams has greatly increased over the past few years and has become a hot topic in synthetic and medicinal chemistry.<sup>1-4</sup> Some cyclic  $\beta$ -amino acids themselves exhibit biological activity (e.g., cispentacin, with antibacterial activity;<sup>5,6</sup> or PLD-118, with antifungal activity, which is currently under clinical investigation).<sup>7</sup> Such compounds can be used as precursors of  $\beta$ -lactams;<sup>8,9</sup> as building blocks for the synthesis of modified peptides with increased activity and stability<sup>10,11</sup> or in drug research.<sup>12–14</sup> Cyclic  $\beta$ -lactams are also important as intermediates in the synthesis of  $\beta$ -amino acids,<sup>15</sup> short peptide segments,<sup>16</sup> taxoid antitumour agents,<sup>17</sup> alkaloids<sup>18</sup> and heterocycles of biological and medicinal importance.<sup>19</sup>

A number of enzymatic syntheses of  $\beta$ -amino acids or their derivatives and  $\beta$ -lactams in enantiomerically pure form have been elaborated over the past few years.<sup>20–28</sup> We have recently developed a direct enzymatic method for the preparation of enantiopure valuable  $\beta$ -amino acids (e.g., cispentacin) through the lipase-catalysed enantioselective hydrolysis of  $\beta$ -lactams in an organic solvent.<sup>29,30</sup>

Herein, we report an easy and efficient lipase-catalysed enantioselective ring opening of racemic 6-azabicyclo[3.2.0]hept-3-en-7-one,  $(\pm)$ -1, 7-azabicyclo[4.2.0]oct-4-en-8-one  $(\pm)$ -2, 7-azabicyclo[4.2.0]oct-3-en-8-one,  $(\pm)$ -3 and 9-azabicyclo[6.2.0]dec-4-en-10-one,  $(\pm)$ -4, in an organic solvent.

### 2. Results and discussion

# 2.1. Syntheses of $(\pm)-1-(\pm)-4$

The racemic  $\beta$ -lactams 1–4 were prepared by the cycloaddition of chlorosulfonyl isocyanate (O=C=N–SO<sub>2</sub>Cl; CSI) to the corresponding cycloalkadiene 5–8 (Scheme 1).<sup>1,31–34</sup> The 1,2-dipolar cycloaddition of CSI to 1,2cyclopentadiene 5 and 1,3-cyclohexadiene 6 takes place regioselectively, in accordance with the Markovnikov orientation,<sup>1,31,32</sup> resulting in  $\beta$ -lactams (±)-1 and (±)-2.



Scheme 1.

<sup>\*</sup> Corresponding author. Tel.: +36-62-545564; fax: +36-62-545705; e-mail: fulop@pharma.szote.u-szeged.hu

#### 2.2. Lipase-catalysed enantioselective ring opening of $(\pm)-1-(\pm)-4$

We previously resolved model compounds  $(\pm)$ -2 and  $(\pm)$ -3 through an indirect enzymatic method: lipase PS (lipase from Pseudomonas cepacia)-catalysed enantioselective acylation of the N-hydroxymethylated derivatives in acetone at room temperature (E > 200, yield<sub>alcohol</sub>  $\ge 32\%$ , yield<sub>ester</sub>  $\geq$  34%),<sup>21</sup> and through a direct enzymatic method: Novozym 435 (lipase B from Candida antarctica, immobilised on a macroporous acrylic resin)-catalysed enantioselective ring opening alcoholysis in disporopyl ether at 60 °C (E > 200, yield<sub>β-lactam</sub>  $\geq 39\%$ , yield<sub>β-amino acid</sub>  $\leq 11\%$ ).<sup>25</sup> We also recently reported a simple and efficient direct method for the enantioselective ring opening of unactivated  $\beta$ -lactams in an organic medium: lipolase (lipase B from C. antarctica, produced by submerged fermentation of a genetically modified Aspergillus oryzae microorganism and adsorbed on a macroporous resin) proved to be applicable for the enantioselective ring opening of 6-azabicyclo[3.2.0]heptan-7-one and the homologues,<sup>29,30</sup> when the reactions were performed with H<sub>2</sub>O (1 equiv) in diisopropyl ether at 60 and 70 °C, respectively (E > 200, yield<sub> $\beta$ -lactam</sub>  $\geq 36\%$ , yield<sub> $\beta$ -amino acid</sub>  $\geq 43\%$ ). These results<sup>29,30</sup> on the lipasecatalysed enantioselective hydrolysis of  $\beta$ -lactams suggested the possibility of the enantioselective ring opening of  $(\pm)$ -1– $(\pm)$ -4 with H<sub>2</sub>O (1 equiv) in diisopropyl ether at high temperature (Scheme 2).

We started the ring-opening experiments on  $(\pm)$ -2 with lipolase at 60 °C, but also tested the reactions at 25, 40 and 70 °C. In all cases, the enantioselectivities were high (E > 200) while the reaction rates gave the following temperature dependence: after 16 h, the conversion was 10% at 25 °C, 30% at 40 °C and 49% at 60 °C, while it was 50% at 70 °C after 5 h. In spite of the high temperature (70 °C), the enzyme did not lose any significant activity. Also it could be reused successfully for a second ring-opening reaction of  $(\pm)$ -2 (conversion 48% after 5 h, at 70 °C).

COOH Lipolase *i*Pr<sub>2</sub>O 70 °C (±)-1, n = 1 (1*R*,2*S*)-**1a**, n = 1 (1S,5R)-1b, n = 1 (1*R*,2*S*)-**2a**, n = 2 (±)-2, n = 2 (1*S*,6*R*)-2b, n = 2 COOH Lipolase *i*Pr<sub>2</sub>O 70 °C (±)-3. n = 1 (1R,2S)-3a, n = 1 (1S,6R)-3b, n = 1 (±)-4, n = 2 (1*R*,2*S*)-**4a**, n = 2 (1S,8R)-4b, n = 2



Even though the E value in the lipolase-catalysed ring opening of  $(\pm)$ -2 with H<sub>2</sub>O (0 or 1 equiv) at 70 °C is excellent (E > 200), in order to enhance the reaction rate (conversion 43-46% after 4 h), several nucleophiles (1 equiv of 2-octanol, Et<sub>3</sub>N and N,N-diisopropylethylamine) were also tested for the ring opening of  $(\pm)$ -2. Since no significant changes in enantioselectivity (E > 200) or reaction rate (conversion 45–46% after 4 h) were observed, we concluded that the water present in the reaction medium (<0.1%) or in the enzyme preparation (<5%) was responsible for the lactam ring opening.

On the basis of the preliminary results, the gram-scale resolutions of  $(\pm)$ -1– $(\pm)$ -4 were performed with 1 equiv of water in the presence of lipolase in diisopropyl ether at 70 °C. The products were characterised by excellent enantiomeric excesses at 50-51% conversion. The results are reported in Table 1.

Evans et al.27 investigated the enzymatic ring-opening reactions of  $(\pm)$ -6-azabicyclo[3.2.0]hept-3-en-7-one  $(\pm)$ -1 and found that ENZA-1 (Rhodococcus equi NCIB 40213), catalysed the enantioselective ring opening of this compound in water. After two consecutive incubations of the lactam with *Rhodococcus equi*, the (1R,5S)-

Table 1. Lipolase-catalysed ring opening of  $(\pm)$ -1– $(\pm)$ -4

	Time (h)	Conv. (%)	Ε	β-Lactam 1b–4b				β-Amino acid 1a–4a			
				Yield <sup>a</sup> (%)	Isomer	Ee <sup>b</sup> (%)	$\left[\alpha\right]_{\mathrm{D}}^{25}$	Yield <sup>a</sup> (%)	Isomer	Ee <sup>c</sup> (%)	$\left[\alpha\right]_{\mathrm{D}}^{25}$
(±)-1	5	51	>200	48	1S,5R	99	-34.8 <sup>d</sup>	45	1 <i>R</i> ,2 <i>S</i>	96	+96.7 <sup>e</sup>
(±)- <b>2</b>	5	50	>200	48	1 <i>S</i> ,6 <i>R</i>	99	$+161.1^{d,f}$	46	1R, 2S	98	+121.1 <sup>g,h</sup>
(±)- <b>3</b>	4.5	50	>200	48	1 <i>S</i> ,6 <i>R</i>	99	-29.1 <sup>d,i</sup>	45	1R, 2S	99	-38.8 <sup>g,j</sup>
(±)- <b>4</b>	7	51	>200	47	1 <i>S</i> ,8 <i>R</i>	99	-24.9 <sup>k</sup>	46	1 <i>R</i> ,2 <i>S</i>	95	+23.9 <sup>g</sup>

<sup>a</sup> Yield 100% at 50% conversion.

<sup>b</sup>According to GC.

<sup>c</sup> Determined by GC [after double derivatization (i) diazomethane; (ii) acetic anhydride in the presence of 4-dimethylaminopyridine and pyridine]. <sup>d</sup> c 0.45, CHCl<sub>3</sub>.

<sup>e</sup> c 0.3, H<sub>2</sub>O.

 ${}^{f}[\alpha]_{D}^{25} = +164 \ (c \ 0.13, \ CHCl_{3}) \ for \ (1S, 6R)-2b.^{21}$ 

<sup>g</sup> c 0.5, H<sub>2</sub>O.

<sup>h</sup>  $[\alpha]_{D}^{25} = +120 \ (c \ 0.25, \ H_2O) \ for \ (1R,2S)-2a.^{21}$ 

 ${}^{i}[\alpha]_{D}^{25} = -26.3 \ (c \ 0.5, \ \text{CHCl}_{3}) \ \text{for} \ (1S, 6R) - 3b.^{21}$ 

 ${}^{j}[\alpha]_{D}^{25} = -36.2 \ (c \ 0.5, \ H_{2}O) \ for \ (1R,2S)-3a.^{21}$   ${}^{k} c \ 0.3, \ CHCl_{3}; \ [\alpha]_{D}^{25} = -19 \ (c \ 0.3, \ MeOH) \ for \ (1S,8R)-4b. \ [\alpha]_{D}^{25} = -20.2 \ (c \ 0.35, \ MeOH) \ for \ (1S,8R)-4b.^{23}$ 

lactam enantiomer was isolated with high ee. Lloyd et al.<sup>28</sup> reported the resolution of 7-azabicyclo[4.2.0]oct-4-en-8-one (±)-**2** using a lactamase (NCIMB 41042) in phosphate buffer at pH 7. Although they specified the formation of β-amino acid in the ring-opening reaction, they isolated and characterised only the (1*R*,6*S*)-β-lactam. Our method not only provides the β-lactams **1b**-**4b** with high ees (≥99%) in high yields (≥46%), but also the β-amino acids **1a**-**4a** with high ees (≥95%) in high yields (≥45%). It is important to stress that the lipolase-catalysed ring opening of (±)-**1** and (±)-**2** in organic solvent favours the opposite enantiomer than that in the case of NCIMB 41042 or NCIMB 40213 lactamase under aqueous conditions.

#### 2.3. Transformations of the enantiomers

The transformations involving the ring opening of  $\beta$ -lactams **1b–4b** with 18% aqueous HCl resulted in the enantiomers of the  $\beta$ -amino acid hydrochlorides **1c**·HCl–**4c**·HCl (Scheme 3). Treatment of amino acids **1a–4a** with 18% aqueous HCl resulted in enantiopure hydrochlorides **1a**·HCl–**4a**·HCl. The physical data on the enantiomers prepared are reported in Table 2.





The absolute configurations were proven by comparing the  $[\alpha]_D$  values with the literature data<sup>21,23,27</sup> (Table 1). The absolute configuration for **1a–4a** was (1*R*,2*S*), for **1b** it was (1*S*,5*R*), for **2b** and **3b** it was (1*S*,6*R*), while that for **4b** was (1*S*,8*R*).

### 3. Conclusions

An efficient and simple direct enzymatic method was developed for the enantioselective ring opening of racemic 6-azabicyclo[3.2.0]hept-3-en-7-one ( $\pm$ )-1, 7-azabicyclo[4.2.0]oct-4-en-8-one ( $\pm$ )-2, 7-azabicyclo[4.2.0]-oct-3en-8-one ( $\pm$ )-3 and 9-azabicyclo[6.2.0]dec-4-en-10-one ( $\pm$ )-4 with 1 equiv of H<sub>2</sub>O in *i*Pr<sub>2</sub>O at 70 °C (*E* >200). The enantioselective ring-opening reactions of unsaturated 1–4 reached 50% conversion in a much shorter time (4.5– 7 h) than that for their saturated analogues (141–249 h).<sup>29</sup> The lipolase-catalysed ring opening of ( $\pm$ )-1 and ( $\pm$ )-2 in organic solvent favours the opposite enantiomer than

Table 2. Physical data on enantiomers prepared

Amino acid hydrochloride	Ee (%)	$[\alpha]_{\mathrm{D}}^{25}$
1 <i>R</i> ,2 <i>S</i> -1a·HCl	99	+81.6 (c 0.3, H <sub>2</sub> O)
1 <i>S</i> ,2 <i>R</i> -1c·HCl	98	-80.2 (c 0.3, H <sub>2</sub> O)
1R,2S-2a·HCl	99	+121.7 (c 0.4, H <sub>2</sub> O)
1 <i>S</i> ,2 <i>R</i> -2c·HCl	99	-121.4 (c 0.4, H <sub>2</sub> O)
1R,2S-3a·HCl	99	-26 (c 0.25, H <sub>2</sub> O)
1 <i>S</i> ,2 <i>R</i> -3c·HCl	99	$+25.8 (c 0.4, H_2O)$
1R,2S-4a·HCl	95	$+14.2 (c 0.35, H_2O)$
1 <i>S</i> ,2 <i>R</i> -4c·HCl	98	-15.9 (c 0.3, H <sub>2</sub> O)

that in the case of lactamase present in the whole cells of *Rhodococcus globerulus* (NCIMB 41042) or *R. equi* in phosphate buffer at pH7. The  $\beta$ -lactam (yield = 48%) and  $\beta$ -amino acid (yield  $\geq 45\%$ ) products can be easily separated. Ring opening of the  $\beta$ -lactams **1b**–**4b** with 18% aqueous HCl resulted in the enantiomers of the  $\beta$ -amino acid hydrochlorides **1c**·HCl–**4c**·HCl (ee >95%). All the produced unsaturated enantiomers can be used for further valuable transformations (i.e., they can be used to prepare information-rich chiral scaffolds for elaboration into single enantiomer compounds libraries as the alkene functionality is amenable to a range of transformations.<sup>35</sup>

#### 4. Experimental

# 4.1. Materials and methods

Lipolase (lipase B from *C. antarctica*), produced by submerged fermentation of a genetically modified *A. oryzae* microorganism and adsorbed on a macroporous resin, was from Sigma-Aldrich (Catalog no L4777). Chlorosulfonyl isocyanate, and 1,3- and 1,4-cyclohexadiene were purchased from Aldrich. The solvents were of the highest analytical grade.

In a typical small-scale experiment, racemic 2-azetidinone (0.05 M solution) in diisopropyl ether (2 mL) was added to lipolase (50 mg/mL). Water (1 equiv) was added and the mixture was shaken at 70 °C. The progress of the reaction was followed by taking samples from the reaction mixture at intervals and analysing them by gas chromatography. The ee values for the unreacted  $\beta$ -lactam enantiomers were determined by gas chromatography on a Chromopak Chiralsil-Dex CB column (25 m) [140 °C for  $7 \min \rightarrow 190$  °C (rate of temperature rise 20 °C/min; 100 kPa), retention times (min): **1b**: 6.17 (antipode: 5.64); **2b**: 9.49 (antipode: 9.26); **3b**: 10.04 (antipode: 9.68); 4b: 11.95 (antipode: 11.85)], while the ee values for the ring-opened amino acids produced (during the preliminary experiments) were calculated by using *n*-hexadecane as an internal standard. The ee values for the isolated  $\beta$ -amino acid enantiomers were determined by using a gas chromatograph equipped with a chiral column after double derivatization with (i) diazomethane; (ii) acetic anhydride in the presence of 4-dimethylaminopyridine and pyridine (Chirasil-L-Val

column, 100 °C for 10 min  $\rightarrow$  160 °C, rate of temperature rise 10 °C/min; 30 kPa), retention times (min): **1a**: 18.2 (antipode: 18.38); (Chirasil-L-Val column, 120 °C for 4 min  $\rightarrow$  190 °C, rate of temperature rise 20 °C/min; 140 kPa), retention times (min): **2a**: 6.2 (antipode: 6.3); **3a**: 5.84 (antipode: 5.94); (CP-Chirasil-Dex CB column, 120 °C for 7 min  $\rightarrow$  160 °C, rate of temperature rise 10 °C/min; 30 kPa), retention times (min): **4a**: 36.92 (antipode: 37.45).

Optical rotations were measured with a Perkin–Elmer 341 polarimeter. <sup>1</sup>H and <sup>13</sup>C NMR spectra were recorded on a Bruker Avance DRX 400 spectrometer. Melting points were determined on a Kofler apparatus.

#### 4.2. Gram-scale resolution of 7-azabicyclo[4.2.0]oct-3-en-8-one, (±)-1

Crystalline racemic 1 (1g, 9.16 mmol) was dissolved in diisopropyl ether (40 mL). Lipolase (2 g, 50 mg/mL) and water (0.16 mL, 9.16 mmol) were added and the mixture shaken in an incubator shaker at 70 °C for 5 h. The reaction was stopped by filtering off the enzyme at 50% conversion (ee-1b = 99%). The solvent was evaporated off and the residue (1S,5R)-1b crystallized out  $\{0.4g,$ 48%; recrystallized from diisopropyl ether,  $[\alpha]_D^{25} = -34.8$  (*c* 0.45, CHCl<sub>3</sub>); mp 76–77 °C, lit.<sup>27b</sup> mp 76–77 °C; ee 99%}. The filtered-off enzyme was washed with distilled water  $(3 \times 15 \text{ mL})$  and the water evaporated off, yielding the crystalline  $\beta$ -amino acid (1*R*,2*S*)-1a {0.52 g, 45%;  $[\alpha]_{D}^{25} = +96.7 (c \ 0.3, H_2O); mp > 240 \,^{\circ}C$  with sublimation (recrystallized from methanol); ee = 96%}. When **1a** (0.1 g) was treated with 18% aqueous HCl (3 mL), (1R,2S)-1a·HCl was obtained  $\{0.11 \text{ g}, 85\%; [\alpha]_{\text{D}}^{25} =$ +81.6 (c 0.3, H<sub>2</sub>O); mp 178–183 °C, ee = 99%}.

<sup>1</sup>H NMR (400 MHz, D<sub>2</sub>O)  $\delta$  (ppm) for **1a**: 2.54–2.56 (1H, m, H-5<sub>A</sub>) 2.67–2.74 (1H, m, H-5<sub>B</sub>) 3.27 (1H, q, J = 8.4 Hz, H-1) 4.28–4.29 (1H, m, H-2) 5.80–5.81 (1H, m, H-3) 6.23–6.24 (1H, m, H-4). <sup>13</sup>C NMR (100.62 MHz, D<sub>2</sub>O)  $\delta$  (ppm) 35.3, 46.0, 56.5, 126.7, 139.6, 179.8. Analysis: calculated for C<sub>6</sub>H<sub>9</sub>NO<sub>2</sub>: C, 56.68; H, 7.13; N, 11.02; found: C, 56.51; H, 6.88; N, 10.91.

<sup>1</sup>H NMR (400 MHz, D<sub>2</sub>O)  $\delta$  (ppm) for **1a**·HCl: 2.76–2.80 (2H, m, H-5<sub>A</sub> and H-5<sub>B</sub>) 3.51–3.58 (1H, m, H-1) 4.46–4.48 (1H, m, H-2) 5.83–5.85 (1H, m, H-3) 6.28–6.29 (1H, m, H-4). <sup>13</sup>C NMR (100.62 MHz, D<sub>2</sub>O)  $\delta$  (ppm) 34.57, 43.91, 56.25, 126.21, 139.05, 175.9. Analysis: calculated for C<sub>6</sub>H<sub>9</sub>NO<sub>2</sub>·HCl: C, 44.05; H, 6.16; N, 8.56; found: C, 43.79; H, 6.26; N, 8.55.

<sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) for **1b**: 2.42–2.49 (1H, m, H-2<sub>A</sub>) 2.69–2.76 (1H, m, H-2<sub>B</sub>) 3.82–3.84 (1H, m, H-1) 4.50–4.51 (1H, m, H-5) 5.93–5.95 (1H, m, H-4) 6.01–6.03 (1H, m, H-3) 6.48 (1H, br s, NH). <sup>13</sup>C NMR (100.62 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 31.5, 54.0, 59.9, 131.3, 137.7, 173.0. Analysis: calculated for C<sub>6</sub>H<sub>7</sub>NO: C, 66.04; H, 6.47; N, 12.84; found: C, 66.12; H, 6.42; N, 12.98.

# 4.3. Gram-scale resolution of 7-azabicyclo[4.2.0]oct-4-en-8-one, (±)-2

Following the procedure described above, the reaction of racemic **2** (2 g, 16.26 mmol) and water (0.29 mL, 16.26 mmol) in diisopropyl ether (80 mL) in the presence of lipolase (4 g, 50 mg/mL) at 70 °C afforded the unreacted (1*S*,6*R*)-**2b** {0.96 g, 48%;  $[\alpha]_D^{25} = +161.1$  (*c* 0.45, CHCl<sub>3</sub>); mp 113–114 °C (recrystallized from diisopropyl ether), lit.<sup>21</sup> mp 106–108 °C; ee = 99%} and β-amino acid (1*R*,2*S*)-**2a** {1.02 g, 46%;  $[\alpha]_D^{25} = +121.1$  (*c* 0.5, H<sub>2</sub>O); mp 233–236 °C (recrystallized from water–acetone) lit.<sup>25</sup> mp 236–238 °C; ee = 98%} in 5 h. When **2a** (0.2 g) was treated with 18% aqueous HCl (3 mL), (1*R*,2*S*)-**2a** ·HCl was obtained {0.24 g, 96%;  $[\alpha]_D^{25} = +121.7$  (*c* 0.4, H<sub>2</sub>O); mp 190–212 °C (slow melting), ee = 99%}.

<sup>1</sup>H NMR (400 MHz, D<sub>2</sub>O)  $\delta$  (ppm) for **2a**: 1.84–2.18 (4H, m, 2×CH<sub>2</sub>) 2.72–2.77 (1H, m, H-1) 3.98–3.99 (1H, m, H-2) 5.73–6.14 (2H, m, CHCH). <sup>13</sup>C NMR (100.62 MHz, D<sub>2</sub>O)  $\delta$  (ppm) 22.1, 24.4, 42.4, 47.1, 122.0, 135.2, 181.2. Analysis: calculated for C<sub>7</sub>H<sub>11</sub>NO<sub>2</sub>: C, 59.56; H, 7.85; N, 9.92; found: C, 59.47; H, 7.88; N, 9.82.

<sup>1</sup>H NMR (400 MHz, D<sub>2</sub>O)  $\delta$  (ppm) for **2a**·HCl: 1.90– 2.20 (4H, m, 2×CH<sub>2</sub>) 3.04–3.08 (1H, m, H-1) 4.09 (1H, m, H-2) 5.73–6.16 (2H, m, CHCH). <sup>13</sup>C NMR (100.62 MHz, D<sub>2</sub>O)  $\delta$  (ppm) 20.9, 23.9, 40.8, 46.6, 121.3, 135.4, 179.5. Analysis: calculated for C<sub>7</sub>H<sub>11</sub>NO<sub>2</sub>·HCl: C, 47.33; H, 6.81; N, 7.89; found: C, 47.47; H, 6.79; N, 7.88.

<sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) for **2b**: 1.63–2.12 (4H, m, 2×CH<sub>2</sub>) 3.51 (1H, m, H-1) 4.02–4.04 (1H, m, H-6) 5.93–6.14 (2H, m, CHCH) 5.95 (1H, br s, NH). <sup>13</sup>C NMR (100.62 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 22.4, 22.5, 45.0, 50.5, 126.6, 135.2, 172.7. Analysis: calculated for C<sub>7</sub>H<sub>9</sub>NO: C, 68.27; H, 7.37; N, 11.37; found: C, 68.13; H, 7.32; N, 11.48.

# 4.4. Gram-scale resolution of 7-azabicyclo[4.2.0]oct-3en-8-one, (±)-3

Following the procedure described above, the reaction of racemic **3** (2 g, 16.26 mmol) and water (0.29 mL, 16.26 mmol) in diisopropyl ether (80 mL) in the presence of lipolase (4 g, 50 mg/mL) at 70 °C afforded the unreacted (1*S*,6*R*)-**3b** (0.96 g, 48%;  $[\alpha]_D^{25} = -29.1$  (*c* 0.45, CHCl<sub>3</sub>); mp 152–154 °C (recrystallized from diisopropyl ether), lit.<sup>21</sup> mp 152–153 °C; ee = 99%) and β-amino acid (1*R*,2*S*)-**3a** {1 g, 45%;  $[\alpha]_D^{25} = -38.8$  (*c* 0.5, H<sub>2</sub>O); mp 232–234 °C (recrystallized from water–acetone) lit.<sup>25</sup> mp 233–235 °C; ee = 99%} in 4.5 h. When **3a** (0.2 g) was treated with 18% aqueous HCl (3 mL), (1*R*,2*S*)-**3a**·HCl was obtained {0.22 g, 88%;  $[\alpha]_D^{25} = -26$  (*c* 0.25, H<sub>2</sub>O); mp 191–220 (slow melting), ee = 99%}.

<sup>1</sup>H NMR (400 MHz, D<sub>2</sub>O)  $\delta$  (ppm) for **3a**: 2.23–2.51 (4H, m, 2×CH<sub>2</sub>) 2.74–2.78 (1H, m, H-1) 3.77–3.80 (1H, m, H-2) 5.64–5.84 (2H, m, CHCH). <sup>13</sup>C NMR (100.62 MHz, D<sub>2</sub>O)  $\delta$  (ppm) 25.2, 27.8, 41.3, 47.5, 122.6,

126.7, 181.3. Analysis: calculated for  $C_7H_{11}NO_2$ : C, 59.56; H, 7.85; N, 9.92; found: C, 59.52; H, 8.07; N, 9.92.

<sup>1</sup>H NMR (400 MHz, D<sub>2</sub>O)  $\delta$  (ppm) for **3a**·HCl: 2.27–2.55 (4H, m, 2×CH<sub>2</sub>) 3.07–3.11 (1H, m, H-1) 3.90–3.91 (1H, m, H-2) 5.66–5.83 (2H, m, CHCH). <sup>13</sup>C NMR (100.62 MHz, D<sub>2</sub>O)  $\delta$  (ppm) 24.5, 27.9, 39.8, 46.9, 122.8, 125.9, 176.8. Analysis: calculated for C<sub>7</sub>H<sub>11</sub>NO<sub>2</sub>·HCl: C, 47.33; H, 6.81; N, 7.89; found: C, 47.59; H, 6.82; N, 7.93.

<sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) for **3b**: 1.59–2.11 (4H, m, 2×CH<sub>2</sub>) 3.49–3.51 (1H, m, H-1) 4.02–4.04 (1H, m, H-6) 5.94–6.17 (2H, m, CHCH) 6.01 (1H, br s, NH). <sup>13</sup>C NMR (100.62 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 22.0, 27.8, 47.7, 48.8, 124.8, 126.8, 171.2. Analysis: calculated for C<sub>7</sub>H<sub>9</sub>NO: C, 68.27; H, 7.37; N, 11.37; found: C, 68.22; H, 7.32; N, 11.40.

# 4.5. Gram-scale resolution of 9-azabicyclo[6.2.0]dec-4en-10-one, $(\pm)$ -4

Following the procedure described above, the reaction of racemic **4** (1 g, 6.61 mmol) and water (0.12 mL, 6.61 mmol) in diisopropyl ether (40 mL) in the presence of lipolase (2 g, 50 mg/mL) at 70 °C afforded the unreacted (1*S*,8*R*)-**4b** (0.47 g, 47%;  $[\alpha]_D^{25} = -24.9$  (*c* 0.4, CHCl<sub>3</sub>);  $[\alpha]_D^{25} = -19.9$  (*c* 0.3, MeOH); mp 136–140 °C (recrystallized from diisopropyl ether), lit.<sup>23</sup> mp 117–119 °C; ee = 99%) and β-amino acid (1*R*,2*S*)-**4a** {0.51 g, 46%;  $[\alpha]_D^{25} = +23.9$  (*c* 0.3, H<sub>2</sub>O); mp 218–220 °C (recrystallized from water–acetone); ee = 95%} in 7 h. When **4a** (0.1 g) was treated with 18% aqueous HCl (3 mL), (1*R*,2*S*)-**4a** HCl was obtained {0.1 g, 82%;  $[\alpha]_D^{25} = +14.2$  (*c* 0.35, H<sub>2</sub>O); mp 194–205 °C, ee = 95%}.

<sup>1</sup>H NMR (400 MHz, D<sub>2</sub>O)  $\delta$  (ppm) for **4a**: 1.81–2.58 (8H, m, 4×CH<sub>2</sub>) 2.79 (1H, m, H-1) 3.70–3.73 (1H, m, H-2) 5.72–5.73 (2H, m, CHCH). <sup>13</sup>C NMR (100.62 MHz, D<sub>2</sub>O) (ppm) 22.9, 24.7, 27.2, 29.8, 45.9, 52.5, 129.6, 129.9, 181.8. Analysis: calculated for C<sub>9</sub>H<sub>15</sub>NO<sub>2</sub>: C, 63.88; H, 8.93; N, 8.28; found: C, 63.99; H, 8.81; N, 8.08.

<sup>1</sup>H NMR (400 MHz, D<sub>2</sub>O)  $\delta$  (ppm) for **4a**·HCl: 1.65– 2.35 (8H, m, 4×CH<sub>2</sub>) 2.94 (1H, m, H-1) 3.67 (1H, m, H-2) 5.55–5.62 (2H, m, CHCCH). <sup>13</sup>C NMR (100.62 MHz, D<sub>2</sub>O)  $\delta$  (ppm) 23.0, 23.7, 26.9, 30.2, 44.6, 51.8, 129.6, 130.9, 177.0. Analysis: calculated for C<sub>9</sub>H<sub>15</sub>NO<sub>2</sub>·HCl: C, 52.56; H, 7.84; N, 6.81; found: C, 52.32; H, 7.99; N, 6.79.

<sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) for **4b**: 1.83–2.02 and 2.30–2.37 (8H, m, 4×CH<sub>2</sub>) 3.21–3.24 (1H, m, H-1) 3.74–3.78 (1H, m, H-8) 5.58–5.62 (2H, m, CHCH) 6.16 (1H, br s, NH). <sup>13</sup>C NMR (100.62 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 23.3, 24.4, 24.9, 31.3, 54.0, 54.6, 130.9, 131.6, 172.2. Analysis: calculated for C<sub>9</sub>H<sub>13</sub>NO: C, 71.49; H, 8.67; N, 9.26; found: C, 71.62; H, 8.67; N, 9.23.

# 4.6. Ring opening of $\beta$ -lactam enantiomers 1b–4b with aqueous HCl

(1S,5R)-1b (0.1 g, 0.92 mmol), (1S,6R)-2b (0.1 g, 0.1 g)1.63 mmol), (1S,6R)-3b (0.1 g, 1.63 mmol) or (1S,8R)-4b (0.1 g, 0.67 mmol) was dissolved in 18% HCl (7 mL) and the solution refluxed for 3h. The solvent was then evaporated off and the product recrystallized from ethanol-diethyl ether, which afforded white crystals of (1*S*,2*R*)-1c·HCl {0.13 g, 87%,  $[\alpha]_D^{25} = +81.6$  (*c* 0.3, H<sub>2</sub>O); (168–171 °C); ee 99%}, (1*S*,2*R*)-2c·HCl {0.13 g, 90%;  $[\alpha]_{D}^{25} = -121.4 \ (c \ 0.4, \ H_2O); \ slow \ melting (192-209 \ ^{\circ}C);$ ee 99%}, (1*S*,2*R*)-3c·HCl {0.11 g, 76%,  $[\alpha]_{\rm D}^{25} = +25.8$  (*c* 0.4, H<sub>2</sub>O); slow melting (180–224 °C); ee 99%}, and (1*S*,2*R*)-4c HCl {0.1 g, 73%;  $[\alpha]_{D}^{25} = -15.9$  (*c* 0.3, H<sub>2</sub>O); (a slowly crystallizing oil); ee 98%}. The <sup>1</sup>H NMR (400 MHz, D<sub>2</sub>O)  $\delta$  (ppm) data for (1S,2R)-1c·HCl, (1S,2R)-2c·HCl, (1S,2R)-3c·HCl and (1S,2R)-4c·HCl are similar to those for (1R,2S)-1a·HCl, (1R,2S)-2a·HCl, (1R,2S)-3a·HCl and (1R,2S)-4a·HCl. Anal. found for (1S,2R)-1c·HCl: C, 44.11; H, 6.11; N, 8.27. Anal. found for (1S,2R)-2c·HCl: C, 47.22; H, 6.84; N, 7.87. Anal. found for (1S,2R)-3c·HCl: C, 47.49; H, 6.77; N, 7.65. Anal. found for (1S,2R)-4c·HCl: C, 52.51; H, 7.96; N, 6.67.

#### Acknowledgements

The authors acknowledge the receipt of OTKA grants TS 040888 and T 046440, FKFP grant 0115/2001 and a Békésy Fellowship for EF (grant no. 181/2002).

#### **References and notes**

- Besada, P.; González-Moa, M. J.; Terán, C.; Santana, L.; Uriarte, E. Synthesis 2002, 16, 2445–2449.
- González-Moa, M. J.; Besada, P.; Teijeira, M.; Terán, C.; Uriarte, E. Synthesis 2004, 4, 543–548.
- Gardiner, J.; Anderson, K. H.; Downard, A.; Abell, A. D. J. Org. Chem. 2004, 69, 3375–3382.
- 4. Abell, A. D.; Gardiner, J. Org. Lett. 2002, 4, 3663-3666.
- Fülöp, F. In *Studies in Natural Product Chemistry*; Atta-ur-Rahman, Ed.; Elsevier Science Publishers, 2000; Vol. 22, pp 273–306.
- 6. Fülöp, F. Chem. Rev. 2001, 101, 2181–2204, and references cited therein.
- Sorbera, L. A.; Castaner, J.; Bozzo, J. Drugs Fut. 2002, 27, 1049–1055.
- Krishnaswamy, D.; Govande, V. V.; Gumaste, V. K.; Bhawal, B. M.; Deshmukh, A. R. A. S. *Tetrahedron* 2002, 58, 2215–2225.
- Sleeman, M. C.; MacKinnon, C. H.; Hewitson, K. S.; Schofield, C. J. *Bioorg. Med. Chem. Lett.* 2002, *12*, 597– 599.
- Steer, D. L.; Lew, R. A.; Perlmutter, P.; Smith, A. I.; Aguilar, M. I. Curr. Med. Chem. 2002, 9, 811–822.
- Brashear, K. M.; Hunt, C. A.; Kucer, B. T.; Duggan, M. E.; Hartman, G. D.; Rodan, G. A.; Rodan, S. B.; Leu, C.; Prueksaritanont, T.; Fernandez-Metzler, C.; Barrish, A.; Homnick, C. F.; Hutchinson, J. H.; Coleman, P. J. *Bioorg. Med. Chem. Lett.* 2002, *12*, 3483–3486.

- 12. Mittendorf, J.; Benet-Buchholz, J.; Fey, P.; Mohrs, K. H. Synthesis 2003, 136–140.
- Mittendorf, J.; Kunisch, F.; Matzke, M.; Militzer, H. C.; Schmidth, A.; Schönfeld, W. *Bioorg. Med. Chem. Lett.* 2003, 13, 433–436.
- Porter, E. A.; Wang, X. F.; Lee, H. S.; Weisblum, B.; Gellman, S. H. Nature 2000, 404, 565.
- Palomo, C.; Aizpurua, J. M.; Ganboa, I.; Oiarbide, M. Synlett 2001, 1813–1826.
- Palomo, C.; Ganboa, I.; Oiarbide, M.; Sciano, G. T.; Miranda, J. L. Arkivoc 2002, v, 8–16.
- Juaristi, E. Enantioselective Synthesis of β-Amino Acids; Wiley-VHC: New York, 1997.
- Wasserman, H. H.; Matsuyama, H.; Robinson, R. P. *Tetrahedron* 2002, 58, 7177–7190.
- Alcaide, B.; Almendros, P.; Alonso, J. M.; Aly, M. F.; Pardo, C.; Saez, E.; Torres, M. R. J. Org. Chem. 2002, 67, 7004–7013.
- 20. Csomós, P.; Kanerva, L. T.; Bernáth, G.; Fülöp, F. *Tetrahedron: Asymmetry* **1996**, *7*, 1789–1796.
- 21. Kámán, J.; Forró, E.; Fülöp, F. Tetrahedron: Asymmetry 2000, 11, 1593–1600.
- 22. Fülöp, F.; Palkó, M.; Kámán, J.; Lázár, L.; Sillanpää, R. *Tetrahedron: Asymmetry* **2000**, *11*, 4179–4187.
- 23. Forró, E.; Árva, J.; Fülöp, F. *Tetrahedron: Asymmetry* **2001**, *12*, 643–649.
- Gyarmati, Z. C.; Liljeblad, A.; Rintola, M.; Bernáth, G.; Kanerva, L. T. *Tetrahedron: Asymmetry* 2003, 14, 3805– 3814.
- Park, S.; Forró, E.; Grewal, H.; Fülöp, F.; Kazlauskas, R. J. Adv. Synth. Catal. 2003, 345, 986–995.

- 26. Forró, E.; Fülöp, F. Mini Rev. Org. Chem. 2004, 1, 93-102.
- (a) Evans, C.; McCague, R.; Roberts, S. M.; Sutherland, A. G.; Wisdom, R. J. Chem. Soc., Perkin Trans. 1 1991, 2276–2277; (b) Evans, C. T.; Roberts, S. M.; Sutherland, A. G. PCT Int. Appl. (1992), WO 92/18477; Chem. Abstr. 1993, 118, 168892.
- 28. Lloyd, R. C.; Lloyd, M. C.; Smith, M. E. B.; Holt, K. E.; Swift, J. P.; Keene, P. A.; Taylor, S. J. C.; McCague, R. *Tetrahedron* **2004**, *60*, 717–728. In this article, the regioisomer of ( $\pm$ )-**2** was given as the only product formed on CSI cycloaddition to the 1,3-cyclohexadiene **4**; however, the formulae relating to the enzymatic step are represented erroneously. Although the formation of β-amino acid was specified in the ring-opening reaction, only the (1*R*,6*S*)-βlactam was isolated and characterised (ee >95%, yield = 31%); no data are available in the Experimental part with regard to the β-amino acid.
- 29. Forró, E.; Fülöp, F. Org. Lett. 2003, 5, 1209-1212.
- 30. Forró, E.; Fülöp, F. Tetrahedron: Asymmetry 2004, 15, 573–575.
- 31. Malplass, J. R.; Tweddle, N. J. J. Chem. Soc., Perkin Trans. 1 1977, 874–884.
- Furet, P.; Garcia-Echeverria, C.; Gay, B.; Schoepfer, J.; Zeller, M.; Rahuel, J. J. Med. Chem. 1999, 42, 2358–2363.
- Singh, R.; Cooper, R. D. G. Tetrahedron 1994, 50, 12049– 12064.
- Parsons, P. J.; Camp, N. P.; Underwood, J. M.; Harvey, D. M. J. Chem. Soc., Chem. Commun. 1995, 1461–1462.
- Taylor, S. J. C.; Keene, P. A. PCT Int. Appl. (2000), WO 00/58283; Chem. Abstr. 2000, 133, 266652.